

## REFERENCES

- [1] Beer, P. D.; Gale, P.A.; Smith, D. K., *Supramolecular Chemistry*, Oxford university press, 1999.
- [2] Steed, J. W.; Atwood, J. L., *Supramolecular Chemistry*, John Wiley & Son, Ltd, New York, USA, 2001.
- [3] Gale, P. A.; Quesada, R., Anion Coordination and Anion-templated Assembly: Highlights from 2002 to 2004., *Coord. Chem. Rev.*, 250 (2006): 3219-3244.
- [4] Gunnlaugsson, T.; Glynn, M.; Tocci, G. M., Kruger, P. E.; Pfeffer, F. M., Anion Recognition and Sensing in Organic and Aqueous Media using Luminescent and Colorimetric Sensors, *Coord. Chem. Rev.*, 250 (2006) 3049-3117.
- [5] Fabbrizzi, L.; Poggi, A., Sensors and Switches from Supramolecular Chemistry, *Chem. Soc. Rev.*, 197 (1995): 197-201.
- [6] Beer, P.D.; Gale, P.A., Anion Recognition and Sensing: The state of the Art and Future Perspectives, *Angew. Chem., Int. Ed.*, 40 (2001): 486-516.
- [7] Kim, S. Y.; Hong, J.-I., Chromogenic and Fluorescent Chemodimeter for Detection of Fluoride in Aqueous Solution, *Org. Lett.*, 9 (2007): 3109-3112.
- [8] Lee, D. H.; Lee, K., H.; Hong, J.-I., An Azophenol-Based Chromogenic Anion Sensor, *Org. Lett.*, 3 (2001): 5-8.
- [9] Lohr, H., -G.; Vogtle, F., Chromo- and Fluoroionophores. A new Class of Dye Reagents, *Acc. Chem. Res.*, 18 (1985): 65-72.
- [10] Kolly, T., R.; Kim, M., H., *Relative Binding Affinity of Carboxylate and Its Isoster: Nitro, Phosphate, Sulfonate and delta-lactone*, *J. Am. Chem. Soc.*, 116 (1994): 7072-7080.
- [11] Lin, Z. -H.; Zhao, Y. G.; Duan, C.-Y., Zhang, B.-G.; Bai, Z. -P., A Highly Selective Chromo- and Fluorogenic Dual Responding Fluoride Sensor: Naked-Eye Detection of F<sup>-</sup> Ion in Natural Water via A Test Paper, *Dalton Trans.* (2006): 3678-3684.
- [12] Rov., -L. J. V.; Martinez, -M. R.; Benito, A.; Soto, J., Naphthoquinone Derivatives as Receptors for The Chromogenic Sensing of Metal Cation and Anions, *Polyhedron*, 25 (2006): 1585-1591.
- [13] Helga, M.; Nikol, N.; Vogler, A., Photochemistry of Barium(II) and Lead (II) 2,4-Dinitrostilbene-4'-Monoaza-18-Crown-6-Complexes, *J. Photochem. Photo biol. A: Chem.*, 84 (1998): 227-231.

- [14] Liang, Z.; Liu, Z.; Gao, Y., A Selective Colorimetric Based on Calixarene Framework for Lanthanide Ion-Dy<sup>3+</sup> and Er<sup>3+</sup>, *Tetrahedron Lett.*, 48 (2007): 3587-3591.
- [15]. Liu, H.; Li, B.; Liu, D.; Xu, Z., Photo-Physical Properties of The Transition Metal Ions Complexes of The Calix[4]arene Derivatives, *Chem. Phys. Lett.*, 350 (2001): 441-446.
- [16] Goodnew, T., T.; Reddigton, M. V.; Stoddart, J. F.; Kaifer, A. E. , Cyclobis(paraquat-p-phenylene): A Novel Synthetic Receptor for Amino Acids with Electron-rich Aromatic Moiety, *J. Am. Chem. Soc.*, 113 (1991): 4335-4337.
- [17] Rebek, J., Jr; Askew, B.; Nemeth, D.; Paris, K., Convergent Function Groups. 4. Recognition and Transport of Amino Acids Across A Liquid Membrane, *J. Am. Chem. Soc.*, 109 (1987): 2432-2434.
- [18] Peacoak, S. C.; Walba, D. M.; Gaota, F. C. A.; Helqesson, R. C.; Cram, D. C., Host-guest complexation. 22. Reciprocal Chiral Recognition between Amino Acids and Dilocular Systems, *J. Am. Chem. Soc.*, 102 (1980): 2043-2052.
- [19] Karoda, Y.; Kato, Y.; Higashioji, T.; Ogashi, H., New Chiral Porphyrins. Syntheses and Molecular Recognition of Amino Acids Ester, *Angew. Chem. Int. Ed. Engl.*, 32 (1993): 723-724.
- [20] Behr, J.-P.; Lehn, J. -M., Transport in Organic Chemistry. I. Transport of Amino Acids through Organic Liquid Membrane, *J. Am. Chem. Soc.*, 95 (1973): 6108-6110.
- [21] Tabushi, I.; Kuroda, Y.; Mizutani, T., Artificial Receptors for Amino Acids in Water Local Environmental Effect on Polar Recognition by 6A-Amino-6B-Caboxy- and 6B-Amino-6a-Carboxy-Beta-Cyclodextrins, *J. Am. Chem.. Soc.*, 108 (1986): 4514-4518.
- [22] Crego, M.; Partearroyo, A.; Raposo, C.; Musson, M. L.; Lopez, J. L.; Alcazar, V.; Moran, J., Chiral Receptor for N-benzyloxycarbonyl Amino Acid Derivatives, *Tetrahedron Lett.*, 35 (1994): 1435-1438.
- [23] Ballistreri, F. P.; Notti, A., Pappalarolo, S.; Paris, M. F. ; Disagatti, L., Multipoint Molecular Recognition of Amino Acids and Biogenic Amines by Ureido calix[5]arene Receptor, *Org. Lett.*, 5 (2003): 1071-1074.

- [24] Moulinoux, J.-P.; Quemener, V.; Delcros, J. -G.; Cipolla, B., *In Polyamines in cancer: Basic Mechanisms and Clinical Approaches*, Nishioka, K., Ed., Springer: New York, 1996, Chapter 10, pp 233.
- [25] Debroy, P.; Banerjee, M.; Prasad, M., Oulik, S.P.; Roy, S., Binding of Amino Acids into a Novel Multiresponsive Ferrocene Receptor Having an Ene Backbone, *Org. Lett.*, 7 (2005): 403-406.
- [26] Gupta, N.; Linschitz, H., Hydrogen-bonding and Protonation Effects in Electrochemistry of Quinones in Aprotic Solvent, *J. Am. Chem. Soc.*, 119 (1997): 6384-6391.
- [27] Chamber, J. Q., *In The Chemistry of the Quinone Compound*; Patai, S.,; Rappoport, Z.,; Eds; Wiley: New York, 1998, Vol II, Chapter 12, pp 719.
- [28] Klothoff, I. M.; Lingane, J. I., *Polarography*, 2<sup>nd</sup> ed., Interscience: New York, 1952, Vol. 1, Chapter XIIV and XV.
- [29] Swallon, A. J., *In Function of Quinone in Energy Conserving System*; Trumpower, B. L., Ed.; Academic Press: New York; 1982; Chapter 3, pp 66.
- [30] Carneiro, P. A., Boralle, N., Stradiotto, N., R., Furlan, M. And Zanoni, M., V., B., Decolourization of Anthraquinone Reaction Dye by Electrochemical Reduction on Reticulated Glassy Carbon Electrode, *J. Braz. Chem. Soc.*, 15 (2004): 587-92.
- [31] Brooks, S., J., Birkin, P., R.; Gale, P. A., Electrochemical measurement of Switchable Hydrogen Binding in An Anthraquinone Based Anion receptor, *Electro. Chem.*, 7 (2005): 1351-1356.
- [32] [Http: // www. osmatech.plc.uk](http://www.osmatech.plc.uk). (17 /05 /2007)
- [33] Uchiyama, S.; McClean, G. D.; Iwai, K.; de Silva, A. P. , Membrane Media Create Small Nanospaces for Molecular Computation, *J. Am. Chem. Soc.*, 127 (2005): 8920-8921.
- [34]. de Silva, A. P., Signaling Recognition Events with Fluorescent Sensor and Switches, *Chem. Rev.*, 57 (1997): 1515-1566.
- [35] Balzani V.; Venturi M.; Credi, A., *Molecular Devices and Machines*, Wiley-VCH, Weinheim, 2003.
- [36] de Silva, A., P.; Uchiyama, S.; Vance, T., P.; Wannalorse, B., A Supramolecular Chemistry Basis for Molecular Logic and Computation, *Coord. Chem. Rev.*, 251 (2007): 1023-1032.

- [37] Magri, D. C.; Vance, T. P.; de Silva, A. P., From complexation to computation: Recent progress in molecular, *Inorg. Chim. Acta.*, 360 (2007): 751-764.
- [38] Bianchi, A.; Bowman, -J., K.; Garcia, -E., E., *In supramolecular Chemistry of Anions*, Wiley-VCH, New York, 1997.
- [39] Dugas, H., *Bioorganic Chemistry*, 3<sup>rd</sup> ed. Spinger, New York, 1996.
- [40] Hamilton, A.D., *Bioorganic Chemistry Frontier*, Dugus, H. Ed; Springer; New York, 1992, Vol 2.
- [41] Sambrook, M. R.; Beer, P. D.; Wisner, J. A.; Paul, R. L., Cowley, A. R., Szemes, F.; Drew, M. G. B., Anion-templated Assembly of Pseudorotaxanes: Importance of Anion Template, Strength of Ion-Pair Thread Association and Macrocyclic Ring Size, *J. Am. Chem. Soc.*, 127 (2005): 2293-2303.
- [42] Khatri, V. K.; Upreti, S.; Pandey, P. S., Novel Bile Acids-Based Cyclic Bisimidazolium Receptors for Anion Recognition, *Org. Lett.*, 8 (2006): 1755-1758.
- [43] Yun, S.; Ihm, H.; kim, H. G.; Lee, C. W.; Indrajit, B.; Oh, K. S.; Gony, Y. J.; Lee, J. W.; Yoon, J.; Lee, H. C.; Kim, K., S., Molecular Recognition of Fluoride Anion; Benzene-Based Tripodal Imidazolium Receptor, *J. Org. Chem.*, 68 (2003): 2467-2470.
- [44] Gutman, V., *The Donor-Acceptor Approach to Molecular Interaction Phenomena*, New York, 1978.
- [45] Singh, N. J.; Jun, J., J.; Chellappan, K.; Thangadurai, D.; Chandran, R. P.; Hwang, I. -C., Yoon, J.; Kim, K. S., Quinoxaline-Imidazolium Receptors for Unique sensing of Pyrophosphate and Acetate by Charge Transfer, *Org. Lett.*, 9 (2007): 485-489.
- [46] Choa, T. -C.; Liao, K. -C.; Lin, J. -J., Synthesis and Luminescence Properties of U-Shaped Polycyclic Molecules Containing Syn-Facial Functionalized Quinoxaline Rings, *Org. Lett.*, 7 (2005): 4843-4847.
- [47] Ihm, H.; Yun, S.; Kim, H., G.; Kim, K. S.; Tripodal Nitro-Imidazolium Receptor for Anion Binding Driven by (C-H)<sup>+</sup>...X<sup>-</sup> Hydrogen Bonds, *Org. Lett.*, 4 (2002): 2897-2900.
- [48] Yoon, J.; Kim, S. K.; Sigh, N. J.; Lee, J. W. ; Yang, Y. J.; Chellappon, K.; Kim, K. S., Highly Effective Fluorescent Sensor for H<sub>2</sub>PO<sub>4</sub><sup>-</sup>, *J. Org. Chem.*, 69 (2004): 581-585.



- [49] Katritzky, A. R.; Jiang, J.; Greenhill, J., 1,2- and 1,3-Monoazabissylides as Novel Synthons, *J. Org. Chem.*, 58 (1993): 1987-1988.
- [50] Katritzky, A. R.; Jiang, J.; Urogdi, L., N-triphenylphosphorylidene-1-(benzotriazol-1-yl) methylamine, A Novel Synthon Equivalent to  $^+\text{CH}_2\text{NH}_2$ : The preparation of Primary Amines, *Tetrahedron Lett.*, 30 (1993): 3303-3306.
- [51] Katritzky, A. R.; Lan, X.; Yang, J. Z.; Denisko, O. V., Properties and Synthetic Utility of N-Substituted Benzotriazoles, *Chem. Rev.*, 98 (1998): 409.
- [52] Herrmann, W. A.; Kocher, C., N-Heterocyclic carbenes, *Angew. Chem. Int. Ed. Engl.*, 36 (1997): 2162-2187.
- [53] Pfeffer, F. M.; Gunlaugsson, T.; Jensen, P.; Kruger, P. E., Anion Recognition Using Preorganized Thiourea Functionalized [3] Polynorbornane Receptors, *Org. Lett.*, 7 (2005): 5357-5360.
- [54] Vazquez, M.; Fabbrizzi, L.; Taglietti, A.; Pedrido, R. M.; Gonzalez-Noya, A. M.; Bermejo, M. R., A Colorimetric Approach to Anion Sensing: A Selective Chemosensor of Fluoride Ions, in Which Color is Generated by Anion-Enhanced  $\pi$  Delocalization, *Angew. Chem. Int. Ed.*, 43 (2004): 1962-1965.
- [55] Miyuji, H.; Sessler, J. L., OFF-the-Self Colorimetric Anion Sensor, *Angew. Chem. Int. Ed. Engl.*, 40 (2001): 154-157.
- [56] Esteban-Gomez, D.; Fabbrizzi, L.; Licchelli, M., Why, on Interaction of Urea-Based Receptors with Fluoride, Beautiful Colors Develop, *J. Org. Chem.*, 70 (2005): 5717-5720.
- [57] Nielsen, K. A.; Cho, W. -S; Jeppesen, J. O.; Lynch, V. M.; Becher, J.; Sessler, J. L., Tetra-TTF Calix[4]pyrrole ; A rationally Designed Receptor for Electron-Deficient Neural Guests, *J. Am. Chem. Soc.*, 126 (2004): 16296-16297.
- [58] Martinez, M. R.; Sancenon, F., Fluorogenic and Chromogenic Chemosensors and Reagents for Anions, *Chem. Rev.*, 103 (2003): 4419-4476.
- [59] Jose, D. A.; Kumar, K. D.; Ganguly, B.; Das, A., Efficient and Simple Colorimetric Fluoride Ion Sensor Based Receptors Having Urea and Thiourea Binding Sites, *Org. Lett.*, 6 (2004): 3445-3448.
- [60] Gomez, M.; Gonzalez, F. J.; Gonzalez, I., Intra and Intermolecular Hydrogen Bonding Effect in The Electrochemical Reduction of  $\alpha$ -Phenolic-Naphthaquinone, *J. Electrochem. Chem.*, 578 (2005): 193-202.

- [61] Sedden, J. M.; Templer, R. H., *Polymorphism of Lipid-Water System, The Handbook of Biological Physics*, vol. 1, ed. R. Lipowsky, and E. Sackmann., 1995, Elsevier Science B.V.
- [62] Baeurle, S. A.; kroener, J., Modeling Effective Interactions of Micellar Aggregates of Ionic Surfactants with The Gause-Core Potential, *J. Math. Chem.*, 36 (2004): 409-421.
- [63] Hartley, G. S., *Aqueous Solution of Paraffin Chain salts, A Study in Micelle Formation*, Hermamm et Cie, Paris, 1936.
- [64] Mancin, F.; Rampazzo, E.; Tecilla, P.; Tonellato, U., Self-assembled Fluorescent Chromosensor, *Chem. Eur.,J.*, 12 (2006): 1844-1850.
- [65] Kleiner, K., *Molecular Computers Act as Tiny ID Tags*, New Scientist, 2006.
- [66] Aviram, A., Molecular for Memory Logic and Amplification, *J. Am. Chem. Soc.*, 110 (1988): 5687-5688.
- [67] Lam, K. S., A New Type of Synthetic Peptide Library for Identifying Ligand-Binding Activity , *Nature*, 354 (1991): 82-84.
- [68] Brummel, C. L.; Lee, I. N. W.; Zhou, Y.; Benkovic, S.J.; Winograd, N. A., Mass Spectromicro Solution to The Address Problem of Combinatorial Libraries., *Science*, 264 (2004): 399-402.
- [69] Ohlmeyer, M. H. J., Complex Synthetic Chemical Libraries Index with Molecular Tags, *Proc. Natl. Acad. Sci. USA.*, 90 (1993): 10922-10926.
- [70] Nicolaon, K. C.; Xiao, X. Y.; Paradoosh, Z.; Denyei, A.; Nova, M., Radiofrequency Encoded Combinatorial Chemistry, *Angew. Chem. Int. Ed.*, 34 (1995): 2289-2291.
- [71] Grandini, P.; Fabrizio, M.; Tecilla, P.; Scrimin, P.; Tonellato, U., Exploiting the Self-assembly Strategy for Selective Cu(II) Ion Chemosensor, *Angew. Chem. Int. Ed.* 38 (1999): 3061-3063.
- [72] Canary, J. W.; Gibb, B. C., For A Different Strategy to Self-Assembling System, *Prog. Inorg. Chem.*, 45 (1997): 1-5.
- [73] Nakahara, Y.; Kida, T.; Nakatsuji, Y.; Akashi, M., Fluoreometric Sensing of Alkaline Metal and Alkaline Earth Metal Cations by Novel Photosensitive Monoazacryptand Derivatives in Aqueous Micellar Solution, *Org. Biomol. Chem.*, 3 (2005): 1787-1794.

- [74] Pallavicini, P.; Diaz-fernandez, Y. a.; Foti, F.; Mangano, C.; Patroni, S., Fluorescent Sensor for  $Hg^{2+}$  in Micelles; A New Approach that Transforms An ON-OFF into An OFF-ON Response as A Function of The Lipophilicity of The Receptor, *Chem. Eur. J.*, 13 (2007): 178-187.
- [75] Wang, P. ; Onozawa,-K. N.; Himeda, Y.; Sugihara, H.; Arakawa, H.; Kasuga, K., 3-(2-Pyridyl)-2-pyrazoline Derivatives: Novel Fluorescent Probe for  $Zn^{2+}$  Ion, *Tetrahehron Lett.*, 42 (2001): 9199-9201.
- [76] de Silva, A. P.; Dioxan. I. M.; Ganarate, N. H. Q.; Gunlaugsson, T.; Maxwell, P. R. S.; Rice, T. E., Integration of Logic Function and Sequential Operation of Gates at The Molecular-Scale, *J. Am. Chem. Soc.*, 6 (1993): 1393-1394.
- [77] Diaz-Fernandez, Y.; Foti, F.; Mangano, C.; Pallavicini, P.; patroni, S.; Perez-Gramatges, A.; Rodriguez-Calvo, S., Micells for The Self-Assembly of "OFF-ON-OFF" Fluorescent Sensor for pH Windows, *Chem. Eur. J.*, 12 (2006): 921-930.
- [78] Gomez, M.; Gomez,-C. C. Z.; padilla, -M. I. I.; Martinez, -M. F. J.; Gonzalez, F. J. Hydrogen Bonding Effects on the Association Process between Chloranil and a Series of Amides, *J. Electro. Chem.*, 567 (2004): 269-276.
- [79] Ballardini, R.; Varani, G.; Indell, M. T.; Scandola, F.; Balzani, V., Free Energy Correlation of Rate Constants for Electron Transfer Quenching of Excited Transition Metal Complexes, *J. Am. Chem. Soc.*, 100 (1978): 7219-7223.
- [80] Grigg, R., Nerbert, W. D. J. A., Luminescent pH Sensors Based on Di (2,2'-bipyridyl)(5,5'-diaminomethyl-2,2' bipyridyl)-ruthenium (II) complexes, *J. Chem. Soc., Chem. Commun.*, (1992): 1300-1302.
- [81] Onoda, M.; Uchiyama, S.; Santa, T.; Imai, K., The Effects of Spacer length on the Fluorescence Quantum yields of the Benzofuran Compounds Bearing a Donor-Acceptor System, *Luminescence*, 17 (2002): 11-14.
- [82] Sumarn, K.; Matsuoka, H.; Yamaoka, H., Evaluation of the counterion Distribution around Spherical Micelles in solution by Small-Angle Neutron Scattering, *Phy. Rev.*, 53 (1996): 1744-1752.
- [83] Fernandez, M. S.; fromherz, P., Lipid pH Indicators as probes of Electrochemical Potential and Polarity in micelles, *J. Phys. Chem.*, 81 (1977): 1755-1761.

- [84] Bissell, R. A.; de Silva, A. P.; Gunaratne, H. Q. N.; Macoy, C. P., fluorescent PET(Photoinduced electron Transfer) Sensors with Targeting Anchoring Modules as Molecular Version of Submarine Periscopes for Mapping Membrane-bounded Photons, *J. Chem. Soc., Chem. Commun.*, (1994): 405-408.
- [85] R. C. Weast(Ed.), *CRC Handbook of Chemistry and Physics*, 56<sup>th</sup> edn., CRC Press, Cleveland, OH, 1975.
- [86] Rajagopal, S.; Gnanaraj, G. A.; Srinivasan, A. M. C., Excited State Electron Transfer Reaction of Tris(4,4'-dialkyl-2,2'-bipyridine)ruthenium (II) Complexes with Phenolate Ions; Structural and Solvent Effects, *J. Photochem. Photobiol. A. Chem.*, 69 (1992): 83-89.
- [87] Miedlar, K.; Das, P. K., Tris(4,4'-dialkyl-2,2'-bipyridine)ruthenium (II)-Sensitized Photooxidation of Phenols. Environmental effects on Effects on Electron Transfer Yields and Kinetics, *J. Am. Chem. Soc.*, 104 (1982): 7462-7469.
- [88] Grigg, R.; Holmes, J. M.; Jones, S. K.; Norbert, W. D. J. A., Luminescent pH Sensor Based on p-Tert-butylcalix[4]arene-Linked Ruthenium(II) Trisbipyridyl Complexes, *J. Chem. Soc., Chem. Commun.*, (1994): 185-187.
- [89] Galster, H., *pH Measurement*, VCH, Weinheim, 1991.
- [90] Kalyanasundaram, K., Photoredox Redction in Micellar Solutions Sensitized by Surfactant Derivative of Tris(2,2'-bipyridine)ruthenium (II), *J. Chem. Soc., Chem. Commun.*, (1976): 628-632.
- [91] Rodgers, M. A. J.; Becher, J. C., Electron-Transfer Quenching of the Luminescent State of the Tris(bipyridine)ruthenium (II) complex in micellar media, *J. Phys. Chem.*, 84 (1980): 2762-2768.
- [92] Perin, D. D., *Dissociation Constants of Organic Bases in Aqueous Solution*, Butterworths, London, 1972.
- [93] de Silva, A. P.; Gunaratne, H. Q. N.; Maquire, G., 'OFF-ON Fluorescent Sesors for Physiological Levels of Magnesium Ions Based on Photoinduced Electron Transfer (PET) , Which also Behave as Photoionic OR Logic Gates, *J. Chem. Soc., Chem. Commun.*, (1994): 1213-1214.
- [94] de Silva, A. P.; Gunaratne, H. Q. N.; MaCoy, C. P., A Molecular Photoionic AND Gated Base on Fluorescent Signalling, *Nature*, 364 (1993): 42-44.



- [95] Bissell, R. A.; de Silva, A. P.; Thilak, W.; Fernando, M. L.; Patuwathavithana, S. T.; Shantha, T. K.; Samarasinghe, D., Fluorescent PET (Photoinduced Electron Transfer) Indicators for Solvent Polarity with Quasi-Step Functional Response, *Tetrahedron Lett.*, 32 (1991): 425-428.
- [96] de Silva, A. P.; Gunaratne, H. Q. N.; Jayasekera, K. R.; Challaghan, S. O.; Sandanayake, K. R. A. S., Temperature dependent Fluorescence of Tunable Fluorophore-Fissile Bond Systems Based on 1-Phenyl, 3-Aryl  $\Delta^2$ -Pyrazolines as a Means of Quantitating Photofission Processes, *Chem. Lett.*, (1995): 123-124.
- [97] de Silva, A. P.; James, M. R.; McKinney, B.O.F; Pears, D.A.; Weir, S.M., Molecular Computational Elements Encode Large Populations of Small Objects, *Nature Mater.*, 5 (2006): 787-788.
- [98] de Silva, A. P.; de Silva, S. S. K.; Goonesekera, N. C. W.; Gunaratne, H. Q. N.; Lynch, P. L. M.; Nesbitt, K. R.; Patuwathavithana, S. T.; Ramyalal, N. L. D., Analog Parallel Processing of Molecular Sensor Information, *J. Am. Chem. Soc.*, 129 (2007): 3050-3051.
- [99] He, H.; Mortellaro, M. A.; Leiner, M. J. P.; Young, S. Y.; Fraatz, R. J.; Tusa, J. K., A Fluorescent Electron Transfer, *Anal. Chem.*, 75 (2003): 549-555.
- [100] Tusa, J. K.; He, H., Critical Care Analyzer with Fluorescent Optical Chemmosensor for Blood Analytes, *J. Mater. Chem.*, 15 (2005): 2640-2647.

**APPENDICES**

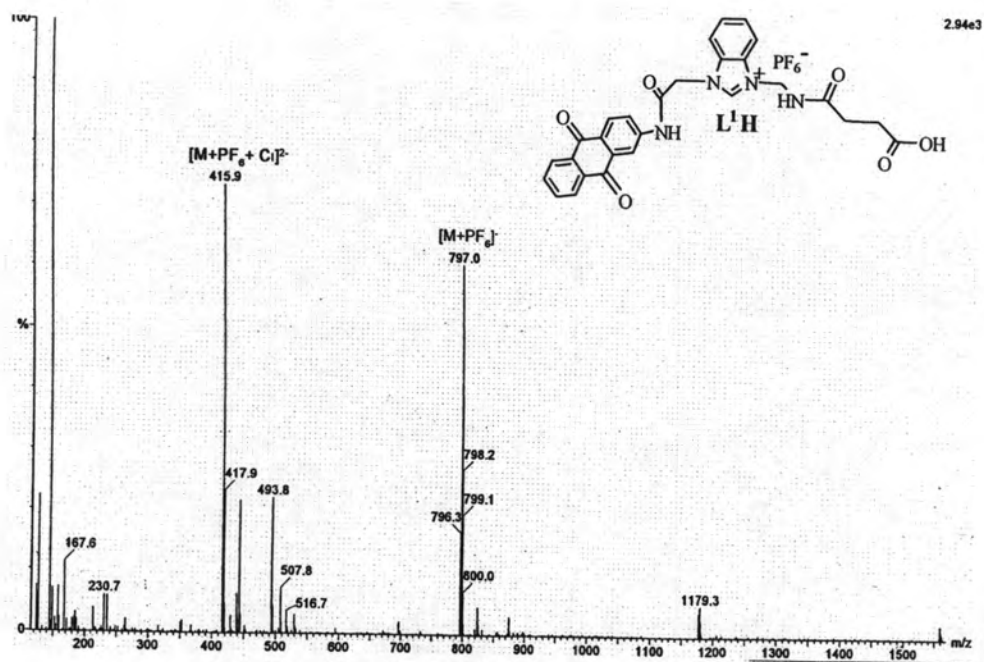


Figure A1 ESI MS (ES<sup>-</sup>) result of L<sup>1</sup>H (M= L<sup>1</sup>H)

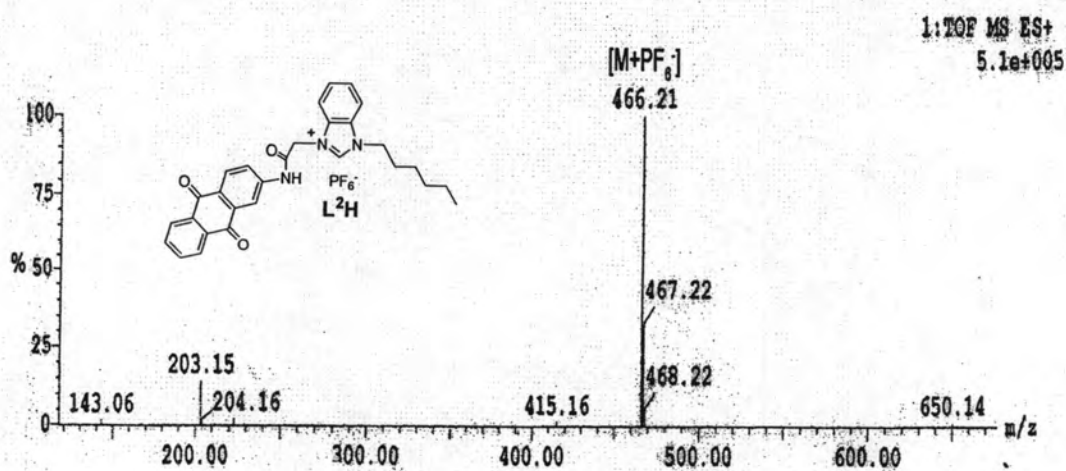


Figure A2 ESI MS (ES<sup>+</sup>) result of L<sup>2</sup>H (M= L<sup>2</sup>H)

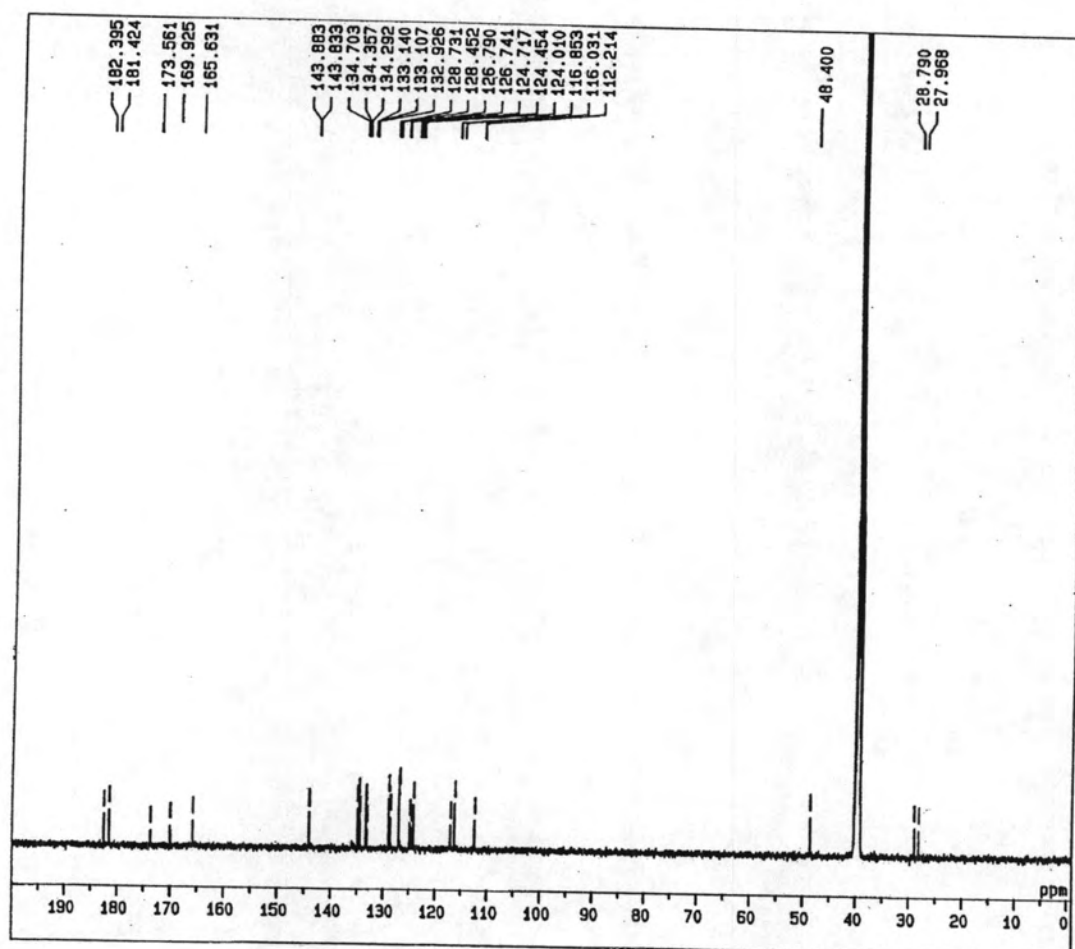


Figure A3  $^{13}\text{C}$  NMR spectrum of  $\text{L}^1\text{H}$  in  $\text{DMSO-}d_6$ .



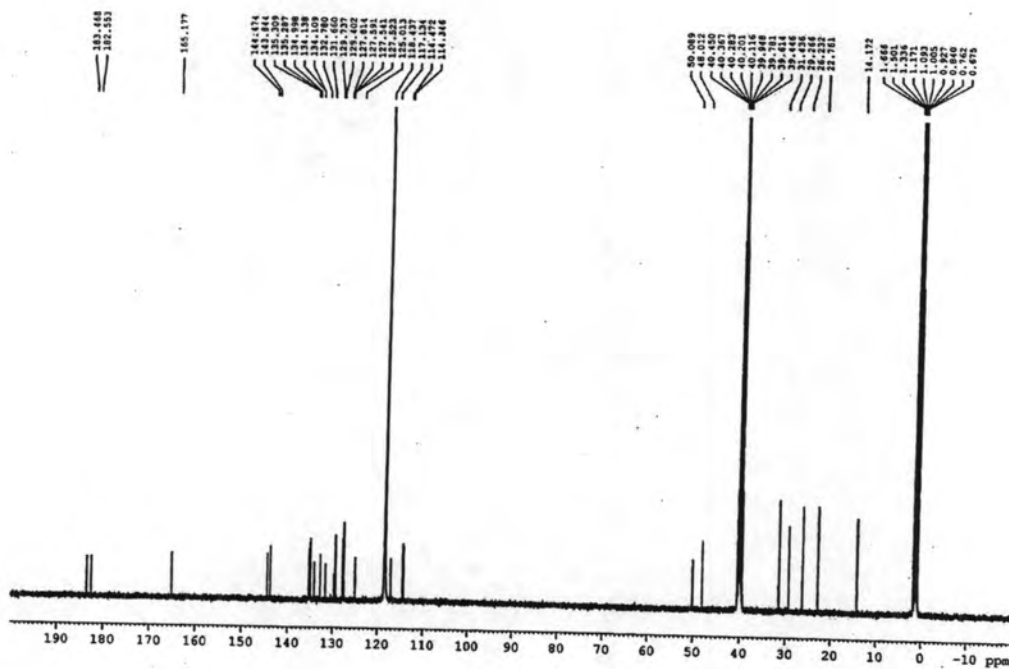


Figure A4  $^{13}\text{C}$  NMR spectrum of  $\text{L}^2\text{H}$  in  $\text{DMSO}-d_6$

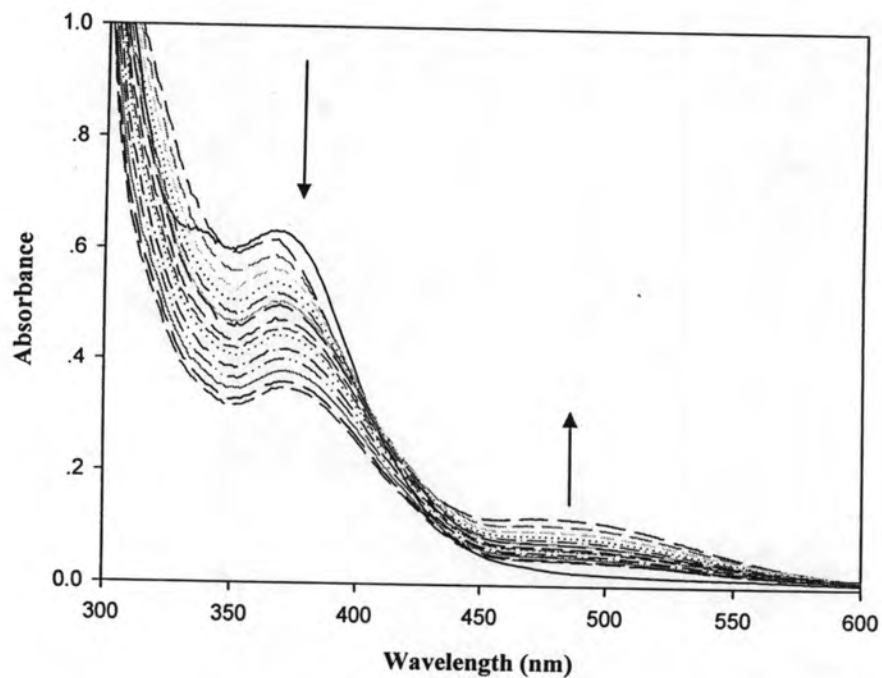


Figure A5 UV titration of  $\text{L}^2\text{H}$  with  $\text{Bu}_4\text{NCH}_3\text{COO}$  in  $\text{DMSO}$ .  $[\text{L}^2\text{H}] = 7.5 \times 10^{-4} \text{ M}$  and  $[\text{Bu}_4\text{NCH}_3\text{COO}] = 0-0.02 \text{ M}$ .

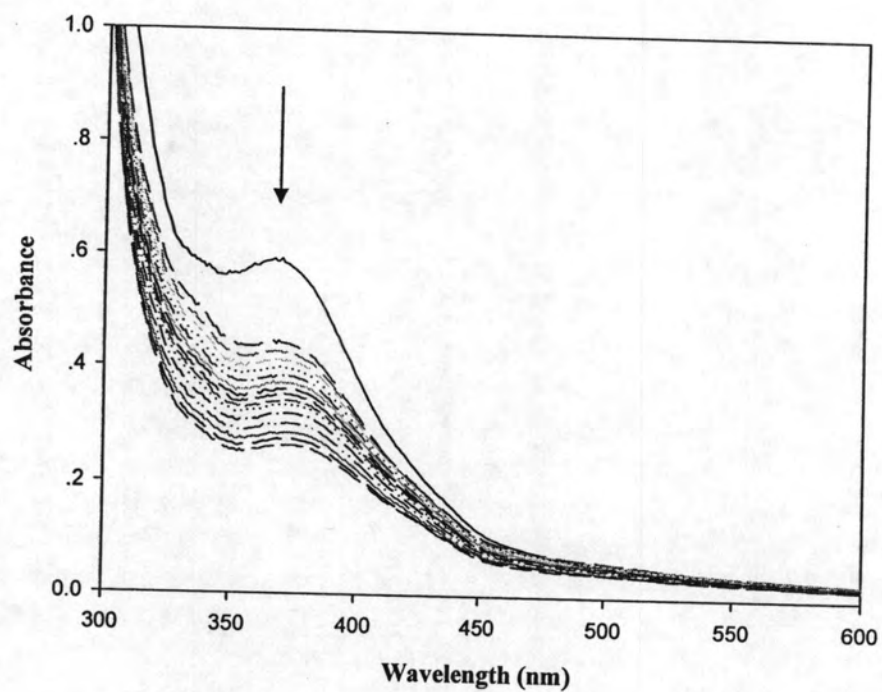


Figure A6 UV titration of L<sup>2</sup>H with Bu<sub>4</sub>NC<sub>6</sub>H<sub>5</sub>COO in DMSO. [L<sup>2</sup>H] = 7.5 \* 10<sup>-4</sup> M and [Bu<sub>4</sub>NC<sub>6</sub>H<sub>5</sub>COO] = 0-0.02 M.

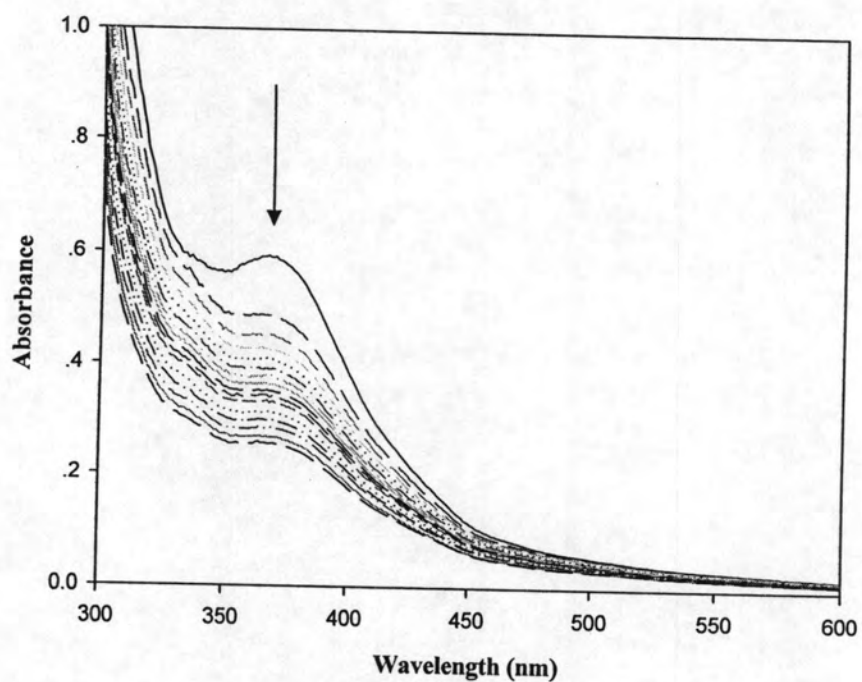
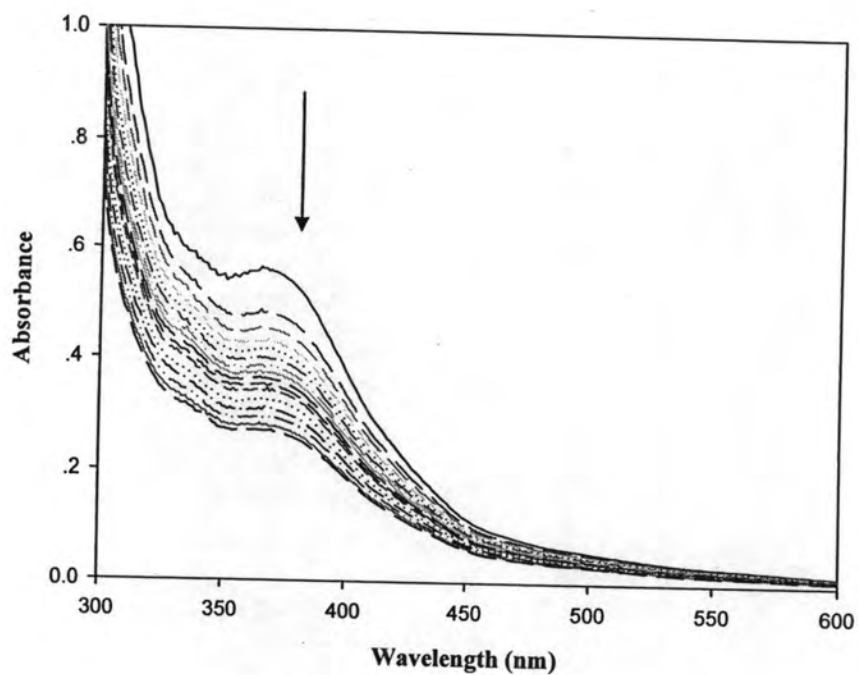
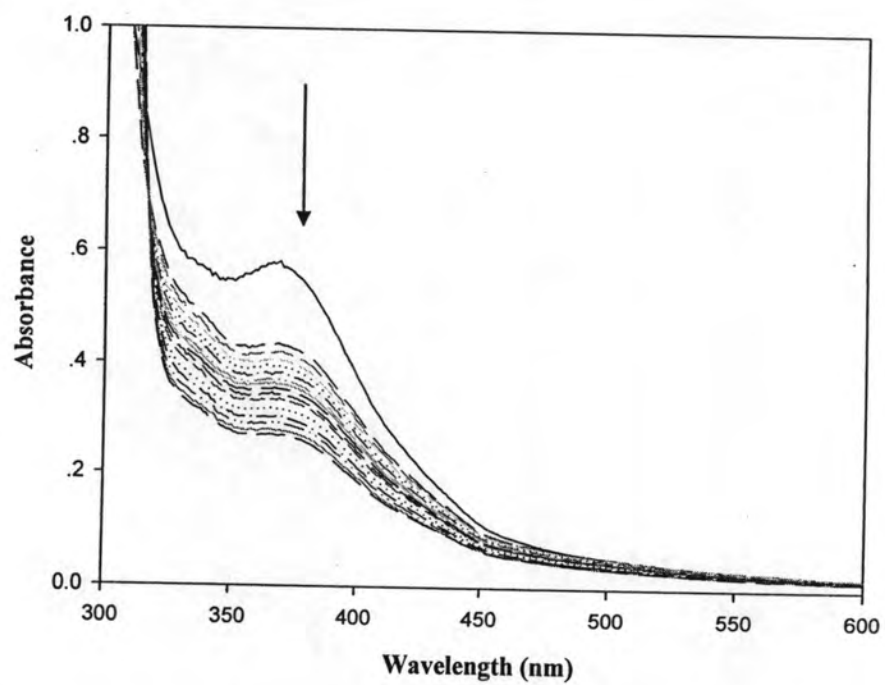


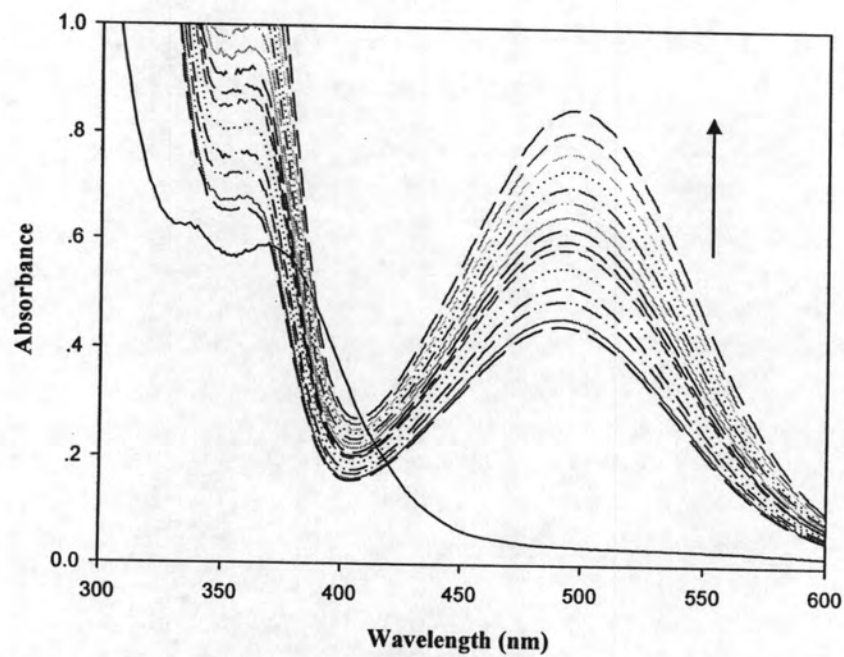
Figure A7 UV titration of L<sup>2</sup>H with Bu<sub>4</sub>NBr in DMSO. [L<sup>2</sup>H] = 7.5 \* 10<sup>-4</sup> M and [Bu<sub>4</sub>NBr] = 0-0.02 M.



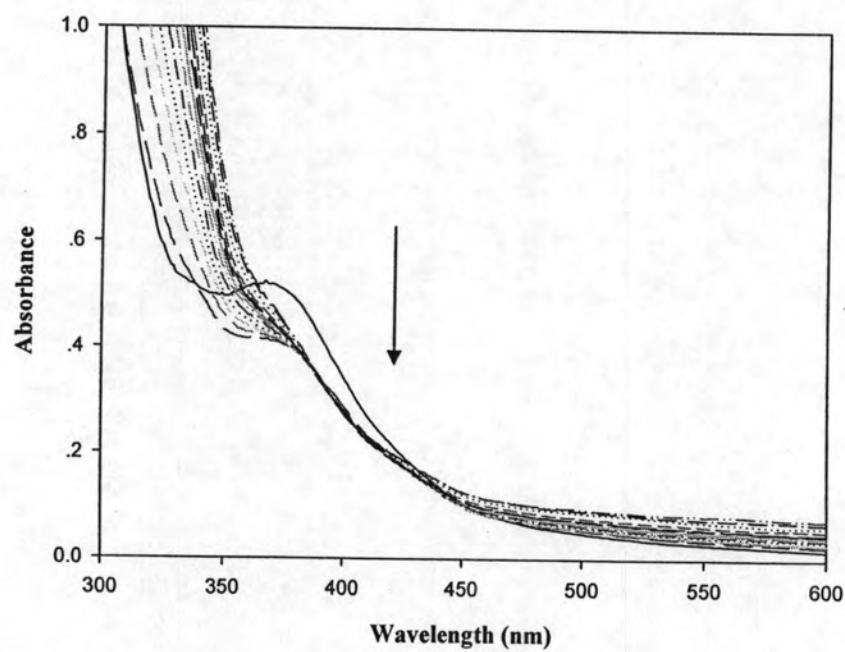
**Figure A8** UV titration of L<sup>2</sup>H with Bu<sub>4</sub>NCl in DMSO. [L<sup>2</sup>H] =  $7.5 \times 10^{-4}$  M and [Bu<sub>4</sub>NCl] = 0-0.02 M.



**Figure A9** UV titration of L<sup>2</sup>H with Trp in DMSO. [L<sup>2</sup>H] =  $7.5 \times 10^{-4}$  M and [Trp] = 0-0.02 M.

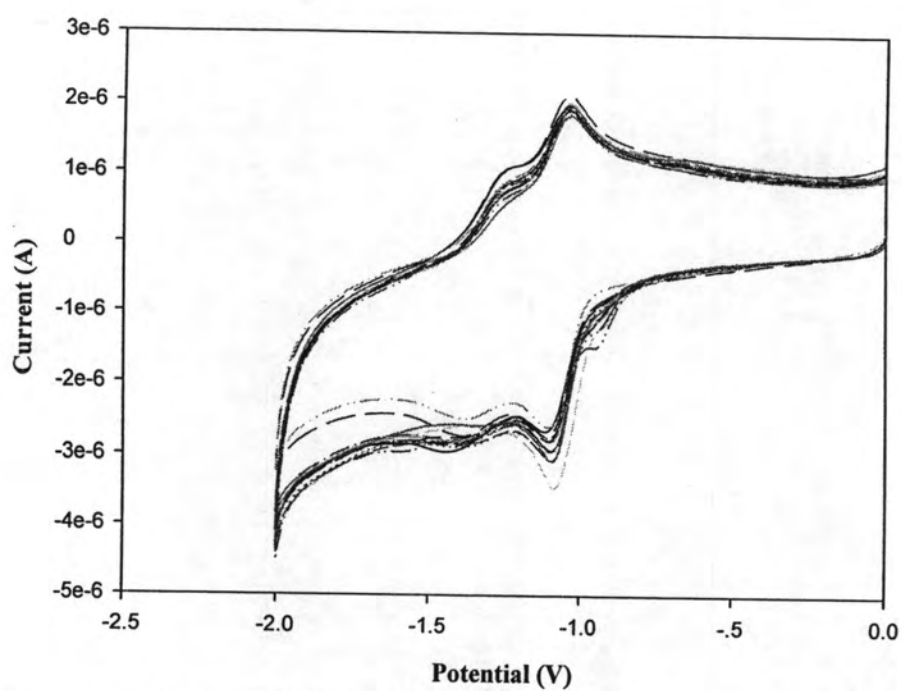


**Figure A10** UV titration of  $L^2H$  with  $Bu_4NOH$  in DMSO.  $[L^2H] = 7.5 \times 10^{-4}$  M and  $[Bu_4NOH] = 0-0.02$  M.

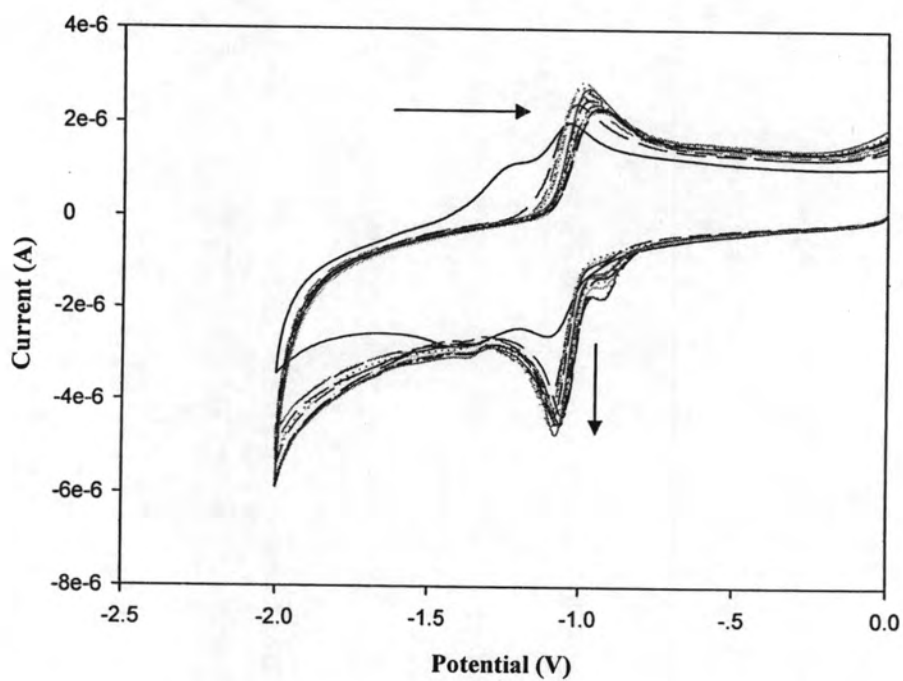


**Figure A11** UV titration of  $L^2H$  with Phe in DMSO.  $[L^2H] = 7.5 \times 10^{-4}$  M and  $[Phe] = 0-0.02$  M.

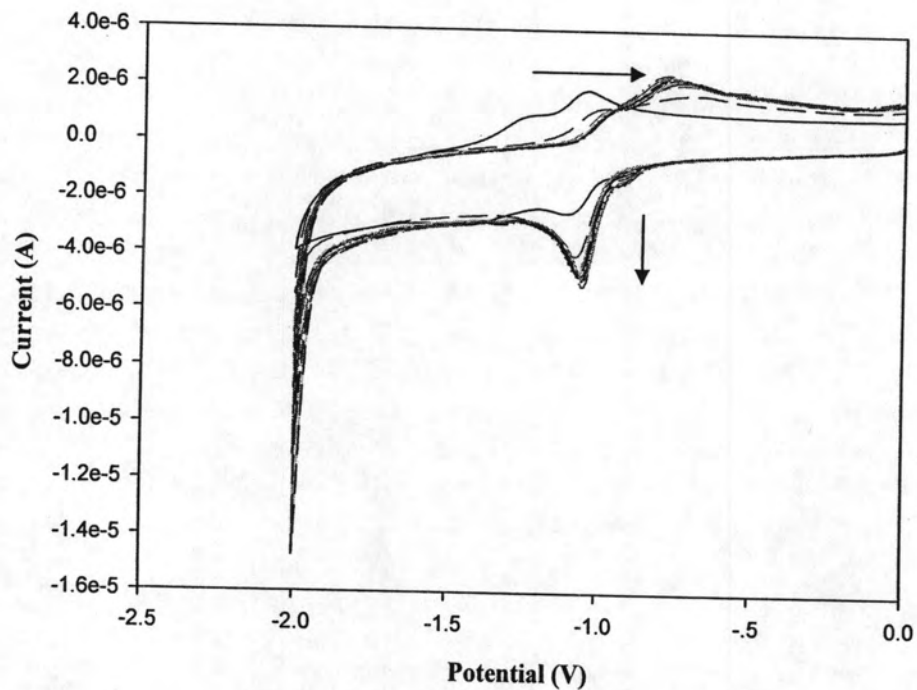




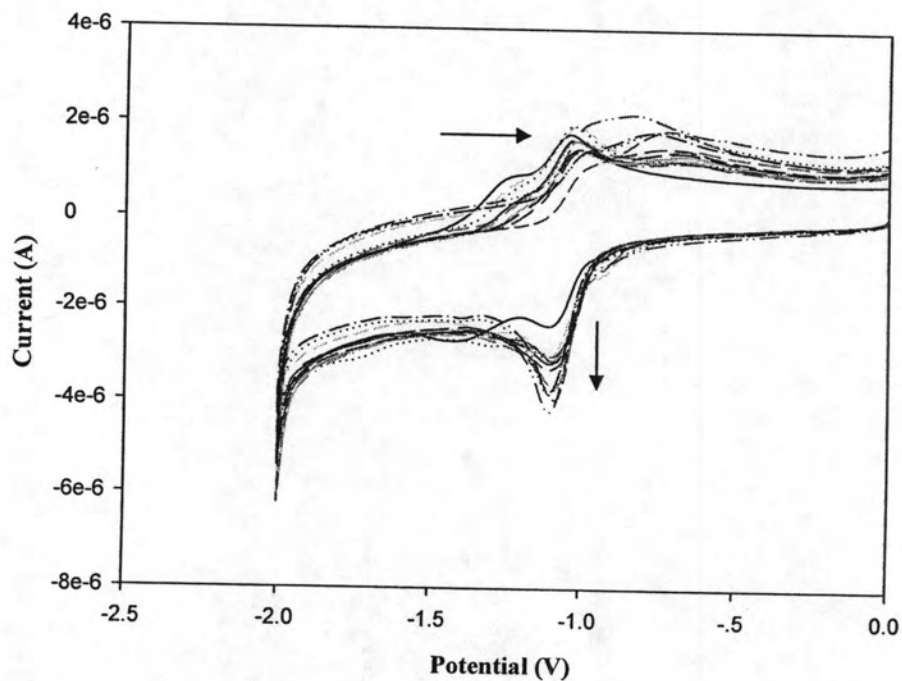
**Figure A12** CV titration of  $L^2H$  with  $Bu_4NCl$  in DMSO.  $[L^2H] = 1 * 10^{-3}$  M and  $[Bu_4NCl] = 0.1$  M.



**Figure A13** CV titration of  $L^2H$  with  $Bu_4NCH_3COO$  in DMSO.  $[L^2H] = 1 * 10^{-3}$  M and  $[Bu_4NCH_3COO] = 0.1$  M.



**Figure A14** CV titration of  $L^2H$  with  $Bu_4NC_6H_5COO$  in DMSO.  $[L^2H] = 1 \times 10^{-3}$  M and  $[Bu_4NC_6H_5COO] = 0.1$  M.



**Figure A15** CV titration of  $L^2H$  with  $Bu_4NH_2PO_4$  in DMSO.  $[L^2H] = 1 \times 10^{-3}$  M and  $[Bu_4NH_2PO_4] = 0.1$  M.

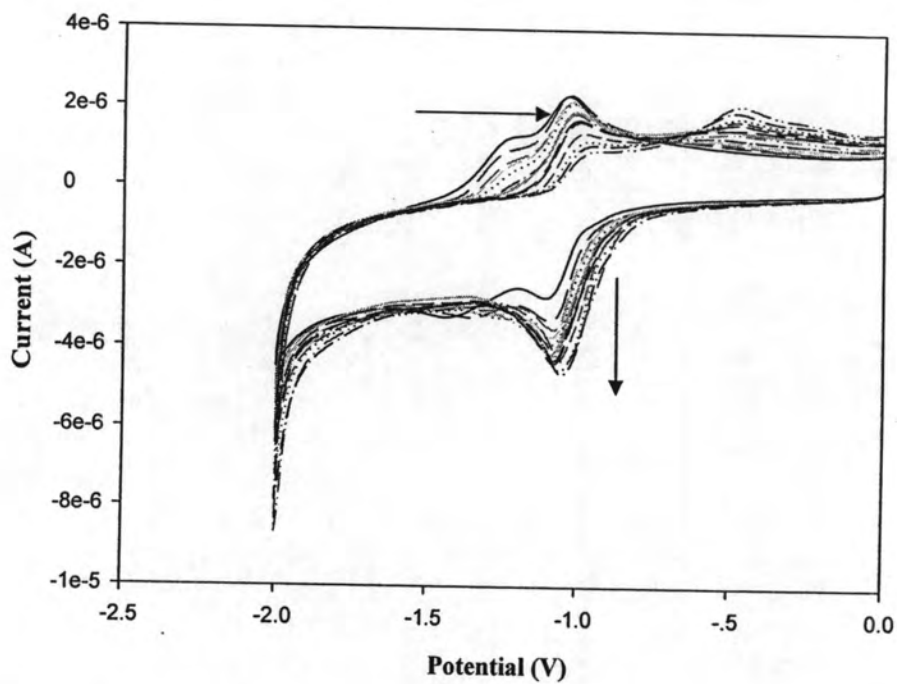


Figure A16 CV titration of  $L^2H$  with Phe in DMSO.  $[L^2H] = 1 \times 10^{-3}$  M and  $[Phe] = 0.1$  M.

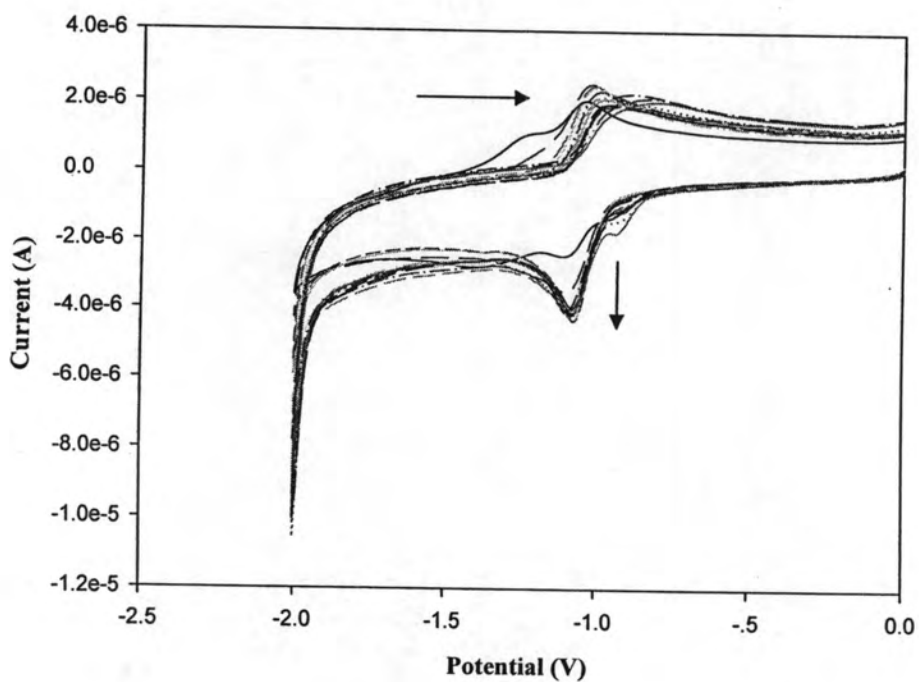


Figure A17 CV titration of  $L^2H$  with Trp in DMSO.  $[L^2H] = 1 \times 10^{-3}$  M and  $[Trp] = 0.1$  M.

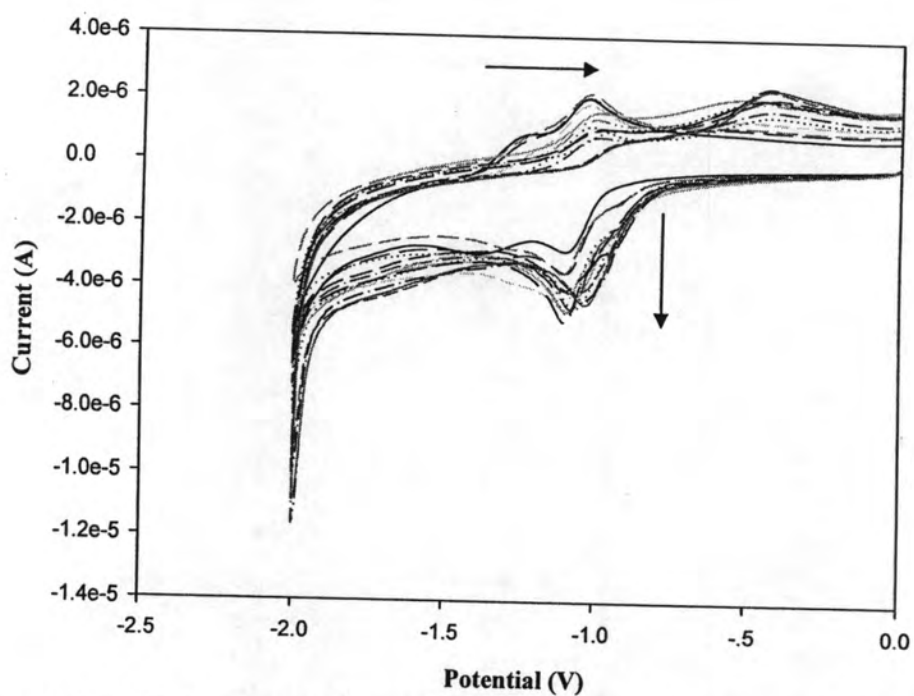


Figure A18 CV titration of  $L^2H$  with Ala in DMSO.  $[L^2H] = 1 \times 10^{-3}$  M and  $[Ala] = 0.1$  M.

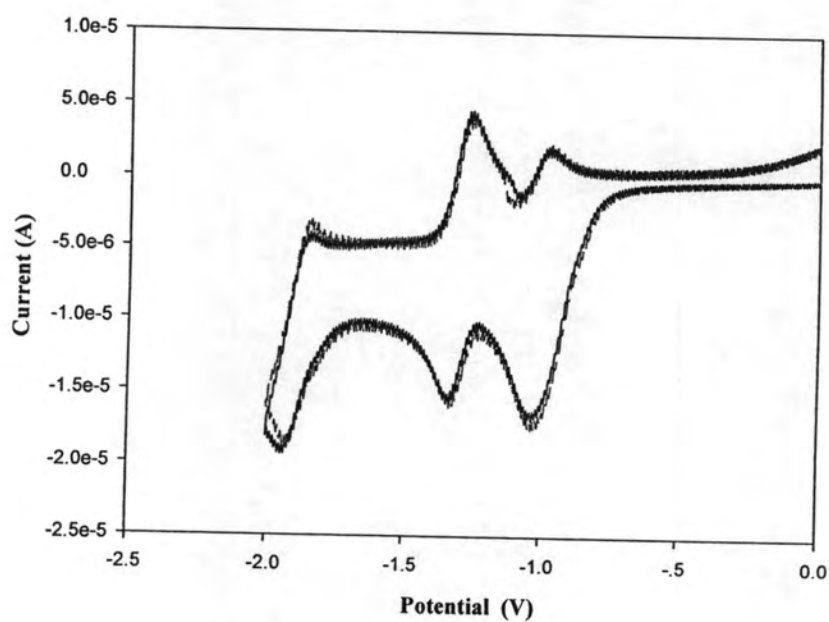
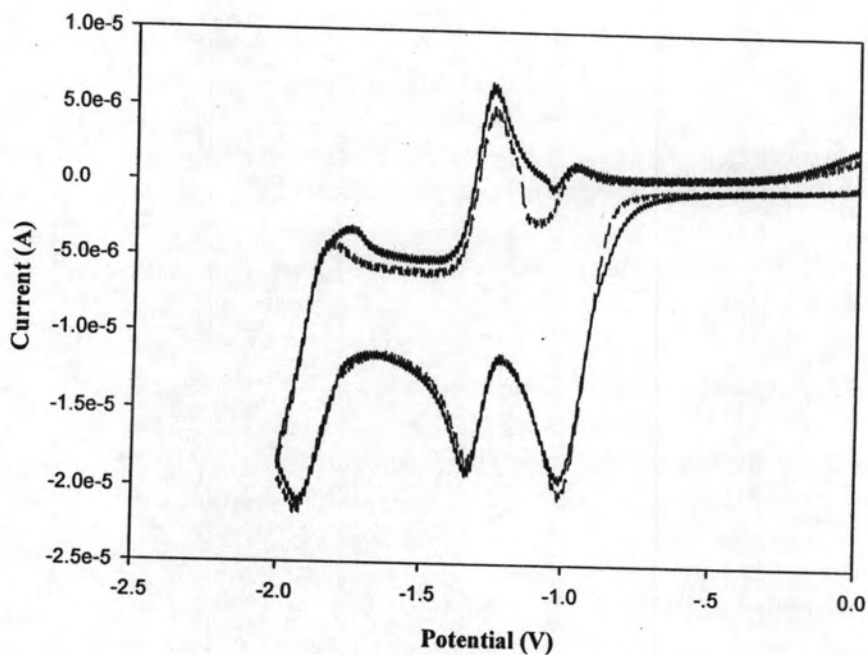
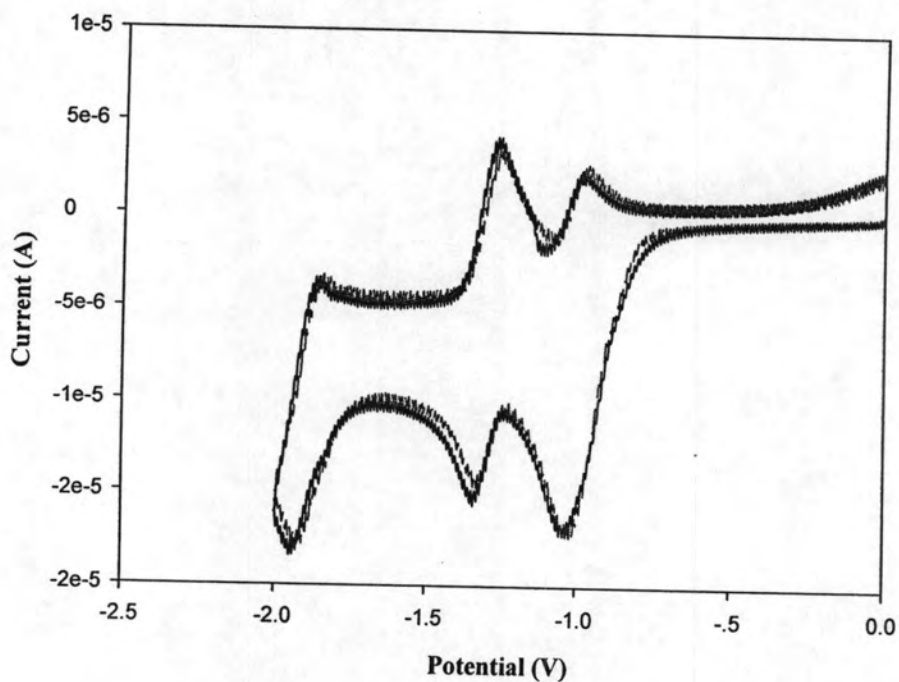


Figure A19 CV titration of  $L^3H$  with 4 equiv.  $Bu_4NCl$  in DMSO.  $[L^3H] = 1 \times 10^{-3}$  M and  $[Bu_4NCl] = 0.1$  M.

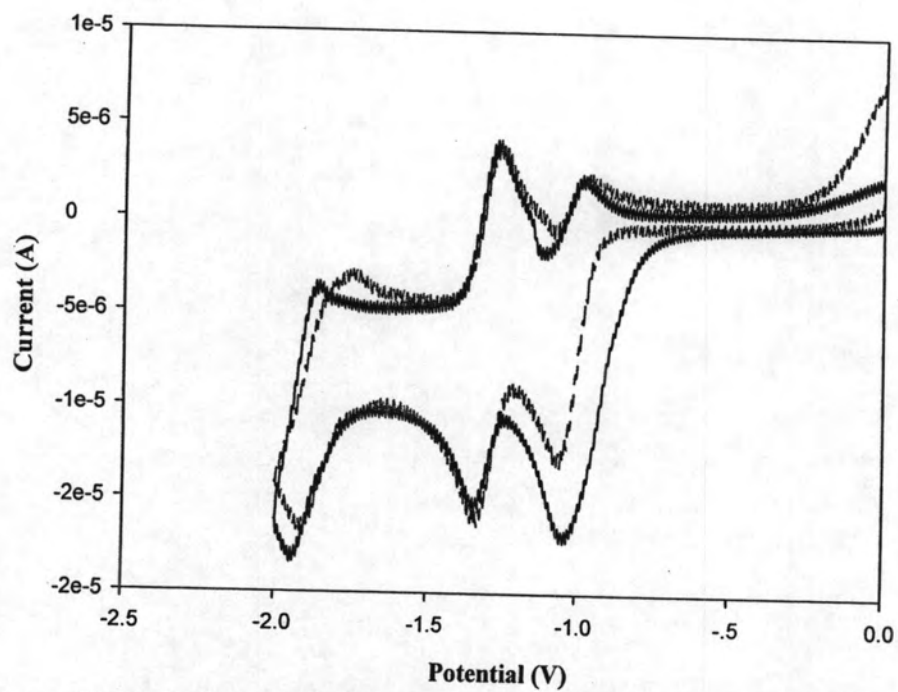




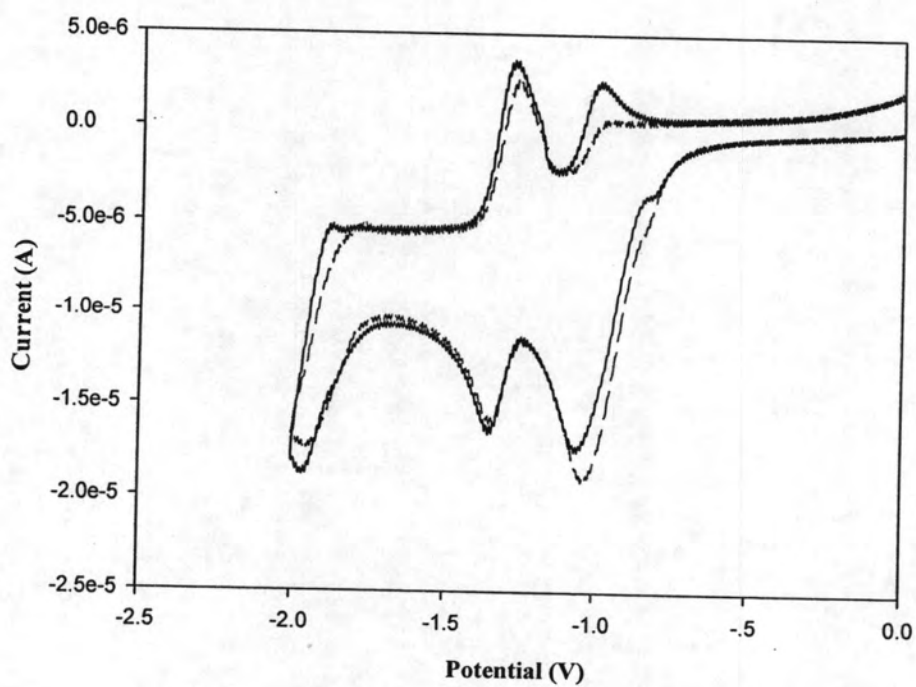
**Figure A20** CV titration of  $L^3H$  with 4 equiv.  $Bu_4NCH_3COO$  in DMSO.  
 $[L^2H] = 1 * 10^{-3}$  M and  $[Bu_4NCH_3COO] = 0.1$  M.



**Figure A21** CV titration of  $L^3H$  with 4 equiv.  $Bu_4NC_6H_5COO$  in DMSO.  
 $[L^2H] = 1 * 10^{-3}$  M and  $[Bu_4NC_6H_5COO] = M$  mol/L.



**Figure A22** CV titration of  $L^3H$  with 4 equiv.  $Bu_4NF$  in DMSO.  $[L^2H] = 1 * 10^{-3}$  M and  $[Bu_4NF] = 0.1$  M.



**Figure A23** CV titration of  $L^3H$  with 4 equiv. Ala in DMSO.  $[L^2H] = 1 * 10^{-3}$  M and  $[Ala] = 0.1$  M.

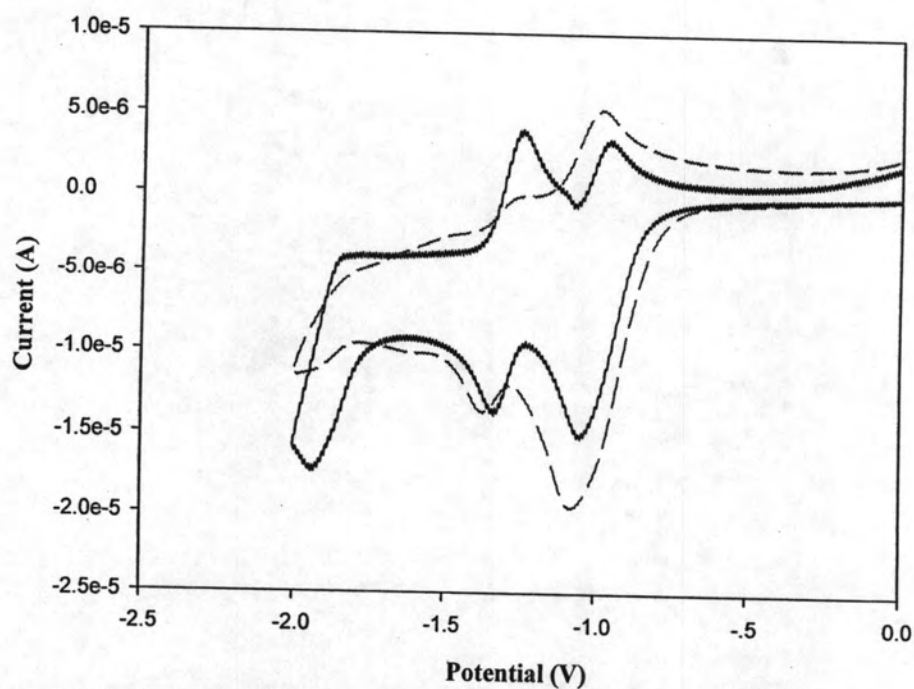


Figure A24 CV titration of  $L^3H$  with 4 equiv. Phe in DMSO.  $[L^2H] = 1 \times 10^{-3}$  M and  $[Phe] = 0.1$  M.

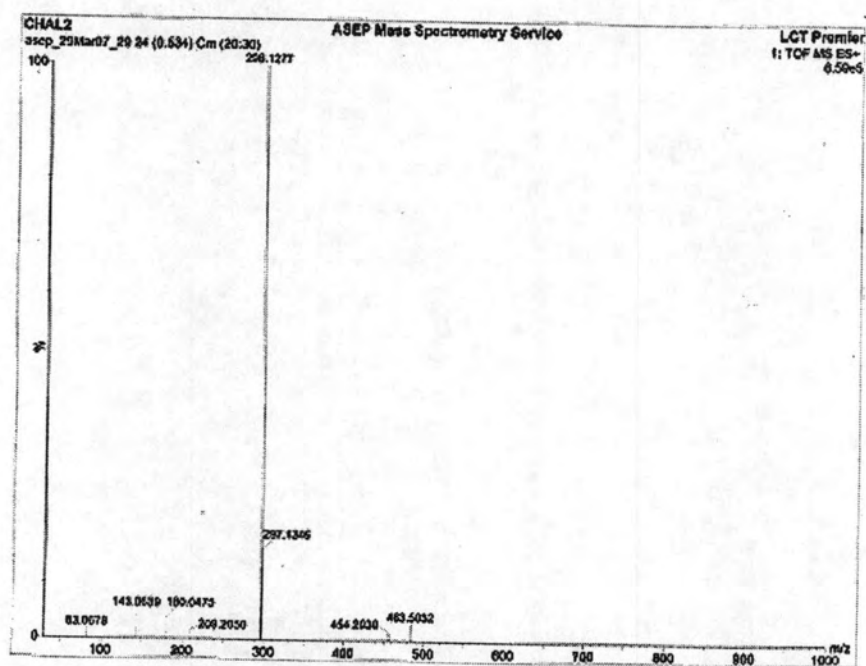


Figure A25 ESI MS ( $ES^+$ ) result of compound 11b

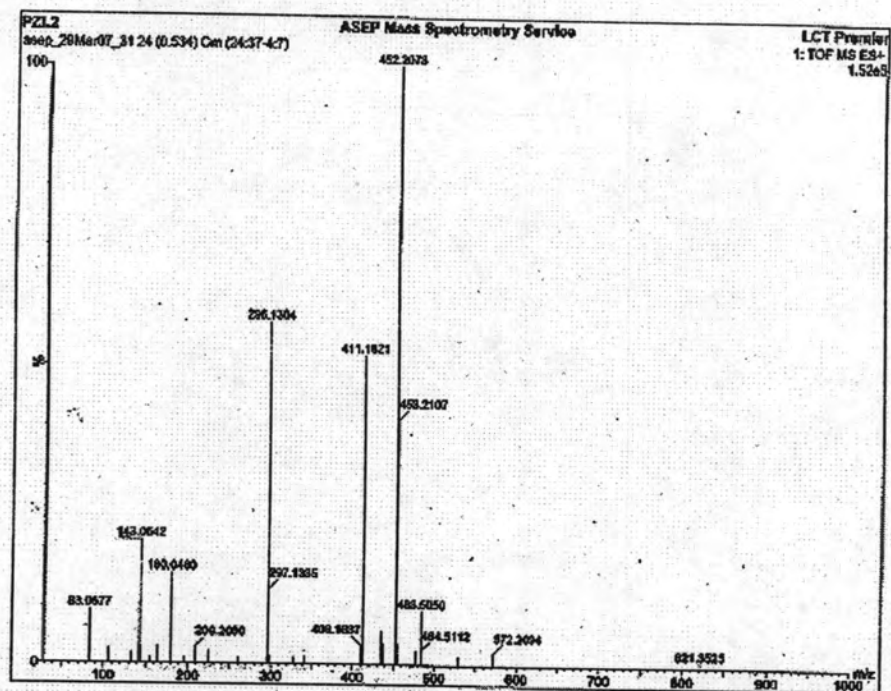


Figure A26 ESI MS ( $ES^+$ ) result of compound 12b

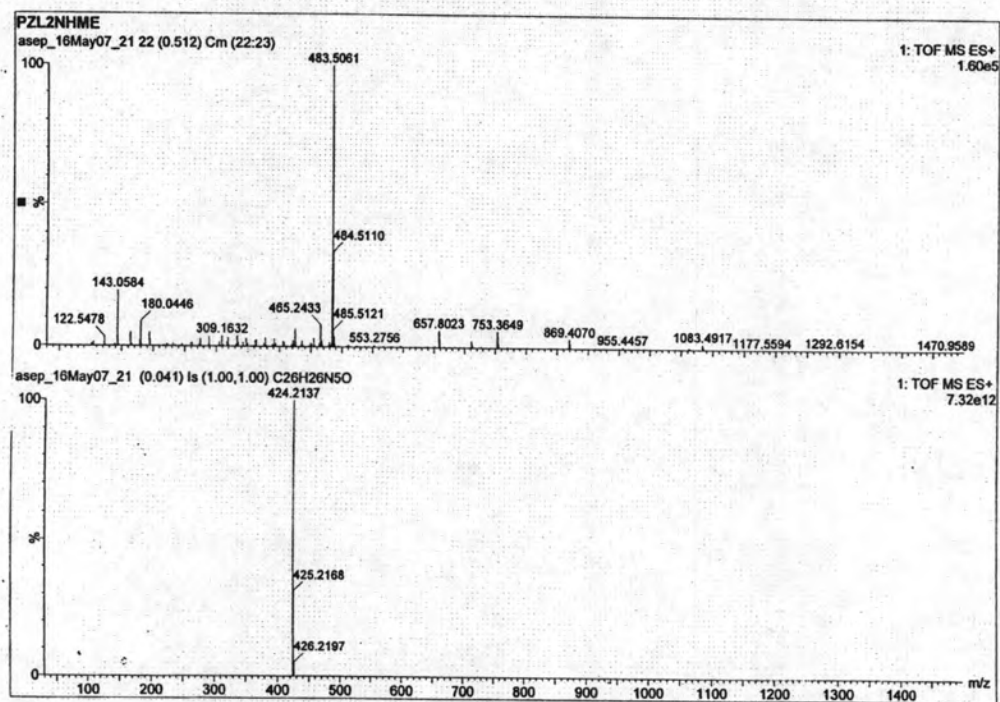


Figure A27 ESI MS ( $ES^+$ ) result of compound 13b



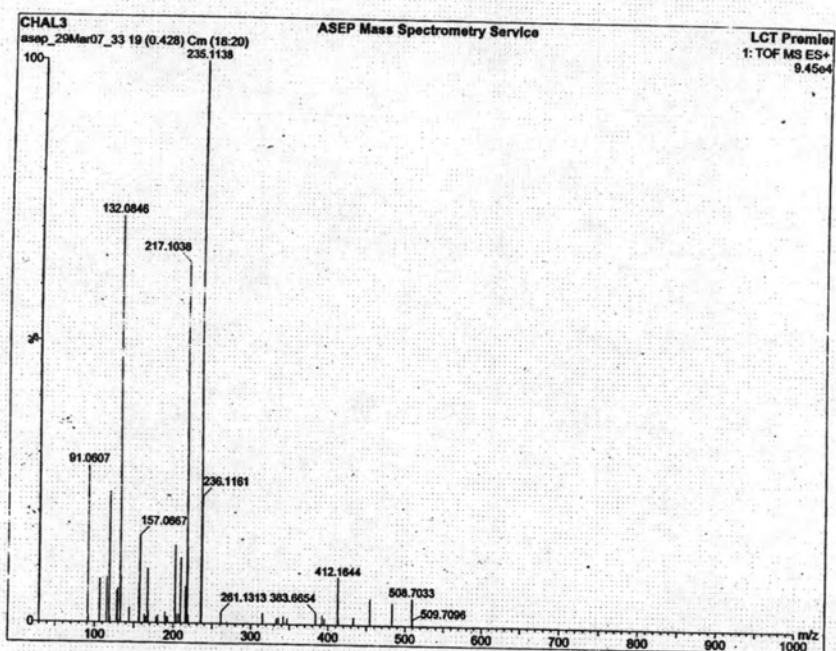


Figure A28 ESI MS ( $ES^+$ ) result of compound 17b

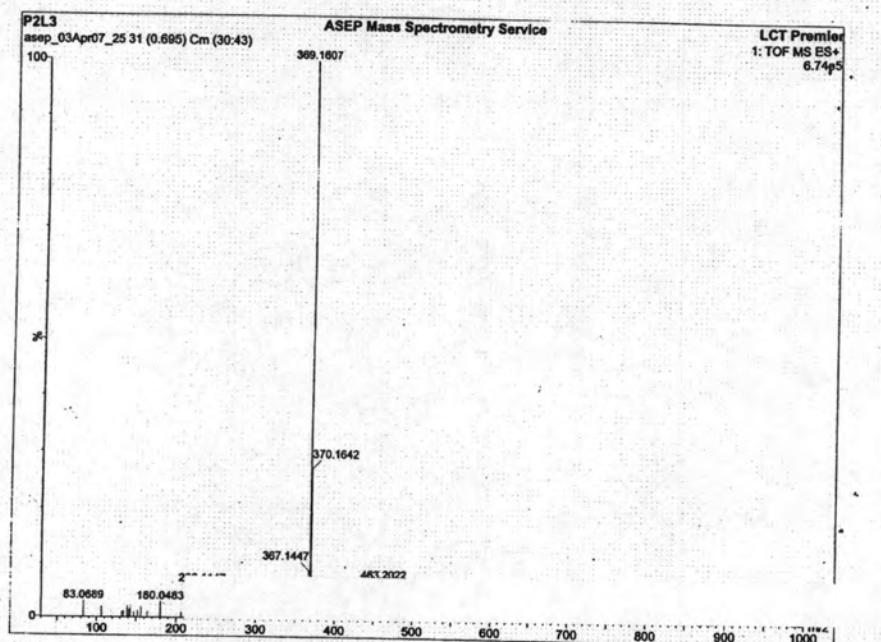


Figure A29 ESI MS ( $ES^+$ ) result of compound 18b

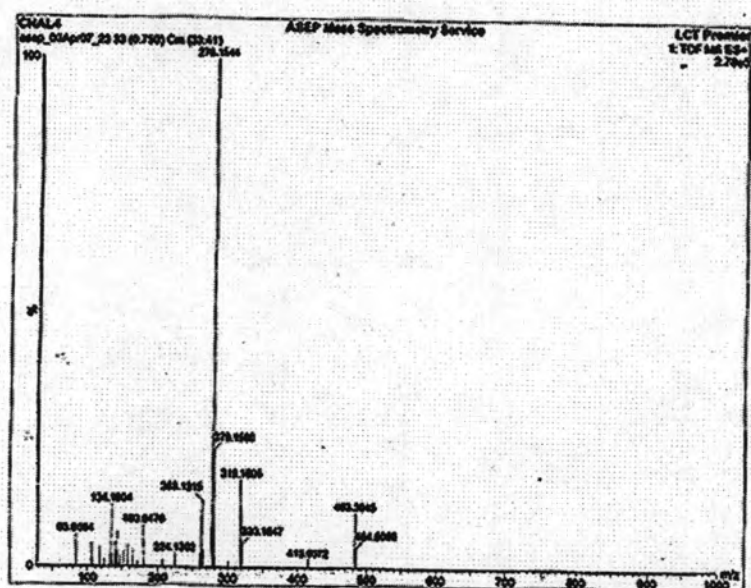


Figure A30 ESI MS ( $ES^+$ ) result of compound 19b

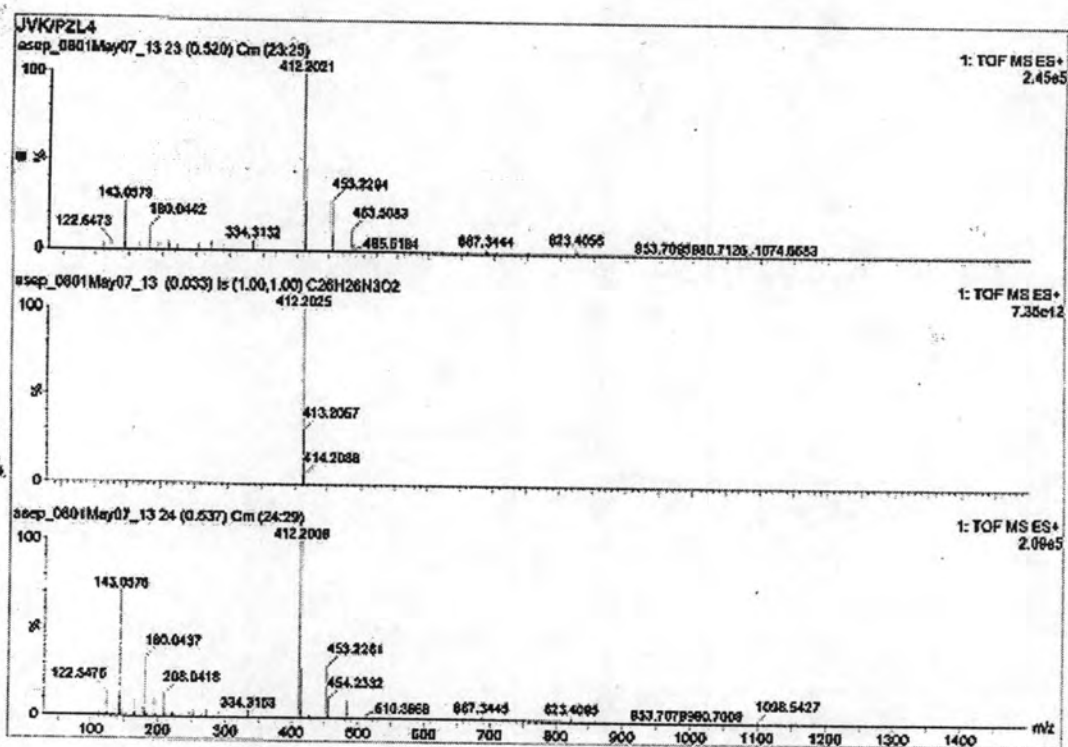


Figure A31 ESI MS ( $ES^+$ ) result of compound 20b

## VITA

Mr. Boontana Wannalarse was born on April 7, 1978 in Chiang mai, Thailand. He graduated with high school degree from The Prince Royal College, Chiang mai in 1995. He received his Bachelor's and Master's degree of Science in Chemistry from Chiang mai University in 1999 and 2002, respectively. Since then, he has been a postgraduate student studying in inorganic chemistry and become a member of Supramolecular Chemistry Research Unit under the supervision of Assoc. Prof. Thawatchai Tuntulani and co-supervision of Assist Prof. Boosayarat Tomapatanaget. During this period, his financial support is granted by "The Commission on Higher Education". This scholarship gave his an opportunity to do the research under the supervision of Prof. A. Prasanna de Silva at Queen's University, Northern Ireland, United Kingdom during 2006-2007. He graduated with a doctorate degree in chemistry in the academic year 2007.

### Publication

1. de Silva, A., P.; Uchiyama, S.; Vance, T., P.; Wannalarse, B., A Supramolecular Chemistry Basis for Molecular Logic and Computation, *Coord. Chem. Rev.*, 251 (2007): 1023-1032.