

REFERENCES

1. Emission Inventory Branch, Technical Support Division, Office of Air Quality Planning and Standards. Locating and Estimating Air Emissions from Sources of Xylene. North Carolina: U.S. Environmental Protection Agency, 1994
2. Chung, H., Lee, J. S., and Ku, M. S. Feasibility of Simultaneous Measurement of Xylene Isomers and Other Hydrocarbons in p-Xylene Production Processes Using Near-Infrared Spectroscopy. Applied Spectroscopy 52 (1998): 885-889.
3. Iizuka, K. and Aishima, T. Starch Gelation Process Observed by FT-IR/ATR Spectrometry with Multivariate Data Analysis. Journal of Food Science 64 (1999): 653-658.
4. Fredericks, P. M., Lee, J. B., Osborn P. R., and Swinkels, D. A. J. Material Characterization Using Factor Analysis of FT-IR Spectra. Part 1: Results. Applied Spectroscopy 39 (1985): 311-316.
5. Tyson, L. L.; Ling, Y. C.; and Mann, C. K. Simultaneous Multicomponent Quantitative Analysis by Infrared Absorption Spectroscopy. Applied Spectroscopy 38 (1984): 663-668.
6. Haaland, D. M.; Eastering, R. G.; and Vopika, D. A. Multivariate Least-Square Method Applied to the Quantitative Spectral Analysis of Multicomponent Samples. Applied Spectroscopy 39 (1985): 73-84
7. Haward, M. Principles and Practice of Spectroscopic Calibration. New York: John Wiley & Sons, 1991.
8. Smith, A. L. Applied Infrared Spectroscopy: Fundamentals, Techniques, and Analytical Problem-solving. New York: John Wiley & Son. 1979.
9. Stuart, B. Modern Infrared Spectroscopy. New York: John Wiley & Son. 1994.
10. Brereton, R. G.; Chemometrics applications of mathematics and statistics to laboratory systems. London: Ellis Horwood Limited. 1990.

11. Ritter, G. L.; Lowry E. R.; and Isrenhour, T. L. Factor Analysis of the Mass Spectra of Mixtures. Applied Spectroscopy 48 (1976): 591-595
12. Fredericks, P. M., Lee, J. B., Osborn P. R., and Swinkels, D. A. J. Material Characterization Using Factor Analysis of FT-IR Spectra. Part 2: Mathematical and Statistical Consideration. Applied Spectroscopy 39 (1985): 311-316.
13. Pelikan, P. Application of Numerical Methods in Molecular Spectroscopy. New York: CRC Press, Inc.1988.
14. Massart, D. L., Vandeginste, B. G. M., Deming, S. N., Michotte, Y. and Kaufman, L. Chemometrics: a textbook. New York: Elsevier Science Publishers B. V.1988.
15. Kramer, R. Chemometric Techniques for Quantitative Analysis. New York: Marcel Dekker Inc.1988.
16. Haaland, D. M.; and Eastering, R. G. Application of New Least-squares Methods for the Quantitative Infrared Analysis of Multicomponent Samples. Applied Spectroscopy 36 (1982): 665-673
17. Tyson, L. L.; Ling, Y. C.; and Mann, C. K. Simultaneous Multicomponent Quantitative Analysis by Infrared Absorption Spectroscopy. Applied Spectroscopy 38 (1984): 663-668
18. Fredericks, C.; and Compton, S. Multivariate Calibration of Infrared Spectra for Quantitative Analysis Using Designed Experiment. Applied Spectroscopy 42 (1988): 865-871.
19. Sasic, S.; Jovanovic, A. A.; Kuzmanovic M.; and Jeremie, M. Quantitative analysis of the Raman spectra of mixtures of weakly interacting components by factor analysis methods. Analyst 124, (1999): 1481-1487.



APPENDICES

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APPENDIX A

FACTOR ANALYSIS PROGRAM

%This program is used for Quantitative Analysis of Mixed Xylene

%by applying Factor Analysis with FT- IR Spectra

%Modified by Pomphen Neamjoy

1. Input_Data.mat (Data Loading)

%%%%%%%%%% FILE IN DATA %%%%%%%%%%%

%%%%%%%% Load experimental data in ASCII formatted file%%%%%%%%

file_in=char('Cal.01','Cal.02', 'Cal.03','Cal.04' 'Mix.01');



2. Mixture_Matrix.mat (Manage the experimental data into matrix form)

%%%%%%%%%%Construct Calibration Set %%%%%%%%%%%

Input_Data;

d=[];

for i=1:size(file_in,1);

temp=load(file_in(i,:));

d=[S temp(:,2)];

end

%%%%%%%%%%Plot All Mixture Spectra %%%%%%%%%%%

X = temp(:,1);

Y = S (:, :);

%%%%%%%% Check number of Mixture in Mixture Matrix %%%%%%%%%

size (d)

%%%%%%%% Number of column is the number of mixtures %%%%%%%%%

3. TestPCA.mat (Principle component analysis process)

```
% PCA Process to find the number of component
% and reproduce spectra.

clear;

Mixture_Matrix;

[m,n] = size (D);

Z = d * d';

[L,lamda] = eig (Z);

%1. Determination of Activecomponent in the spectra set
% number of nonzero element of N equal to
% the number of component.

N = diag(lamda);
```

4. TestEFA.mat (Evolving factor analysis process)

```
function [e,efl,eb]=efa(d,ns)

% ns is the number of rows of this matrix
% e is the abstract concentration profiles found by EFA

clear;

Mixture_Matrix;

ns=size(d,2); % number of factor ( column ) in matrix d

x=d(:,1:ns);

[nw,ns]=size(x);

minn=min(nw,ns);

ef=ones(ns-1,ns-1).*1e-33;

eb=ef;

% forward analysis

disp('FORWARD ANALYSIS')
```

```

n=2;
while n<=ns
l=svd(x(:,1:n));
nl=size(l);
ef(1:nl,n-1)=l(1:nl,1);
efl(1:nl,n-1)=2*log10(ef(1:nl,n-1));
eflmin=min(min(efl));
efl(efl==0)=ones(size(find(efl==0)))*eflmin;
n=n+1;
end
ef=ef';
efl=efl';
% backward analysis
disp('BACKWARD ANALYSIS')
x=x(:,ns:-1:1);
n=2;
while n<=ns
l=svd(x(:,1:n));
nl=size(l);
eb(1:nl,n-1)=l(1:nl,1);
ebl(1:nl,n-1)=2*log10(eb(1:nl,n-1));
eblmin=round(min(min(ebl)));
ebl(ebl==0)=ones(size(find(ebl==0)))*eblmin;
n=n+1;
end
eb=eb';
ebl=ebl';
% evolving factor analysis plots

```

```

% 1) plot the singular values (ef,eb)

xforward=[2:ns];

xbackward=[ns :-1 :2];

hold off;

subplot (211) , plot ( xforward,ef,xbackward,eb)

plot(xforward,ef,xbackward,eb)

title('evolving factor analysis')

% 2) plot the log of eigenvalues (ef ef^2, eb eb^2)

maxvalue=round(max(max(ebl)))+1;

minvalue=round(min(min(ebl)))-1;

[maxvalue,minvalue];

minvalue= 0 ;% input(' min. value of log efa plots ? ');

efl(efl<minvalue)=ones(size(find(efl<minvalue)))*minvalue;

ebl(ebl<minvalue)=ones(size(find(ebl<minvalue)))*minvalue;

subplot (212) ,plot (xforward,efl,xbackward,ebl)

plot(xforward,efl,xbackward,ebl)

title('evolving (log eig) factor analysis')

% 3) the arranged conc. profiles for the num. of factors

%% Num. of factors to be considered ? result from PCA %%%

nf=3; ← Insert the number of component in the mixture

axis;

while nf~=0

    else,

        e(i,j)=eb(ii,jj);

    end

    if e(i,j)==0.0, e(i,j)=1.0e-30; end

end

end

```



```

plot(xforward,e)

title('arranged efa (svd) profiles')

nf=0; %('other num. of components to be considered:?' );

end

    e(i,j)=ef(i,j);

e(2:ns,:)=e(1:ns-1,:);

```

5. SLR.mat (Simple linear regression process)

```

% Iteration process to find the true result
% from the rough concentration (e) from TesEFA .mat
clear ;

Mixture_Matrix;

Test_EFA;

Ci = ans';

%Simple Linear Regression
nonnegativity = 1;

normalize = 1;

k=0;

Residual = 1;

%% Factor to be consideration before run program%%

itmax=50; % number of iteration process

n=14; % number of mixture ← Insert number of mixture ( Calibration +Test)

while norm( Residual )> 1.0e-8 & k< itmax;

    k = k+1;

    if k == 1

        Sr = (d* Ci)*(inv(Ci*Ci'));

    else

        Sr = (d* Cr)*(inv(Cr*Cr'));

```



```

end

if nonnegativity ==1;
    Sr(Sr < 0) = ones(size(find (Sr< 0))).*1e-33;
for j=1:nf
    jj=nf+1-j;
for i=1:ns-1,
    ii=ns-i;
    if ef(i,j)<=eb(ii,jj),
end
    Cr =(inv(Sr'*Sr)* Sr' *d ;
    if nonnegativity ==1;
Cr(Cr < 0) = ones(size(find (Cr< 0))).*1e-33;
    end
    if normalize ==1;
        Crsum = sum(Cr);
        for i=1:n
            Cr (:,i) = Cr (:,i)/ Crsum(i) ← Result ( Volume Fraction)
        end
    end
end

%%%% Recheck reproduce spectra.
X = temp(:,1);
Y1= Sr (:,1 );
Y2= Sr (:,2 );
Y3= Sr (:,3 );

subplot (311) , plot ( X,Y1)
subplot (312) , plot ( X,Y2) ← Result (Original Spectra)
subplot (313) , plot ( X,Y3)

```

APPENDIX B
SPECIFICATION OF THREE XYLENE ISOMER

1. ***o*-xylene (Carlo erba reagent)**

$C_6H_4(CH_3)_2$	Cas. No. 96-47-6
-FW	106.17
-Minimum assay	99 % (GLC)
-Freezing point	-25 °C ± 1.5 °C
-Boiling point	144.0 °C (b.r. ± 1.0 °C)
-Density @ 20 °C/ 4 °C	0.880 ± 0.005
-Refractive index at 20 °C/D	1.5058 ± 0.0030

2. ***p*-xylene (Carlo erba reagent)**

$C_6H_4(CH_3)_2$	Cas. No. 106-42-3
-FW	106.17
-Minimum assay	99 % (GLC)
-Melting point	13.5 °C ± 1.0 °C
-Boiling point	138.0 °C (b.r. ± 1.5 °C)
-Density@20 °C/ 4 °C	0.861 ± 0.005
-Refractive index at 20 °C/D	1.4958 ± 0.0030

3. ***m*-xylene (Carlo erba reagent)**

$C_6H_4(CH_3)_2$	Cas. No. 108-38-3
-FW	106.17
-Minimum assay	99 % (GLC)
-Melting point	-47.4 °C ± 1.5 °C
-Boiling point	139.0 °C (b.r. ± 1.0 °C)
-Density@20 °C/ 4 °C	0.864 ± 0.003
-Refractive index at 20 °C/D	1.4973 ± 0.0030

APPENDIX C

PREDICTED ERROR

1. Predicted residual error of sum of square (PRESS)

$$\text{PRESS} = \sum_{i=1}^n (y^* - y)^2$$

where:

y^*	=	The actual value
y	=	The predicted value
n	=	number of testing sample

2. Variance of prediction (s^2)

$$s^2 = \sum_{i=1}^n \frac{(y^* - y - \text{bias})^2}{n-1}$$

where:

y^*	=	The actual value
y	=	The predicted value
n	=	number of testing sample
bias	=	mean of error. This can be written as $\frac{\sum (y^* - y)}{n}$

APPENDIX D

GAS CHROMATOGRAPHY TEST

Gas Chromatography Operating Conditions:

Standard References:

p-xylene	Merck 808691	98%
m-xylene	Merck 808688	99%
o-xylene	Merck 808697	99%

Experimental Setup:

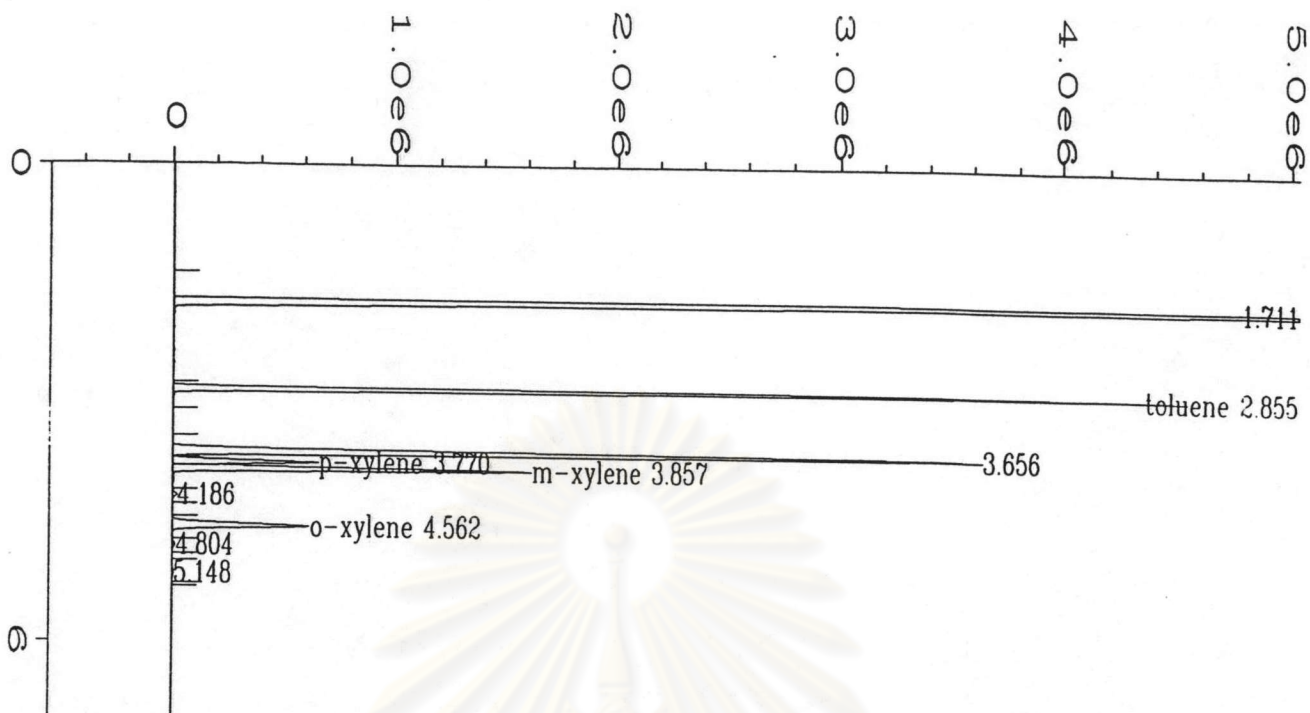
Column	Stabiwax 30 m length/ 0.53 mm ID./ 1.0 μ m df
Flow rate	3 ml/min
Injector	150 °C 1.0 μ l
Detector	FID 230 c

Gas Chromatography Test Result:

Mixed Xylene:	o-xylene	% weight	10.9
	m-xylene	% weight	23.2
	p-xylene	% weight	9.9

Remarks:

The chromatogram and detail were illustrated in page 62-63.



=====
 Internal Standard Report
 =====

Data File Name : C:\HPCHEM\1\DATA\KHANIT\VH759019.D
 Operator : Khanit
 Instrument : ANALYZER1
 Sample Name : Mixed xylene
 Run Time Bar Code:
 Acquired on : 11 Jul 01 11:42 AM
 Report Created on: 11 Jul 01 02:47 PM
 Last Recalib on : 11 Jul 01 02:39 PM
 Multiplier : 0.025

Page Number : 1
 Vial Number : 2
 Injection Number :
 Sequence Line :
 Instrument Method: XYLE_STA.
 Analysis Method : XYLE_STA.
 Sample Amount : 0.8695
 ISTD Amount : 17.71

Sig. 2 in C:\HPCHEM\1\DATA\KHANIT\VH759019.D

Ret Time	Area	Type	Width	Ref#	Amount %	Name
2.855	9821491	PV T	0.034	1-I	50.914	toluene
3.770	1870500	VV T	0.044	1	9.901	p-xylene
3.857	4658004	VV T	0.045	1	23.329	m-xylene
4.562	2209841	BV T	0.056	1	10.924	o-xylene

=====
 =====



ที่ วว 0504/ 8660

ถึง สาขาวิชา ปิโตรเคมี และวิทยาศาสตร์พอลิเมอร์ จุฬาลงกรณ์มหาวิทยาลัย

กรมวิทยาศาสตร์บริการขอส่งรายงานการตรวจ วิเคราะห์ ทดสอบ วัตถุตัวอย่าง ตาม คำร้อง

ลงวันที่ 5 มิถุนายน 2544 เลขรับ 3080 วันที่ 5 มิถุนายน 2544

ซึ่งกรมวิทยาศาสตร์ฯ ได้รับเมื่อวันที่ 5 มิถุนายน 2544

กอง เคมี

โทร. 2017227-8



รายงานการตรวจ วิเคราะห์ ทดสอบ

ชื่อวัตถุตัวอย่าง เครื่องหมาย หมายเลข
 ตามที่ผู้ส่งเรียก ที่ระบุตัวอย่าง ปฏิบัติการ

ไซลีนผสม	-	VH.759	O-xylene	ร้อยละ	10.9
(Mixed xylene)			m-xylene	ร้อยละ	23.2
			p-xylene	ร้อยละ	9.9

(นางสาวชนิษฐา พานชูวงศ์)

นักวิทยาศาสตร์ 4

หมายเหตุ ค่าธรรมเนียมการวิเคราะห์ 2,000.00 บาท (สองพันบาทถ้วน)

รายงานนี้ : - รับรองเฉพาะวัตถุตัวอย่างที่ได้ตรวจ วิเคราะห์ ทดสอบ เท่านั้น
 - ไม่รับรองวัตถุหรือสินค้าที่ใช้รายงานนี้ในการโฆษณาหรืออ้างถึง

CIRRICULUM VITAE

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