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## APPENDIX

### Preparation of Concentrated Ammonium Molybdate Solution

Twenty five grams of  $(\text{NH}_4)_6\text{Mo}_7\text{O}_{24} \cdot 4\text{H}_2\text{O}$  are dissolved in 200 cm<sup>3</sup> of lukewarm distilled water, 290 cm<sup>3</sup> of concentrated sulphuric acid are diluted to 750 cm<sup>3</sup> with distilled water. After cooling, ammonium molybdate solution is added slowly with stirring to sulphuric acid solution and the volume is made to 1000cm<sup>3</sup> and then stored in a dark glass bottle in a refrigerator.

### Preparation of Dilute Ammonium Molybdate Solution

The concentrated ammonium molybdate solution is diluted by adding slowly 300 cm<sup>3</sup> of distilled water to 200 cm<sup>3</sup> of that reagent with stirring and stored in a dark glass bottle in a refrigerator. The dilute solution has to be freshly prepared every one or two months.

### Preparation of Concentrated Stannous Chloride Solution

Ten grams of stannous chloride are added to 25 cm<sup>3</sup> of concentrated hydrochloric acid solution and stored in a dark glass bottle in a refrigerator. The solution also has to be freshly prepared every one or two months.

### Preparation of Dilute Stannous Chloride Solution

Two cm<sup>3</sup> of the concentrated stannous chloride solution are added to 266 cm<sup>3</sup> of distilled water. The solution has to be prepared after 4 hours of storage.



## VITA



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