

การสังเคราะห์อนุพันธ์ของกรดแอลฟ้าไอดรอกซีฟอสฟอโนิก

นางสาวจินตนา นามมูลน้อย

ศูนย์วิทยทรัพยากร
มหาวิทยาลัย

วิทยานิพนธ์นี้เป็นส่วนหนึ่งของวิชาการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรมหาบัณฑิต

สาขาวิชาเคมี ภาควิชาเคมี

คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย

ปีการศึกษา 2546

ISBN: 974-17-5169-9

ลิขสิทธิ์ของจุฬาลงกรณ์มหาวิทยาลัย

SYNTHESIS OF α -HYDROXYPHOSPHONIC ACID DERIVATIVES

Miss Jintana Nammoonnoy

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

A thesis Submitted in Partial Fulfillment of the Requirements
for the Degree of Master of Science in Chemistry

Department of Chemistry

Faculty of Science

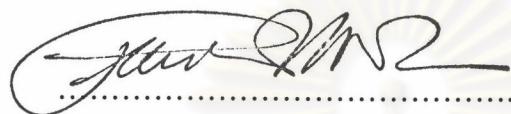
Chulalongkorn University

Academic Year 2003

ISBN: 974-17-5169-9

Thesis title Synthesis of α -Hydroxyphosphonic Acid Derivatives
By Miss Jintana Nammoonnoy
Field of Study Chemistry
Thesis Advisor Assistant Professor Worawan Bhanthumnavin, Ph.D.

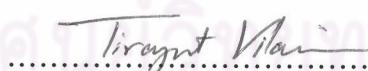
Accepted by the Faculty of Science, Chulalongkorn University in Partial
Fulfillment of the Requirements for the Master's Degree


..... Dean of Faculty of Science
(Professor Piamsak Menasveta, Ph.D.)

Thesis Committee


..... Chairman
(Professor Udom Kokpol, Ph.D.)


..... Thesis Advisor
(Assistant Professor Worawan Bhanthumnavin, Ph.D.)


..... Member
(Assistant Professor Tirayut Vilaivan, D.Phil.)


..... Member
(Soamwadee Chaianansutcharit, Ph.D.)

จินตนา นามมูลน้อย: การสังเคราะห์อนุพันธ์ของกรดแอลฟ้าไฮดรอกซีฟอสฟonic

(SYNTHESIS OF α -HYDROXYPHOSPHONIC ACID DERIVATIVES)

อ. ที่ปรึกษา: ผศ. ดร. วราวรรณ พันธุ์มนนาวิน

121 หน้า, ISBN: 974-17-5169-9

ได้เตรียมแอลฟ้าไฮดรอกซีฟอสฟonic เตามากจากปฏิกิริยาระหว่างแอลดีไฮด์กับไดออกซิลฟอสไฟต์โดยใช้ตัวเร่งปฏิกิริยาหลายชนิดได้แก่ ไตรเอтиlamine อะลูมิโน ทิเทเนียมไอกาโรฟอกไซด์ และลิเทียมอะลูมิเนียมไอกาโรด พบว่าตัวเร่งปฏิกิริยาที่เหมาะสมที่สุดในการเตรียมแอลฟ้าไฮดรอกซีฟอสฟonic เป็นไดออกซิลฟอสไฟต์ได้แก่ ลิเทียมอะลูมิเนียมไอกาโรด โดยให้เปอร์เซ็นต์ผลิตภัณฑ์ประมาณ 43-84 % นอกจากนี้ยังได้ศึกษาปฏิกิริยาระดับลิติกอะซิมเมตريكไอกาโรฟอสฟonic เหล่านี้ของแอลดีไฮด์กับไดออกซิลฟอสไฟต์โดยใช้ตัวเร่งปฏิกิริยาที่เป็นสารประกอบเชิงช้อนของลิเทียมอะลูมิเนียมไอกาโรดกับลิแกนด์ต่างๆ ได้แก่ ลิแกนด์กสุ่มชิฟเบส เปปไทด์ชิฟเบสและกลุ่มเอ็นชาลิชิลเบตาอะมิโนแอลกอฮอล์ จากผลการทดลองพบว่าตัวเร่งปฏิกิริยาระหว่างอัลดีไฮด์กับไดออกซิลฟอสไฟต์ที่มีประสิทธิภาพที่สุดคือตัวเร่งปฏิกิริยาที่เตรียมได้จากปฏิกิริยาระหว่างลิเทียมอะลูมิเนียมไอกาโรดกับลิแกนด์กลุ่มเอ็นชาลิชิลเบตาอะมิโนแอลกอฮอล์โดยให้เปอร์เซ็นต์ผลิตภัณฑ์ประมาณ 50-75 % และสามารถให้สารที่มีเปอร์เซ็นต์อิเอนชิโนเมอริกเอ็กเซลปานกลางคือประมาณ 32-70 % ซึ่งเปอร์เซ็นต์อิเอนชิโนเมอริกเอ็กเซลวิเคราะห์โดยใช้เทคนิคแก๊สโครมაตอกราฟีบันไดรัลคลอเลมัน จากการศึกษาพบว่าความเก lokale ของหมู่แทนที่ในส่วนที่เป็นเอ็นชาลิชิลและเบตาอะมิโนแอลกอฮอล์จะมีผลต่อการเหนี่ยวนำให้เกิดซีเลกติวิตี นอกจากนี้ยังพบว่าปฏิกิริยาของอะลิฟาติกอัลดีไฮด์จะให้อิเอนชิโนเมอริกเอ็กเซลตี่สูงกว่าของอะโรมาติกอัลดีไฮด์

ภาควิชา.....เคมี.....รายมีอชื่อนิสิต.....
สาขาวิชา.....เคมี.....รายมีอชื่ออาจารย์ที่ปรึกษา.....
ปีการศึกษา.....2546.....รายมีอชื่ออาจารย์ที่ปรึกษาร่วม.....

4372232523: MAJOR CHEMISTRY

KEY WORD: α -HYDROXYPHOSPHONIC ACID, HYDROPHOSPHONYLATION

JINTANA NAMMOONNOY: SYNTHESIS OF α -HYDROXYPHOSPHONIC ACID DERIVATIVES

THESIS ADVISOR: ASSISTANT PROFESSOR WORAWAN BHANTHUMNAVIN, Ph.D.

121 pp. ISBN: 974-17-5169-9

The synthesis of α -hydroxyphosphonates by the reaction between dialkyl phosphites and aldehydes using various catalysts such as triethylamine, alumina, titanium isopropoxide, and lithium aluminium hydride were studied. The use of lithium aluminium hydride as catalyst resulted in the desired products in moderate to high yields (43-84%). The asymmetric synthesis of α -hydroxyphosphonates by asymmetric hydrophosphonylation reaction of aromatic aldehyde with dialkyl phosphites in the presence of a catalytic amount of complexes of lithium aluminium hydride and chiral ligands was carried out. The ligands of interest are chiral Schiff's base, peptide Schiff's base, and *N*-salicyl- β -aminoalcohol ligands. The reaction of aldehydes and dialkyl phosphites in the presence of Li-Al-*N*-salicyl- β -aminoalcohol complexes proceeded efficiently to give the corresponding α -hydroxyphosphonate in moderate yields (50-75%) and moderate enantioselectivity (32-70% *ee*). The enantiomeric excess was observed by chiral gas chromatography. The bulkiness of substituents on the salicyl and β -amino alcohol moiety plays a significant role in the induction of enantioselectivities. In addition, the aliphatic aldehydes gave product in higher enantiomeric excess than aromatic aldehydes.

ศูนย์วิทยกรรมการ
จุฬาลงกรณ์มหาวิทยาลัย

Department.....Chemistry.... Student's signature..... Jintana Nammoonnoy...

Field of Study.....Chemistry.... Advisor's signature..... Mr. Bleale.....

Academic year.....2003..... Co-advisor's signature.....

ACKNOWLEDGEMENT

The author wishes to express her deepest gratitude to her advisor, Assistant Professor Worawan Bhanthumnavin, for her help, advice, encouragement, and patience throughout the course of this thesis. Her kind suggestions gave the author the strength and perseverance to do this thesis. Although, there were obstacles, this thesis could finally be completed by her advice. As with the highest honor dedicated to Assistant Professor Tirayut Vilaivan, the author received his help in such a way that he helped her to pursue many of the specific problems such as selection of catalysts used in the reactions, and suggestion of reaction conditions.

Her appreciation is also extended to Professor Udom Kokpol as the chairman, and to Assistant Professor Tirayut Vilaivan and Dr. Soamwadee Chaianansutcharit, the thesis examiners, for their valuable constructive comments and suggestions.

The author would like to thank Associate Professor Somsak Ruchirawat of Chulabhorn Research Institute (CRI) for his courtesy of instrument usage (200 MHz NMR spectrometer and mass spectrometry facilities). In addition, the author would like to thank Dr. Aroonsiri Shitangkoon and Miss Jirawit Yanchinda for their kind hospitality in the determination of enantiomeric excess by chiral GC. Moreover, the author would like to thank Dr. Yongsak Sritana-anant for his kindness, advice, and suggestions. Furthermore, the permission to use low pressure vacuum system by Assistant Professor Varawut Tangpasuthadol and the help in operating such system by Miss Weerawan Sunsaneeyametha is gratefully acknowledged.

The author gratefully thanks the Development and Promotion for Science and Technology Talents Project of Thailand (DPST) for granting her a financial support to study and to this research. Research financial supports from the Ministry of University Affairs are greatly appreciated.

Special thanks are extended to Miss Paethong Srikaenjun for providing the stimulating idea of using heterobimetallic complexes in this reaction. Special thanks are also extended to Miss Woraluk Mansawat, Miss Saowaluk Chaleawlerttumpon, Miss Siriporn Jiwpanich, and Mr. Vorawit Banphavichit for providing some catalysts or starting materials.

Finally, the author would like to thank her parents, her friends, WB, YS and TV group members for strong moral support.

CONTENTS

	Pages
Abstract in Thai.....	iv
Abstract in English.....	v
Acknowledgement.....	vi
Contents.....	vii
List of Figures.....	
List of Tables.....	
List of Abbreviations.....	
CHAPTER I: INTRODUCTION.....	1
1.1 α -Hydroxyphosphonic acid derivatives.....	1
1.2 Non asymmetric synthesis of α -hydroxyphosphonates.....	2
1.3 Asymmetric synthesis.....	4
1.4 Asymmetric hydrophosphonylation.....	8
1.5 Objective of this research.....	23
CHAPTER II: EXPERIMENTAL.....	24
2.1 General and materials.....	24
2.2 Synthesis of dialkyl phosphite.....	25
2.3 General procedure for the synthesis of racemic α -hydroxyphosphonates.....	26
2.3.1 General procedure for the synthesis of racemic α - hydroxyphosphonate using triethylamine.....	26
2.3.2 Synthesis of racemic α -hydroxyphosphonate using $Ti(O^iPr)_4$	30
2.3.3 General procedure for the synthesis of racemic α -hydroxyphosphonate <i>via</i> lithium aluminium hydride catalyzed reaction.....	31
2.4 General method for optical purity determination of α -hydroxyphosphonates.....	42
2.4.1 Analysis of α -hydroxyphosphonates using a chiral GC column.....	42

CONTENTS (Continued)

	Pages
2.4.2 General procedure for preparation of MTPA ester of α-hydroxyphosphonates	42
2.5 General procedure for asymmetric synthesis of α-hydroxyphosphonates.....	43
2.5.1 Representative procedure for Pudovik reactions with $Ti(O^iPr)_4 : N$ -salicyl-β-aminoalcohol complex.....	43
2.5.2 Representative procedure for Pudovik reactions with $LiAlH_4 : N$ -salicyl-β-aminoalcohol complex.....	43
CHAPTER III: RESULTS AND DISCUSSION.....	45
3.1 Analytical methods for the determination of enantiomeric purity of α-hydroxyphosphonates.....	45
3.1.1 NMR spectroscopy.....	45
3.1.1.1 chiral derivatizing agents (CDAs).....	45
3.1.1.2 chiral solvating agents (CSAs).....	47
3.1.1.3 chiral lanthanide shift reagents (CLSRs).....	48
3.1.2 Chromatographic methods.....	48
3.1.2.1 Gas chromatography.....	48
3.1.2.2 High performance liquid chromatography.....	49
3.2 The results of analytical methods for the determination of enantiomeric purity of α-hydroxyphosphonates.....	49
3.2.1 NMR spectroscopic analysis.....	50
3.2.1.1 Analysis employing chiral solvating agents.....	50
3.2.1.2 Analysis employing chiral derivatizing agents.....	51
3.2.2 Chiral GC analysis.....	55
3.3 The synthesis of α-hydroxyphosphonates.....	56
3.3.1 Synthesis of racemic α-hydroxyphosphonates.....	56
3.3.2 Asymmetric synthesis of α-hydroxyphosphonates.....	58
3.3.2.1 Pudovik reactions promoted by titanium complexes.....	58

CONTENTS (Continued)

	Pages
3.3.2.2 Asymmetric Pudovik reactions promoted by heterobimetallic complexes.....	66
CHAPTER IV: CONCLUSION.....	80
REFERENCES.....	81
APPENDIX.....	88
VITAE.....	121



**ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย**

LIST OF FIGURES

Figures	Pages
1.1 Phosphates, phosphonates, and α -substituted phosphonates.....	2
1.2 The Abramov reaction.....	2
1.3 The Pudovik reaction.....	3
1.4 Phosphite-phosphonate tautomers.....	3
1.5 The addition of dialkyl phosphite to unsaturated molecules and by using TMG as catalysts.....	4
1.6 Known disconnections to nonracemic α -hydroxyphosphonates.....	9
1.7 Synthesis of α -hydroxyphosphonates from <i>O</i> -benzyl lactaldehyde.....	10
1.8 The addition of phosphorus nucleophile to aldehyde derived from amino acids.....	10
1.9 The stereoselective ring opening of homochiral acetal 26 with triethyl phosphite.....	11
1.10 The addition of oxazaphosphite to aldehyde.....	12
1.11 The addition of the bicyclic chiral phosphorus acid diamide to aldehyde.....	13
1.12 Heterobimetallic catalysts.....	15
1.13 The reduction of diethyl α -ketophosphonates with catecholborane 44 and oxazaborolidine 45	19
3.1 Preparation of diastereomeric Mosher's esters.....	46
3.2 Preparation of <i>O</i> -methyl mandelate esters.....	47
3.3 Preparation of Naproxan [®] and Ibuprofen [®] derivatives of the diethyl hydroxyphosphonates.....	47
3.4 The ¹ H NMR spectra of: (a) α -hydroxyphosphonate 50b before derivatizing with Mosher's acid; (b) (<i>R</i>)-MTPA ester of 50b	52
3.5 The ³¹ P NMR spectra of: (a) α -hydroxyphosphonate 50b before derivatizing with Mosher's acid; (b) (<i>R</i>)-MTPA ester of 50b	54
3.6 GC chromatograms on a 15.092 m long, 0.25mm i.d., capillary, coated with 0.25 μ m film of 10% BSiMe in PS255. Temperature program: 140°C (12.5 min) 15°C/min 240°C of: (a) racemic mixture of 50aa ; (b) product mixture of 50aa obtained from a catalytic asymmetric reaction	55

LIST OF FIGURES (Continued)

Figures	Pages
3.7 Schematic representation for the catalytic Pudovik reaction catalyzed by metal alkoxides.....	59
3.8 Various interesting chiral ligands.....	63
3.9 Amino alcohol ligands synthesized in this laboratory.....	64
3.10 The heterobimetallic complex of aluminium lithium BINOL (ALB)....	66
3.11 The Michael addition reaction of cyclic enone and malonate.....	67
3.12 racemic α -hydroxyphosphonates.....	71



LIST OF TABLES

Tables	Pages
1.1 Diastereoselectivity of the addition of (<i>S</i>)- 21	11
1.2 Enantiomeric excess of the phosphonates.....	12
1.3 The enantioselectivities in the hydrophosphonylation by using titanium alkoxides as catalysts.....	14
1.4 Catalytic asymmetric hydrophosphonylation of aldehydes catalyzed by ALB.....	16
1.5 Enantioselectivities in the reactions between DMHP (1 mmol) and ArCHO (1 mmol) catalyzed by 38 at 298 K in THF solvent.....	17
1.6 Enantioselective reduction of diethyl α -ketophosphonates 40 in the presence of chiral catalyst (<i>S</i>)- 41 or (<i>R</i>)- 41 and borane.....	18
1.7 Enantioselective reduction of α -ketophosphonates 43 in the presence of catecholborane 44 and (<i>S</i>)-5,5-diphenyl-2-butyl-3,4-propano-1,3,2-oxazaborolidine 45	20
1.8 Enantioselective reduction of α -ketophosphonates using (-)-Ipc ₂ B-Cl 47 in THF at -20 °C.....	21
1.9 Asymmetric dihydroxylation of a racemic mixture of 2(<i>E</i>)-alkenylphosphonates 48 with AD- <i>mix</i> - β or - α reagent.....	22
1.10 The oxidation of phosphonates anion with oxaziridine.....	23
3.1 Synthesis of racemic α -hydroxyphosphonates using triethylamine.....	57
3.2 Hydrophosphonylation of dimethyl phosphite (DMHP) and benzaldehyde in THF under various amount of DMHP and Ti(O <i>i</i> Pr) ₄	60
3.3 Hydrophosphonylation of benzaldehyde with diethyl phosphite by various Lewis acid catalysts at room temperature.....	62
3.4 Asymmetric hydrophosphonylation of dimethyl phosphite to aldehyde with titanium-complex catalyst.....	65
3.5 Hydrophosphonylation of diethyl phosphite and benzaldehyde in difference solvents.....	68
3.6 Synthesis of racemic α -hydroxyphosphonates using LiAlH ₄	69

LIST OF TABLES (Continued)

Tables	Pages
3.7 Reactions of benzaldehyde and diethyl phosphite with various catalysts.....	72
3.8 Enantioselective additions of diethyl phosphite to benzaldehyde with the complex of LAH and (<i>S</i>)- 71k ligand.....	75
3.9 Hydrophosphonylation of diethyl phosphite and benzaldehyde in difference solvents in the presence of chiral ligand (<i>S</i>)- 71k	76
3.10 Catalytic asymmetric hydrophosphonylation of various aldehydes catalyzed by LAH -(<i>S</i>)- 71k complex.....	78



 ศูนย์วิทยทรัพยากร
 จุฬาลงกรณ์มหาวิทยาลัย

List of Abbreviations

Ar	aromatic	m	multiplet
bs	broad singlet	Me	methyl
Bu	butyl	MHz	megaHertz
°C	degree Celsius	mg	milligram(s)
CDA	chiral derivatizing agent	mL	milliliter(s)
CDCl ₃	deuterated chloroform	mmol	millimole
CLSR	chiral lanthanide shift reagent	m.p.	melting point
CSA	chiral solvating agent	MS	molecular sieves
d	day(s)	MTPA	α-methoxy-α-(trifluoro methyl)phenylacetic acid,
d	doublet (NMR)		Mosher's acid
DBU	1,8-diazabicyclo[5.4.0]undec- 7-ene	NMR	nuclear magnetic resonance
DCC	dicyclohexylcarbodiimide	Tf	trifluoromethanesulfonyl
DMAP	4-N,N-dimethylamino pyridine	Ph	phenyl
DEHP	diethyl phosphite	ppm	parts per million
DMHP	dimethyl phosphite	Pr	propyl
dd	doublet of doublet (NMR)	rt	room temperature
de	diastereomeric excess	s	singlet
dr	diastereomeric ratio	t	triplet
ds	diastereoselectivity	THF	tetrahydrofuran
ee	enantiomeric excess	TBD	1,5,7-triazabicyclo[4.4.0]dec- 5-ene
equiv	equivalent	TMG	trimethylguanidine
Et	ethyl	μL	microliter
g	gram	δ	chemical shift
h	hour(s)	%	percent
Hz	Hertz		
J	coupling constant		
K	Kelvin		
lit.	literature		
LAH	lithium aluminium hydride		