

REFERENCES

- Antony, P., Bandyopadhyay, S., and De, S.K. (2000). Synergism in properties of ionomeric polyblends based on zinc salts of carboxylated nitrile rubber and poly(ethylene-co-acrylic acid). Polymer, 41, 787-793
- Clagett, D.C. Engineering plastic. Encyclopedia of Polymer Science and Engineering, 6, 121-131
- Feng, Y., Schmidt, A., and Weiss, R.A. (1996). Compatibilization of polymer blends by complexation. 1. Spectroscopic characterization of ion-amide interactions in ionomer/polyamide blends. Macromolecules, 29, 3909-3917
- Fox, D.W., and Allen, R.B. Compatibility. Encyclopedia of Polymer Science and Engineering, 3, 758-775
- Gadekar, R., Kulkarni, A., and Jog, J.P., (1998). Blends of nylon with polyethylene: effect of compatibilization on mechanical and dynamic mechanical properties. Journal of Applied Polymer Science, 69, 161-168
- Gebel, G., and Loppinet, B. (1997). Small-angle neutron scattering study of polyethylene-co-methacrylate ionomer aqueous solutions. Journal of Physical Chemistry B., 101, 3980-3987
- Grady, B.P. (2000). Relative size of ionic aggregates determined by X-ray absorption spectroscopy. Polymer, 41, 2325-2328
- Grady, B.P., Floyd, J.A., Genetti, W.B., Vanhoorne, P., and Register, R.A. (1999). X-ray absorption spectroscopy studies of zinc-neutralized ethylene-methacrylic acid ionomers. Polymer, 40, 283-288
- James, D.E., Low density polyethylene (high pressure). Encyclopedia of Polymer Science and Engineering, 6, 386-454

- Kelar, K., and Jurkowski, B. (2000). Preparation of functionalised low-density polyethylene by reactive extrusion and its blend with polyamide 6. *Polymer*, 41, 1055-1062
- Li, H., Chiba, T., Higashida, N., Yang, Y., and Inoue, T. (1997). Polymer-polymer interface in polypropylene/polyamide blends by reactive processing. *Polymer*, 38(15), 3921-3925
- Lim, S., and White, J.L., (1994). Influence of a compatibilizing agent on the phase morphology of a polyethylene-polyamide 6 blend in a modular intermeshing co-rotating twin screw extruder. *Journal of Polymer Engineering and Science*, Mid-Feb, 34(3), 221-228
- Loveless, H.K. (1984). Ionomer. *Modern Plastics Encyclopedia*, 70-71
- Lundberg, R.D. Ionic polymer. *Encyclopedia of Polymer Science and Engineering*, 8, 393-423
- Morgan, R.L., Hill, M.J., and Barham, P.J. (1999). Morphology, melting behaviour and co-crystallization in polyethylene blends: the effect of cooling rate on two homogeneously mixed blends. *Polymer*, 40, 337-348
- Neoh, K.G., Tay, B.K., and Kang, E.T. (2000). Oxidation and ion migration during synthesis and degradation of electroactive polymer-nylon 6 composite films. *Polymer*, 41, 9-15
- Ohisson, B., Hassander, H., and Tornell, B. (1998). Improved compatibility between polyamide and polypropylene by the use of maleic anhydride grafted SEBS. *Polymer*, 39(26), 6705-6714
- Paul, D.R., Barlow, J.W., and Keskkula, H., Polymer blends, *Encyclopedia of Polymer Science and Engineering*, 12, 399-461
- Skrovanek, D.J., Howe, S.E., Painter, P.C., and Coleman, M.M. (1985). Hydrogen bonding in polymers: infrared temperature studies of an amorphous polyamide. *Macromolecules*, 18, 1676-1683

- Supapol, P. (1999). Methods for the determination of equilibrium melting temperature of semi-crystalline polymers: a review. Journal of Engineering Kasetart University, 38, 41-51
- Tomba, J.P., Puente, E., and Pastor, J.M. (2000). Calculation of polymer blend compositions from raman spectra: a new method based on parameter estimation techniques. Journal of Polymer Science: Part B: Polymer Physics: 38, 1013-1023
- Vankan R., Teyssie Ph., and Jerome R. (1997). Design of polymer blend rheology: 4. effect of polymethacrylic ionomers on the melt viscosity of polyamide mxD,6. Polymer, 38(15), 3861-3867
- Vernon, J. (1992). Testing of Materials. Hong Kong: Macmillan
- Whelan, T. Polymer Technology Dictionary. Chapman and Hall
- Willis, J.M., and Favis, B.D. (1988). Processing-morphology relationships of compatibilized polyolefin/polyamide blends. Part I: the effect of an ionomer compatibilizer on blend morphology. Polymer Engineering and Science, Mid-November, 28(21), 1416-1426
- Zhou, X., Goh, S.H., Lee, S.Y., and Tan, K.L. (1998). APS and FTIR studies of the interactions in poly(carboxylic acid)/poly(vinylpyridine) complexes. Polymer, 39(16), 3631-3640

APPENDIX A

Table A1 Tensile strength of nylon 6/LDPE blends (without compatibilizer).

Blend composition (nylon 6:LDPE)	0.8:0.2	0.6:0.4	0.4:0.6	0.2:0.8
Tensile strength (MPa)	22.3152	19.0539	10.8809	9.0128

Table A2 Tensile strength of nylon 6/ LDPE blends (with compatibilizer).

Surlyn (%)	Nylon 6:LDPE 0.8:0.2	Nylon 6: LDPE 0.6:0.4	Nylon 6:LDPE 0.4:0.6	Nylon 6:LDPE 0.2:0.8
0.0	22.3152	19.0539	10.5059	9.0128
0.1	24.1762	15.5943	13.4223	9.3148
0.5	30.9013	20.6966	16.8343	10.0900
1.0	37.4830	22.3738	20.0096	10.6044
2.5	39.9093	23.3651	20.2414	10.2783
3.5	42.4245	24.5020	21.1041	16.0391
5.0	50.2236	27.1184	28.2518	17.3687
10.0	53.1261	31.1802	31.5508	20.7611
15.0	53.3407	33.1028	34.3257	24.2092
25.0	56.0036	35.7210	35.6895	26.2404
35.0	55.5188	37.8798	36.5173	27.5689

Table A3 Tensile modulus of nylon 6/LDPE blends (without compatibilizer).

Blend composition (nylon 6:LDPE)	0.8:0.2	0.6:0.4	0.4:0.6	0.2:0.8
Tensile modulus (MPa)	440.7501	422.7666	309.4116	232.6506

Table A4 Tensile modulus of nylon 6/ LDPE blends (with compatibilizer).

Surlyn (%)	Nylon 6:LDPE 0.8:0.2	Nylon 6:LDPE 0.6:0.4	Nylon 6:LDPE 0.4:0.6	Nylon 6:LDPE 0.2:0.8
0.0	440.7501	422.7666	309.4116	232.6506
0.1	462.3784	460.5826	478.9633	278.7642
0.5	489.3908	480.6399	495.5820	285.1813
1.0	513.8695	494.0600	577.4787	294.4807
2.5	527.5716	522.5641	617.6398	295.2549
3.5	536.2504	554.0303	627.9969	309.6579
5.0	570.5046	577.5322	666.6833	305.6296
10.0	580.5721	595.2834	659.8730	310.7202
15.0	618.0514	615.5441	664.6011	312.6351
25.0	648.2263	626.2108	661.7623	321.5476
35.0	668.7946	661.0136	664.4730	321.8743

Table A5 Impact strength of nylon 6/LDPE blends (without compatibilizer).

Blend composition (nylon 6:LDPE)	0.8:0.2	0.6:0.4	0.4:0.6	0.2:0.8
Impact strength (kJ/m ²)	5.11	6.35	9.40	13.73

Table A6 Impact strength of nylon 6/ LDPE blends (with compatibilizer).

Surlyn (%)	Nylon 6:LDPE	Nylon 6:LDPE	Nylon 6:LDPE
	0.8:0.2	0.6:0.4	0.4:0.6
0.0	6.35	9.40	13.73
0.1	6.97	10.25	15.34
0.5	7.08	10.82	16.77
1.0	9.02	11.29	17.59
2.5	25.39	11.26	18.04
3.5	58.11	11.74	18.68
5.0	78.89	12.61	23.83
10.0	81.00	26.54	64.90
15.0	87.58	64.60	95.20
25.0	-	84.40	106.79
35.0	-	90.74	-

APPENDIX B

Table B1 Hardness of Nylon 6/LDPE Blends (without compatibilizer).

Blend composition (nylon 6:LDPE)	0.8:0.2	0.6:0.4	0.4:0.6	0.2:0.8
Hardness (Shore D)	56.4000	42.9545	42.9545	35.0500

Table B2 Hardness of Nylon 6/ LDPE Blends (with compatibilizer).

Surlyn (%)	Nylon 6:LDPE 0.8:0.2	Nylon 6: LDPE 0.6:0.4	Nylon 6:LDPE 0.4:0.6	Nylon 6:LDPE 0.2:0.8
0.0	56.4000	58.1154	42.9545	35.0500
0.1	57.0455	57.8214	46.9545	41.2000
0.5	57.7500	59.0714	48.7273	43.0000
1.0	57.7941	60.9667	52.3185	47.0385
2.5	59.6000	62.6000	52.5455	47.5500
3.5	60.9667	63.3929	53.2273	50.8500
5.0	62.8750	66.5333	53.9167	55.0500
10.0	63.7143	67.7500	57.6154	56.1500
15.0	64.5385	68.2500	63.2727	57.0909
25.0	64.5357	69.9722	65.4167	57.8636
35.0	66.8071	72.5833	67.6667	58.4091

APPENDIX C

Table C1 Particle Size of Dispersed Phase of Nylon 6/LDPE Blends (without compatibilizer).

Blend composition (nylon 6:LDPE)	0.8:0.2	0.6:0.4	0.2:0.8
Particle size (μm)	7.1831	5.5134	7.7942

Table C2 Particle Size of Dispersed Phase of Nylon 6/ LDPE Blends (with compatibilizer).

Surlyn (%)	Nylon 6:LDPE 0.8:0.2	Nylon 6: LDPE 0.6:0.4	Nylon 6:LDPE 0.2:0.8
	7.1831	5.5134	7.7942
0.0	2.6623	1.1526	2.6250
0.5	1.4805	0.6626	1.5641
1.0	1.2612	0.6723	1.3824
2.5	1.0389	1.7143	0.6500
3.5	0.8242	0.8283	0.6000

CURRICULUM VITAE

Name: Rungravee Pattanaolarn

Date of Birth: Febuary 10, 1974

Nationality: Thai

University Education:

1995-1999 Bachelor Degree of Science in Chemistry

Thammasat University