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ลิขสิทธิ์ของจุฬาลงกรณ์มหาวิทยาลัย

SYNTHESIS AND *IN VITRO* DEGRADATION
OF POLY(L-LACTIDE-*BLOCK*-GLYCIDOL)

Miss Pattarapond Gonil

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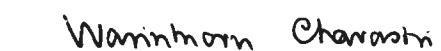
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บล็อกโคโพลิเมอร์ของเออล-แลกไทด์ (LLA) และ ไกลชิดอล (G) สังเคราะห์โดยปฏิกิริยา พอกลิเมโนไรเซ็นแบบเปิดวงของ LLA โดยใช้ทินออกโตเอด [Sn(Oct)₂] และพอลิไกลชิดอลแบบ กึ่งเป็นระบบบริเริ่ม น้ำหนักโมเลกุลของบล็อกโคโพลิเมอร์ขึ้นกับอุณหภูมิ, เวลาในการทำปฏิกิริยา, และอัตราส่วนเริ่มต้นของ LLA ต่อ G ผลจากจีพีซีโครามาติแกรมบ่งบอกว่ามีชนิดของพอลิเมอร์ที่ ต่างๆ กัน รวมทั้งไโอลิเออล-แลกไทด์ (PLLA) อาจเกิดขึ้นระหว่างการทำปฏิกิริยา เมื่อใช้ อัตราส่วนเริ่มต้นของ LLA:G เป็น 10:1 พบร่วมกับการลดลงของอุณหภูมิในเมทานอล ต่างกัน ส่วนที่ไม่ละลายในเมทานอลมีน้ำหนักโมเลกุลสูงกว่าส่วนที่ละลายในเมทานอล ภาวะที่ เหมาะสมของปฏิกิริยาการเปิดวงของ LLA ด้วย PG คือการทำปฏิกิริยาโดยปราศจากตัวทำ ละลายที่ 130 องศาเซลเซียสเป็นเวลา 24 ชั่วโมง ปริมาณ Sn(Oct)₂ ที่ใช้คือ 5 เปอร์เซ็นต์โดยโม ลของปริมาณหมุ่ยครอกซิลในพอลิไกลชิดอล อุณหภูมิเปลี่ยนสถานะคล้ายแก้ว (T_g) ของบล็อก โคโพลิเมอร์มีค่าอยู่ระหว่าง T_g ของ PG และ PLLA และเพิ่มขึ้นตามการเพิ่มน้ำหนัก โมเลกุลของโคโพลิเมอร์ บล็อกโคโพลิเมอร์มีความชอบน้ำมากกว่า PLLA และความชอบน้ำมาก ขึ้นเมื่อมีปริมาณ LLA ในบล็อกโคโพลิเมอร์ลดลงซึ่งเป็นผลจากการวัดค่ามุ่งสัมผัสของอากาศ กับน้ำ อัตราการสลายตัวของบล็อกโคโพลิเมอร์ในฟอสเฟตบัฟเฟอร์ชาลีนที่ pH 7.4 มากกว่าไโอลิเมอร์ PLLA ที่มีน้ำหนักโมเลกุลใกล้เคียงกัน แต่การสูญเสียน้ำหนักของโคโพลิเมอร์มากกว่าไโอลิเมอร์ PLLA ซึ่งอาจเป็นเพราะบล็อกของ PG ที่ละลายน้ำได้นั้นสลายตัวละลายออกมาก

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PATTARAPOND GONIL: SYNTHESIS AND *IN VITRO* DEGRADATION OF POLY(L-LACTIDE-BLOCK-GLYCIDOL). THESIS ADVISOR: ASSISTANT PROFESSOR VARAWUT TANGPASUTHADOL, Ph.D.; 110 pp. ISBN: 974-17-2240-7

Block copolymers of L-lactide (LLA) and glycidol (G) were synthesized by ring-opening polymerization of LLA using tin octoate $[Sn(Oct)_2]$ and branched polyglycidol (PG) as an initiation system. The molecular weight of the block copolymers depend on the reaction temperature, reaction time, and monomer feed ratio (LLA:G). Results from GPC chromatograms suggested that different polymeric species including homo PLLA might occur during the polymerization. At high LLA:G feed ratio (10:1) two types of polymeric products having different solubilities in methanol were obtained. It was found that the methanol-insoluble portion having higher molecular weight than the methanol-soluble portion. The suitable polymerization condition was carried out without solvent at 130°C for 24 hours. The amount of $Sn(Oct)_2$ was 5%mol of the total hydroxyl group content in PG. The glass transition temperatures (T_g) of the block copolymers were found to be between those of PG and PLLA and increased with increasing the copolymer molecular weight. The block copolymer was more hydrophilic than homo PLLA and increased with decreasing LLA content as examined by air-water contact angle measurement. The obtained block copolymer degraded in phosphate buffer saline at pH 7.4 at a slower rate than the homo PLLA having the same molecular weight. The weight loss of the copolymer was however slightly more than that of homo PLLA, probably due to dissolution of the cleaved water-soluble PG block.

Field of study Petrochemistry and Polymer Science Student's signature Pattarapond Gonil

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LIST OF ABBREVIATIONS

ROP	: Ring-opening polymerization
CROP	: Cationic ring-opening polymerization
AROP	: Anionic ring-opening polymerization
ACE	: Active chain end
AM	: Activated monomer
FAD	: American Food and Drug Administration
LLA	: L-lactide, L,L-lactide
DLA	: D-lactide, D,D-lactide
PLLA	: Poly(L-lactide), poly(L-lactic acid)
PGL	: Polyglycolide
ϵ -CL	: ϵ -caprolactam
G	: Glycidol
PG	: Polyglycidol
GL	: Glycerol
SB	: Sorbitol
KO ^t Bu	: Potassium <i>tert</i> -butoxide
Mg(OEt) ₂	: Magnesium diethoxide
Al(O ⁱ Pr) ₃	: Aluminium triisopropoxide

SnPh ₄	: Tetraphenyl tin
Sn(Oct) ₂	: Tin(II) 2-ethylhexanoate, Stannous Octoate
NMR	: Nuclear magnetic resonance
GPC	: Gel permeation chromatography
MALDI-TOF	: Matrix assisted laser desorption ionization time of flight
MS	: Mass spectroscopy
\overline{M}_n	: Number-average molecular weight
\overline{M}_w	: Weight-average molecular weight
MWD	: Molecular weight distribution
PDI	: Polydispersity index