

BIBLIOGRAPHY

- ASA-SSSA. 1982. Methods of Soil Analysis, Part II Chemical and Microbiological Properties, (Page, A.L. ed.) 2 nd ed. 677 S. Segoe Rd., Madison, Wi 53711, USA.
- Abd Elfattah, Aly. and K. Wada. 1981. Adsorption of Lead, Copper, Zinc, Cobolt, and Cadmium by Soils that differ in Cation Exchange Meterials. J. Soil Sci. 32, 271-283.
- Adams, T. McM. 1985. The Effect of pH on the Uptake of Zine, Copper, and Nickel from Chloride Solutions by an Uncontaminated Sewage Sludges. Environ, Pollut. Ser. B 151-161.
- Black, C.A., D.D. Evans, L.E. Ensminger, J.L. White, and F.E. Clark. 1965. Methods of Soil Analysis part 2 Chemical and Microbiological Properties. American Society of Agronomy.
- Bohn, H., B.L. Mcneal, and G.A. O'connor. 1979. Soil Chemistry.

 John Wiley & Sons.
- Brady, Nyle C.. 1974. <u>The Nature and Properties of Soils</u>. 8 th. ed. New York. Macmillan.
- Brummer, G., K.G Tiller, U. Herms, and P.M. Clayton. 1983.

 Adsorption Desorption and/or Precipitation-Dissolution

 Processes of Zinc in Soils., Geoderma 31, 337-354.
- Bunzl, K., W. Schmidt, and B. Sansoni. 1976. Kinetics of Ion Exchange in Soil Organic Matter. IV. Adsorption and Desorption of Pb²⁺, Cu²⁺, Cd²⁺, Zn²⁺ and Ca²⁺ by Peat. J. Soil Sci. 27, 32-41.
- Burchill, S., M.H.B. Hayes, and D.J. Greenland. 1981. Adsorption in Greenland, D.J. and M.H.B. Hayes (eds.) <u>The Chemistry of Soil Processes</u>, pp.221-400, John Wiley and Son.

- Christensen, T.H. 1985. Cadmium Soil Sorption at Low Concentrations: III. Prediction and Observation of Mobility. Water, Air, and Soil Pollut. 26, 255-264.
- Dhilton, S.K., P.S. Sidhu, and M.K. Sinha, 1981. Copper Adsorption by Alkaline Soils J. Soil Sci 32, 571-578.
- Davies, Brian E. 1980. Applied Soil Trace Elements. John Willey and Sons. pp.259-282.
- Egozy, Y.. 1980. Adsorption of Cadmium and Cobalt on Montmorillonite as a Function of Solution Composition.

 Clays and Clay Minerals 28 (4), 311-318.
- Eliott, H.A. 1983. The Adsorption Behavior of Cadmium in Response to Soil Surface Charge. Soil Sci, 136(5), 317-321.
- Farrah, H. and W.F. Pickering. 1977. Influence of Clay-Solute Interactions on Aqueous Heavy Metal Ion Levels. <u>Water</u>, <u>Air and Soil Pollut.</u>, 8, 189-197.
- _____. 1978. Extraction of Heavy Metal Ions Sorbed on Clays.

 Water, Air and Soil Pollut. 9, 491-498.
- Forbes, E.A., A.M. Posner, and J.P. Quirk, 1976. The Specific Adsorption of Divalent Cd, Co, Cu, Pb and Zn on Geothite <u>J. Soil Sci</u>. 27, 154-166.
- Frank, R., K.I. Stonefield, and P. Suda. 1985. Background Elemental Content of Animal Feeds, Ontario, Canada, 1978-82 Bull. Environ Contam. Toxicol., vol.35, pp.689-695.
- Gadde, R. Rac and H.A. Laitinen. 1974. Studies of Heavy Metal Adsorption by Hydrous Iron and Manganese Oxides," Anal. Chem. 46(13), 2022-2026.
- Harter, Robert D. 1983. Effect of Soil pH on Adsorption of Lead, Copper, Zinc, and Nickel. Soil, Sci. Am. J. 47-51.
- Jones, L.H.P., and S.C.Jarvis. 1981. The Fate of Heavy Metals in Greenland, D.J., and Hayes, M.H.B. <u>The Chemisty of Soil Processes</u>, pp.593-620, Jonh Wiley and Sons.
- Kinniburgh, David G. 1986. General Purpose Adsorption Isotherms. Environ. Sci. Technol. 20(9), 895-904.

- Kinnikurgh, D.G. and M.L. Jackson. 1982. Concentration and pH

 Dependence of Calcium and Zinc Adsorption by Iron

 Hydrous Oxide Gel. Soil Sci. Soc. Am. J. 46, 56-61.
- Levy, R. and C.W. Francis. 1976. Adsorption and Desorption of Cadmium by Synthetic and Natural Organo-Clay Compleses Geoderma 15, 361-370.
- Lindsay, Willard L. 1979. <u>Chemical Equilibria in Soils</u>. New York: John Wiley & Sons.
- McKenney, D.J., and J.R. Vriesacker. 1985. Effect of Cadmium Contamination On Denitrification Processes in Brookston Clay and Fox Sandy Loam. Envi. Poll. (series B) 38, 221-233.
- Miller, J.E., W.L.Nelson, and D.E. Koeppe. 1976. Uptake of Cadmium by Soy beans as Influenced by Soil Cation Exchange Capacity, pH, and Available Phosphorus.

 J. Environ Qual. vol 5(2), 157-160.
- Mortvedt, J.J. 1985. Plant Upatke of Heavy Metals in Zinc Fertilizers Made from Industial By-Products. <u>J.Environ Qual vol.14</u>, no.3, pp 424-427.
- Plailamoon, S. 1984. <u>Influence of Cadmium in Activated Sludge on Ammonifier Activity and Rice.</u> Master's Thesis, Department of Environ mental Science, Kasertsart University.
- Ram Nand and M. Verloo. 1985. Effect of Various Organic Materials on the Mobility of Heavy Metal in Soil. Environ Pollut Ser. B 10,241-248.
- Ratanaphaithun, J. 1983. Characteristics of Wastewater from Small Industries in Rajburana District, Bangkok Metropolis. Master's Thesis, Department of Sanitary Engineering, Faculty of Engineering, Chulalongkorn University.
- Reddy, M.R. and S.J.Dunn. 1986. Distribution Coefficients for Nickel and Zinc in Soil. <u>Environmental Pollution (Series B)</u> 11, 303-313.

- Sanders, J.R.M.I.E.l Kherbawy. 1987. The Effect of pH on Zn Adsorption Equilibria and Exchangeable Zinc Pools in Soils. Environmental Pollution 44 (3). 165-176.
- Sanders, J.R., S.R. McGrath, and T. McM Adams. 1987. Zinc, Copper and Nickel Concentrations in Soil Extracts and Crops Grown on Four Soils Treated with Metal-loaded Sewage Sludges. Environ Pollut. 44(3), 193-210.
- Soon, Y.K.. 1987. Solubility and Sorption of Cadmium in Soils
 Amended with Sewage Sludge. J. Soil Science. 32, 85-95.
- Spositio Garrison. 1984. The Surface Chemistry of Soils, pp.113-148 Oxford University Press, U.S.A..
- Talibudeen, O. 1981. Cation Exchange in Soils. in Greenland, D.J. and M.H.B. Hayes, (eds) <u>The Chemistry of Soil Processes</u> pp.115-177, John Wiley and Son.
- Tiller, Kevin G., J. Gerth and G. Brummer. 1984. The Sorption of Cd,
 Zn and Ni by Soil Clay Fractions: Procedures for Partion
 of Bound Forms and their Interpretation. Geoderma
 34,1-16.
- ________ 1984. The Relative Affinities of Cd, Ni and Zn for Different Soil Clay Fractions and Goethite. Geoderma, 34, 17-35.
- Tisdale, S.L., W.L. Nelson. 1975. Soil Fertility and Fertilizers. pp. 325-331, New York: Macmillian Publishing Co..
- Wada, K., and Aly Abd-Elfattah. 1978. Characterization of Zinc Adsorption SItes in Two Mineral SOils Soil Sci Plant
 Nutri 24(3), 417-426.
- _____. 1979. Effects of Cation Exchange Material on Zinc Adsorption by Soils. J. of Soil Science 30, 281-290.
- Wada, K. and Y. Kakuto. 1980. Selective Adsorption of Zinc on Halloysite. Clays and Clay Minerals. vol. 28(5). 321-327.
- Walpole, Ronald E. and Mayers, Raymond H.. 1972. <u>Probability and Statistics for Engineers and Scientists</u>. 2nd ed. Macmillan Publish.

- Xian, X.. 1989. Response of Kidney Bean to Concentration and Chemical Form of Cadmium, Zinc and Lead in Polluted Soils. Environ, Pollut. 57, 127-137.
- Zhan, Tao. 1986. A Laboratory Study of the Immobilization of Cadmium in Soils. Environ, Pollut (Ser. B) 12, 265-280.

APPENDIX A

BAN MI SERIES (Bm)

<u>Distribution</u>: Moderate to small extent along the borders of the Central Plain.

Setting: Ban Mi soils are formed from montmorillonitic clays. They occur on semi-recent terraces and along the boundary between these terraces and the recent alluvial plain. Relief is nearly flat with slopes of less than 1%. Elevation ranges from 4 m. to 20 m. above sea level. The climate is Tropical Savanna (Koppen 'Aw'). Annual precipitation ranges from 1,000 mm. to 1,400 mm.

<u>Drainage and Permeability</u>: Poorly to somewhat poorly drained. Runoff is slow and permeability is estimated as slow. These soils are flooded by impounded rainwater to depths of 30 cm. to 40 cm. for several months during the wet season. However, these soils also dry out deeply with groundwater level falling below 2 m. during the peak of the dry season when deep wide cracks may be observel.

<u>Vegetation and Land Use</u>: Mainly used for transplanted rice cultivation.

Characteristic Profile Features: The Ban Mi series is a member of the very fine clayey, montmorillonitic family of Grumusols (National), Entic Pelluderts (USDA). They are deep, slightly acid to neutral soils characterized by a thick, dark coloured A horizon overlying a slightly paler subsoil (B ?) containing predominantly dark yellowish brown mottles. The occurrence of slickensides is characteristic. Cracks are at least 1 cm. wide at 50 cm. depth.

Typifying Pedon : Ban Mi clay - Rice field from Amphoe Ban Mo, Saraburi Province - Code C4/11 (Type Location) (moist colours unless otherwise stated)

Ap -- 0-10 cm. Dark grey (10YR4/1), many, fine yellowish red mottles as coatings along root channnels and on ped faces; clay; strong fine angular blocky, granular in upper 2 cm; hard dry; many very fine tubular pores; common very fine roots; clear, wavy boundary; pH 5.5.

A12 -- 10-26 cm. Dark grey (10YR4/1), many, fine and medium, distinct yellowish red mottles as coatings along root channels and on ped faces; clay; moderate coarse angular blocky, breaking into small blocks; firm moist; common pressure faces; common very fine tubular and very fine and fine interstitial pores; common very fine roots; clear, smooth boundary; pH 6.5.

B21 -- 26-70 cm. Dark grey (10YR4/1), many, medium and coarse, distinct dark yellowish brown mottles; clay; weak to moderate coarse angular blocky; many pressure faces and common intersecting slickensides; many very fine interstitial, few very fine tubular pores; common fine manganese nodules; common fine and very fine roots; gradual, smooth boundary; pH 7.0.

B22 -- 70-160 cm.Dark grey (5Y 4/1), many fine and medium, distinct brown mottles; clay; weak to moderate medium angular blocky; firm moist; many intersecting slickensides; common very fine

interstitial pores; common fine manganese nodules and secondary lime concretions in the deeper part; pH 8.0.

C1 -- 160-250 cm. Brown (10YR4/3), common, medium, faint dark brown mottles; clay; common fine manganese nodules and fine calcite crystals; pH 8.0.

Range of Profile Features: The A horizon is from 20 cm. to 40 cm. thick, has dark grey (10YR4/1) matrix colours and has a strong fine blocky structure becoming granular in the uppermost layer. Field pH ranges from 5.5 to 6.5.

The subsoil, provisionally termed B horizon, has a lower boundary deeper than 150 cm. from the soil surface. Matrix colours are dark grey becoming grey in the deeper layers. Hues range from 10YR to 5Y. Field pH values range from 6.5 to 7.0 in the upper B horizon; but below approximately 1 m. pH values may in crease to 8.0. Limestone concretions, when found, are restricted to the deeper layers below approximately 1 m. from the soil surface.

Similar Soil Series:

Wathana - similar profile but dries out for longer periods (Usterts) and is derived from basalt, limestone and andesite.

Chong Kae - has lower pH values and characteristic fine red mottles.

<u>Principal Associated Soils</u>: These include Lop Buri series soils which occupy higher positions and are better drained, and Chong Kae series soils which occupy similar positions on the lower parts of the sem-recent terraces transitional to the flood plains.

KAMPHAENG SAEN SERIES (Ks)

<u>Distribution</u>: Occupies large extent in the southwestern part of the Central Plain.

Setting: Kamphaeng Saen soils are formed from semi-recent alluvium and occur on old levees and breach deposits of the semi-recent terrace. Relief is flat to nearly flat, wity a slightly undulating micro-relief. Slopes are 1% or less. Elevation ranges from 6 to 20 m. above sea level. The climate is Tropical Savanna (Koppen 'Aw'). Mean annual precipitation ranges from 800 to 1,600 mm. Mean annual temperature is 27°C.

<u>Drainage and Permeability</u>: Well drained. Permeability is moderate and runoff is slow. Groundwater level is below 1.5 m. from the soil surface throughout the year.

<u>Vegetation and Land Use</u>: Mainly used for settlement sites, gardens and orchards; or are put to upland crops such as maize, cotton and sugar cane.

Characteristic Profile Features: Kamphaeng Saen series is a mamber of the fine loamy, mixed family of Nan Calcic Brown Soils (National), Udic Haplustalfs (USDA). They are deep, friable, slightly acid over neutral to mildly alkaline soils. They are characterized by a brown or dark brown loam or clay loam A horizon, overlying a brown or strong brown clay loam, weakly developed, argillic B horizon. Fine mica flakes occur in all horizona; but not enough for the micaceous family.

Typifying Pedon: Kamphaeng Saen clay loam -- Orchard

form Amphoe Muang, Nakhon Pathom province - Code SW 53/6 (Type

Location) (moist colours unless otherwise stated)

Ap -- 0-30 cm. Brown to dark brown (10YR4/3); clay loam; weak coarse subangular blocky; nonsticky and nonplastic, friable moist, hard dry; many very fine interstitial and tubular pores; common very fine roots; diffuse, smooth boundary; pH 6.5.

B2t -- 30-65 cm. Dark yellowish brown (10YR4/4); clay loam to clay; weak to moderate medium subangular blocky; slightly sticky and slightly plastic, friable moist, hard dry; thin, broken, brown to dark brown clay coatings in pores and on ped faces; many fine and very fine tubular and interstitial pores; few mica flakes; many fine and medium and common very fine roots; gradual, smooth boundary; pH 7.5.

B3 -- 65-90 cm. Brown to dark brown (7.5YR4/4); clay loam; moderate medium and fine subangular blocky; slightly sticky and slightly plastic, friable moist; thin patchy clay coatings; many fine and very fine interstitial and tubular pores; many white lime pseudo-mycelia, strongly calcareous, common mica flakes; common fine and very fine roots; gradual, smooth boundary; pH 8.0.

C -- 90-130 cm.Strong brown (7.5YR5/6); loam; moderate medium subangular blocky; nonsticky and nonplastic, friable moist; many very fine interstitial mycelia, many mica flakes; few very fine roots; pH 8.0. Range of Profile Features: The A horizon is from 20 to 40 cm. thick, has 10YR or 7.5YR hues, values of 3 through 5 and chromas of 2 and 4 in 7.5YR hue, and 2 or 3 in 10YR hue. Texture may be sandy clay loam and structure is weak coarse blocky. Field pH values range from 6.0 to 6.5.

The B horizon has 10YR or 7.5YR hues, values of 4 and 5 and chromas of 4 through 8. Reddish brown (5YR5/4 or 4/4) colours may also occur in the deeper subsoil. Textures may be silty clay loam and structure is weak to moderate, medium blocky. Field pH values range from 7.0 to 8.0. The lower B and C horizons commonly contain white, soft, powdery lime; but its presence is not diagnostic for the series.

Similar Soil Series:

Pran Buri -- does not contain micas or soft, powdery lime and has somewhat lower pH values.

Ruso -- has a similar profile but lower pH values throughout, no soft, powdery lime and has an udic moisture regime.

<u>Principal Associated Soils</u>: These include Nakhon Pathom and Nakhon Phanom series on the semi-recent terrace, and Saraburi series on the transition between the semi-recent terrace and flood plain.

PAK CHONG SERIES (Pc)

<u>Distribution</u>: Occupies moderate extent in Central Highlands and small extent in North Thailand.

Setting: Pak Chong soils are formed from residuum and local colluvium from shale in association with limestone and occur on the dissected erosion surface or partial peneplain. Relief is undulating to rolling which slopes range from 2 to 8 percent. Elevation is variable above sea level, but mainly not above 400 m. The climate is Tropical Savanna (Koppen 'Aw'). Average annual precipitation varies from 1,100 mm. to 1,400 mm. Mean annual air temperature is from 26°C to 28°C.

<u>Drainage</u>, <u>Permeability and Runoff</u>: is well drained soils. Permeability is moderate to slow. Runoff is rapid to medium.

<u>Vegetation and Land Use</u>: Originally mixed diciduous forest, but mainly cleared for upland crop cultivation such as corn, cotton, beans, sorghum, castor bean and some fruit crops.

Characteristic Profile Features: The Pak Chong series is a member of the very fine clayey, Kaolinitic, acid family of Peddish Brown Lateritic soils (National), oxic Paleustults? (USDA). They are deep soils and are characterized by a dark reddish brown clay or silty clay A horizon overlying a dark reddish brown or reddish brown clay upper argillic B horizon which in turn overlies red or dark red clay lower argillic B horizon. Reaction is slightly acid to neutral over medium to very strongly acid.

Typifying Pedon: A profile taken from 200 m. northwest of Kasetsart University Training Farm, Ban Pang Sok, Amphoe Pak Chong, Nakhon Ratchasima Province: Province; Profile Code NE-S

20/30. (moist colors unless otherwise stated).

Ap -- 0-30 cm. Dark reddish brown (2.5YR3/4) clay; moderate medium subangular blocky structure; friable, slightly sticky, plastic; few fine interstitial pores; common fine roots; slightly acid; gradual smooth boundary to B21t

B21t-- 30-60 cm. Dark reddish brown (2.5YR3/4) clay; moderate medium subangular blocky structure; friable, slightly sticky, plastic; common moderately thisck continuous clay coating on ped faces; many fine roots; very strongly acid; gradual smooth boundary to B22t

B22t-- 60-100 cm.Red (2.5YR4/6) clay; moderate medium subangular blocky structure; friable, sticky, plastic; many moderately thick continuous clay coating on ped faces; few fine roots; very strongly acid.

Type Location: The Pak Chong series was name for Amphoe Pak Chong in which soils of this series were first described by R.L. Dendleton in 1929 and shown on the map as "Pak Chong loam". Then these soils were revised by F.R. Moormann et.al. in 1964 and were called as "Pak Chong series".

Range of Profile Features: The thickness of the A horizon range from 10 to 25 cm. and has 2.5YR or 5YR hues, values of 3 or 4, and chromas of 3 or 4 in 5YR hue and 2 to 6 in 2.5YR hues. Textures of silty clay loam or clay loam may occur. Structure is moderate fine to medium granular at upper most of layer and moderate medium blocky at lower parts of the horizon. Field pH value is from 6.0 to 8.0.

The B horizon has 2.5YR or 10R hues, values of 3 to 5, and chromas of 4 to 8. Few fine hard subrounded and/or rounded iron-manganese nodules may occur. Field pH value is from 4.5 to 5.5.

Similar Soil Series:

Chok chai Series (Ci) - semilar profile, but are derived from basalt and some andesite.

Ao Luk Series (Ak) - similar profile, but have udic moisture regimes.

Loei Series (Lo) - semilar profile, but are derived from granite and shale and contain lower clay fraction.

<u>Principal Associated Soils</u>: These include Muak Lek, Lop Buri, and Tha Kli Soils. The Muak Lek and Tha Kli soils occupy on the higher topography while the Lop Buri soils occupy on slightly lower topography.

NARATHIWAT SERIES (Nw)

Classification : Organic soils

Described by : P.Vijarnsorn

Date : September 12, 1974

Location : Plu Thung Ka, Amphoe Yi-ngo,

Changwat Narathiwat

Elevation : Approximately 5 m. above sea level

Relief and slope : Flat; 0 - 1 % slope

Physiography : Domed bogs

Natural Vegetation or Land Use : Swamp forest

Climate :

- Climate type : Tropical monsoon climate (Koppen "Am")

- Annual rainfall : 2,644.8 mm.

- Mean temperature : 27.0°C

Parent material : Organic material

Drainage : Very poorly drained

Permeability : Estimated to be slow

Run off : Slow

Ground water depth : Near the surface

Other : -----

Horizon Depth (cm) Description

0 - 60 Dark reddish brown (5YR 3/2, 5YR

3/3) peaty muck; peaty part containing very fine and fine roots mostly; structureless; non sticky,

non plastic; very strongly acid (pH

5.0).

0, 60 - 120 Dark reddish brown (5YR 3/2) peaty

muck; peaty part containing very fine and fine roots mostly; structureless; non sticky non

plastic; very strongly acid (pH 5.0).

- Remark: Soil pH and texture are determined in laboratory.

 Color are for fully moist soil except where designated.
- Source: 1. Soil Survey Division. 1979. <u>Detailed Reconnaissance</u>

 <u>Soil Map of Lop Buri Province</u> Ministry of

 Agriculture and Cooperative, Bangkok.
 - 2. Soil Survey Division. 1979. <u>Detailed Recomaissance</u>

 <u>Soil Map of Nakhon Pathom Province</u> Ministry of

 Agriculture and Cooperative, Bangkok.
 - 3. Soil Survey Division. 1979. <u>Detailed</u>

 <u>Reconnaissance Soil Map of Nakhon Ratchasima</u>

 <u>Province.</u> Ministry of Agriculture and Cooperative,

 Bangkok.
 - 4. Soil Survey Division. 1979. <u>Detailed Reconnaissance</u>

 <u>Soil Map of Narathiwat Province</u> Ministry of Agriculture and Cooperative, Bangkok.

Results of Soil Analysis

Ban Mi Series : Bm

Laboratory No. of	Topsoi	1 : Pb	494 T			Н	orizon	A	Horiz		n B		orizon	C
Horizon Depth (cm)	Ap 0-10	A ₁₂ 10-26	B _n 26-70	B _{az} 70-160	C ₁ 160-250		Min	Ave.	Max	Min	Ave.	Max	Min	Ave.
Particle Size Analysis (%) (USDA Grading)														.1
Sand	3.0	2.0	1.5	6.0	9.5	11.9	2.38	7.98	0.10					
silt	29.0	29.0	26.0	21.5	25.5	41.75		31.45						4.9
Clay	68.0	69.0	72.5	72.5	65.0	68.62	48.58	60.57		51.46				14.6
pH 1:1 H ₂ O	5.4	5.35	6.95	8.3	7.9	6.9	5.37	6.36		6.82	61.98 7.34			80.5
11 KC1	3.85	3.6	4.9	6.6	7.2	5.96	3.7	5.12		5.82	5.93			7.2
CaCO ₃ %	1.79	1.94	1.79	5.51	5.07	3.28	1.4	2.19	4.29	2.5	3.4			5.6
Moisture Air to									7.20		5.4			2.2
Over dry %	6.6	6.8	8.0	8.1	7.9	6.72	5.62	6.17	8.07	. 4.68	6.38			-
1 : 5 EC X 10°	67.0	48.0	29.0	85.0	1000.0	965.31	55.31	404.21	767.49	66 61	417.05			04.0
C *	1.38	0.77	0.35	0.22	0.13	1.2	0.68	0.96	0.36	0.26	0.31			94.0
Exchange										0.20	0.31			0.4
Capacity & Cation														
Meq. 100 g)														
a	27.3	29.8	35.9	45.7	80.6	34.47	25.6	29.64	39.81	31.12	35.47			22.6
lg .	0.5	0.5	0.5	0.7	0.9	16.6	0.5	6.0	0.63	0.43	0.53			
	0.7	0.4	0.3	0.2	0.2	1.0	0.17	0.56	0.03	0.43				23.8
a .	0.7	0.9	1.3	3.6	9.5	2.72	0.32	1.58			0.17			0.7
					0.0		0.02	1.00	2.84	2.78	2.81			3.8

Ban Mi Series : Em (cont.)

				1/4		1		aster Ho						
Laboratory No. of	Laboratory No. of Topsoil : Pb 494 1,336 !					1	Horizon	1 A Horizon B		а В	H	lorizon	С	
Horizon Depth (cm)	Ap 0-10	A ₁₂ 10-26	B ₂ .	B ₂₂	C ₁ 160-250	Max	Min	Ave.	Max	Min	Ave.	Max	Min	Ave.
(B) Sum (Ca+Mg+K+Na) (A) Extr Acidity (B+A) Sum (C) CEC Soil CEC, 100 g clay Base Saturation % 3 X 100	29.2 17.2 46.4 50.0 73.5	31.6 14.2 45.8 49.7 72.0	38.0 8.5 46.5 51.8 71.5	50.2 4.3 54.5 53.3 73.5	91.2 2.9 94.1 46.8 72.0	44.4 15.35 52.7 49.82 93.53	30.68 4.71 42.97 37.79 72.58	9.45 47.23 44.2	5 5.68 51.87 52.81	4.51 38.93 33.45	5.1 45.4 43.13			50.9 8.5 59.4 49.2 60.99
B+A	58 63	64 ; 69	73 82	94	195	100.61	61.69 66.69		104.33		95.72		المترادي	103
(ppm) Bray N.2 (ppm) Ammon	7.1	-5.5	4.3	9.1	15.8	10.3	2.25	6.22	7.52	1.97	88.28		315	9.7
cetate	286	193	131	119	113 2	28.77	78	159.59	122.94	67.75	95.35			250

Number of Profile Sampled : SW 51/3, C 4/11, C 4/33

Date of Compilation : February 1976

Results of Soil Analysis

Kamphaen Saen series : Ks

	100			Mar.							Uo	rizon C		
Laboratory No. of	Topsoil	: M -	543		Horizon A			Н	Horizon B			nor izon o		
Torizon Depth (cm)	Ар	Bet	B _o	С	Max	Min	Ave.	Max	Min	Ave.	Max	Min	Ave.	
Particle Size														
Analysis (%)														
(USDA Grading)														
Sand	42.5	21.5	28.0	60.0	55.88	3.18	37.78	57.46	4.4	18.13	85.0	8.5	51.1	
silt	40.0	50.5	56.0	28.0	62.41	11.01	39.59	58.61	30.46	45.74	67.5	9.5	35.0	
Clay	17.5	.28.0	16.0	12.0	39.38	9.33	22.64	36.99	12.08	24.64	24.0	5.5	13.8	
рН 1:1 Н ₂ О	7.4	6.3	7.5	8.35	7.4	5.65	6.68	8.34	5.19	6.78	8.35	5.4	7.0	
1:1 KC1	6.4	5.7	7.0	7.5	6.7	4.85	6.02	6.7	3.86	5.46	7.5	4.0	6.1	
CaCO ₃ %	0.90	0.30	6.90	3.00	2.12	0.39	1.04	8.62	0.39	2.14	21.45	0.57	8.3	
Moisture Air to														
Over dry %	1.1	1.8	2.7	1.1	4.01	0.79	1.79	2.42	0.36	1.51	1.2	0.3	0.8	
Conductivity														
1 : 5 EC X 10°	10.0	680.0	800.0	700.0	677.81	10.0	194.02	146.66	725.66	398.45	700.0		337.4	
C %	3.22	2.02	1.42	0.82	17.68	0.65	4.94	1.77	0.23	1.0	0.83	0.49	0.7	
Exchange				2.0										
Capacity & Cation														
(Meq. 100 g)													Dr.	
Ca	10.6	16.6	37.2	19.5	25.54	5.12	10.61	37.86	1.34	14.84	37.5	0.7	19.2	
Mg	1.6	2.1	5.1	2.7	3.71	1.18	2.07	4.9	0.51	2.16	2.7	0.3	1.1	
K	0.1	0.2	0.2	0.1	0.53	0.82	0.41	0.8	0.2	0.35	0.2	0.1	0.1	
Na	0.8	1.7	11.0	6.6	1.2	0.2	0.54	5.58	0.23	1.62	6.6	0.2	2.3	

Kamphaen Saen series : Ks (Cont.)

Laboratory No. of	Topsoi	1 : M -	543		Н	Horizon A			Horizon B			Horizon C		
(B)Sum(Ca+Mg+K+Na)	13.4	20.6	53.5	28.9	27.36	7.49	13.63	38.04	2.59	18.98	38.2	1.5	22.8	
(A) Ext.r Acidity	3.3	3.8	0.5	0.2	6.31	3.0	4.0	5.79	1.24	3.36	2.1	. ò	0.77	
(B+A) Sum	16.7	24.4	54.0	29.1	30.76	12.3	18.39	40.18	8.38	24.34	38.2	3.6	23.63	
(C) CEC Soil	15.1	16.1	15.1	11.5	22.88	9.79	15.52	18.6	5.39	13.41	12.1	2.7	0.77	
CEC, 100 g clay	85.7	57.5	94.4	95.8	121.83	69.13	92.7	101.67	44.9	66.18	95.8	49.1	65.1	
Base Saturation %														
3 X 100														
	89	100	100	100	119.0	53.66	88.59	234.12	31.0	110.1	316.0	42.0	152.67	
3 X 100														
BIA	80	100	100	100	88.24	70.69	78.02	100	47.46	77.74	100	55	85	
(ppm) Bray N.2	47.6	82.4	44.7	20.5	75.29	12.33	40.38	60.00	3.4	18.08	20.5	3.6	10.37	
(ppm) Ammon										23.00	20.0		10.31	
Acetate	175	70	88	47	187.71	79.25	129.03	400.58	47.44	151.31	73	38	52.67	

Number of Profile Sampled: N-35/70, N-38/44 NC-42/6, SW-53/6, SW-56/3

Date of Complitation :

Rssults of Soil Analysis

Pak Chong Series : Pc

							- A-5.*					
Laboratory No. of	Topso	il : Pa	- 1273	Но	orizon	A	H	lorizon	В		Horizon	ı C
Horizon Depth (cm)	Лр 0-30	B ₂₁ t.	B ₂₂ t	Max	Min	Ave.	Max	Min	Ave.	Max	Min	Ave
Particle Size Analysis (%) (USDA Grading)												
Sand	7.0	4.0	3.0	29.5	5.5	14.09	17.36	3.3	8.29			
Silt	24.0	16.0	11.0	40.5	11.5	30.41		6.87				
Clay	69.0	80.0	86.0	83.0	30.5	55.49		57.68		85 July 1997		
PH 1:1 H ₂ O	6.1	4.35	4.0	7.1	5.8	6.38		4.15	5.23			
1:1 KC1	5.3	3.65	3.65	6.2	4.9	5.51	1	3.65	4.78			
Caco ₃ %	1.2	0.6	0.75	2.58	0.71	1.23	1.18	0.69	0.88			
loisture Air to												
over dry %	7.7	6.4	6.3	7.7	3.4	5.46	6.34	3.21	4.6			
Conductivity	1 11 14											
: 5 EC X 10°	52.0	64.0	100.0	640.0	18.0	154.18	140.25	16.97	59.89			
% xchange	1.14	0.27	0.62	3.73	0.76	1.56	2.12	0.27	0.77			
apacity & Cation												
a	10.1	3.2	0.7	25.0	7.8	13.6	14.77	1.77	0.00			
g	1.8	1.0	0.7	5.2	1.6	2.5	2.62	0.83	6.99			1
	0.1	0.1	0.1	1.93	0.1	0.83	0.5		1.52			
1	0.2	0.5	0.2	0.2	0.2	0.2	0.33	0.1	0.23	W		

Pak Chong Series : Pc (Cont.)

Laboratory No. of	Topsoi	l : Pa	- 1273	Hor	izon A		Ho	orizon B		ŀ	lorizon	С
Horizon Depth (cm)	Ap	B ₂₁ t.	B ₂₂ t 60-100	Max	Min	Ave.	Max	Min	Ave.	Max	Min	Ave
(B)Sum(Ca+Mg+K+Na)	12.2	4.5	1.7	31.9	10.0	17.13	16.56	2.9	8.92			
(A)Extr Acidity	8.5	13.3	15.6	8.8	4.1	7.16	14.61	8.32	11.47			
(B+A) Sum	20.7	17.8	17.3	36.0	18.8	24.29	26.94	16.96	20.38			
(C) CEC Soil	20.4	16.5	17.3	25.05	16.88	21.01	27.22	14.67	18.00			
CEC, 100 g clay Base Saturation %	29.6	48.5	66.5	81.6	21.4	45.24	58.79	17.54	34.11			
<u>B</u> X 100												
C B X 100 .	60	27	10	129	56	80.17	74.69	17.29	48.27			
B+A	59	25	10.	89	53	67.79	58.09	16.43	40.85			
P (ppm) Bray N.2 K (ppm) Ammon	5.1	2.0	1.8	49.2	3.05	14.07	6.74	1.04	3.29			
Acetate	48	24	24	933.7	48	300.74	227.44	24.00	96.61			

Number of Profile Sampled:

NE-S20/30, NE-N30/61

Date of Compilation : February 1976

APPENDIX B

Table 1 Adsorption of Cadmium by Natural Soil Series

Soil series	Initial Cd conc.* (µeq/l)	Equilibrium Cd conc.* c, (jueq/l)	Adsorbed Cd*
Bm	227.7	51.2	17.9
	533.8	154.7	38.0
	978.6	440.0	54.4
	2063.9	1063.0	95.3
	2295.2	1483.2	82.5
Ks	227.7	179.5	4.8
	533.8	348.7	17.6
	978.6	848.0	12.8
	2063.9	1710.2	34.0
	2295.2	2035.0	26.7
Pc	227.7	53.2	17.9
	533.8	165.0	34.9
	978.6	469.7	48.3
	2063.9	1283.2	77.1
	2295.2	1601.3	70.7
Nw	220.4	155.5	6.1
	449.4	320.0	12.8
	865.8	667.0	19.1
	1674.6	1399.5	25.8
	2307.6	2041.2	25.5

^{*} The mean value from 4 replications

Table 2 Adsorption of Nickel by Natural Soil Series

Soil Series	Initial Ni Conc.*	Equilibrium Ni Conc.	Adsorbed Ni*
Bm	170.4	34.1	13.5
	354.3	34.1	22.8
	681.4	72.4	61.1
	1635.4	339.8	130.6
	1737.6	414.8	136.4
Ks	170.4	111.6	5.7
	354.3	247.0	10.5
	681.4	448.9	22.5
	1635.4	1376.4	25.8
	1737.6	1466.0	27.0
Pc	170.4	34.1	13.8
	354.3	34.1	22.6
	681.4	71.6	59.1
	1635.4	416.5	121.4
	1737.6	397.8	128.1
Nw	221.5	158.2	5.5
	432.7	320.7	9.1
	885.9	655.4	15.8
	1754.7	1345.0	36.1
	2248.7	1659.9	31.4

^{*} The mean value from 4 replications

Table 3 Adsorption of Zinc by Natural Soils

Soil Series	Initial Zn Conc.*	Equilibrium Zn Conc* c, (µeq/l)	Adsorbed Zn	
Bm	281.4	9.2	27.6	
	572.0	27.6	56.2	
	1139.5	42.4	108.9	
	2890.1	102.4	266.5	
	3212.0	310.6	283.7	
Ks	281.4	205.0	7.5	
	572.0	403.0	17.3	
	1139.5	891.0	24.7	
	2890.1	2799.0	39.9	
	3212.0	2344.0	54.1	
Pc	281.4	9.2	26.7	
	572.0	16.3	56.0	
	1139.5	43.5	105.9	
	2890.1	1258.0	189.1	
	3212.0	868.0	197.2	
Nw	215.7	135.4	8.2	
	415.4	282.8	14.1	
	821.6	631.2	24.7	
	1719.2	1277.0	45.5	
	1984.7	1678.0	55.7	

^{*} The mean value from 4 replications

Table 4 Adsorption of Cadmium by Calcium - Saturated Soils

Soil series		Equilibrium Cd cone*	
	(µeq/1)	с, (µeq/1)	n, (µeq/g soil)
Ca-sat.Bm	187.3	37.0	15.3
	415.4	63.0	35.2
	776.6	260.6	51.5
	1568.9	771.6	79.9
	1916.2	1021.4	89.6
Ca-sat.Ks	187.3	75.6	11.0
	415.4	119.7	28.4
	776.6	458.0	31.2
	1568.9	1050.0	49.4
	1916.2	1380.0	52.2
Ca-sat.Pc	187.3	35.2	15.3
	415.4	59.6	35.2
	776.6	315.0	41.31
	1568.9	830.0	74.37
	1916.2	1167.0	75.01

^{*} The means value from 4 replications

Table 5 Adsorption of Nickel by Calcium - Saturated Soils

Soil series	Initial Ni conc.*	Equilibrium Ni conc.*	Adsorbed Ni*
Ca-sat.Bm	187.0	30.2	16.0
	400.0	31.8	37.0
	814.6	93.8	71.3
	1686.5	318.2	137.6
	1994.8	370.6	152.1
Ca-sat.Ks	187.0	68.3	11.9
	400.0	194.0	20.1
	814.6	485.8	32.3
	1686.5	1209.0	46.6
	1914.8	1497.0	41.2
Ca-sat.Pc	187.0	14.4	17.3
	400.0	32.4	36.3
	814.6	77.7	73.6
	1686.5	345.7	134.7
	1914.8	449.0	145.4

^{*} The mean value from 4 replications

Table 6 Adsorption of Zinc by Calcium - Saturated Soils

Soil series	Initial Zn conc.*	Equilibrium Zn conc.*	n, ueq/g	
Ca-sat.Bm	163.6	9.6	15.8	
	452.7	26.8	42.3	
	1079.8	59.4	103.7	
	2138.3	222.5	194.9	
	2523.7	219.4	234.3	
Ca-sat.Ks	163.6	94.5	6.9	
	452.7	93.6	36.1	
	1079.8	526.9	54.5	
	2138.3	1619.2	51.7	
	2523.7	1771.2	74.7	
Ca-sat.Pc	163.6	6.6	15.8	
	452.7	13.6	44.3	
	1079.8	57.2	103.0	
	2138.3	376.2	179.8	
	2523.7	355.6	216.8	

^{*} The mean value from 4 replications

APPENDIX C

Table 7 Effect of pH on the Adsorption of Cd by Bm series

				1
	pH of heavy metal solutions			
	4.0	5.0	6.0	
	heavy m			
	86.03	88.70	69.57	
	87.28	88.03	72.84	
	85.41	95.68	75.28	
	84.29	75.75	76.23	
Total (T)	343.91	348.16	293.92	985.99
Mean (ȳ)	85.98	87.04	73.48	
Variance (S)2	1.93	68.56	8.82	

Example I The test for equality of Variances for Cochran's test (data from Table 7)

1.
$$H_o : S_1^2 = S_2^2 = S_3^2$$

2. H₁ : at least two of the variance are not equal

3. Significant level
$$, \alpha = 0.01$$

4. Critrical value for cachran's test, G = 0.8831

5. Computations

$$G = \frac{\text{largest S}_{i}^{2}}{k S_{i}^{2}}$$

$$i = 1$$

$$= \frac{68.56}{1.93+68.56+8.82} = 0.8644$$

Then, accept Ho, the variances are equal

Example II The test for equality of the means, analysis of variance for the one-way classification

From the assumption of equal variances for the k populations when the samples, n are selected. It will be assumed that the k populations are independent and normally distributed with mean, μ_1 , μ_2 ... μ_k and variance, S^2 (data from Table 4.12).

2. H₁: at least two of the means are not equal

4. Critrical region, F > 8.02 with degree of freedom $V_1 = 2$, $V_2 = 9$

5. Computations:

Total sum of square (SST) =
$$\sum_{i=1}^{k} \sum_{j=1}^{n} y_{ij}^{z} - \frac{T^{z}}{nk}$$

SST =
$$(86.93)^2 + 87.28)^2 + ... + (76.23)^2 - (985.99)^2$$
(4)(3)

= 693.05

Treatment sum of squares (SSA) =
$$\sum_{i=1}^{k} \frac{T_i^2}{n} - \frac{T^2}{nk}$$

SSA =
$$(343.91)^2 + (348.16)^2 + (293.92)^2 - (985.99)^2$$

$$4 \qquad (4)(3)$$

error sum of squares (SSE) = SST-SSA

SSE = 693.05-454.92 = 238.93

treatment mean square (MST) = <u>SSA</u>

k-1

MST = 454.92 = 227.460

2

error mean square (MSE) = SSE

k(n-1)

MSE =
$$\frac{238.13}{3(4-1)}$$
 = 26.459

computed F = MST

MSE

F = 227.460 = 8.596

26.459

reject H_0 , accept H_1 , the means are not equal at p = 0.01

Analysis of Variance for the Date of Table 7 (p = 0.01)

Source	qŧ	SS	MS	F
Treatment	2	454.92	227.460	8.596*
Error	. 9	238.13	26.459	
Total	11	693.05		

Table 8 Effect of pH on the Adsorption of Cd by Ks series

	4.0	5.0	6.0	
	heavy metal	adsorbed (Med/g)		
	18.90	29.90	24.46	
	11.21	29.70	20.67	
	19.79	28.61	23.53	
	19.81	22.68	16.61	
Total (T)	69.71	110.89	85.27	265.87
Mean (\vec{y})	17.43	27.72	21.32	
Variance (S)2.	17.38	11.63	12.46	

Analysis of Variance for the Data of Table 8 (p = 0.01)

Source	df	SS	Ms	F
Freatment	2	216.19	108.095	7.827
Error	9	124.29	13.81	
Total	11	340.48		

Remark : F(2, 9) = 8.02

Table 9 Effect of pH on the Adsorption of Cd by Pc series

	pH of hea	avy metal sol	utions	
1	4.0	5.0	6.0	
-	heavy meta	al adsorbed (
	61.74	60.72	51.69	
	58.75	65.10	55.49	
	63.33	58.37	56.26	
	68.71	64.11	-	
Total (T)	252.53	248.30	163.44	664.27
Mean (ȳ)	63.13	62.07	54.48	
Variance (S) ²	17.39	9.61	6.00	

Analysis of Variance for the Data of Table 9 (p = 0.01)

df	SS	MS	F
2	146.22	73.11	6.28
8	93.12	11.64	
10	239.34		
	2 8	2 146.22 8 93.12	2 146.22 73.11 8 93.12 11.64

Remark : F(2, 8) = 8.65

Table 10 Effect of pH on the adsorption of Ni by Bm series

	pH of heavy metal solutions			
	4.0	5.0	6.0	
	heavy met	al adsorbed	(hed/g)	
	142.08	148.87	127.65	
	154.14	147.35	125.57	
	150.31	146.31	131.31	
	143.43	141.61	126.75	
Total (T)	589.96	584.14	511.28	1685.38
Mean (ȳ)	147.49	146.03	127.82	
Variance (S)2.	32.60	9.80	6.15	

Analysis of Variance for the Data of Table 10 (p = 0.01)

Source	df	SS	MS	F
Treatment	2	961.09	480.545	29.669*
Error	9	145.77	16.197	
Total	11	1106.86		

Remark : F(2, 9) = 8.02

Table 11 Effect of pH on the adsorption of Ni by Ks series

	4.0	₹5.0	6.0	
	heavy meta	l adsorbed (µeq/g)	
	21.57	21.11	25.36	
	22.23	17.03	25.48	
	22.38	22.52	25.12	
	16.87	21.70	23.72	
Total (T)	83.05	82.36	99.68	265.09
Mean (\$\vec{y}\$)	20.76	20.59	24.92	
Variance (S)2.	6.86	5.95	0.66	

Analysis of Variance for the Data of Table 11 (p = 0.01)

Source	df	SS	MS	F	
Treatment	2	48.10	24.05	5.36	
Error	9	40.45	4.49		
Total	11	88.55			

Remark : F(2, 9) = 8.02

Table 12 Effect of pH on the adsorption of Ni by Pc series

	pH of h	eavy metal so	lutions	
	4.0	5.0	6.0	
	heavy me	tal adsorbed	(jueq/g)	
	142.40	122.41	101.05	
	113.28	133.96	94.66	
	122.46	122.29	105.16	
	112.88	-	101.39	
Total (T)	491.02	378.66	402.26	1271.94
Mean (ȳ)	122.76	126.22	100.56	
Variance (S) ²	190.99	44.89	18.92	

Analysis of Variance for the Data of Table 12 (p = 0.01)

Total	10	2167.	39	
Error	8	120.	06 90.00	7
Treatment	2	1447.	33 723.66	5 8.04
Source	df	SS	MS	F

Remark : F(2, 8) = 8.65

Table 13 Effect of pH on the Adsorption of Zn by Bm series

	pH of heavy metal solutions				
	4.0	5.0	6.0	1	
	heavy me	tal adsorbed (
	221.69	199.16	215.09		
	244.81	207.45	224.08		
	234.34	211.84	218.95		
	233.93	212.11	206.88		
Total (T)	934.77	830.56	865.00	2630.33	
Mean (ȳ)	233.69	207.64	216.25		
Variance (S	89.30	36.48	52.56		

Analysis of Variance for the Data of Table 13 (p = 0.01)

Source	df	SS	MS	F
Treatment	2	1409.48	704.740	11.845*
Error	9	535.44	59.493	
Total	11	1944.92		

Remark : F(2, 9) = 8.02

Table 14 Effect of pH on the adsorption of Zn by Ks series

	4.0	5.0	6.0	
	heavy met	al adsorbed (µeq/g)	
	25.27	30.93	32.12	
	26.79	20.13	26.41	
	24.85	21.09	44.97	
	_	21.73	37.92	
Total (T)	76.91	93.88	141.42	312.21
Mean (\overline{y})	25.64	23.47	35.35	
Variance (S	1.04	25.20	63.20	

Analysis of Variance for the Data of Table 14 (p = 0.01)

Source	df	SS	MS	F
Treatment	2	313.61	156.805	4.697
Error	8	267.09	33.386	
Total	10	580.70		

Remark : F(2, 8) = 8.65

Table 15 Effect of pH to the adsorption of Zn by Pc series

	4.0	5.0	6.0	
	heavy metal	l adsorbed (hed/a)	
	178.21	173.35	176.05	
	184.88	170.71	184.16	
	183.85	173.22	194.29	
	188.01	163.91	172.05	
Total (T)	734.86	681.19	726.55	2142.60
Mean (\overline{y})	183.71	170.30	181.64	
Variance(S) ²	17.06	19.62	16.43	

Analysis of Variance for the Date of Table 15 (p = 0.01)

Source	df	SS	MS	F
Treatment	2	417.25	208.625	4.699
Error .	9	399.54	44.393	
Total	11	816.79		*

Remark : F(2, 9) = 8.02

APPENDIX D

Cation - Exchange Capacity by Ammonium Saturation

1. Ammomium Saturation

- Reagents : Ammonium acetate (NH₄OAc) 1 N adjusted to pH 7.0 with NH₄OH
 - Isopropyl alcohol 99%
 - Ammonium chloride 1 N (NH $_4$ Cl) adjust to pH 7.0 with NH $_4$ OH
 - Ammonium oxalate (NH₄)₂ C₂O₄ H₂O 10%
 - Dilute ammonium hydroxide (NH,OH)
 - Silver mitrate AgNO₃ 0.10 N

Procedure

- Place 10 g. of 2 mm., air-dried soil in 500 ml Erlenmeyer flask, add 250 ml of 1 N NH₄OAc.
- 2. Shake the flask thoroughly, allow it to stand overnight.
- 3. Filter the soil and leach with neutral NH₄OAc till no test for calcium can be obtained in the effuent solution.
- 4. Leach the soil four times with neutral, 1N NH₄Cl and once with 0.25 N NH₄Cl.
- 5. Wash out the electrolyte with 150 to 200 ml of isopropyl alcohol
- Allow the soil to drain throughly when chloride test in the leachate becomes neglible.
- Determine the adsorbed ammonium by the acid - NaCl method

- 2. Displacement by the acid-NaCl.
 - Reagents: Sodium chloride, NaCl (acidified) 10% approximately 0.005 N with respect to acidity
 - Sodium hydroxide, NaOH, 1N
 - Boric acid, H₃BO₃ 2% solution
 - Standard Sulfuric acid, H2SO4 0.1 N
 - Bromocresol green methyl red mixed indicator

Procedure

- Leach the ammonium-saturated soil form section 1 with 10% acidified NaCl till 225 ml have passed.
- 2. Transfer the leachate quantitatively to an 800 ml Kjeldahl flask, add 25 ml of 1N NaOH and distill 60 ml of the soluton into 50 ml of 2% H₃BO₃.
- 3. Titrate the boric acid solution with standard 0.1 N H₂SO₄, by using 10 drops of bromocresol green-methyl red mixed indicator At the end point, the color change is from Bluish green through bluish purple to pink.
- 4. Run blanks on the reagents for the titration figure correction and calculate the milliequivalents of ammonium in 100 g of soil.

Free Iron Oxide

(SCS; 1972)

Reagents

- Sodium dithionite powder, Na S O4
- Sodium citrate pentahydrate powder Na_3 C_6 H_5 $O_7 \cdot 5H_2O$
- Superfloc flocculating agent, 0.2% in water

Procedure

- Weight 1 to 4 g of 80-mesh soil into nursing bottle.
- 2. Add 2 g of Na₂ S₂ O₄ and 20 to 25 g of Na₃ C₆ H₅ O₇ \cdot 5H₂O.
- 3. Add deionized water to the 4-oz. (118 ml) level.
- 4. Shake overnight, transfer the suspension to a 250 ml volumetric flask, add 5 drops of supperfloc, stopper and shake vigorously for 15 sec. make to volume & allow to settle 1 hr.
- 5. Run blank as the same procedure for analysis correction.
- 6. Calculate the Fe content of the sample as follows:
- % Free Fe oxide = 250 x (ppm Fe in sample solution-ppm Fe in blank solution)/sample weight (g) x 10,000

Remark: citted from Black et.al. (1965) and ASA - SSSA (1982)

APPENDIX E

Zinc.

The average Zn content of the lithosphere is 80 ppm while the Zn content of soil ranges from 10 to 300 ppm. Zinc is considered a trace element in soils. The solubilization of Zn minerals usually affects the availability to plants. The Zn hydrolysis species in aqueous solution are given by reactions.

$Zn^{2+} + H_2O \Longrightarrow ZnOH^+ + H^+$	logK° = -7.691.
$Zn^{2+} + 2H_2O \Longrightarrow Zn(OH)^{\circ} + 2H^{+}$	logK = -16.80 2.
$\operatorname{Zn}^{2+} + 3\operatorname{H}_2\operatorname{O} \Longrightarrow \operatorname{Zn}(\operatorname{OH})_3 + 3\operatorname{H}^+$	logK = -27.68 3.
$Zn^{2+} + 4H_2O \rightleftharpoons Zn(OH)^{2-} + 4H^+$	logK = -38.294.
	(Lindsay, 1979)

Zn²⁺ is the predominant species in solution below pH 7.7. Several investigators have reported that montmorillonite was capable of adsorbing Zn in amount greater than other clays. The mechanism of the adsorption is unknown, the specific adsorption could be replaced by nondestructive dilute acid extraction to release the Zn

Cadmium.

The average Cd content of the lithosphere is 0.2 ppm while the Cd content of soils from 0.01 to 0.70 ppm. The principal use of cadmium in industry is for metal plating. Cadmium is hazard pollutant because of its toxicity to nearly all living organism and its lack of useful biological function. The hydrolysis reaction of Cd²⁺ are given by reactions.

$$Cd^{2+} + H_2O \Longrightarrow CdOH^{+} + H^{+}$$
 log K = -10.10 ____5.
 $Cd^{2+} + 2H_2O \Longrightarrow Cd(OH)_2 + 2H^{+}$ log K = -20.30 ____6.

Source: Lindsay (1979)



Biography

Name : Krongkaew Jujun

Birth Date : July 25, 1960

Birth Place : Phitsanulok Province

Education : Bachelor Degree of Science (Chemistry)

Faculty of Science, Chiang Mai University.

(1981)

Working Experience : Teacher; under Vocational Department,

The Ministry of Education (1981-present).

Working Place : Sukothai Vocational College