

REFERENCES

- Adams, J. D., T. Pickett, and T. Nilsen. 1999. Arsenic treatment technology, <http://www.bioprocess.com/arsenic.htm>
- Ahmann, Dianne, A. L. Roberts, L.R. Krumholz, and Francois M.M. Morel. 1994. Microbe grows by reducing arsenic. Nature. 371: 750.
- Ahmann, D., L. R. Krumholz, H. F. Hemond, D. R. Loveley, and F. M. M. Morel. 1997. Microbial Mobilization of Arsenic from Sediments of the Aberjona Watershed. EST. 31:2923-2930.
- American Public Health Association. 1975. Standard Methods for the Examination of Water and Wastewater (14 eds.), pp. 499-507. Washington D.C. : American Public Health Association.
- Anderson, G. L., J. Williams, and R. Hille. 1992. The Purification and Characterization of arsenite oxidase from *Alcaligenes faecalis*, a Molybdenum containing Hydroxylase. J. Biol. Chem. 267(33):23674-23682.
- Astolfi, E. A. Maccagno, J C. Fernandez, R. Vaccara, and R. Stimola. 1981. Relation between arsenic in drinking water and skin cancer. Biol. Trace.El Res 3: 139-143.
- Bagla, P. and J. Kaiser. 1996. India's Spreading Health Crisis Draws Global Arsenic Experts. Science. 274: 174-175.
- Baker, M. D., W. E. Inniss, C. I. Mayfield, P. T. S. Wong and Y. K. Chau. 1983. Effect of pH on the methylation of mercury and arsenic by sediment microorganisms. Environ. Technol. Lett. 4:89-100.

- Battersloy, N.S. 1988. Sulphate—Reducing Bacteria. In B. Austin (ed.), *Method in aquatic bacteria*, pp. 269-299. NY : John Wiley & Sons.
- Belin, D. D., B. E. Dinsdale, and T. B. Altringer. 1993. Arsenic removal from mining waste waters using sulfate-reducing bacteria in a 2-stage bioreactor. *Biohydrometallurgical Technologies*. 613-620.
- Bellaek, E. 1971. Arsenic Removal from Potable Water. *JAWWA*, 63: 454-458.
- Belliveau, B. H., M. E. Starodub, C. Cotter, and J. T. Trevors. 1978. Metal resistance and accumulation in bacteria. *Biotechnol. Adv.* 5(1):102-103.
- Beveridge, T. J. and others. 1997. Metal-Microbe Interaction: Contemporary Approaches. In R. K. Poole (ed.), *Advances in Microbial Physiology*, pp. 177-243. NY: Academic Press.
- Bhattacharyya, D., A. B. Jumawan. and R. B. Grieves. 1979. Separation of Toxic Heavy Metals by Sulfide Precipitation. *Sep. Sci and Technol.* 14 (5): 441-452.
- Bhumbla, D. K., and R. F. Keefer. 1994. Arsenic mobilization and bioavailability in soils. In J. O. Nriagu (ed.), *Arsenic in the environment. I. Cycling and characterization*, pp. 51-82. NY: John Wiley & Sons, Inc.
- Blum, J. S., A. B. Bindi, J. Buzzelli, J. F. Stolz, and R. S. Oremland. 1998. *Bacillus arsenicoselenatis*, sp. nov., and *Bacillus selenitireducens*, sp. nov.: two haloalkaliphiles from MonoLake, California that respire oxyanions of selenium and arsenic. *Arch. Microbiol.* 171:19-30.

- Borgono, J. M., H. Venturiono, and P. Vicent. 1980. Clinical and epidemiological study of arsenism in northern Chile. Rev. Med. Chile. 108: 1039-1048.
- Bottino, N. R., Newman, R. D., E. R. Cox, Stokton, R. A., M. Hoban, R. A. Zingaro, and K. J. Irgolic. 1978. The effects of arsenate and arsenite on the growth and morphology of the marine unicellular algae *Tetraselmis chuii* chlorophyta and *Hymenomonas carterae* (chrysophyta). J. Exp. Mar. Biol. Ecol. 33; 153-168.
- Brannon, J. M., and W. H. Patrick, Jr. 1987. Fixation, Transformation, and Mobilization of Arsenic in Sediments. Environ. Sci. Technol. 21: 450-459.
- Brierley, C.L., J. A. Brierley, and M. S. Davidson. 1989. Applied Microbial Processes for Metals Recovery and Removal from Wastewater. In T. J. Beveridge and R. J. Doyle (eds.), *Metal Ions and Bacteria*, pp. 359-382. NY: John Wiley & Sons.
- Broer, S., G. Ji, A. Broer, and S. Silver. 1993. Arsenic efflux governed by arsenic resistance determinant of *Staphylococcus aureus* plasmid pI258. J. Bacteriol. 175(11):3480-3485.
- Bruhn, D. F., J. Li, S. Silver, F. Roberto, and B. P. Rosen. 1996. The arsenical resistance operon of InCN plasmid R46. FEMS Microbiol. Lett. 139:149-153.
- Calmon, C. 1973. Comment. J. Am. Wat. Wks. Ass. 65: 568-569.
- CasareH and Doull's. 1991. Toxic effects of metals. In Amdor, M. O, Doull John and C. D Klaassin, *Toxicology the Basic Science of Poisons* (4eds.), pp. 629-633. NY: Pergamon Press.

- Cebrian M. E., A. Albores, M. Aquilar, and E. Blakely. 1983. Chronic arsenic poisoning in the north of Mexico. Hum. Toxicol. 2: 121-133.
- Cervantes, C., G. Ji, J. L. Romerez, and S. Silver. 1994. Resistance to arsenic compounds in microorganisms. FEMS Microbiol. Rev. 15: 355-367.
- Challenger, F. 1945. Biological Methylation. Chem. Rev. 36: 315-361.
- Chen, C. M., T. K. Misra, S. Silver, and B. P. Rosen. 1986. Nucleotide sequence of the structural genes for an anion pump: the plasmid-encoded arsenical resistance operon. J. Biol.Chem. 261:15030-15038.
- Cheng, C. CV. and D.D.Focht. 1979. Production of Arsine and Methylarsines in Soil and in Culture. Appl. Environ. Microbiol. 38(3): 494-498.
- Chilvers, D. C and Peterson, P. J. 1987. Chapter 17. In Hutchison, TC. and Meema, K. M. (eds), Lead, Mercury Cadmium and Arsenic in the environment. NY: John Wiley.
- Clement, W. H., and S. D. Faust. 1981. The release of arsenic from contaminated sediments and muds. J. Environ. Sci. Health, A 16(1): 81-122.
- Cline, J. D. 1969. Spectrophotometric Determination of Hydrogen Sulfide in Natural Waters. Limno. Oceano. 14: 454-459.
- Cox, D. P., and M. Alexander. 1973a. Effect of phosphate and other anions on trimethylarsine formation by *Candida humicola*. Appl. Microbiol. 25: 408-413.
- Cox D. P., and M. Alexander. 1973 b. Production of trimethylarsine gas from various arsenic compounds by three sewage fungi. Bull. Environ. Contant. Toxicol. 9: 84-88.

- Cullen, W. R., A. E. Edman, and B. C. McBride. 1983. The identification of dimethylphenylarsine as a microbial metabolite using a simple method of chemofocusing. J. Microbiol. Method. 1: 297-303.
- Cullen, W. R, B. C. McBride, and A. W. Pickett. 1979. The transformation of arsenicals by *Candida humicola*. Can. J. Microbiol. 25: 1201-1205.
- Cullen, W. R, B. C. McBride, and A. W. Pickett, and J. Reglinski. 1984. The wood preservative chromated copper arsenate is a substrate for trimethylarsine biosynthesis. Appl. Environ. Microbiol. 47(2): 443-444.
- Cullen, W. R., B. C. McBride, and M. Reimer. 1979. Induction of the Aerobic Methylation of Arsenic by *Candida humicola*. Bull. Environ. Contam. Toxicol. 21: 157-161.
- Cullen, W. R. and others. 1977. The aerobic methylation of arsenic by microorganisms in the presence of L-methio nine-methyl-d3. J. Organometal. Chem. 139: 61-69.
- Cullen, W. R. and K. J. Reimer. 1989. Arsenic Speciation in the Environment. Chem. Rev. 89: 713-764.
- Diorio, C., J. Cai, J. Mormor, R. Shinder, and M. S. Dubow. 1995. An *Escherichia coli* chromosomal ars operon homolog is functional in arsenic detoxification and is conserved in gram-negative bacteria. J. Bacteriol. 177(8):2050-2056.
- Dowdle, P. R., A. M. Laverman, and R. S. Oremland. 1996. Bacterial Dissimilatory Reduction of Arsenic (V) to Arsenic (III) In Anoxic Sediments. Appl. Environ. Microbiol. 62(5): 1664-1669.
- Ehrlich, H. L. 1997. Microbes and metals. Appl. Microbiol. Biotechnol. 48: 687-692 .

- Ferguson, J. F., and J. Gavis. 1972. A review of the arsenic cycle in natural waters. Water Res. 6: 1259-1274.
- Francis, J., C. J. Dodge, W. Rose, A. J. Ramirez. 1989. Aerobic and Anaerobic Microbial Dissolution of Toxic Metals from Coal Wastes: Mechanism of Action. Environ. Sci. Technol. 23(4): 435-441.
- Gadd, G. M. 1992. Microbial control of heavy metal pollution. In J. C. Fry, and G. M. Gadd, Microbial control of pollution, pp. 59-88. Cambridge University Press.
- Guha, S. and M. Chaudhuri. 1990. Removal of arsenic (III) from groundwater by low-cost materials. Asian. Environ. 12(1): 42-50.
- Gupta, S. K. and KY. Chen. 1978. Arsenic removal by adsorption. J.WPCF 50: 493-506.
- Hamasaki, T., H. Nagase, Y. Yoshioka, and T. Sato. 1995. Formation, Distribution, and Ecotoxicity of Methylmetal of Tin, Mercury, and Arsenic in the Environment. Crit. Rev. Environ. Sci. Technol. 25(1): 45-91.
- Hammack, R. W., and H. M. Edenborn. 1992. The removal of nickel from mine waters using bacterial sulfate reduction. Appl. Microbiol. Biotechnol. 37:674-678.
- Hanaoka, K., S. Hasegawa, N. Kawabe, S. Tagawa and T. Kaise. 1990. Aerobic and anaerobic degradation of several arsenicals by sedimentary micro-organisms. Appl. Organomet. Chem. 4: 239-243.
- Hansen, T. A. 1994. Metabolism of sulfate-reducing prokaryotes. Antonie van leeuwenhoek 66: 165-185.

- Hassler, R. A., D. A. Klein, and R. F. Meglen. 1984. Microbial Contributions to Soluble and Volatile Arsenic Dynamics in Retorted Oil Shale. J. Environ. Qual. 13(3): 466-470.
- Hering, J. G., P. Y. Chen, J. A. Wilkie, M. Elimelech, and S. Liang. 1996. Arsenic removal by ferric chloride. Journal AWWA, 88(4): 155-167.
- Hespell, R. B. 1990. Isolation of Anaerobic Microorganisms. In D. P. Labeda, Isolation of Biotechnological Organisms from Nature, pp. 117-140. NY: McGraw-Hill Publishing Company.
- Honschopp, S., N. Brunken, A. Nehr Korn, and H. J. Breunig. 1996. Isolation and Characterization of a new arsenic methylating bacterium from soil. Microbiol. Res. 151: 37-41.
- Hsia, T.H. and K. L. Lo. 1990. The pollution problems and treatment methods of arsenic in water. Water supply. 86:32-44.
- Huysmans, K. D. and W. T. Frankenberger. 1990. Arsenic resistant microorganisms isolated from agricultural drainage water and evaporation pond sediments. Water, Air, Soil, Pollut. 53:159-168.
- Huysmans, K. D. and W. T. Frankenberger, Jr. 1992. Microbial Resistance to Arsenic. In I. K. Iskander and S. M. Selin (eds), *Engineering Aspects of Metal-Waste Management*, pp. 93-116. London: Lewis Publishers.
- Iverson, W. P. and Brinkman, F. E. 1978. Microbial metabolism of heavy metals. In: Mitchell, R., *Water Pollution Microbiology* (ed.), Vol.2, pp. 201-232. NY: John Wiley.
- Ji, G. and Silver, S. 1992. Reduction of arsenate to arsenite by Ars C protein of the arsenic resistance operon of *Staphylococcus aureus* plasmid pI258. Proc.Natl.Acad.Sci. 89: 9474-9478.

- Ji, G. and Silver, S. 1992. 1995. Bacterial resistance mechanisms for heavy metals of environmental concern. J. Ind. Microbiol. 14: 61-75.
- Johnson, D. L. 1971. Simultaneous Determination of Arsenate and Phosphate in Natural Waters. Environ. Sci-Technol. 5(5): 411-414.
- Kaur, P., and B. P. Rosen. 1992. Plasmid-Encoded Resistance to arsenic and Antimony. Plasmid 27: 29-40.
- Kirt Ottmer. 1978. The Encyclopedia of the Chemical Technology, V. 33, pp. 243-251. USA: John Wiley & Sons.
- Korte, N. E., and Q. Fernando. 1991. A Review of Arsenic (III) in Groundwater. Crit. Rev. Environ Cont. 21(1): 1-39.
- Krafft, T. and J. M. Macy. 1998. Purification and characterization of the respiratory arsenate reductase of *Chrysiogenes arsenatis*. Eur. J. Biochem. 255; 647-653.
- Laverman, A. M. and others. 1995. Growth of Strain SES-3 with Arsenate and Other Diverse Electron Acceptors. Appl. Environ. Microbiol. 61 (10): 3556-3561.
- Lemmo, N. V., S. D. Faust, T. Belton, and R. Tucker. 1983. Assessment of the chemical and biological significance of arsenical compounds in a heavily contaminated watershed. Part I. The fate and speciation of arsenical compounds in aquatic environments-A Literature review. J. Environ. Sci. Health A18(3): 355-387.
- Ljungdahl, L. G. and J. Wiegel. 1986. Working with Anaerobic Bacteria. In A. L. Demain and N. A. Solomon (eds.), *Manual of Industrial Microbiology and Biotechnology*, pp. 84-96. Washington, D. C: American Society for Microbiology.

- Logsdon, G. S. and Symons, J. M. 1973. Removal of heavy metals by conventional treatment. Conference on Traces of Heavy Metals in Water: Removal Processes and Monitoring, pp. 15-16. New Jersey: Princeton University.
- Logsdon, G. S., T. J. Sorg and J. M. Symons. 1974. Removal of Heavy Metals by Conventional Treatment, Proc. 16th Water Quality Conference, pp. 111-133. USA: Univ. of Illinois.
- Lovely, D. R. 1995. Bioremediation of organic and metal contaminants with dissimilatory metal reduction. J. Ind. Microbiol. 14: 85-93.
- Lovley, D. R., and E. J. P. Phillips. 1988. Novel Mode of Microbial Energy Metabolism: Organic Carbon Oxidation Coupled to Dissimilatory Reduction of Iron or Manganese. Appl. Environ. Microbiol. 54(6): 1472-1480.
- Lovley, D. R., and E. J. P. Phillips. 1992. Reduction of Uranium by *Desulfovibrio desulfuricans*. Appl. Environ. Microbiol. 58(3): 850-856.
- Lovely, R. D., and J. D. Coates. 1997. Bioremediation of metal contamination. Current Opinion in Biotechnol. 8(3): 285-289.
- Macy, J. M., K. Nunan, K. D. Hagen, D. R. Dixon, P. J. Harbour, M. Cahill, and L. I. Sly. 1996. *Chrysiogenes arsenatis* gen. Nov., sp. Nov., a New Arsenate-Respiring Bacterium Isolated from Gold Mine Wastewater. Int. J. Syst. Bacteriol. 46(4): 1153-1157.
- Maeda, S. 1994. Biotransformation of arsenic in the freshwater environment. In J. O. Nriagu (ed.), Arsenic in the environment. I. Cycling and characterization, pp. 155-187. NY: John Wiley & Sons.

- Maeda, S., S. Nakashima, and T. Takeshita 1985. Bioaccumulation of Arsenic by Freshwater Algae and the Application to the Removal of Inorganic Arsenic from an Aqueous Phase. Part II. By *Chlorella vulgaris* Isolated from Arsenic-Polluted Environment. Sep. Sci. Technol. 20(223): 153-161.
- Maedda, S, A. Ohki, K. Miyahara, T. Takeshita., and S. Higashi. 1990. Growth characteristics and arsenic metabolism of two species of arsenic-tolerant bacteria. Appl. Organomet. Chem. 4: 245-250.
- McBride, B. C., Merilees, H., Cullen, W. R., and PickeH, W. 1978. Anaerobic and aerobic alkylation of arsenic. In Brickman, F. B. and Bellan J. M., *Organometal and Organometalloid: Occurrence and Fate in the Enuironment: ACS Symposium Series 82*, (eds), pp. 94-129. NY: American Chemical Society.
- McBride, B. C., and R. S. Wolfe. 1971. Biosynthesis of Dimethylarsine by *Methanobacterium*. Biochem. 10(23): 43312-4317.
- McClung, L. S. and R. B. Lindberg. 1957. The Study of Obligately anaerobic Bacteria. In M. J. Pelczar, Jr., *Manual of Microbiological Methods*, pp. 120-139. NY: McGraw-Hill Book Company.
- Merrill, D. T., M. A. Manzione, J. J. Peterson, D. S. Parker, W. Chow, A. O. Hobbs. 1986. Field evaluation of arsenic and selenium removal by iron coprecipitation. JWPCF 58(1): 18-26.
- Murphy, J. and J. P. Riley. 1986. A modified single solution method for the determination of phosphate in natural waters. Anal. Chim. Acta. 27: 31-36.
- National Academy of Science. 1977. *Medical and Biologic Effects of Environmental Pollutants: Arsenic*, Washington D. C.

- Nealson, K. H., and D. A. Saffarini. 1994. Iron and manganese in anaerobic respiration: environmental significance, physiology, and regulation. Ann. Rev. Microbiol. 48:311-343.
- Newman, D. K., D. Ahmann, and Francois M. M. Morel. 1998. A Brief Review of Microbial Arsenate Respiration. Geomicrobiol. J. 15: 255-268.
- Newman, D. K., T. J. Beveridge, and Francois M. M. Morel. 1997a. Precipitation of Arsenic Trisulfide by *Desulfotomaculum auripigmentum*. Appl. Environ. Microbiol. 63(5): 2022-2028 .
- Newman, D. K. and others. 1997b. Dissimilatory arsenate and sulfate reduction in *Desulfotomaculum auripigmentum* sp. Nov. Arch. Microbiol. 168: 380-388.
- O'Neill, P. 1995. Arsenic, p. 105-121. In Alloway, B. J. (ed.), Heavy metal in soil. Blackie Academic & Professional, London.
- Oremland, R. S. and others. 1994. Isolation, Growth, and Metabolism of an Obligately Anaerobic, Selenate-Respiring Bacterium, Strain SES-3. Appl. Environ. Microbiol. 60(8):3011-3019.
- Osborne, F. H., and H. L. Ehrlich. 1976. Oxidation of Arsenite by Soil Isolate of *Alcaligenes*. J. Appl. Bacteriol. 41: 295-305
- Panchanaddikar, V. V., and R. N. Kar. 1993. Precipitation of copper using *Desulfovibrio* sp. World. J. Microbiol. Biotechnol. 9: 280-281.
- Phillips, S. E., and M. L. Taylor. 1976. Oxidation of Arsenite to Arsenate by *Alcaligenes faecalis*. Appl. Environ. Microbiol. 32(3): 392-399.
- Quinn, J. P., and G. McMullan. 1995. Carbon-arsenic bond cleavage by a newly isolated Gram-negative bacterium, strain ASV2. Microbiol. 41: 721-727.

- Rittle, K. A., J. I. Drever, and P. J. S Colberg. 1995. Precipitation of Arsenic During Bacterial Sulfate Reduction. Geomicrobiol.J. 13: 1-11.
- Sanford, R. A. and D. A. Klen. 1988. Environmental bioremediation for organometallic compounds: microbial growth and arsenic volatilization from soil and retorted shale. Appl. Organomet. Chem. 2: 159-169.
- Shariatpanahi, M. and others. 1981. Biotransformation of the pesticide sodium arsenate. J. Environ. Sci. Health. B16(1): 35-47.
- Shen, Y. S. 1973. Study of Arsenic Removal From Drinking Water. JAWWA. 65: 543-548.
- Silver, S. 1996. Bacterial resistances to toxic metal ions-a review. Gene. 179: 9-19.
- Silver, S. 1994. Exploiting heavy metal resistance systems in bioremediation. Res.-Microbiol. 145(1): 61-66 .
- Silver, S. and others. 1996. MicroReview: Orphan enzyme or patriarch of a new tribe : the arsenic resistance ATPase of bacterial plasmids. Molec. Microbiol. 8(4): 637-642.
- Silver, S. and others. 1981. Inducible plasmid determined resistance to arsenate, arsenite, and antimony(III) in *Escherichai coli* and *Staphylococcus aureus*. J. Bacteriol. 146 : 983-996.
- Silver, S. and M. Walderhaug. 1992. Gene Regulation of Plasmid-and Chromosome-Determined Inorganic Ion Transport in Bacteria. Microbiol. Rev. 56(1): 195-228.
- Silver, S., T. K. Misra, and R. A. Laddaga. 1989. Bacterial Resistance to Toxic Heavy Metals. In T. J. Beveridge and R. J. Doyle (eds.), *Metals Inos and Bacteria*, pp. 121-140. NY: John Wiley&Sons.

- Sorg, T. J. and G. S. Logsdon. 1978. Treatment Technology to Meet the Interim Primary Drinking water Regulations for Inorganics: Part 2. Jurnal AWWA, 70: 379-393.
- Spliethoff, H. M., R. P. Mason, and H. F. Hemond. 1995. Interannual Variability in the Speciation and Mobility of Arsenic in a Dimictic Lake. Environ. Sci. Technol. 29: 2157-2161.
- Strickland, J. D. H and T. R. Parsons. 1968. A Practical Handbook of Seawater Analysis. Fish. Res. Board. Can. Bull. 167, Queens Printer, Ottawa. p. 49-62.
- Summers, A. O., and S. Silver. 1978. Microbial Transformations of Metals. Annual Rev. Microbiol. 32: 637-672.
- Tamaki, S and W. T. Frankenberger Jr. 1992. Environmental Biochemistry of Arsenic. Rev. Environ. Contam. Toxicol. 124: 79-110.
- Thomson, G. A., and R. J. Watling. 1987. Bioaccumulation Potential of Heterotrophic Bacteria for Lead, Selenium, and Arsenic. Bull. Environ. Contam. Toxicol. 38: 1049-1054.
- Tokunaga., S., S. A. Wasay and S. W. Park. 1997. Removal of arsenic (V) ion from aqueous solutions by lanthanum compounds. Wat. Sci. Tech. 35(7): 71-78.
- Uhrie, J. L., J. I. Drever, P. J. S. Colberg, and C. C. Nesbitt. 1996. In situ immobilization of heavy metals associated with uranium leach mines by bacterial sulfate reduction. Hydrometallurgy. 43: 231-239.
- US Environmental Protection Agency. 1993. Mercury and Arsenic Waste: Removal, Recovery, Treatment, and Disposal. Noyes Data Corporation, New Jersey. p. 59-127.

- Vidal, F. V., and V. M. V. Vidal. 1980. Arsenic Metabolism in Marine Bacteria and Yeast. Marine Biol. 60: 1-7.
- Viraraghavan, T., Y. C. Jin, and P. M. Toneta. 1992. Arsenic in water supplies. J. Environ. Studies. 41: 159-167.
- Von Endt, D. W., P. C. Kearney, and D. D. Kaufman. 1968. Degradation of Monosodium Methanearsonic Acid by Soil Microorganisms. J. Agr. Food Chem. 16(1): 17-20.
- Wakao, N., H. Koyatsu, Y. Komai, H. Shimokawara, Y. Sakurai, and H. Shiota. 1988. Microbial Oxidation of Arsenite and Occurrence of Arsenite-Oxidizing Bacteria in Acid Mine Water from a Sulfur-Pyrite Mine. Geomicrobiol. J. 6: 11-24.
- Wakao, N., H. Kumagai, H. Haga, T. Yasuda, and T. Hattori. 1995. Analysis of arsenic-tolerant bacterial populations in a paddy field soil according to rates of development on solid media. J. Gen. Appl. Microbiol. 41: 437-448.
- Wentz, C. A. 1989. Hazardous Waste Management in Chemical, Physical and Biological Treatment. McGraw-Hill Book Company, NY.
- Whang, J. J. 1982. Soluble-Sulfide Precipitation for Heavy Metals Removal from Waste waters. Environ. Progress. 1(2): 110-113.
- White, C., J. A. Sayer, G. M. Gadd. 1997. Microbial solubilization and immobilization of toxic metals: key biogeochemical processes for treatment of contamination. FEMS. Microbiol. Rev. 20: 503-516.
- WHO. 1981. Environmental Health Criteria 18: Arsenic. World Health Organization, Geneva. p. 1-41.
- Williams, J. W. and S. Silver. 1984. Bacterial resistance and detoxification of heavy metal. Enz. Microb. Technol. 6(6): 530-537.

- Williams, M., F. Fordyce, A. Pajitprapapon, and P. Charoenchaisri. 1996. Arsenic contamination in surface drainage and groundwater in part of the southeast Asian tin belt, Naknon Si Thammarat Province, southern Thailand. Environ. Geol. 27: 16-33.
- Willsky, G. R. and M. H. Malamy. 1980. Effect of Arsenate on Inorganic Phosphate Transport in *Escherichia coli*. J. Bacteriol. 144: 366-374.
- Wood, J. M., and H. K. Wang. 1983. Microbial resistance to heavy metals. Environ. Sci. Technol. 17(12): 582A-590A.
- Woolson, E. A. 1977. Generation of Alkylarsines from Soil. Weed. Sci. 25 (5): 412-416.
- Woolson, E. A., and P. C. Kearney. 1973. Persistence and Reactions of ^{14}C -Cacodylic Acid in Soils. Environ. Sci. Technol. 7(1): 47-50.
- Wrench, J. J. and Addison, R. F. 1981. Reduction, methylation, and incorporation of arsenic into lipids by the marine phytoplankton *Dunaliella tertiolecta*. Can. J. Fish. Aquat. Sci. 38:518-523.
- Xu, C., T. Zhou, M. Kuroda, and B. P. Rosen. 1998. Metalloid Resistance Mechanisms in Prokaryotes. J. Biochem. 123: 16-23.
- Zetibox, J. L. Jr., M. W. Doughten, D. J. Grimes, and R. R. Colwell. 1987. Testing for Bacterial Resistance to Arsenic in Monitoring Well Water by the Direct Viable Counting Method. Appl. Environ. Microbiol. 53 (12): 2929-2934.



APPENDIX

สถาบันวิทยบริการ
จุฬาลงกรณ์มหาวิทยาลัย

APPENDIX A

RATE OF ARSENIC AND ARSENIC COMPOUNDS IMPORTED IN THAILAND DURING 1995 TO 1999

Year	Country	Quantity/kg	Value/bath
1995	China	274200	1670723
	France	39600	904582
	Germany	13	40290
	USA	4	5360
	UK	-	238
	Total	313817	2621193
	1996	China	51850
France		39600	949396
Germany		6	20542
Total		91456	1900203
1997	China	99000	2354048
	France	59400	1763489
	Germany	192	122136
	Total	158592	4239673

Appendix A (cont.)

Year	Country	Quantity/kg.	Value/bath
1998	China	88550	3894684
	France	19800	724440
	USA	-	27081
	Total	108350	4646205
1999	China	119150	3345426
	France	19800	844695
	Belgium	16000	739727
	Korea	3	745
	USA	6	21264
	Japan	2	18218
	Total	154961	4956051

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APPENDIX B

BACTERIAL SOURCES

In this present study 50 samples were chosen as bacterial source, they were collected during May 1998 to December 1999 and were categorized into various groups as follows:

Type of samples	Site	Characteristic
Soil	S-1	Roughly sand, black
	S-2	Roughly sand, brown
	S-3	Sand, light brown
	S-4	Roughly sand, dark brown
	S-5	Clay, black
	S-6	Clay and mud, brown
	S-7	Clay, dark brown
	S-8	Clay, brown
	S-9	Clay and sand, black
	S-10	Sand, reddish brown
	S-11	Clay, dark brown
	S-12	Mud and sand, black
	S-13	Clay, black and odd odor
Sediment	Sd-1	Black mud, odd odor
	Sd-2	Black mud and humus, odd odor
	Sd-3	Clay, mud and humus
	Sd-4	Black mud, bad smell
	Sd-5	Clay and sand, brownish
	Sd-6	Black mud and humus
	Sd-7	Brown clay and black mud
	Sd-8	Mud, light brown
	Sd-9	Black mud
	Sd-10	Clay, dark brown
	Sd-11	Black mud and humus
	Sd-12	Black mud
	Sd-13	Black clay and mud
	Sd-14	Clay, brownish
	Sd-15	Black clay, oil suspension
	Sd-16	Clay, dark brown
Sd-17	Brownish mud	
Sd-18	Clay and sand, dark brown	
Sd-19	Mud and sand, black	

Type of samples	Site	Characteristic
Sediment	Sd-20	Mud and sand, black
	Sd-21	Clay and sand, black
	Sd-22	Black mud
	Sd-23	Clay and mud, dark gray
	Sd-24	Clay and mud, dark brown
	Sd-25	Mud, dark brown
	Sd-26	Sand, dark brown
	Sd-27	Mud, light brown
	Sd-28	Mud, dark brown
	Sd-29	Mud, black
	Sd-30	Mud and humus, black
	Sd-31	Clay, light brown
	Sd-32	Clay, brown
	Sd-33	Mud and sand, black
	Sd-34	Clay and sand, black
	Sd-35	Clay and sand, black
	Sd-36	Sand and small stone, black
	Sd-37	Mud, dark brown and odd odor
Sd-38	Mud and sand, black	
Waste water	Ww-1	Light brown water; mud and humus
	Ww-2	Reddish brown water and humus
	Ww-3	Black water and humus
Natural water	Nw-1	Turbidity, light brown water

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APPENDIX C: MEDIA

Freshwater minimal medium

Salt

Formula in gram per liter of distilled water

KH_2PO_4	0.14
NH_4Cl	0.25
KCl	0.50
$\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$	0.15
NaCl	1.0
$\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$	0.62

Trace elements

Formula in milligram per liter of distilled water

Conc. HCl	0.001 ml
$\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$	0.1
$\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$	0.12
ZnCl_2	0.07
H_3BO_3	0.06
$\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$	0.025
$\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$	0.015
$\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$	0.025
$\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$	1.5

Vitamins

Formula in milligram per liter of distilled water

p-Aminobonzoic Acid	0.05
Biotin	0.02
Nicotinic Acid	0.05
Calcium Pantothenate	0.05
Thiamine Hydrochloride	0.05
Pyridoxine Hydrochloride	0.1
Cyanocobalamin	0.001

Sterilized by 0.45 μ millipore fitter, freezed and stored in a refrigerator in the absence of light until used.

Reducing agent : Cysteine HCl 0.05%, NaHCO₃ 0.085%

Carbon source: Sodium lactate 10 mM.

Preparation:

The 500 ml of salt solution of ingredients was added to a 1-liter ground joint round bottom flask^a and heated. When the solution was boiled, color turned from blue to reddish-pink. During boiling the solution for several minutes, the pink color rather than disappeared imply, indicating reduction. The mask was suddenly stopper with a rubber stopper, covered with cloth and tied tightly, and autoclaved at 121°C for 15 minutes. After transferring to the anaerobic chamber; sodium lactate, bicarbonate cysteine HCl, trace elements and vitamins was added. Before used, any plate or vial contained certain medium was incubated over night.

^A The round shape of the flask prevents breakage due to the sterilization pressuse.

Tryptic Soy Broth (TSB)

Formula: ingredient per liter

Bacto Tryptone	17	g
Bacto Soytone	3	g
Bacto Dextrose	2.5	g
Sodium Chloride	5	g
Dipotassium Phosphate	2.5	g
(Bacto Agar	15	g; TSA)

Final pH 7.3 ± 0.2 at 25 °C

Direction: Suspend 30 grams in 1 liter distilled or deionized water and warm slightly to dissolve completely. Sterilize at 121 - 124 °C for 15 min.

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Brewer anaerobic agar

Formula: Ingredients per liter

Bacto-Tryptone	5	g
Proteose Peptone No.3, Difco	10	g
Bacto-Yeast Extract	5	g
Bacto-Dextrose	10	g
Sodium chloride	5	g
Bacto-Agar	20	g
Sodium Thioglycollate, Difco	2	g
Sodium Formaldehyde Sulfoxylate	1	g
Resazurin, Certified	0.002	g

Direction: Suspend 58 grams in 1 liter distilled or deionized water and warm slightly to dissolve completely. Sterilize at 121 - 124 °C for 15 min.

Nutrient Gelatin (Formula: Ingredients per liter)

Bacto Beef Extract	3	g
Bacto Peptone	5	g
Bacto Gelatin	120	g

PH 6.8 at 25°C

To rehydrate the medium, suspend 128 grams in 1000 ml. Cold distilled water. Warm to about 50°C to dissolve the medium completely. Sterilize in the autoclave for 15 minutes at 15 ponds pressure (121°C).

MacConkey agar

For isolating and differentiating lactose-fermenting from lactose-non fermenting gram negative enteric bacilli.

Formula: ingredient per liter

Bacto Peptone	17	g
Bacto Proteose Peptone	3	g
Bacto Lactose	10	g
Bacto Bile Salt No.3	1.5	g
Sodium Chloride	5	g
Bacto Agar	13.5	g
Neutral Red	30	mg
Bacto Crystal Violet	1	mg

Final pH 7.1 ± 0.2 at 25°C

Direction: suspend 50 grams in 1 liter, distilled or deionize water and boil to dissolve completely. Sterilize in the autoclave for 15 minutes at 15 pounds pressure. Avoid overheating.

Motility test medium (Formula: ingredient per liter)

Beef extract	3	g
Peptone	10	g
NaCl	5	g
Agar	4	g

Final pH = 7.3

MR/VP broth

Formula: ingredient per liter

Polypeptone	7	g
Glucose	5	g
Dipotassium phosphate	5	g

Final pH 6.9 ± 0.2 **Salmonella-Shigella Agar (SS)**

Formula: ingredient per liter

Bacto Beef Extract	5	g
Bacto Proteose Peptone	5	g
Bacto Lactose	10	g
Bacto Bile Salt No.3	8.5	g
Sodium Citrate	8.5	g
Sodium Thiosulfate	8.5	g
Ferric Citrate	1	g
Bacto Agar	13.5	g
Brilliant Green	0.33	mg
Neutral Red	25	mg

Final pH 7.0 ± 0.2 at 25°C

Direction: Suspend 60 grams in 1 liter distilled or deionized water and boil carefully for no more than 2-3 min to dissolve completely. Avoid overheating. Do not autoclave.

Simmons Citrate Agar

Formula: ingredient per liter

Magnesium Sulfate	0.2	g
Ammonium Dihydrogen Phosphate	1	g
Dipotassium Phosphate	1	g
Sodium Citrate	2	g
Sodium Chloride	5	g
Bacto Agar	15	g
Bacto Brom Thymol Blue	0.08	g

Final pH 6.8 at 25 °C

Direction: To rehydrate the medium, suspend 24.2 grams in 1L, cold freshly distilled water and heat to boiling to dissolve the medium completely. Sterilize in the autoclave for 15 minutes at 15 pounds pressure (121°C)

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Triple Sugar Iron Agar (TSI)

Formula: ingredient per liter

Bacto Beef Extract	3	g
Bacto Yeast Extract	3	g
Bacto Peptone	15	g
Proteose Peptone	5	g
Bacto Dextrose	1	g
Bacto Lactose	10	g
Saccharose	10	g
Ferrous Sulfate	0.2	g
Sodium Sulfate	5	g
Sodium Thiosulfate	0.3	g
Bacto Agar	12	g
Bacto Phenol Red	24	mg

Final pH 7.4 at 25 °C

Direction: To rehydrate the medium, suspend 65 grams in 1000 ml, cold freshly distilled water and heat to boiling to dissolve the medium completely. Sterilize in the autoclave for 15 minutes at 15 pounds pressure. Allow the tubes to solidify in a slanting position in a manner which will give a generous butt.

APPENDIX D

Equilibrium	Equation	pK
Arsenic (ortho)		
$\text{H}_3\text{AsO}_4 + \text{H}_2\text{O}$	$\text{H}_3\text{O}^{++} + \text{H}_2\text{AsO}_4^-$	3.6
$\text{H}_2\text{AsO}_4^- + \text{H}_2\text{O}$	$\text{H}_3\text{O}^{++} + \text{HAsO}_4^{2-}$	7.3
$\text{HAsO}_4^{2-} + \text{H}_2\text{O}$	$\text{H}_3\text{O}^{++} + \text{HAsO}_3^{3-}$	12.5
Arsenous (meta)		
$\text{HAsO}_2 + \text{H}_2\text{O}$	$\text{H}_3\text{O}^{++} + \text{AsO}_2^-$	9.2

Source : Lemmo, et. al., 1983

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APPENDIX E

MOLYBDENUM BLUE METHOD

Reagents:

(i) Mixed reagent ;

- $(\text{NH}_4)_6\text{Mo}_7\text{O}_{24}\cdot 4\text{H}_2\text{O}$ 3 g in 100 ml distilled water
- $\text{C}_4\text{H}_4\text{KO}_7\text{SbO}\cdot 5\text{H}_2\text{O}$ 0.068 g in 50 ml distilled water
- Conc. H_2SO_4 38.9ml make volume to 100 ml in distilled water
- ascorbic acid 5.4 g in 100 ml distilled water

Mix these compound together. Prepare this reagent for use and discard any excess. Do not store for more than about 6 hr.

(ii) Reducing reagent ;

- 20 ml of 3.5 N H_2SO_4
- $\text{Na}_2\text{S}_2\text{O}_5$ 14% (w/w) in 40 ml deionized water. Prepare fres for use.
- $\text{Na}_2\text{S}_2\text{O}_3\cdot 5\text{H}_2\text{O}$ 1.4% (w/w) in 40 ml deionized water

Mix 20 ml of 3.5 N sulfuric acid solution into 40 ml of the sodium metabisulfite solution. This should be done slowly to avoid excessive bubbling caused by the liberation of SO_2 . Finally, mix in 40 ml of the sodium thiosulfate solution. Again, prepare this reagent for use and discard any excess. This reagent is stable for up to 24 hr.

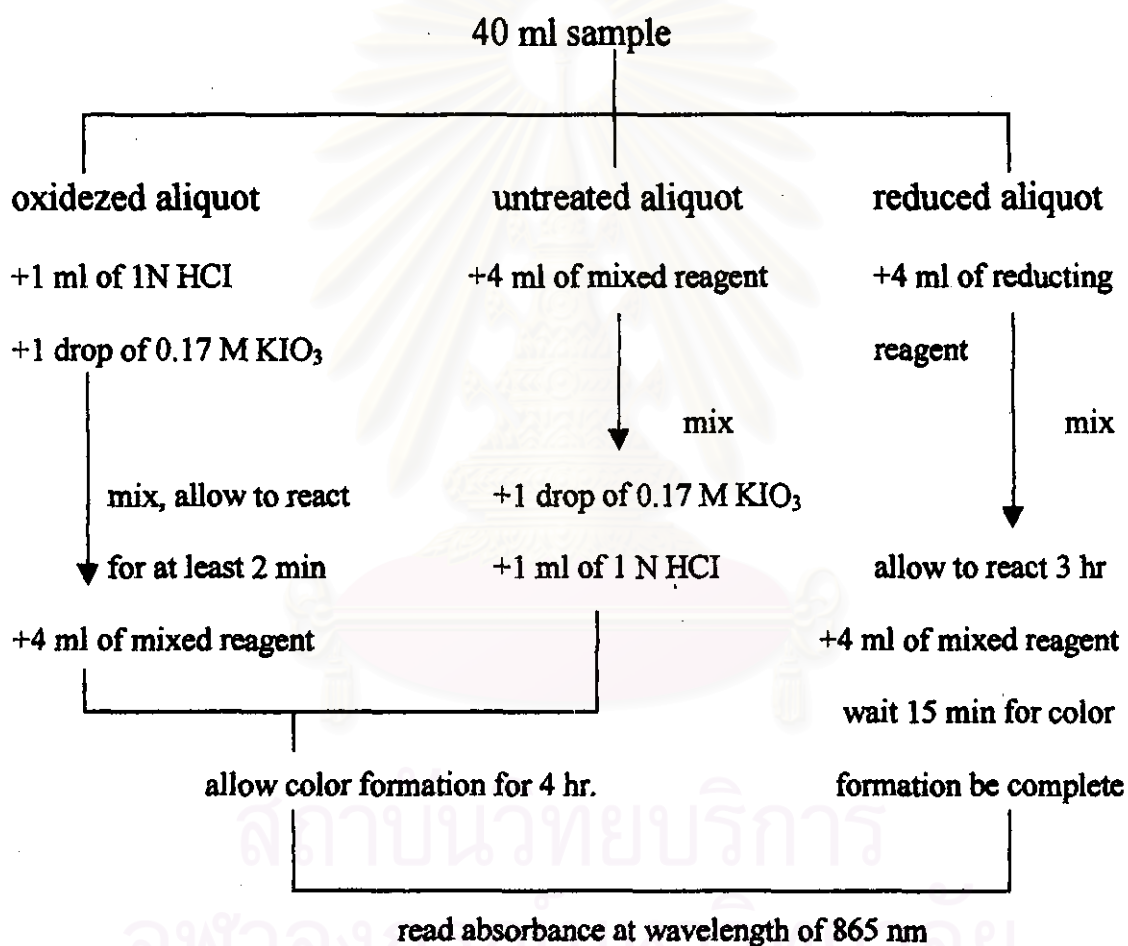
(iii) Potassium iodate solution 0.17 M.

(iv) 1 N HCl

Standard

Prepare stock solution of arsenic and phosphorus having final concentrations of 1mM from reagent grade arsenic trioxide, Sodium arsenate and potassium dihydrogen phosphate. Make appropriate dilutions of this standard solution to 0-40 μm .

Procedure



Run deionized water blank, by the above procedures, along with the samples.

Calculation

Since the final volume of the "reduced" aliquotes is greater than that of the other aliquots, a dilution correction must be made to obtain corrected absorbances. In this procedure, the correction

$$\frac{\text{Final volume "reduced"}}{\text{Final volume "oxidized or untreated"}} = \frac{48}{45} = 1.067$$

would be applied to make the "reduced" absorbances comparable to those of the other aliquots.

After dilution correction, absorbances of the blanks are subtracted to give corrected absorbance (C.A.). Absorbance factors for arsenite, arsenate and phosphate are determined from standard. Arsenite, arsenate and phosphate concentration are calculated as follows:

$$\text{Arsenite} = (\text{C.A. 'oxidized'} - \text{C.A. 'untreated'}) \times \text{abs.factor As (III)}$$

$$\text{Arsenate} = (\text{C.A. 'untreated'} - \text{C.A. 'reduced'}) \times \text{abs.factor As (V)}$$

$$\text{Phosphate} = \text{C.A. 'reduced'} \times \text{abs.factor PO}_4^{3-}$$

$$\text{* absorbance factor} = \frac{\mu \text{ moles/l}}{\text{absorbance}}$$

$$\text{absorbance factor of As(III)} = 49.12$$

$$\text{absorbance factor of As(V)} = 45.36$$

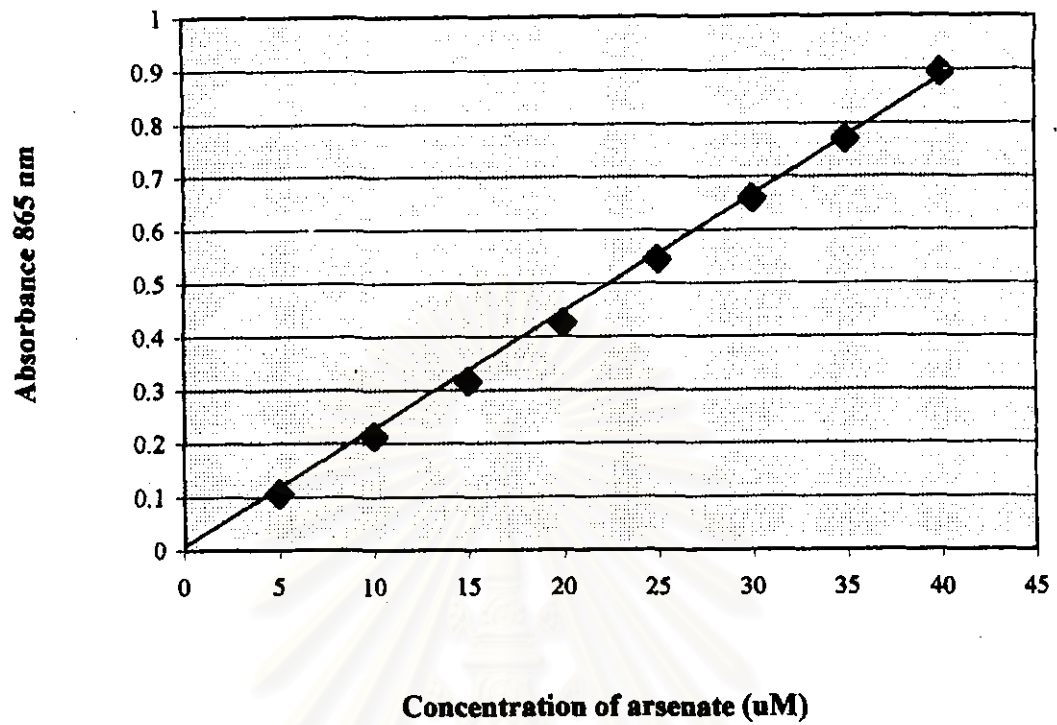
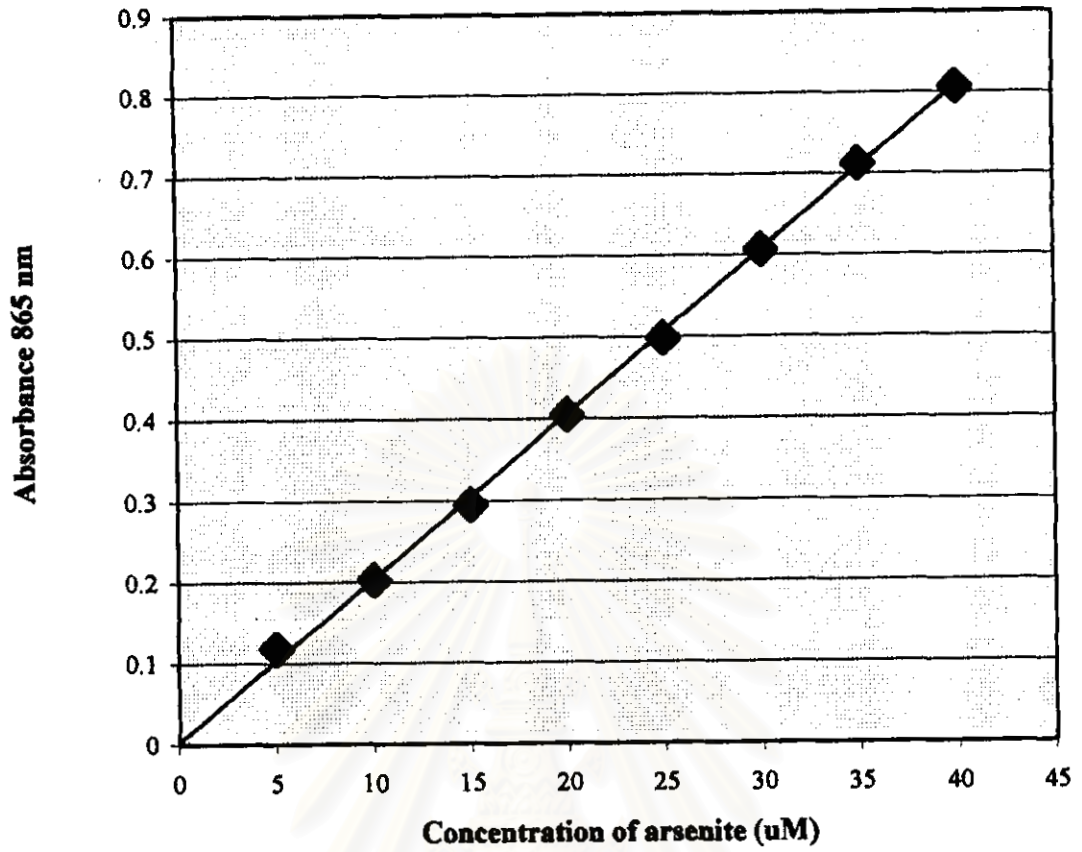


Figure E-1: Standard curve of arsenate in oxidized aliquot

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DETERMINATION OF HYDROGEN SULFIDE

Reagents:

- Mixed diamine reagent : 2 g N, N-dimethyl-p-phenylene diamine sulphate in 50 ml 50% (v/v) HCl + 3 g anhydrous ferric chloride in 50 ml 50% (v/v) HCl are mixed together immediately before analysis; if the solutions are not made fresh, they should be stored separately in dark bottles

Standard

Sulfide standard solution prepared in deoxygenated distilled water at range 0-1 mM.

Procedure

- Add 50 ml of sample to 4 ml of mixed diamine reagent, mix
- Place 1 ml of sample into 50 ml deoxygenated distilled water, and wait for 20 min.
- Read at 670 nm.
- Run deoxygenated distilled water blank, by the above procedures, along with the samples.

Calculation

$$\text{Conc. of H}_2\text{S (mM)} = F \times \text{absorbance of sample}$$

$$F; \quad \frac{\text{concentration of sulfide}}{\text{Absorbance}}$$

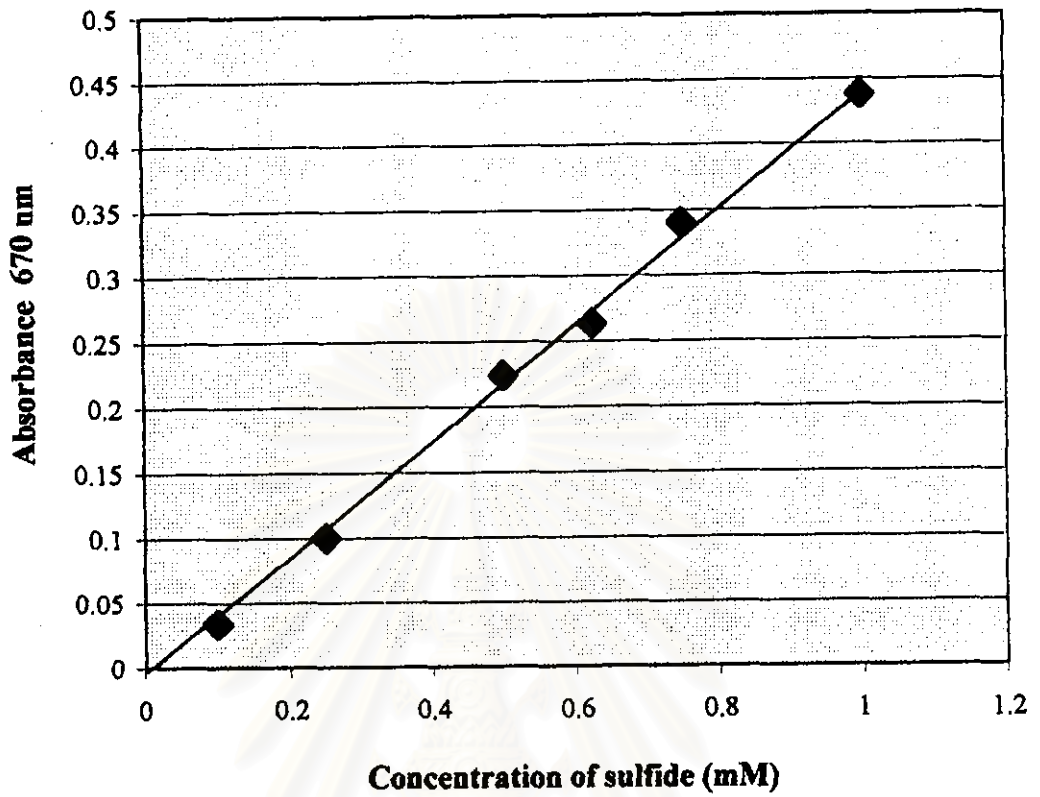


Figure E-3: Standard curve of sulfide

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Appendix F

Some characteristic of the selected bacterial isolates;
AsR-17, AsR-19 and AsR-20.

Some Characteristics	The Selected Bacterial Strains		
	AsR-17	AsR-19	AsR-20
1. Source of sample	S-2, smelting area	S-11, acrylic dye industry	
2. Type of organisms	Obligately anaerobic bacteria	Facultative anaerobic bacteria	
3. Colony	2 mm in diameter, yellowish, convex	2mm in diameter, pale to brown, clear, convex	4 mm in diameter, white, convex
4. Morphology	Rod shape, gram negative ~ 0.5 by 1.8	Rod shape, gram negative ~ 0.7 by 2.0	Rod shape, gram negative ~ 0.5 by 1.8
5. Expected genus	Unknown strain	Unknown strain	<i>Citrobacter</i>
6. Maximum resistance to arsenic: ($\mu\text{g/ml}$)			
6.1 As (V)	> 2,400	>2,400	>2,400
6.2 As (III)	> 500	< 500	<500
7. Maximum resistance to other metal ions: ($\mu\text{g/ml}$)			
7.1 Ag	ND	<100	<100
7.2 Cd	ND	<100	<100
7.3 Cr	ND	<100	<200
7.4 Cu	ND	<100	<100
7.5 Mn	ND	<800	>800
7.6 Ni	ND	<100	<100
7.7 Zn	ND	<200	<200
8. Optimum for growth			
8.1 pH	7	7	7
8.2 temperature ($^{\circ}\text{C}$)	35	40	35
9. Type of organism to precipitate arsenic	By itself	Consortium	

Appendix F (cont.)

Some Characteristics	Selected Bacterial Strains		
	AsR-17	AsR-19	AsR-20
10. Optimum condition for arsenic precipitation			
10.1 arsenic, As(V) concentration; ($\mu\text{g/ml}$)	100		200
10.2 pH	7		7
10.3 temperature ($^{\circ}\text{C}$)	35		35
11. Percentage of arsenic removal; (8-day of incubation)			
11.1 arsenic, As(V) concentration ; ($\mu\text{g/ml}$)			
11.1.1 100	49.21		41.60
11.1.2 200	48.27		45.24
11.1.3 300	-		25.86
11.2 pH			
11.2.1 6	-		26.93
11.2.2 7	35.02		42.21
11.2.3 8	-		40.75
11.3 temperature			
11.3.1 30	13.12		29.29
11.3.2 35	45.08		46.24
11.3.3 40	28.45		39.68

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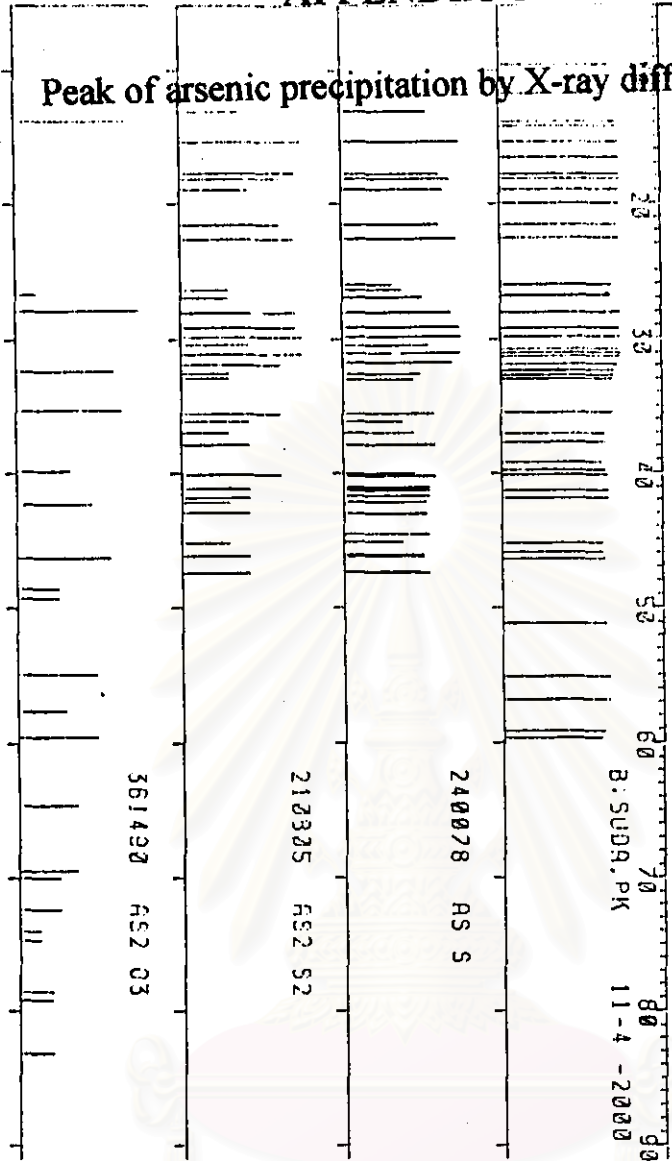
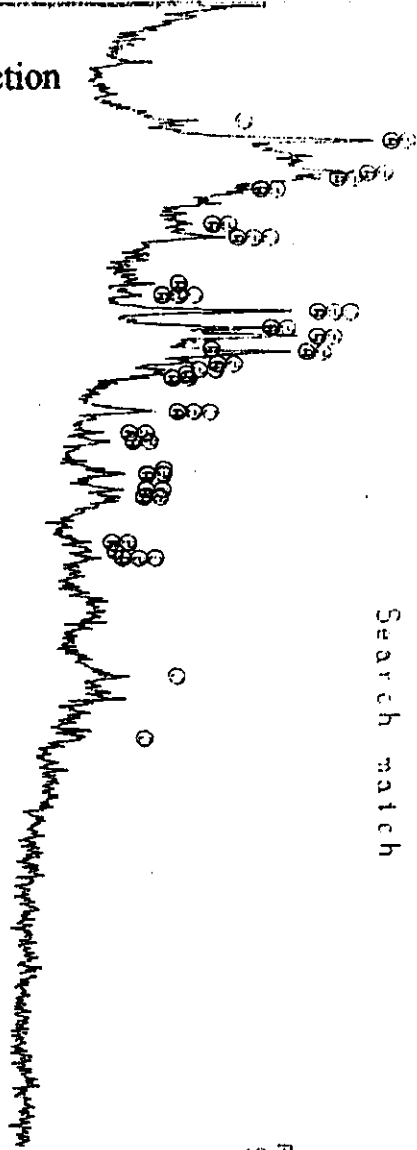
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File name : B.SUD9.PK
Sample name :

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APPENDIX G

Peak of arsenic precipitation by X-ray diffraction



ARSENOLITE, SYM

*4

0

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APPENDIX G (cont.)

COUNTS

500

250

0

5.0 15.0 25.0 35.0 45.0 55.0 65.0 75.0 85.0

ANGLE

-9.561

-7.494

-6.6375

-5.743

-5.543

-4.870.973

-4.435.667

-4.111.935

-3.31924

-3.198

-3.09336

-2.833

-2.552

-2.554

-2.4339

-2.392045

-2.199

-2.1957

-1.787

-1.666

-1.622

-1.550

-1.550

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BIOGRAPHY

Miss Suda Ittisupornrat was born in Bangkok on the 19 July 1973. She entered Kasetsart University in June 1992 and graduated a Bachelor of Science (Fisheries) in March 1995. She furthered her education at the programme of Biotechnology, Graduate School of Chulalongkorn University, in 1996.



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