#### CARBON DIOXIDE GASIFICATION OF CELLULOSE

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A Thesis Submitted in Partial Fulfilment of the Requirements for the Degree of Master of Science The Petroleum and Petrochemical College, Chulalongkorn University in Academic Partnership with The University of Michigan, The University of Oklahoma, Case Western Reserve University and Institut Français du Pétrole

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#### ABSTRACT

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Keywords: Biomass/ Carbon dioxide/ Cellulose/ Gasification/ Dry reforming

Biomass can be converted to gaseous products via the gasification process. Major components of biomass are cellulose, hemi-cellulose and lignin. This work focused on the gasification of cellulose using CO<sub>2</sub> as the gasifying agent. Cellulose was gasified at specified temperatures ranging from 600 to 900 °C with ZSM-5, Fe, Co, and Co-Fe bimetallic as catalysts. The main gasification products were CO, CH<sub>4</sub> and H<sub>2</sub>. An increase in gasifying temperature led to higher total gas yield and lower liquid yield. The use of CO<sub>2</sub> produced CO rich gas products. When steam was used as the gasifying agent, H<sub>2</sub> was produced more, while CO was produced less. The use of catalysts produced more CO and H<sub>2</sub> yields in the gas phase , and reduced the amount of liquid yield. Some light hydrocarbons such as ethylene, ethane, and propane were also found in the gaseous products. The liquid products from both CO<sub>2</sub> and steam gasification contained mainly aliphatic and alicyclic compounds, alcohols, esters, nitrogenated and oxygenated compounds.

# บทคัดย่อ

วสันต์ ชีวาสุขถาวร : กระบวนการแกสิฟิเคชั่นของเซลลูโลสในสภาวะที่มีแก๊ซ การ์บอนไดออกไซด์ (Carbon dioxide Gasification of Cellulose) อ. ที่ปรึกษา : ผศ.คร. บุนขรัชต์ กิติขานันท์, ศ.คร. โจฮานเนส ชวางค์, รศ.คร. วิษณุ มีอยู่, รศ.คร. ปราโมช รังสรรค์วิจิตร และ รศ. คร. ธีรศักดิ์ ฤกษ์สมบูรณ์ 96 หน้า ISBN 974-9937-07-4

งานวิจัยนี้ศึกษาถึงการแปรรูปของเซลลูโลสเป็นผลิตภัณฑ์ที่มีจุณค่าเพิ่มจาก กระบวนการแกสิฟิเคชันในสภาวะที่มีแก๊ซการ์บอนไดออกไซด์และใช้ตัวเร่งปฏิกิริยาได้แก่ เหล็ก, โคบอลท์, โลหะผสมของเหล็กกับโคบอลท์ และ ZSM-5 การศึกษานี้ได้ศึกษาถึงผลของอุณหภูมิ ดัวเร่งปฏิกิริยา และไอน้ำ จากการศึกษาพบว่าผลิตภัณฑ์หลักเกิดในรูปของก๊าซได้แก่ ไฮโดรเจน การ์บอนมอนอกไซด์ และมีเทน การเพิ่มอุณหภูมิส่งผลให้ปริมาณผลิตภัณฑ์ในรูปก๊าซเพิ่มขึ้น ในขณะที่ผลิตภัณฑ์ในรูปของเหลว และของแข็งลดลง สำหรับกระบวนการแกสิฟิเคชันภายใด้ บรรยากาศของก๊าซการ์บอนไดออกไซด์จะให้สัดส่วนของก๊าซการ์บอนมอนอกไซด์มากที่สุด ในขณะที่กระบวนการแกสิฟิเคชันภายใต้บรรยากาศของไอน้ำจะให้สัดส่วนของก๊าซไฮโครเจน มากที่สุด ผลิตภัณฑ์ในรูปของเหลวจะลดลงเมื่อใช้ตัวเร่งปฏิกิริยามีส่วนช่วยส่งเสริมการเกิด ปฏิกิริยารีฟอร์มมิ่งของผลิตภัณฑ์ที่เป็นของเหลว จากการศึกษายังพบว่ามีสารไฮโครคาร์บอน น้ำหนักเบาเช่น เอธิลีน อีเทน และ โพรเพนเกิดขึ้นในผลิตภัณฑ์ที่เป็นก๊าซ สำหรับผลิตภัณฑ์ที่เป็น ของเหลวจะประกอบไปด้วยสารประกอบหลายประเภทเช่น อะลิฟาติก, อะลิไซคลิก, เอสเทอร์, แอลกฮอล์, สารประกอบที่มีออกซิเจน และสารประกอบที่มีในโตรเจน

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