

CHAPTER III

Nuclear and Electromagnetic Interactions of Neutrons

III.1 General Remarks.

There are two different kinds of the interactions of high energy particles with matter. One is the electromagnetic interactions, the other is nuclear interactions. The examples of the electromagnetic interactions are the collisions between the particles with atomic electrons, the production of photons by charged particles and the ionization of photons. The examples of nuclear interactions are the collisions between neutrons and neutrons, neutrons and protons, and protons and protons. High energy nuclear interactions is sometimes called the spallation reaction.

In the electromagnetic interactions, we would consider a charged particle passes near an atom. According to Bohr(4), if the distance of closest approach is large, the interaction will involve the passing particle and the atom as a whole. In this case, an excitation or an ionization of the atom is considerably the most important. If the distance is closer to the order of the atomic dimension, the interaction is between the passing particle and one of the

atomic electrons. The phenomenon is called the knock-out process, when the distance is smaller than the atomic radius, the passing particle will be deflected by the atomic field, this phenomenon is called scattering.

11.2 Interaction of Gamma Ray Particles with Particles of the Medium.

a. Ionization The atom that loses one or more electrons is said to be ionized. The energy required to ionize the atom is called the ionization energy. The ionization may be caused by radiations or collisions. When gamma ray particles pass through a medium, its energy may be lost by ionization. According to Richtmyer(11), the ionization loss is given by the relation,

$$-\frac{dE}{dx} = BZ^2 Z(V)$$

where B is a known constant, Z is the charge of incident particle, Z(V) is a function of its velocity which varies as $\frac{1}{v^2}$.

b. Scattering When a charged particle passes near a nucleus it will change the direction of motion known as scattering. According to Rossi(4), Rutherford scattering formula is given

by the relation;

$$P(\theta)d\Omega = \frac{4\pi Z^2 r_0^2}{\Lambda} \left(\frac{m_0 c^2}{\beta \gamma} \right)^2 \frac{d\Omega}{\theta^4} \quad \text{where } r_0 \text{ is the radius of electron and } m_0 \text{ is the mass of electron.}$$

$P(\theta)d\Omega$ represent the probability of scattering of a particle of momentum p and velocity βc traversing a thickness of 1 gm cm^{-2} of the medium and undergoing a collision with the target nucleus which deflects the trajectory of the particle into the solid angle $d\Omega$ at an angle θ to the original motion. Z is the Avogadro's number. The probability is dependent of Z_0 , charge of the target nucleus. therefore the heavy nucleus has more probability of scattering than the light nucleus. The energy of the incident particle implies a related momentum transfer, leading to multiple scattering.

c. Energy Loss Electromagnetic: In the radiation loss of the incident particles. The emission of radiation is due to the acceleration or retardation in the atomic fields. Consider a particle of mass m passing near a nucleus with impact parameter b , according to Rossi(4), the radiation loss Q is related by;

$$Q = \frac{2}{3} \frac{h}{\Lambda} \left(\frac{v}{c} \right)^2 \frac{v^2}{c^2} \log \frac{b_{max}}{b_{min}} \quad \text{where } \frac{1}{3} \text{ is the fine-structure}$$

constant, v_0 is the electrostatic wave, the energy of the incident particle of mass $m = m_e$, will be lost more than that

of the heavier particles, so that ionization is to be expected in case of high energy electrons.

6. Nuclear collisions Cosmic ray particles may lose the energy by ionization and elastic collisions with the nuclei of the medium. If the incident particle approaches a nucleus, it may be absorbed by the nucleus to form spallation reaction. The spallation reaction is responsible for the formation of stars.



II.5 Cosmic Ray Stars.

Cosmic ray stars are seen as photographic emulsions and in cloud chambers. Stars have been observed in which several tracks meeting at a point resemble like a star in shape. The stars are related to nuclear disintegrations. The tracks of the stars, called the prongs, are caused by ionizing particles produced in the nuclear disintegrations. Neutral particles leave no trace in the nuclear emulsions. The nuclei which produce stars in the emulsions are probably those of silver or of bromine. The reactions that cause the formation of nuclear stars are sometimes called "Spallation Reaction".

II.4 Spallation Reactions

Spallation reaction is a type of nuclear reactions in

which several small particles are ejected from the nucleus. This reaction is being produced only with particles of high energy of order 100 MeV or more, incident upon a nucleus. The reaction is divided into two steps. The first step is the nuclear cascade collision of the nucleus inside the nucleus, which is the cause of direct emission from the nucleus of one or several particles. According to Rossi (4) and Fermi (5) this step is in the time of order of 10^{-20} sec after the incident particles have been absorbed. The second step is nuclear evaporation, which takes time of the order of 10^{-21} sec.

In such process, a high energy incident particle, which may be, say a 5 MeV proton, collides with the target nucleus produces a large star. If a neutron of 100 MeV hits a nucleus it will probably cause a small star in which two or four neutrons may be evaporated isotropically.

The star may be initiated from the high energy protons of about 100 MeV. For example, Balvin and Klüber (12) had obtained 4- prong stars produced in a cloud chamber by gamma ray beam from betatron operated at 100 MeV.

II.5 Nuclear Cascade Collisions and Nuclear Evaporation.

Spallation reaction is clearly explained by using Fermi gas model as shown by Fermi (13). The nucleus of a

nucleons in this model are contained in a well of constant diameter and length. The well depth V_0 is determined by the Fermi kinetic energy ϵ_F , and the average binding energy ϵ_B , i.e.,

$$V_0 = \epsilon_F - \epsilon_B$$

The charged particles are affected by the central nucleus V_C

$$V_C = \frac{Z_1 Z_2 e^2}{2r}$$

where Z_1 and Z_2 are number of electronic charge of the nucleus and the particle respectively. The nucleus radius is denoted by R .

A high energy particle can initiate within the nucleus a cascade collision. According to Wilson (20) particles with energy below about 1 eV have a very low probability of generating other particles, assuming that only elastic collisions are important. However, some of the nucleons will have increased in energy after relative few collisions. The energetic nucleons or particles usually cause gray tracks, i.e., those tracks with specific resolution between $1.5 \mu_{min}$ and $10 \mu_{min}$ and are mostly protons, neutrons and alphas of deuterons.

Some of the nucleons can transfer energy outside to other particles in the nucleus. The excitation energy of the nucleus at the end of the cascade, E_0^* can show by H. Nishijima,

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H. Divin and Stern (15), to be

$$E^{\circ} = E_1 = \sum_{i=0}^n E_i = (n-1)E$$

where E_1 is the energy of the fissioning particle (in the laboratory system), n is the number of outgoing particles, and E is the average kinetic energy of the n outgoing particles. The excitation energy content of the nucleus may be compared with the heat energy of a liquid droplet. The disintegration is analogous to a process of evaporation.

The excitation energy is considered to be heated to a nuclear temperature T in the walls of the nucleus. According to Wilson(14), the temperature is varied as $(A)^{2/3}$, so that

$$E^{\circ} = KA^{2/3}$$

where A is the total number of nucleons in the nucleus, K is a constant of nearly 0.1.

The nucleus and particles of atomic number 5 to 20, i.e., protons, neutrons, alpha particles and heavier nuclei are evaporated out of the nucleus. These nucleus and the particles usually cause black tracks of specific ionization greater than 10 G_{min} in the emulsions. The particles must have excitation energy more than their binding energies.

The probability of neutrons being emitted with energy between E and $E+dE$ was shown by Kulsrud (19), as shown

$$P(E)dE = \frac{2}{E^2} \exp\left(-\frac{E}{kT}\right) dE$$

where T is the nuclear temperature measured in eV or kT of MeV.

The probability of charged particles is,

$$P(E)dE = \frac{E - V_0}{E^2} \exp\left(-\frac{(E - V_0)}{kT}\right) dE$$

where V_0 is the Coulomb barrier.

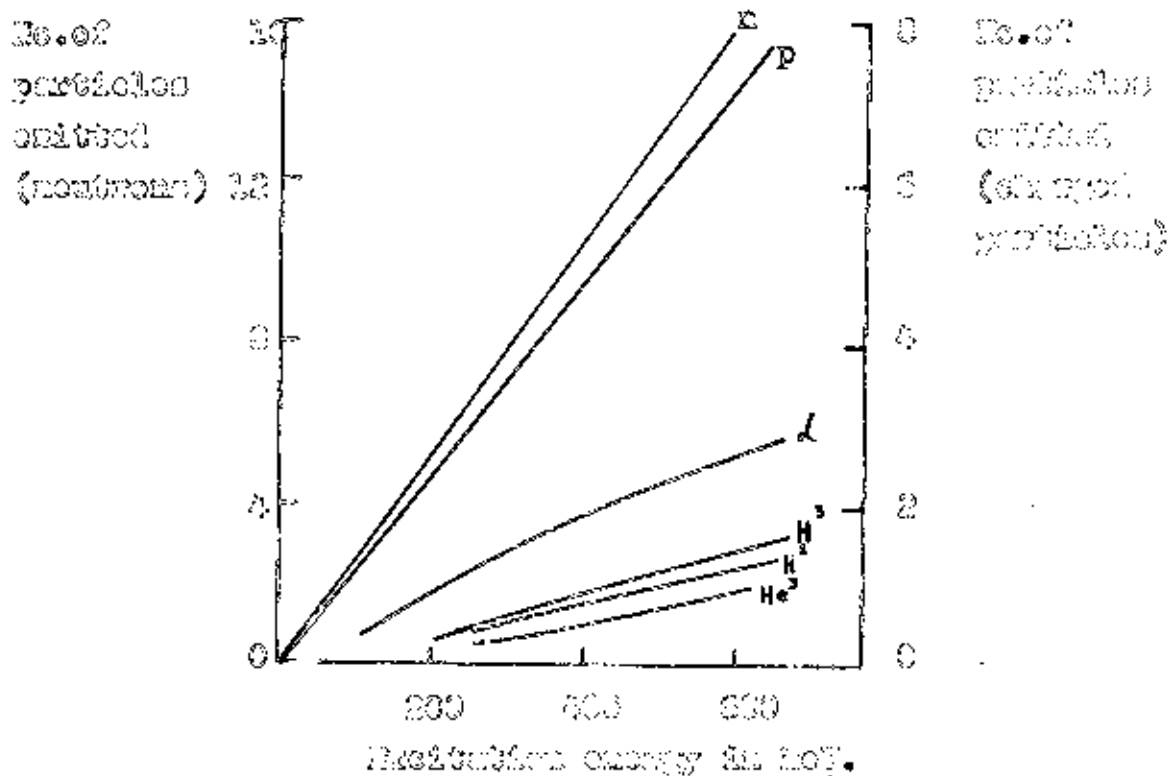


Fig. II . The number of evaporated particles as a function of initial excitation energy [after Bohr (19)]

II.6 Probability of Emission for Various Types of Compressed Particles.

The ratio of emission for various types of compressed particles as shown in Wilson(11) is as follows:

$$\frac{N(\alpha)}{N(p)} = 0.95 \pm 0.10$$

$$\frac{N(p)}{N(\alpha)}$$

$$\frac{N(d)}{N(\alpha)} = 0.90$$

$$\frac{N(t)}{N(\alpha)}$$

$$\frac{N(\pi^+ + \pi^-)}{N(\pi^+ + \pi^- + \pi^0)} = 0.81 \pm 0.10 \quad \text{for } n < 7 \text{ groups}$$

$$= 0.65 \pm 0.15 \quad \text{for } n > 7 \text{ groups}$$

where $\frac{N(\alpha)}{N(p)}$ denotes the ratio of emission of alpha particles,

$$\frac{N(p)}{N(\alpha)}$$

to protons, $\frac{N(d)}{N(\alpha)}$ denotes the ratio of emission of deuterium

$$\frac{N(t)}{N(\alpha)}$$

to tritium.

As low emission energy (< 10 eV) deuterium and tritium ions are emitted, particles are a small quantity. The frequency of the particles emitted at various energies is shown in Table II in Wilson(11) is as follows:

$$\text{Emission energy } \begin{matrix} 100 \text{ eV} & 10 \text{ eV} & 1 \text{ eV} & 0.1 \text{ eV} & 0.01 \text{ eV} \end{matrix}$$

$$\text{Frequency (groups)} \begin{matrix} 0.4 \times 10^3 & 0.3 \times 10^3 & 0.2 \times 10^3 & 0.1 \times 10^3 & 0.05 \times 10^3 \end{matrix}$$

The frequency with which these ions are emitted is estimated to be as follows:

	2	3	4	5	6	7	8	9	10
Number	1483	70429	1786	1036	624	624.3	624	624	624

where Z denotes the atomic number of the critical component.