การกำจัดทองแดงที่ละลายอยู่ในน้ำโดยเอคโซโพลีแชคคาไรด์จากแบคทีเรียสายพันธุ์คัดที่ทนต่อทองแดง

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REMOVAL OF SOLUBLE COPPER BY EXOPOLYSACCHARIDE ISOLATED FROM COPPER-RESISTANT BACTERIAL ISOLATES

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แบคทีเรียที่สามารถทนต่อโลหะทองแดงและสามารถสร้างสารเอคโซโพลีแซคคาไรด์จำนวน 2 สายพันธุ์ ซึ่งคัดเลือก มาจาก 350 สายพันธุ์ได้นำมาใช้ในการทดลอง โดยให้ชื่อสายพันธุ์ว่า CuR-38 และ CuR-40 จากการทดลองพบว่าน่าจะเป็นแบคทีเรีย ที่จัดอยู่ในกลุ่ม Zoogloea sp. และ Bacillus sp. ตามลำดับ ช่วงค่าความเป็นกรด-ด่างที่ 7 และอุณหภูมิที่ 37 องศาเซลเซียสเป็น สภาวะที่เหมาะสมต่อการเจริญเติบโตและการสร้างสารเอคโซโพลีแซคคาไรด์ การผลิตสารเอคโซโพลีแซคคาไรด์จะผลิตได้บริมาณ มากที่สุดที่เวลา 48 ชั่วโมง แต่การเติมโลหะทองแดงในอาหารเลี้ยงเชื้อพบว่าไม่มีผลทำให้การสร้างสารเอคโซโพลีแซคคาไรด์เพิ่มขึ้น ในแบคทีเรียสายพันธุ์คัดทั้ง 2 สายพันธุ์ การดูดซับโลหะทองแดงโดยสารสกัดเอคโซโพลีแซคคาไรด์กับเซลทั้งหมดมีประสิทธิภาพสูง และไม่แตกต่างกันมากนัก การดูดซับโลหะทองแดงโดยเอคโซโพลีแซคคาไรด์ที่อบให้แห้งให้ประสิทธิภาพมากกว่า 80 เปอร์เซ็นต์ และ สามารถใช้ในการดูดซับโลหะชนิดอื่น (สังกะสี, แมงกานีสและแคดเมียม) ในแบคทีเรียสายพันธุ์คัดทั้ง 2 สายพันธุ์ การซะเอาโลหะออก จากสารสกัดเอคโซโพลีแซคคาไรด์สามารถทำได้โดยใช้กรดไฮโดรคลอริกและสามารถนำสารสกัดเอคโซโพลีแซคคาไรด์ไปใช้ใหม่ได้

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Two strains of 350 strains of copper-resistant bacterial isolates were selected and named CuR-38 and CuR-40. Both of them resisted to 700 μ g/ml Cu and EPS-producing. By some identification test, they might be classified as *Zoogloea* sp. and *Bacillus* sp., respectively. Optimum pH and temperature for growth and EPS production were 7 and 37°C. EPS production appeared maximally during the stationary phase (48 hr.). Addition of Cu induction was not induced EPS production in both bacterial isolates. The copper accumulation capacities between wet EPS and wet whole cell is highest, but not much different. The percentage of Cu accumulation by dried EPS was greater than 80% of both strains. The dried EPS has been used to remove other metal (zinc, manganese and cadmium). The release of metal from EPS achieved by HCl and can regenerated to reuse in the future.

ภาควิชา	Environmental	Science	ลายมือชื่อนิสิต	TORPOW	CHANN	ARONG
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ABBREVATION AND SYMBOL

Ag = Silver

 $AgNO_3$ = Silver Nitrate

 Ag_2S = Silver sulfide

A.N. = Atomic Number

BOD = Biochemical oxygen demand

B.P. = Boiling Point

Cd = Cadmium

 $CdCl_2.H_2O$ = Cadmium Chloride

cm = Centimeter

Cr = Chromium

Cu = Copper, Chalcite

 Cu^{2+} = Cupric ion

 $Cu(C_2H_3O_2)_2.H_2O =$ Copper acetate

 $CuCl_2$ = Copper chloride

 $Cu(CN)_2$ = Copper cyanide

 $CuCO_3.Cu(OH_2)$ = Copper carbonate

 $2CuCO_3.Cu(OH)_2 = Azurite$

CuO = Copper oxide

 Cu_2O = Cuprite

CuR- = Copper-Resistant Bacterial

Isolates

CuS = Covellite

 Cu_2S = Cuprous sulfide, Chalcocite

 $CuSO_4.5H_2O$ = Copper Sulfate, Chalcanthite

°C = Degree Celsius

Exopolysaccharide or

Extracellular polysaccharide

or Exopolymer

g = Gram

g/L = Gram/liter

Hg = Mercury

 HNO_3 = Nitric acid

 H_2S = Hydrogen sulfide

 H_2SO_4 = Sulfuric acid

KCN = Potassium cyanide

 K_2CrO_4 = Potassium Chromate

Kg = Kilogram

Kb = Kilobase

L = Liter

LPS = Lipopolysaccharide

m = Meter

mg = Milligram

mg/kg = Milligram/kilogram

mg/l = Milligram/liter

min = Minute

ml = Milliliter

MLVSS = Mixed liquor volatile

suspended solids

mmol = Millimole

Mn = Manganese

 $MnSO_4.H_2O$ = Manganese Sulfate

mol = Mole

M.P. = Melting Point

M.W. = Molecular Weight

μg = Microgram

 $\mu g/g$ = Microgram/gram

 $\mu g/L$ = Microgram/liter

 $\mu g/ml$ = Microgram/milliliter

 μ mol = Micromole

 $\mu mol/L$ = Micromole/liter

μmol/mg = Micromole/milligram

 NH_4Cl = Ammonium chloride

Ni = Nickel

 $NiSO_4.H_2O$ = Nickel Sulfate

nm = Nanometer

ppb = Parts per Billion

ppm = Parts per Million

Temp = Temperature

TSA = Tryptic Soy Agar

TSB = Tryptic Soy Broth

t/yr = Ton/year

U = Uranium

V = Volume

W = Weight

Zn = Zinc

ZnS = Sphalerite

 $ZnSO_4.7H_2O = Zinc Sulfate$