

การสังเคราะห์สารเรื่องโยงบางชนิดที่มีหมู่ฟังก์ชันหลายหมู่สำหรับพอลิยูรีเทน

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**SYNTHESIS OF SOME MULTIFUNCTIONAL CROSSLINKING AGENTS  
FOR POLYURETHANE**

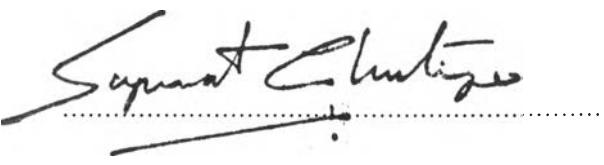
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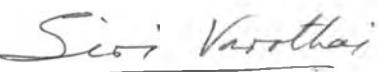
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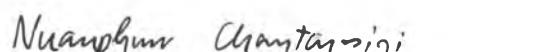
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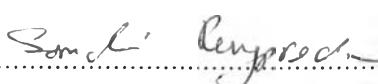
  
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พิมพ์ต้นฉบับทัศนศิลป์อวิทยานิพนธ์ภาษาไทยในกรอบสีเขียวนี้เพียงแผ่นเดียว

นารี พากะจ่าง : การสังเคราะห์สารเชื่อมโยงบางชนิดที่มีหมุนฟังก์ชันหลายหมู่สำหรับพอลิยูรีเทน  
(SYNTHESIS OF SOME MULTIFUNCTIONAL CROSSLINKING AGENTS FOR POLYURETHANE)  
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สารประกอบ 5 ตัว คือ Bis-(3-allyloxy-2-hydroxy-1-propoxy)diphenoxyp propane (3), Bis-[3-(4-allyloxy-2-methoxy)phenoxy-2-hydroxy-1-propoxy]diphenoxyp propane (7), 2,7-Bis-(3-allyloxy-2-hydroxy-1-propoxy)naphthalene (9), Bis-(3-N,N-diallylamino-2-hydroxy-1-propoxy)diphenoxyp propane (11) และ N,N,N',N'-tetrakis-(3-allyloxy-2-hydroxy-1-propoxy)ethylenediamine (13) เตรียมได้จากการเปิดวงอีพอกไซด์ด้วยนิวคลีโอไฟล์ สารประกอบ (7) และ (11) เตรียมได้จากปฏิกิริยาระหว่าง diglycidyl ether of bisphenol A (6) กับ eugenol (5) และ diallylamine (10) ตามลำดับ สารประกอบ (3), (9), และ (13) เตรียมได้จากปฏิกิริยาระหว่าง allyl glycidyl ether (2) กับ bisphenol-A (1), 2,7-dihydroxynaphthalene (8) และ ethylenediamine (12) ตามลำดับ สารประกอบเหล่านี้ประกอบด้วยหมู่ไฮดรอกซีและหมู่ไวนิลหลายหมู่ ซึ่งสามารถพิสูจน์เอกลักษณ์โดยเทคนิค <sup>1</sup>H NMR และ <sup>13</sup>C NMR สเปกตรอสโกรี, อินฟารेड สเปกตรอสโกรี, แมสสเปกตรومetri และการวิเคราะห์ปริมาณธาตุองค์ประกอบ จากการศึกษาปฏิกิริยาของหมู่ไฮดรอกซีและหมู่ไวนิลของสารประกอบ (3), (9), (11) และ (13) พบว่าหมู่ไฮดรอกซีของสารประกอบ (3), (9), (11) และ (13) สามารถเกิดปฏิกิริยากับหมู่ไฮโซเดนต์ของ MDI ได้เป็นพันธะยูรีเทน เชื่อมโยง และหมู่ไวนิลของสารประกอบ (3), (9), (11) และ (13) สามารถเกิดปฏิกิริยาแบบพรีแอดดิคัล เมื่อมี benzoyl peroxide อยู่ด้วย จากผลดังกล่าวสรุปได้ว่า อาจนำสารประกอบ (3), (9), (11) และ (13) มาใช้เป็นสารเชื่อมโยงได้สำหรับการเตรียมพอลิยูรีเทน อีเลสโตเมอร์

ภาควิชา ..... เคมี  
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CROSSLINKING AGENTS FOR POLYURETHANE. THESIS ADVISOR :  
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Five compounds, Bis-(3-allyloxy-2-hydroxy-1-propoxy)diphenoxyp propane (3), Bis-[3-(4-allyloxy-2-methoxy)phenoxy-2-hydroxy-1-propoxy]diphenoxyp propane (7), 2,7-Bis-(3-allyloxy-2-hydroxy-1-propoxy)naphthalene (9), Bis-(3-N,N-diallylamino-2-hydroxy-1-propoxy)diphenoxyp propane (11) and N,N,N',N'-tetrakis-(3-allyloxy-2-hydroxy-1-propoxy)ethylenediamine (13) were synthesized by ring opening of epoxides with nucleophiles. Compounds (7) and (10) were synthesized by the reaction of diglycidyl ether of bisphenol A (6) with eugenol (5) and diallylamine (10), respectively. Compounds (3), (9), and (13) were obtained from the reaction of allyl glycidyl ether (2) with bisphenol-A (1), 2,7-dihydroxynaphthalene (8) and ethylenediamine (12), respectively. These compounds, which contain multiple hydroxyl and vinyl groups, were characterized by <sup>1</sup>H NMR and <sup>13</sup>C NMR spectroscopy, mass spectrometry, infrared spectroscopy and elemental analysis. The reactivity of the hydroxyl and vinyl groups of compounds (3), (9), (11), and (13) were studied. It was found that the hydroxyl groups of compounds (3), (9), (11), and (13) could react with the isocyanate group of MDI to give a urethane linkage. It was also found that the vinyl groups of compounds (3), (9), (11), and (13) can undergo free radical reactions in the presence of benzoyl peroxide. These results suggest that compounds (3), (9), (11), and (13) may be used as crosslinking agents in the preparation of polyurethane elastomers.

ภาควิชา เคมี

ลายมือชื่อนิสิต 

สาขาวิชา เคมี

ลายมือชื่ออาจารย์ที่ปรึกษา  ๑๒๗๐๙๖

ปีการศึกษา ๒๕๔๐

ลายมือชื่ออาจารย์ที่ปรึกษาร่วม



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## LIST OF ABBREVIATIONS

Analysis calc.	Analysis calculation
$^{13}\text{C}$ NMR	Carbon-13 Nuclear Magnetic Resonance
$\text{cm}^{-1}$	unit of wavenumber
d	doublet
dd	double doublet
DGEBA	Diglycidyl ether of bisphenol A
DSC	Differential Scanning Calorimetry
FT-IR	Fourier Transform Infrared Spectroscopy
$^1\text{H}$ NMR	Proton Nuclear Magnetic Resonance
IPNs	Interpenetrating Polymer Networks
IR	Infrared
J	coupling constant
m	multiplet
$\text{M}^+$	Molecular ion in mass spectrum
MDI	4,4'-diphenylmethane diisocyanate
MS	Mass spectrometry
ppm	parts per million
$R_f$	retention factor in chromatography
s	singlet
tdd	triple double doublet
TLC	Thin Layer Chromatography
TGA	Thermogravimetric Analysis
$\delta$	Chemical Shift