

ผลงานตัวเริมปฏิกริยาแบบรีดออกซ์ต่อกราฟต์โภคโลลิเมอ ไรเซชันของเมทิลเมทาคริเลต
บันยางธรรมชาติ

นางสาวธีรานุช กษาทองรัศมี

ศูนย์วิทยทรัพยากร

วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตร์มหาบัณฑิต
สาขาวิชาปีโตรเคมีและวิทยาศาสตร์พอลิเมอร์ หลักสูตรปีโตรเคมีและวิทยาศาสตร์พอลิเมอร์
คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย

ปีการศึกษา 2544

ISBN 974-03-1515-1

ลิขสิทธิ์ของจุฬาลงกรณ์มหาวิทยาลัย

THE EFFECT OF REDOX INITIATOR ON GRAFT COPOLYMERIZATION OF
METHYL METHACRYLATE ONTO NATURAL RUBBER

Miss Teeranuch Kochthongrasamee

A Thesis Submitted in Partial Fulfillment of the Requirements
for the Degree of Master of Science in Petrochemistry and Polymer Science

Program of Petrochemistry and Polymer Science

Faculty of Science

Chulalongkorn University

Academic Year 2001

ISBN 974-03-1515-1

Accepted by the Faculty of Science, Chulalongkorn University in Partial Fulfillment of
the Requirements for the Master 's Degree

Pipat Karntiang Deputy Dean for Administrative Affairs
(Associate Professor Pipat Karntiang, Ph.D.) Acting Dean, Faculty of Science

THESIS COMMITTEE

Chairman
(Associate Professor Sophon Roengsumran, Ph.D)

Suda Kiatkamjornwong Thesis Advisor
(Professor Suda Kiatkamjornwong, Ph.D.)


..... Thesis Co-advisor
(Professor Pattarapan Prasassarakich, Ph.D.)

Varaporn Kajornchaiyakul
Member
(Varaporn Kajornchaiyakul, M.Phil.)

..... ov-Trakarnpruk Member
(Associate Professor Wimonrat Trakarnpruk, Ph.D.)

ชีรานุช กษทองรัศมี : ผลของตัวริเริ่มปฏิกิริยาแบบบรีดอคซ์ต่อกราฟต์โคพอลิเมอไรเซชันของเมทิล เมทาคริเลตบนยางธรรมชาติ (THE EFFECT OF REDOX INITIATOR ON GRFAT COPOLYMERIZATION OF METHYL METHACRYLATE ONTO NATURAL RUBBER)

อ.ที่ปรึกษา : ศ. ดร. สุดา เกียรติกำจารวงศ์

อ.ที่ปรึกษา : ศ. ดร. ภัทรพรรณ ประศาสน์สารกิจ; 131 หน้า. ISBN 974-03-1515-1.

งานวิจัยนี้เป็นการศึกษาผลของตัวริเริ่มปฏิกิริยาแบบบรีดอคซ์ต่อกราฟต์เมทิลเมทาคริเลตบนยางธรรมชาติโดย อิมัลชันพอลิเมอไรเซชัน โดยตัวริเริ่มปฏิกิริยาระบบรีดอคซ์ 3 ชนิดที่แตกต่างกัน คือ คิวมีนไไฮโตรเพอร์ออกไซด์/เททระเออทิลีนเพนทามีน เทอร์-บิวทิลไไฮโตรเพอร์ออกไซด์/เททระเออทิลีนเพนทามีน และ โพแทสเซียมเพอร์ชัลเฟต/โซเดียมไทโอลชัลเฟต สำหรับตัวริเริ่มปฏิกิริยาแต่ละชนิดได้ศึกษาภาวะที่เหมาะสมในการเตรียมยางธรรมชาติกราฟต์ โดยตรวจสอบหนุ่ฟังก์ชันด้วย FT-IR และ NMR หน้าหนักโมเลกุลเฉลี่ยและการกระจายหน้าหนักโมเลกุลด้วย GPC ตรวจสอบสมบัติทางความร้อนของยางธรรมชาติกราฟต์ด้วย DSC ศึกษาสัณฐานวิทยาของยางกราฟต์ด้วย TEM พบว่า เมทิลเมทาคริเลตเกิดปฏิกิริยากราฟต์โคพอลิเมอไรเซชันที่บริเวณผิวนูภาคยาง

ตัวริเริ่มปฏิกิริยาคิวมีนไไฮโตรเพอร์ออกไซด์คล้ายได้ดีมากในน้ำมัน และเทอร์-บิวทิลไไฮโตรเพอร์ออกไซด์คล้ายได้ดีปานกลางในน้ำมัน ทำปฏิกิริยาร่วมกับเททระเออทิลีนเพนทามีนที่คล้ายได้ดีปานกลางในน้ำ พบร่วมกิวมีนไไฮโตรเพอร์ออกไซด์ให้ค่าประสิทธิภาพการกราฟต์ที่สูงกว่า ตัวริเริ่มปฏิกิริยาแบบที่คลายน้ำได้ดี คือ โพแทสเซียมเพอร์ชัลเฟต/โซเดียมไทโอลชัลเฟต ดังนั้นจึงเติมไวนิลนีโอดีเคลโนเอตลงในตัวริเริ่มปฏิกิริยาโพแทสเซียมเพอร์ชัลเฟตเพื่อส่งเสริมให้เกิดกราฟต์ของเมทิลเมทาคริเลตบนยางธรรมชาติและลดปริมาณพอลิเมทิลเมทาคริเลตซึ่งเกิดในน้ำ พบร่วมอยู่ด้วยของประสิทธิภาพการกราฟต์ของ เมทิลเมทาคริเลตบนน้ำยางธรรมชาติโดยคิวมีนไไฮโตรเพอร์ออกไซด์ เทอร์-บิวทิลไไฮโตรเพอร์ออกไซด์ และ โพแทสเซียมเพอร์ชัลเฟต มีค่า 84.4 74.5 และ 61.1 ตามลำดับ ซึ่งสอดคล้องกับค่าร้อยละของพอลิเมทิล เมทาคริเลตที่เกิดในน้ำ ซึ่งมีค่า 7.2 12.0 และ 17.8 สำหรับการริเริ่มปฏิกิริยาด้วยคิวมีนไไฮโตรเพอร์ออกไซด์ เทอร์-บิวทิลไไฮโตรเพอร์ออกไซด์ และ โพแทสเซียมเพอร์ชัลเฟต ตามลำดับ การเติมไวนิลนีโอดีเคลโนเอตทำให้เกิดแอลลิลิกแรคคิคลับโน่พอลิไโอลพรีนซึ่งเอื้อต่อการเกิดกราฟต์ด้วยมอนомерรชนิดอื่น งานวิจัยนี้สรุปได้ว่าคิวมีนไไฮโตรเพอร์ออกไซด์/เททระเออทิลีนเพนทามีนเป็นระบบของตัวริเริ่มปฏิกิริยาที่ดีของการกราฟต์น้ำยางธรรมชาติด้วยไวนิลมอนomer นอกจากนี้ยังได้อธิบายกลไกของการส่งเสริมการเกิดปฏิกิริยากราฟต์โคพอลิเมอไรเซชัน

หลักสูตร....ปีตรเคมีและวิทยาศาสตร์พอลิเมอร์.... ลายมือชื่อนิสิต.....จิตาภา บด్యูงวัฒน์.....
สาขาวิชา....ปีตรเคมีและวิทยาศาสตร์พอลิเมอร์.... ลายมือชื่ออาจารย์ที่ปรึกษา.....ดร. ภัทรพรรณ ประศาสน์.....
ปีการศึกษา.....2544..... ลายมือชื่ออาจารย์ที่ปรึกษาร่วม.....ดร. สุดา เกียรติกำจารวงศ์.....

##4272303323 : MAJOR PETROCHEMISTRY AND POLYMER SCIENCE

KEYWORD : NATURAL RUBBER / METHYL METHACRYLATE / REDOX INITIATOR / GRAFT COPOLYMERIZATION

TEERANUCH KOCHTHONGRASAMEE : THE EFFECT OF REDOX INITIATOR ON GRAFT COPOLYMERIZATION OF METHYL METHACRYLATE ONTO NATURAL RUBBER.

THESIS ADVISOR : PROF. SUDA KIATKAMJORNWONG, Ph.D., THESIS CO-ADVISOR : PROF. PATTARAPAN PRASASSARAKICH, Ph.D. 131 pp. ISBN 974-03-1515-1.

This research investigated the effect of redox initiator on natural rubber graft methyl methacrylate by emulsion polymerization. Three different types of redox initiator namely cumene hydroperoxide (CHPO)/tetraethylene pentamine (TEPA), tert-butyl hydroperoxide (TBHPO)/TEPA, and potassium persulfate ($K_2S_2O_8$)/sodium thiosulfate ($Na_2S_2O_3$). The optimum reaction condition for each redox initiator on grafting of natural rubber was studied. Functional groups of grafted natural rubber by FT-IR and NMR, molecular weights and their distribution by GPC, thermal properties by DSC, and morphology by TEM were carried out. It was found that the poly(methyl methacrylate) graft copolymerization resides on the surface of rubber particles.

CHPO dissolves very well in the oil phase, and TBHPO dissolves moderately in the oil phase as well. Each can interact with TEPA in aqueous phase. CHPO was found to give a higher grafting efficiency. $K_2S_2O_8$ / $Na_2S_2O_3$ is water-soluble initiator. To promote a greater grafting efficiency and lower homopolymer content of poly(methyl methacrylate) in the aqueous phase, vinyl *neo*-decanoate (VneoD), a grafting promoter was then added in the latter. It was found that the percentages of grafting on methyl methacrylate on natural rubber latex initiated by CHPO, TBHPO, and $K_2S_2O_8$ are 84.4, 74.5, and 61.1, respectively, which are in good agreement with percentages of poly(methyl methacrylate) formed in the aqueous phase as 7.2, 12.0, and 17.9 by CHPO, TBHPO, and $K_2S_2O_8$. The role of added VneoD is to produce allylic radicals on polyisoprene chains that favor the grafting reaction with other vinyl monomers. This research concludes that CHPO/TEPA is a good redox system for grafting of vinyl monomers on natural rubber latex. Additionally, this research also describes the reaction mechanism that promotes such a graft copolymerization.

Program Petrochemistry and Polymer Science... Student's signature.....Teeranuch Kochthongrasamee
Field of study... Petrochemistry and Polymer Science... Advisor's signature.....Suda Kiatkamjornwong
Academic year.....2001.....Co-Advisor's signature.....Pattaranaporn Prasassarakich

ACKNOWLEDGMENTS

First of all, I wish to express my deep gratitude to my advisor Professor Suda Kiatkamjornwong and to my co-advisor Professor Pattarapan Prasassarakich for their suggestion, guidance, encouragement, reviewing the whole thesis, and kindness throughout the course of study. In addition, I also thank the thesis committee for their suggestion on the thesis.

I also wish to thank the Department of Imaging and Printing Technology, Faculty of Science, and the Graduate School of Chulalongkorn University for research facilities and partial financial support, respectively.

The author is also thankful for Thai Rubber Latex Corporation (Thailand) Public Co., Ltd., Bangplee Samutprakarn and Siam Chemical Industry Co., Ltd., who provided the materials used in this research.

Finally, I am grateful to my parents for their love and endless support. Thanks are also extended to my friends for their helpful support during this entire study.

CONTENTS

	PAGE
ABSTRACT (IN THAI).....	iv
ABSTRACT (ENGLISH).....	v
ACKNOWLEDGMENTS.....	vi
CONTENTS	vii
LIST OF TABLES.....	xi
LIST OF FIGURES.....	xv
ABBREVIATIONS.....	xxv
CHAPTER 1 : INTRODUCTION.....	1
1.1 The Purpose of the Investigation.....	1
1.2 Objectives.....	2
1.3 Scope of the Investigation.....	2
CHAPTER 2 : THEORY AND LITERATURE REVIEW.....	3
2.1 Natural Rubber.....	3
2.1.1 Natural Rubber Latex.....	3
2.1.2 The Preservation of Natural Rubber Latex.....	4
2.1.3 The Concentration of Natural Rubber Latex.....	5
2.2 Modified Natural Rubber.....	6
2.3 Emulsion Polymerization.....	9
2.3.1 Ingredients and Processes.....	10
2.4 Graft Copolymer.....	13
2.4.1 Synthesis of Graft Copolymer of Natural Rubber.....	14
2.4.2 Natural Rubber-Graft-Methyl Methacrylate.....	18
2.5 Literature Review.....	19

CONTENTS (CONTINUED)

	PAGE
CHAPTHER 3 : EXPERIMENTAL.....	23
3.1 Chemicals.....	23
3.2 Glassware.....	24
3.3 Equipment.....	24
3.4 Procedure.....	25
3.4.1 Purification of Monomer.....	25
3.4.2 Preparation of Grafted Natural Rubber.....	25
3.5 Characterization of the Grafted Natural Rubber.....	28
3.5.1 Determination of Conversion.....	28
3.5.2 Determination of Grafted Natural Rubber and Grafting Efficiency.....	29
3.5.3 Determination of the Gel Content of Graft Copolymer.....	29
3.5.4 The Morphology of Grafted Natural Rubber.....	30
3.5.5 Determination of Average Molecular Weight of Grafted Natural Rubber.....	30
3.5.6 Determination of Grafted Natural Rubber.....	31
3.5.7 Thermal Properties of the Grafted Natural Rubber.....	31
CHAPTER 4 : RESULTS AND DISCUSSION.....	33
4.1 Properties of Natural Rubber Latex.....	33
4.2 Preparation of the Grafted Natural Rubber.....	34

CONTENTS (CONTINUED)

	PAGE
4.2.1 The Preparation of the Grafted Natural Rubber using CHPO or TBHPO as Redox Initiator.....	35
4.2.1.1 Effect of Initiator Concentration.....	35
4.2.1.2 Effect of Reaction Temperature.....	49
4.2.1.3 Effect of MMA Monomer Concentration.....	52
4.2.1.4 Effect of Reaction Time.....	56
4.2.2 The Preparation of the Grafted Natural Rubber using Potassium Persulfate.....	59
4.2.2.1 Effect of Grafting Agent.....	59
4.2.2.2 Effect of Initiator Concentration.....	64
4.2.2.3 Effect of Reaction Temperature.....	71
4.2.2.4 Effect of MMA Monomer Concentration.....	74
4.2.2.5 Effect of Reaction Time.....	76
4.2.3 Types of Redox Initiator on Grafting of Natural Rubber.....	79
4.2.4 Mechanism of Grafting.....	81
4.3 Characterization of Grafted Natural Rubber	88
4.3.1 Grafted Natural Rubber Morphology.....	88
4.3.1.1 Locus of Polymerization.....	92
4.3.2 Effect of \bar{M}_w , \bar{M}_n , and Polydispersity of the Grafted Natural Rubber.....	94
4.3.3 Effect of Glass Transition of the Grafted Natural Rubber.....	98
4.3.4 Functional Groups in the Grafted Natural Rubber.....	100

CONTENTS (CONTINUED)

	PAGE
CHAPTER 5: CONCLUSIONS AND SUGGESTION.....	109
5.1 Conclusions.....	109
5.1.1 Preparation of Grafted Natural Rubber.....	109
5.2 Suggestion.....	111
REFFERENCES.....	112
APPENDICES.....	116
APPENDIX A.....	117
APPENDIX B.....	129
VITA.....	131

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

LIST OF TABLES

TABLE	PAGE
2.1 Typical proportion of composition in natural rubber latex.....	4
2.2 Typical properties of high ammonia type of centrifuged natural rubber latex.....	6
2.3 Some monomers used in commonly by commercial emulsion polymerization.....	12
3.1 Experimental conditions of graft copolymerization.....	28
4.1 The properties of natural rubber latex (High ammonia).....	34
4.2 Effect of initiator concentration of TBHPO and CHPO on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at 60°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	36
4.3 Effect of initiator concentration of TBHPO and CHPO on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at 50°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	37
4.4 Effect of initiator concentration of TBHPO and CHPO on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at 40°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	37
4.5 Effect of initiator concentration of TBHPO and CHPO on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at 30°C, MMA monomer concentration and 100 phr, reaction time 8 hr.....	38

LIST OF TABLES (CONTINUED)

TABLE	PAGE
4.6 Effect of reaction temperature on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at initiator concentration 0.5 phr, MMA monomer concentration 100 phr, and reaction time 8 hr.....	50
4.7 Effect of the MMA monomer concentration on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at 50°C, initiator concentration 0.5 phr, and reaction time 8 hr.....	53
4.8 Effect of reaction time on the percentage conversion, percentage grafting efficiency percentage grafted natural rubber at 60°C, initiator concentration 0.5 phr and MMA monomer concentration 100 phr.....	57
4.9 Effect of grafting agent on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at 55°C, $K_2S_2O_8$ 1.0 phr, and reaction time 8 hr.....	62
4.10 Effect of initiator concentration of $K_2S_2O_8$ on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at 70-60°C, VneoD 20 %wt of MMA monomer concentration, monomer concentration 100 phr, and reaction time 8 hr.....	65

LIST OF TABLES (CONTINUED)

TABLE	PAGE
4.11 Effect of initiator concentration of $K_2S_2O_8$ on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at $55-50^\circ C$, VneoD 20 %wt of MMA monomer concentration, monomer concentration 100 phr, and reaction time 8 hr.....	65
4.12 Effect of reaction temperature on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at VneoD 20 %wt of MMA, $K_2S_2O_8$ 1.0 phr, concentration 100 phr, and reaction time 8 hr.....	72
4.13 Effect of MMA concentration on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at $K_2S_2O_8$ 1.0 phr, and reaction time 8 hr.....	75
4.14 Effect of reaction time on the percentage conversion, percentage grafting efficiency and percentage grafted natural rubber at $60^\circ C$, $K_2S_2O_8$ 1.0 phr, VneoD 20 %wt of MMA, and monomer concentration 100 phr.....	77
4.15 Effect of redox initiator system on percentage grafting efficiency, percentag grafted natural rubber, and percentage grafting efficiency at the optimum condition.....	80
4.16 The gel content, \bar{M}_w , \bar{M}_n , and \bar{M}_n/\bar{M}_w from the optimum condition of different redox initiators.....	96

LIST OF TABLES (CONTINUED)

TABLE	PAGE
4.17 The gel content, \bar{M}_w , \bar{M}_n , and \bar{M}_n/\bar{M}_w of various initiators.....	97
4.18 The important characteristic peaks for the FT-IR spectra of the grafted natural rubber.....	101
4.19 The important signals for the NMR spectra of the grafted natural rubber.....	104
A-1 Effect of CHPO concentration, reaction temperature, and reaction time on the percentage conversion, percentage grafted natural rubber, and percentage grafting efficiency.....	119
A-2 Effect of TBHPO concentration, reaction temperature, and reaction time on the percentage conversion, percentage grafted natural rubber, and percentage grafting efficiency.....	122
A-3 Effect of $K_2S_2O_8$ concentration, reaction temperature, and reaction time on the percentage conversion, percentage grafted natural rubber, and percentage grafting efficiency.....	125
B-1 The calibration data of Shodex polystyrene standard S-66.0.....	130

LIST OF FIGURES

FIGURE	PAGE
3.1 Apparatus for copolymerization of MMA monomer onto natural rubber.....	27
3.2 The overall schematic experimental process.....	32
4.1a Effect of initiator concentration of TBHPO on the percentage free NR, percentage free PMMA , and percentage grafted natural rubber at 60°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	39
4.1b Effect of initiator concentration of TBHPO on the percentage conversion and percentage grafting efficiency at 60°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	39
4.2a Effect of initiator concentration of TBHPO on the percentage free NR, percentage free PMMA , and percentage grafted natural rubber at 50 °C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	39
4.2b Effect of initiator concentration of TBHPO on the percentage conversion and percentage grafting efficiency at 50°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	40
4.3a Effect of initiator concentration of TBHPO on the percentage free NR, percentage free PMMA , and percentage grafted natural rubber at 40°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	40

LIST OF FIGURES (CONTINUED)

FIGURE	PAGE
4.3b Effect of initiator concentration of TBHPO on the percentage conversion and percentage grafting efficiency at 50°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	41
4.4a Effect of initiator concentration of TBHPO on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 30°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	41
4.4b Effect of initiator concentration of TBHPO on the percentage conversion and percentage grafting efficiency at 50°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	42
4.5a Effect of initiator concentration of CHPO on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 60°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	42
4.5b Effect of initiator concentration of CHPO on the percentage conversion and percentage grafting efficiency at 60°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	43
4.6a Effect of initiator concentration of CHPO on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 50°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	43

LIST OF FIGURES (CONTINUED)

FIGURE	PAGE
4.6b Effect of initiator concentration of CHPO on the percentage conversion and percentage grafting efficiency at 50°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	44
4.7a Effect of initiator concentration of CHPO on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 40 °C MMA monomer concentration 100 phr, and reaction time 8 hr.....	45
4.7b Effect of initiator concentration of CHPO on the percentage conversion and percentage grafting efficiency at 40°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	45
4.8a Effect of initiator concentration of CHPO on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 30°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	45
4.8b Effect of initiator concentration of CHPO on the percentage conversion and percentage grafting efficiency at 30°C, MMA monomer concentration 100 phr, and reaction time 8 hr.....	46
4.9 Effect of initiator concentration of TBHPO on the percentage conversion (30-60 °C, MMA monomer concentration 100 phr, and reaction time 8 hr).....	46
4.10 Effect of initiator concentration of TBHPO on the percentage grafted natural rubber (30-60 °C, MMA monomer concentration 100 phr, and reaction time 8 hr).....	47

LIST OF FIGURES (CONTINUED)

FIGURES	PAGE
4.11 Effect of initiator concentration of TBHPO on the percentage grafting efficiency (30-60 °C, MMA monomer concentration 100 phr, and reaction time 8 hr).....	47
4.12 Effect of initiator concentration of CHPO on the percentage conversion at(30-60 °C, MMA monomer concentration 100 phr, and reaction time 8 hr).....	48
4.13 Effect of initiator concentration of CHPO on the percentage grafted natural rubber (30-60 °C, MMA monomer concentration 100 phr, and reaction time 8 hr).....	48
4.14 Effect of initiator concentration of CHPO on the percentage grafting efficiency (30-60 °C, MMA monomer concentration 100 phr, and reaction time 8 hr).....	49
4.15 Effect of reaction temperature on the percentage conversion (30-60 °C, MMA monomer concentration 100 phr, and reaction time 8 hr).....	51
4.16 Effect of reaction temperature on the percentage grafted natural rubber (30-60 °C, MMA monomer concentration 100 phr, and reaction time 8 hr).....	51
4.17 Effect of reaction temperature on the percentage grafting efficiency (30-60 °C, MMA monomer concentration 100 phr, and reaction time 8 hr).....	52

LIST OF FIGURES (CONTINUED)

FIGURE	PAGE
4.18a Effect of the MMA monomer concentration on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 50°C, TBHPO 0.5 phr, and reaction time 8 hr.....	54
4.18b Effect of monomer concentration on the percentage conversion, and percentage grafting efficiency at 50°C, TBHPO 0.5 phr, and reaction time 8 hr.....	54
4.19a Effect of monomer concentration on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 50°C, CHPO 0.5 phr, and reaction time 8 hr.....	55
4.19b Effect of monomer concentration on the percentage conversion, and percentage grafting efficiency at 50°C, CHPO 0.5 phr, and reaction time 8 hr.....	55
4.20a Effect of reaction time on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 50°C, TBHPO 0.5 phr, and MMA monomer concentration 100 phr.....	57
4.20b Effect of reaction time on the percentage conversion, and percentage grafting efficiency at 50°C, TBHPO 0.5 phr, and MMA monomer concentration 100 phr.....	58
4.21a Effect of reaction time on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 50°C, CHPO 0.5 phr, and MMA monomer concentration 100 phr.....	58

LIST OF FIGURES (CONTINUED)

FIGURE	PAGE
4.21b Effect of reaction time on the percentage conversion, and percentage grafting efficiency at 50°C, CHPO 0.5 phr, and MMA monomer concentration 100 phr.....	59
4.22 Chemical structures of vinyl acetate (VAc) and vinyl <i>neo</i> -decanoate.....	60
4.23 Reaction scheme for hydrogen abstraction form poly(cis-isoprene) to from graft sites.....	61
4.24a Effect of grafting agent on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at 55°C , $K_2S_2O_8$ 1.0 phr, MMA monomer concentration 100 phr, and reaction time 8 hr.....	63
4.24b Effect of grafting agent on the percentage conversion and percentage grafting efficiency at 55°C, $K_2S_2O_8$ 1.0 phr, MMA monomer concentration 100 phr, and reaction time 8 hr	63
4.25a Effect of initiator concentration of $K_2S_2O_8$ on the percentage free NR, percentage free PMMA , and percentage grafted natural rubber at 70°C, VneoD 20 %wt of MMA , MMA monomer concentration 100 phr, and reaction time 8 hr.....	66
4.25b Effect of initiator concentration of $K_2S_2O_8$ on the percentage conversion, and percentage grafting efficiency at 70°C, VneoD 20 %wt of MMA , MMA monomer concentration 100 phr, and reaction time 8 hr.....	66

LIST OF FIGURES (CONTINUED)

FIGURE	PAGE
4.26a Effect of initiator concentration of $K_2S_2O_8$ on the percentage free NR percentage free PMMA, and percentage grafted natural rubber at $60^\circ C$, VneoD 20 %wt of MMA, MMA monomer concentration 100 phr, and reaction time 8 hr.....	66
4.26b Effect of initiator concentration of $K_2S_2O_8$ on the percentage conversion, and percentage grafting efficiency at $60^\circ C$, VneoD 20 %wt of MMA, MMA monomer Concentration 100 phr, and reaction time 8 hr.....	67
4.27a Effect of initiator concentration of $K_2S_2O_8$ on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at $55^\circ C$, VneoD 20 %wt of MMA, MMA monomer concentration 100 phr, and reaction time 8 hr.....	68
4.27b Effect of initiator concentration of $K_2S_2O_8$ on the percentage conversion, and percentage grafting efficiency at $55^\circ C$, VneoD 20 %wt of MMA, MMA monomer concentration 100 phr, and reaction time 8 hr.....	68
4.28a Effect of initiator concentration of $K_2S_2O_8$ on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at $50^\circ C$, VneoD 20 %wt of MMA, MMA monomer concentration 100 phr, and reaction time 8 hr.....	69

LIST OF FIGURES (CONTINUED)

FIGURE	PAGE
4.28b Effect of initiator concentration of $K_2S_2O_8$ on the percentage conversion, and percentage grafting efficiency at 50°C, VneoD 20 %wt of MMA, monomer concentration 100 phr, and reaction time 8 hr.....	69
4.29 Effect of initiator concentration of $K_2S_2O_8$ on the percentage conversion (50-70 °C, VneoD 20 %wt of MMA, monomer concentration 100 phr, and reaction time hr).....	70
4.30 Effect of initiator concentration of $K_2S_2O_8$ on the percentage Grafted Natural rubber (50-70 °C, VneoD 20 %wt of MMA, monomer concentration 100 phr, and reaction time 8 hr).....	70
4.31 Effect of initiator concentration of $K_2S_2O_8$ on the percentage Grafting efficiency (50-70 °C, VneoD 20 %wt of MMA, monomer concentration 100 phr, and reaction time 8 hr).....	71
4.32a Effect of reaction temperature on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at $K_2S_2O_8$ 1.0 phr, VneoD 20 %wt of MMA, monomer concentration 100 phr, and reaction time 8 hr.....	73
4.32b Effect of reaction temperature on the percentage conversion and percentage grafting efficiency at $K_2S_2O_8$ 1.0 phr, VneoD 20 %wt of MMA, monomer concentration 100 phr, and reaction time 8 hr.....	73

LIST OF FIGURES (CONTINUED)

FIGURE	PAGE
4.33a Effect of monomer concentration on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at $K_2S_2O_8$ 1.0 phr, VneoD 20 %wt of MMA, 60°C, and reaction time 8 hr.....	75
4.33b Effect of monomer concentration on the percentage conversion and percentage grafting efficiency at $K_2S_2O_8$ 1.0 phr, VneoD 20 %wt of MMA, 60°C, and reaction time 8 hr.....	76
4.34a Effect of reaction time on the percentage free NR, percentage free PMMA, and percentage grafted natural rubber at $K_2S_2O_8$ 1.0 phr, VneoD 20 %wt of MMA, 60°C, and monomer concentration 100 phr.....	78
4.34b Effect of reaction time on the percentage conversion and percentage grafting efficiency at $K_2S_2O_8$ 1.0 phr, VneoD 20 %wt of MMA, 60°C, and monomer concentration 100 phr.....	78
4.35 Effect of types of redox initiator on percentage conversion, percentage grafted natural rubber, and percentage grafting efficiency at the optimum condition.....	80
4.36 Transmission electron micrographs of grafted natural rubber latex; MMA 100 phr, TBHPO 0.5 phr, 50°C. a.) 4 hr (x 30,000), b.) 6 hr (x30,000), c.) 8 hr (x45,000), and d.) 10 hr (x30,000).....	89
4.37 Transmission electron micrographs of grafted natural rubber latex; MMA 100 phr, CHPO 0.5 phr, 50°C. a.) 4 hr (x 30,000), b.) 6 hr (x30,000), c.) 8 hr (x45,000), and d.) 10 hr (x16,500).....	90

LIST OF FIGURES (CONTINUED)

FIGURE	PAGE
4.38 Transmission electron micrographs of grafted natural rubber latex; 60°C, K ₂ S ₂ O ₈ 1.0 phr, VneoD 20 %wt of MMA, and monomer concentration 100 phr a.) 4 hr (x 45,000), b.) 6 hr (x45,000), c.) 8 hr (x45,000), and d.) 10 hr (x45,000).....	91
4.39 Site of reaction in modified natural rubber latex in the bipolar redox initiator.....	93
4.40 DSC thermograms of natural rubber.....	98
4.41 DSC thermograms of grafted natural rubber from CHPO initiation	99
4.42 DSC thermograms of grafted natural rubber from TBHPO initiation.....	99
4.43 DSC thermograms of grafted natural rubber from K ₂ S ₂ O ₈ initiation	100
4.44 The FT-IR spectrum of natural rubber.....	102
4.45 The FT-IR spectrum of grafted natural rubber initiated by CHPO.....	102
4.46 The FT-IR spectrum of grafted natural rubber initiated by TBHPO.....	103
4.47 The FT-IR spectrum of grafted natural rubber initiated by K ₂ S ₂ O ₈	103
4.48 The ¹ H-NMR spectrum of grafted natural rubber initiated by CHPO	105
4.49 The ¹ H-NMR spectrum of grafted natural rubber initiated by TBHPO	105
4.50 The ¹ H-NMR spectrum of grafted natural rubber initiated by K ₂ S ₂ O ₈	106
4.51 The ¹³ C-NMR spectrum of grafted natural rubber initiated by CHPO	107
4.52 The ¹³ C-NMR spectrum of grafted natural rubber initiated by TBHPO.....	107
4.53 The ¹³ C-NMR spectrum of grafted natural rubber initiated by K ₂ S ₂ O ₈	108
B-1 The calibration curve of Shodex polystyrene standard S-66.0.....	130

ABBREVIATIONS

NR	: Natural Rubber
MMA	: Methyl Methacrylate
PMMA	: Poly(methyl methacrylate)
DRC	: Dry Rubber Content
FT-IR	: Fourier Transform Infrared Spectroscopy
NMR	: Nuclear Magnetic Resonance Spectroscopy
TEM	: Transmission Electron Microscopy
DSC	: Differential Scanning Calorimetry
phr	: Parts per hundred of rubber
hr	: Hour
Tg	: Glass Transition Temperature
GE	: Grafting Efficiency
pp.	: Page
VneoD	: Vinyl <i>neo</i> -decanoate