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APPENDICES

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APPENDIX A

The average grafting yield

The grafting yield of graft polymerization was calculated by following equation;

$$\text{Grafting yield (\%)} = \frac{(\text{Weight}_{\text{after grafting}} - \text{Weight}_{\text{(before grafting)}})}{\text{Weight}_{\text{(before grafting)}}} \times 100 \quad \text{A-1}$$

Table A-1. The average grafting yield of vulcanized and unvulcanized rubber as a function of time using 0.5 M PEGMA.

Grafting time (min)	Average grafting yield (%)	
	Unvulcanized rubber	Vulcanized rubber
30	0.348±0.167	0.146±0.128
60	0.522±0.077	0.237±0.024
90	0.730±0.166	0.528±0.044
120	1.369±0.380	0.791±0.032
150	1.496±0.335	0.846±0.103
180	2.592±0.518	1.096±0.197

Table A-2. The average grafting yield of vulcanized and unvulcanized rubber as a function of PEGMA concentration for 150 min.

Monomer concentration (M)	Average grafting yield (%)	
	Unvulcanized rubber	Vulcanized rubber
0.1	0.505±0.080	0.307±0.043
0.2	0.733±0.150	0.389±0.033
0.3	1.112±0.117	0.571±0.040
0.4	1.030±0.519	0.629±0.052
0.5	1.496±0.335	0.846±0.103

Table A-3. The average grafting yield of vulcanized and unvulcanized rubber as a function of time with 0.5 M VPy.

Grafting time (min)	Average grafting yield (%)	
	Unvulcanized rubber	Vulcanized rubber
30	0.113±0.167	0.000±0.000
60	0.228±0.080	0.101±0.076
90	0.284±0.056	0.153±0.042
120	0.318±0.070	0.155±0.024
150	0.353±0.045	0.189±0.085
180	0.411±0.008	0.242±0.027

Table A-4. The average grafting yield of vulcanized and unvulcanized rubber as a function of VPy concentration for 150 min.

Monomer concentration (M)	Average grafting yield (%)	
	Unvulcanized rubber	Vulcanized rubber
0.1	0.209±0.053	0.081±0.017
0.2	0.251±0.074	0.105±0.016
0.3	0.269±0.015	0.136±0.074
0.4	0.292±0.042	0.165±0.073
0.5	0.353±0.045	0.189±0.085

Table A-5. Water contact angle of vulcanized NR and unvulcanized NR after being graft copolymerized by 0.5 M PEGMA and VPy as a function of time.

Grafting time (min)	Water contact angle (degree)			
	U-g-PEGMA	V-g-PEGMA	U-g-VPy	V-g-VPy
0	86.18 ± 1.82	88.45 ± 3.72	86.18 ± 1.82	88.45 ± 3.72
30	79.38 ± 3.29	81.43 ± 1.71	81.08 ± 1.38	87.85 ± 1.13
60	74.23 ± 2.05	79.73 ± 2.80	79.75 ± 0.81	84.70 ± 1.54
90	73.80 ± 1.35	74.10 ± 1.90	78.1 ± 1.48	81.48 ± 3.67
120	72.50 ± 4.54	72.53 ± 1.58	77.70 ± 1.59	79.10 ± 1.42
150	71.28 ± 1.42	70.30 ± 4.60	75.15 ± 1.75	77.85 ± 1.47
180	59.23 ± 3.90	66.95 ± 1.77	66.95 ± 2.25	74.48 ± 3.32

Table A-6. Water contact angle of vulcanized NR and unvulcanized NR after being graft copolymerized by PEGMA and VPy for 150 min as a function of monomer concentration.

Monomer Concentration (M)	Water contact angle (degree)			
	U-g-PEGMA	V-g-PEGMA	U-g-VPy	V-g-VPy
0	86.18 ± 1.82	88.45 ± 3.72	86.18 ± 1.82	88.45 ± 3.72
0.1	76.50 ± 2.64	76.68 ± 1.13	80.00 ± 2.03	82.28 ± 2.87
0.2	76.33 ± 1.61	74.90 ± 2.22	79.03 ± 1.74	81.83 ± 2.95
0.3	74.63 ± 2.79	73.4 ± 0.94	78.25 ± 1.05	79.43 ± 1.77
0.4	73.78 ± 3.89	72.35 ± 2.09	76.90 ± 3.04	78.40 ± 0.88
0.5	71.28 ± 1.42	70.3 ± 4.60	75.15 ± 1.75	77.85 ± 1.47

Determination of Benzophenone residue

Table A-7 Standard benzophenone solution, for the calibration curve, was prepared according to the following table.

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Standard	Solution (mL)	Ethanol (mL)	Benzophenone conc ($\mu\text{g/mL}$)
S ₁	10.0 of Benzophenone solution (100 ($\mu\text{g/mL}$) ^a)	30	25
S ₂	12 of S ₁	3.0	20
S ₃	9.0 of S ₂	3.0	15
S ₄	6.0 of S ₃	3.0	10
S ₅	4.0 of S ₄	4.0	5
S ₆	4.0 of S ₅	4.0	2.5
S ₇	4.0 of S ₆	6.0	1.0
S ₈	4.0 of S ₇	4.0	0.5
S ₉	4.0 of S ₇	4.0	0.25

After reading the UV absorbance of the samples and standard benzophenone solution at $\lambda = 251$ nm. the calibration curve was plotted as shown in Figure A-1. The benzophenone concentration was determined from the calibration curve (C; $\mu\text{g/mL}$). The amount of benzophenone (B) in the original solution (5 mL) from the sampling sample (1 mL)

$$\text{Total amount of benzophenone (B)} = C (\mu\text{g/mL}) \times 5 (\text{mL}) \quad \text{A-2}$$

After that the determined of the amount of adsorbed protein/surface area was calculated as follow;

$$\text{Adsorbed protein/surface area } B_{\text{ads}} = B/\text{surface area (2 sides)} (\mu\text{g/cm}^2) \quad \text{A-3}$$

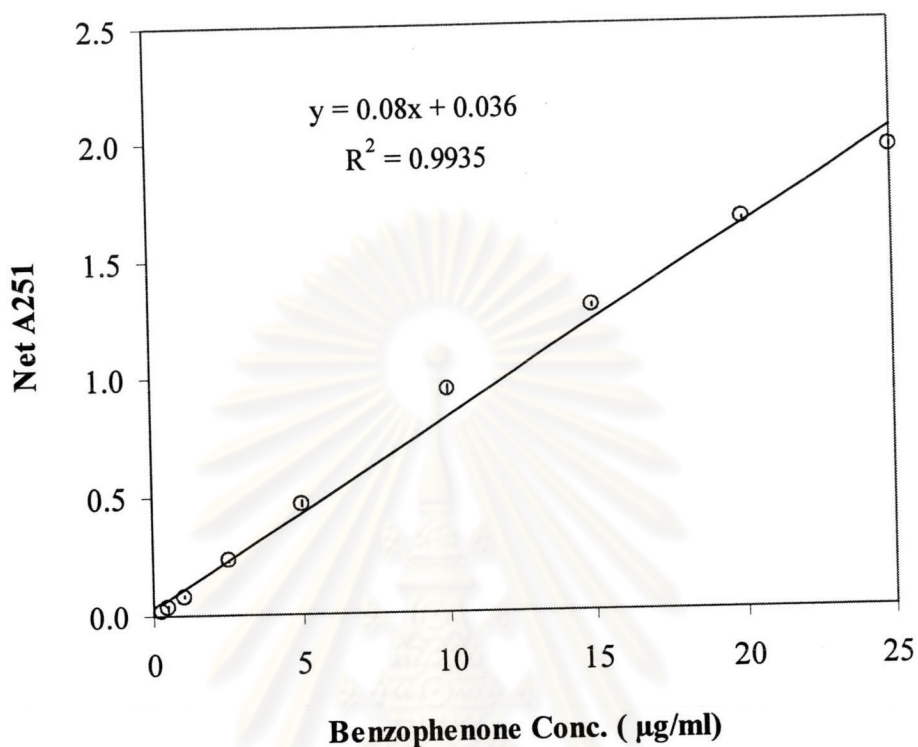


Figure A-1 A calibration curve of the amount of benzophenone as a function of UV absorbance at 251 nm.

Table A-8 The amount of benzophenone per surface area ($\mu\text{g}/\text{cm}^2$) dissolving out from vulcanized and unvulcanized NR before and after graft copolymerization.

Grafting condition	Amount of Benzophenone residue ($\mu\text{g}/\text{ml}$)	
	Unvulcanized rubber	Vulcanized rubber
Before grafting	32.14 ± 1.77	42.91 ± 1.78
After grafting for 30 min	5.09 ± 0.45	3.58 ± 0.84
After grafting for 180 min	4.80 ± 1.20	4.30 ± 0.73

APPENDIX B

Bicinchoninic Acid Assay

Bicinchoninic acid assay is a method used for determination of the amount of proteins. The standard reagents used in this method are reagent A, reagent B and reagent C. Reagent A consists of an aqueous solution of $\text{Na}_2\text{tartrate}$, Na_2CO_3 , NaHCO_3 in 0.2 M NaOH , pH 11.25. Reagent B is 4% (W/V) bicinchoninic acid solution, pH 8.5. Reagent C is 4% $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ in deionized water.

The principle of the bicinchoninic assay (BCA) relies on the formation of a Cu^{2+} -protein complex under alkaline conditions, followed by reduction of the Cu^{2+} to Cu^{1+} . The amount of reduction is proportional to protein present. It has been shown that the peptide bond is able to reduce Cu^{2+} to Cu^{1+} . BCA forms a purple-blue complex with Cu^{1+} in alkaline environments, thus providing a basis to monitor the reduction of alkaline Cu^{2+} by proteins.³⁰ Figure B-1 shows complexation between bicinchoninic acid and Cu^{1+} .

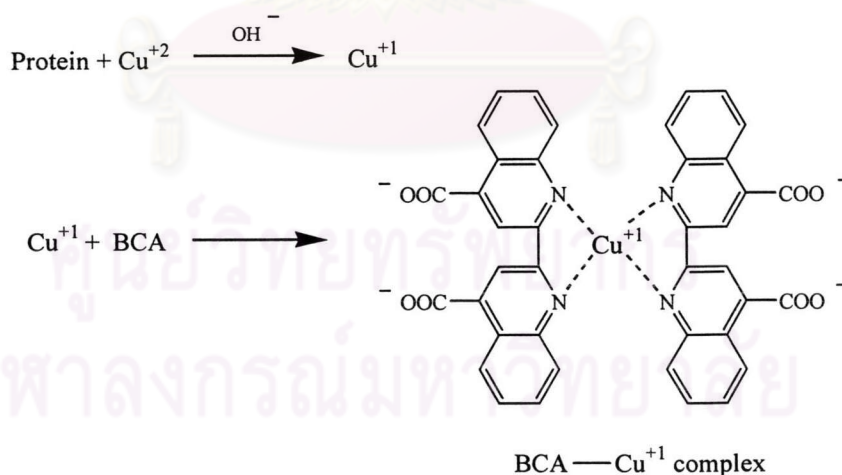


Figure B-1. Formation of purple complex between BCA and cuprous ion generated from the biuret reaction.

Calculation of Protein Adsorption

Table B-1 Standard BSA solution, for the calibration curve.

Standard	Solution (mL)	SDS (mL)	BSA conc ($\mu\text{g/mL}$)
S ₁	0.5 of BSA (1000 ($\mu\text{g/mL}$) ^a)	4.5	100
S ₂	4.0 of S ₁	4.0	5.0
S ₃	4.0 of S ₂	4.0	25
S ₄	4.0 of S ₃	6.0	10
S ₅	4.0 of S ₄	4.0	5
S ₆	4.0 of S ₅	4.0	2.5
S ₇	4.0 of S ₆	6.0	1.0
S ₈	4.0 of S ₇	4.0	0.5

a : standard BSA was pipette from 1 mg/mL ampule

After reading the UV absorbance of the samples and standard BSA solution at $\lambda = 562 \text{ nm.}$, the result was then calculated for the net absorbance by subtracting the absorbance of the blank (SDS).

$$\text{Net } A_{562} = \text{recorded } A_{562} - A_{562}(\text{blank}) \quad \text{B-1}$$

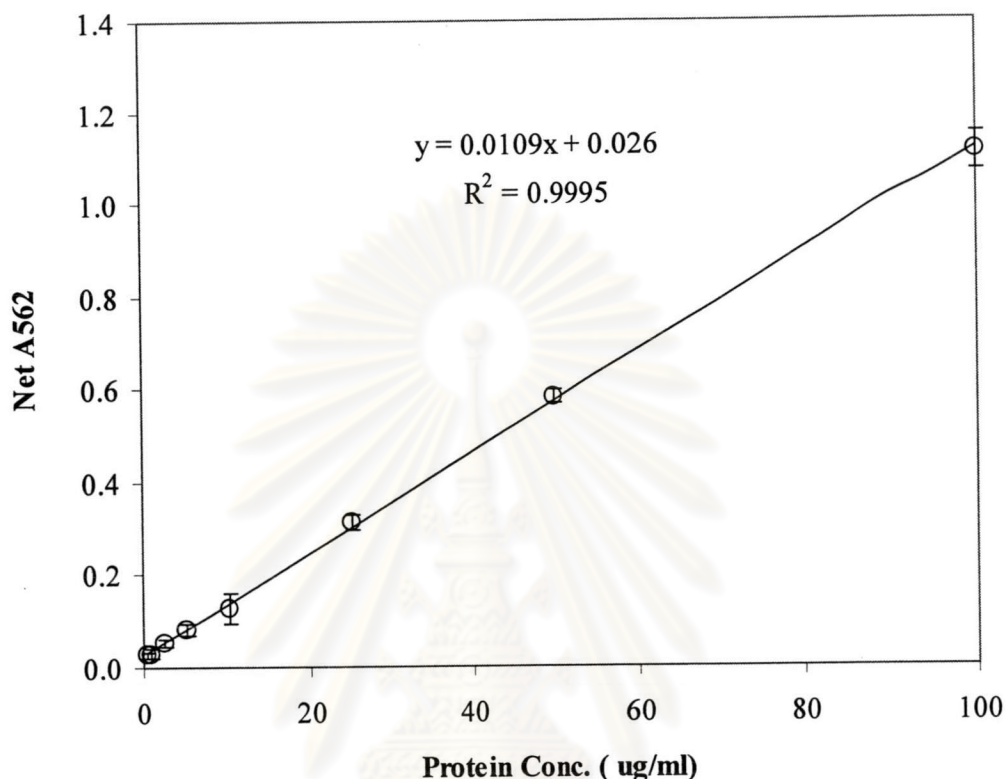


Figure B-2 A calibration curve of the amount of albumin adsorbed and the absorbance obtained from BCA microassay.

The protein concentration (C ; $\mu\text{g/mL}$) in each well was determined from the calibration curve. The total amount of protein (P) in the original solution (1 mL) was calculated from the sampling sample (100 μL) + BCA working solution (100 μL)

$$\text{Total amount of protein (P)} = \frac{C (\mu\text{g/mL}) \times 200 (\mu\text{L})}{1000 (\mu\text{L/mL})} \times \frac{1000 (\mu\text{L})}{100 (\mu\text{L})} \quad \text{B-2}$$

$$\text{Adsorbed protein/surface area } P_{\text{ads}} = P/\text{surface area (2 sides)} (\mu\text{g/cm}^2) \quad \text{B-3}$$

Table B-2 The amount of plasma protein adsorption per surface area ($\mu\text{g}/\text{cm}^2$) of Vulcanized and unvulcanized rubber before and after graft copolymerization with PEGMA and VPy as function of time.

Grafting time (min)	The amount of plasma protein adsorption ($\mu\text{g}/\text{cm}^2$)			
	U-g-PEGMA	V-g-PEGMA	U-g-VPy	V-g-VPy
0	26.396 \pm 3.980	12.834 \pm 1.570	26.396 \pm 3.980	12.834 \pm 1.570
30	8.380 \pm 4.100	11.184 \pm 4.100	19.419 \pm 0.711	12.437 \pm 3.115
60	3.286 \pm 0.999	10.648 \pm 0.999	13.673 \pm 3.799	11.069 \pm 1.375
90	0.596 \pm 0.531	6.424 \pm 0.531	7.807 \pm 3.174	8.016 \pm 3.497
120	-0.304 \pm 1.153	2.275 \pm 1.153	5.536 \pm 0.887	7.671 \pm 2.923
150	-1.027 \pm 0.844	0.551 \pm 0.844	3.046 \pm 0.431	6.449 \pm 1.053
180	-0.767 \pm 2.839	-0.174 \pm 2.839	2.612 \pm 0.862	3.743 \pm 0.526

Table B-3 The amount of plasma protein adsorption per surface area ($\mu\text{g}/\text{cm}^2$) of Vulcanized and unvulcanized rubber before and after graft copolymerization with PEGMA and VPy as function of monomer concentration.

Monomer Concentration (M)	The amount of plasma protein adsorption ($\mu\text{g}/\text{cm}^2$)			
	U-g-PEGMA	V-g-PEGMA	U-g-VPy	V-g-VPy
0	26.396 \pm 3.980	12.834 \pm 1.570	26.396 \pm 3.980	12.834 \pm 1.570
0.1	4.939 \pm 0.612	10.596 \pm 2.358	8.102 \pm 4.347	11.220 \pm 1.620
0.2	4.536 \pm 1.669	7.064 \pm 3.177	7.013 \pm 3.085	10.447 \pm 1.966
0.3	0.099 \pm 0.642	3.261 \pm 0.714	5.723 \pm 1.156	9.134 \pm 1.083
0.4	-0.347 \pm 0.311	1.122 \pm 1.059	3.269 \pm 1.077	8.443 \pm 0.547
0.5	-1.027 \pm 1.794	0.551 \pm 0.774	3.046 \pm 0.431	6.449 \pm 1.053

APPENDIX C

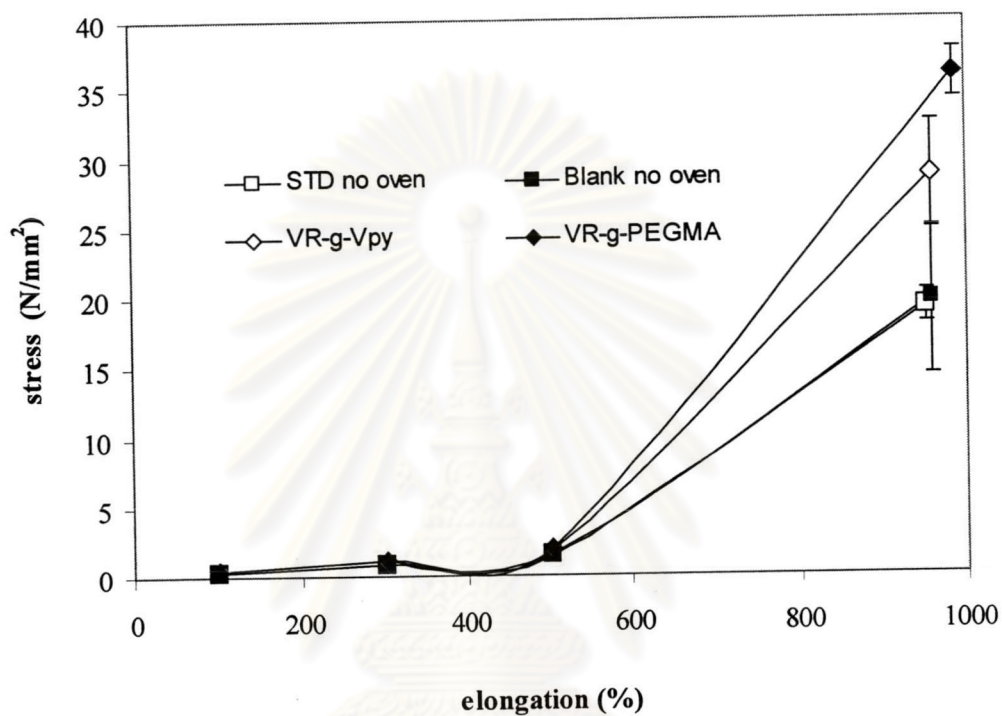


Figure C-1 Stress-strain curves of NR latex films before and after graft copolymerization using 0.5 M PEGMA and VPy for 150 min

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VITAE

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