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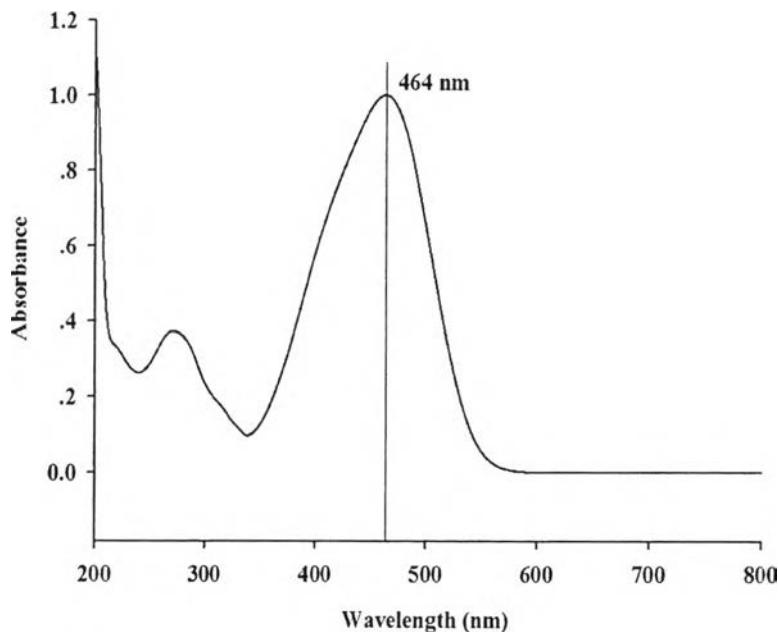
## APPENDIX

### Appendix A UV-Visible Spectrometer

Methyl orange concentration was identified by measuring light absorption of sampling solution. The absorption spectrum of methyl orange was taken to find a suitable wavelength to determine the methyl orange content. From Figure A, the 464 nm wavelength was selected to study. Concentration of the methyl orange was determined using the standard calibration curve and the equation shown below.

$$y = -0.0017 + 0.0489x$$

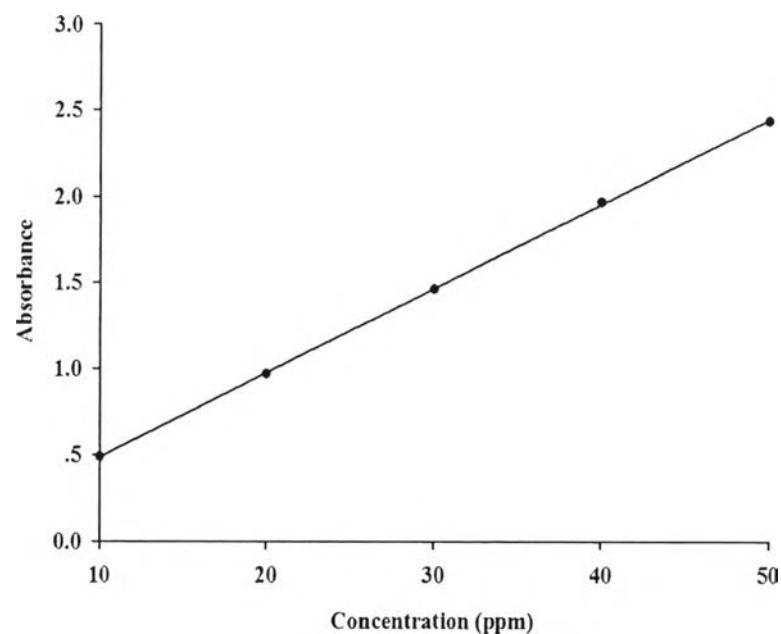
y = absorbance, x = methyl orange concentration



**Figure A1** UV-visible absorption spectrum (200-800) of methyl orange solution.

**Table A1** Absorbance and concentration of standard methyl orange solution

Concentration (ppm)	Absorbance
10	0.48912
20	0.96872
30	1.46216
40	1.96609
50	2.43301

**Figure A2** Calibration curve of methyl orange at various concentrations.

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**Proceedings:**

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**Presentations:**

1. Piwnuan, C.; Chaisuwan, T.; Luengnaruemitchai, A.; and Wongkasemjit, S. (2012, March 25-29) Catalytic Activity of Metal Loaded Titania Nanotubes. Paper presented at the 243<sup>rd</sup> ACS National Meeting & Exposition, San Diego, CA, USA.
2. Piwnuan, C.; Chaisuwan, T.; Luengnaruemitchai, A.; and Wongkasemjit, S. (2012, April 24) Catalytic Activity of Metal Loaded Titania Nanotubes. Paper presented at the 3<sup>rd</sup> Research Symposium on Petrochemical and Materials Technology and the 18<sup>th</sup> PPC Symposium on Petroleum, Petrochemicals, and Polymers, Bangkok, Thailand.

