

CHAPTER II

HISTORICAL REVIEW

Alkaloidal Distribution in Genus Annona

Since 1930, the first study of Annona alkaloids from
A. reticulata Linn(50), there have been about 40 alkaloids of known structures from 11 out of 207 species. It is striking, that for its size the genus Annona is perhaps one of the chemically least known.

The alkaloidal distribution in the genus Annona is summarized in Table 2.

Table 2 Alkaloidal Distribution in genus Annona

Plant (Part)	Alkaloid	Structure No.	Туре	Refer- ence
Annona acuminata Saff.	Liriodenine	34	Oxoaporphine	10
Sari:	Lysicamine	35	TT	51
	Homomoschatoline	32	"	52
	(O-methylmoschatoline			
	Liridine)			



Table 2 (continued)

Plant (Part)	Alkaloid	Structure No.	Туре	Refer- ence
Annona cherimolia Mill.	(-)-Anonaine	14	Aporphine	1
	Caffeine	40	Xanthine	53-55
(Seeds, Twigs)	(+)-Corytuberine	17	Aporphine	72
and the same of	(+)-Isoboldine	20	п	
	Lanuginosine	. 33	Oxoaporphine	
	Liriodenine	34	n	
	(+)-Nornantenine	25	Aporphine	
	(-)-Norushinsunine	28	7-substituted-	
	(Michelalbine)		aporphine	
	(+)-Reticuline	8	Tetrahydro-	
			benzylisoquinoline	
	(-)-Stepholidine	10	Tetrahydro —	
			protoberberine	
Annona crassiflora Mart.	(-)-Anonaine	14	Aporphine	1 .
Marc.	Asimilobine	15	. "	56
(Leaves, Stem-	Liriodenine	34	Oxoaporphine	
barks)	(+)-Reticuline		Cetrahydro =	
			penzylisoquinoline	

Table 2 (continued)

Plant (Part)	Alkaloid	Structure No.	Type	Refer ence
Annona cristalensis	Liriodenine	34 ·	Oxoaporphine	35
(Aerial parts)	(R)-Coclaurine	5	Tetrahydro -	7
			benzylisoquinoline	
Annona elliptica R.E. Fries	(-)-Annonelliptine	2	Tetrahydro -	57
R.E. Fries			benzylisoquinoline	
(Leaves, Stems)				
Annona glabra Linn.	(-)-Anolobine	13	Aporphine	1
(Leaves, Seeds,	(-)-Anonaine	14	n .	58-60
Stem barks)	Asimilobine .	15	n	
	(+)-Isoboldine	20	п	
	Liriodenine	34	Oxoaporphine	
	(-)-N-methylactinodaphnine	22	Aporphine	
	(-)-Nornuciferine	26	п	
	Norushinsunine	28	7-substituted-	
	(Michelalbine)		aporphine	
	(+)-Reticuline	8	Tetrahydro -	
			benzylisoquinoline	
-	(+)-Roemerine	30	Aporphine	

Table 2 (continued)

Plant (Part)	Alkaloid	Structur No.	∈ Туре	Refer- ence
Annona montana Macfad.	Annomontine	41	Pyrimidine-B-	1
(Stem barks,	(-)-Anonaine	14	Aporphine	61,62
Root barks)	Argentinine	39	Phenanthrene	
	Asimilobine	15	Aporphine	100
	Atherosperminine	38	Phenanthrene	7.75
	(-)-Coreximine	9	Tetrahydro -	
			protoberberine	
	(-)-Isoboldine	20	Aporphine	
	Liriodenine	34	Oxoaporphine	
	Methoxyannomontine	42	Pyrimidine-B- carboline	
	(+)-Reticuline	8	Tetrahydro -	
			oenzylisoquinoline	
Annona muricata Linn.	Anomuricine	3	Tetrahydro -	1
			penzylisoquinoline	
(Leaves, Root-	Anomurine	4		63-65
barks, Stem-	Atherosperminine	38	henanthrene	
barks)	Coclaurine	5	Cetrahydro -	
		t	enzylisoquinoline	

Table 2 (continued)

Plant (Part)	Alkaloid	Structur No.	Туре	Refer ence
Annona muricata Linn.	Coreximine	9	Tetrahydro - protoberberine	
(Leaves, Root-	(+)-Reticuline	8	Tetrahydro-	
barks, Stem barks)			benzylisoquinoline	
	(+)-Stepharine	12	Proaporphine	
Annona purpurea	(+)-Glaziovine	11	Proaporphine	1
	(-)-Isocorydine	21	Aporphine	36
(Leaves, Stems)	Norpurpureine	27	11	
	0-Demethylpurpureine	18		
	Oxoglaucine	36	Oxoaporphine	
	Oxopurpureine	37	n	
	Purpureine	29	Aporphine	120
	(+)-Stepharine	12	Proaporphine	
Annona reticulata	(-)-Anonaine	14	Aporphine	1
Linn.	Coclaurine		Tetrahydro -	66-68
(Root barks,			benzylisoquinoline	
Stem barks)	Liriodenine		Oxoaporphine	
	(-)-Norushinsunine		7-substitute	
	(Michelalbine)		aporphine	

Table 2 (continued)

Plant (Part)	ALkaloid	Structu No.	re Type	Refer- ence
Annona reticulata Linn.	(+)-Reticuline	8	Tetrahydro -	
(Root barks,	Salsolinol	1	benzylisoquinoline Simple Tetrahydro- isoquinoline	
Annona squamosa Linn.	(-)-Anolobine	13	Aporphine	1-9
(Stems, Leaves,	(-)-Anonaine :	14 16	n *	
Roots, Barks,	(-)-Isocorydine	21	II .	
Seeds)	(+)-Glaucine	19	"	
	Higenamine	6	Tetrahydro - benzylisoquinoline	
	Lanuginosine (Oxoxylopine)	33	0xoaporphine	
	Liriodenine	34	n .	
	(+)-Norcorydine	23	Aporphine	
	(+)-Norisocorydine	24	"	
	(-)-Norushinsunine	28	7-substituted-	
	(Michelalbine) (+)-0-methylarmepavine	7	aporphine	
	- Industrial inches in E		Tetrahydro - benzylisoquinoline	

Table 2 (continued)

Plant (Part)	Alkaloid	Structure No.	Туре	Refer- ence
Annona squamosa Linn.	(+)-Reticuline	8	Tetrahydro -	1-9
(Stems, Leaves,	(-)-Roemerine	30	benzylisoquinoline Aporphine	
Roots, Barks,	(-)-Xylopine	31		
Seeds)			100 mg/s	



Chemistry of Annona Alkaloids

The majority of alkaloids found in Annona spp. possess an isoquinoline derived structure (39 out of 42 alkaloids). They are simple tetrahydroisoquinolines, tetrahydrobenzylisoquinolines, protoberberines, proaporphines, aporphines, oxoaporphines and phenanthrenes. Only three alkaloids found in Annona spp. are non-isoquinoline. The figure in square bla following the name of alkaloid type indicating the number of alkaloid in that type.

The numbering and structure of the alkaloids are as follows.

1. Simple Tetrahydroisoquinoline [1]

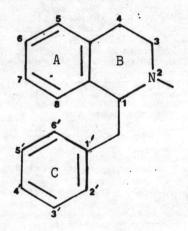
Although many of the simple tetrahydroisoquinolines have been found in Cactaceae and are also present in some members of the Papaveraceae and Fumariaceae (69), one of them, salsolinol (#1), surprisingly found in Annona reticulata Linn., together with dopamine, its biogenetic precursor (1).

#1 Salsolinol

2. Benzylisoquinolines [7]

Seven tetrahydrobenzylisoquinolines were isolated from some species of Annona as shown in Table 3. The most frequently occurring tetrahydrobenzylisoquinoline is reticuline (#8). The alkaloid isolated by Meyer in 1941 from Annona muricata Linn, under the name " muricinine ", was later identified to be reticuline by Santos et al. (1).

Anomuricine (#3), anomurine (#4) from Annona muricata Linn. (63), and annoelliptine (#2) from Annona elliptica R.E. Fries (57), are special in that they are substituted at C-5 (1).



Tetrahydrobenzylisoquinoline

Table 3 Tetrahydrobenzylisoquinolines isolated from Annona spp.

Alkaloid Substitution						
	2	5	6	7	3'	4'
Annoelliptine	CH ₃	OCH ₃	OCH ₃	ОН	-	ОН
Anomuricine	Н	ОН	OCH ₃	OCH ₃	-	OCH ₃
Anomurine	Н	OCH ₃	OCH ₃	OCH ₃	-	OCH ₃
Coclaurine	Н	-	OCH ₃	ОН	<u></u>	ОН
Higenamine	Н	-	ОН	ОН	-	ОН
O-methylarmepavine	CH ₃	-	OCH ₃	OCH ₃	-	OCH ₃
Reticuline	CH ₃	(- :	OCH ₃	ОН	ОН	OCH ₃
	Annoelliptine Anomuricine Anomurine Coclaurine Higenamine O-methylarmepavine	Annoelliptine CH ₃ Anomuricine H Anomurine H Coclaurine H Higenamine H O-methylarmepavine CH ₃	Annoelliptine CH ₃ OCH ₃ Anomuricine H OH Anomurine H OCH ₃ Coclaurine H - Higenamine H - O-methylarmepavine CH ₃ -	Annoelliptine CH ₃ OCH ₃ OCH ₃ OCH ₃ Anomuricine H OCH ₃ OCH ₃ OCH ₃ H OCH ₃	2 5 6 7 Annoelliptine CH ₃ OCH ₃ OCH ₃ OH Anomuricine H OH OCH ₃ OCH ₃ Anomurine H OCH ₃ OCH ₃ OCH ₃ Coclaurine H - OCH ₃ OH Higenamine H - OH OH O-methylarmepavine CH ₃ - OCH ₃ OCH ₃	Annoelliptine CH3 OCH3 OCH3 OH - Anomuricine H OH OCH3 OCH3 - Anomurine H OCH3 OCH3 OCH3 - Coclaurine H - OCH3 OH - Higenamine H - OH OH - O-methylarmepavine CH3 - OCH3 OCH3 -

The benzylisoquinoline alkaloids are either of the 1,2,3,4, - tetrahydroisoquinoline type, such as coclaurine (#5), or of the completely aromatic type, as in the case of papaverine (#43).

#5 Coclaurine

#43 Papaverine

Ring A in the benzylisoquinoline alkaloids may possess two or three oxygenated substituents, while ring C has only one or two substituents. No C-4 hydroxylated benzylisoquinoline has yet been isolated from nature, although it is more than probable that they do exist (70).

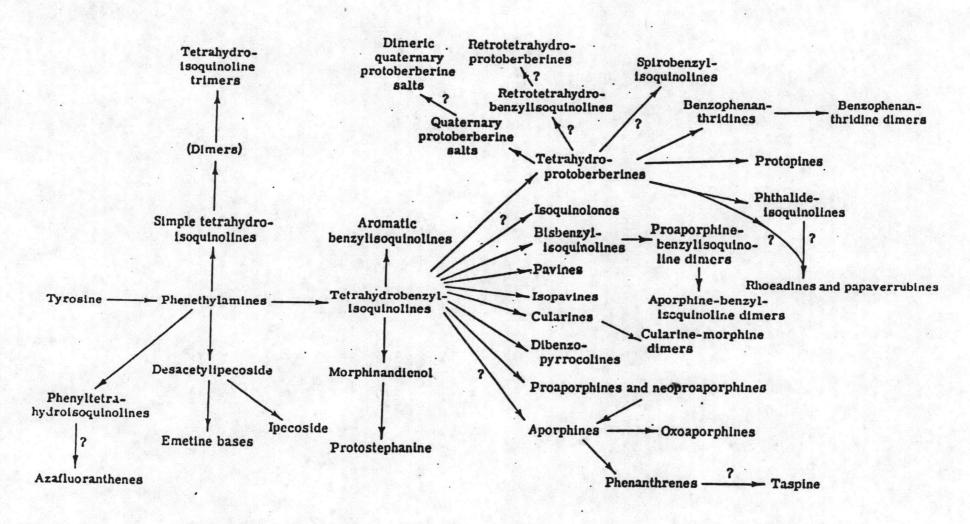
The alkaloids of the benzylisoquinoline type, are a vast and complex group of alkaloids (77). They occupy a paramount position in alkaloid chemistry because they act as in vivo precursors to so many of the other natural occuring isoquinolines: isoquinolones, pavines, isopavines, bisbenzylisoquinolines, cularines, dibenzopyrrocolines, morphines, cularine-morphine dimers, proaporphines, aporphines, protoberberines and others (70). Some of proven or probable biogenetic loci for the formation of the isoquinoline alkaloids are illustrated in Scheme I (71). (Page 26)

In addition the biogenetic relationship of the major alkaloid groups derived from a tetrahydrobenzylisoquinoline precursor are shown in Scheme II (77). (Page 27)

3. Protoberberines

Tetrahydroprotoberberines [2]

Only two tetrahydroprotoberberines described in Annona species. They are coreximine (#9) from Annona montana Macfad. and Annona muricata
Linn., and stepholidine (#10) from Annona cherimolia Mill. These alkaloid structures are shown in Table 4.



Scheme I Proven or Probable Biogenetic Loci for the Formation of the Isoquinoline Alkaloids

Scheme II Biogenetic relationships of the major alkaloid groups derived from a tetrahydrobenzylisoquinoline precursor

Tetrahydroprotoberberines

Table 4 Tetrahydroprotoberberines isolated from Annona spp.

Struç			S	ubstitut	cions	
No.	Alkaloid	2	:3	9	10	11
9	Coreximine	ОН	OCH ₃		OCH ₃	ОН
10	Stepholidine	ОН	OCH ₃	OCH ₃	ОН	-

Most protoberberine alkaloids exist in nature either as tetrahydroprotoberberines or quaternary protoberberine salts, but some dihydroprotoberberines are also known. Substituents are usually present at C-2 and C-3 and either at C-9 and C-10 or at C-10 and C-11. In some instances a hydroxyl or methoxyl substituent may be present at C-1. A methyl group is sometimes found at C-13, while in a few cases an alcoholic hydroxyl is treated at C-13 or C-5 (72)

The protoberberines are of the most widely distributed of the isoquinoline alkaloid groups, being present in at least nine plant families, particularly the Annonaceae, Berberidaceae, Lauraceae, Menispermaceae, Papaveraceae and Rutaceae (78). Over forty protoberberines are known but only two tetrahydroprotoberberines were

Quaternary protoberberine salts

4. <u>Proaporphines</u> [2]

Tetrahydroprotoberberine

isolated from Annona spp.

There are over 30 proaporphine alkaloids distributed in the plant families:— Annonaceae, Euphorbiaceae, Lauraceae, Menispermaceae, Monimiaceae, Nymphaeaceae and Papaveraceae (1,79). However, there seem to be few proaporphines in genus Annona. Only two representatives (#11,12), of this group have been reported in two species i.e. Annona muricata Linn., and A. purpurea Linn. They are shown in Table 5.

Proaporphine

<u>Table 5</u> Proaporphine isolated from Annona spp.

ruc			Substitutio	ns
o. Al	kaloid	1 :	2	6
11 (+)-G1	aziovine	ОН	OCH ₃	CH ₃
12 Stepha	rine	OCH ³	OCH ₃	Н

Fundamentally they are of two structural types, those with a dienone system, such as (+)-glaziovine (#11), and those in which the dienone system has been completely or partially reduced, such as - (+)-linearisine (#44) and (-)-oreoline (#45). Compounds having either stereochemistry at C-6 are known and the two sides of the cyclohexadienone system are not equivalent because of this asymmetry (79).

5. Aporphines [19]

Half of the alkaloids (19 out of 42) isolated from genus Annona are aporphines (#13-31), as shown in Table 6. They have very different structures and none is 8-substituted. Anonaine (#14) is the most common aporphine in this genus, since it has been reported in five species of Annona. Norushinsunine (michelalbine) (#28) is the only one 7-substituted aporphine found in Annona species (1).

Aporphine

Table 6 Aporphines isolated from Annona spp.

Struc- ture	Alkaloid		S	ubstitutions					
No.		1	2	3	6	7	9	10	11
13	Anolobine	OC	H ₂ 0		Н	-	ОН		
14	Anonaine	OCI	H ₂ 0	_	Н	-	-		_
15	Asimilobine	OCH ₃	ОН	_	Н	1	-	<u>-</u>	_
16	Corydine	ОН	OCH,	_	CH3	_	_	OCH ₃	OCH,
17	(+)-Corytuberine	ОН	OCH ₃	-	CH ₃	-	-	OCH ₃	
18	0-demethylpurpureine	OCH ₃	осн 3	→ ОН	CH ₃	-	OCH ₃	OCH ₃	

Struc- ture	Alkaloid	Substitutions							
No.	MIKATOTO	. 1	2	3	6	7	9	10	11
19	Glaucine	OCH3	OCH ₃	-	CH ₃	-	OCH ₃	осн ₃	•
20	Isoboldine	ОН	OCH ₃	-	CH ₃	-	ОН	OCH ₃	-
21	Isocorydine	OCH ₃	OCH ₃	-	CH ₃	-	-	OCH ₃	ОН
22	N-methylactinodaphnine	OCI	H ₂ 0	-	CH ₃	-	ОН	OCH ₃	_
23	Norcorydine	ОН	OCH ₃	·-	Н	-	-	OCH ₃	OCH ₃
24	Norisocorydine	OCH ₃	OCH ₃	-	Н	-	-	осн ₃	OH
25	Nornantenine	OCH ₃	OCH ₃	-	Н	· •	осн	20	-
26	Nornuciferine	OCH ₃	OCH ₃	-	Н	-	-	-	-
27	Norpurpureine	OCH ₃	OCH ₃	OCH ₃	Н	-	OCH ₃	OCH ₃	-
28	Norushinsunine	oc	H ₂ 0	-	Н	ОН		-	-
29	Purpureine	OCH ₃	OCH ₃	осн3	CH ₃	_	OCH ₃	OCH ₃	-
30	Roemerine	oc	H ₂ 0	<u>-</u>	CH ₃	-	-	-	-
31	Xylopine	oc	H ₂ 0	-	Н	-	OCH ₃	-	-

The aporphines constitute the next to the largest group of isoquinoline alkaloids, being second only to the bisbenzylisoquinolines. They are distributed in at least 18 plant families, of which the most important are the Papaveraceae, Annonaceae, Lauraceae and Monimiaceae (80).

The nitrogen atom of aporphines is usually methylated, and although some noraporphines are known, they are not very stable and

are often characterized as their N-acetyl derivatives. Aporphines are known with the C-6 a stereochemistry either α or β .

The most diverse structural feature of the aporphines is the oxygenation pattern. Positions 1 and 2 are always oxygenated, either by hydroxy, methoxy or methylene dioxy groups. It is common to find further oxygen substituents at C-9, C-10 and C-11, and occasionally at C-8. It is rare to find oxygenation at C-7, except in oxoaporphines, and even rarer to find any oxygenation in ring B (80).

6. Oxoaporphines [6]

Six oxoaporphines (#32-37) were isolated from various species of Annona, liriodenine being ubiquitous. From the structural point of view, these oxoaporphines are rarely substituted at positions 3,9,10 and 11 (1). The oxoaporphines found in Annona &pp. are shown in Table 7.



Oxoaporphine

Table 7 Oxoaporphines isolated from Annona spp.

Struc ture No.	Alkaloid	Substitutions				
		1	· 2	3	9	10
32	Homomoschatoline	осн3	ОН	осн3	2-1	-
33	Lanuginosine	OCH	20	-	OCH ₃	-
34	Liriodenine	OCH	20	-	-	-
35	Lysicamine	OCH3	OCH ₃	-	-	-
36	Oxoglaucine	och3	OCH ₃	- **	OCH ₃	OCH ₃
37	Oxopurpureine	OCH ₃				

The oxoaporphines represent the most highly oxidized state of the aporphine skeleton. They commonly cooccur with aporphine alkaloids. Therefore, the oxoaporphines are most probably derived in plants by oxidation of the corresponding aporphines. The free bases possess a bright yellow or orange color which turns pink or red upon the addition of mineral acid (73,81).

Oxoaporphines are high melting and show a decomposition point rather than a melting point (73).

The numbering system of oxoaporphine is the same as that of the aporphines. Like the aporphines, beyond the 1,2-dioxygenation, they exhibit a variety of oxygen substitution patterns. In this series, however, there is a tendency toward 3-substitution and not 11-substitution (81).

7. Phenanthrenes [2]

Two members (#38-39), as shown in Table 8, belonging to the limited group of aminoethylphenanthrene derivatives ("open aporphine"), have been found in Annona montana Macfad. and Annona muricata Linn.

None possess substituents on C-5 and C-6 (corresponding to positions 11 and 10 of the aporphines) (1).

Phenanthrene

Table 8 Phenanthrenes isolated from Annona spp.

truc- ture	Alkaloid	Substitutions			
No.		N	2	3	4
38	Atherosperminine	(CH ₃) ₂	- -	OCH ₃	осн ₃
39	Argentinine	(CH ₃) ₂	-	ОН	OCH ₃

The phenanthrene alkaloids are a small group of optically inactive tertiary bases probably derived biogenetically from the Hofmann elimination of quaternary aporphine salts. They can, therefore, be included among the isoquinoline alkaloids.

The phenanthrene alkaloids are always substituted at C-3, C-4 since their precursors, the aporphines, are found with substituents at these two positions corresponding to C-2, C-1 of the aporphine skeleton (74).

8. Non-isoquinoline Alkaloids [3]

8.1 Purine [1]

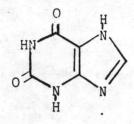
Surprisingly, caffeine (#40) has been found to be one of the major constituents of Annona cherimolia Mill. seeds (1,54).

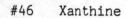
Normally it was found in Coffee (Coffea arabica Linn.), Cola (Cola nitida (Vent.) Schott & Endl.), Maté (Ilex paraguariensis A.St-Hil.),

Tea (Camellia sinensis (Linn.) O. Kuntze) and Cocoa (Theobroma cacao Linn.) (82).

#40 Caffeine

The most important purine alkaloids are derived from the xanthine nucleus (#46).







Xanthine itself has not yet been found naturally, but several simple N-alkyl derivatives are of quite considerable significance. Prime among these is caffeine, which is 1,3,7-trimethylxanthine (82).

8.2 Pyrimidine-B-carboline [2]

Recently the isolation of two new alkaloids of an unusual type, namely annomontine (#41), and methoxyannomontine (#42), from the stem and root bark of Annona montana Macfad.was reported. They are the first examples of a new class of pyrimidine-\beta-carboline alkaloids, composed by an harman moiety linked to 2-aminopyrimidine (61,62).

Pyrimidine-B-carboline

#41 Annomontine R = H

#42 Methoxyannomontine $R = OCH_3$

Biosynthesis of Aporphines

The biogenetic pathways leading to isoquinoline alkaloids are derived from tyrosine. Tyrosine is first elaborated to a suitable hydroxylated and derivatised phenethylamine which is then combined with a second building block. This second building block can vary widely to give various types of isoquinolines (85). The proven or probable biogenetic loci for the formation of the isoquinoline alkaloids is shown in scheme I (71). (Page 26)

The biosynthesis of isoquinoline alkaloids is conveniently treated in two parts, starting with the generation of benzylisoquinolines and then elaborating to more complicated structures (85).

Benzylisoquinolines derive from two molecules of L-tyrosine. They occupy a paramount position in alkaloid chemistry because they act as in vivo precursors to so many of the other naturally occurring isoquinolines. Reticuline is regarded as the key intermediate in the biosynthesis of the alkaloids based on the benzylisoquinoline nucleus, and as a result of the study of these alkaloids, much has been learned of the biosynthesis of reticuline (70,84).

The example of the formation of reticuline in Papaver somniferum Linn. is shown in Scheme 3 (86).

Scheme III The formation of 1-benzyl-1,2,3,4-tetrahydroisoquinolines

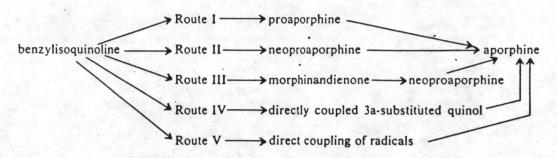
^{*}SAM = S-adenosylmethionine

The first step (A) involves the conversion of two molecules of L-tyrosine into two molecules of DOPA. One molecule of DOPA is converted into dopamine (B) whilst the other is converted into 3,4-dihydroxyphenyl-pyruvic acid (C). These two compounds then combine with the elimination of water and carbon dioxide in a Mannich-type reaction to yield a molecule of norlaudanosoline (D). O- and N-methylation (E) then lead to (-)-reticuline (86).

More recent experiments have definitely show that benzylisoquinolines are natural precursor of aporphine alkaloids. For instance, reticuline occupied a key position in the biosynthesis of morphine alkaloids (66) and also was the precursor of other alkaloids such as boldine, isoboldine and bulbocapnine (15).

From some biogenetic results of aporphines, it would suggest that boldine is produced by two different pathways in two different plants. If this is a general trend, the biosynthesis of aporphine alkaloids may never be established. Clearly this is an area in need of considerable further study, and at this time it is not clear which, or how many, of the possible routes may be operating in order to produce the various aporphine alkaloids.

Hence, the discussion on the biogenetic synthesis of the aporphine alkaloids has raised a number of points concerning the biosynthesis of this group. In particular and depending on the orientation of phenolic and methoxy groups, we might envisage any of at least five routes being in operation from a benzylisoquinoline precursor (Scheme IV) (83).



Scheme IV Biogenesis of aporphines from benzylisoquinolines.

Route I : Benzylisoquinoline \longrightarrow Proaporphine \longrightarrow Aporphine

From Scheme V aporphine alkaloids lacking an oxygen substituent in ring D are derived from a benzylisoquinoline by oxidation to a dienone, reduction to dienol and dehydration with rearrangement to the fully aromatic compound (87).

$$R^{1}O$$
 $R^{2}O$
 R

Benzylisoquinoline Dienone Dienol Aporphine

Scheme V Reaction of benzylisoquinoline to aporphine

Barton and co-workers (88) investigated the formation of (+)-roemerine in *Papaver dubium* Linn. and found that tritium-labeled (+)-N-methylcoclaurine was well incorporated. The position of the hydroxy group in the precursor suggests that a proaporphine intermediate is involved as in Scheme VI (75,83,89).

Scheme VI Biosynthesis of (+)-roemerine

Similarly, Battersby and co-workers (90) showed that labeled (+)-orientaline, when fed to *Papaver orientale* Linn., was incorporated into the aporphine alkaloid (+)-isothebaine. The (+)-isothebaine can be envisaged as occuring through (-)-orientalinone, a co-constituent, as shown in Scheme VII (75,83,89,91).



Scheme VII Biosynthesis of (+)-isothebaine

Route II: Benzylisoquinoline—Neoproaporphine—Aporphine

More definitive results have been obtained by Battersby and

co-workers (90) concerning the formation of corydine, dicentrine and
glaucine in Dicentra eximia (Ker.) Torr. Reticuline and orientaline were

not precursors, but 4'-O-methylnorlaudanosoline which originated from

norlaudanosoline was effective precursor. 4'-O-methylnorlaudanosoline
is converted into norprotosinomenine which must be incorperated into

the three aporphine alkaloids by way of the two neoproaporphine

intermediates as indicated in Scheme VIII (76,83,89).

Scheme VIII Biosynthesis of Corydine, Dicentrine and Glaucine in Dicentra eximia (Ker.) Torr.

Route III : Benzylisoquinoline → Morphinandienone → →
Neoproaporphine → Aporphine

Route IV : Benzylisoquinoline ---> Directly coupled 3a -substituted quinol ---> Aporphine

It is very difficult in some instances to say what the probable mechanism is unless variously methylated benzylisoquinoline precursors are used. For example, as in Scheme IX, N-methyl labeled (+)-reticuline (A) was a precursor of bulbocapnine in Corydalis cava Schweigg.et Korte, but is this evidence of direct phenol coupling? Even if demethylation-remethylation is assumed not to occur, direct coupling of a 3 -substituted quinol not involving radicals is still a reasonable alternative (i.e. B). It is pertinent to note that the morphinandienone sinoacutine is also a constituent of this plant. Could sinoacutine be a key precursor via the intermediate neoproaporphine (C) ? (83)

Scheme IX Biosynthesis of bulbocapnine

Route V: Benzylisoquinoline \longrightarrow Direct coupling of radicals \longrightarrow Aporphine

Aporphines are formed in plants by intramolecular phenolic oxidative coupling of benzylisoquinolines (92). Isoboldine has been shown to be formed in *Papaver somniferum* Linn. by a straightforward intramolecular oxidative coupling between the two aryl rings of reticuline. The site of coupling (marked by dots in Scheme X) both bear a hydrogen and therefore the two rings can rearomatise to give the biphenyl system characteristic of the aporphine alkaloids, though as we shall see the biosynthetic routes leading to members of this groups of alkaloids are not always so straightforward (86).

$$H_3^{CO}$$
 H_3^{CO}
 H_3^{CO}

Reticuline

Isoboldine

Scheme X Biosynthesis of Isoboldine

In other experiments, it was found that (-+)-coclaurine could be a precursor for anomaine in Anona reticulata Sieber, ex. A.DC. (76).

$$\begin{array}{c} \text{H}_3^{\text{CO}} \\ \text{HO} \\ \text{T} \\ \text{HO} \\ \text{T} \\ \text{labeled ($\stackrel{+}{-}$)-coclaurine} \end{array}$$

The evidence in favor of direct coupling of a benzylisoquinoline to an aporphine is more ambiguous. Blaschke has found that feeding (+)-reticuline labeled on the N-methyl group to the larkspur, Corydalis cava Schweigg.et Korte, had incorporated to labeled bulbocapnine as follows:-

$$H_3^{CO}$$
 H_3^{CO}
 H_3^{CO}

labeled (+)-reticuline

labeled (+)-bulbocapnine

Benzylisoquinolines are not only the precursor of aporphines, but they are also the precursor of many other isoquinoline alkaloids which is summarized in Scheme II (Page 27) (77).

Biological and Pharmacological Activity of Annona Alkaloids

Considerable studies have been carried out to investigate the active principle for biological and pharmacological action of Annona alkaloids, as shown in Table 9.

Table 9 Biological and Pharmocological Activity of Annona Alkaloids

Alkaloid	Activity	Species found	Reference
Annomontine	weak sedative, antiamoebic.	Annona montana Macfad	93
Anonaine	Parasiticide.	 A. cherimolia Mill. A. crassiflora Mart. A. glabra Linn. A. montana Macfad. A. reticulata Linn. A. squamosa Linn. 	1,9
Caffeine	increase central - nervous system activity	A. cherimolia Mill.	82

Table 9 (continued)

Alkaloid	Activity	Species found	Referenc
Corydine	cytotoxic, antitumor,	Annona purpurea Linn.	3,5,
	adrenolytic action,	A. squamosa Linn.	64,
	C.N.S. depressant,		82,
	hypertensive,		94
	block nerve impulse-		
	transmission.		
Corytuberine	respiratory stimulant	A. cherimolia Mill.	82,
	secretion stimulant,		95
	slow the pulse.		
Glaucine	reduce blood pressure,	A. squamosa Linn.	82,
	inhibit respiration,		94,
	antitussive,		95
	active adrenolytic,		
	hypoglycemia.		
	·		
Glaziovine	antidepressant,	A. purpurea Linn.	63,
	Cytotoxic.		95

Table 9 (continued)

Alkaloid	Activity	Species found	Reference
Higenamine	effective B-adrenergic agonist (positive inotropic action)	Annona squamosa Linn.	2
Homomoschatoline	Cytotoxic .	A. acuminata Saff.	10
Isoboldine	insect-feeding - inhibitor	A. cherimolia Mill.A. glabra Linn.A. montana Macfad.	96
Isocorydine	adrenolytic action	A. squamosa Linn.	94
Liriodenine	cytotoxic, antitumor, antibacterial, antifungal.	A. acuminata Saff. A. cherimolia Mill. A. glabra Linn. A. montana Macfad. A. purpurea Linn. A. reticulata Linn. A. squamosa Linn.	1,10, 57, 97

Table 9 (continued)

Alkaloid	Activity	Species found	Reference
Lysicamine	cytotoxic	Annona acuminata Saff.	10
Methoxyannomontine	weak sedative,	A. montana Macfad.	93
O-methylatheroline	cytotoxic	A. purpurea Linn.	63
Xylopine	sedative,	A. squamosa Linn.	95
Alkaloid from leaves and stem bark	analgesic, spasmolytic, antibacterial	A. crassiflora Mart.	52
Petroleum ether extract of root bark (C/M ₂)	antineoplastic, antibacterial against Staphylococcus aureus, Salmonella typhosa (in vitro)	A. senegalensis Pers.	98 , 99

Table 9 (continued)

Alkaloid	Activity	Species found	Reference
Aqueous and	cardiorespiratory effect	Annona squamosa Linn.	10-14,
Ethanol	(cat),		100
extract	insecticidal,		
	oxytocic, uterotonic,		
	spasmogenic (guinea pig		
	ileum),		
	spasmolytic (rabbit		
	duodenum)		

