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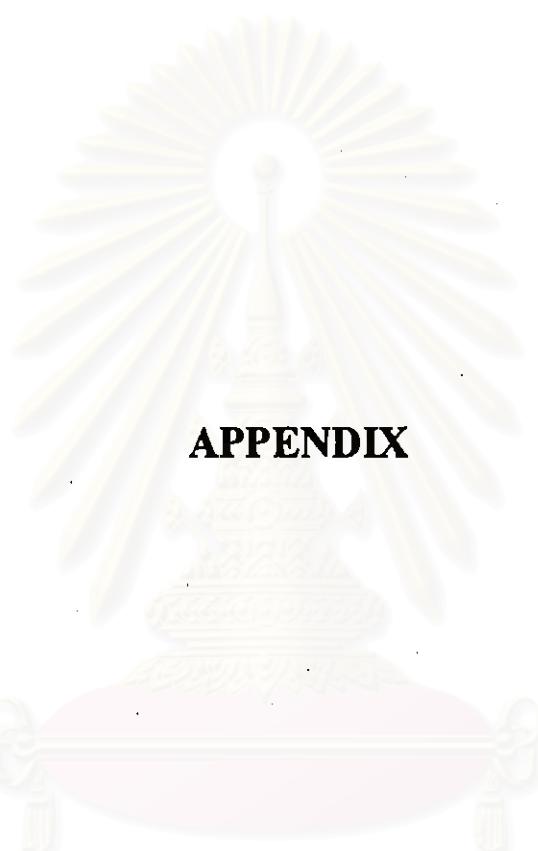
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APPENDIX

สถาบันวิทยบริการ
จุฬาลงกรณ์มหาวิทยาลัย

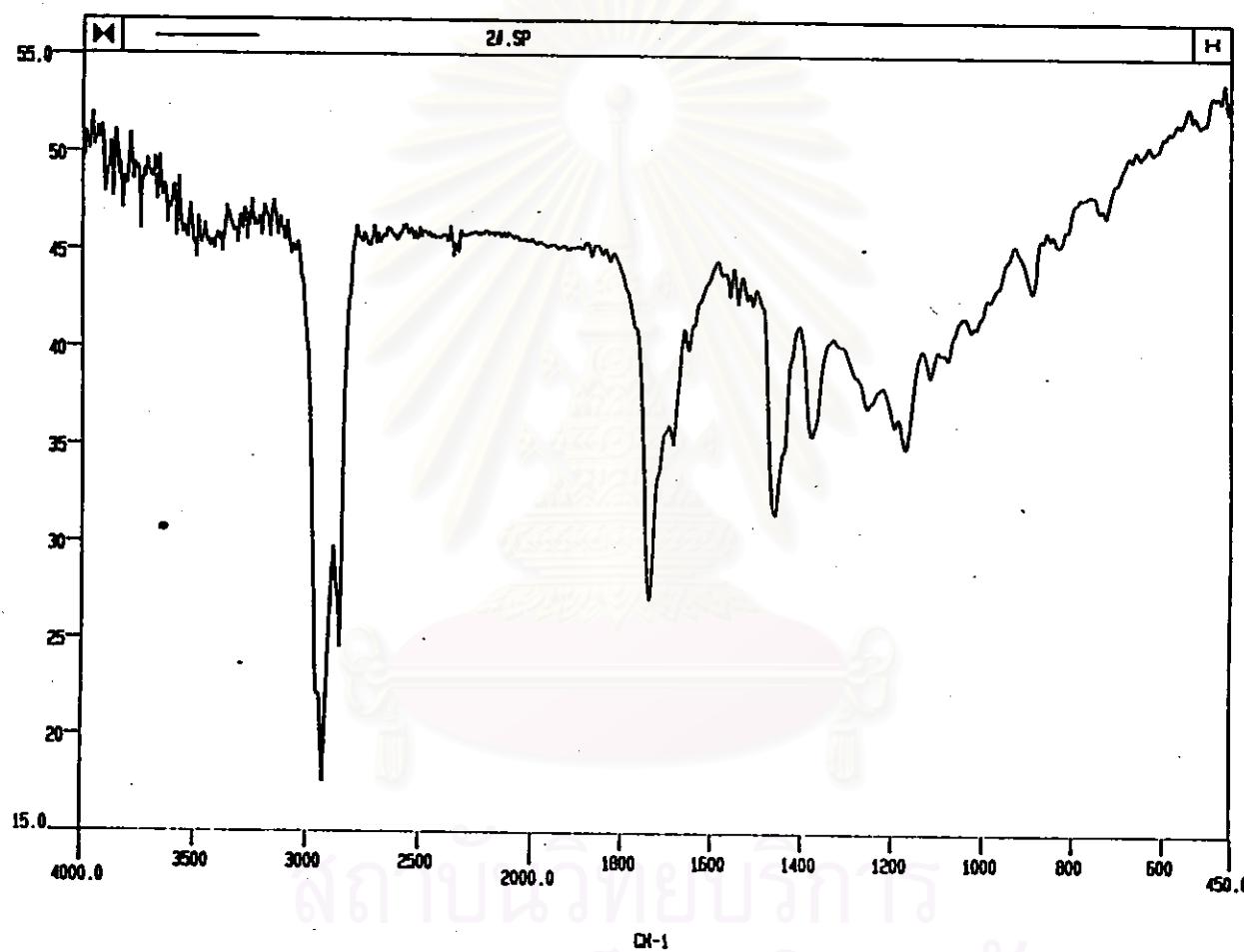


Figure 3 The IR spectrum of mixture I

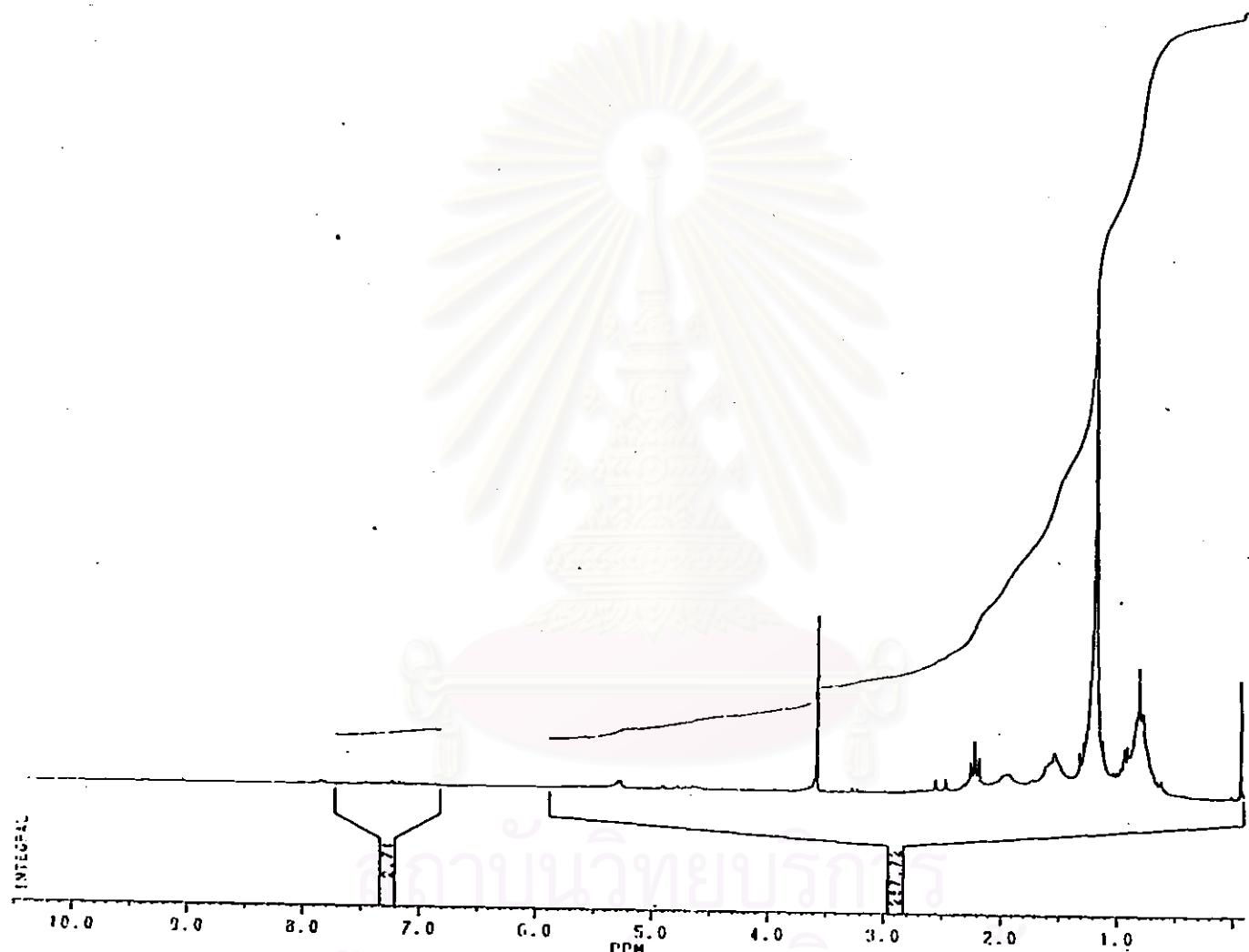


Figure 4 The ${}^1\text{H}$ -NMR spectrum of mixture I

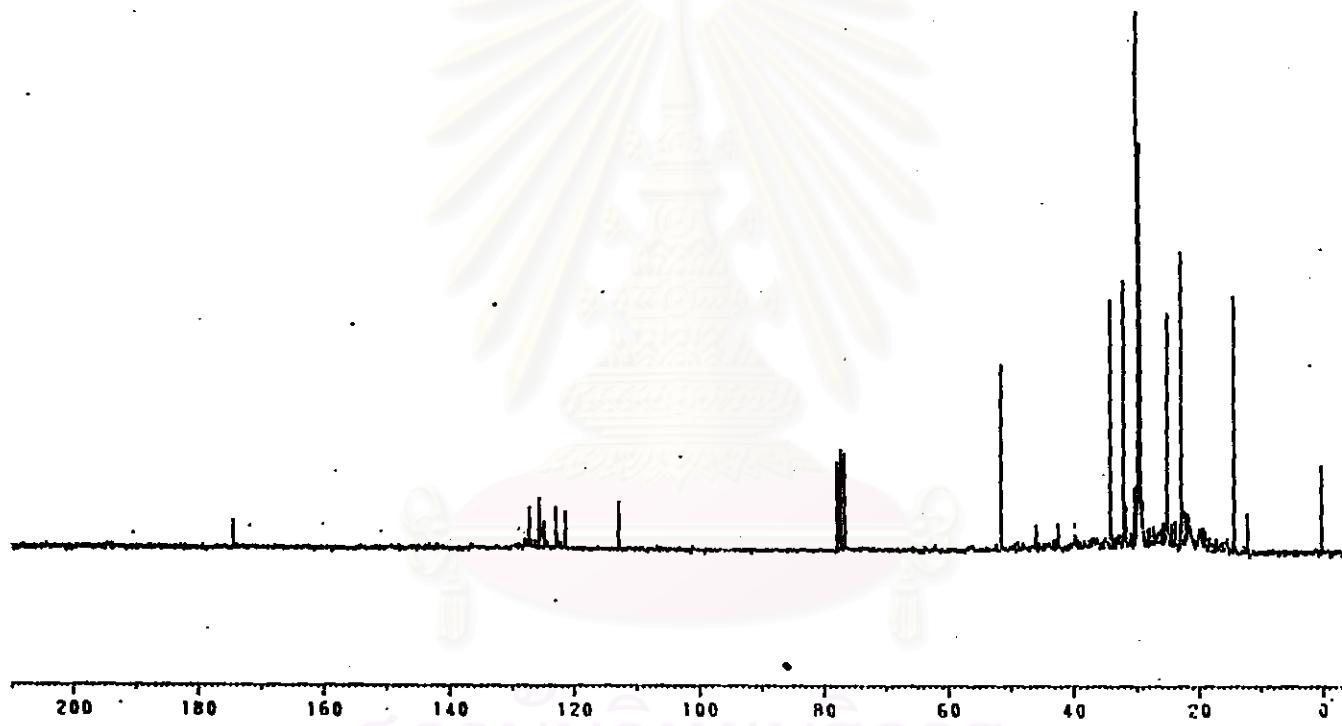


Figure 5 The ^{13}C -NMR spectrum of mixture 1

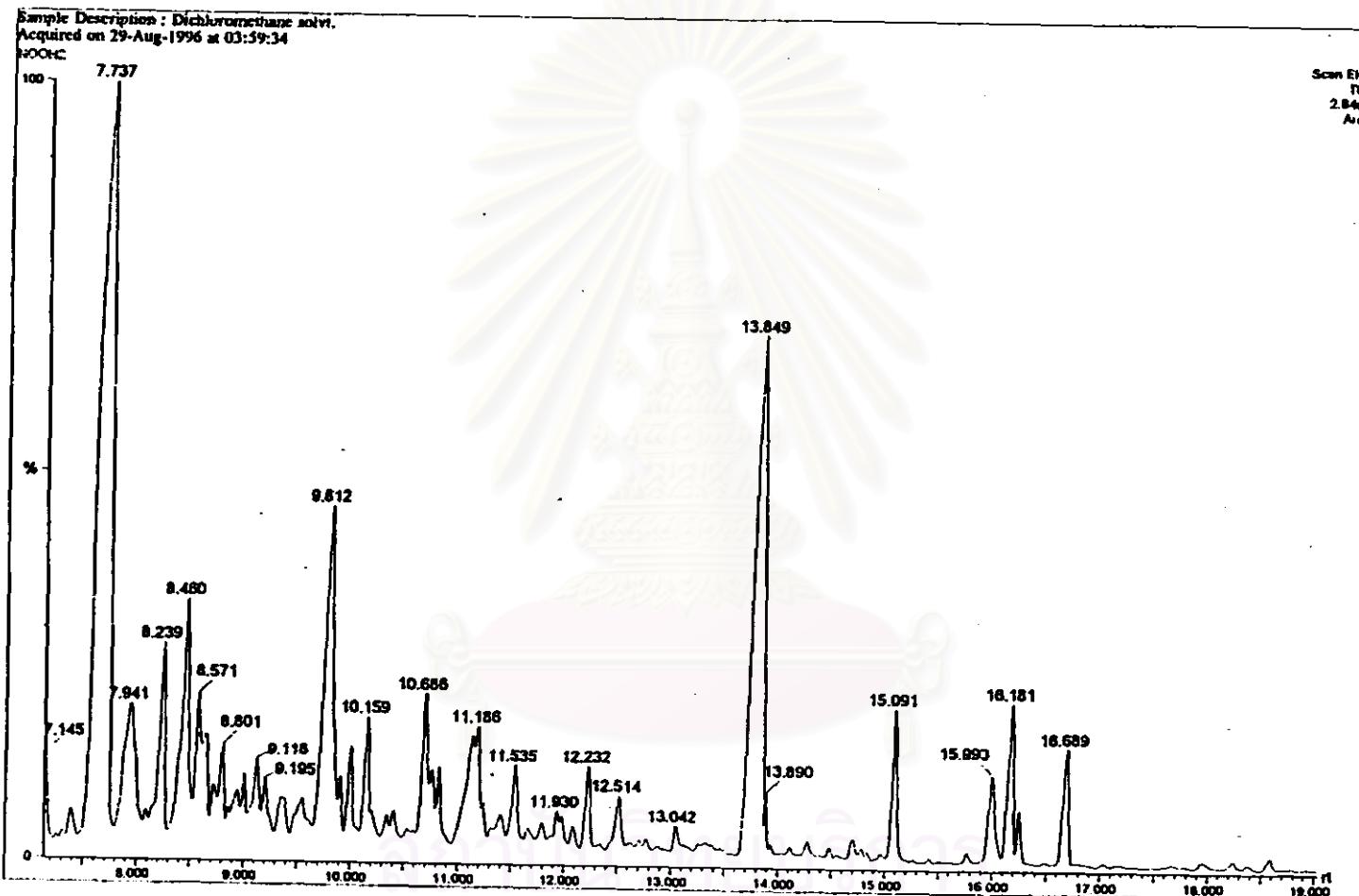


Figure 6 The Gas-Liquid chromatogram of mixture I

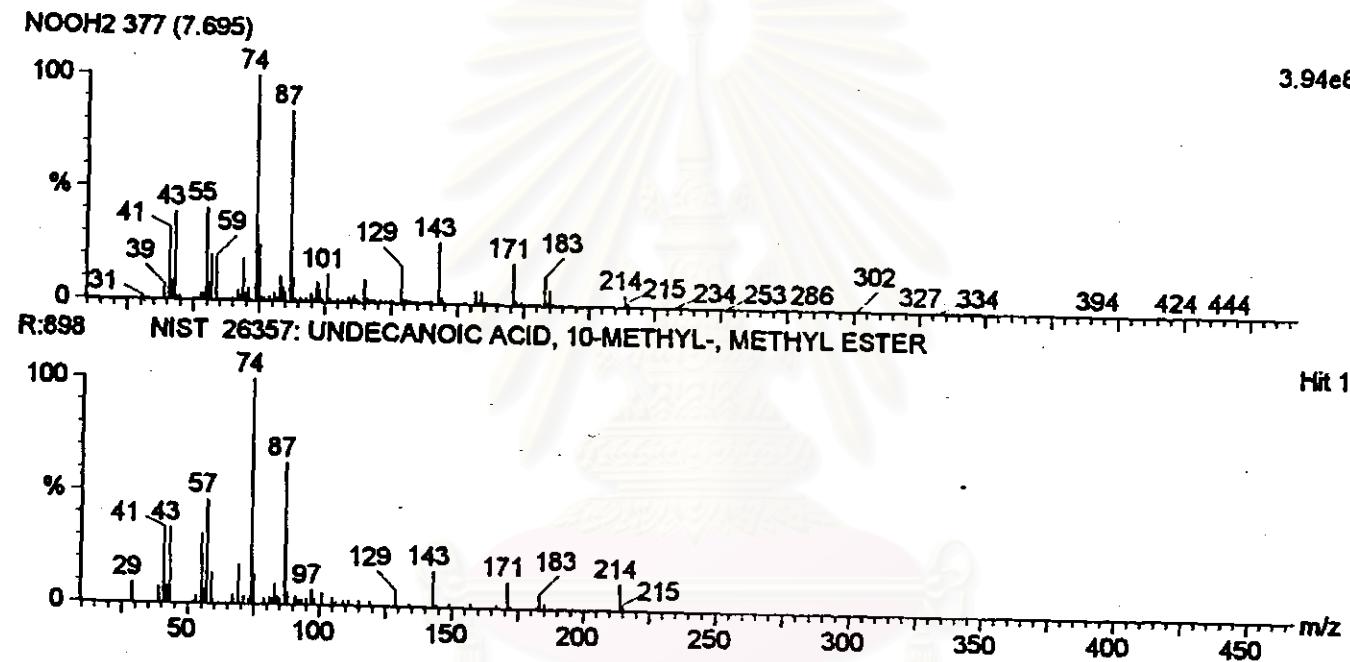


Figure 7 The mass spectrum of mixture I at retention time 7.74

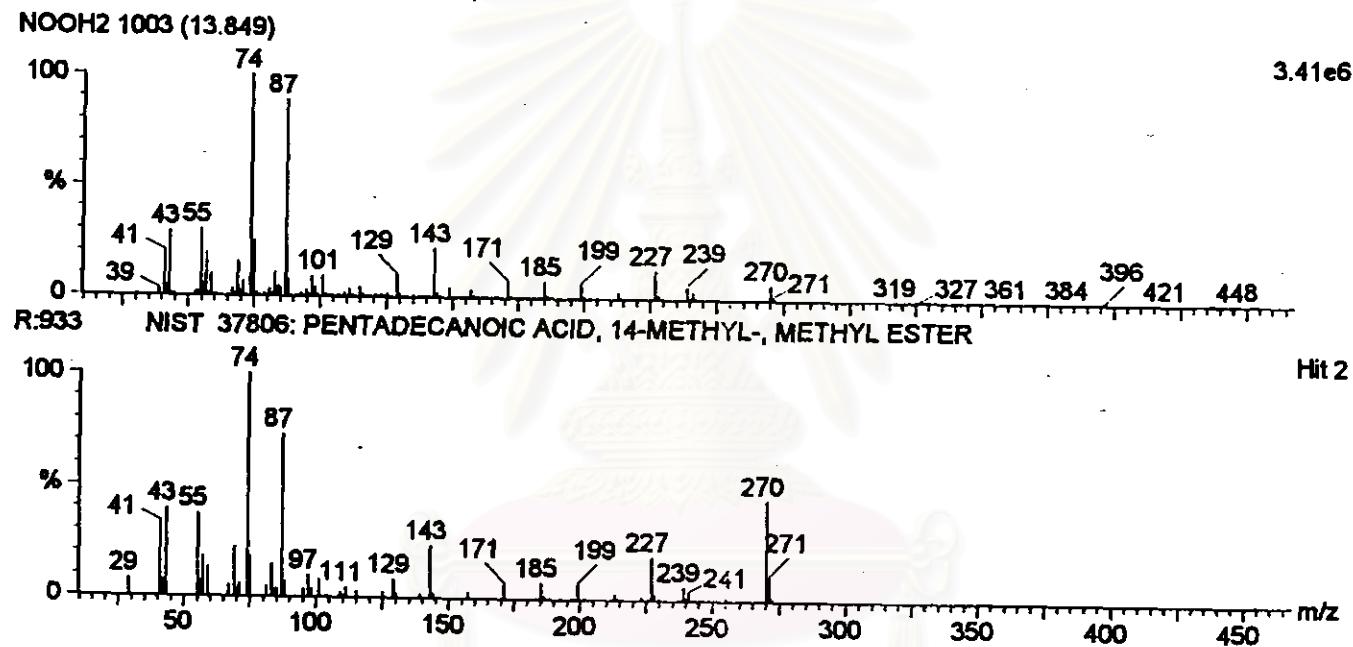


Figure 8 The mass spectrum of mixture I at retention time 14.85

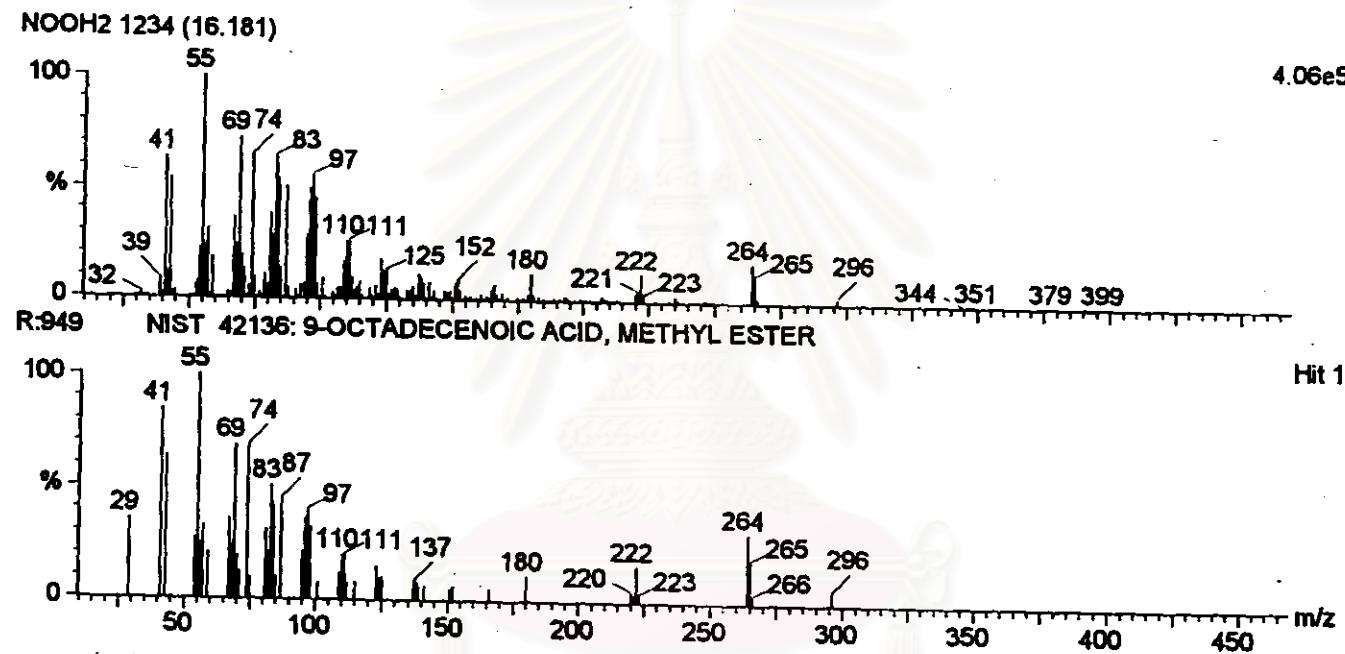


Figure 9 The mass spectrum of mixture I at retention time 16.18

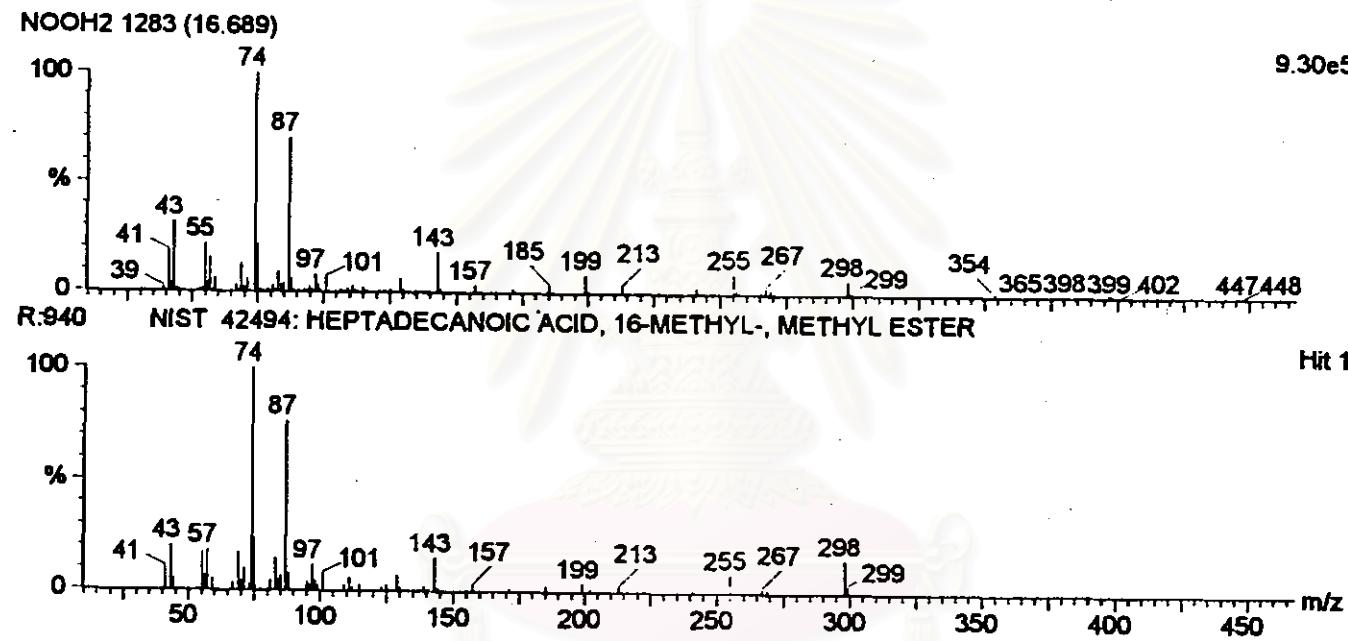


Figure 10 The mass spectrum of mixture I at retention time 16.70

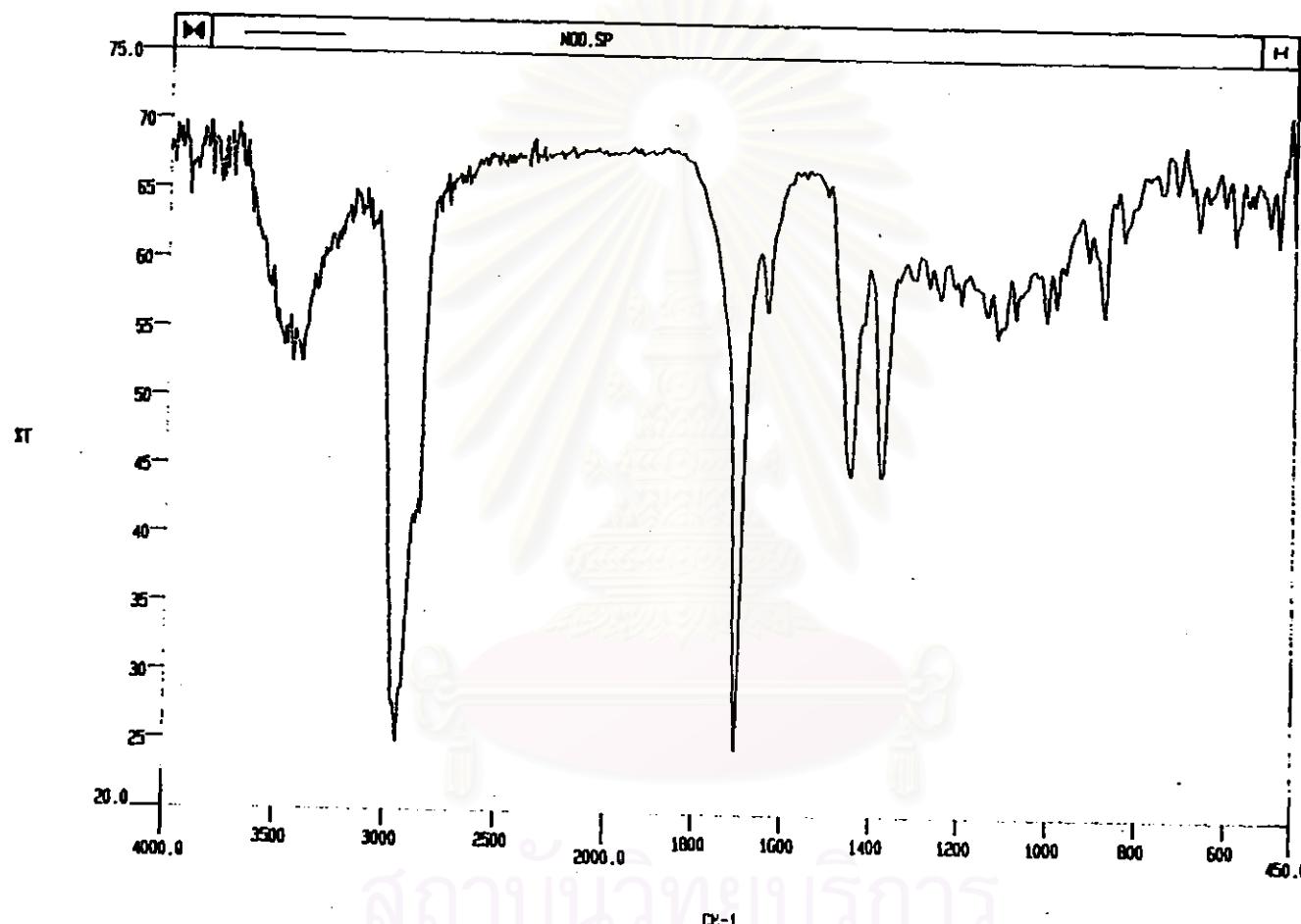
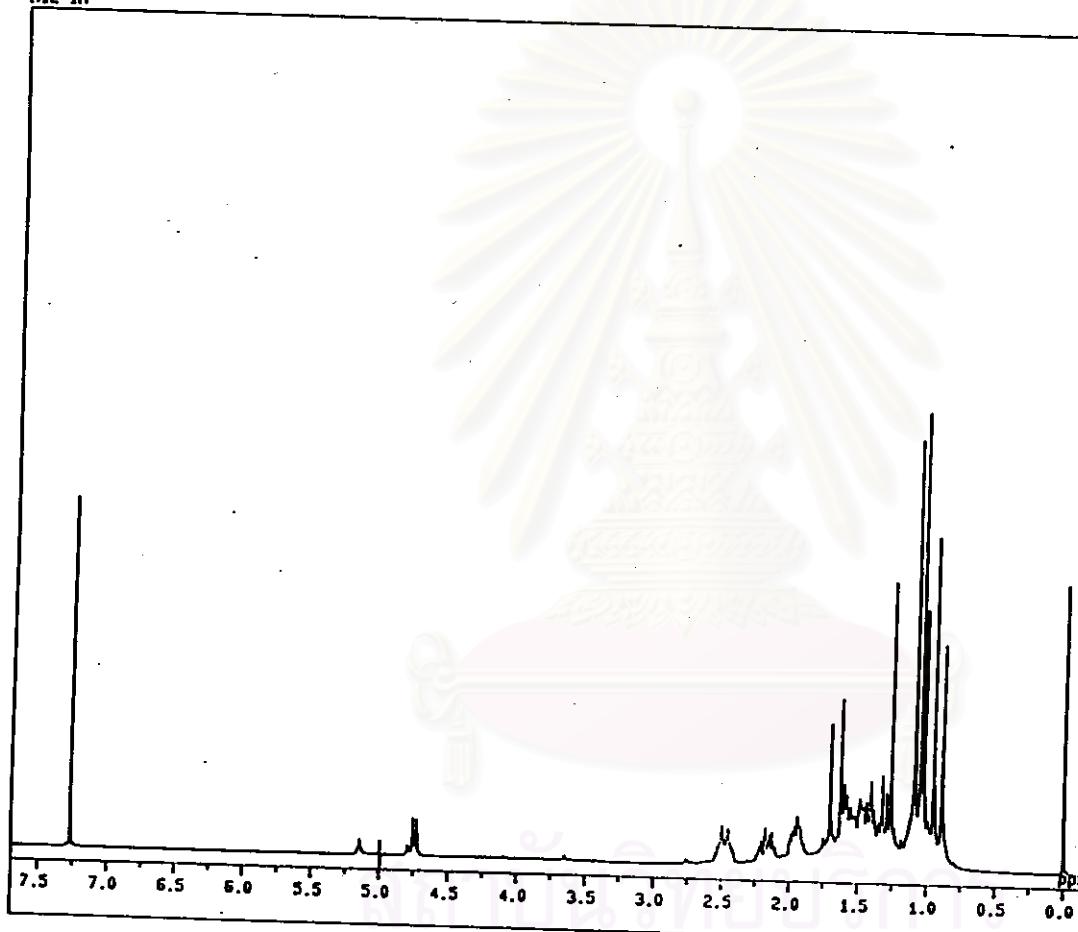


Figure 11. The IR spectrum of compound II

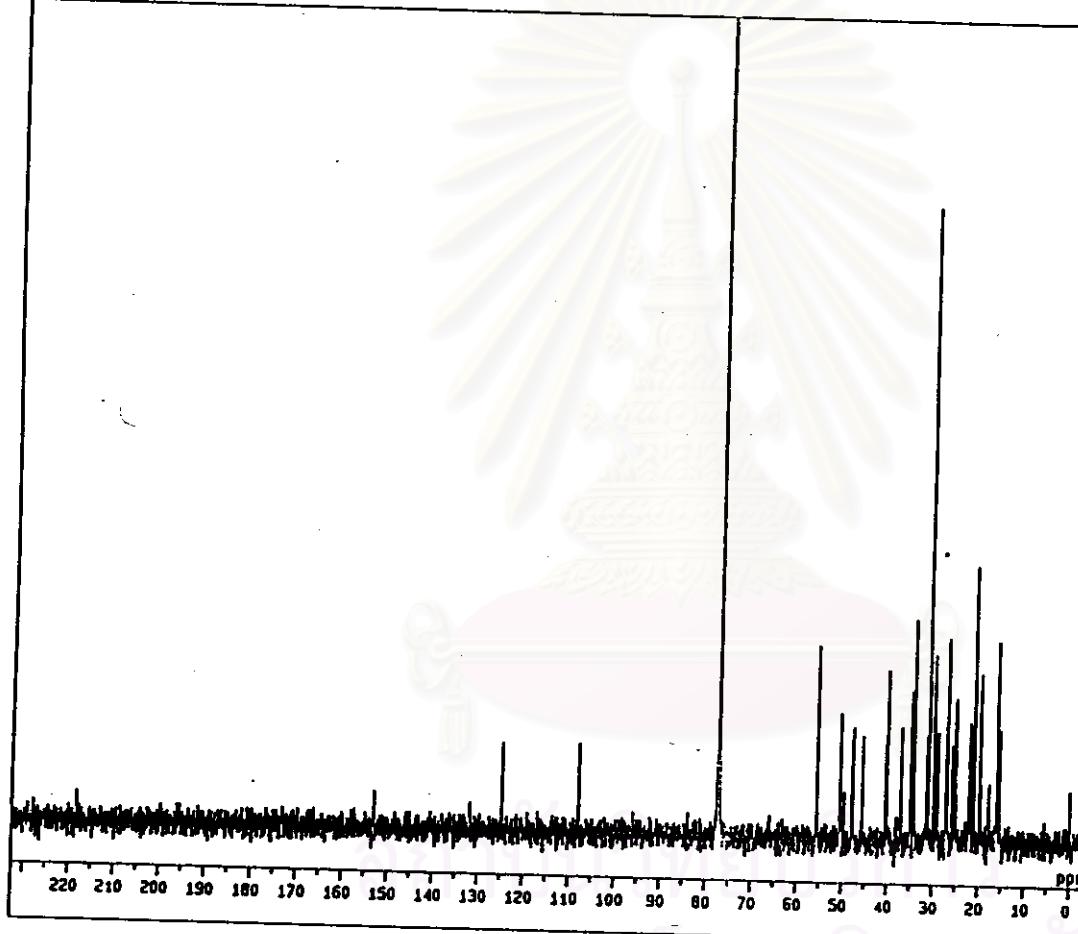
H12-1H



12-NOV-1996 10:20:02.93
CHULALONGKORN UNIVERSITY
JMR-AS00
SFFILE : (.SS)H12-H
COMNT : H12-1H
EXHOD : SINGL
IRHOU : MON
POINT : 32768
FREQU : 10000.00 Hz
SCANS : 24
DURATY : 3.2768 sec
PD : 3.0000 sec
RGAIN : 17
PWI : 3.40 usec
DEDEC : 3H
DEFRQ : 500.00 MHz
DSET : 162410.00 Hz
INAVC : 1H
INFRQ : 500.00 MHz
ISSET : 162410.00 Hz
IRATN : 120
IRAPM : 30.0 usec
IRBP1 : 25
IRBP2 : 2
IRBPG : 0
ADDT : 16
CTEESP : 29.4 c
CSPED : 13 Hz
SILVNT : CDCl₃
RESOL : 0.31 Hz
RF : 0.31 Hz
REFVL : 0.00 ppm
XE : 3451.72 Hz
IS : 618.44 Hz
OPERATOR :

Figure 12 The ¹H-NMR spectrum of compound II

H12-13C



12-NOV-1996 08:23:35.13

CHULALONGKORN UNIVERSITY
JMH-A500

FILE : (.3G)H12-BCN
CUNIT : H12-13C

EXNU : 3164L
IRATE : BCM
POINT : 16384
FREQU : 33898.31 Hz
SCANS : 2500
DURAT : 1
ACQST : 0.4833 sec
PR : 2.0000 sec
PAWA : 23
PMI : 4.99 msec
DEPH : 13C
DEPHU : 125.63 Hz
DSET : 127958.00 Hz

INNUC : 1H
INNUU : 500.00 MHz
DSET : 162410.00 Hz
IAATM : 120
IADPM : 55.0 msec
IADP1 : 30
IADP2 : 6
IADW : 9

ADUIT : 16
CTEW : 29.8 c
CPDPD : 11 Hz
CLVLT : CDLL3
RESOL : 2.07 Hz
SF : 2.07 Hz
REFVL : 77.00 ppm
XE : 29799.64 Hz
XG : -1683.19 Hz

OPERATOR :

Figure 13 The ¹³C-NMR spectrum of compound II

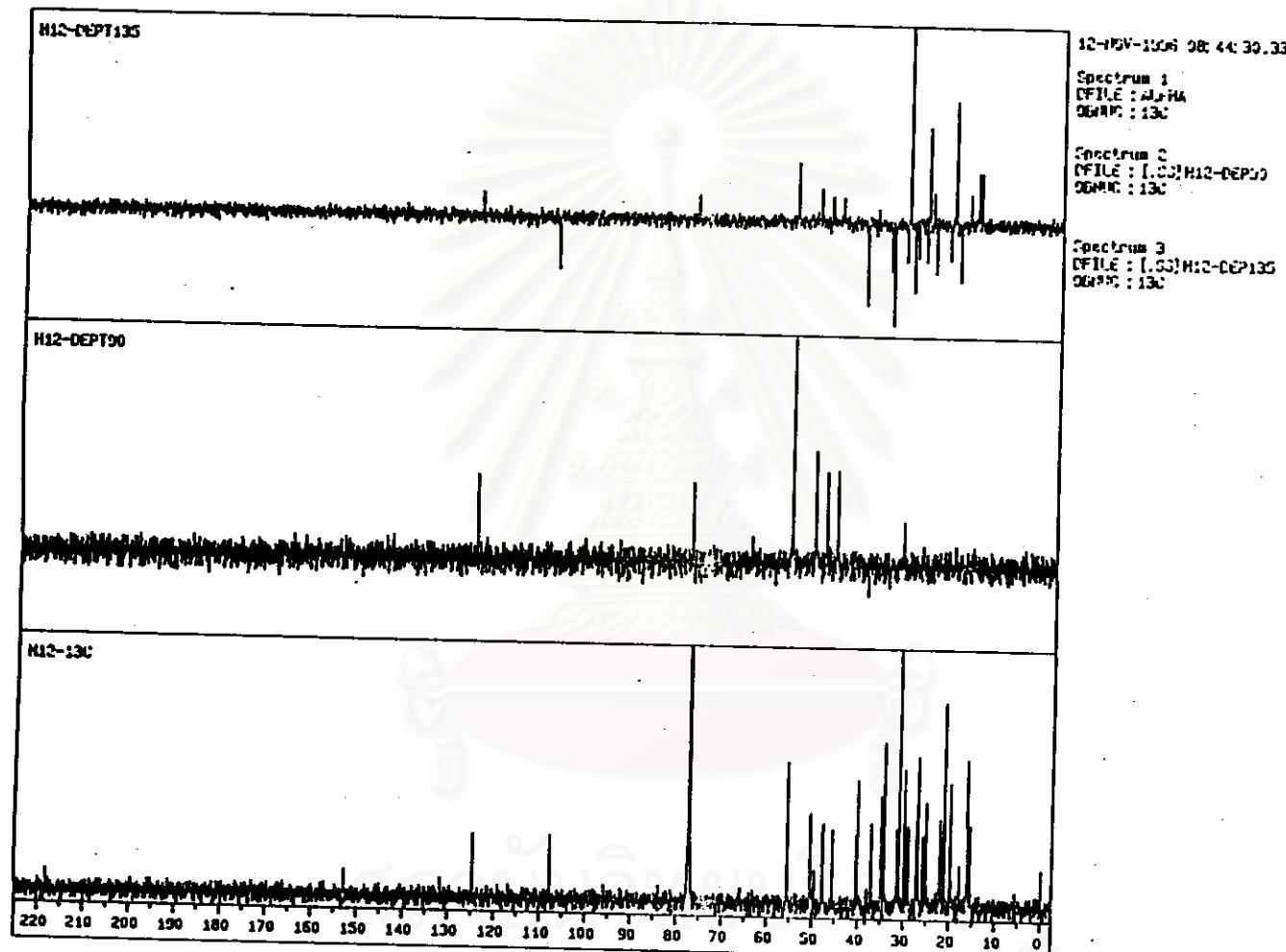


Figure 14 The DEPT 90 and DEPT 135 ^{13}C -NMR spectrum of compound II

Sample Description : Dichloromethane solv.
Acquired on 29-Aug-1996 at 04:51:01
NCOOTR 2455 (0.331) Cm (2421-2484)

Scan El+
2.75e4

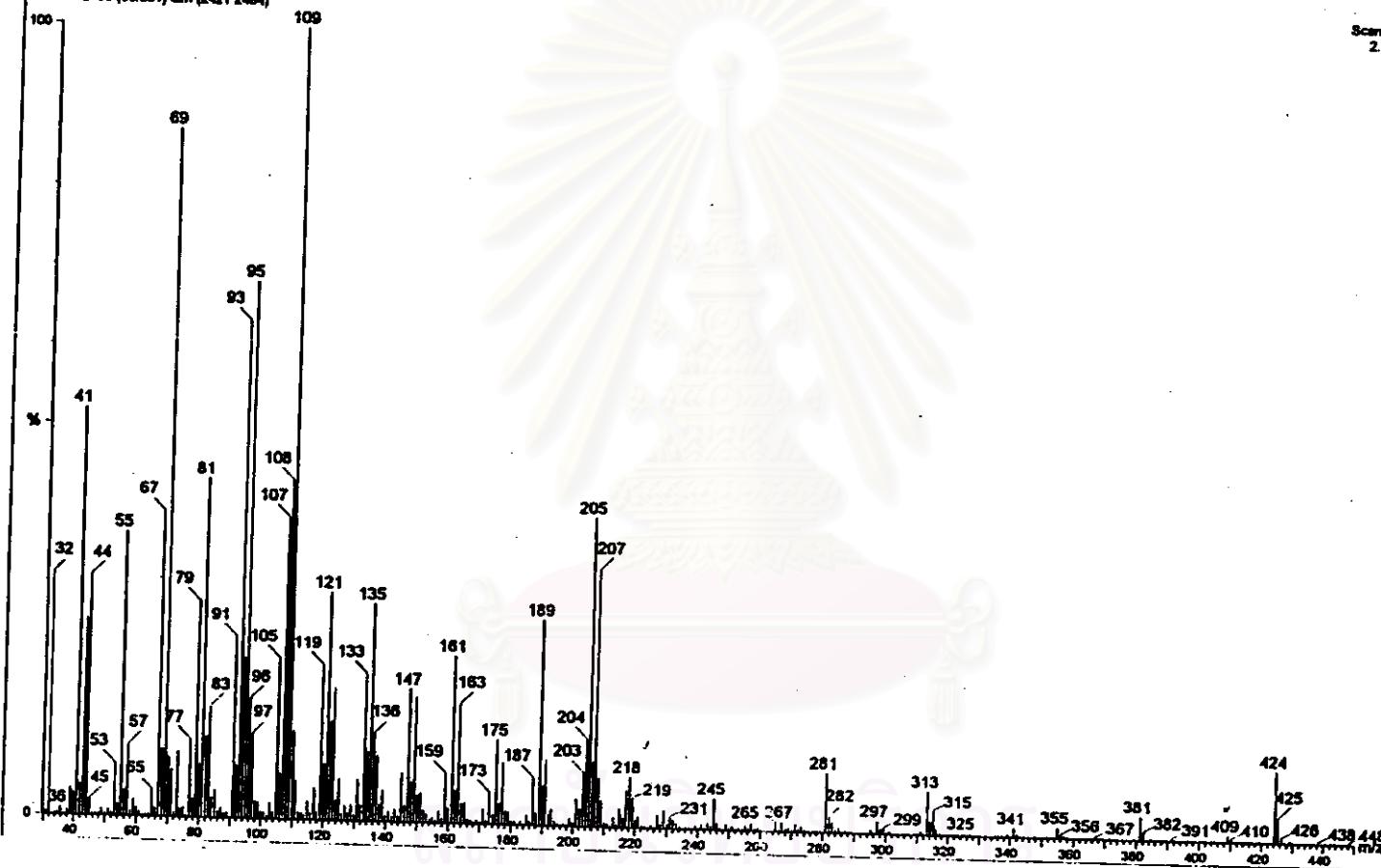


Figure 15 The mass spectrum of compound II

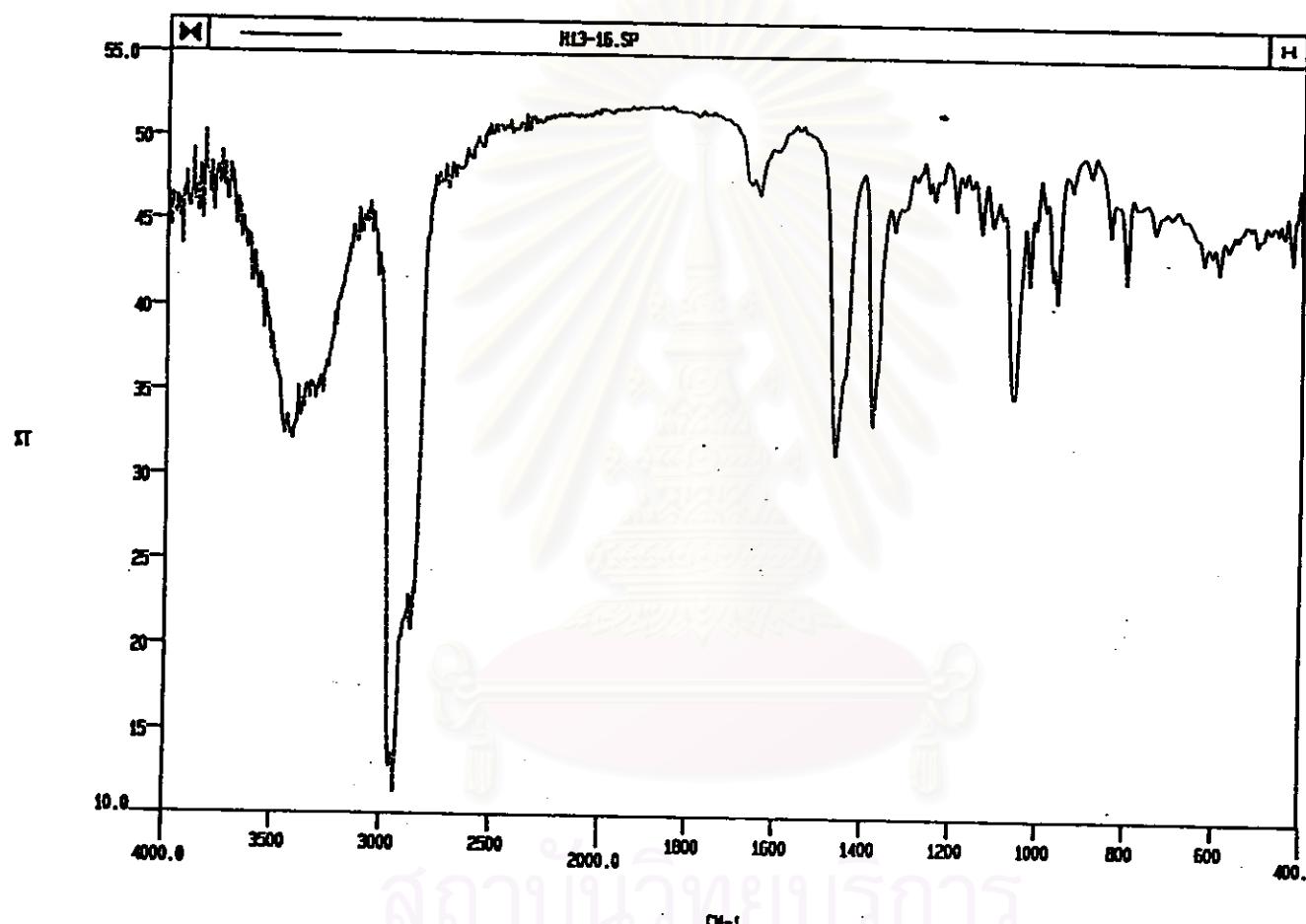


Figure 16 The IR spectrum of mixture III

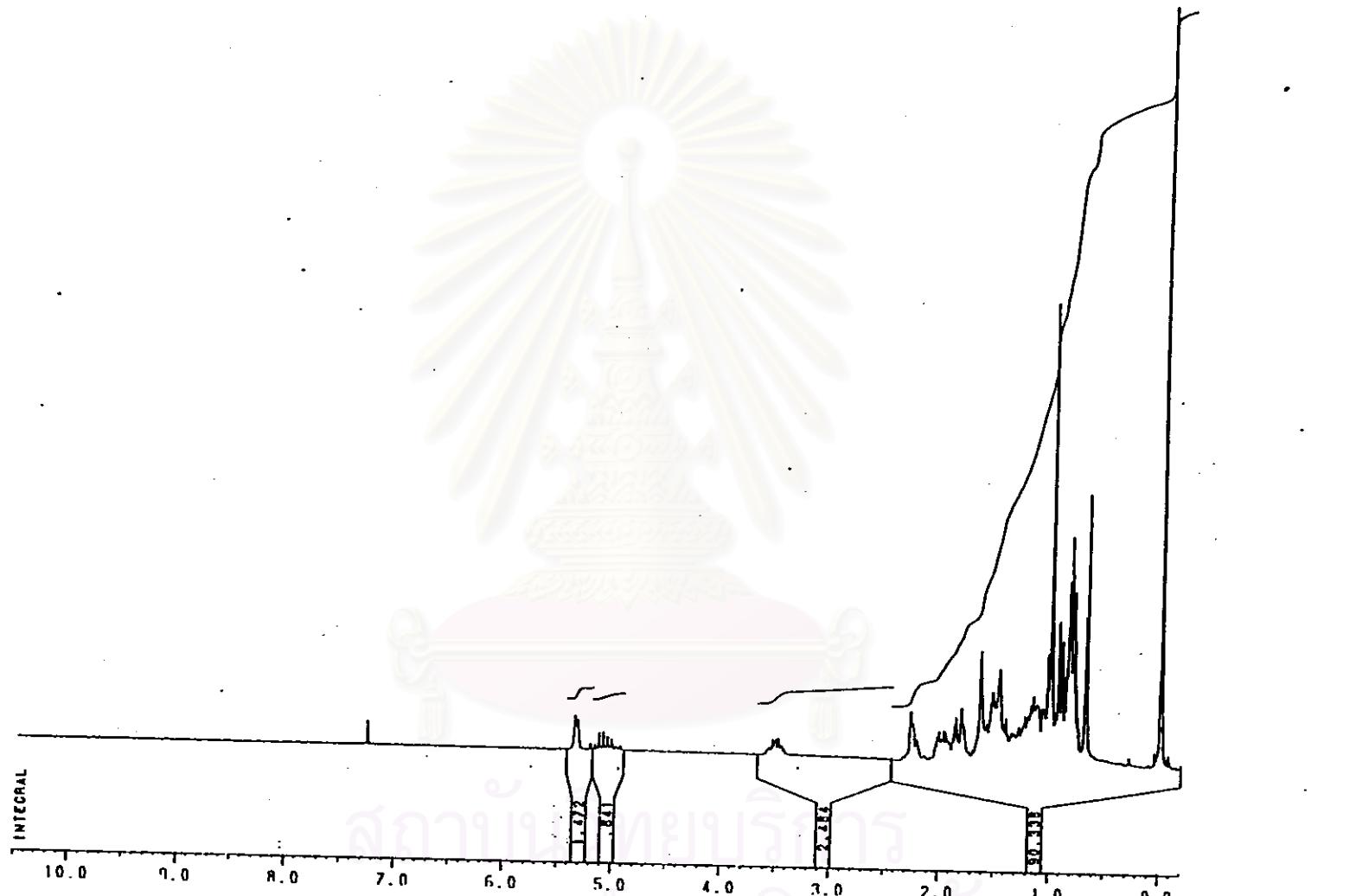


Figure 17 The ^1H -NMR spectrum of mixture III

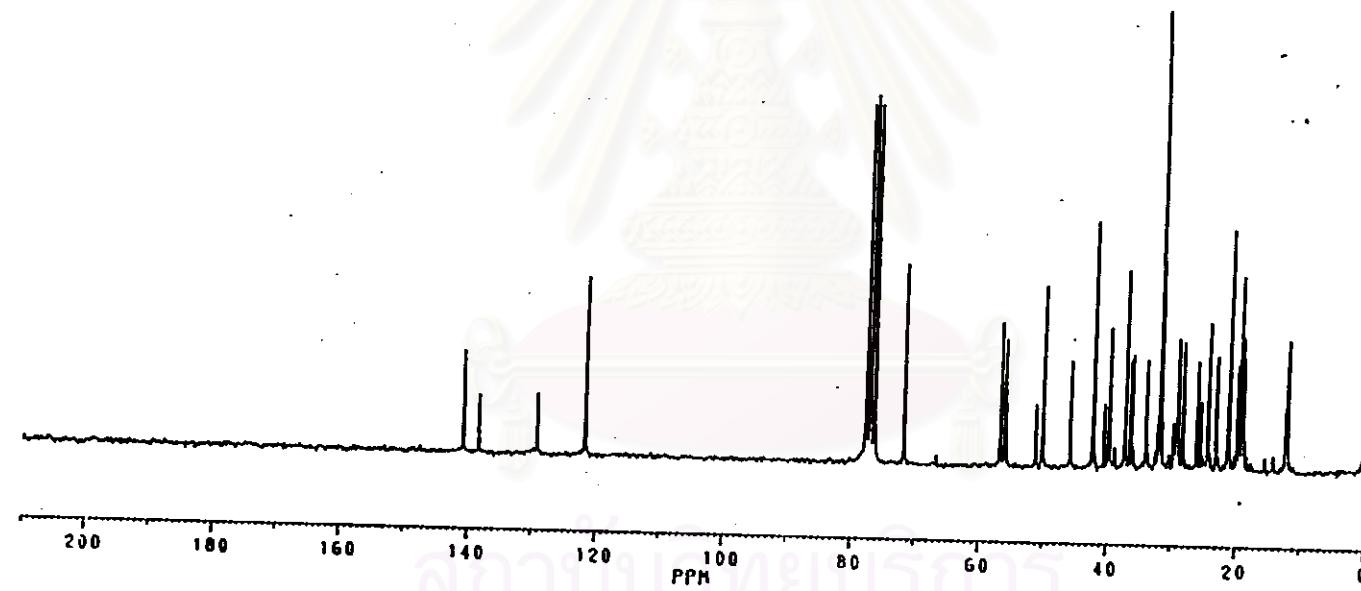


Figure 18 The ^{13}C -NMR spectrum of mixture III

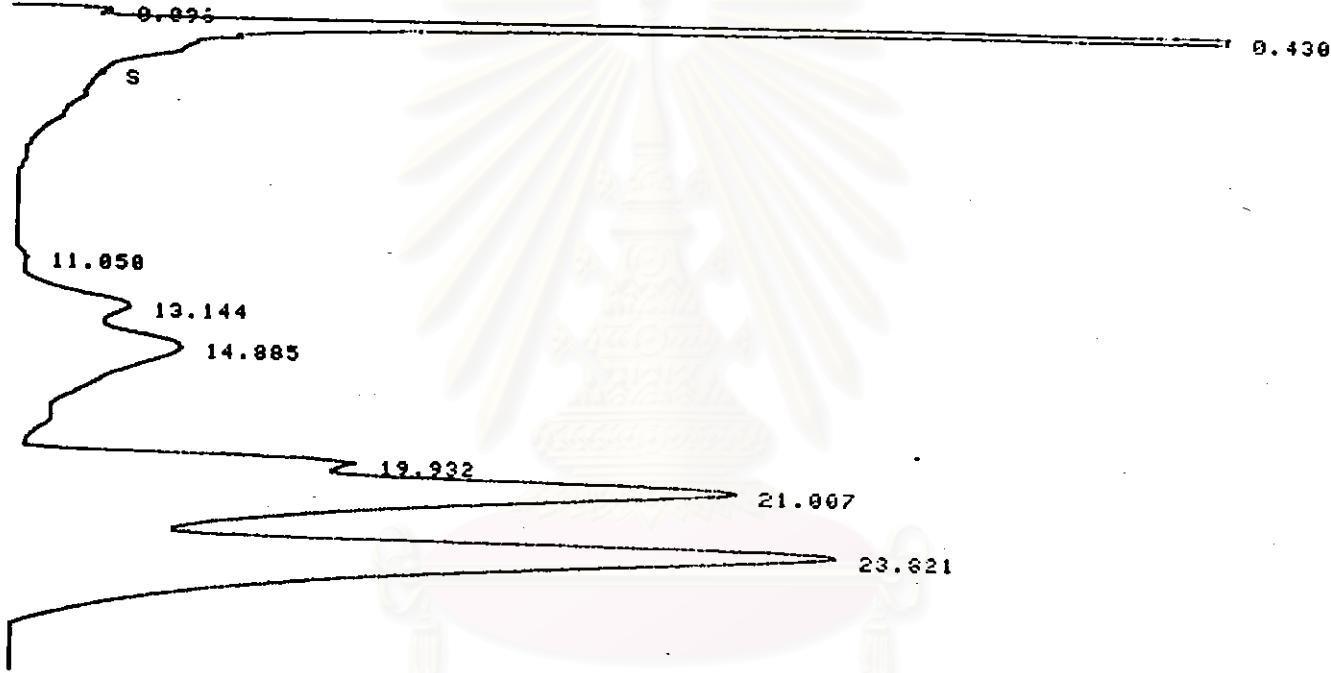


Figure 19 The Gas-Liquid chromatogram of mixture of standard ateroid

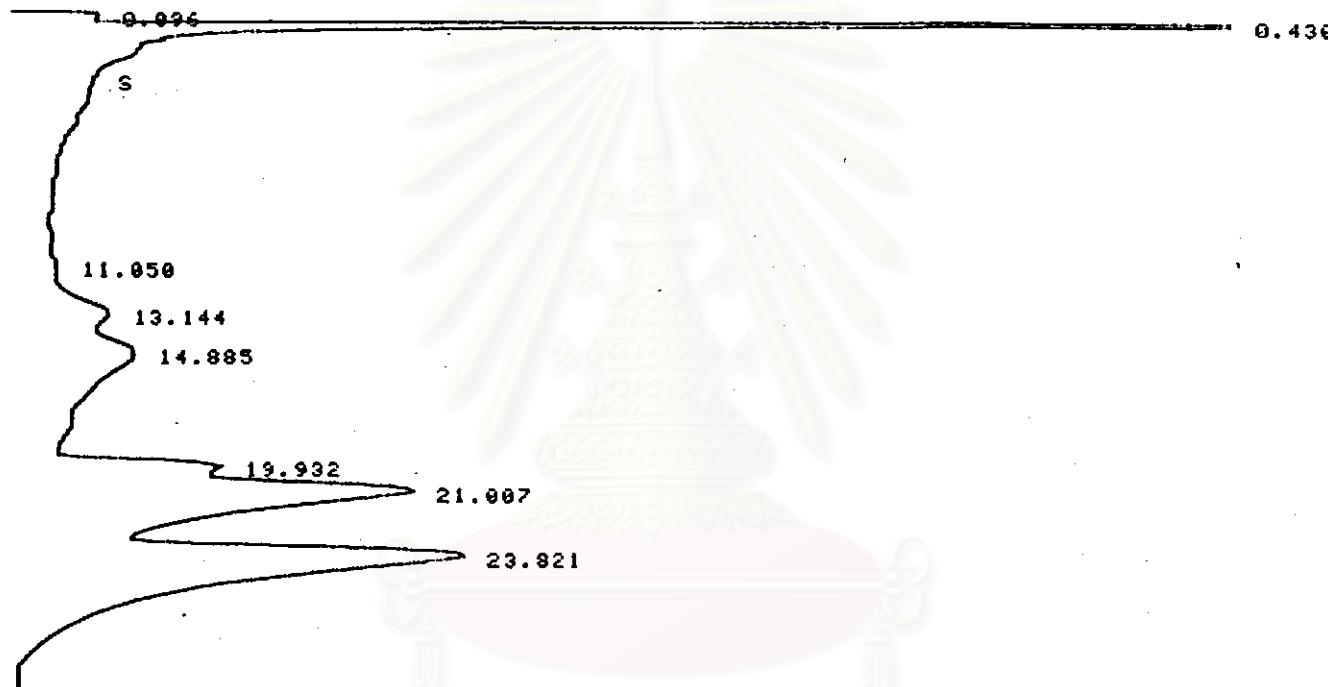


Figure 20 The Gas-Liquid chromatogram of mixture III

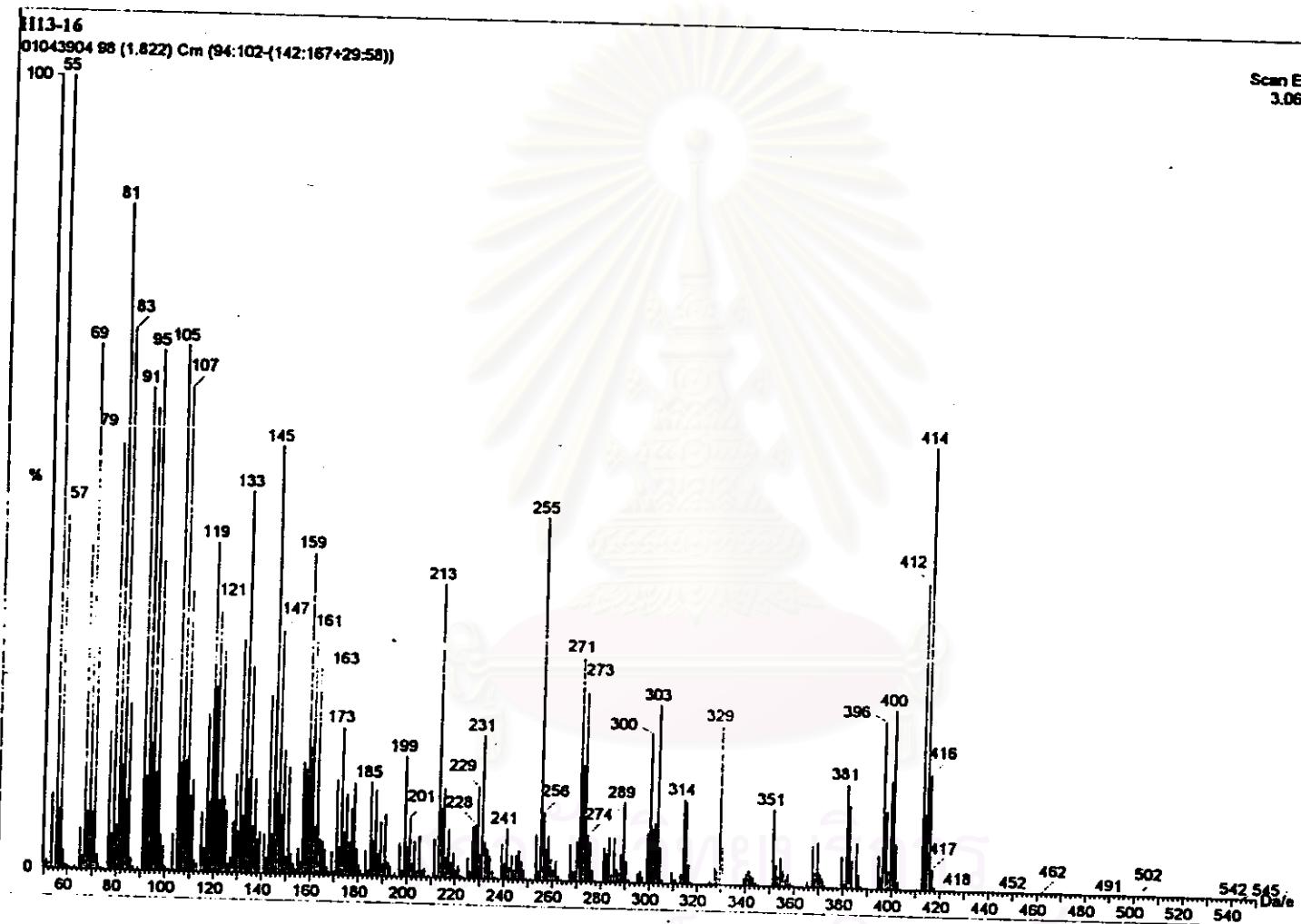


Figure 21 The mass spectrum of mixture III

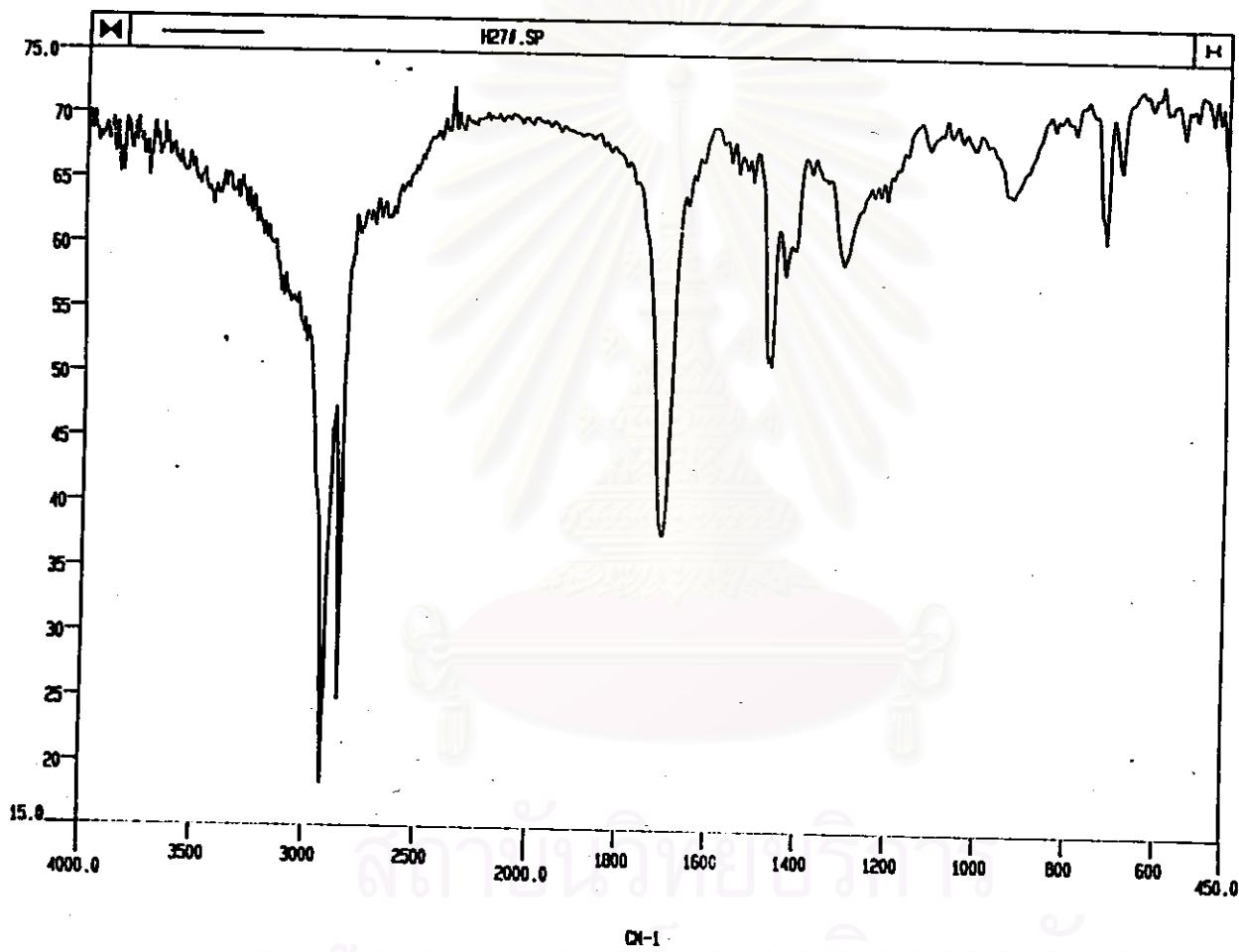


Figure 22 The IR spectrum of mixture IV

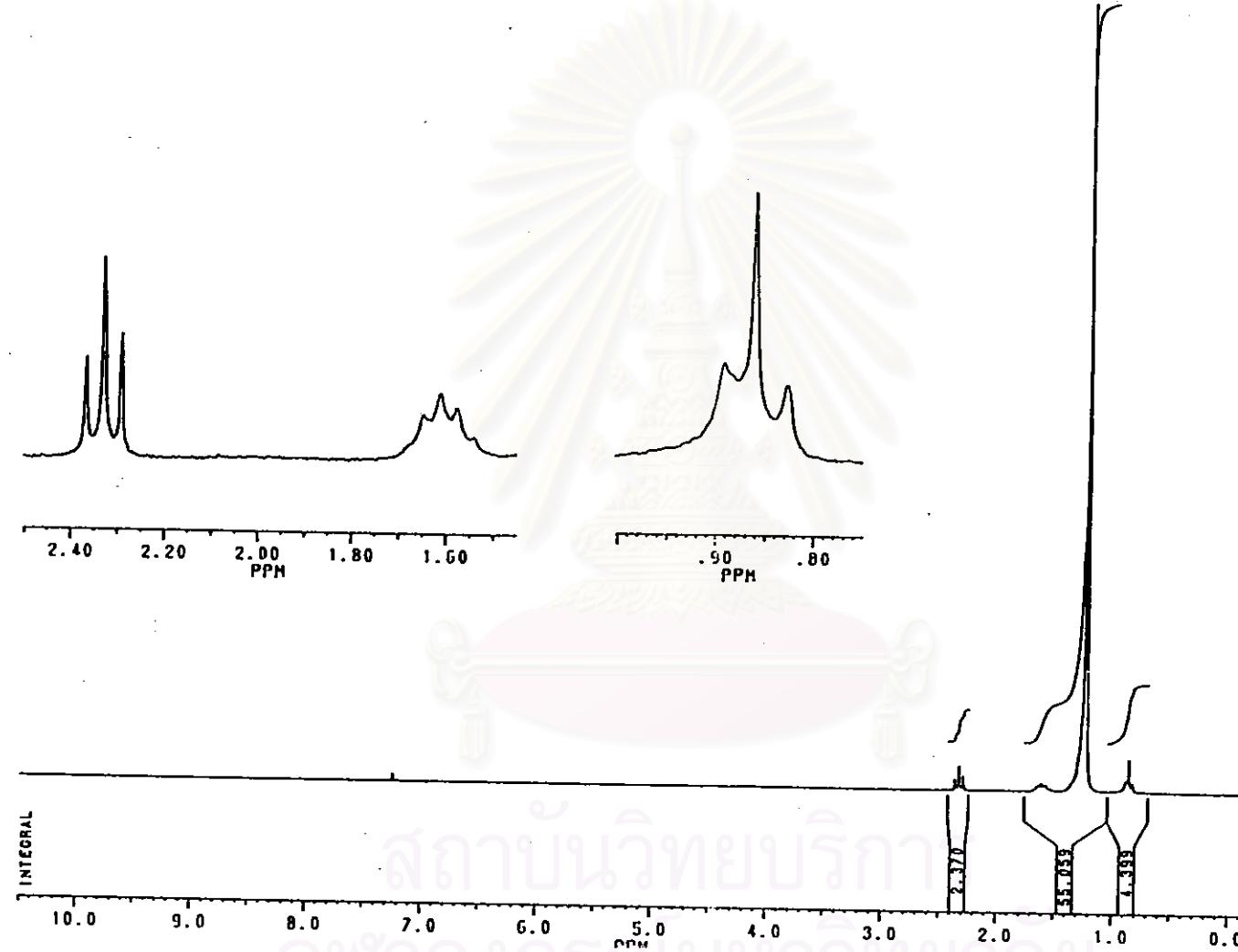


Figure 23 The ^1H -NMR spectrum of mixture IV

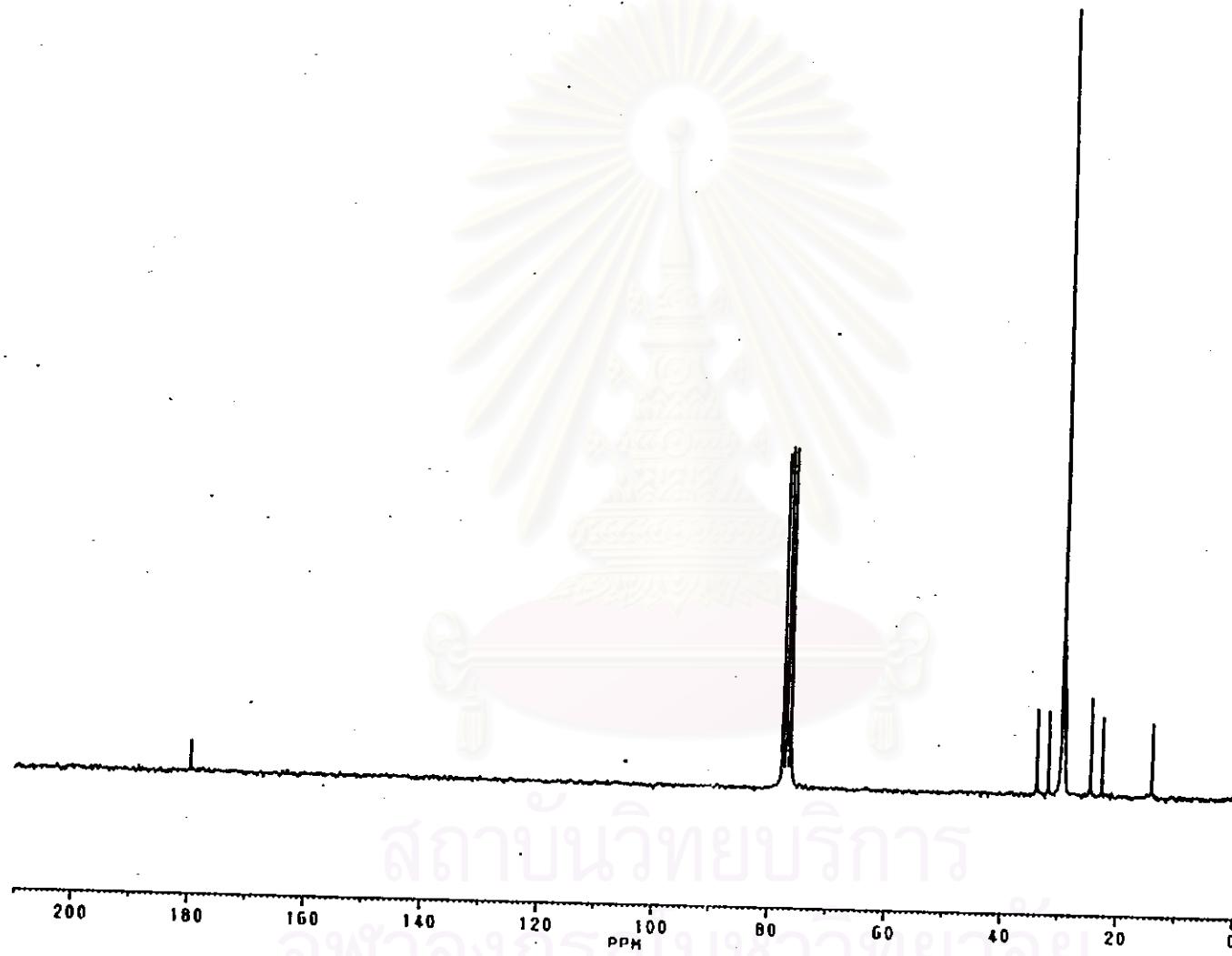


Figure 24 The ^{13}C -NMR spectrum of mixture IV

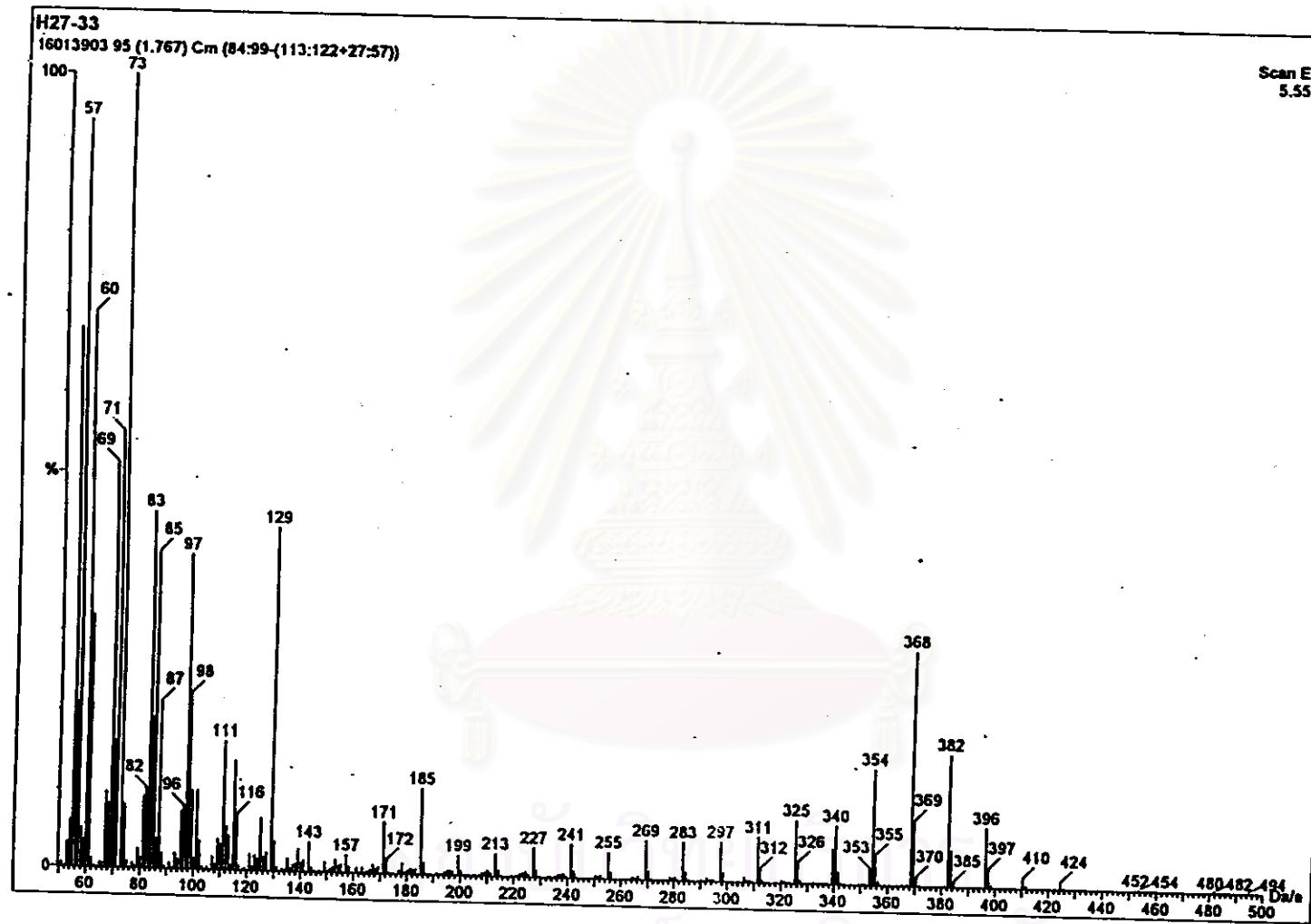


Figure 25 The mass spectrum of mixture IV

Sample Description : Dichloromethane solv.
Acquired on 29-Aug-1996 at 03:08:07
NOOT

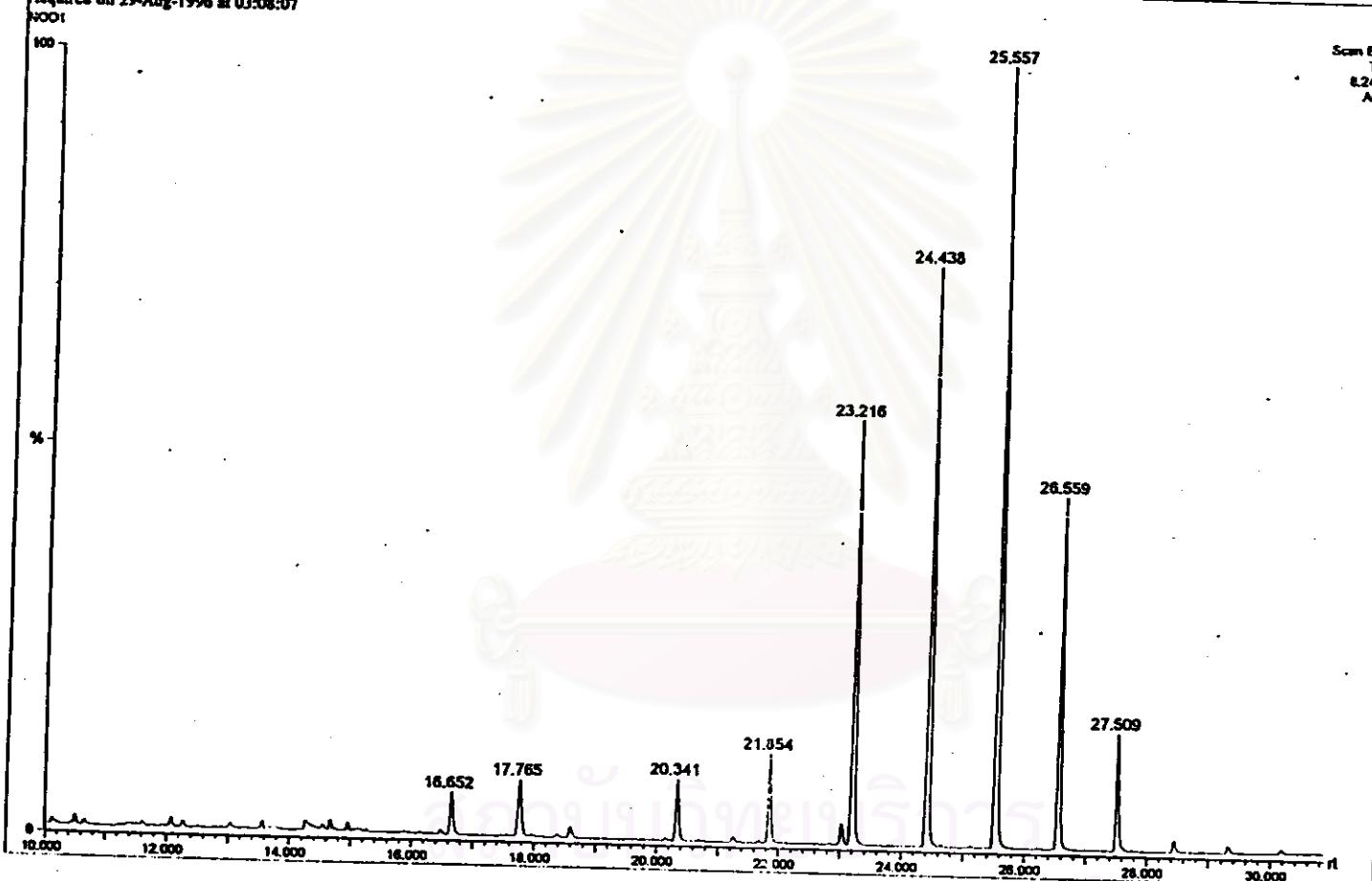


Figure 26 The Gas-Liquid chromatogram of methyl ester of mixture IV

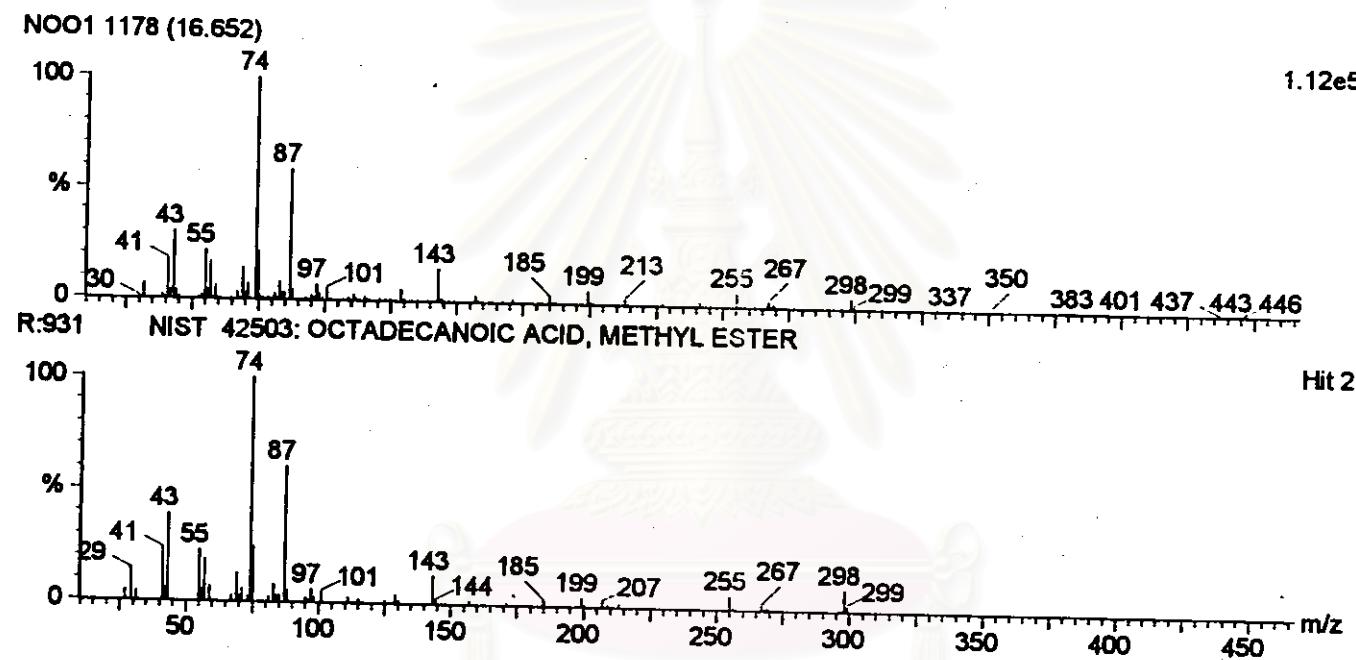


Figure 27. The mass spectrum of methyl ester at retention time 16.65

Sample Description : Dichloromethane solvt.
Acquired 29-Aug-1996 at 03:08:07
Library: NIST

NOO1

Forward Fit: 844, Reverse Fit: 927

5.09e5

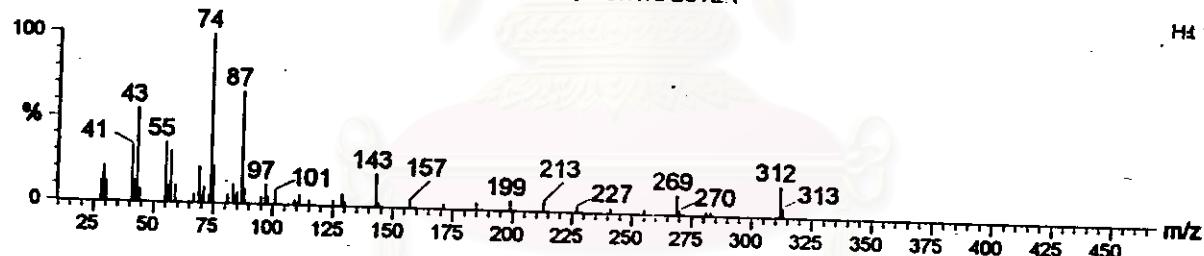
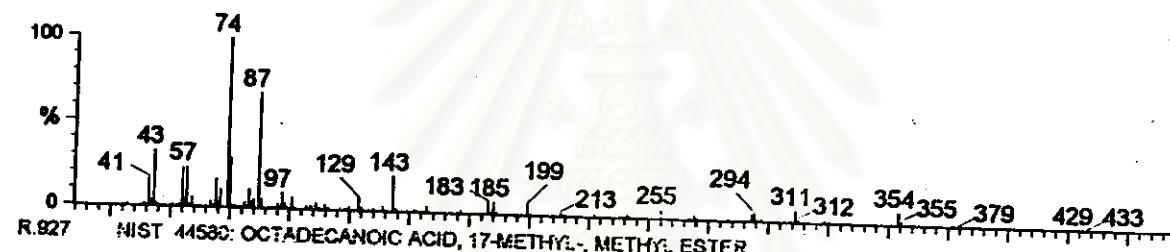


Figure 28 The mass spectrum of methyl ester at retention time 17.77

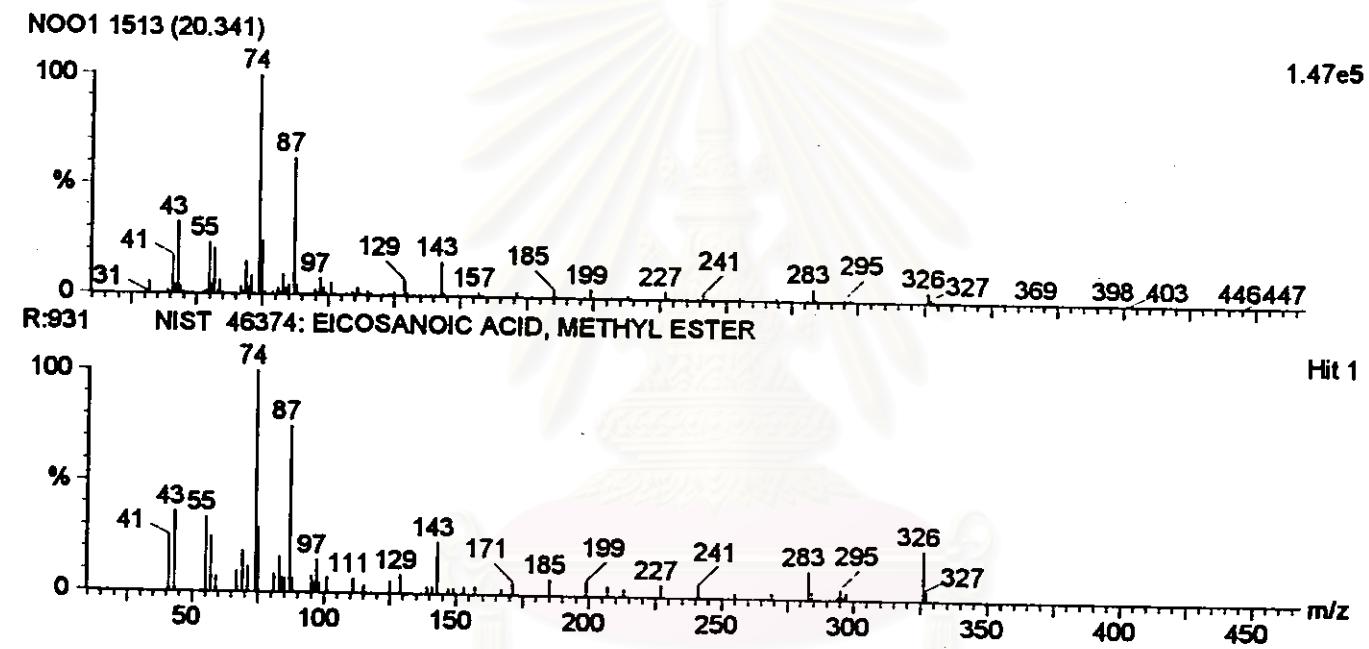


Figure 29 The mass spectrum of methyl ester at retention time 20.34

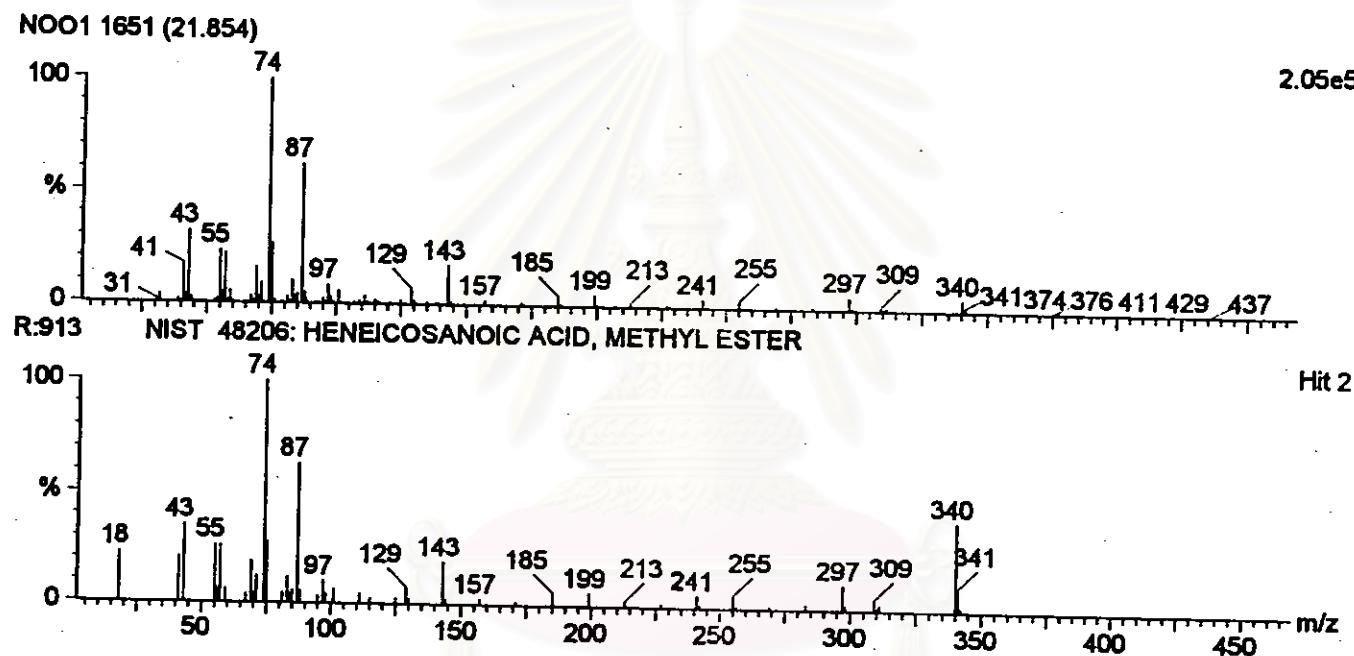


Figure 30 The mass spectrum of methyl ester at retention time 21.85

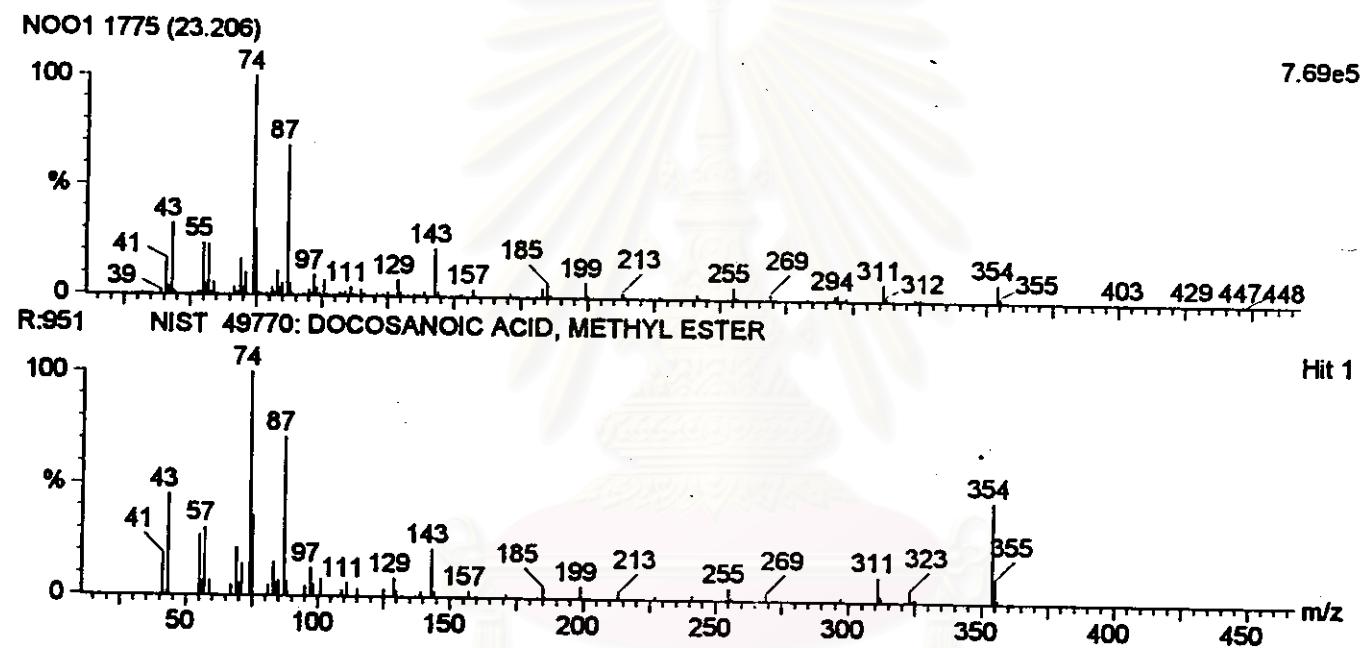


Figure 31 The mass spectrum of methyl ester at retention time 23.22

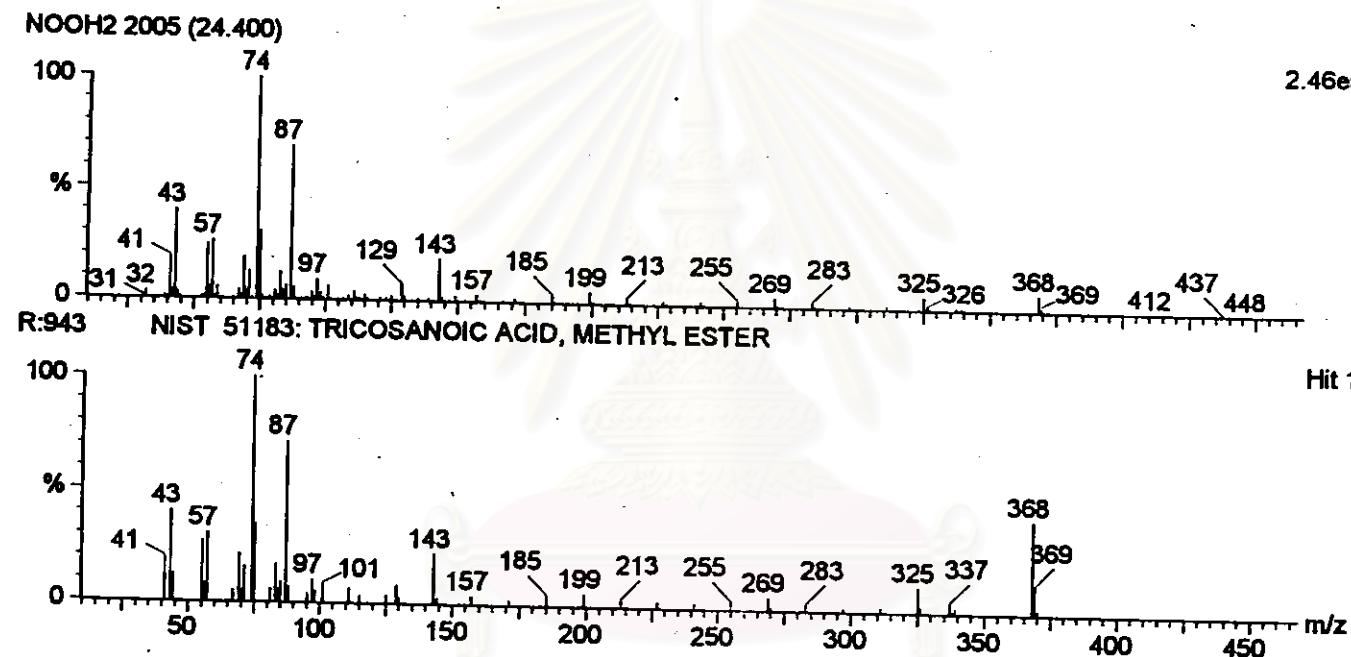


Figure 32 The mass spectrum of methyl ester at retention time 24.44

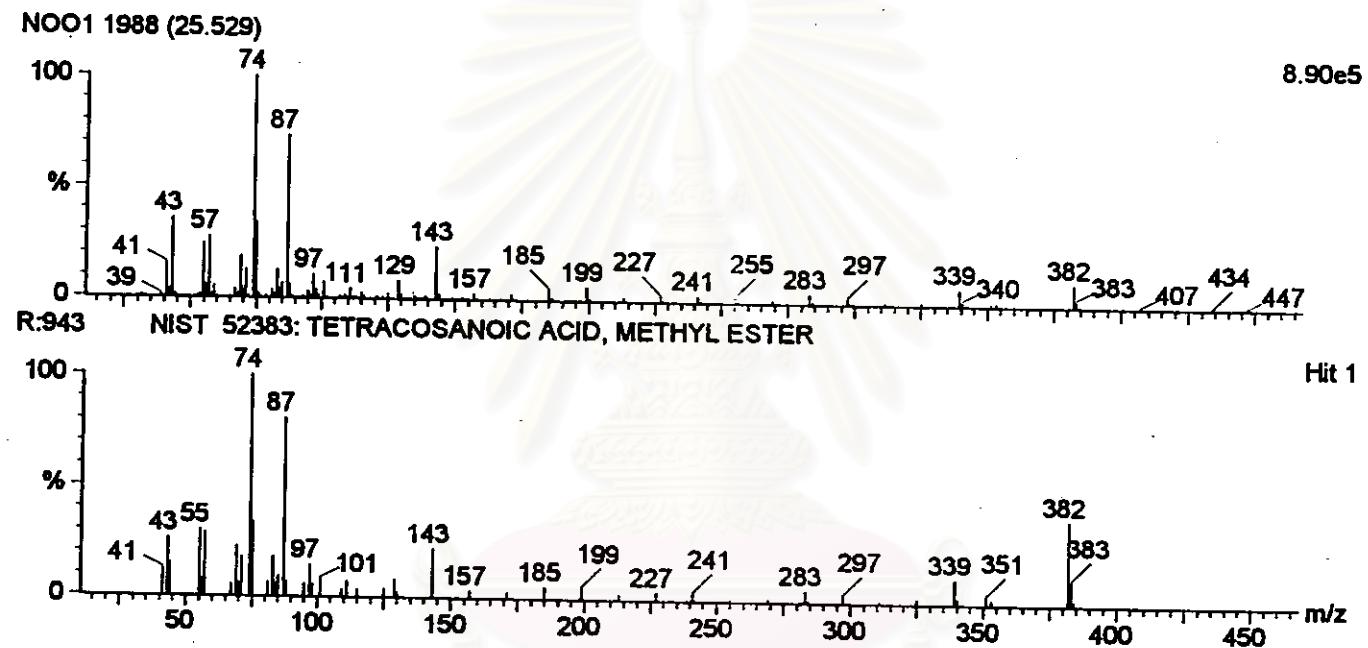


Figure 33 The mass spectrum of methyl ester at retention time 25.56

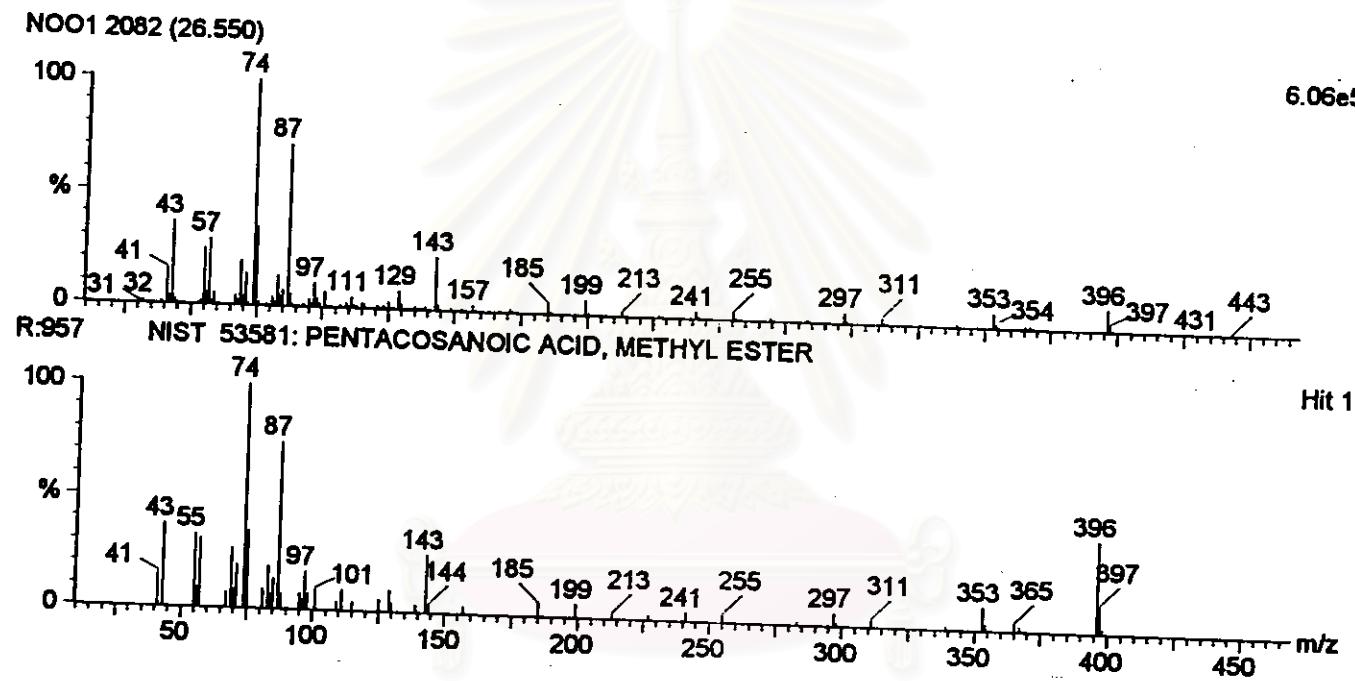


Figure 34 The mass spectrum of methyl ester at retention time 26.56

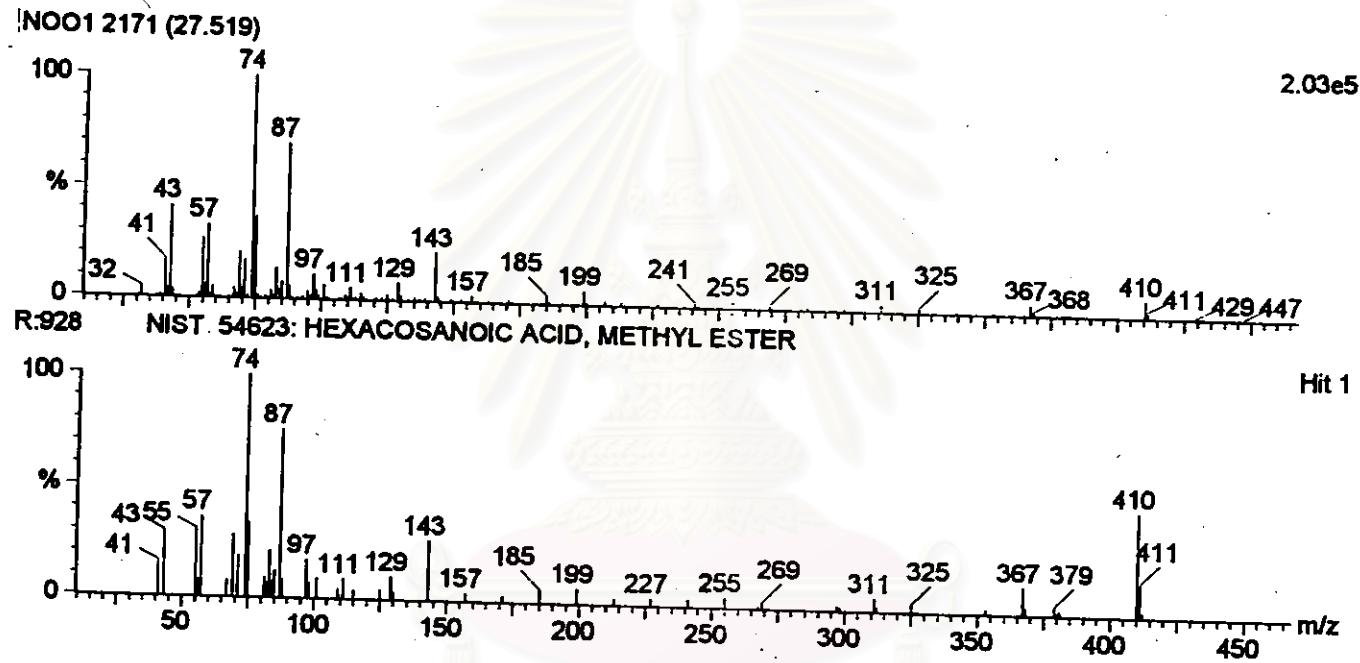


Figure 35 The mass spectrum of methyl ester at retention time 27.51

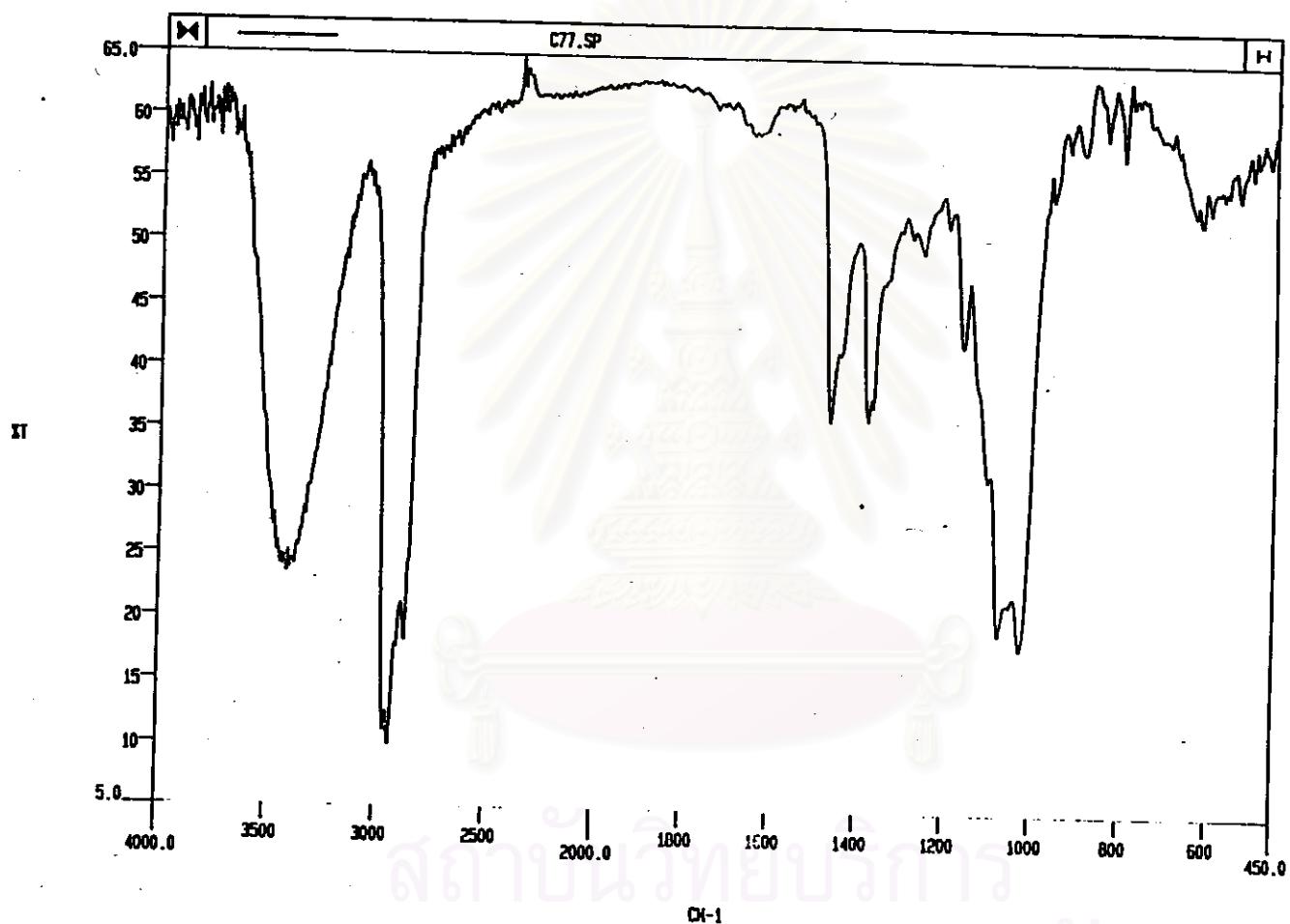


Figure 36 The IR spectrum of mixture V

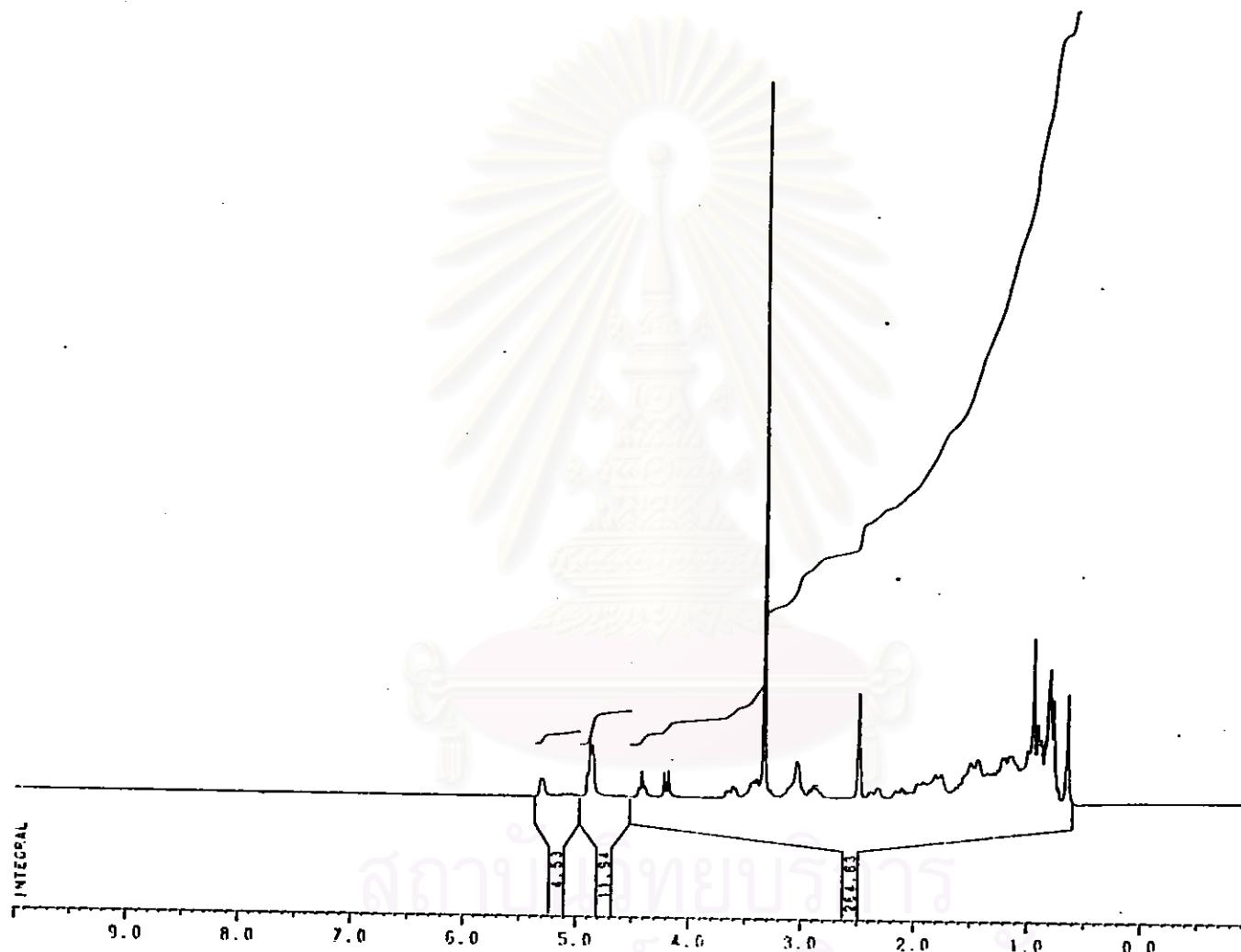


Figure 37 The ^1H -NMR spectrum of mixture V

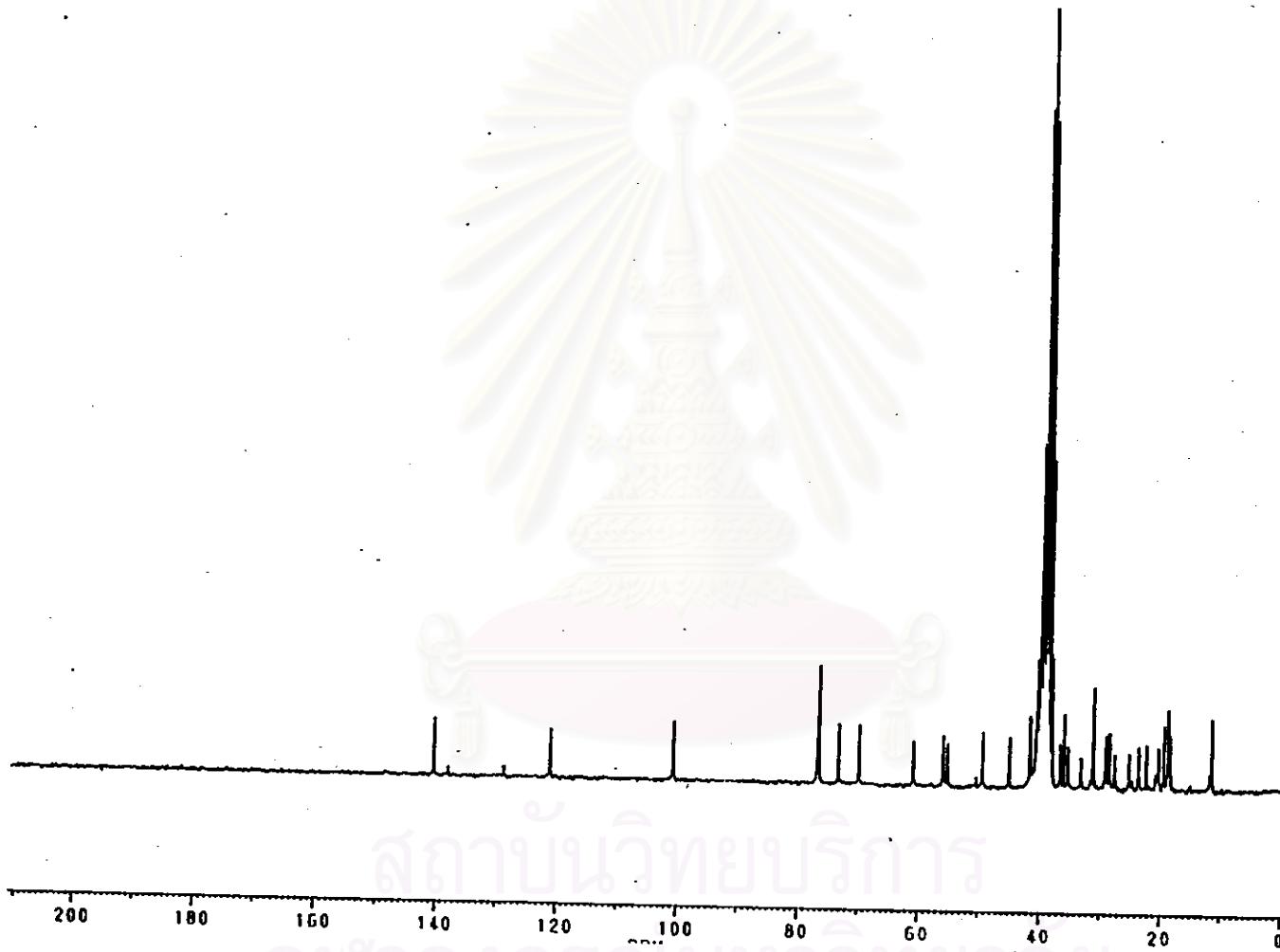


Figure 38 The ^{13}C -NMR spectrum of mixture V

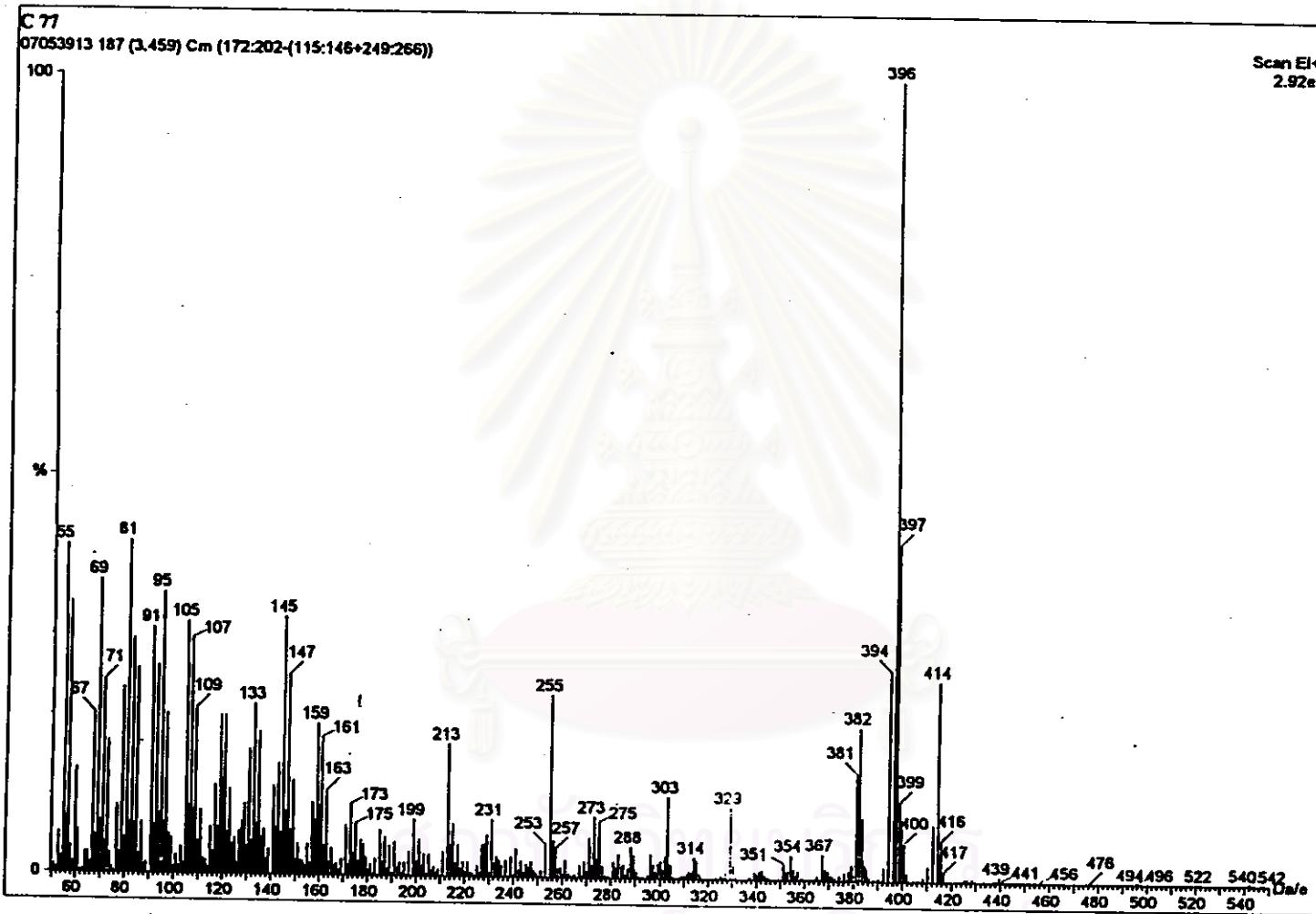


Figure 39 The mass spectrum of mixture V

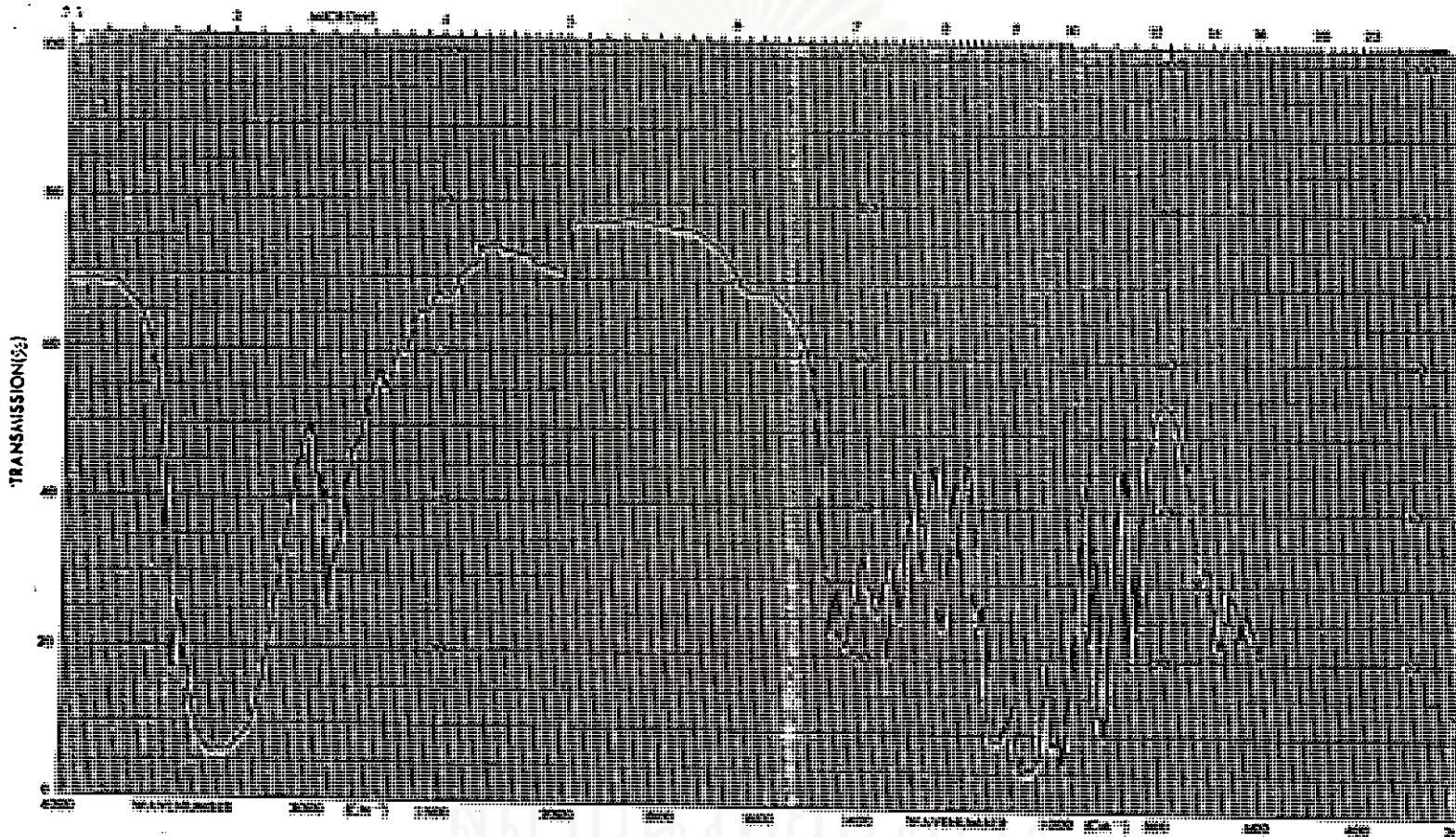


Figure 40 The IR spectrum of compound VI

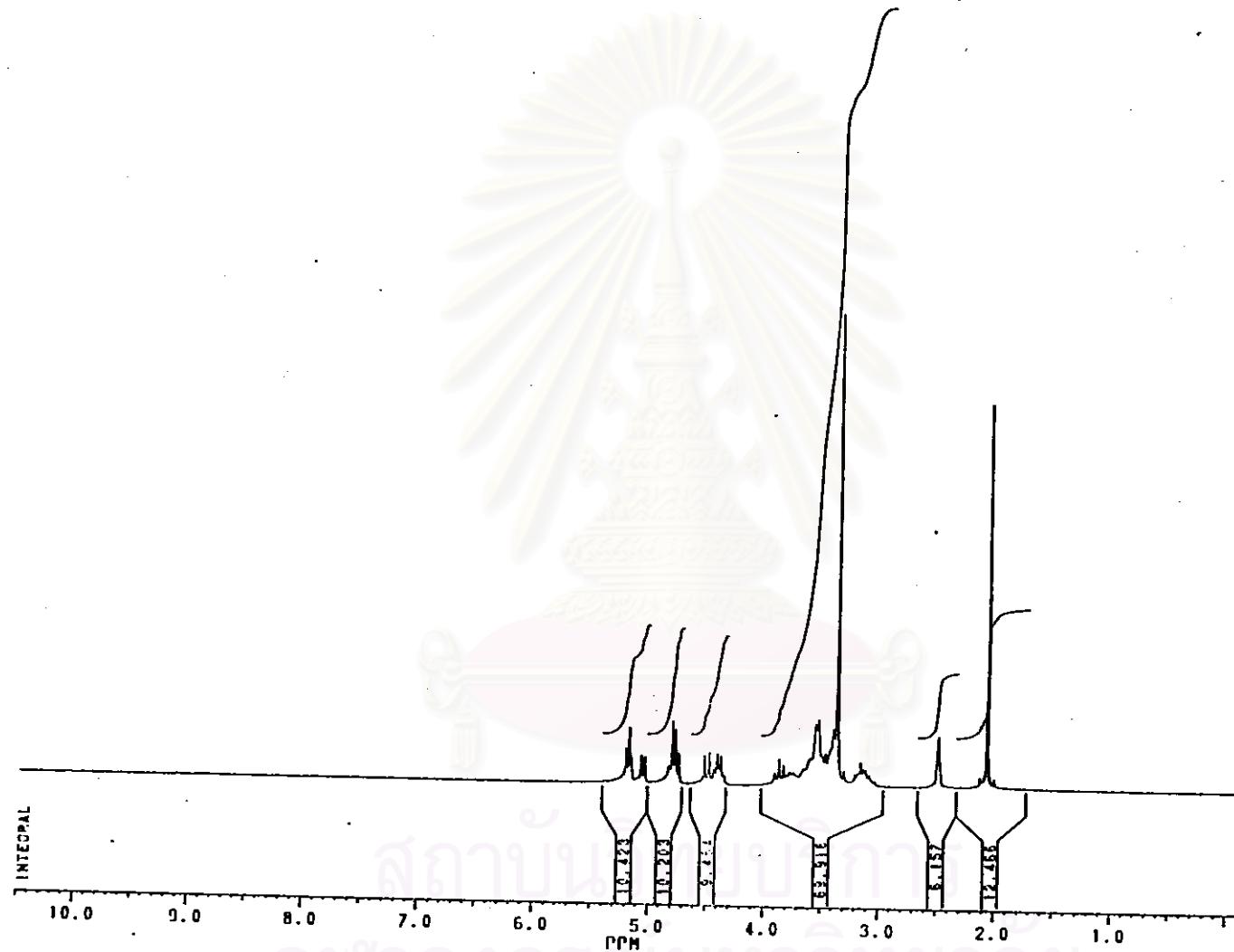


Figure 41 The ^1H -NMR spectrum of compound VI

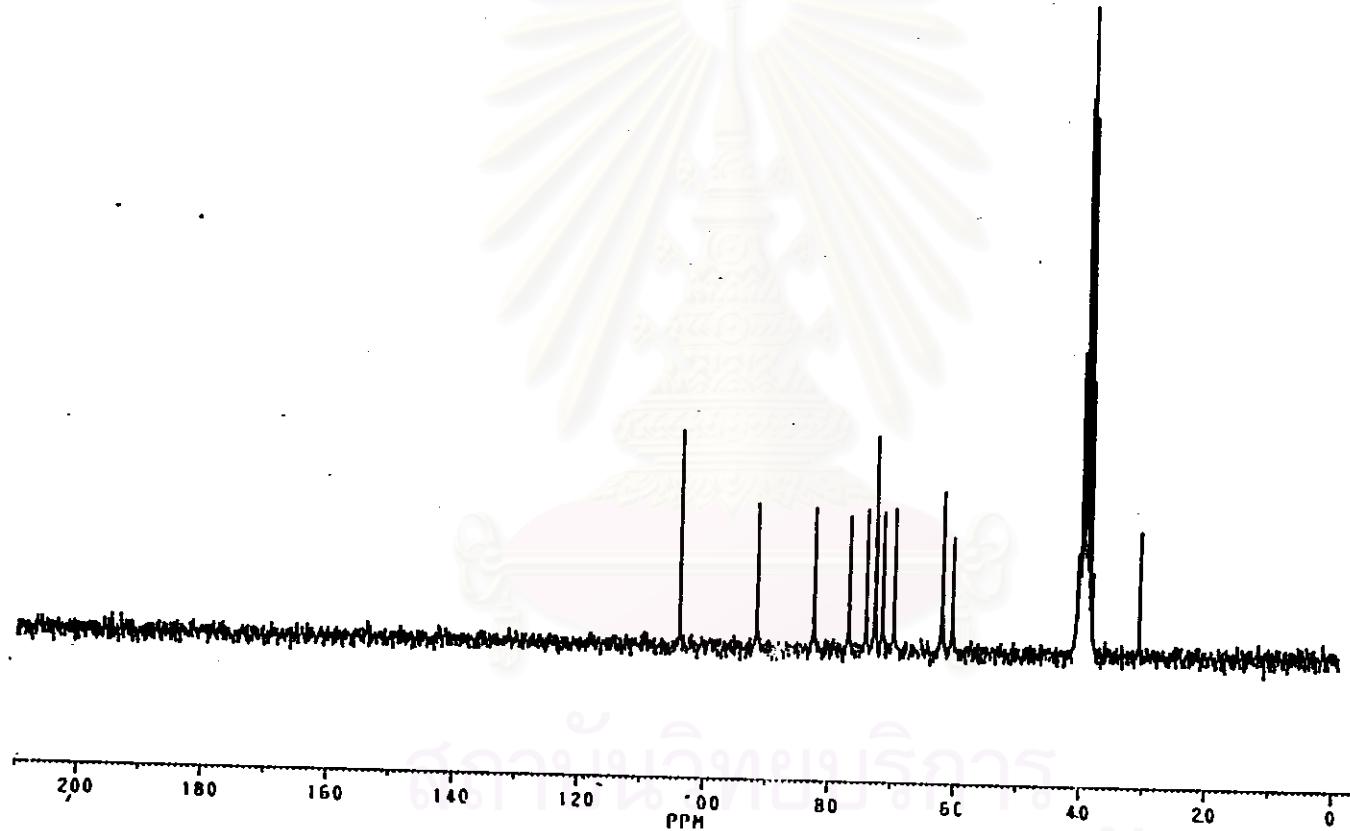


Figure 42 The ^{13}C -NMR spectrum of compound VI

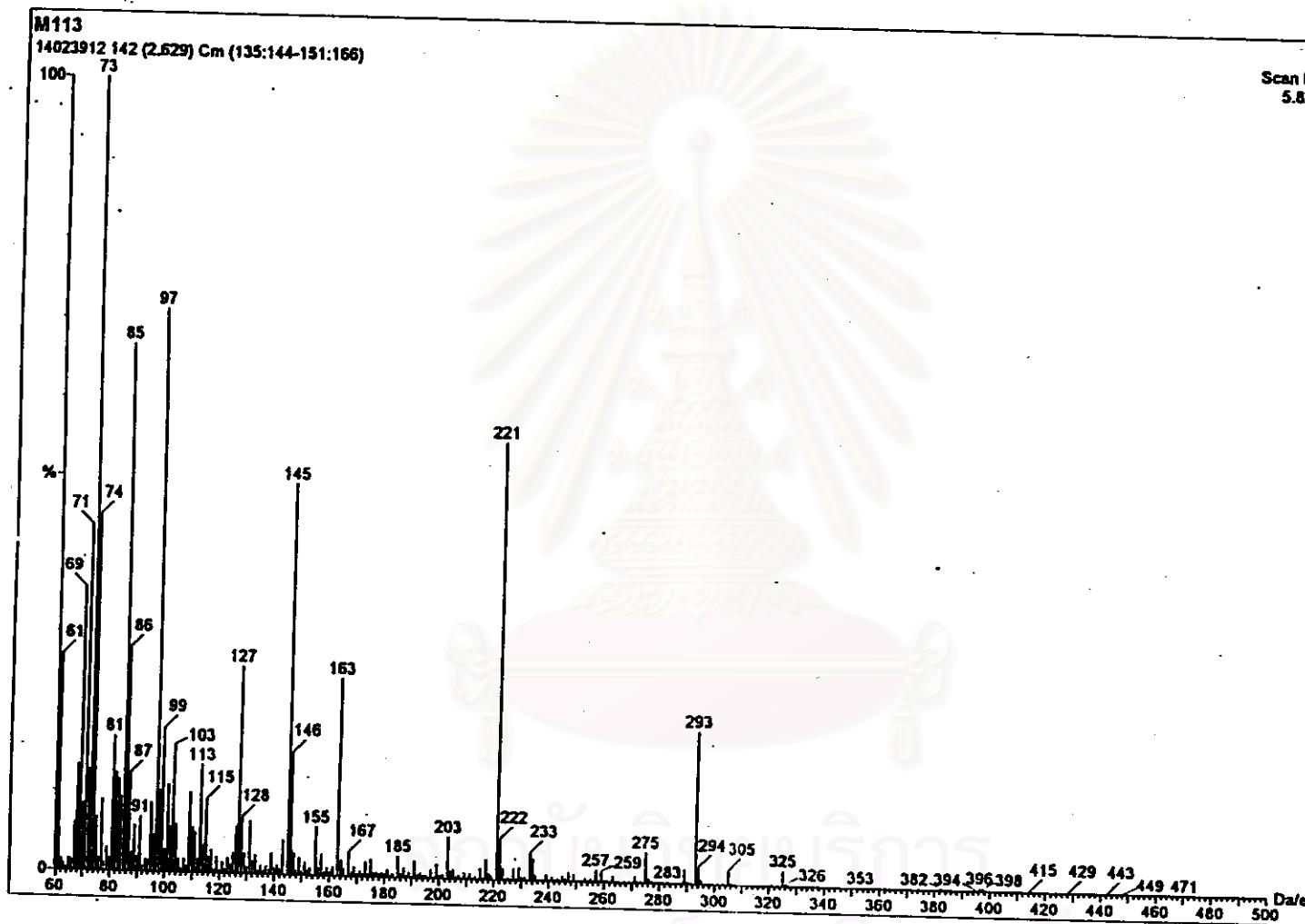


Figure 43 The mass spectrum of compound VI

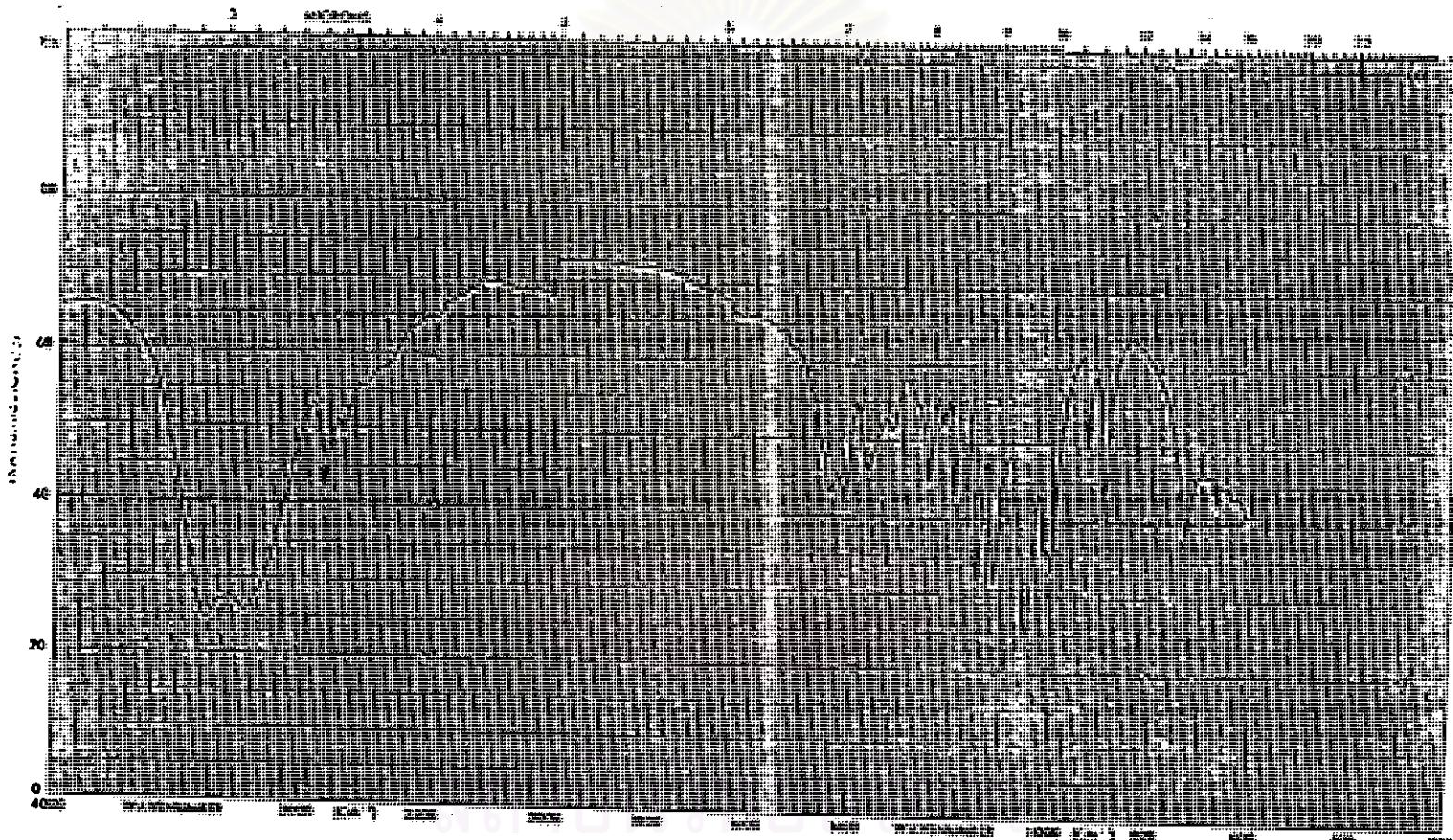


Figure 44 The IR spectrum of compound VII

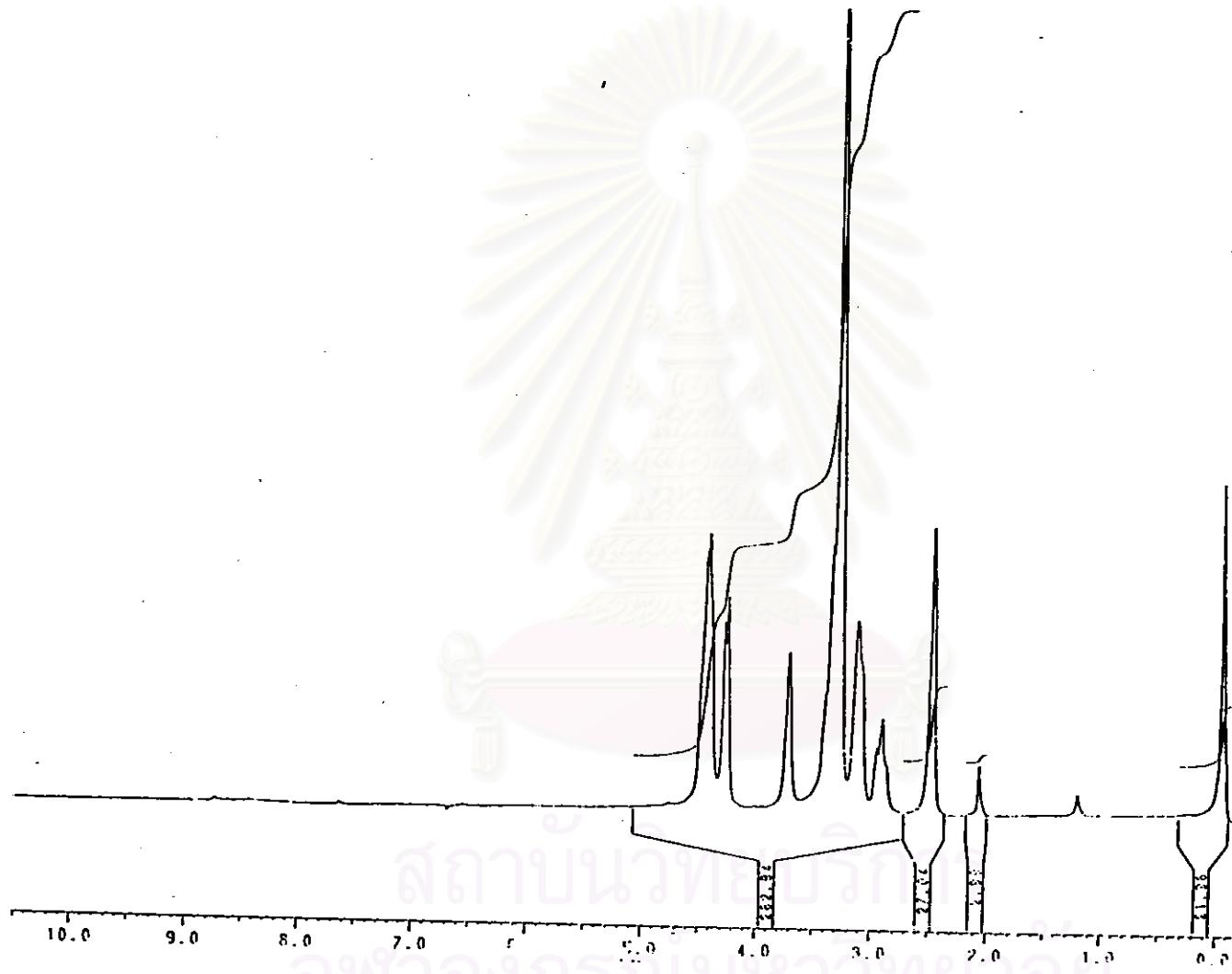


Figure 45 The $^1\text{H-NMR}$ spectrum of compound VII

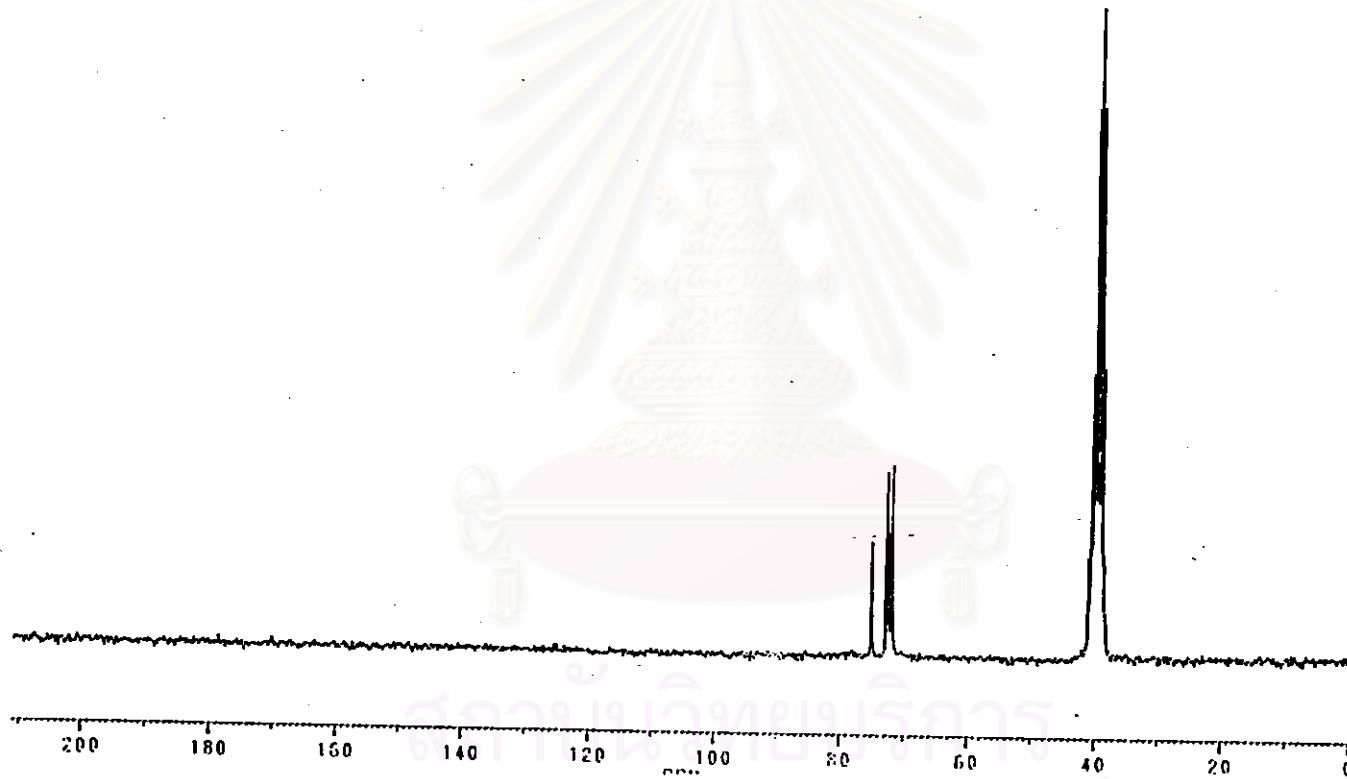


Figure 46 The ^{13}C -NMR spectrum of compound VII

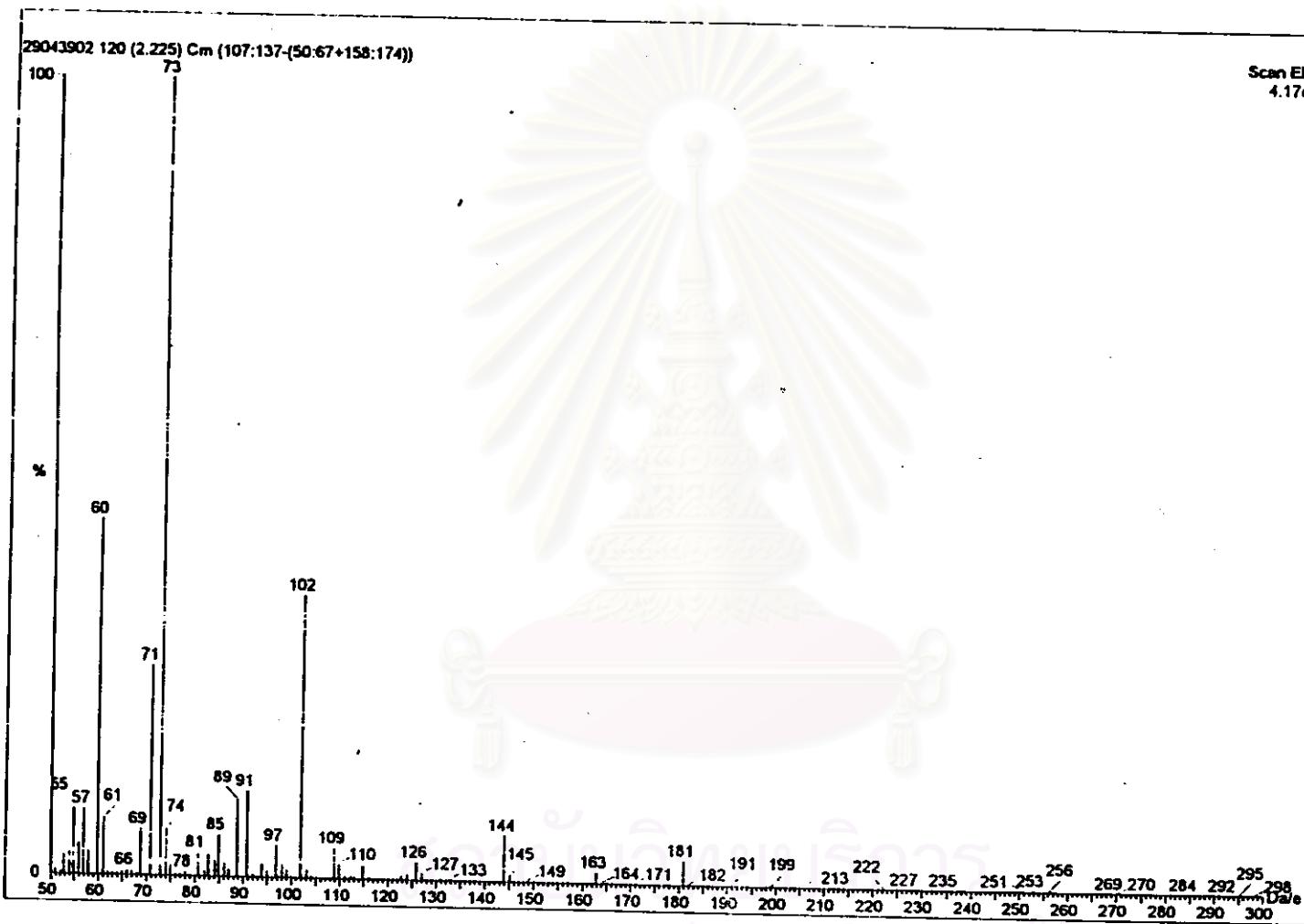


Figure 47 The mass spectrum of compound VII

VITA

Miss Jiraporn Ansomboon was born on April 13, 1972 in Bangkok, Thailand. She received a Bachelor Degree of Science (Chemistry) from Chulalongkorn University in 1994. In the same year, she was admitted into a Master Degree program in organic chemistry in the faculty of science, Chulalongkorn University.



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