

REFERENCES

- Arayawongkul, S. (2002) Characterization of polystyrene produced by admicellar polymerization. M.S. Thesis, The Petroleum and Petrochemical College, Chulalongkorn University.
- Aumsuwan, N. (2003) Characterization of polystyrene formed via admicellar polymerization: the effect of initiator concentration. M.S. Thesis, The Petroleum and Petrochemical College, Chulalongkorn University.
- Chaisirimahamorakot, S. (2001) Modification of silica surface for rubber reinforcement using a continuous admicellar polymerization system. M.S. Thesis, The Petroleum and Petrochemical College, Chulalongkorn University.
- Chinpan, N. (1996) Comparison of rubber reinforcement using various surface modified silicas. M.S. Thesis, The Petroleum and Petrochemical College, Chulalongkorn University.
- Cross, J., and Singer, E.J. (1994) Cationic Surfactants (Analytical and Biological Evaluation). New York: Marcel Dekker, Inc.
- Dickson, J., and O'Haver, J.H. (2002) Adsolubilization of naphthalene and α -naphthol in C_nTAB admicelles. Langmuir, 18, 9171-9176.
- Harwell, J.H., Hoskins, J.C., Schechter, R.S., and Wade, W.H. (1985) Pseudophase separation model for surfactant adsorption: isomerically pure surfactants. Langmuir, 1, 251-262.
- Iler, R.K. (1979) The Chemistry of Silica. New York: John Wiley and Sons Inc.
- Grady, B.P., O'Rear, E.A., Penn, L.S., and Pedicini, A. (1998) Polymerization of styrene-isoprene on glass cloth for use in composite manufacture. Polymer Composites, 19(5), 579-587.
- Kitiyanan, B., O'Haver, J.H., Harwell, J.H., and Osuwan, S. (1996) Adsolubilization of styrene and isoprene in cetyltrimethylammonium bromide admicelle on precipitated silica. Langmuir, 12, 2162-2168.
- Kudisri, R. (1997) Comparison of surface modified fillers to clay for natural rubber composites. M.S. Thesis, The Petroleum and Petrochemical College, Chulalongkorn University.

- Lai, C.L., Harwell, J.H., and O'Rear, E.A. (1995) Formation of poly(tetrafluoroethylene) thin films on alumina by admicellar polymerization. *Langmuir*, 11, 905-911.
- Nonthasorn, P. (2002) Improvement of natural rubber properties by modification of silica surface using a continuous admicellar polymerization system. *M.S. Thesis*, The Petroleum and Petrochemical College, Chulalongkorn University.
- O'Haver, J.H., Harwell, J.H., O'Rear, E.R., Snodgrass, L.J., and Waddell, W.H. (1994) In situ formation of polystyrene in adsorbed surfactant bilayers on precipitatedsilica. *Langmuir*, 10, 2588-2593.
- O'Haver, J.H., Harwell, J.H., Evans, L.R., and Waddell, W.H. (1996) Polar copolymer-modified precipitated silica. *Journal of Applied Polymer Science*, 59, 1427-1435.
- Rosen, M.J. (1989) *Surfactants and Interfacial Phenomena*. 2nd edition. New York: John Wiley and Sons, Inc.
- Rubingh, D.N., and Holland. P.M. (1990) *Cationic Surfactants (Physical Chemistry)*. New York: Marcel Dekker, Inc.
- Scamehorn, J.F., Schechter, R.S., and Wade, W.H. (1982) Adsorption of surfactants on mineral oxide surfaces from aqueous solution. *J. Colloid and Interface Science*, 85(2), 463-477.
- Scamehorn, J.F., and Harwell, J.H. (1988) Surfactant-based treatment of process streams. *In Surfactant in Chemical/Process Engineering*. New York: Marcel Dekker, Inc.
- See, C.H., and O'Haver, J.H. (2003) Atomic force microscopy studies of admicellar polymerization polystyrene-modified amorphous silica. *Journal of Applied Polymer Science*, 87, 290-299.
- See, C.H., and O'Haver, J.H. (2003) Atomic force microscopy characterization of ultrathin polystyrene films formed by admicellar polymerization on silica disks. *Journal of Applied Polymer Science*, 89, 36-46.
- See, C. H. and O'Haver, J. H. (2004) Two-dimensional phase transition of styrene adsolubilized in cetyltrimethylammonium bromide admicelles on mica. *Colloids and Surfaces A: Physicochem. Eng.*, 243, 169-183.

- Thammathadanukul, V., O'Haver, J.H., Harwell, J.H., Osuwan, S., Na-Ranong, N., and Waddell, W.H. (1996) Comparison of rubber reinforcement using various surface-modified precipitated silicas. Journal of Applied Polymer Science, 59, 1741-1750.
- Thakulsukanant, C., Lobban, L.L., Osuwan, S., and Waritswat, A. (1997) Adsolubilization and stability characteristics of hydrocarbon aggregates chemically bonded to porous silica. Langmuir, 13, 4595-4599.
- Waddell, W.H., O'Haver, J.H., Evans, L.R., Harwell, J.H. (1995) Organic polymer-surface modified precipitated silica. Journal of Applied Polymer Science, 55, 1627-1641.
- Wu, J., Harwell, J.H., and O'Rear, E.A. (1987) Two-dimensional reaction solvents: surfactant bilayers in the formation of ultrathin films. Langmuir, 3, 531-537.
- Yeskei, M.A., and Harwell, J.H. (1988) On the structure of aggregates of adsorbed surfactant: The surface charge density at the hemimicell/admicell transition. Physical Chemistry, 92, 2346-2352.

APPENDICES

APPENDIX A CTAB Adsorption Measurement

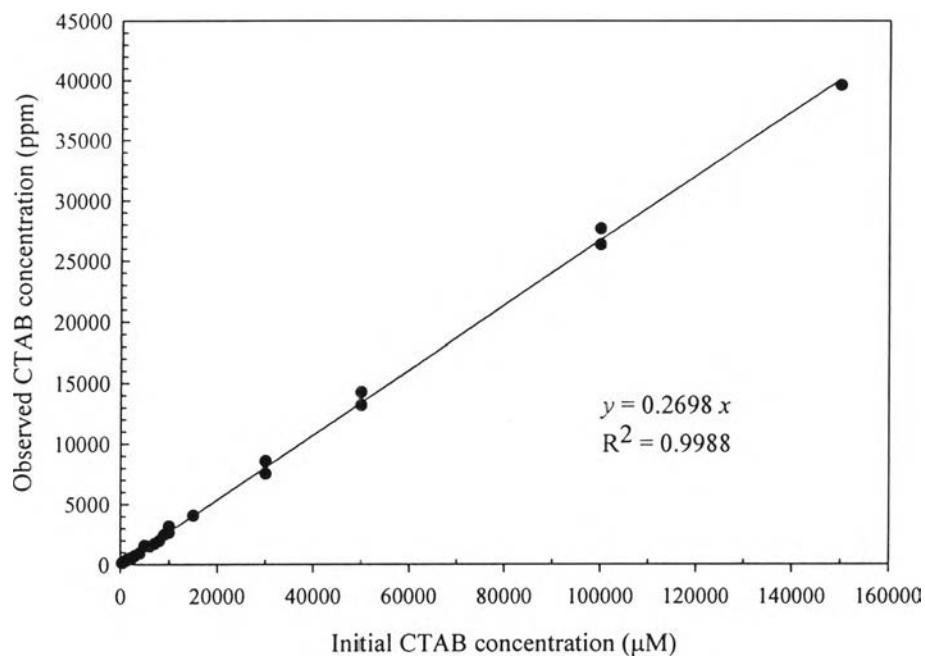


Figure A1 Calibration curve of CTAB solution by Total Organic Carbon analyzer (TOC).

Table A1 Data from CTAB adsorption isotherm on Aerosil® OX50

Initial CTAB concentration (μM)	Observed initial CTAB concentration, (μM)	Equilibrium CTAB concentration (μM)	CTAB adsorption (μmol/g)
400	566.42	266.38	3.75
600	743.96	278.69	5.82
800	927.06	282.39	8.06
1400	1458.19	278.47	14.75
1600	1643.88	279.28	17.06
1800	1821.05	285.80	19.19
2000	1977.84	293.96	21.05

Initial CTAB concentration (μM)	Observed initial CTAB concentration (μM)	Equilibrium CTAB concentration (μM)	CTAB adsorption (μmol/g)
2200	2031.95	268.01	22.05
2500	2139.81	328.84	72.44
3000	2660.19	355.52	92.19
4000	3475.61	488.95	119.47
4200	3738.40	570.13	126.73
4600	4057.15	964.86	123.69
4800	4327.72	1428.54	115.97
5000	5837.73	2436.69	136.04
6000	5606.45	1739.88	154.66
7000	6361.08	2610.16	150.04
8000	7262.49	3255.82	160.27
9000	9127.58	5072.72	162.19
15000	15057.89	11195.77	154.48
30000	31733.21	26266.20	218.68
50000	52804.37	48023.05	191.25
100000	102507.86	94724.31	311.34
150000	146836.99	137163.16	386.95

APENDIX B Styrene Adsolubilization Measurement

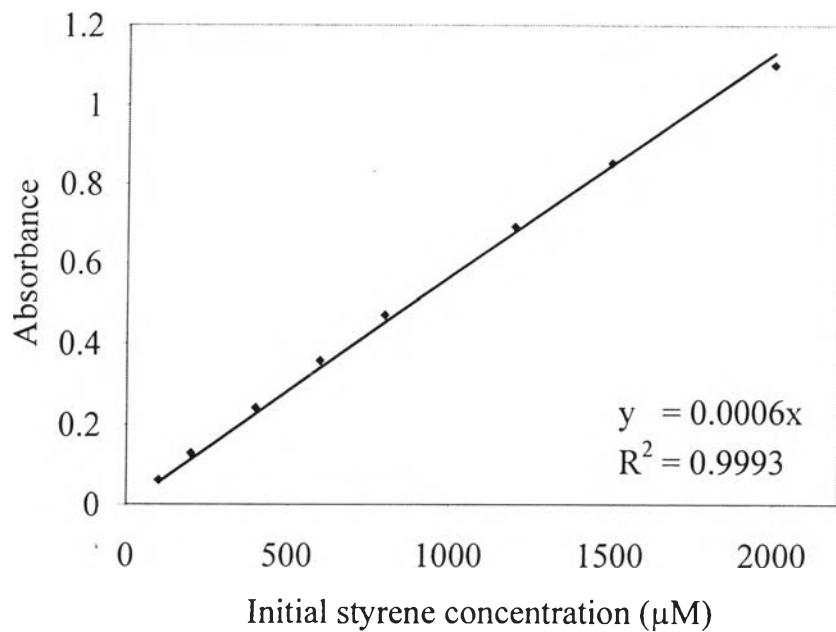


Figure B1 Calibration curve of styrene in CTAB solution by UV-Vis at 280cm^{-1} .

Ratio of mixture = Silica 0.5 g : 20 ml solution

Table B1 Data from styrene adsolubilization into CTAB adsorption 20 $\mu\text{mol/g}$ on Aerosil®OX50

Initial styrene concentration (μM)	Equilibrium styrene concentration (μM)	Styrene adsolubilization (μmol/g)
100	43.86	2.25
250	77.02	6.92
500	151.93	13.92
1000	376.11	24.96
1500	608.07	35.68
2000	785.44	48.58

Table B2 Data from styrene adsolubilization into CTAB adsorption 100 $\mu\text{mol/g}$ on Aerosil[®]OX50

Initial styrene concentration (μM)	Equilibrium styrene concentration (μM)	Styrene adsolubilization ($\mu\text{mol/g}$)
1000	315.33	27.39
2000	690.67	52.37
3000	809.33	87.63
4000	1119.67	115.21
5000	1485.67	140.57
7000	2190.00	192.40

APENDIX C Isoprene Adsolubilization Measurement

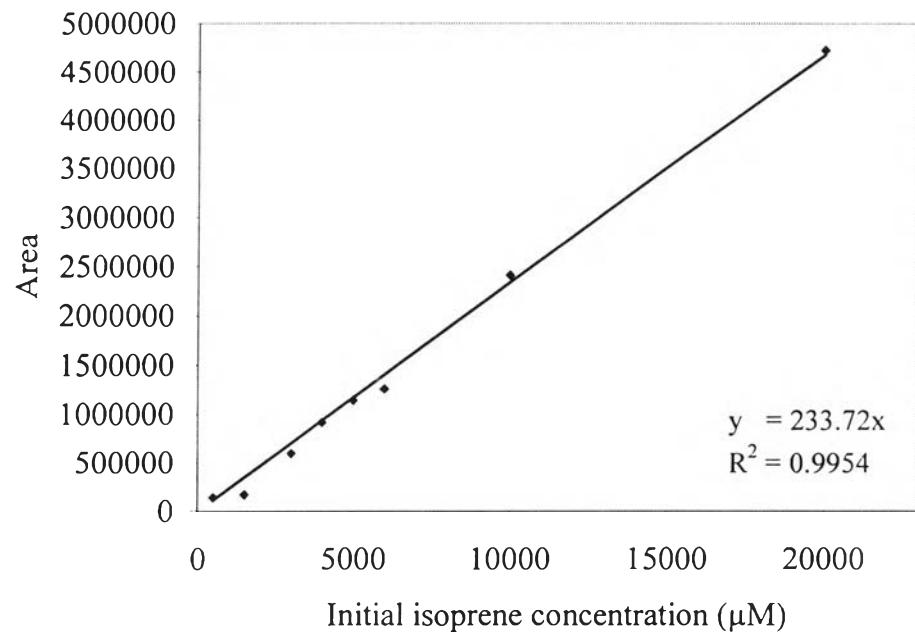


Figure C1 Calibration curve of isoprene in CTAB solution by headspace GC.

Table C1 Data from isoprene adsolubilization into CTAB adsorption 20 μmol/g on Aerosil®OX50

Initial isoprene concentration (μM)	Equilibrium isoprene concentration (μM)	Isoprene adsolubilization (μmol/g)
225	25.81	7.97
450	41.90	16.32
1800	134.39	66.62
3600	358.85	129.65
5400	462.22	197.51
7200	542.62	266.30

Table C2 Data from isoprene adsolubilization into CTAB adsorption 100 $\mu\text{mol/g}$ on Aerosil[®]OX50

Initial isoprene concentration (μM)	Equilibrium isoprene concentration (μM)	Isoprene adsolubilization ($\mu\text{mol/g}$)
1750	40.24	68.39
7000	165.02	273.40
14000	301.29	547.95
21000	442.21	822.31
28000	688.86	1092.45
31500	786.07	1228.56

APPENDIX D Calculation for Amount of CTAB Loading, Comonomer Loading, and AIBN Loading for Admicellar Polymerization

System; Silica 15 g : Solution 250 ml

CTAB

Molecular weight : 364.46 gmol⁻¹

Styrene

Molecular weight: 104.15 gmol⁻¹

Density: 0.906 ml/g

Isoprene

Molecular weight: 68.12 gmol⁻¹

Density: 0.681 ml/g

AIBN

Molecular weight : 164.21 gmol⁻¹

D1 CTAB Loading Calculation

Table D1 Calculation of initial CTAB concentration for CTAB adsorption 20 and 100 $\mu\text{mol/g}$ silica in the system

CTAB adsorption		Equilibrium CTAB concentration		Initial CTAB loading in the system (μmol)	Total weight of CTAB (g)
($\mu\text{mol/g}$)	($\mu\text{mol}/15\text{g}$)	(μM)	(μmol in 250 ml)		
20	300	300	75	375	0.1367
100	1500	400	100	1600	0.5831

D 2 Comonomer Loading Calculation

Ratio of styrene : isoprene = 1:3

Table D2 Calculation of initial styrene loading into CTAB adsorption 20 $\mu\text{mol/g}$ silica in the system at ratio of S:I=1:3

Styrene adsolubilization		Equilibrium styrene concentration		Initial styrene loading in the system (μmol)	Total volume of styrene (μl)
($\mu\text{mol/g}$)	($\mu\text{mol}/15 \text{ g}$)	(μM)	(μmol in 250 ml)		
5	75	80.52	20.13	95.13	10.94
10	150	161.03	40.26	190.26	21.87
15	225	241.55	60.39	285.39	32.81

Table D3 Calculation of initial isoprene loading into CTAB adsorption 20 $\mu\text{mol/g}$ silica in the system at ratio of S:I=1:3

Isoprene adsolubilization		Equilibrium isoprene concentration		Initial isoprene loading in the system (μmol)	Total volume of isoprene (μl)
($\mu\text{mol/g}$)	($\mu\text{mol}/15 \text{ g}$)	(μM)	(μmol in 250 ml)		
15	225	33.75	8.44	233.44	23.35
30	450	67.49	16.87	466.87	46.70
45	675	101.24	25.31	700.31	70.05

Table D4 Calculation of initial styrene loading into CTAB adsorption 100 $\mu\text{mol/g}$ silica in the system at ratio of S:I=1:3

Styrene adsolubilization		Equilibrium styrene concentration		Initial styrene loading in the system (μmol)	Total volume of styrene (μl)
($\mu\text{mol/g}$)	($\mu\text{mol}/15 \text{ g}$)	(μM)	(μmol in 250 ml)		
25	375	271.15	67.79	442.79	50.90
50	750	542.30	135.57	885.57	101.80
75	1125	813.45	203.36	1328.36	152.70

Table D5 Calculation of initial isoprene loading into CTAB adsorption 100 $\mu\text{mol/g}$ silica in the system at ratio of S:I=1:3

Isoprene adsolubilization		Equilibrium isoprene concentration		Initial isoprene loading in the system (μmol)	Total volume of isoprene (μl)
($\mu\text{mol/g}$)	($\mu\text{mol}/15 \text{ g}$)	(μM)	(μmol in 250 ml)		
75	1125	46.00	11.50	1136.50	113.68
150	2250	99.01	23.00	2273.00	227.37
225	3375	138.01	34.50	3409.50	341.05

Ratio of styrene : isoprene = 1:1

Table D6 Calculation of initial styrene loading into CTAB adsorption 20 $\mu\text{mol/g}$ silica in the system at ratio of S:I=1:1

Styrene adsolubilization		Equilibrium styrene concentration		Initial styrene loading in the system (μmol)	Total volume of styrene (μl)
($\mu\text{mol/g}$)	($\mu\text{mol}/15 \text{ g}$)	(μM)	(μmol in 250 ml)		
10	150	161.03	40.26	190.26	21.87
20	300	322.06	80.52	380.52	43.74
30	450	438.09	120.77	570.77	65.61

Table D7 Calculation of initial isoprene loading into CTAB adsorption 20 $\mu\text{mol/g}$ silica in the system at ratio of S:I=1:1

Isoprene adsolubilization		Equilibrium isoprene concentration		Initial isoprene loading in the system (μmol)	Total volume of isoprene (μl)
($\mu\text{mol/g}$)	($\mu\text{mol}/15 \text{ g}$)	(μM)	(μmol in 250 ml)		
10	150	22.50	5.62	155.62	15.57
20	300	44.99	11.25	311.25	31.13
30	450	67.49	16.87	466.87	46.70

Table D8 Calculation of initial styrene loading into CTAB adsorption 100 $\mu\text{mol/g}$ silica in the system at ratio of S:I=1:1

Styrene adsolubilization		Equilibrium styrene concentration		Initial styrene loading in the system (μmol)	Total volume of styrene (μl)
($\mu\text{mol/g}$)	($\mu\text{mol}/15 \text{ g}$)	(μM)	(μmol in 250 ml)		
50	750	542.30	135.57	885.57	101.80
100	1500	1084.60	271.15	1771.15	203.60
150	2250	1626.90	406.72	2656.72	305.41

Table D9 Calculation of initial isoprene loading into CTAB adsorption 100 $\mu\text{mol/g}$ silica in the system at ratio of S:I=1:1

Isoprene adsolubilization		Equilibrium isoprene concentration		Initial isoprene loading in the system (μmol)	Total volume of isoprene (μl)
($\mu\text{mol/g}$)	($\mu\text{mol}/15 \text{ g}$)	(μM)	(μmol in 250 ml)		
50	750	30.67	7.67	757.67	75.79
100	1500	61.34	15.33	1515.33	151.58
150	2250	92.01	23.00	2273.00	227.37

Ratio of styrene : isoprene = 3:1

Table D10 Calculation of initial styrene loading into CTAB adsorption 20 µmol/g silica in the system at ratio of S:I=3:1

Styrene adsolubilization		Equilibrium styrene concentration		Initial styrene loading in the system (µmol)	Total volume of styrene (µl)
(µmol/g)	(µmol/15 g)	(µM)	(µmol in 250 ml)		
15	225	241.55	60.39	285.39	32.81
30	450	483.09	120.77	570.77	65.61
45	675	724.64	181.16	856.16	98.42

Table D11 Calculation of initial isoprene loading into CTAB adsorption 20 µmol/g silica in the system at ratio of S:I=3:1

Isoprene adsolubilization		Equilibrium isoprene concentration		Initial isoprene loading in the system (µmol)	Total volume of isoprene (µl)
(µmol/g)	(µmol/15 g)	(µM)	(µmol in 250 ml)		
5	75	11.25	2.81	77.81	7.78
10	150	22.50	5.62	155.62	15.57
15	225	33.75	8.44	233.44	23.35

Table D12 Calculation of initial styrene loading into CTAB adsorption 100 µmol/g silica in the system at ratio of S:I=3:1

Styrene adsolubilization		Equilibrium styrene concentration		Initial styrene loading in the system (µmol)	Total volume of styrene (µl)
(µmol/g)	(µmol/15 g)	(µM)	(µmol in 250 ml)		
75	1125	813.45	203.36	1328.36	152.70
150	2250	1626.90	406.72	2656.72	305.41
225	3375	2440.35	610.09	3985.09	458.11

Table D13 Calculation of initial isoprene loading into CTAB adsorption 100 $\mu\text{mol/g}$ silica in the system at ratio of S:I=3:1

Isoprene adsolubilization		Equilibrium isoprene concentration		Initial isoprene loading in the system (μmol)	Total volume of isoprene (μl)
($\mu\text{mol/g}$)	($\mu\text{mol}/15\text{ g}$)	(μM)	(μmol in 250 ml)		
25	375	15.33	3.83	378.83	37.89
50	750	30.67	7.67	757.67	75.79
75	1125	46.00	11.50	1136.50	113.68

D 3 AIBN Loading Calculation

Ratio of AIBN = 1 mole AIBN : 25 mole comonomer

Ratio of styrene : isoprene = 1:3

Table D14 Calculation of AIBN loading at CTAB adsorption 20 $\mu\text{mol/g}$ silica at ratio of S:I=1:3

CTAB _{adsorp} : Comonomer _{adsol} ($\mu\text{mol/g}$) : ($\mu\text{mol/g}$)	Total comonomer (μmol)	AIBN loading (μmol)	Total weight AIBN (g)
1 : 1	328.57	13.14	0.00216
1 : 2	657.13	26.29	0.00432
1 : 3	985.70	39.43	0.00647

Table D15 Calculation of AIBN loading at CTAB adsorption 100 $\mu\text{mol/g}$ silica at ratio of S:I=1:3

CTAB _{adsorp} : Comonomer _{adsol} ($\mu\text{mol/g}$) : ($\mu\text{mol/g}$)	Total comonomer (μmol)	AIBN loading (μmol)	Total weight AIBN (g)
1 : 1	1579.29	63.17	0.01037
1 : 2	3158.58	126.34	0.02075
1 : 3	4737.87	189.51	0.03112

Ratio of styrene : isoprene = 1:1

Table D16 Calculation of AIBN loading at CTAB adsorption 20 $\mu\text{mol/g}$ silica at ratio of S:I=1:1

CTAB _{adsorp} : Comonomer _{adsol} ($\mu\text{mol/g}$) : ($\mu\text{mol/g}$)	Total comonomer (μmol)	AIBN loading (μmol)	Total weight AIBN (g)
1 : 1	345.88	13.83	0.00227
1 : 2	691.76	27.67	0.00454
1 : 3	1037.65	41.51	0.00682

Table D17 Calculation of AIBN loading at CTAB adsorption 100 $\mu\text{mol/g}$ silica at ratio of S:I=1:1

CTAB _{adsorp} : Comonomer _{adsol} ($\mu\text{mol/g}$) : ($\mu\text{mol/g}$)	Total comonomer (μmol)	AIBN loading (μmol)	Total weight AIBN (g)
1 : 1	1643.24	65.73	0.01079
1 : 2	3286.48	131.46	0.02159
1 : 3	4929.73	197.19	0.03238

Ratio of styrene : isoprene = 3:1

Table D18 Calculation of AIBN loading at CTAB adsorption 20 $\mu\text{mol/g}$ silica at ratio of S:I=3:1

CTAB _{adsorp} : Comonomer _{adsol} ($\mu\text{mol/g}$) : ($\mu\text{mol/g}$)	Total comonomer (μmol)	AIBN loading (μmol)	Total weight AIBN (g)
1 : 1	363.20	14.53	0.00239
1 : 2	726.40	29.06	0.00477
1 : 3	1089.60	43.58	0.00716

Table D19 Calculation of AIBN loading at CTAB adsorption 100 $\mu\text{mol/g}$ silica at ratio of S:I=3:1

CTAB _{adsorp} : Comonomer _{adsol} ($\mu\text{mol/g}$) : ($\mu\text{mol/g}$)	Total comonomer (μmol)	AIBN loading (μmol)	Total weight AIBN (g)
1 : 1	1707.20	68.29	0.01121
1 : 2	3414.39	136.58	0.02243
1 : 3	5121.59	204.86	0.03364

APENDIX E Data of Gel Permeation Chromatography

Table E1 Sample name for gel permeation analysis

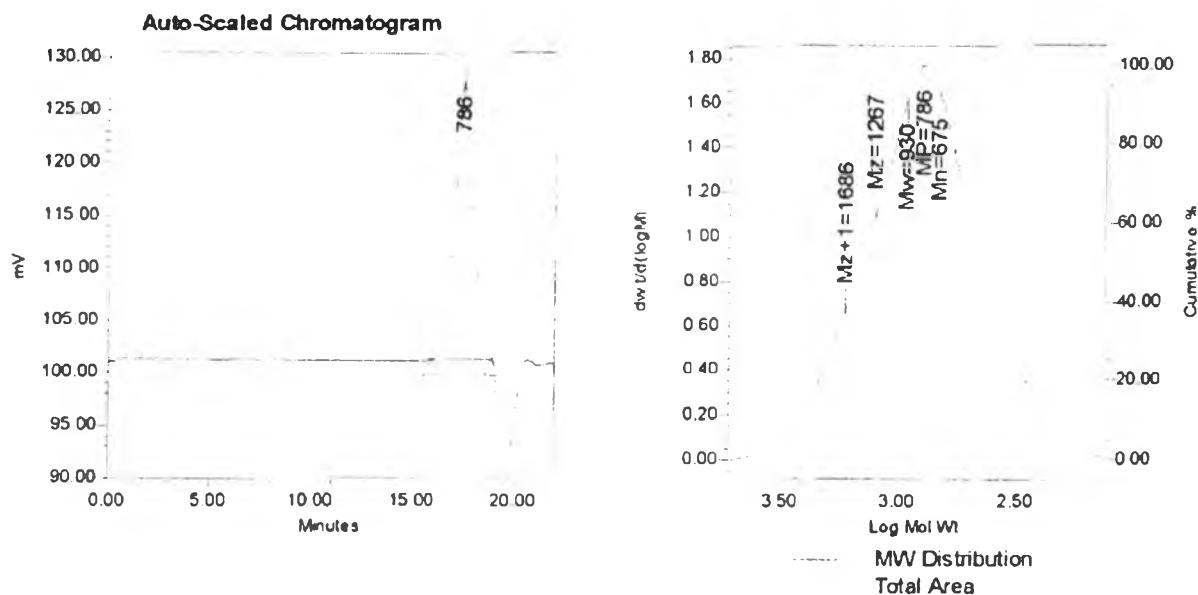
CTAB _{adsorp} : Comonomer _{adsol} (μmol/g) : (μmol/g)	CTAB adsorption (μmol/g of silica)	Comonomer adsolubilization (μmol/g of silica)	Ratio of styrene:isoprene	Sample name
1:1	20	20	1 : 3	1:3L1:1
			1 : 1	1:1L1:1
			3 : 1	3:1L1:1
1:2	20	40	1 : 3	1:3L1:2
			1 : 1	1:1L1:2
			3 : 1	3:1L1:2
1:3	20	60	1 : 3	1:3L1:3
			1 : 1	1:1L1:3
			3 : 1	3:1L1:3
1:1	100	100	1 : 3	1:3H1:1
			1 : 1	1:1H1:1
			3 : 1	3:1H1:1
1:2	100	200	1 : 3	1:3H1:2
			1 : 1	1:1H1:2
			3 : 1	3:1H1:2
1:3	100	300	1 : 3	1:3H1:3
			1 : 1	1:1H1:3
			3 : 1	3:1H1:3

Current Date 1/12/05

Sample Information

SampleName 1.3L1:1
 Vial 8
 Injection 1
 Injection Volume 100.00 μ l
 Channel SATIN
 Run Time 22.0 Minutes

Sample Type Broad Unknown
 Date Acquired 1/11/05 7:07:29 PM
 Acq Method Set Y2005_MethR_ THF_30C_2
 Processing Method Y2005_ProcR_ THF_30C_2
 Date Processed 1/12/05 8:35:29 AM

**Peak Results**

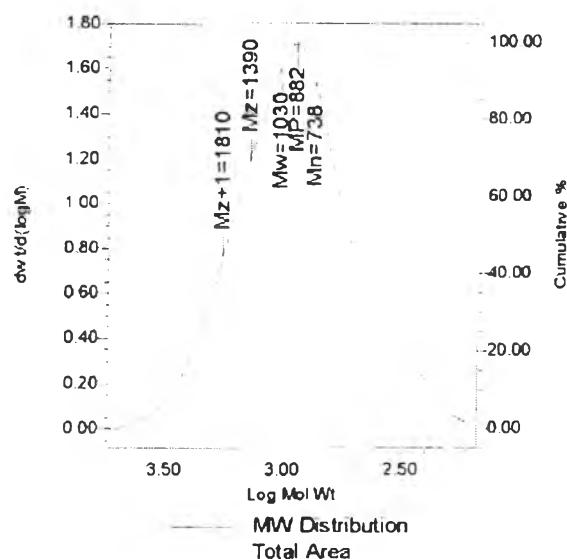
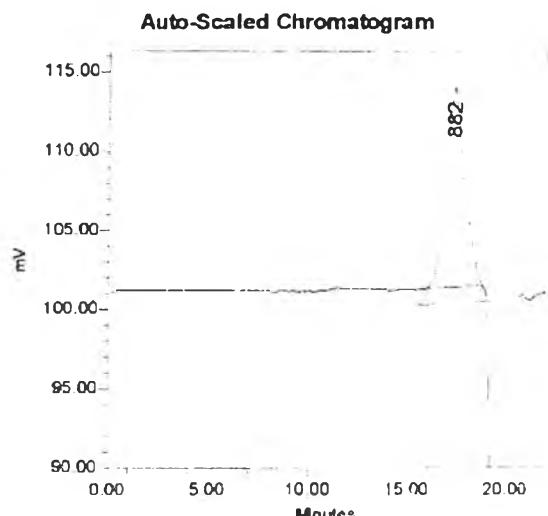
#	Mn	Mw	Mp	Mz	Mz+1	Polydispersity
1						
2	675	930	786	1267	1686	1.377369

Current Date 1/12/05

Sample Information

SampleName 1:3L1:2
 Vial 11
 Injection 1
 Injection Volume 100.00 ul
 Channel SATIN
 Run Time 22.0 Minutes

Sample Type Broad Unknown
 Date Acquired 1/11/05 8:24:37 PM
 Acq Method Set Y2005_MethR_THF_30C_2
 Processing Method Y2005_ProcR_THF_30C_2
 Date Processed 1/12/05 8:36:06 AM

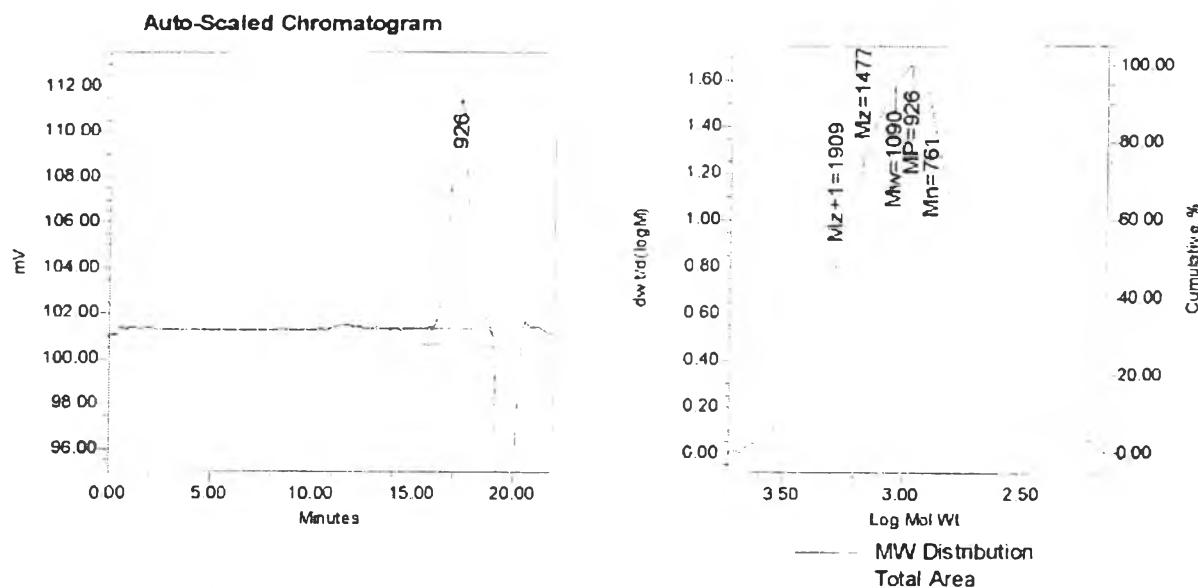
**Peak Results**

	Mn	Mw	Mp	Mz	Mz+1	Polydispersity
1						
2	738	1030	882	1390	1810	1.394901

Current Date 1/12/05

Sample Information

SampleName	1:3L1:3	Sample Type	Broad Unknown
Vial	14	Date Acquired	1/11/05 9:41:46 PM
Injection	1	Acq Method Set	Y2005_MethR_THF_30C_2
Injection Volume	100.00 μ l	Processing Method	Y2005_ProcR_THF_30C_2
Channel	SATIN	Date Processed	1/12/05 8:37:22 AM
Run Time	22.0 Minutes		



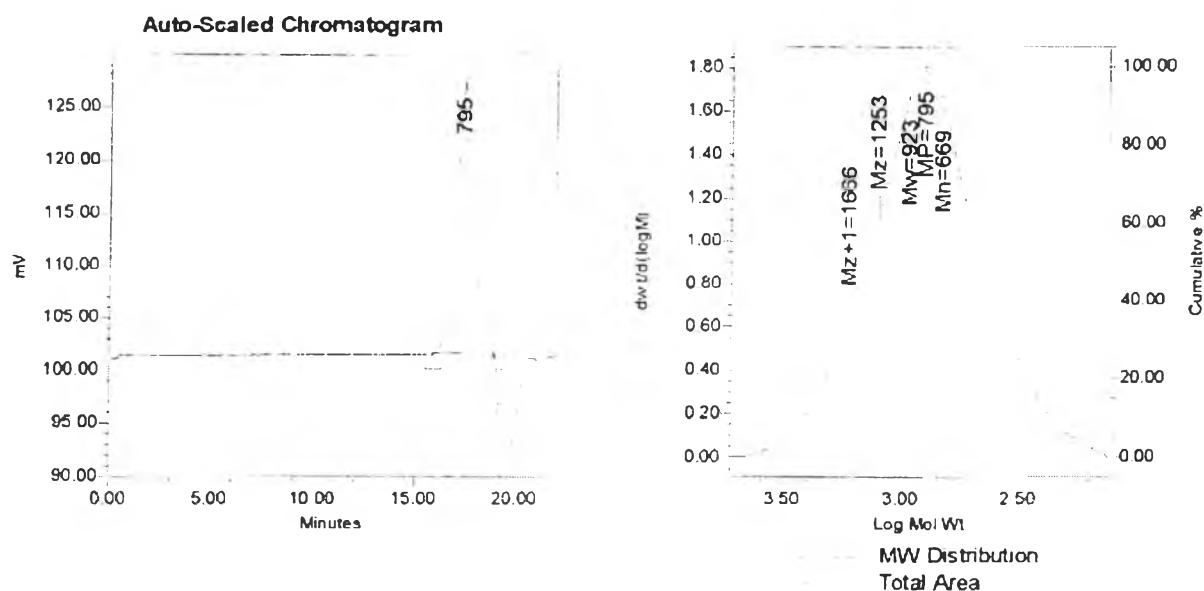
Peak Results

	Mn	Mw	MP	Mz	Mz+1	Polydispersity
1						
2	761	1090	926	1477	1909	1 433472

Current Date 1/12/05

Sample Information

SampleName	1.1L1:1	Sample Type	Broad Unknown
Vial	9	Date Acquired	1/11/05 7:33:11 PM
Injection	1	Acq Method Set	Y2005_MethR_THF_30C_2
Injection Volume	100.00 ul	Processing Method	Y2005_ProcR_THF_30C_2
Channel	SATIN	Date Processed	1/12/05 8:35:46 AM
Run Time	22.0 Minutes		



Peak Results

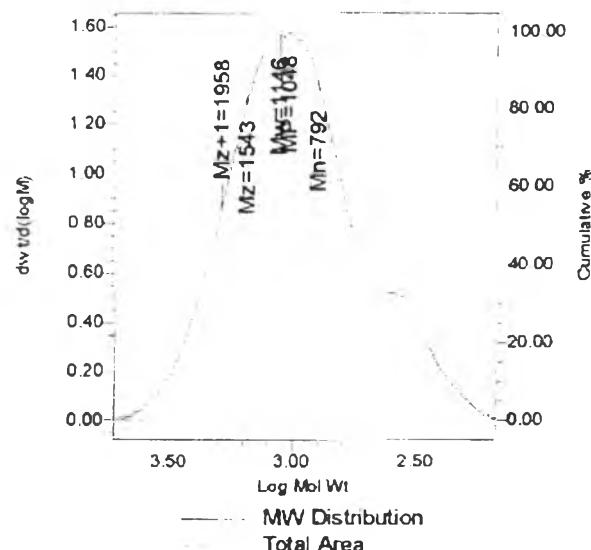
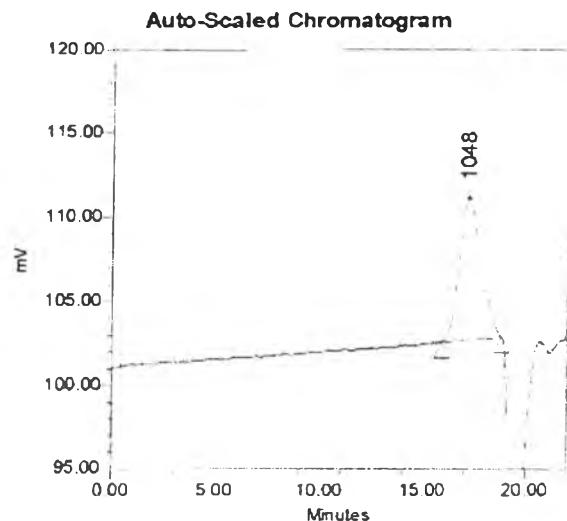
	Mn	Mw	Mp	Mz	Mz+1	Polydispersity
1						
2	669	923	795	1253	1666	1.378594

Current Date 1/12/05

Sample Information

SampleName 1 1L1.2
 Vial 2
 Injection 1
 Injection Volume 100.00 μ l
 Channel SATIN
 Run Time 22.0 Minutes

Sample Type Broad Unknown
 Date Acquired 1/11/05 3:30:27 PM
 Acq Method Set Y2005_MethR_THF_30C_2
 Processing Method Y2005_ProcR_THF_30C_2
 Date Processed 1/12/05 8:22:11 AM

**Peak Results**

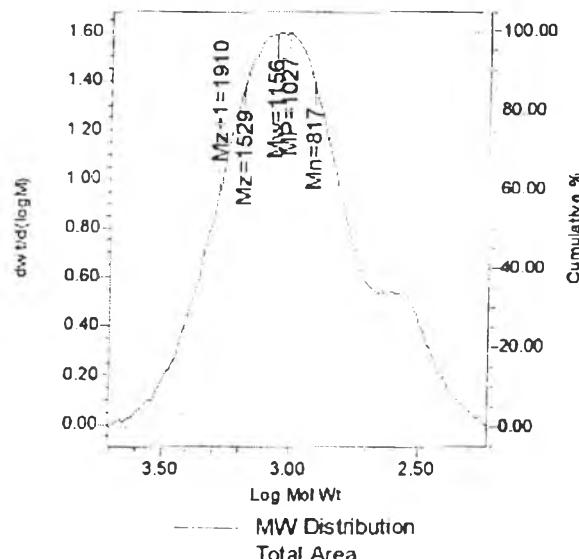
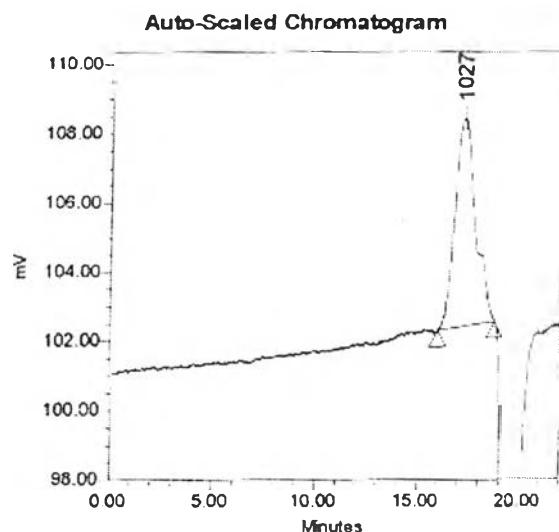
	Mn	Mw	MP	Mz	Mz+1	Polydispersity
1						
2	792	1146	1048	1543	1958	1.446658

Current Date 12/14/04

Sample Information

SampleName 1:1L(1:3)
 Vial 6
 Injection 1
 Injection Volume 100.00 ul
 Channel SATIN
 Run Time 22.0 Minutes

Sample Type Broad Unknown
 Date Acquired 12/14/04 2:01:40 PM
 Acq Method Set Y2004_1_MethR_THF_30C_4
 Processing Method Y2005_ProcR_THF_30C_1
 Date Processed 12/14/04 4:02:18 PM

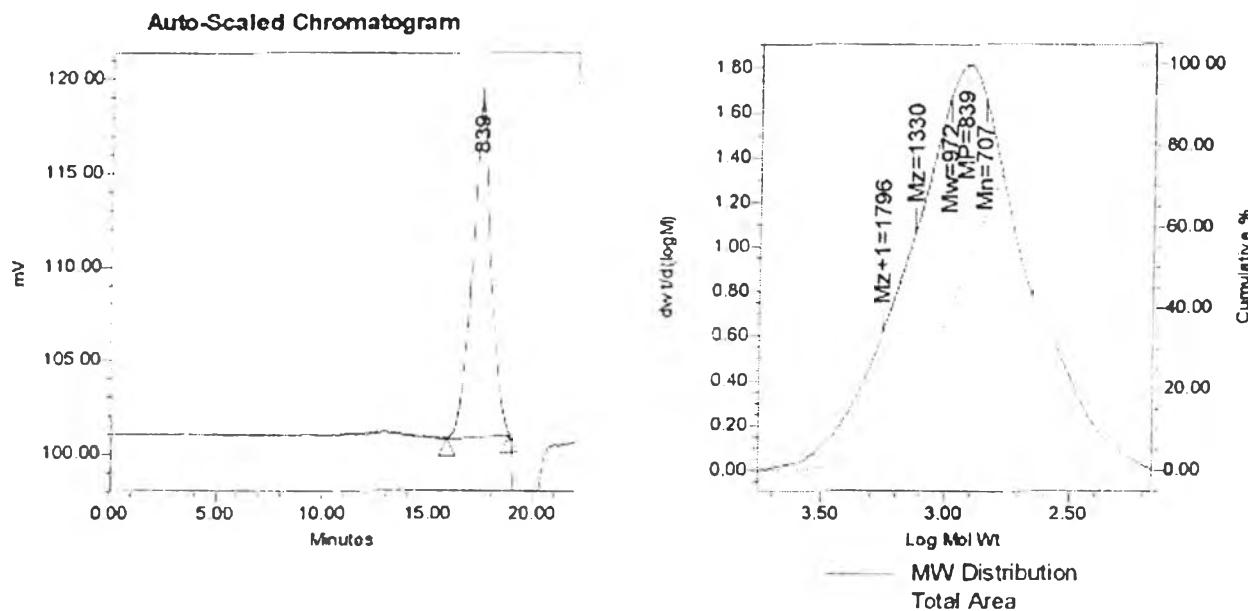
**Peak Results**

Pk	Mn	Mw	MP	Mz	Mz+1	Polydispersity
1						
2	817	1156	1027	1529	1910	1.414734

Current Date 12/14/04

Sample Information

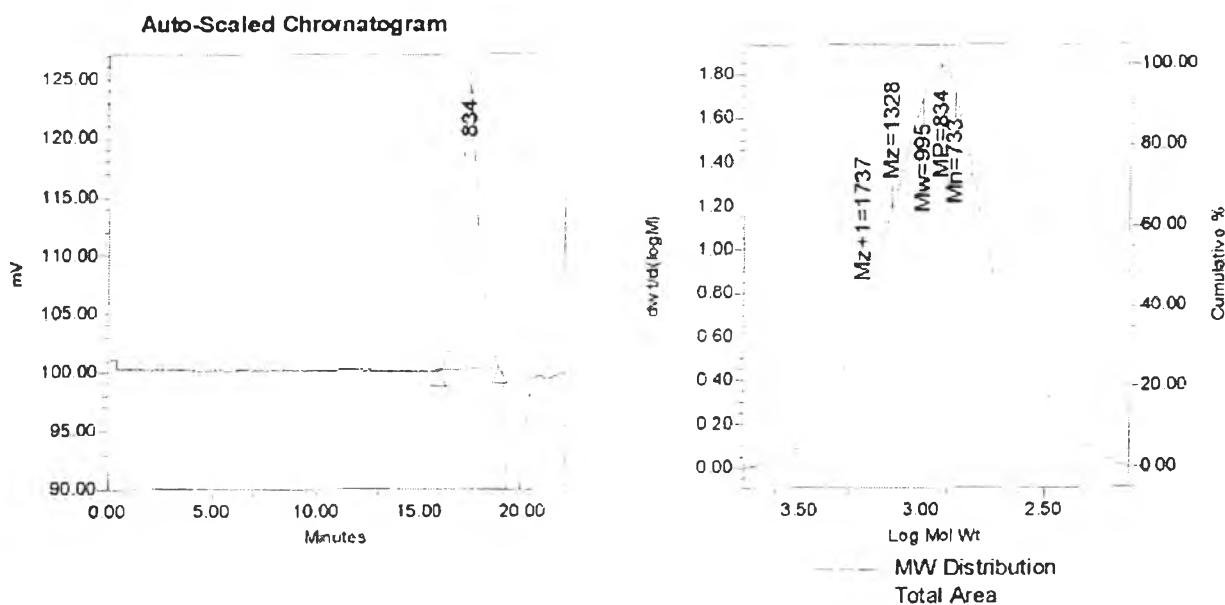
SampleName	3:1L1:1	Sample Type	Broad Unknown
Vial	3	Date Acquired	12/14/04 3:28:22 PM
Injection	1	Acq Method Set	Y2004_1_MethR_THF_30C_4
Injection Volume	100.00 ul	Processing Method	Y2005_ProcR_THF_30C_1
Channel	SATIN	Date Processed	12/14/04 4:05:19 PM
Run Time	22.0 Minutes		



Current Date 1/12/05

Sample Information

SampleName	3:1L1:2	Sample Type	Broad Unknown
Vial	10	Date Acquired	1/11/05 7:58:55 PM
Injection	1	Acq Method Set	Y2005_MethR_THF_30C_2
Injection Volume	100.00 μ l	Processing Method	Y2005_ProcR_THF_30C_2
Channel	SATIN	Date Processed	1/12/05 8:35:55 AM
Run Time	22.0 Minutes		



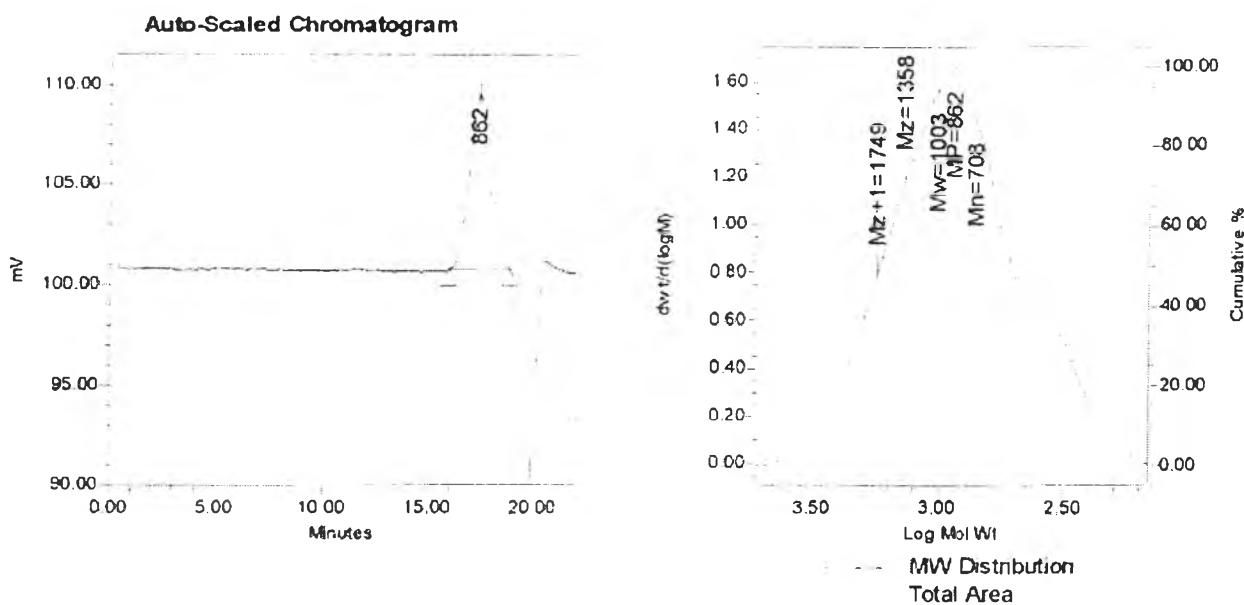
Peak Results

#	Mn	Mw	Mp	Mz	Mz+1	Polydispersity
1						
2	733	995	834	1328	1737	1.357290

Current Date 1/12/05

Sample Information

Sample Name	3:1L1:3	Sample Type	Broad Unknown
Vial	12	Date Acquired	1/11/05 8:50:21 PM
Injection	1	Acq Method Set	Y2005_MethR_THF_30C_2
Injection Volume	100.00 μ l	Processing Method	Y2005_ProcR_THF_30C_2
Channel	SATIN	Date Processed	1/12/05 8:36:21 AM
Run Time	22.0 Minutes		



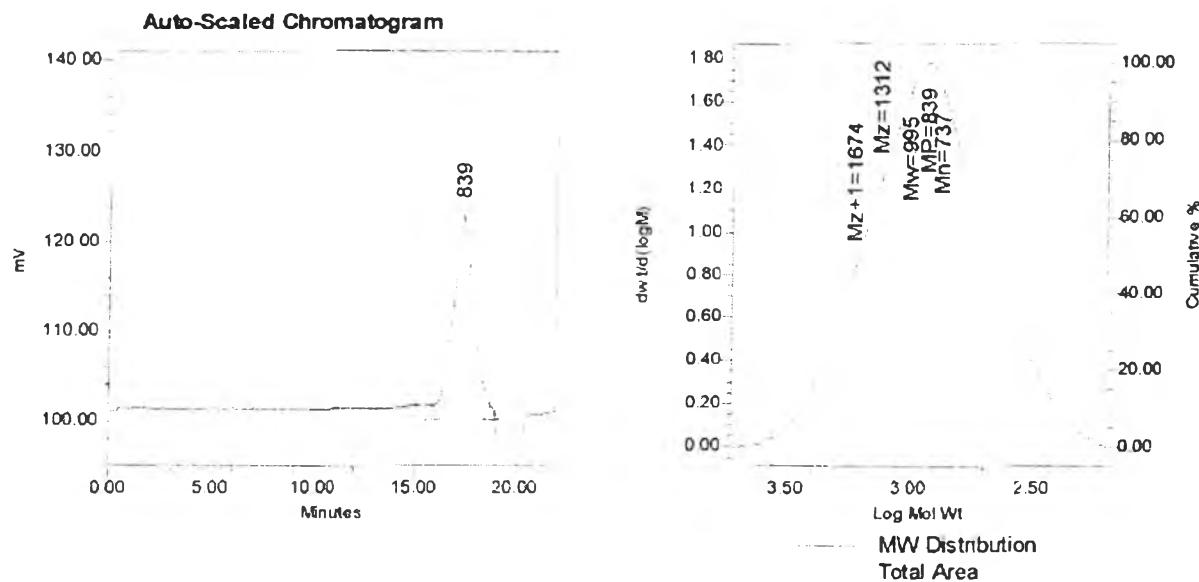
Peak Results

	Mn	Mw	MP	Mz	Mz+1	Polydispersity
1						
2	708	1003	862	1358	1749	1.415891

Current Date 1/12/05

Sample Information

SampleName	1:3H1:1	Sample Type	Broad Unknown
Vial	15	Date Acquired	1/11/05 10:07:30 PM
Injection	1	Acq Method Set	Y2005_MethR_THF_30C_2
Injection Volume	100.00 ul	Processing Method	Y2005_ProcR_THF_30C_2
Channel	SATIN	Date Processed	1/12/05 8:37:41 AM
Run Time	22.0 Minutes		

**Peak Results**

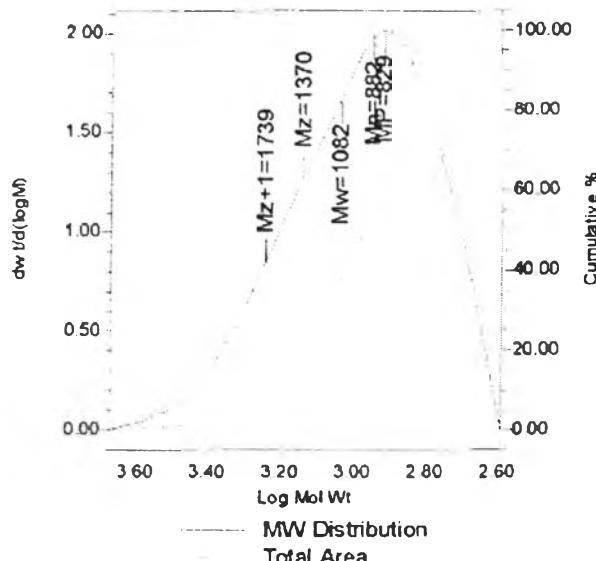
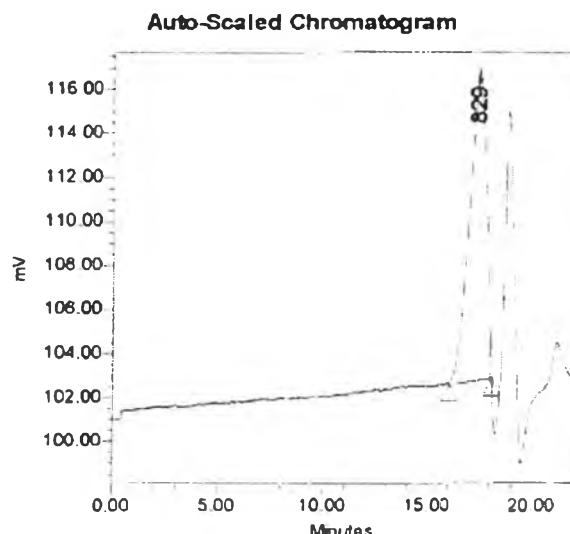
	Mn	Mv	MP	Mz	Mz+1	Polydispersity
1						
2	737	995	839	1312	1674	1 349638

Current Date 1/19/05

Sample Information

SampleName 1:3H1:2
 Vial 4
 Injection 1
 Injection Volume 100.00 ul
 Channel SATIN
 Run Time 22.0 Minutes

Sample Type Broad Unknown
 Date Acquired 1/19/05 11:08:13 AM
 Acq Method Set Y2005_MethR_THF_30C_2
 Processing Method Y2005_ProcR_THF_30C_2
 Date Processed 1/19/05 3:15:05 PM

**Peak Results**

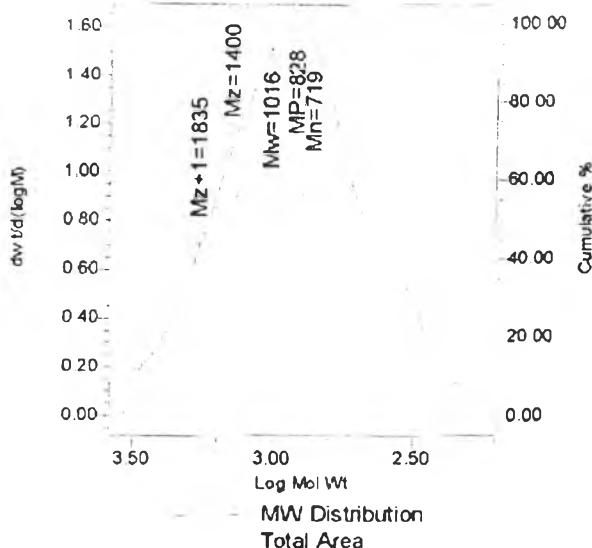
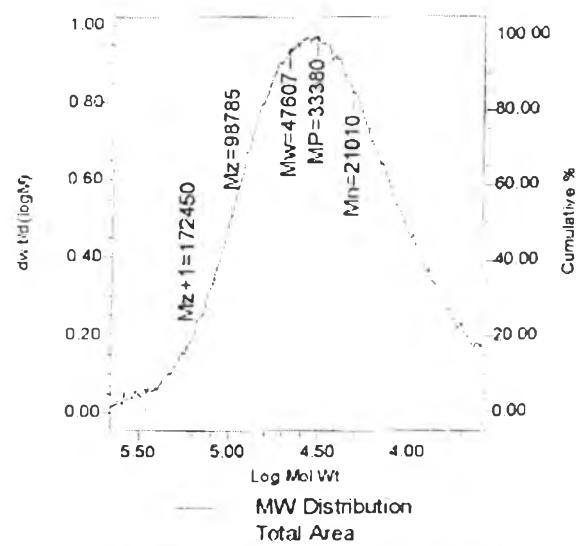
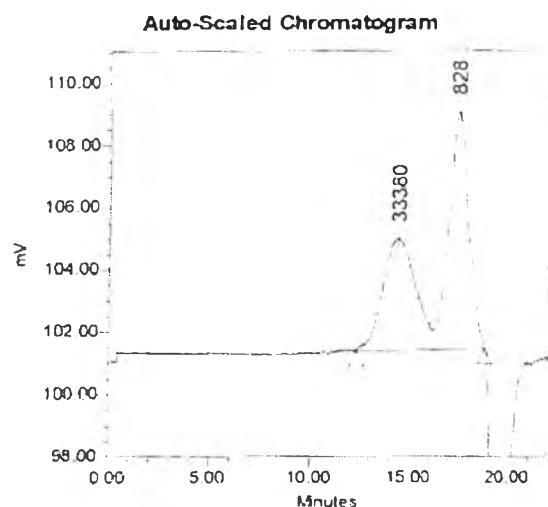
	Mn	Mw	Mp	Mz	Mz+1	Polydispersity
1						
2	882	1082	829	1370	1739	1.226502

Current Date 12/14/04

Sample Information

SampleName 1:3H1:3
 Vial 2
 Injection 1
 Injection Volume 100.00 ul
 Channel CATTIN
 Run Time 22.0 Minutes

Sample Type Broad Unknown
 Date Acquired 12/14/04 3:02:40 PM
 Acq Method Set Y2004_1_MethR_THF_30C_4
 Processing Method Y2005_ProcR_THF_30C_1
 Date Processed 12/14/04 4:06:37 PM

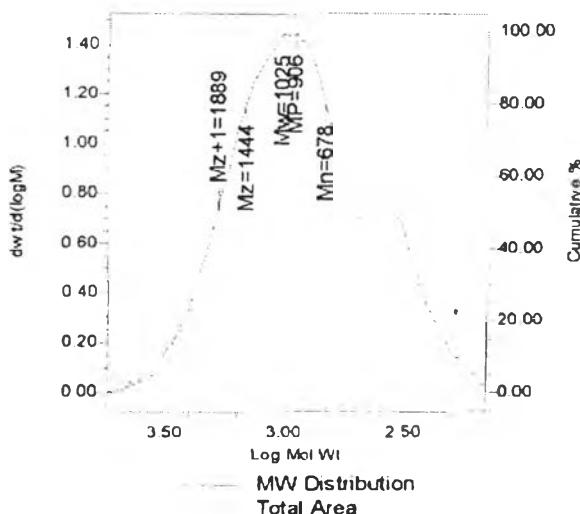
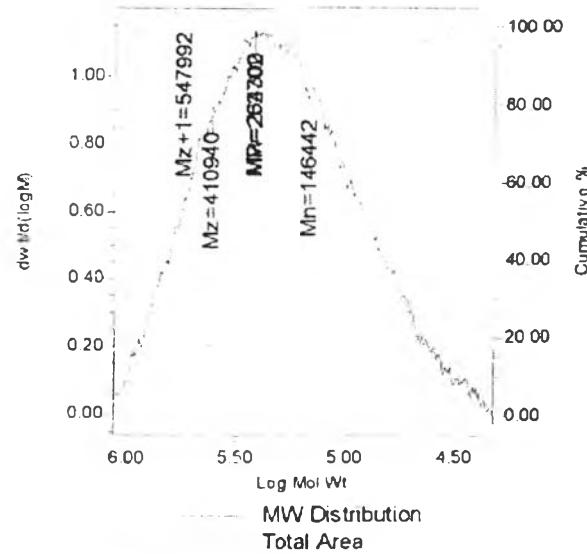
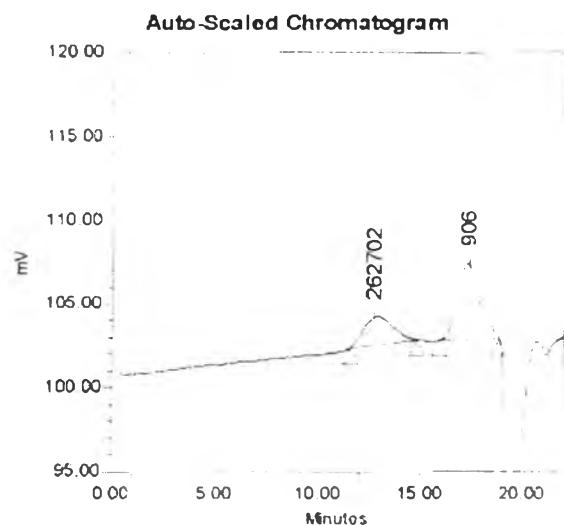
**Peak Results**

	Mn	Mw	Mp	Mz	Mz+1	Polydispersity
1	21010	47607	33380	98785	172450	2.265989
2	719	1016	828	1400	1835	1.412051

Current Date 1/12/05

Sample Information

SampleName	1:1H1:1	Sample Type	Broad Unknown
Vial	1	Date Acquired	1/11/05 3:04:45 PM
Injection	1	Acq Method Set	Y2005_MethR_THF_30C_2
Injection Volume	100.00 ul	Processing Method	Y2005_ProcR_THF_30C_2
Channel	SATIN	Date Processed	1/12/05 8:21:58 AM
Run Time	22.0 Minutes		

**Peak Results**

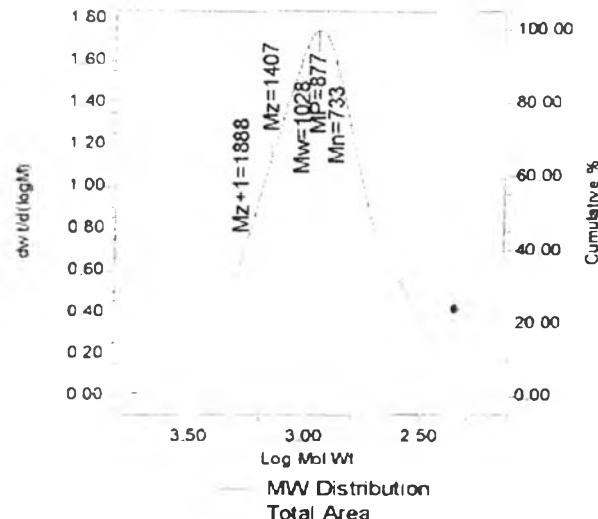
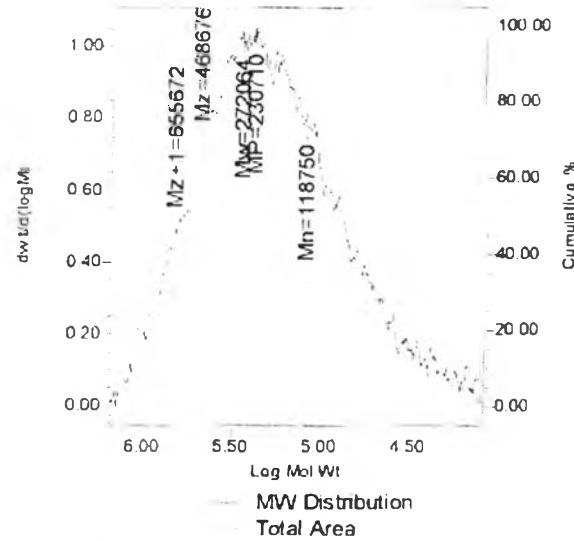
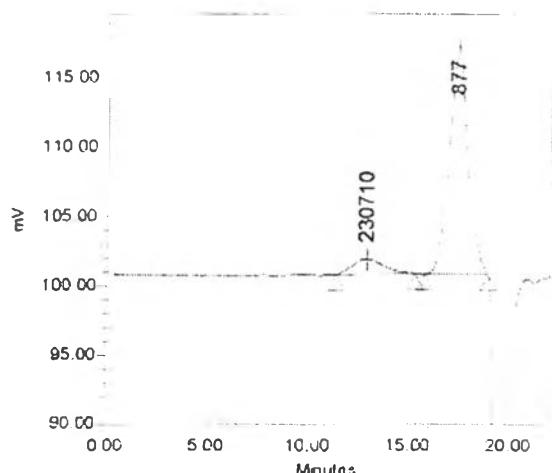
#	Mn	Mw	MP	Mz	Mz+1	Polydispersity
1	146442	254300	262702	410940	547992	1.804813
2	678	1025	906	1444	1889	1.512110

Current Date 1/12/05

Sample Information

SampleName	1:1H1:2	Sample Type	Broad Unknown
Vial	13	Date Acquired	1/11/05 9 16.04 PM
Injection	1	Acq Method Set	Y2005_MethR_THF_30C_2
Injection Volume	100.00 ul	Processing Method	Y2005_ProcR_THF_30C_2
Channel	SATIN	Date Processed	1/12/05 8 36:58 AM
Run Time	22.0 Minutes		

Auto-Scaled Chromatogram



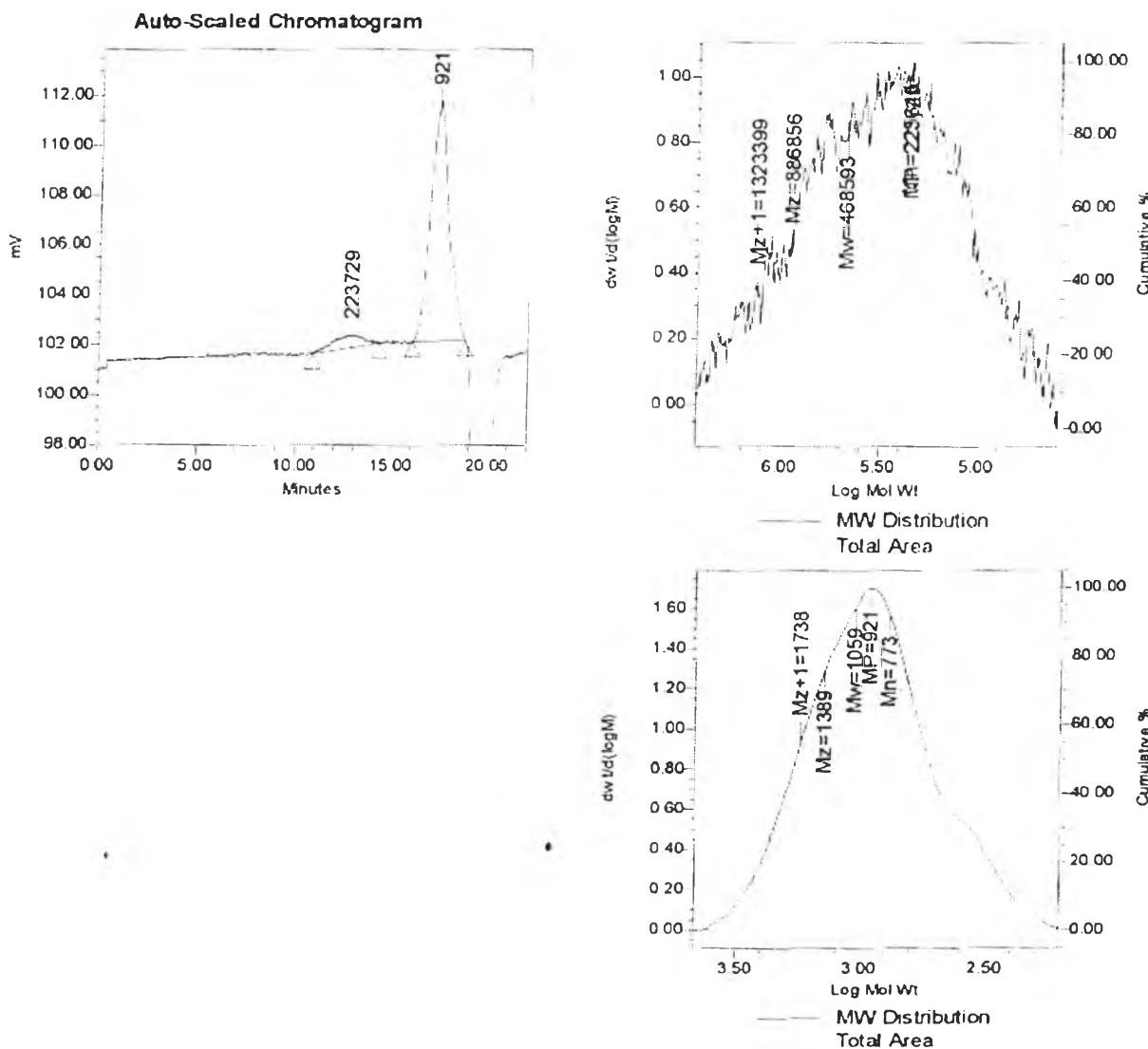
Peak Results

	Mn	Mw	Mz	Mz+1	Polydispersity
1	118750	272064	230710	468676	655672
2	733	1028	877	1407	1888

Current Date 12/14/04

Sample Information

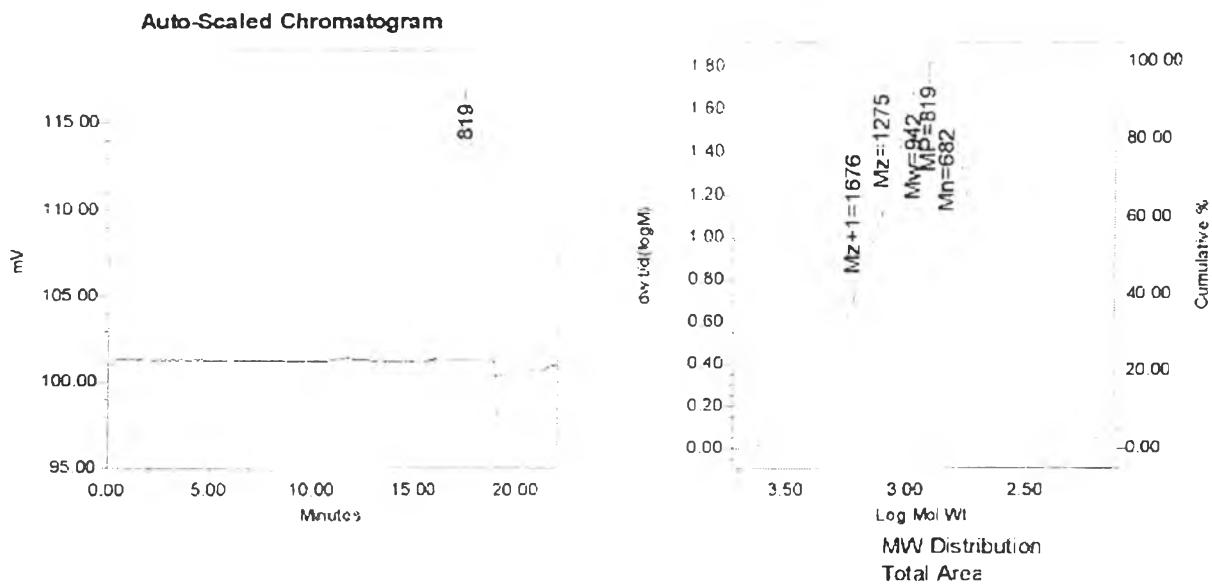
SampleName	1:1H1:3	Sample Type	Broad Unknown
Vial	1	Date Acquired	12/14/04 2:36:56 PM
Injection	1	Acq Method Set	Y2004_1_MethR_THF_30C_4
Injection Volume	100.00 μ l	Processing Method	Y2005_ProcR_THF_30C_1
Channel	SATIN	Date Processed	12/14/04 4:03:25 PM
Run Time	22.0 Minutes		



Current Date 1/12/05

Sample Information

SampleName	3:1H1:1	Sample Type	Broad Unknown
Vial	16	Date Acquired	1/11/05 10:33:12 PM
Injection	1	Acq Method Set	Y2005_MethR_THF_30C_2
Injection Volume	100.00 ul	Processing Method	Y2005_ProcR_THF_30C_2
Channel	SATIN	Date Processed	1/12/05 8:38:12 AM
Run Time	22.0 Minutes		



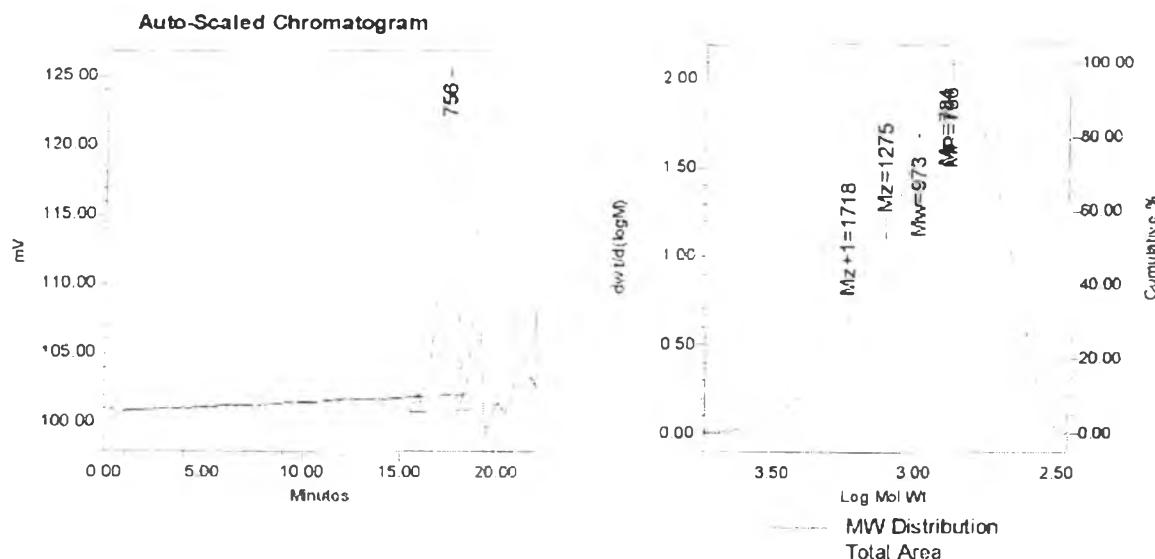
Peak Results

	Mn	Mw	Mp	Mz	Mz+1	Polydispersity
1						
2	682	942	819	1275	1676	1.381010

Current Date 1/19/05

Sample Information

SampleName	3.1H1_2	Sample Type	Broad Unknown
Vial	5	Date Acquired	1/19/05 11:33:55 AM
Injection	1	Acq Method Set	Y2005_MethR_THF_30C_2
Injection Volume	100.00 μ l	Processing Method	Y2005_ProcR_THF_30C_2
Channel	SATIN	Date Processed	1/19/05 3:15:14 PM
Run Time	22.0 Minutes		



Peak Results

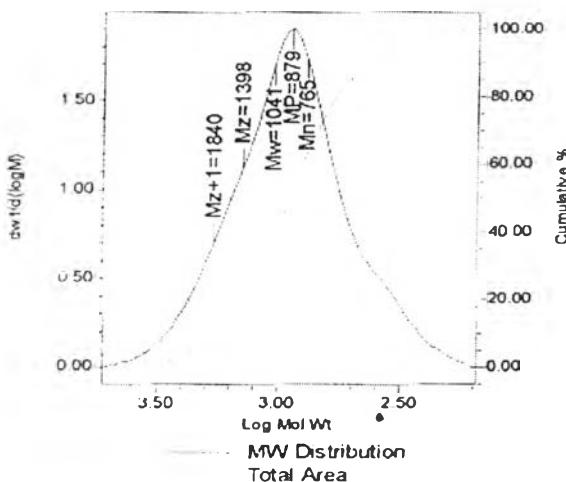
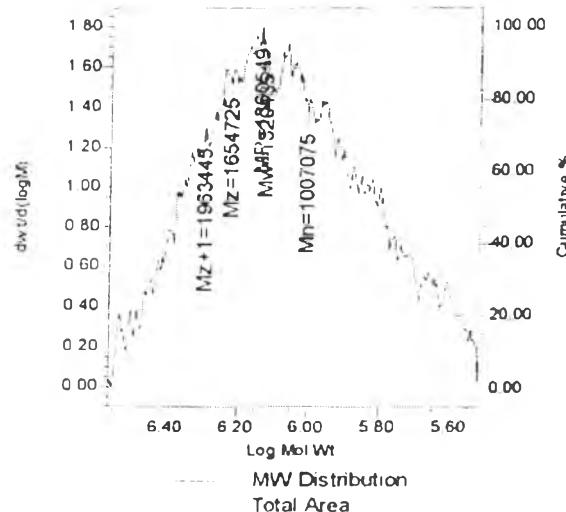
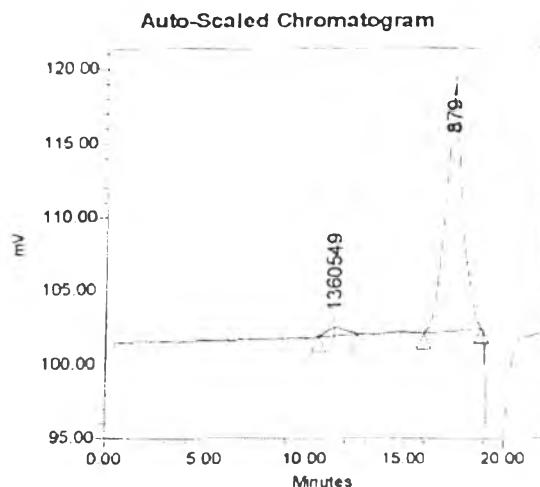
	M_n	M_w	M_p	M_z	$M_z + 1$	Polydispersity
1						
2	784	973	756	1275	1718	1.240947

Current Date 12/14/04

Sample Information

SampleName 3:1H1.3
 Vial 5
 Injection 1
 Injection Volume 100.00 ul
 Channel SATIN
 Run Time 22.0 Minutes

Sample Type Broad Unknown
 Date Acquired 12/14/04 1:36 00 PM
 Acq Method Set Y2004_1_MethR_THF_30C_4
 Processing Method Y2005_ProcR_THF_30C_1
 Date Processed 12/14/04 2 21:27 PM

**Peak Results**

	Mn	Mw	MP	Mz	Mz+1
1	1007075	1320435	1360549	1654725	1963445
2	765	1041	879	1398	1840

Peak Results

	Polydispersity
1	1.311159
2	1.361098

CURRICULUM VITAE

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Poster Presentation:

1. Katchamart, S., Arayawongkul, S., Nithitanakul, M. and O'Haver J. H. (2005) Characterization of Polymer formed via Admicellar Polymerization: Copolymerization of Styrene/Isoprene. Poster Presentation at the 7th World Congress of Chemical Engineering, Glasgow, Scotland.