

REFERENCES

1. Almog, Y., Levy, M. Free radical polymerization of styrene and methyl methacrylate in dispersed systems and the effect of carbon black. Ind. Eng. Chem. Prod. Res. Dev. 21 (1982): 163-170.
2. Lok, K. P., and Ober, C. K. Particle size control in dispersion polymerization of styrene. Can. J. Chem. 63 (1985): 209-216.
3. Lovelace, A. M., Vanderhoff, J. W., Micale, F. J., El-Aasser, M. S., and Kommfeld, D. M. Process for preparation of large-particle-size monodisperse. U.S. Pat 4,247,434 (Jan. 27, 1981).
4. Ugelstad, J. Monodisperse polymer particles and dispersions thereof. U.S. Pat 4,459,378 (July, 10, 1984).
5. Paine, A. J. Process for the preparation of toner compositions. U.S. Pat 5,304,450 (July 31, 1991).
6. Horak, D., Svec, F., and Frechet, J. M. J. Preparation of colored poly(styrene-co-butyl methacrylate) micrometer size beads with narrow size distribution by dispersion polymerization in presence of dyes. J. Polym. Sci. A. 33 (1995): 2,961-2,968.
7. Gruber, J. R., and Julin, C. P. Dry Toner Technology. In B. A. Diamond (ed.) Handbook of Imaging Materials, pp. 159-178. New York: Marcel Dekker, 1991.
8. Kamiyama, M., Maeda, M., Totsuka, H., and Hamanaka, T. Properties of polymerized toners. J. Imag. Sci & Technol. 39 (1995): 433-438.
9. Barrett, K. E. J. Dispersion Polymerization in Organic Media. New York: John Wiley & Sons, 1975. pp. 115-116.
10. Barrett, K. E. J. Dispersion Polymerization in Organic Media. New York: John Wiley & Sons, 1975. pp. 205-206.

11. Walbridge, D. J. The design and synthesis of dispersants for dispersion polymerization in organic media. In K. E. J. Barrett (ed.) Dispersion Polymerization in Organic Media. pp. 45-47. New York: John Wiley & Sons, 1975.
12. Napper, D. H., and Hunter, R. J. In M. Kerker (ed.) Surface Chemistry and Colloids. Vol. 7. London: Butterworths, 1972.
13. Barrett, K. E. J. Dispersion Polymerization in Organic Media. New York: John Wiley & Sons, 1975. pp. 130-177.
14. Fitch, R. M., and Tsai, C. H. In R. M. Fitch (ed.) Polymer Colloid I. p. 73. New York: Plenum, 1971.
15. Barrett, K. E. J., and Thomas, H. R. Kinetics of dispersion polymerization of soluble monomers. I. Methyl methacrylate. J. Polym. Sci. A. 7 (1969): 2612-2650.
16. Barrett, K. E. J. Dispersion Polymerization in Organic Media. New York: John Wiley & Sons, 1975. pp. 4-6.
17. Apiwattananon, S. Synthesis and Characterization of Novel super-fine Particle-sized Polymers. Master's Thesis Chulalongkorn University, 1994.
18. Odian, G. Principles of Polymerization. 3rd ed. New York: John Wiley & Sons, 1991. pp. 453-454.
19. Barrett, K. E. J. Dispersion polymerization in organic media. Br. Polym. J. 5 (1973): 259-271.
20. Almog, Y., Reich, S., and Levy, M. Monodisperse polymeric spheres in the micron size range by a single step process. Br. Polym. J. 14 (1982): 131-136.
21. Tseng, C. M., Lu, Y. Y., El-Aasser, M. S., and Vanderhoff, J. W. Uniform polymer particles by dispersion polymerization in alcohol. J. Polym. Sci. A. 24 (1986): 2995-3007.

22. Ober, C. K., and Hair, M. L. Monodispersed, micron-sized polystyrene particles by dispersion polymerization. J. Polym. Polym. Chem. Ed. 23 (1985): 103-108.
23. Ober, C. K., and Hair, M. L. The effect of temperature and initiator levels on the dispersion polymerization of styrene. J. Polym. Sci. A. 25 (1987): 1395-1407.
24. Williamson, B., Lukas, R., Winnik, M. A., and Croucher, M. D. The preparation of micron-size polymer particles in non polar media. J. Colloid Interface Sci. 119 (1987): 559-564.
25. Paine, A. J., Luymes, W., and McNulty, J. Dispersion polymerization of styrene in polars solvents. 6. Influence of reaction parameters on particle size and molecular weight in poly(*N*-vinylpyrrolidone)-stabilized reactions. Macromolecules. 23 (1990): 3104-3109.
26. Paine, A. J. Dispersion polymerization of styrene in polar solvents 7. A simple mechanistic model to predict particle size. Macromolecules. 23 (1990): 3109-3117.
27. Shen, S., Sudol, E. D., and El-Aasser, M. S. Control of particle size in dispersion polymerization of methyl methacrylate. J. Polym. Sci. A. 31 (1993): 1393-1402.
28. Shen, S., Sudol, E. D., and El-Aasser, M. S. Dispersion polymerization of methyl methacrylate: mechanism of particle formation. J. Polym. Sci. A. 32 (1994): 1087-1100.
29. Ho, C., Chen, S., Amiridis, M. D., and Zee, J. W. V. Dispersion polymerization of styrene in alcohol media. J. Polym. Sci. A. 35 (1997): 2907-2905.
30. Ober, C. K., and Lok, K. P. Formation of large monodisperse copolymer particles by dispersion polymerization. Macromolecules. 20 (1987): 268-273.
31. Horak, D., Svec, F., and Frechet, J. M. J. Preparation and control of surface properties of monodisperse micrometer size beads by dispersion

- copolymerization of styrene and butyl methacrylate in polar media. J. Polym. Sci. A. 33 (1995): 2329-2338.
32. Kiatkamjornwong, S., and Anantapiphat, A. Syntheses of super-fine mono-dispersed particles of poly(styrene-co-methyl methacrylate) by dispersion copolymerization. J. Sci. Soc.Thailand. 24 (1998):169-192.
33. Lok, K. P., and Ober, C. K. Stable polymeric dispersion and methods for making. U.S. Pat 4,524,199. (Nov. 9, 1983).
34. Tuncel, A., Kahraman, R., and Piskin, E. Monosize polystyrene microbeads by dispersion polymerization. J. Appl. Polym. Sci. 50 (1993): 303-319.
35. Dawkins, J. V., and Neep, D. J. Non-aqueous polystyrene dispersions:steric stabilization by partially hydrolysed poly(vinyl alcohol) in methanolic media. Polymer. 35 (1994): 5366-5368.
36. Rosen, S. L. Fundamental Principles of Polymeric Material. 2 nd ed. New York: John Wiley & Sons, 1993. p. 88.
37. Brandrup, J., Immergut, E. H. Polymer Handbook. 2 nd.ed. New York: John Wiley & Sons, 1975. pp. IV-337-341.
38. Saenz, J. M., and Ausa, J. M. Kinetics of dispersion copolymerization of styrene and butyl acrylate. Macromolecules. 31 (1998): 5125-5222.
39. Barton, A. F. M. CRC Handbook of Polymer-liquid interaction Parameters and Solubility Parameters. pp. 73-74. Florida: CRC Press, 1990.
40. Odian, G. Principles of Polymerization. 3 rd ed. New York: John Wiley & Sons, 1991. pp. 286-290
41. Stevens, M. P. Polymer Chemistry. 2 nd ed. New York: Oxford Press, 1990. pp. 204.
42. Baines, F. L., Dionisio, S., Billingham, N. C., and Armes, S. P. Use of block copolymer stabilizers for dispersion polymerization of styrene in alcoholic media. Macromolecules. 29 (1996): 3096-3102.

43. Brandrup, J., Immergut, E. H. Polymer Handbook. 2 nd.ed. New York: John Wiley & Sons, 1975. pp. III-144-145.
44. Thomson, B., Rudin, A., Lajoie, G. Dispersion copolymerization of styrene and divinylbenzene. II. Effect of crosslinker on particle morphology. J Appl. Polym. Sci. 59 (1996): 2009-2028.
45. Li, K., and Stover, H. D. H. Highly crosslinked micron-range polymer microspheres by dispersion polymerization of divinylbenzene. J. Polym. Sci. A. 31 (1993): 2473-2479.
46. Hattori, M., Sudol, E. D., and El-Aasser. M. S. Highly crosslinked polymer particles by dispersion polymerization. J. Appl. Polym.Sci. 50 (1993): 2027-2034.
47. Rosen, S. L. Fundamental Principles of Polymeric Material. 2 nd ed. New York: John Wiley & Sons, 1993. p. 109.

APPENDIX A

NUCLEAR MAGNETIC RESONANCE SPECTRA

Nuclear magnetic resonance spectroscopy was used to determine the copolymer composition. The copolymer composition was investigated by the resonance signal area of ^1H -NMR spectra for each component shown as follows:

signal at 6.8-7.2 ppm due to aromatic protons of styrene

3.3-3.8 ppm due to OCH_2 protons of *n*-butyl acrylate and 2-ethylhexyl acrylate

The relative mole ratio of styrene/*n*-butyl acrylate and styrene/2-ethylhexyl acrylate was calculated from the ratio of characteristic signal area for each component divided by the number of protons per component molecule giving rise to the resonance. The results were shown in Table A-1 and A-2, respectively.

Table A-1 The mole ratio of styrene/*n*-butyl acrylate in the copolymer

St / BuA (mol %) in feed	signal area at (ppm)		St / BuA (mol %) in copolymer
	3.3-3.8	6.8-7.2	
95/5	2.5489	0.0534	95.1/4.9
90/10	2.5565	0.0903	91.9/8.1
85/15	2.5745	0.1553	86.9/13.1
80/20	2.4448	0.2030	82.8/17.2

Table A-2 The mole ratio of styrene/2-ethylhexyl acrylate in the copolymer

St / 2-EHA (mol %) in feed	signal area at (ppm)		St / 2-EHA (mol %) in copolymer
	3.3-3.8	6.8-7.2	
95/5	2.5535	0.0582	94.6/5.4
90/10	2.5724	0.1090	90.4/9.6
85/15	2.5615	0.1267	89.0/11.0
80/20	2.6230	0.2492	80.8/19.2

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APPENDIX B

GEL PERMEATION CHROMATOGRAPHY

Average molecular weights and molecular weight distribution of the copolymer were measured by gel permeation chromatography. This technique uses a bed of cross-linked polystyrene-divinylbenzene gel for the separation of the polymer taking place in the voids present in the gel particles. The pore volume is available to smaller molecules. This makes the path traveled by the larger molecules shorter than that of the small molecules, and they (the former) are eluted from the column first, because the sorting process takes place on the basis of molecular size, a true size distribution of the polymer species takes place.

Before analysis of the copolymer, the standard calibration curve was set up using standard polystyrene (S-66.0) supplied by Showa denko. The calibration data and calibration curve were shown in Table B-1 and Figure B-1, respectively.

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Table B-1 The calibration data of Shodex polystyrene standard S-66.0

Average Molecular Weights	Retention Times, min.
560,000	15.782
156,000	17.198
28,500	19.100
11,600	20.060
2,950	21.258

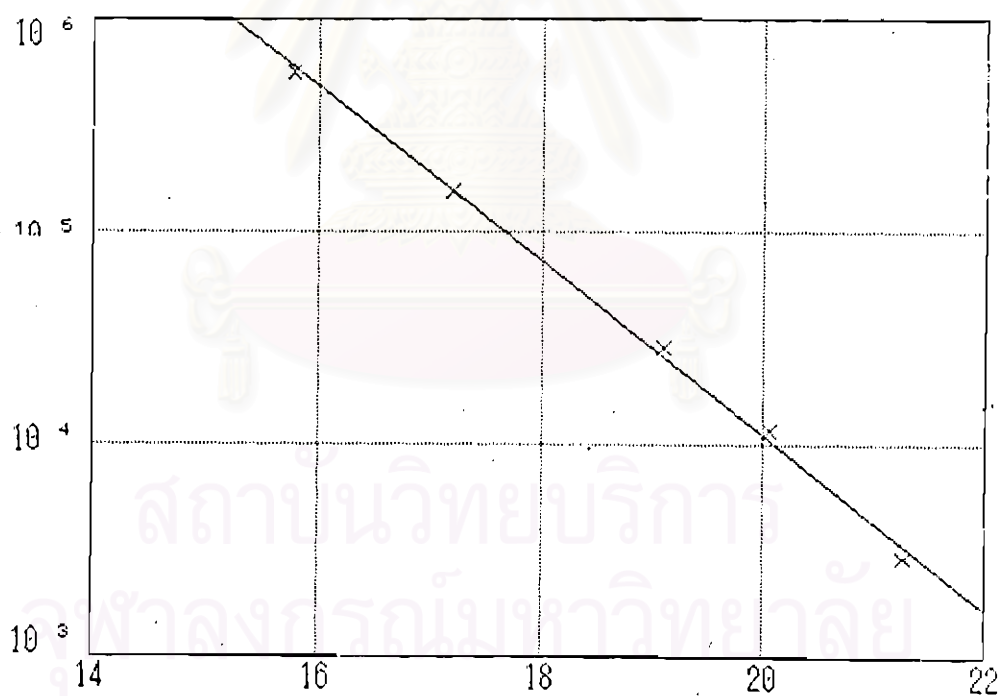


Figure B-1 The calibration curve of Shodex polystyrene standard S-66.0

VITA

Miss Cattleeya Kongsupapsiri was born on January 13, 1975, in Samutsakhon, Thailand. She received the Bachelor's Degree in Industrial Chemistry from the Faculty of Science, Srinakharinwirot University in 1994. She has worked as a chemist in the Research and Development Division, TNP Industry Co., Ltd. from April, 1994 to May, 1996. She has continued her Master's Degree in the Program of Petrochemistry and Polymer Science, Graduate School, Chulalongkorn University since 1996 and finished her study in May 1999.



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