

การตั้งตำรับ ศึกษาการปลดปล่อย และความคงตัวของยาเตรียม ไดโครฟีแนค สำหรับใช้กับผิวหนัง

นาย สุดเขตต์ ด้วงชนะ [\]



วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาเภสัชศาสตรมหาบัณฑิต ภาควิชาเภสัชอุตสาหกรรม บัณฑิตวิทยาลัย จุฬาลงกรณ์มหาวิทยาลัย

พ.ศ. 2537

ISBN 974-582-160-08 ลิขสิทธิ์ของบัณฑิตวิทยาลัย จุฬาลงกรณ์มหาวิทยาลัย

FORMULATION, RELEASE AND STABILITY STUDIES OF DICLOFENAC TOPICAL DOSAGE FORMS

Mr. SOODKATE DUANGCHANA

A Thesis Submitted in Partial Fulfillment of the Requirements for the Degree of Master of Science in Pharmacy

Department of Manufacturing Pharmacy

Graduate School

Chulalongkorn University

1994

ISBN 974-582-160-08

Thesis Title

Formulation, Release and Stability

Studies of Diclofenac Topical

Dosage Forms

By

Mr. Soodkate Duangchana

Department

Manufacturing Pharmacy

Thesis Advisor Assoc. Prof. Garnpimol C. Ritthidej, Ph.D.

Accepted by the Graduate School. Chulalongkorn University in Partial Fulfillment of the Requirement for the Master's Degree

> Santi Throngsunan Dean of Graduate School (Assoc. Prof. Santi Thoongsuwan, Ph.D.)

Thesis Committee

P. Mount Chairman

(Assist. Prof. Poj Kulvanich, Ph.D.)

Lampinol C. fittidy... Thesis Advisor

(Assoc. Prof. Garnpimo / C. Ritthidej, Ph.D.)

Parimer Thanon kist Member

(Assoc. Prof. Parunee Thanomkiat, M.Pharm. St.)

.. allen this Nimmannit Member

(Assoc. Prof. Ubolthip Nimmannit, Ph.D.)

สุดเขตต์ ด้วงชนะ: การตั้งตำรับ คึกษาการปลดปล่อย และความคงตัวของยาเตรียม ไดโครพีแนค สำหรับใช้กับผิวหนัง (FORMULATION, RELEASE AND STABILITY STUDIES OF DICLOFENAC TOPICAL DOSAGE FORMS) อ.ที่ปรึกษา: รคํ.ดร.กาญลน์พิมล ฤทธิเดช, 250 หน้า. ISBN 974-582-160-08

จากการศึกษาพบว่า ไดโครที่แนคโซเดียม และไดโครพี่แนคโดเอททิลแอมโมเนียม มีรูปร่าง และขนาดผลึกทางจุลภาค อินฟราเรดส์เปคตัม ค่าการละลาย ความคงตัวและปริมาณที่ถูกปลดปล่อยจากยาเตรียมแตกต่างกัน ลำดับความคงตัวของยาคือ พีเอช่ 9 > พีเอช่ 8 > พีเอช่ 7 ∿ พีเอช่ 6 >> พีเอช่ 5 องค์ประกอบของส่วนน้ำมันในตำรับครีมชนิดน้ำมันในน้ำ ชนิดและปริมาณของสำรก่อเจลในตำรับเจล รวมทั้งปริมาณน้ำมันในตำรับครีมและเจลน้ำ-น้ำมัน มีผลต่อการปลดปล่อยตัวยาอย่างมีนัยสำคัญ ไดโคร - พี่แนคโซเดียม ถูกปลดปล่อยจากตำรับครีมชนิดปี (ประกอบด้วย กรดส่เตียริค ส่เปอร์มาลิติ และ ไวท์ปีล์แวค) น้อยกว่าไดโครพี่แนคโดเอททิลแอมโมเนียมแต่ในเจลโซเดียมซีเอ็มซีจะถูกปลดปล่อยออกมามากกว่า อย่างมีนัยสำคัญ ลำดับการปลดปล่อยตัวยา จากยาเตรียมรูปแบบต่าง ๆ คือ เจลโซเดียมซีเอ็มซี > ผลิตภัณฑ์ในท้องตลาด > เจลน้ำ-น้ำมัน ∿ เจลโพลอคซาเมอร์ > ครีมชนิดน้ำมันในน้ำ

ภาควิชาเภลัยอุตลำหกรรม	
สาขาวิชา เภลัยอุตลำหกรรม	
ปีการศึกษา 2537	

าายมือชื่อนิสิต	(C 300
ายมือชื่ออาจารย์ที่ปรึกษา	
ายมือชื่ออาจารย์ที่ปรึกษาร่วม	9 //

##C375190 MAJOR MANUFACTURING PHARMACY
KEY WORD: DICLOFENAC/ RELEASE/ TOPICAL/ STABILITY/ SOLUBILITY
SOODKATE DUANGCHANA: FORMULATION, RELEASE AND STABILITY STUDIES OF
DICLOFENAC TOPICAL DOSAGE FORMS. THESIS ADVISOR: ASSO. PROF.
GARNPIMOL C. RITTHIDEJ, Ph.D. 250 pp. ISBN 974-582-160-08

Diclofenac sodium (DS) and diclofenac diethylammonium (DE) were different in microscopic appearance, IR spectra, solubility, stability and the release from topical preparations. The general rank order of drug stability was pH 9 > pH 8 > pH 7 $^{\circ}$ pH 6 >> pH 5. The composition of oil phase in o/w creams, type and percent of gelling agent in hydrophilic gels significantly affected the release of drug. The percent of oil phase in o/w creams and oil-water gels also significantly affected the release. The release of DS from oil phase B cream containing spermaceti, stearic acid and white beeswax was significantly less than of DE but shown the reverse in sodium CMC gel. The general rank order of the release of drug was sodium CMC gel > commercial products (A, B and C) > oil-water gel $^{\circ}$ poloxamer gel > o/w cream.

ภาควิชา เภสัชอุตลำหกรรม สาขาวิชา เภสัชอุตลำหกรรม ปีการศึกษา 2537

ลายมือชื่ออาจารย์ที่ปรึกษา มีภาษาทาง (. คำในค่า ลายมือชื่ออาจารย์ที่ปรึกษาร่วม



I would like to express my sincere gratitude to Associate Professor Dr. Garnpimol C. Ritthidej for her excellent supervisor, invaluable guidance, and encouragement that enable me to carry out my thesis completely.

Gratitude is extened to Assistant Professor Dr. Poj Kulvanich, Head of the Department of Manufacturing Pharmacy, for his encouragement and permission of using testing instrument and apparatus all the time.

I am indebted to the Department of Manufacturing Pharmacy, Faculty of Pharmaceutical Science, Khon Kaen University for the use of computer facilities and graduate school, Chulalongkorn University for supporting this project.

I would like to express my sincere gratitude to my parent for their encouragement and compassion. Special thanks to all lecturer colleague in Khon Kaen for their encouragement. As well, many thanks to all of my friends for their friendship throughout the period of my studies. Finally, I would like to give many thanks to all of my young brothers and sisters in Angthong for their interesting and care during my writing and interpretation of this project along 2 months in Angthong.

CONTENTS

		Page.
Abstract (T	HAI)	iv
Abstract (E	NGLISH)	٧
Acknowledgem	ents	v i
List of Tabl	es	viii
List of Figu	res	×i
Abbreviation		xvii
Chapter		
I	Introduction	1
II	Experimental	34
III	Results	75
ΙV	Discussion and Conclusion	116
Refferences.		166
Appendix		184
Vita		250

Table List

Table		Page
1	Composition of DS Creams	4 1
2	Composition of DE Creams	42
3	Composition of DS Carbopol Gels	44
4	Composition of DE Carbopol Gels	45
5	Composition of DS and DE Hydrophilic Gels.	45
6	Composition of DS and DE Emulsions	48
7	Composition of DS Oil-Water Gels	48
8	Composition of DE Oil-Water Gels	50
9	Calibration Data from HPLC Chromatogram of	
	DS in Solubility and Stability Study	59
10	Calibration Data from HPLC chromatogram of	
	DE in solubility and stability study	62
11	Calibration Data from HPLC Chromatogram of	
	DS in release study	66
12	Calibration Data from HPLC Chromatogram of	
	DE in release study	69
13	UV Absorbance of DS and DE in 0.1 N NaOH	
	at Wavelength of 276 nm	72
14	Solubility of DS and DE at 35 °C	81
15	The percent of DS remained in pH 5,6,7,8	01
4.7	and 9 solution after 5 months of storage	
	at 35, 45, 55 and 65 ^O C	82

Table		Page
16	The percent of DE remained in pH 5,6,7,8	
	and 9 solution after 5 months of storage	
	at 35, 45, 55 and 65 OC	94
17	Physical appearance of DS and DE topical	
	preparations	105
18	Percent label amount of diclofenac in	
	topical preparations	109
19	The percent of diclofenac released from	
	o/w creams	110
20	The percent of diclofenac released from	
	hydrophilic gels	112
21	The percent of diclofenac released from	
	oil-water gels	113
22	The percent of diclofenac released from	
	commercial products	115
23	Degradation rate of diclofenac in pH 5, 6,	
	7, 8 and 9 buffer solution at 35, 45, 55	
	and 65 °C	122
24	Activated energy (Ea) of diclofenac in pH	
i	5, 6, 7, 8 and 9 buffer solution	130
25	Student 's t distribution test of	
	hypothesis and significance for DS creams.	136
26	Student 's t distribution test of	
ř	hypothesis and significance for DE creams.	136

Table		Page.
27	Student 's t distribution test of	
	hypothesis and significance for DS	
	hydrophilic gels	145
28	Student 's t distribution test of	
	hypothesis and significance for DE	
	hydrophilic gels	145
29	Analysis of variance and the F test for	
	the null hypothesis of equal means of DS	
	oil-water gels	151
30	Analysis of variance and the F test for	
	the null hypothesis of equal means of DE	
	oil-water gels	152
31	Student 's t distribution test of	
	hypothesis and significance for DS and DE	
	topical preparations	159

Figure List

Figure.		Page.
1	Schematic illustration of Franz-diffusion	
	cell	14
2	Schematic illustration of Chien-Valia	
	diffusion cell	16
3	Schematic illustration of USP dissolution	
	apparatus 3 : Paddle over Disk	18
4	Schematic illustration of a Jar with semi-	
	permeable membrane diffusion cell	19
5	Schematic illustration of diffusion cell	
	of Key Method	21
6	Schematic illustration of diffusion cell	
	of Searle Method	23
7	Schematic illustration of Polano and Ponac	
	diffusion cell	24
8	Schematic Diagram of Enhancer Cell	26
9	Schematic Diagram of 200 mL Flask Centering	
	Ring Assembly	27
10	Schematic illustration of diffusion cell	
	disk modified from disk of USP	
	dissolution apparatus 3	53
11	HPLC chromatogram of propyl paraben,	
	diclofenac and mefenamic acid	55

Figure.		Page
12	A HPLC chromatograms of DS and PP in	
	standardization of solubility and	
	stability study	58
13	A HPLC calibration curve of DS solubility	
	and stability study	60
14	A HPLC chromatograms of DE and PP in	
	standardization of solubility and	
	stability study	61
15	A HPLC calibration curve of DE solubility	
	and stability study	63
16	A HPLC chromatograms of DS and MFA in	
	standardization of release study	65
17	A HPLC calibration curve of DS release	
	study	67
18	A HPLC chromatograms of DE and MFA in	
	standardization of release study	68
19	A HPLC calibration curve of DE release	
	study	70
20	A UV spectrometrical calibration curve of	70
	DS and DE in 0.1 N NaOH solution at	
¥	wavelenght of 276 nm	7.
21		74
21	Photomicrograph under scanning electron	-
22	microscope of DS	76
44	Photomicrograph under scanning electron	
	microscope of DE	77
23	Infrared spectrum of DS	78

Figure.		Page
24	Infrared spectrum of DE	79
25	Plots of Percent label amount of DS in	, 5
	pH 5 buffer solution versus time of storage	
	at 35, 45, 55 and 65 OC	84
26	Plots of Percent label amount of DS in	
	pH 6 buffer solution versus time of storage	
	at 35, 45, 55 and 65 OC	85
27	Plots of Percent label amount of DS in	
	pH 7 buffer solution versus time of storage	
	at 35, 45, 55 and 65 °C	86
28	Plots of Percent label amount of DS in	
	pH 8 buffer solution versus time of storage	
	at 35, 45, 55 and 65 °C	87
29	Plots of Percent label amount of DS in	
	pH 9 buffer solution versus time of storage	
	at 35, 45, 55 and 65 °C	88
30	Plots of Percent label amount of DS in	
	pH 5, 6, 7, 8 and 9 buffer solution versus	
	time of storage at 35 °C	89
31	Plots of Percent label amount of DS in	
	pH 5, 6, 7, 8 and 9 buffer solution versus	
	time of storage at 45 °C	90
32	Plots of Percent label amount of DS in	
	pH 5, 6, 7, 8 and 9 buffer solution versus	
	time of storage at 55 °C	91

Figure		Page.
		rage.
33	Plots of Percent label amount of DS in	
	pH 5, 6, 7, 8 and 9 buffer solution versus	
	time of storage at 65 °C	92
34	Plots of Percent label amount of DE in	
	pH 5 buffer solution versus time of storage	
	at 35, 45, 55 and 65 °C	95
35	Plots of Percent label amount of DE in	
	pH 6 buffer solution versus time of storage	
	at 35, 45, 55 and 65 °C	96
36	Plots of Percent label amount of DE in	
	pH 7 buffer solution versus time of storage	
	at 35, 45, 55 and 65 °C	97
37	Plots of Percent label amount of DE in	
	pH 8 buffer solution versus time of storage	
	at 35, 45, 55 and 65 °C	98
38	Plots of Percent label amount of DE in	
	pH 9 buffer solution versus time of storage	
	at 35, 45, 55 and 65 °C	99
39	Plots of Percent label amount of DE in	
	pH 5, 6, 7, 8 and 9 buffer solution versus	
	time of storage at 35 °C	100
40	Plots of Percent label amount of DE in	
	pH 5, 6, 7, 8 and 9 buffer solution versus	
	time of storage at 45 °C	101

Figure.		Page.
41	Plots of Percent label amount of DE in	
	pH 5, 6, 7, 8 and 9 buffer solution versus	
	time of storage at 55 °C	102
42	Plots of Percent label amount of DE in	
	pH 5, 6, 7, 8 and 9 buffer solution versus	
	time of storage at 65 °C	103
43	pH-rate profile for the degradation of	
	DS at 35, 45, 55 and 65 OC	123
44	pH-rate profile for the degradation of	
	DE at 35, 45, 55 and 65 °C	124
45	Schematic illustration of chemical	
	degradation reaction mechanism of	
	diclofanac	125
46	Arrhenius plots for the degradation of DS	
	at pH 5, 6, 7, 8 and 9	128
47	Arrhenius plots for the degradation of DE	
	at pH 5, 6, 7, 8 and 9	129
48	Plots of percent DS release from cream	
	versus time of release study	137
49	Plots of percent DE release from cream	
	versus time of release study	138
50	Plots of percent DS and DE release from	
	creams which contained 17 percent of oil	
	phase versus time of release study	140

Figure.		Page.
51	Plots of percent DS and DE release from	
	creams which contained 27 percent of oil	
	phase versus time of release study	141
52	Plots of percent DS release from	
	hydrophilic gels versus time of release	
	study	143
53	Plots of percent DE release from	
	hydrophilic gels versus time of release	
	study	144
54	Plots of percent DS release from	
	oil-water gels versus time of release study	149
55	Plots of percent DE release from	
	oil-water gels versus time of release study	150
56	Plots of percent diclofenac release from	
	commercial products	155
57	Plots of percent DS release from	
	formulation 7, 53, 57, 79 and product C	
	versus time of release study	156
58	Plots of percent DE release from	
	formulation 17, 59, 62, 89 and product A	
	versus time of release study	157
59	Plots of percent release of DS and DE from	
	oil phase B cream	160
60	Plots of percent release of DS and DE from	
	sodium CMC gel	161
9		

ABBREVIATION

alc.

A1coho1

 $^{\circ}$ C

Degree Cencius

Ceto. 1000

Cetomacrogol 1000

cm

Centimetre

DE

Diclofenac dietylammonium

DS

Diclofenac sodium

Ea

Energy of activation

g.

Gram

HLB

Hydrophilic-Lipophilic Balance

hr

Hour

IPA

Isopropyl alcohol

IPM

Isopropyl myristate

K

Kelvin (absolute temperature)

mcg

Microgram

mcL

Microlitre

MFA

Mefenamic acid

mg

Milligram

min

Minute

mL

Mililitre

mm

Milimetre

mh

Month

N

Normality

PHC

PEG-40-hydrogenated castor oil

PG

Propylene glycol

PP

Propyl paraben

Sor. Sol.

Sorbitol solution

Sodium CMC

Sodium Carboxymethylcellulose

TEA

Triethanolamine

Temp.

Temperature

wk

Week

WSP

White soft paraffin