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APPENDICES

ศูนย์วิทยทรัพยากร
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APPENDIX A

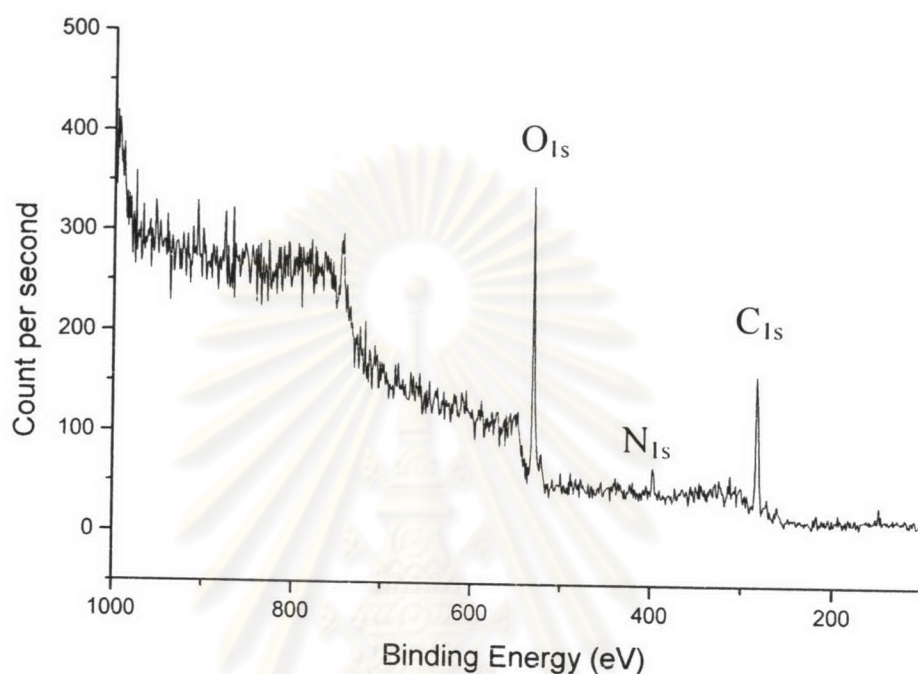


Figure A-1 XPS survey spectrum of chitosan film immersed in solvent combined NaI.

Table A-1 XPS atomic composition of modified and unmodified chitosan film

Surface type	Percent atomic composition					
	C	O	N	I	S	Na
Chitosan-film	69.8	25.3	4.9	-	-	-
QAC-film	63.0	28.6	6.8	1.6	-	-
Chitosan film (QAC-control)	67.1	26.9	5.9	-	-	-
SFC-film	62.2	29.9	4.9	-	0.9	2.1

APPENDIX B

Bicinchoninic Acid Assay

Bicinchoninic acid assay is a method used for determination of the amount of proteins. The standard reagents used in this method are reagent A, reagent B and reagent C. Reagent A consists of an aqueous solution of $\text{Na}_2\text{tartrate}$, Na_2CO_3 , NaHCO_3 in 0.2 M NaOH , pH 11.25. Reagent B is 4% (W/V) bicinchoninic acid solution, pH 8.5. Reagent C is 4% $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ in deionized water.

The principle of the bicinchoninic assay (BCA) relies on the formation of a Cu^{2+} -protein complex under alkaline conditions, followed by reduction of the Cu^{2+} to Cu^{1+} . The amount of reduction is proportional to protein present. It has been shown that the peptide bond is able to reduce Cu^{2+} to Cu^{1+} . BCA forms a purple-blue complex with Cu^{1+} in alkaline environments, thus providing a basis to monitor the reduction of alkaline Cu^{2+} by proteins.³⁰ Figure B-1 shows complexation between bicinchoninic acid and Cu^{1+} .

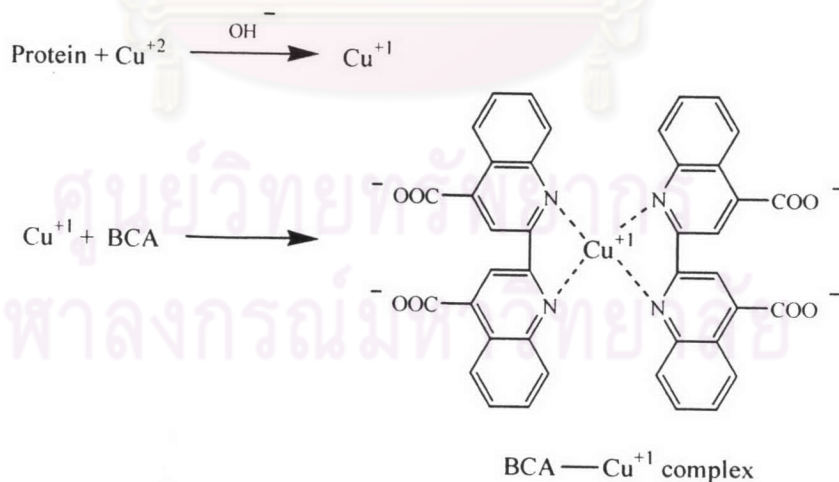


Figure B-1. Formation of purple complex between BCA and cuprous ion generated from the biuret reaction.

Calculation of Protein Adsorption

Table B-1 Standard BSA solution, for the calibration curve

Standard	Solution (mL)	SDS (mL)	BSA conc ($\mu\text{g/mL}$)
S ₁	0.5 of BSA (1000 ($\mu\text{g/mL}$) ^a)	4.5	100
S ₂	4.0 of S ₁	4.0	5.0
S ₃	4.0 of S ₂	4.0	25
S ₄	4.0 of S ₃	6.0	10
S ₅	4.0 of S ₄	4.0	5
S ₆	4.0 of S ₅	4.0	2.5
S ₇	4.0 of S ₆	6.0	1.0
S ₈	4.0 of S ₇	4.0	0.5

a : standard BSA was pipette from 1 mg/mL ampule

After reading the UV absorbance of the samples and standard BSA solution at $\lambda = 562 \text{ nm.}$, the result was then calculated for the net absorbance by subtracting the absorbance of the blank (SDS).

$$\text{Net } A_{562} = \text{recorded } A_{562} - A_{562} (\text{blank})$$

B-1

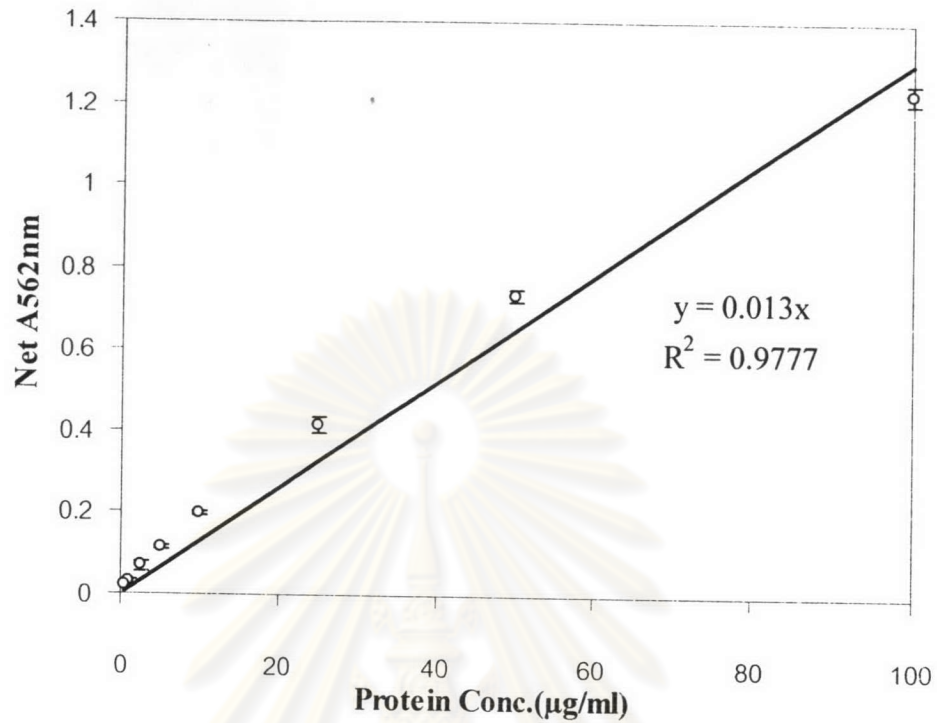


Figure B-2 A calibration curve of the amount of albumin adsorbed and the absorbance obtained from BCA microassay.

The protein concentration (C ; $\mu\text{g/mL}$) in each well was determined from the calibration curve. The total amount of protein (P) in the original solution (2 mL) was calculated from the sampling sample ($100\ \mu\text{L}$) + BCA working solution ($100\ \mu\text{L}$)

$$\text{Total amount of protein (P)} = \frac{C (\mu\text{g/mL}) \times 200 (\mu\text{L})}{1000 (\mu\text{L/mL})} \times \frac{2000 (\mu\text{L})}{100 (\mu\text{L})} \quad \text{B-2}$$

$$\text{Adsorbed protein/surface area } P_{\text{ads}} = P / \text{surface area (2 sides)} (\mu\text{g/cm}^2) \quad \text{B-3}$$

Table B-2 The amount of protein adsorption per surface area ($\mu\text{g}/\text{cm}^2$) of modified chitosan film, as initial concentration 1mg/ml.

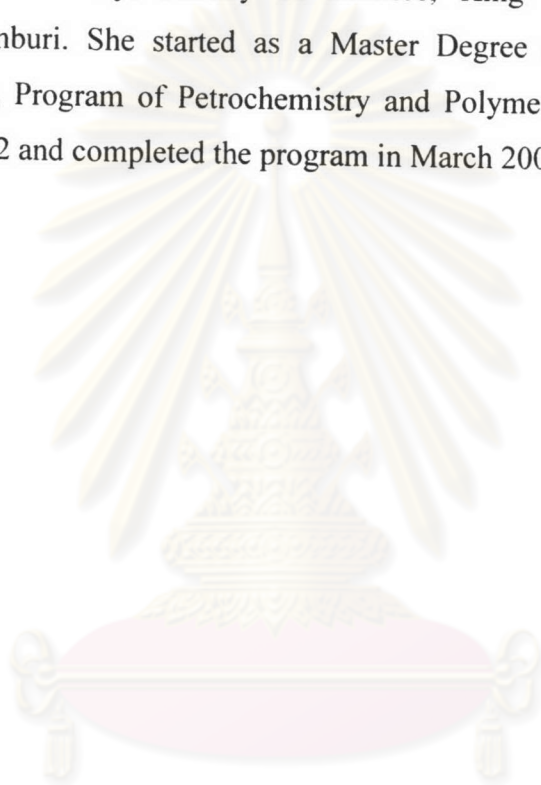
Surface	The amount of protein adsorption ($\mu\text{g}/\text{cm}^2$)			
	BSA	FIB	LYZ	RNase
SFC-film 1	1.54 \pm 0.17	1.84 \pm 0.45	4.57 \pm 0.18	0.50 \pm 0.15
SFC-film 2	1.07 \pm 0.28	1.20 \pm 0.40	2.86 \pm 0.44	1.19 \pm 0.19
SFC-film 3	0.46 \pm 0.26	0.88 \pm 0.33	3.36 \pm 0.38	1.23 \pm 0.3
Chitosan film	0.88 \pm 0.26	1.41 \pm 0.19	2.54 \pm 0.41	1.07 \pm 0.15
QAC-film 1	1.38 \pm 0.45	1.77 \pm 0.29	2.37 \pm 0.42	1.58 \pm 0.26
QAC-film 2	1.75 \pm 0.23	2.09 \pm 0.23	3.44 \pm 0.43	2.10 \pm 0.23
QAC-film 3	1.91 \pm 0.43	2.71 \pm 0.59	9.86 \pm 0.85	2.77 \pm 0.61

Table B-3 Air-Water contact angle of modified and unmodified chitosan film

Surface	Condition		Water contact angle ($^\circ$)
	Rt (h)	Ratio	
SFC-film 1	24	1:0.5	71.2 \pm 2.6
SFC-film 2	24	1:1	67.8 \pm 3.8
SFC-film 3	24	1:5	62.0 \pm 1.9
Chitosan film	-	-	79.6 \pm 1.1
QAC-film1	2	1:3	74.2 \pm 2.2
QAC-film2	8	1:3	63.2 \pm 3.6
QAC-film3	8	1:12	61.0 \pm 1.7

VITAE

Yaowamand Angkitpaiboon was born in Bangkok, Thailand, on August 14th, 1980. She received Bachelor Degree of Science (Industrial Chemistry) in 1997 from Department of Chemistry, Faculty of Science, King Monkut's University of Technology Thonburi. She started as a Master Degree student with a major in Polymer Science, Program of Petrochemistry and Polymer Science, Chulalongkorn University in 2002 and completed the program in March 2005.



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