

## รายการอ้างอิง

1. ณัฐริตา ชวนเกริกกุล , การขึ้นรูปและสมบัติของชิ้นงานซิลิกอนคาร์ไบด์ที่มีรูพรุนสำหรับใช้ในภากรอง. วิทยานิพนธ์ระดับมหาบัณฑิต เทคโนโลยีเซรามิก คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย, 1999.
2. Rainer T. Boride and Carbide Ceramics. In Cahn R. W. Haasen P. and Kramer E. J. (eds). Materials Science and Technology, Vol 11, Structure and Properties of Ceramics, p.181. New York: Weinheim, 1994.
3. R.A. Alliegro, L.B. Coffin, and J.R. Tinklepaugh. Pressure-Sintered Silicon Carbide . J. Am.Ceram.Soc.39 [11], 1956.
4. S. Prochazka and R.J. Charles. Strength of Boron-Doped,HotPressed Silicon Carbide. Am.Ceram.Soc.Bull. 52 [12], 1973.
5. K. Yamada and M. Mouri. Properties and Applications of SiC Ceramics, Silicon Carbide Ceramics : p.14-29.
6. E.R. Maddrell. Pressureless sintering of silicon carbide, Journal of Materials Science letters ., 6(1987) : p.486-488.
7. H. Suzuki and T. Hase. Boron Transport and Change of Lattice Parameter During Sintering of  $\beta$ -SiC. J.Am.Ceram.Soc. 63(1980) :p.349-350.
8. R.M. Williams, B.N. Juterbock, C.R. Peters and T.J. Whalen. Communications of the American Ceramic Society 1984 : C-62-C-64.
9. W. Bocker, H. Landfermann and H. Hausner. Sintering of Alpha Silicon Carbide with Addition of Aluminum. Powder metallurgy International. 11(2)(1979) : p.83-85.
10. S. Shinozaki, R.M. Williams,B.N. Juterbock,W.T. Donlon,J. Hangas and C.R. peters. Microstructural Developments in Pressureless-Sintered  $\beta$ -SiC Materials with Al,B, and C Additions. Am.Ceram.Soc.Bull. 64(10)(1985) : p.1389-93.
11. M. Omori and H. Takei. Pressureless Sintering of SiC. J.Am.Ceram.Soc. 65(1982) : C92.

12. J.H. She and K. Ueno. Effect of Additive Content on Liquid-Phase Sintering on Silicon Carbide Ceramics. Materials Research Bulletin 34(10/11) (1999) :p.1629-1636.
13. D.H. Kim and C.H. Kim. Toughening Behavior of Silicon Carbide with Additions of Ytria and Alumina. J.Am.Ceram.Soc. 73 (5)(1990) : p1431-34.
14. T. Noguchi and M. Mizuno. Kogyo Kagaku Zasshi 70 (6)(1967) : p.839.
15. Y.W. Kim, H. Tanaka, M. Mitomo and S. Otani, Influence of Powder Characteristics on Liquid Phase Sintering of Silicon Carbide. J.Ceram.Soc.Jpn. 103(1994) : p.260-64.
16. J.Y. Kim, Y.W. Kim, M. Mitomo, GD. Zhan and J.G.Lee. Microstructure and Mechanical Properties of  $\alpha$ -Silicon Carbide Sintered with Yttrium-Aluminum Garnet and Silica. J.Am.Ceram.Soc. 82 (2)(1999) : p.441-44.
17. I.A. Bondar and F.Ya. Galakhov. Isv. Akad. Nauk. SSSR. Ser.Khim. 7(1963) : 1325.
18. M. Mitomo, Y.W. Kim and H. Hirotsumu. Fabrication of Silicon Carbide nanoceramics. J. Mater.Res. 11 (7)(1996) : p.1601-04.
19. H. Miyazaki, A. Hakomori, K. Yasuda, Y. Matsuo, T. Yano and T. Iseki. Densification and Thermal, Mechanical and Electrical Properties of SiC Ceramics Sintered with Addition of MgO. J.Ceram.Soc.Jpn. 109 (3)(2001) : p.227-31.
20. จุฑามาส จิตต์เจริญ. การศึกษาระบบการขึ้นเทอริงซิลิกอนคาร์ไบด์. วิทยานิพนธ์ระดับมหาบัณฑิต เทคโนโลยีเซรามิก คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย, 1993.
21. J.J. Cao, W.J. Moberlychan, L.C. De Jonghe, C.J. Gilbert and R.O. Ritchie. In Situ Toughened Silicon Carbide with Al-B-C Additions. J.Am.Ceram.Soc. 79(2)(1996) : p.461-69.
22. S.K. Lee and C.H. Kim. Effects of  $\alpha$ -SiC versus  $\beta$ -SiC Starting Powder on Microstructure and Fracture Toughness of SiC Sintered with  $Al_2O_3$ - $Y_2O_3$  Additive. J.Am.Ceram.Soc. 77(6)(1994) : p.1655-58 .

23. G.R. Artis, P. Chantikul, B.R. Lawn, and D.B. Marshall. A Critical Evaluation of Indentation Techniques for Measuring Fracture Toughness : I, Direct Crack Measurements. J.Am.Ceram.Soc. 64 (9)(1981) : p.533-38.
24. D.B. Marshall, T. Noma, and A.G. Evans. A Simple Method for Determining Elastic-Modulus-to-Hardness Ratios Using Knoop Indentation Measurements. J.Am.Ceram.Soc. 65 (10)(1982) : C-175.
25. T. Nagano, K. Kaneko, G.D. Zhan and M. Mitomo. Effect of Atmosphere on Weight Loss in Sintered Silicon Carbide during Heat Treatment. J.Am.Ceram.Soc. 85(11)(2000) : p.2781-87.
26. T. Grande, H. Sommerset, E. Hagen, K. Wiik, and M.A. Einarsrud. Effect of Weight Loss on Liquid-Phase-Sintered Silicon Carbide. J.Am.Ceram.Soc. 80(4)(1997) : p.1047-52.



ศูนย์วิทยทรัพยากร  
จุฬาลงกรณ์มหาวิทยาลัย



ภาคผนวก

ศูนย์วิทยทรัพยากร  
จุฬาลงกรณ์มหาวิทยาลัย

## ภาคผนวก

## การวัดมาตรฐาน JCPDS

29-1131		Wavelength= 1.54184									
a-SiC		d Å	Int	h	k	l					
Silicon Carbide		2.62100	40	1	0	1					
		2.51100	00	1	0	2					
		2.35200	20	1	0	3					
Moissanite-611, syn		2.17400	10	1	0	4					
Rad.: CuKα1: 1.5405 Filter: d-sp: Debye-Scherrer		1.53700	35	1	1	0					
Cut off Int. Film I/teor.:		1.41800	15	1	0	9					
Ref: Hannam, Shaffer, J. Appl. Crystallogr., 2, 45 (1969)		1.31100	40	1	1	6					
		1.28600	15	2	0	3					
		1.25600	7	0	0	12					
		1.08700	15	2	0	8					
		1.04200	7	2	0	9					
Sys: Hexagonal S.G.: P6 <sub>3</sub> mc (186)		1.00400	15	2	1	1					
a: 3.073	b:	c: 15.08	A:	C: 4.9073							
		0.972700	15	1	1	12					
		0.887600	15	2	1	8					
		0.862400	7	2	1	9					
Ref: Thibault, N., Am. Mineral., 29, 327 (1944)		0.837000	9	2	1	10					
		0.802100	9	2	0	15					
Dx: 1.239 Dm: 3.218 SS/FOM <sub>1/2</sub> -13(0186, 73)											
m: 2.648(1) r: 2.691(2) Sign: 2V:											
Ref: Ibid.											
Color: Yellow, black											
CAS #: 409-21-2 IAI corrected for tungsten carbide contamination. C-Si type. Wurtzite group, moissanite subgroup. PSC: hP12. To replace 22-1273 and validated by calculated patterns 29-1128 and 31-1232. Mwt: 40.10. Volume[CD]: 123.33.											

©1996 JCPDS-International Centre for Diffraction Data. All rights reserved.

10-0173		Wavelength= 1.5405									
Al <sub>2</sub> O <sub>3</sub>		d Å	Int	h	k	l	d Å	Int	h	k	l
Aluminum Oxide		3.47900	75	0	1	2	0.934500	4	3	1	8
		2.55200	90	1	0	4	0.917800	4	2	2	9
		2.37900	40	1	1	0	0.907600	14	3	2	4
Corundum, syn		2.16500	<1	0	0	6	0.905200	4	0	1	4
Rad.: CuKα1: 1.5405 Filter: Ni Beta M d-sp:		2.08500	100	1	1	3	0.899100	8	4	1	0
Cut off Int. Diffract. I/teor.: 1.00		1.96400	2	2	0	2	0.888400	<1	2	3	5
Ref: Natl. Bur. Stand. (U.S.), Circ. 539, 9, 3 (1960)		1.74000	45	0	2	4	0.880400	4	4	1	3
		1.60100	80	1	1	6	0.869800	2	0	4	8
		1.54600	4	2	1	1	0.858000	12	1	3	10
		1.51400	6	1	2	2	0.850200	4	3	0	12
		1.51000	8	0	1	8	0.846000	4	2	0	14
Sys: Rhombohedral S.G.: R3c (167)		1.40400	30	2	1	4	0.830300	22	1	4	6
a: 4.758	b:	c: 12.991	A:	C: 2.7303							
		1.37400	50	3	0	0	0.815700	4	1	1	15
		1.33700	2	1	2	5	0.807200	11	4	0	10
		1.27600	4	2	0	8	0.798800	7	0	5	4
Ref: Ibid.		1.23900	16	1	0	10	0.797000	14	1	0	16
		1.23430	8	1	1	9	0.793100	13	3	3	0
		1.18980	8	2	2	0					
Dx: 3.989 Dm: 4.050 SS/FOM <sub>3/2</sub> -50(0188, 32)		1.16000	<1	3	0	6					
		1.14700	6	2	2	3					
		1.13820	2	1	3	1					
		1.12550	6	3	1	2					
Ref: Dana's System of Mineralogy, 7th Ed., I, 520		1.12460	4	1	2	8					
		1.09880	8	0	2	10					
		1.08310	4	0	0	12					
		1.07810	8	1	3	4					
		1.04260	14	2	2	6					
		1.01750	2	0	4	2					
		0.997600	12	2	1	10					
		0.985700	<1	1	1	12					
		0.981900	4	4	0	4					
		0.943100	<1	3	2	1					
		0.941300	<1	1	2	11					
Color: Blue, colorless, yellow											
Pattern taken at 26 C. Sample annealed at 1400 C for four hours in an Al <sub>2</sub> O <sub>3</sub> crucible. Spectroscopic analysis showed <0.1% K, Na, Si; <0.01% Ca, Cu, Fe, Mg, Pb; <0.001% B, Cr, I, Mn, Ni. Also called: ruby. Also called: sapphire. Al <sub>2</sub> O <sub>3</sub> type. Corundum group, corundum subgroup. Also called: alumina. Also called: diamonite. PSC: hR10. Mwt: 101.96. Volume[CD]: 254.70.											

©1996 JCPDS-International Centre for Diffraction Data. All rights reserved.

41-1105		Wavelength= 1.540598									
Y2O3		d Å	Int	h	k	l	d Å	Int	h	k	l
Yttrium Oxide											
		5.30700	<1	2	0	0	1.24960	1	6	6	0
		4.33000	13	2	1	1	1.23270	3	8	3	1
		3.06100	100	2	2	2	1.21630	6	6	6	2
		2.83500	<1	3	2	1	1.20070	<1	7	5	2
		2.65100	24	4	0	0	1.18560	4	8	4	0
Rad.: CuKα1: 1.5405 Filter: Graph Monoc-d-sp: Diffractometer		2.49900	5	4	1	1	1.17110	1	8	3	3
Cut off: 15.0 Int.: Diffract. I/cor.: 9.1		2.37150	1	4	2	0	1.15700	1	8	4	2
Ref: Martin, K., McCarthy, G., North Dakota State University, Fargo, North Dakota, USA, ICDD Grant-in-Aid, (1989)		2.26050	6	3	3	2	1.14350	2	9	2	1
		2.16450	1	4	2	2	1.13050	<1	6	6	4
		2.07880	8	4	3	1	1.11780	2	8	5	1
		1.93570	2	5	2	1	1.09380	2	9	3	2
Sys.: Cubic S.G.: Ia3 (206)		1.87400	39	4	4	0	1.08230	4	8	4	4
a: 10.6041(5)b:		1.81830	2	4	3	3	1.07120	2	9	4	1
c:		1.76770	<1	6	0	0	1.06040	1	10	0	0
α:		1.72000	5	6	1	1	1.05000	1	10	1	1
β:		1.67670	1	6	2	0	1.03990	3	10	2	0
γ:		1.63610	4	5	4	1	1.03000	1	9	4	3
Z: 16 mp: 2440		1.59840	25	6	2	2	1.02040	3	10	2	2
Ref: Ibid.		1.56330	5	6	3	1	1.01110	2	10	3	1
Dx: 5.031 Dm: SS/FOM3(=160(0060, 31))		1.53040	4	4	4	4	993200	2	8	7	1
Color: White		1.49950	2	5	4	3	984500	2	10	4	0
Peak height intensity. Sample obtained from Research Chemicals, Phoenix, Arizona, USA. Annealed for 48 hours at 1200 °C. Average relative standard deviation in intensity of the ten strongest reflections for three specimen mounts - 2%. Validated by a calculated pattern. Dixbyite, Mn2 O3 type. Also called: yttria. Silicon used as an internal stand. Single-crystal data used. PNC: 0180. To replace 25-1200. Mwt: 225.81. Volume[CD]: 1192.40.		1.47050	1	6	4	0	976200	1	10	3	3
		1.44300	3	7	2	1	968000	2	10	4	2
		1.41700	1	6	4	2	960100	1	9	5	4
		1.34670	2	6	5	1	944800	2	11	2	1
		1.32550	4	8	0	0	937400	1	8	8	0
		1.30520	3	8	1	1	923100	1	10	4	4
		1.28590	2	8	2	0	916000	1	11	3	2
		1.26740	2	6	5	3	909300	1	10	6	0

d Å	Int	h	k	l
902700	1	11	4	1
896300	2	10	6	2
889700	<1	9	6	3

©1996 JCPDS-International Centre for Diffraction Data. All rights reserved.

33-0040		Wavelength= 1.540598									
Al5Y3O12		d Å	Int	h	k	l	d Å	Int	h	k	l
Aluminum Yttrium Oxide											
		4.90500	27	2	1	1	1.26560	1	8	5	1
		4.24700	7	2	2	0	1.23880	2	9	3	2
		3.21000	19	3	2	1	1.22570	<1	8	4	4
		3.00200	27	4	0	0	1.21280	<1	9	4	1
Rad.: CuKα1: 1.5405 Filter: Mono d-sp: Diffractometer		2.68700	100	4	2	0	1.20110	<1	8	6	0
Cut off: Int.: Diffract. I/cor.:		2.56100	<1	3	3	2	1.18890	<1	10	1	1
Ref: Natl. Bur. Stand. (U.S.) Monogr. 25, 19, 11 (1982)		2.45200	20	4	2	2	1.17760	2	10	2	0
		2.35500	6	4	3	1	1.16650	<1	9	4	3
		2.19200	23	5	2	1	1.14510	3	10	3	1
		2.12200	5	4	4	0	1.12490	<1	8	7	1
		1.94740	26	5	3	2	1.11510	14	10	4	0
Sys.: Cubic S.G.: Ia3d (230)		1.89940	<1	6	2	0	1.10560	2	9	6	1
a: 12.0089(3)b:		1.85360	<1	5	4	1	1.09640	6	10	4	2
c:		1.77050	2	6	3	1	1.08740	<1	9	5	4
α:		1.73300	17	4	4	4	1.06980	3	10	5	1
β:		1.69880	<1	5	4	3	1.06160	6	8	8	0
γ:		1.66520	31	6	4	0	1.03730	2	9	7	2
Z: 8 mp		1.63380	9	7	2	1	1.02980	1	10	6	0
Ref: Ibid.		1.60460	28	6	4	2	1.02230	<1	11	4	1
Dx: 4.553 Dm: SS/FOM3(=89(0112, 30))		1.52470	4	6	5	1	1.01500	<1	10	6	2
Color: White		1.50060	10	8	0	0	1.00770	<1	9	6	5
Pattern at 25 °C. The sample was prepared at NBS, Gaithersburg, Maryland, USA. CAS #: 12005-21-9. Stoichiometric amounts of the constituent oxides were blended and calcined at 1650 °C for two hours. After grinding, the resultant product was placed in an iridium crucible, fused in an induction heater and several single-crystal boules grown using the Czochralski technique. Silicon used as an internal stand. PNC: 0160. To replace 8-178, 30-51. Mwt: 593.62. Volume[CD]: 1731.85.		1.47800	<1	7	4	1	1.00070	1	12	0	0
		1.45610	<1	8	2	0	993900	<1	9	7	4
		1.43520	1	6	5	3	987200	3	12	2	0
		1.41570	1	6	6	0	980500	2	11	5	2
		1.39620	<1	7	4	3	974100	6	10	6	4
		1.35980	<1	7	5	2	967600	<1	12	3	1
		1.34230	7	8	4	0					
		1.31020	17	8	4	2					
		1.29490	2	7	6	1					
		1.28000	6	6	6	4					

©1996 JCPDS-International Centre for Diffraction Data. All rights reserved.

## ประวัติผู้เขียนวิทยานิพนธ์

นายกานต์ เสรีวัลย์สถิตย์ เกิดวันที่ 14 กรกฎาคม พ.ศ. 2521 ที่กรุงเทพมหานคร สำเร็จการศึกษาปริญญาตรี วิทยาศาสตร์บัณฑิตจาก ภาควิชาวัสดุศาสตร์ คณะวิทยาศาสตร์ เมื่อปี การศึกษา 2542 และได้เข้าศึกษาต่อในหลักสูตรวิทยาศาสตรมหาบัณฑิต สาขาเทคโนโลยีเซรามิก จุฬาลงกรณ์มหาวิทยาลัย ในปีการศึกษา 2543 สำเร็จการศึกษาในภาคการศึกษาปลาย ปีการศึกษา 2546



ศูนย์วิทยทรัพยากร  
จุฬาลงกรณ์มหาวิทยาลัย