

REFERENCES

1. L.Montanaro, N. Bianchini, J.Ma. Rincon and M. Romero. Sintering behaviour of pressed red mud wastes from zinc hydrometallurgy. Ceramic International 27 (2001) : 29-37.
2. Mario Pelino, Carlo Cantalini and Carlo Abbruzzese. Treatment and recycling of goethite waste arising from the hydrometallurgy of zinc. Hydrometallurgy 40 (1996) : 25-35.
3. Mauro Porcu, Roberto Orru and Giacomo Cao. On the used of industruil scraps for the treatment of zinc hydrometallurgical wastes by self-propagating reactions. Chemical Engineering Journal 99 (2004) : 247-256.
4. M.Romero and J.Ma Rincon. Microstructural characterization of a goethite waste from zinc hydrometallurgical process. Materials Letters 31 (1997) : 67-73.
5. M. Pelino. Recycling of zinc-hydrometallurgy wastes in glass and glass ceramic materials. Waste Management 20 (2000) : 561-568.
6. Iron oxide red. Available from: www.ceramic-materials.com. July 2005.
7. The critical role of networks in embedded industrial production systems lessons from waste management in zinc production. Available from: www.deft2001.tudelft.nl/paper/20files/paper1169.doc. 2001.
8. M.K. Jha, V. Kumar and R.J. Singh. Review of hydrometallurgical recovery of zinc from industrial wastes. Resources, Conservation and Recycling 33 (2001) : 1-22.
9. M.R.C. Ismael and J.M.R. Carvalho. Iron recovery from sulphate leach liquors in zinc hydrometallurgy. Minerals Engineering 16 (2003) : 31-39.
10. International Trad Center UNCTAD/GATT. Unglazed and glazed ceramic tiles. Geneva : The Center, 1982.
11. Cullen W. Parmelee. Ceramic glazes. 3th edition. The United States of America: The Maple Press, 1973.
12. Robin Hopper. The ceramic spectrum. The United States of America : Chilton Book, 1984.
13. ไพจิตร อังศิริวัฒน์. สีเซรามิก. กรุงเทพมหานคร : สำนักพิมพ์โอเดียนสโตร์, 2546.

14. An overview of ceramic stains. Available from: www.ceramic-materials.com. July 2005.
15. V. Mymrin and A. Vazquez Vaamonde. Technical note new construction materials from spanish jarosite processing wastes. Materials Engineering Vol.12, No.11 (1999) : 1399-1402.
16. Vsevolod A. Mymrin, Haroldo A. Ponte and Patricio R. Impinnisi. Potential application of acid jarosite wastes as the main component of construction materials. Construction and Building Materials 19 (2005) : 141-146.
17. J.L.T. Hage and R.D. Schuiling. Comparative column elution of jarosite waste and its autoclaved product-evidence for the immobilization of deleterious elements in jarosite. Minerals Engineering Vol 13, No. 3 (2000) : 287-296.
18. Asokan Pappu, Mohini Saxena and Shyam R. Asolekar. Jarosite characteristics and its utilization potentials. Science of the Total Environment Article in press (2005).
19. M. Pelino. Recycling of jarosite waste in the production of glass and glass ceramic materials . Interceram 47 (1998) : 22-26.
20. M. Romero and J. Ma. Rincon. Preparation and properties of high iron oxide content glasses obtained from industrial wastes. Journal of the European Ceramic Society 18 (1998) : 153-160.
21. Romualdo R. Menezes, Heber S. Ferreira, Gelmires A. Neves, Helio de L. Lira and Heber C. Ferreira. Use of granite sawing wastes in the production of ceramic bricks and tiles. Journal of the European Ceramic Society 25 (2005) : 1149-1158.
22. S.N. Monteiro, L.A. Pecanha and C.M.F. Vieira. Reformulation of roofing tiles body with addition of granite waste from sawing operations. Journal of the European Ceramic Society 24 (2004) : 2349-2356.
23. M. Romero, J. Ma. Rincon and A. Acosta. Effect of iron oxide content on the crystallization of a diopside glass-ceramic glaze. Journal of the European Ceramic Society 22 (2002) : 883-890.
24. P. Appendino, M. Ferraris, I. Matekovits and M. Salvo. Production of glass-ceramic bodies from the bottom ashes of municipal solid waste incinerators. Journal of the European Ceramic Society Article in press (2003).

25. T.W. Cheng, T.H. Ueng, Y.S. Chen and J.P. Chiu. Production of glass-ceramic from incinerator fly ash. Ceramic International 28 (2002) : 779-783.
26. Standard test method for water absorption, bulk density, apparent porosity, and apparent specific gravity of fired whiteware products. ASTM Standard C373-88 (1998).
27. Determination of resistance to deep abrasion for unglazed tiles . International standard ISO 13006 (1998).
28. Standard test method for resistance of ceramic tile to chemical substances. ASTM Standard C650-97 (1997).
29. Method 1311 toxic characteristic leaching procedure(online).Available from: <http://www.epa.gov/epaoswer/hazwaste/test/pdfs/1311>. July 1992.
30. Padaeng Industry Public Co. Ltd. The brochure of zinc in everyday life.
31. W.D. Kingery, H.K. Bowen and D.R. Uhlmann. Introduction to ceramics. Singapore : John Wiley & Sons, 1991.
32. V.K. Marghussan and A. Maghsoodipoor. Fabrication of unglazed floor tiles containing Iranian copper slags. Ceramics International 25 (1999) : 617-622.
33. Felix singer and Sonja S.Singer. Industrial ceramics. The United States of Kingdom: Chapman & Hell, 1963.
34. William E. Lee Dphil and W. Mark Rainforth PhD. Ceramic microstructures. The United States of Kingdom: Chapman & Hell, 1994.

APPENDICES

APPENDIX A

Table A-1 Percent linear shrinkage of specimens

Temp.	(%)	Formula										
		T1	T2	T3	T4	T5	T6	T7	T8	T9	T10	T11
1075°C	Shrinkage	2.36	2.92	2.26	2.59	3.77	10.77	8.99	11.84	6.36	6.33	3.66
	S.D.	0.16	0.11	0.10	0.18	0.33	0.79	0.68	0.57	0.27	0.21	0.07
1100°C	Shrinkage	3.73	5.07	3.77	5.25	8.67	14.55	13.61	13.73	8.91	12.00	6.91
	S.D.	0.30	0.22	0.08	0.35	0.47	0.07	0.04	0.18	1.01	2.20	1.43
1125°C	Shrinkage	7.41	7.90	5.27	9.63	11.76	13.96	12.15	13.15	11.92	13.03	10.82
	S.D.	0.12	0.43	0.20	2.29	0.04	0.20	0.29	0.26	0.07	0.10	0.20
1150°C	Shrinkage	10.93	11.64	8.60	11.60	11.91	8.45	10.54	11.59	11.67	12.48	12.73
	S.D.	0.48	0.25	0.40	0.12	0.04	0.58	0.34	0.21	0.09	0.26	0.23
1175°C	Shrinkage	12.55	11.38	10.29	10.42	*	*	*	*	*	*	*
	S.D.	0.03	0.34	0.24	0.38	*	*	*	*	*	*	*

Table A-2 Bulk density of specimens

Temp.		Formula										
		T1	T2	T3	T4	T5	T6	T7	T8	T9	T10	T11
1075°C	B.D. (g/cm ³)	1.72	1.81	1.81	1.76	1.88	2.27	2.22	2.35	1.92	1.88	1.73
	S.D. (%)	0.01	0.01	-	0.01	0.02	0.06	0.04	0.03	0.01	0.01	0.07
1100°C	B.D. (g/cm ³)	1.82	1.94	1.86	1.93	2.18	2.54	2.50	2.50	2.14	2.17	1.80
	S.D. (%)	0.02	0.02	0.08	0.02	0.04	0.01	-	0.01	0.05	0.06	0.07
1125°C	B.D. (g/cm ³)	2.04	2.14	2.00	2.19	2.44	2.46	2.43	2.44	2.34	2.36	2.22
	S.D. (%)	0.03	0.03	0.02	0.07	0.01	0.02	0.02	0.03	0.01	0.01	0.02
1150°C	B.D. (g/cm ³)	2.31	2.45	2.15	2.42	2.35	1.96	2.09	2.20	2.28	2.41	2.31
	S.D. (%)	0.04	0.02	0.04	-	0.01	0.03	0.02	0.01	0.01	-	0.02
1175°C	B.D. (g/cm ³)	2.44	2.39	2.36	2.28	*	*	*	*	*	*	*
	S.D. (%)	0.01	0.03	0.02	0.03	*	*	*	*	*	*	*

The specimens were shape distortion and bloating.

Table A-3 Percent water absorption of specimens

Temp.	(%)	Formula										
		T1	T2	T3	T4	T5	T6	T7	T8	T9	T10	T11
1075°C	W.A.	20.32	17.63	17.79	19.58	16.52	5.83	8.35	5.26	16.38	17.86	23.55
	S.D.	0.35	0.25	0.13	0.30	0.62	2.04	1.40	0.81	0.32	0.16	0.34
1100°C	W.A.	16.84	13.89	16.28	14.04	7.62	0.11	0.14	0.10	9.78	9.64	20.98
	S.D.	0.62	0.58	0.25	0.85	1.39	0.05	0.02	0.04	1.39	1.63	2.31
1125°C	W.A.	10.14	7.68	11.35	3.63	0.09	0.06	0.09	0.16	0.14	2.84	9.66
	S.D.	0.79	1.05	0.50	0.87	0.03	0.03	0.06	0.11	0.03	0.96	0.05
1150°C	W.A.	2.13	0.14	3.01	0.08	0.17	0.78	0.65	1.25	1.24	0.18	5.47
	S.D.	1.04	0.09	1.17	0.02	0.06	0.62	0.39	0.39	0.15	0.01	1.24
1175°C	W.A.	0.09	0.09	0.10	0.08	*	*	*	*	*	*	*
	S.D.	0.03	0.04	0.05	0.02	*	*	*	*	*	*	*

Table A-4 Percent apparent porosity of specimens

Temp.	(%)	Formula										
		T1	T2	T3	T4	T5	T6	T7	T8	T9	T10	T11
1075°C	A.P.	35.01	32.01	32.35	34.63	31.19	13.18	18.59	12.35	31.53	33.63	40.93
	S.D.	0.039	0.017	0.32	0.85	0.35	4.27	2.88	1.74	0.41	0.17	2.22
1100°C	A.P.	30.73	26.98	30.41	27.22	16.64	0.29	0.34	0.26	20.95	20.92	37.69
	S.D.	0.80	0.84	0.34	1.34	2.71	0.12	0.05	0.10	2.16	2.96	2.82
1125°C	A.P.	20.78	16.46	22.82	7.95	0.22	0.15	0.21	0.38	0.32	6.73	21.52
	S.D.	1.47	1.97	0.83	1.65	0.08	0.07	0.14	0.26	0.07	2.26	0.95
1150°C	A.P.	4.91	0.35	6.60	0.20	0.39	1.54	1.36	2.75	2.83	0.43	12.65
	S.D.	2.78	0.22	2.48	0.05	0.15	1.23	0.81	0.84	0.35	0.03	2.78
1175°C	A.P.	0.23	0.22	0.24	0.18	*	*	*	*	*	*	*
	S.D.	0.08	0.09	0.12	0.04	*	*	*	*	*	*	*

The specimens were shape distortion and bloating.

APPENDIX B

Table B-1 Bending strength of Specimens

Temp.		Formula					
		T1	T4	T5	T6	T7	T8
1100°C	MOR (MPa)	24.33	43.94	60.36	94.70	101.94	107.26
	S.D. (%)	2.58	3.73	11.49	16.82	10.08	22.86
1125°C	MOR (MPa)	33.11	57.15	84.09	82.42	95.51	101.61
	S.D. (%)	6.73	4.76	3.26	14.79	11.12	20.87
1150°C	MOR (MPa)	36.43	61.60	64.06	30.12	42.71	49.17
	S.D. (%)	4.57	12.58	15.49	2.62	7.29	5.75
1175°C	MOR (MPa)	21.55	43.67	*	*	*	*
	S.D. (%)	6.70	9.72	*	*	*	*

* The specimens were shape distortion and bloating.

BIOGRAPHY

Miss Laksana Kreethawate was born in Nakhon Sawan on 22th November 1976. In 2002, after she had finished her Bachelor's Degree from Department of Physics, Faculty of Science, Chiang Mai University, she continued to study in Master's Degree in the field of Ceramic Technology at Chulalongkorn University. In 2003-2004, she had received a scholarship from Thailand Graduate Institute of Science and Technology (TGIST), National Science and Technology Development Agency (NSTDA).

