

REFERENCES

1. อรรถวิมล สุขวัฒน์, นันทนา แก้วบุญ และวิภา คียมมณี รายงานผลของการศึกษา เรื่อง "การเก็บรักษาและคุณภาพเปลี่ยนแปลงของน้ำมะนาวในระยะเวลาการเก็บ" กองวิทยาศาสตร์ชีวภาพ กรมวิทยาศาสตร์ กระทรวงอุตสาหกรรม, ๒๕๐๘
2. Anusornpanich, "Preservation of Fresh Limes by Controlled Atmosphere Storage and Lime Juice Concentrate Processing". Master's Thesis, Department of Chemical Technology, Graduate School, Chulalongkorn University, 1979.
3. Bissett, D.W., Tatum, J.H., Wagner, C.J., Veldhuis, M.K., Graham, R.P. and A.I. Morgan, Jr. "Foam-Mat Dried Orange Juice". Food Technology 12(1963): 92-95.
4. Notter, G.K., Taylor, D.H., and Brekke, J.E. "Pineapple Juice Powder" Food Technology 12(1958): 363-366.
5. Ammu, K., Radhakrishna, K., Subramanian, V., Sharma, T.R. and Nath, H. "Storage Behavior of Freeze Dried Fruit Juice Powder". Journal of Food Technology 12(1977): 541-544.
6. Nagy, S., Shaw, P.E., Matthew and Veldhuis Citrus Science and Technology. Vol.1. Westport, Connecticut. The AVI Publishing Company, Inc., 1977.

7. Chanda, S.K., Hirst, E.L., and Perceval, E.G.V. "The constitution of orange cell-wall. Journal of Chemical Society 1240-1246. Cited in (6).
8. Schulman, Y., and Monselise, S.P. "Some Studies of the Cuticular Wax of Citrus Fruits". Journal of Hortical Science. 45 (1970): 471-478. Cited in (6).
9. Albrigo, I.G. "Distribution of Cutomata and Epicuticular wax on Oranges as related to Stem and rind Breakdown and Water loss". Journal of the American Society of Hortical Science. 97 (1972):220-223. Cited in (6).
10. U.S.Dep.Agric. Chemistry and Technology of Citrus, Citrus Products and Byproducts. U.S.Dep.Agric., Agric. Handbook, 1962: 98. Cited in (6).
11. Scott, W.C., Kew, T.J., and Veldhuis, M.K. "Composition of Orange Juice cloud". Journal of Food Science 30(1965): 883-837.
12. Curl, A.L., and Veldhuis, M.K. "The Composition of Sugars in Florida Valencia Orange Juice". Products Journal 27 (1948): 342-343, 361. Cited in Nagy, Shaw and Veldhuis, 1977.
13. Stepak, Y., and Lifshitz, A. "Identification and Determination of Sugar in Some Fruit Juices". Journal of Association Official and Analytical Chemistry 54 (1971): 1215-1217. Cited in (6).

14. U.S. Department of Health education and Welfare, Food Composition Table for use in East Asia. Food and Agriculture Organization of the United Nation, Food Policy and Nutrition Division. Public Health Service, 1972.
15. Joslyn, M.A., and Heid, J.L. "Food Processing Operations" vol 3. Westport, Connecticut. The AVI Publishing Company, Inc., (1964): 309-334.
16. Heikal, H.A., Manawaty, H.I., Shaker, G. and Gamali, L. "Concentration of Citrus Juice. I. Factors affecting the Quality and Stability of Concentrated Lime Juice by Vacuum Method". Agricultural Research Review 50(1972): 139-147.
17. Pruthi, J.S., Chakraborty, R.N., Sandhi, S.P., Sastry, L.V.L. and Siddappa, G.S. "Studies on Concentrating the Juice of the Cashew Apple". Food Technology 17; 11(1963): 95-98.
18. Muller, J.G. "Freeze Concentration of Food Liquids Theory, Practice, and Economics" Food Technology 21 (January 1967): 49-60.
19. Joslyn, M.A. and Tressler, D.K. Fruit and Vegetable Juice Processing Technology. The AVI Publishing Company Inc., 1961.
20. Notter, G.K., Taylor, D.H. and Brekke, J.E. "Orange Juice Powder" Food Technology 12(1958): 113.

21. Schultz, T., Dinick, K. and Makower, B. "Incorporation of Natural Fruit Flavors into Fruit Juice Powders" Food Technology (1956) 10: 57-59.
22. Schultz, T. and Talburt, W. "Preparation of Locked-in Citrus Oils with "Mixed Sugars" Food Technology (1961) 15: 188-190.
23. Sugigawa, Kilson and Moyls "Determination of Equilibria in Dehydrated foods". Food Technology 15(1961): 536-539.
24. Thijssen, H.A.C. and Kulken, W.H. "Retention of Aromas in Drying Food Liquids" Ingenieur 80, 47 (1968): 45-56.
25. Flink, J. and Karel, M. "Retention of Organic Volatiles in Freeze-Dried Solutions of Carbohydrates" Journal of Agricultural and Food Chemistry 18: 2 (1970): 295-297.
26. Monzini, A., Maltini, E. "An approach to the Freeze Drying of frozen concentrated Lemon Juices" XIII Congress Intern. Inst. Refrig., Comm. X, Washington, 1971.
27. Maltini, E. "Thermophysical Properties of Frozen Lemon Juice related to Freeze Drying Problems" Annali dell' Istituto Sperimentale per la Valorizzazione Tecnologica dei Prodotti Agricoli 5 (1974): 65-72.
28. Kopelman, I.J., Meydav, S. and Weinberg, S. "Storage Studies of Freeze Dried Lemon Crystals" Journal of Food Technology 12 (1977): 403-405.

29. Notter, G.K., Taylor, D.H. and Downes, W.J. "Orange Juice Powder" Food Technology 13(1959): 113-116.
30. Makower, B., and Dye, W.B. "Equilibrium Moisture Content and Crystallization of Amorphous Sucrose and Glucose" Journal of Agricultural and Food Chemistry 4;1 (1956): 72-73.
31. Eskin, N. A. M., Handerson, H.M. and Townsend, R.J. "Browning Reactions in Foods" Biochemistry of Foods. Academic Press; New York. (1971): 69-108.
32. Shaw, P.E., Tatum, J.H., Kew, T.J., Wagner, C.J. and Berry, R.E. Journal of Agricultural and Food Chemistry 18 (1970):343-349.
33. Karel, M. and Nickerson, J.T.R. "Effects of Relative Humidity, Air, and Vacuum on Browning of Dehydrated Orange Juice". Food Technology (1964)18: 104-108.
34. Lea, C.H. "Chemical changes in the preparation and storage of dehydrated foods in Fundamental Aspects of Dehydration of Food stuffs". Society of Chemical Industry . London(1958): 178.
35. Foda, Y.M., Hamed, M.G.E. and Abd Allah, M.A. "Fruit Products Order" Food Technology 24 (1970): 12-18.
36. Ammu, K. "Fruit Juice Powder" Journal of Food Technology 13 (1978): 541-553.

.....

..... 316

..... 9 NO 25

37. Jolyn, M.A., "Role of Amino acids in the Browning of Orange Juice". Food Research 22(1957): 1-5.
38. Flink, J., and Karel, M., "Effect of Process Variables on Retention of Volatiles in Freeze-Drying". Journal of Food Science 35 (1970): 444-447.
39. Omatete, O.O., and King, C.J., "Volatiles Retention During Rehumidification of Freeze Dried Food Models". Journal of Food Technology 13(1978): 137-144.
40. Islesias, H.A., and Chirife, J., "Delayed Crystallization of Amorphous Sucrose in Humidified Freeze Dried Model Systems". Journal of Food Technology 13(1978): 137-144.
41. Purwadaria, H.K., Heldman, D.R., and Kirk, J.R., "Computer Simulation of Vitamin Degradation in a Dry Model Food System During Storage". Journal of Food Process Engineering 3(1979): 7-28.
42. Bernhard, R.A., and Marr, A.D., "Oxidation of Turpines I. Mechanism and Reactive Products of D-Limonene autooxidation". Food Research 25(1960): 517-530.
43. Tatum, J.H., Shaw, P.E., and Bergy, R.E., "Some Compounds formed during Nonenzymatic Browning of Orange Powder". Journal of Agricultural and Food Chemistry 15(1967): 773-779.
44. Dinsmore, H.L., and Nagy, S., "A rapid Gas Chromatographic Method for Studying Volatile Carbonyl Compounds from Orange Juice and Their Changes During Storage". Journal of Agricultural and Food Chemistry 19(1971): 517-519

45. Shaw., P.E., Tatum, T.H., and Berry, R.E. Journal of Agriculture and Food Chemistry 17(1967): 773. Cited in (28)
46. Huelin, F.B., Goggiola, I.M., Sidhu, G.S. and Kennett, B.H. Journal of Science and Food Agriculture 22(1971) 540. Cited in Papanicolaou et.al. "Volatiles in orange juice powders". Journal of Food Chemistry (1978): -519.
47. El'Ode, K.E., Dornseiffer, T.P., Keith, E.S., and Powers, J.J. "Effects of pH and Temperature on the Carbonyls and Aromas Product in Heated Amino Acid-Sugar Mixtures". Journal of Food Science 31(1966): 351-358.
48. Scalan, R.A., Kayser, S.G., Libbey, L.M., and Morgan, M.E., "Identification of Volatile Compounds from Heated L-Cysteine. HCl/D-Glucose". Journal of Agricultural and Food Chemistry 21(1973): 673-675.
49. Blair, J.S., Godar, E.M., Master, J.E. and Riestu, D.W. Food Research 17(1952).235. Cited in Papanicolaou, D., Rigaud, J., Sauvageot, P., Dubois, P., and Simatos, D. "Behavior of Some Volatile Components During Storage of Orange Juice Powder with Low and Intermediate Moisture Contents". Journal of Food Technology 13(1978): 511-519.
50. Nagy, S. and Randall, V., "Use of Furfural Content as an Index of Storage Temperature abuse in Commercially Processed Orange Juice". Journal of Agricultural and Food Chemistry 21(1973): 272-275.

51. Salwin, H. "The Role of Moisture in Deteriorative Reactions of Dehydrated Foods". Freeze-Drying of Foods International Institute of Refrigeration, Commission X, Lausanne, Switzerland. (1962): 58-74
52. Rockland, L.B. "A new treatment of Hygroscopic Equilibria application to walnuts" Food Research. 22(1957): 604-610.
53. Hodge, J.E. "Chemistry of Browning Reactions in Model Systems". Journal of Agricultural and Food Chemistry 1(1953): 928-935.
54. Marcus Karel and Labuza, T.P., "Nonenzymatic Browning in Model Systems Containing Sucrose". Journal of Agricultural and Food Chemistry. 16;5 (1968): 717-719.
55. Schoebel, T. "Sucrose Hydrolysis at Limited Water Concentration". Master's Thesis, Department of Nutrition and Food Science, Massachusetts Institute of Technology, Cambridge, Mass., 1968.
56. Draudt, R.P., and Huang, C.E., "Browning of dehydrated sulfited vegetables during storage". Food Technology 5(1951): 417.
57. Kirk, J., et al. "Degradation of Ascorbic Acid in a Dehydrated Food System". Journal of Science. 40(1974): 1274-1279.
58. Cox, H.E. and Pearson, D. The Chemical Analysis of Foods. New York, Chemical Publ. Co., 1962.
59. A.O.A.C. "Official Method of Analysis" 9th ed. Assoc. Offic. Agr. Chemists 1960.
60. Bennett, C.A., and Frankliu, N.L., Statistical Analysis in Chemistry and the Chemical Industry. John Wiley and Sons, Inc. London. Chapman and Hall, Limited. U.S.A., 1954.

APPENDIX I

FACTORIAL DESIGN (60)

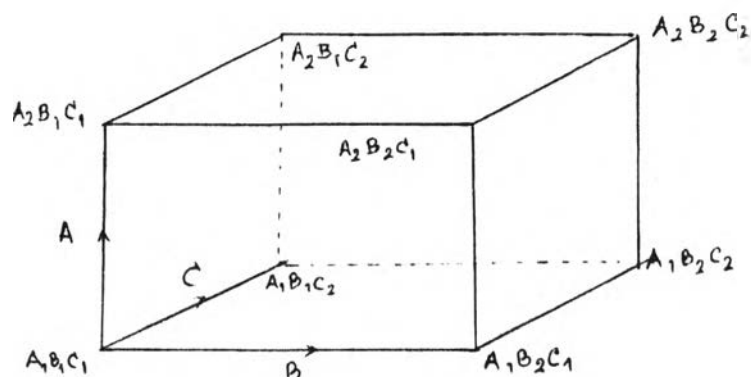
The factorial design is frequently employed by workers who have no interest in statistical analysis, because it provides a picture of a complete field of investigation rather than a detailed examination of a few selected paths. By combining the factorial method of design with a statistical method of analysis of the results it is possible to obtain information on the effects and interactions of the factors involved, together with unbiased significance tests for these factors. The conclusions obtained are likely to be of wider application because the tests carried out cover a greater range of experimental conditions.

As an illustration of the application of the two methods consider an investigation of the efficiency of an extraction column. Suppose that 2 types of packing, helices and Raschig rings, were under consideration, and that 3 levels of the concentration of salting agent were to be examined. By the classical method we should carry out 4 tests, 2 to differentiate between the types of packing at a fixed concentration, and an additional 2 to examine the 2 remaining levels of salting agent for 1 type of packing. A factorial scheme would involve 6 experimental units corresponding to the 6 possible combinations of treatments. If we suppose that each type of design were replicated and that the total number of observations by the 2 methods were equal (e.g., by 3 replicates of the classical method and 2 of the factorial method), then the advantages of the factorial

design may be summarized as

- (1) Greater efficiency in estimating the main effects.
- (2) Greater scope, since the two-factor interaction effects may be estimated.

In order to generalize these conclusions we consider the case of an experiment involving 3 factors each at 2 levels. Denoting the levels by A_1A_2 , B_1B_2 , and C_1C_2 , we have in all 8 experimental units. If 16 tests



are to be carried out, we could replicate the design ($A_1B_1C_1$, $A_2B_1C_1$, $A_1B_2C_1$, $A_2B_2C_1$, $A_1B_1C_2$) 4 times, which would enable us to calculate the overall mean effect, the 3 differences between the means of given treatment combinations, and leave $(16 - 3 - 1) = 12$ degrees of freedom available for an estimate of the residual variance. The estimates of the differences between the means would in each case relate only to the first level of the remaining factors, and the means themselves would be based on 4 observations so that the variance of the estimate of a given effect would be $(\text{estimate of residual variance})/2$.

Alternatively, the full design of 8 treatment combinations could be replicated twice, the resulting observations being illustrated

schematically in Figure 1 where each vertex of the cube represents the two observations made under these conditions. The 8 observations of the upper face of the cube differ from the 8 of the lower face only in having factor A at level 2 in place of level 1. The difference between these means then provides an estimate of the effect due to increasing the level of A from 1 to 2, and all the observations are employed in this comparison. Similarly a comparison of the mean of the 8 observations constituting the right face of the cube with the corresponding mean for the left face gives the effect of change in the level of B; the effect of a change in the level of C may be determined from the front and rear face of the cube. If we suppose that there are no real interaction effects, then the analysis of variance for this experiment takes the form of the accompanying table. The estimate of the residual variance is again based on 12 degrees of freedom, but each treatment effect is computed from means based on 8 observations, so that the estimated variance of any treatment effect is given by (estimated residual variance)/4.

Source of Estimate	D.F.	Average Value of Mean Square
Between treatments	3	$\frac{1}{6} \cdot 2 + 8 \cdot 2$
Residual	12	$\frac{1}{6}$
Total	15	

In the event that real interactions do exist they can be estimated only from the factorial design. The existence of an interaction A x B implies that the magnitude of the effect of a change of level of A (e.g., from A_1 to A_2) depends upon the level of B at which this change is achieved. If we sum the terms in corresponding positions on the front and rear faces of the cube, we obtain

$$\begin{array}{cc} (A_2B_1C_1 + A_2B_2C_1) & (A_2B_2C_1 + A_2B_2C_2) \\ (A_1B_1C_1 + A_1B_1C_2) & (A_1B_2C_1 + A_1B_2C_2) \end{array}$$

The mean value for each vertex of the resultant square is then based on 4 observations. The difference in A effects with B at the first level is given by

$$(A_2B_1C_1 + A_2B_1C_2 - A_1B_1C_1 - A_1B_1C_2)$$

and with B at the second level by

$$(A_2B_2C_1 + A_2B_2C_2 - A_1B_2C_1 - A_1B_2C_2)$$

and the interaction A x B which is the difference of these two is given by

$$(A_2B_2C_1 + A_2B_2C_2 + A_1B_1C_1 + A_1B_1C_2 - A_1B_2C_1 - A_1B_2C_2 - A_2B_1C_1 - A_2B_1C_2)$$

If we interchange the subscripts on A and B in each term of the above expression, the value of the expression is unchanged, indicating that the interactions A x B and B x A are identical and that we could have reversed the roles of A and B in determining the interaction. Similar geometrical derivations can be given for the remaining two-factor interactions, and for the three-factor interaction ABC, but these will be deferred until an alternative nomenclature

has been developed. If the interaction effects are real, the degrees of freedom available for the estimation of the residual variance are reduced by the number necessary to estimate these effects. The scope of the analysis is then increased, but the sensitivity is reduced.

In the analysis of the factorial design it is important to consider the type of population involved when significance tests are being made. Factorial experiments frequently involve scale factors and are carried out to investigate conditions at a number of fixed levels of these factors. Since the inference to be drawn in this case does not relate to a population of possible levels from which those selected represent a random sample, significant interactions should not be used for testing effects of lower order without careful consideration of the individual problem and of the average value of the mean squares involved.

A factorial design involving k factors each at 2 levels will include a total of 2^k treatment combinations. In such an experiment each main effect, and every possible interaction between factors, will have 1 degree of freedom in the analysis of variance, a total of $2^k - 1$ degrees of freedom in all. The analysis of the treatment effects in a factorial experiment can be made by breaking down the treatment sum of squares into components each with a single degree of freedom, and proceeding with an analysis of variance. In the case of the 2^k design it is possible to substitute a more direct method which gives a better understanding of the results of the

experiment. We shall illustrate this method for the case of a 2^3 design, and for this purpose we adopt a notation which has been generally used in this type of experiment.

Let us designate by a, b, and c the presence of the factor A, B, and C at the second level, assigning the term arbitrarily in cases where 2 distinct categories are involved. Hence the treatment a would consist of factor A at the second level and factor B and C at the first level, and the treatment ab would consist of A and B at the second level and C at the first level. If we denote the combination in which all factors are at the first level by (1), then the possible treatment combinations is a 2^3 factorial design are given by (1), a, b, c, ab, ac, bc, and abc. We also designate the result or total for a given experimental unit by these symbols. The letters A, B, C, AB, AC, BC, ABC are used to denote the totals of the effects due to these factors, and also the experimental estimates of these totals. Thus the estimate of the total effect of A would be the difference of the sum of observations from experimental units involving treatment a and the sum of observations from units not involving this treatment :

$$A = abc + ab + ac + a - bc - b - c - (1)$$

and similarly

$$B = abc + ab + bc + b - ac - a - c - (1)$$

$$C = abc + ac + bc + c - ab - a - b - (1)$$

Although these totals are generally described as, e.g., "The effect of A . . . , " they do not represent the mean effect on the

observations of moving from the lower to the upper level of A, but 4 times this effect in a 2^3 design. In the 2^k design there are 2^{k-1} experimental units at the upper level of A and a similar number at the lower level. The difference between the sums of these two groups therefore represents 2^{k-1} times the mean effect of a change of level.

With the above notation the interaction effect AB is given by the difference between the effect of A at the upper level of B and the effect of A at the lower level of B, that is:

$$\begin{aligned} AB &= (abc - bc + ab - b) - (ac - c + a - (1)) \\ &= abc + ab + c + (1) - ac - bc - a - b \end{aligned}$$

and similarly

$$\begin{aligned} AC &= abc + ac + b + (1) - ab - bc - a - c \\ BC &= abc + bc + a + (1) - ab - ac - b - c \end{aligned}$$

As in the previous treatment the same results are obtained if we interchange the roles of the factors involved in the interactions. To obtain the single three-factor interaction we first consider the estimate of the effect due to A for each of the 4 combinations of B and C. These are $(abc - bc)$, $(ab - b)$, $(ac - c)$, and $(a - (1))$. The AB interaction estimates computed separately for the two levels of C are $(abc - bc) - (ac - c)$ and $(ab - b) - (a - (1))$, and the difference between these expressions represents the effect of the level of C upon the AB interaction, i.e., the ABC interaction. Thus we have

$$\begin{aligned} ABC &= (abc - bc) - (ac - c) - (ab - b) + (a - (1)) \\ &= abc - ab - ac - bc + a + b + c - (1) \end{aligned}$$

Since the interchange of the symbols a, b, and c does not affect the estimate ABC, we should obtain the same result by interchanging the roles of A, B, and C in the derivation of this effect. This is generally true for interaction of any order. As with the main effects, the estimates of the interaction effects obtained by this method represent a comparison between observation totals, and in order to estimate the mean effect we divide these totals by 4 in the case of a 2^3 design, or by 2^{k-1} in the case of a 2^k design.

Table 1 gives the signs which must be employed in combining the observations from the experimental units to obtain estimates of the main effects and interactions.

TABLE 1

Treatments	(1)	a	b	ab	c	ac	bc	abc
Effects								
1	+	+	+	+	+	+	+	+
A	-	+	-	+	-	+	-	+
B	-	-	+	+	-	-	+	+
AB	+	+	-	+	+	-	-	+
C	-	-	-	-	+	+	+	+
AC	+	-	+	-	-	+	-	+
BC	+	+	-	-	-	-	+	+
ABC	-	+	+	-	+	-	-	+

The numerical coefficient is unity in each instance so that the comparisons are orthogonal, and therefore independent. The contribution of each effect to the total sum of squares may be obtained by squaring the effect estimates and dividing by the sum of squares of the coefficients involved in the comparison. In a 2^3 design this divisor is 8, since 8 experimental units are involved and each is employed in every comparison. More generally, for a 2^k design the divisor is 2^k .

If a design of this type is carried out in r replicates, the difference between replicates within experimental units provides an estimate of the inherent variability, and the effect estimates may be obtained from Table 1 by employing the sums of the observations for given experimental units as the factors in the comparison. In this case the overall mean will be estimated as $1/2^k r$, and the remaining effects and interactions will be estimated by dividing the total effect estimates by $2^{k-1} r$. The divisor for the components of the sum of squares will be equal to the total number of observations, i.e., $2^k r$.

Example. The data of Table 2 represent results from a factorial experiment on a spinning band laboratory fractionating column. The factors involved in the design are:

- A. Clearance between band and static tube (two levels).
- B. Boil-up rate (two levels).
- C. Rate of rotation of band (two levels).

The observations represent the number of equivalent theoretical plates as determined by computation from the refractive index of the still and condenser lipids and the known characteristics of

the binary mixture examined.

TABLE 2

A. Band clearance	0.05 inch				0.1 inch			
B. Boil-up rate	1		2		1		2	
C. Band speed (rpm)	750	1500	750	1500	750	1500	750	1500
Number of equilibrium theoretical plates	11.8	20.9	8.5	16.2	9.9	18.3	8.1	16.0
Treatment combination	(1)	c	b	bc	a	ac	ab	abc

The main effects and interactions could be computed directly as, e.g.,

$$\begin{aligned}
 A &= abc + ab + ac + a - bc - b - c - (1) \\
 &= 16.0 + 8.1 + 18.3 + 9.9 - 16.2 - 8.5 - 20.9 - 11.8 \\
 &= 5.1
 \end{aligned}$$

but a systematic tabular method due to Yates ⁵² is more convenient. The original data are listed in column (2) of Table 3 in the order indicated in column (1). The first 4 entries in column (3) are the sums (i) + (ii) , (iii) + (iv) , (v) + (vi) , and (vii) + (viii) , and second 4 are the differences (ii) - (i) (iv) - (iii) , (vi) - (v) , and (viii) - (vii) , so that, for example, the term 16.6 represents (ab + b). Operating on column (3)

Table 3

(1) Treatment Combination	(2) Observation	(3)	(4)	(5) Effect	(Effect) ² /8
(1)	11.8 (i)	21.7	38.3	+109.7 I	1,504.261
a	9.9 (ii)	16.6	71.4	-5.1 A	3.251
b	8.5 (iii)	39.2	-2.3	-12.1 B	18.301
ab	8.1 (iv)	32.2	-2.8	+3.9 AB	1.901
c	20.9 (v)	-1.9	-5.1	+33.1 C	136.951
ac	18.3 (vi)	-0.4	-7.0	-0.5 AC	0.031
bc	16.2 (vii)	-2.6	+1.5	-1.9 BC	0.451
abc	16.0 (viii)	-0.2	+2.4	+0.9 ABC	0.101

in the same way gives column (4), of which the first entry is (1) + a + b + ab and the second C + ac + bc + abc. On forming column (5) from (4) in the same way we obtain for the first entry

$$c + ac + bc + abc + (1) + a + b + ab = I$$

and for the fifth

$$c + ac + bc + abc - (1) + a + b + ab = C$$

The other entries may be shown to correspond to the appropriate effects. The final column contains the contribution to the uncorrected sum of squares of each effect. Since the 8 comparisons are orthogonal, this may be obtained by squaring the effect total and dividing by the sum of squares of the coefficients. As a check on the results, the sum of squares of the original observations may be computed and compared with the total of column (7).

The formal analysis of variance for the components, each with a single degree of freedom, is given in Table 4.

Table 4

Source of Estimate	Sum of Squares	D.F.
Effect A	3.25	1
B	18.30	1
C	136.95	1
Interaction AB	1.90	1
AC	0.03	1
BC	0.45	1
ABC	0.10	1
Total (corrected for mean effect)	160.98	7

In view of the small number of tests we cannot hope to learn much about the interactions. Since $F_{1,1,0.05} = 164$, there is no reason to suppose that any of them are disproportionately large, and we pool them to provide an estimate of the residual based on 4 degrees of freedom. This proves to be 0.62, and testing the main effect components of the sum of squares against it we obtain variance ratios:

$$\begin{aligned}
 A : \quad 3.25/0.62 &= 5.2 & (F_{1,4,0.1} &= 4.5) \\
 B : \quad 18.30/0.62 &= 29.5^{**} & (F_{1,4,0.01} &= 21.2) \\
 C : \quad 137/0.62 &= 221^{***} & (F_{1,4,0.001} &= 74.1)
 \end{aligned}$$

The data provide significant evidence of an improvement in fractionation performance with increase in band speed and decrease in boil-up rate. They suggest that within the range examined the column may be more efficient as the clearance is reduced.

APPENDIX II

Determination of Citric acid content

$$\text{Citric acid (mg)/100 (ml)} = \frac{64}{2} \times \frac{\text{ml titre}}{\text{ml sample}} \times 0.1 \text{ NaOH} \times 100$$

$$\text{Normality of NaOH used is} = 0.0712$$

$$\text{ml sample used} = 2$$

$$\begin{aligned} \text{So Citric acid (mg)/100 (ml)} &= \frac{64}{2} \times \frac{\text{ml titre}}{2} \times 0.0712 \times 100 \\ &= 113.92 \times \text{ml titre} \end{aligned}$$

Appendix III

Hedonic scale

Lime juice study

Name

Date

Direction : You will be served with lime juice and will be asked to comment how much you like or dislike it. Put a check (/) at the point of scale which describe your feeling about the lime juice

	Sample No.		
	A	B	C
like extremely
like very much
like moderately
like slightly
neither like nor dislike
dislike slightly
dislike moderately
dislike very much
dislike extremely

Comments :

VITA

Miss Pranee Prakittachakul was born on 15th December 1954 in Bangkok. She received Bachelor Degree of Science in the field of Food Technology from the faculty of Science, Chulalongkorn University in 1976.

She has been working as food and drug technologist in the Food control division, Food and Drug Administration, Ministry of Public Health.

