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## APPENDICES

### Appendix A Determination of methyl red adsorption by porous sample

The amount of methyl red (MR) adsorbed onto porous sample was calculated following equation:

$$q = \frac{V(C_0 - C)}{mS}$$

Where,  $q$  = The amount of MR adsorbed onto porous sample ( $\text{mol}/\text{m}^2$ )

$C_0$  = The initial concentration of MR solution ( $\text{mol}/\text{l}$ )

$C$  = The residual concentration of MR solution ( $\text{mol}/\text{l}$ )

$V$  = The solution volume ( $\text{l}$ )

$m$  = Mass of porous sample ( $\text{g}$ )

$S$  = BET surface area of porous sample ( $\text{m}^2/\text{g}$ ) sample ( $\text{mol}/\text{m}^2$ )

**Table A1** Adsorption of methyl red by APPCH and PCH

Sample	BET Surface area ( $\text{m}^2/\text{g}$ )	Residual Concentration ( $\text{mol}/\text{l}$ )	Amount of Methyl Red Adsorption ( $\text{mol}/\text{m}^2$ )
PCH	524.1	$4.17 \times 10^{-5}$	$1.51 \times 10^{-7}$
APPCH (0.95: 0.05)	275.2	$1.76 \times 10^{-5}$	$3.32 \times 10^{-7}$

## Appendix B Determination of Total Volatile Basic Nitrogen (TVB-N)

TVB-N can be calculated by following equation:

$$\text{TVB-N (mg/100g)} = \frac{(V_S - V_B) \times (N_{HCl} \times A_N) \times [W_S \times (M/100) + V_E] \times 100}{W_S}$$

- Where,  
 $V_S$  = Titration volume of 0.01 N HCl for sample extract (ml)  
 $V_B$  = Titration volume of 0.01 N HCl for blank (ml)  
 $N_{HCl}$  = Normality of HCl ( $= 0.01 \text{ N} \times \text{factor of HCl}$ )  
 $A_N$  = Atomic weight of nitrogen (14.00)  
 $W_S$  = Weight of tissue sample (g)  
 $M$  = Percentage moisture of tissue sample (Assume 80%)  
 $V_E$  = Volume of 4% TCA used in extraction

**Table B1** Change in TVB-N Values of fresh fish during storage at ambient temperature

Hours	0.01 HCl (ml)	TVB-N (mg/100g)
0	0.15	10.08
3	0.20	13.44
6	0.35	23.23
9	0.60	40.32
12	0.80	53.76
15	1.40	94.08
18	1.00	67.20
21	1.00	67.20
24	1.20	80.64

### Appendix C Mechanical Measurement of nanocomposites

**Table C1** Mechanical properties of PP/APPCH-MR nanocomposite

Sample	Young's Modulus (MPa)	Tensile Strength (MPa)	Elongation at Yield (%)
PP	1201.63 ± 98.73	36.03 ± 2.00	9.70 ± 0.72
PP/6%wtSurlyn	1103.01 ± 72.05	33.48 ± 1.55	8.49 ± 0.47
PP/6%wtS/2%wtClay (10:1)	1035.02 ± 61.82	28.95 ± 2.16	7.05 ± 0.57
PP/6%wtS/2%wtClay (20:1)	1071.31 ± 66.21	29.51 ± 1.82	7.46 ± 0.62
PP/6%wtS/2%wtClay (30:1)	1054.14 ± 69.34	29.88 ± 2.51	7.21 ± 0.86

**Table C2** Mechanical properties of LDPE/PCH-BTB nanocomposites

Sample	Young's Modulus (MPa)	Tensile Strength (MPa)	Elongation at Yield (%)
LDPE	200.19 ± 7.17	8.69 ± 0.30	19.13 ± 5.33
LDPE/6%wtSurlyn	215.75 ± 13.35	8.81 ± 0.47	15.35 ± 2.34
LDPE/6%wtS/2%wtClay (10:1)	221.72 ± 8.62	9.00 ± 0.43	14.42 ± 2.00
LDPE/6%wtS/2%wtClay (20:1)	215.26 ± 14.48	8.94 ± 0.51	16.64 ± 2.03
LDPE/6%wtS/2%wtClay (30:1)	216.67 ± 15.94	8.68 ± 0.46	15.46 ± 2.76

## Appendix D Oxygen Gas Transmission Rate of Nanocomposites

**Table D1** Oxygen gas transmission rate of PP and PP/APPCH-MR nanocomposites

Sample	Oxygen gas transmission rate (cc/m <sup>2</sup> .day)
PP	148 ± 3.39
PP/APPCH-MR (30:1)	140 ± 2.97

**Table D2** Oxygen gas transmission rate of LDPE and LDPE/PCH-BTB nanocomposites

Sample	Oxygen gas transmission rate (cc/m <sup>2</sup> .day)
LDPE	257 ± 7.07
LDPE/PCH-BTB (10:1)	225 ± 0.85
LDPE/PCH-BTB (30:1)	230 ± 7.07

## Appendix E Bentonite Clay, Mac-Gel® GRADE SAC

**Table E1** Typical chemical analysis of bentonite on dry basis at 105°C

Element	Percentage
SiO <sub>2</sub>	65 - 70
Al <sub>2</sub> O <sub>3</sub>	13 - 17
Fe <sub>2</sub> O <sub>3</sub>	1.0 - 2.0
Na <sub>2</sub> O	1.5 - 2.5
LOI	10 - 12
MgO	2.0 - 3.0
CaO	1.5 - 2.5
K <sub>2</sub> O	0.4 - 0.8
TiO <sub>2</sub>	0.2 - 0.3

**Table E2** Physical properties of bentonite

Moisture content, %	8 - 12
5% suspension, pH	9.5 - 11.0
Swelling index, ml per 2 g of clay	15
Viscosity dial reading at 600 rpm	12 - 20
Dry particle size (pass 200 meshes), %	80 min
Wet particle size (pass 325 meshes)	98 min
Specific gravity	2.3 - 2.4
CEC, meq/100g of clay	44.5

## CIRICULLUM VITAE

**Name:** Ms. Supatcharee Boonruang

**Date of Birth:** May 12, 1988

**Nationality:** Thai

### **University Education:**

2006 – 2009 Bachelor Degree of Polymer Science, Faculty of Science, Chulalongkorn University, Bangkok, Thailand

### **Proceedings:**

1. Boonruang, S., Magaraphan, R. and Manuspiya, H. (2012, January 11-13) Synthesis of pH Dye Modified Porous Clay Heterostucture for Smart Packaging Film. Proceedings of the 6<sup>th</sup> Pure and Applied Chemistry International Conference, the Empress Hotel, Chiang Mai, Thailand.
2. Boonruang, S., Magaraphan, R. and Manuspiya, H. (2012, April 24) Chromophores Modified Porous Clay Heterostructure for Smart Packaging Films. Proceedings of the 18<sup>th</sup> PPC Symposium in Petroleum, Petrochemical and Polymers, Bangkok, Thailand.

### **Presentations:**

1. Boonruang, S., Magaraphan, R. and Manuspiya, H. (2012, January 11-13) Synthesis of pH Dye Modified Porous Clay Heterostucture for Smart Packaging Film. Paper presented at the 6<sup>th</sup> Pure and Applied Chemistry International Conference, the Empress Hotel, Chiang Mai, Thailand.
2. Boonruang, S., Magaraphan, R. and Manuspiya, H. (2012, April 24) Chromophores Modified Porous Clay Heterostructure for Smart Packaging Films. Paper presented at the 18<sup>th</sup> PPC Symposium in Petroleum, Petrochemical and Polymers, Bangkok, Thailand.

