CHAPTER IV

RESULTS AND DISCUSSION

The pulverized seeds of *Pachyrrhizus erosus* (L.) Urban (2 kg) were extracted with hexane, chloroform and then ethanol. The oil portion from hexane extract was analyzed by Gas-Liquid Column chromatography (GLC) to afford the percentages of fatty acid composition. The chloroform and the ethanol extracts were investigated by several chromatographic techniques to give eight compounds classified as an isoflavone [2], an isoflavanone [15], a 3-phenylcoumarin [17] and five rotenoids [4, 8, 11, 12 and 18]. Their completed structures were determined based on their UV, IR, MS and NMR data, and then discussed by comparison with the literature values. The antimicrobial, anti-HSV, COX-2 inhibitory and cytotoxic activities of these compounds were evaluated.

The ethanol extract of dried stem bark of *Millettia leucantha* Kurz var. *leucantha* (2.7 kg) was investigated by means of chromatographic methods to yield eleven compounds classified as six chalcones [102, 279, 280, 281, 282 and 284] and five flavones [68, 103, 115, 285 and 287]. The structure determinations of these compounds were achieved by interpretation of their UV, IR, MS and NMR data, and then confirmed by comparison with the literature values. Additionally, their antimicrobial, anti-HSV, COX-2 inhibitory and cytotoxic activities were also discussed.



1. Determination of Oil Compositions from Pachyrrhizus erosus seeds

The oil obtained from the seeds of *P. erosus* is clearly yellow. After analysis by GC, the percentages of normal fatty acids were determined and reported in the following table. GC chromatogram (**Figure 12**) were demonstrated in Appendix part.

Peak number	Component name	%Area
1	Capric acid (C10:0) ^A [268]	0.035
2	Myristic acid (C14:0) [269]	0.249
3	Palmitic acid (C16:0) [270]	28.302
4	Haxadecenoic acid (C16:1) [271]	0.193
5	Stearic acid (C18:0) [272]	5.776
6	Oleic acid (C18:1) [273]	29.258
7	Linoleic acid (C18:2) [274]	31.380
8	Linolenic acid (C18:3) [275]	0.790
9	Eicosenoic acid (C20:1) [276]	0.960
10	Docosenoic acid (C22:1) [277]	1.746
11	Docosahexaenoic acid (C22:6) [278]	1.311

A: Abbreviation (C 10:0); C = Carbon atom, 10 = Number of carbon atoms,

0 = Number of double bonds

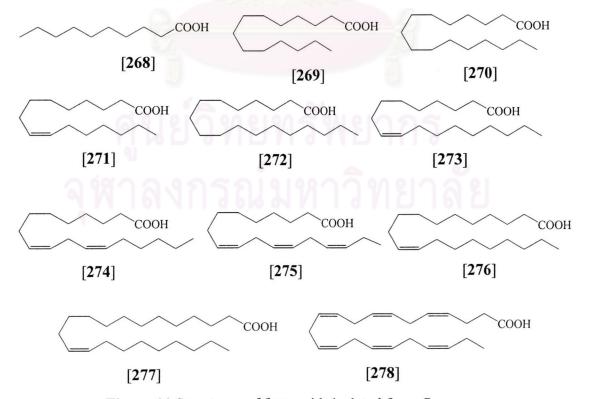


Figure 11 Structures of fatty acids isolated from P. erosus

2. Structure Determination of Compounds Isolated from

Pachyrrhizus erosus

2.1 Structure Determination of Compound 4

Compound 4 was obtained as white crystals. The EI mass spectrum (Figure 13) showed the molecular ion peak at m/z 336, consistent with the molecular formula C₁₉H₁₂O₆. The UV spectrum (**Figure 14**) showed absorption maxima at 341, 302, 275, 236 and 209 nm. The IR spectrum (Figure 15) displayed absorption bands at 1681 (conjugated C=O stretching), 1624-1469 (aromatic ring) and 1154 (C-O stretching) cm⁻¹. This compound was assigned to rotenoid, (+)-dolineone [4], which had the cis-B/C fusion like (-)-rotenone [19]. This conclusion was comfirmed by the $[\alpha]^{27}$ _D (+192, c=0.35 in CHCl₃) and ¹H-NMR spectrum (**Figure 16**), which indicated that the H-1 is located at δ 6.74 ppm and was therefore not being strongly negatively shielded by the carbonyl group. All signals of ¹H-NMR spectrum were exhibited at the locations, corresponding to those of literature (Puyvelde, 1987). The ¹³C-NMR spectrum (Figure 17) showed the signals at 159.8 and 158.6 ppm due to the carbons 7a and 9, respectively. These signals had been conversely assigned to those in the literature (Puyvelde, 1987). The present work completely assigned the ¹H- and ¹³C-NMR data of this compound by HMQC (Figure 18), HMBC (Figure 19) experiments and compared all signals with those of literature (Table 10).

Table 10 The ¹H-and ¹³C-NMR data of Compound 4 in CDCl₃

	δ_{H} (ppm), J (Hz)		$\delta_{\rm C}$ (pp	m)
Н	(+)-Dolineone	Compound 4	C	(+)-Dolineone	Compound 4
1	6.72 (1H,s, overlap)	6.74 (1H, s, overlap)	1	106.9	106.9
			2	143.2	142.3
			3	147.9	147.9
4	6.44 (1H, s)	6.45 (1H, s)	4	98.9	98.6
6	4.19 (1H, <i>d</i> , 12.0)	4.19 (1H, <i>d</i> , 12.0)	6	66.4	66.3
	4.63 (1H, <i>dd</i> , 12.0,	4.64 (1H, dd, 12.0,		8	
	3.2)	3.5)			
8	7.05 (1H, s)	7.06 (1H, s)	8	99.8	99.8
			9	158.6	159.8
			10	123.1	123.0
11	8.21 (1H, s)	8.21 (1H, s)	11	121.0	120.9
			12	190.6	190.5
			4a	148.5	148.4
6a	4.96 (1H, ddd, 3.9,	4.96 (1H, ddd, 4.0,	6a	72.1	72.0
	3.2, 1.0)	3.0, 1.0)			
			7a	159.8	158.6
			11a	116.1	116.0
12a	3.89 (1H, d, 3.9)	3.89 (1H, d, 4.0)	12a	45.3	45.2
			12b	105.3	105.3
2"	7.54 (1H, <i>d</i> , 2.3)	7.54 (1H, d, 2.5)	2"	146.2	146.2
3"	6.73 (1H, dd, 2.3, 1)	6.73 (1H, dd, 2.5, 1)	3"	106.9	106.8
2""	5.80 (1H, d, 1.3)	5.81 (1H, d, 1.3)	2""	101.2	101.1
	5.86 (1H, d, 1.3)	5.87 (1H, <i>d</i> , 1.3)	۵	ω	

^{-:} The bold values are revised assignments.

2.2 Structure Determination of Compound 15

Compound 15 was acquired as pale yellow crystals. The EI mass spectrum (Figure 20) revealed the molecular ion peak at *m/z* 338, corresponding to the molecular formula C₁₉H₁₄O₆. The UV spectrum (Figure 21) showed absorptions at 336, 299, 273, 235 and 207 nm. The IR spectrum (Figure 22) displayed absorption bands at 2893 (CH stretching), 1687 (C=O stretching) and 1625-1475 (aromatic ring) cm⁻¹. This compound was identified as neotenone [15] and has already been isolated from this plant (Krishnamurti and Seshadri, 1966), *Neorautanenia pseudopachyrrhiza* (Crombie and Whiting, 1963) and *N. mitis* (Puyvelde, 1987). Additionally, this compound was always isolated in racemate form (Puyvelde, 1987). The ¹H-NMR (Figure 23) data exhibited close similarity to those in the literature (Puyvelde, 1987). The ¹³C-NMR spectrum (Figure 24) showed the signals of the carbons 6 and 1' at δ 115.5 and 122.6 ppm, respectively. These were revised from previously report (Puyvelde, 1987). This assignment was confirmed by the application of HMQC (Figure 25) and HMBC (Figure 26) experiments. The ¹H- and ¹³C-NMR data were demonstrated in Table 11.

Table 11 The ¹H- and ¹³C-NMR data of Compound 15 in CDCl₃

	δ _H (ppm	J(Hz)		δ _C (ppr	
Н	Neotenone	Compound 15	С	Neotenone	Compound
		5			15
2	4.50(1H,dd,10.8,5.4)	4.50(1H,dd,10.6,5.5)	2	71.3	71.3
	4.58(1H,dd,10.8,11.4)	4.56(1H,dd,10.6,11.3)			
3	4.31(1H,dd, 11.4,5.4)	4.31(1H,dd, 11.3,5.5)	3	48.3	48.3
			4	192.6	192.8
5	8.25(1H, s)	8.25(1H, s)	5	120.8	120.9
			6	115.6	122.6
			7	159.2	159.3
8	7.08(1H, s)	7.08(1H, s)	8	99.6	99.7
			4a	118.8	118.8
		// 9.393.60	8a	159.5	159.9
		// 3. 6 4	1'	122.6	115.5
			2'	152.8	152.7
3'	6.57(1H, s)	6.57(1H, s)	3'	95.4	95.4
		ANGLESIS	4'	147.8	147.8
		(1) (8) (8) (8) (8) (8) (8) (8) (8) (8) (8	5'	141.4	141.3
6'	6.62(1H, s)	6.61(1H, s)	6'	109.8	109.8
2"	7.57(1H, <i>d</i> , 2.3)	7.57(1H, d, 2.2)	2"	146.0	146.0
3"	6.76(1H, dd, 2.3,1.0)	6.76(1H, dd, 2.2,1.0)	3"	107.0	107.0
2""	5.90(2H, s)	5.91(2H, s)	2"'	101.3	101.3
OCH ₃	3.72(3H, s)	3.72(3H, s)	OCH ₃	56.5	56.5

^{-:} The bold values are revised assignments.

2.3 Structure Determination of Compound 18

Compound 18 was obtained as white crystals. The EI mass spectrum (Figure 27) exhibited the molecular ion peak at m/z 366, consistent to the molecular formula $C_{20}H_{14}O_7$. The UV spectrum (Figure 28) displayed absorptions at λ_{max} 344, 283, 243 and 208 nm. The IR spectrum (Figure 29) showed absorption bands at 1676 (conjugated C=O stretching) and 1481-1619 (aromatic ring) cm⁻¹. The $[\alpha]^{27}_D$ showed dextrorotatory dispersion at + 116.44 (c=0.45, CHCl₃), which presents the cis-B/C fusion like (-)-rotenone [19] and (+)-dolineone [4]. This compound was assigned to pachyrrhizone [18] already isolated from this plant (Norton and Hansberry, 1945) and *Neorautanenia* species (Crombie and Whiting, 1963). However, the ¹H- and ¹³C-NMR data (Figure 30 and 31, respectively) have not been reported. Therefore, the present work completely reported these data of compound 18 for the first time (Table 12). This assignment was confirmed by HMQC (Figure 32) and HMBC (Figure 33) experiments.

Table 12 The ¹H- and ¹³C-NMR data of Compound 18 in CDCl₃

$\delta_{\rm H}$ (ppm), J (Hz) of Compound 18	С	δ_{C} (ppm) of Compound 18
6.71 (1H, s)	1	106.9
	2	142.2
	3	147.3
6.42 (1H, s)	4	98.9
4.2 (1H, <i>d</i> , 11.9)	6	66.2
4.6 (1H, <i>dd</i> , 11.9, 3.5)		
	8	133.5
	9	150.9
	10	123.9
7.91 (1H, s)	11	114.0
	12	190.7
	4a	148.6
4.97 (1H, <i>m</i>)	6a	72.3
	7a	149.7
	11a	117.1
3.88 (1H, d, 3.84)	12a	45.2
	12b	105.2
7.55 (1H, <i>d</i> , 2.2)	2"	146.1
6.73 (1H, d, 2.2)	3"	107.3
5.80 and 5.85 (each 1H, d, 1.2)	2'''	101.2
4.13 (3H, s)	OCH ₃	61.1
	6.42 (1H, s) 4.2 (1H, d, 11.9) 4.6 (1H, dd, 11.9, 3.5) 7.91 (1H, s) 4.97 (1H, m) 3.88 (1H, d, 3.84) 7.55 (1H, d, 2.2) 6.73 (1H, d, 2.2)	2 3 6.42 (1H, s) 4 4.2 (1H, d, 11.9) 6 4.6 (1H, dd, 11.9, 3.5) 8 9 10 7.91 (1H, s) 11 12 4a 4a 6a 7a 11a 12a 12b 7.55 (1H, d, 2.2) 2" 6.73 (1H, d, 2.2) 3"

2.4 Structure Determination of Compound 17

Compound 17 was appeared as green needles. The EI mass spectrum (**Figure 34**) displayed the molecular ion peak at m/z 336, consistent with $C_{19}H_{12}O_6$. The UV absorption bands (**Figure 35**) appeared at λ_{max} 348, 292, 242 and 210 nm. The IR absorption spectrum (**Figure 36**) showed v_{max} at 1716 (conjugated C=O stretching) and 1429 and 1625 (aromatic) cm⁻¹. This compound demonstrated identical formula with that of compound 4 but different in other major fragment ions in EIMS appeared at m/z 293, 265 and 179. This compound was identified as pachyrrhizin [17] by analyses of the ¹H- (**Figure 37**) and ¹³C-NMR (**Figure 38**) data, and by comparison with these data in previously reports (Puyvelde, 1987). Additionally, the carbon signal at δ 116.1 was newly assigned to be carbon 1", whereas the carbons 7, 2' and 8a should be located at δ 156.1, 152.9 and 151.6, respectively. **Table 13** showed ¹H- and ¹³C-NMR data from the present work compared to these data from literature. The completed assignment of this compound was successfully done by application of 2D-NMR such as HMQC (**Figure 39**) and HMBC (**Figure 40**) experiments.

Table 13 The ¹H- and ¹³C-NMR data of Compound 17

	δ _H (ppm	J(Hz)		δ _H (pp	om)
Н	Pachyrrhizin	Compound 17	С	Pachyrrhizin	Compound
	(in CDCl ₃)	(in CDCl ₃)		(in CDCl ₃ -	17
				DMSO- d_6 =1:5)	(in CDCl ₃)
			2	173.2	160.7
		s Autoria	3	124.0	123.9
4	7.80 (1H, s)	7.80 (1H, s)	4	142.4	142.4
5	7.69 (1H, s)	7.68 (1H, s)	5	119.6	119.6
	4		6	124.8	124.8
			7	156.2	156.1
8	7.46 (1H, s)	7.49 (1H, s)	8	99.5	99.4
			4a	116.2	116.1
		///////////////////////////////////////	8a	156.2	151.6
		// h [6] A\	1'	-	116.1
			2'	151.7	152.9
3'	6.64(1H, s)	6.63 (1H, s)	3'	95.5	95.4
		ANAS CONTRACTOR	4'	148.8	148.7
		Commence of the Commence of th	5'	141.3	141.2
6'	6.91 (1H, s)	6.89 (1H, s)	6'	110.3	110.3
2"	7.70 (1H, <i>d</i> , 2.2)	7.69 (1H, <i>d</i> , 2.4)	2"	146.7	146.7
3"	6.84(1H,dd,2.2,1.0)	6.82(1H,dd,2.4,0.5)	3"	106.4	106.4
2""	5.98 (2H, s)	5.96 (2H, s)	2"'	101.5	101.5
OCH ₃	3.79 (3H, s)	3.77 (3H, s)	OCH ₃	56.9	56.8

^{-:} The bold values are revised assignments.

2.5 Structure Determination of Compound 8

Compound 8 was acquired as pale yellow crystals. The EIMS (Figure 41) displayed the molecular ion peak at m/z 352, agreeing with the molecular formula C₁₉H₁₂O₇. The UV spectrum (Figure 42) showed the absorption maxima identical with those of compound 4 at λ_{max} 339, 304, 276, 237 and 208 nm. The IR spectrum (Figure 43) revealed absorption bands at 3461 (OH stretching), 2907 (CH stretching), 1682 (C=O stretching) and 1480 and 1625 (aromatic ring) cm⁻¹. The $\left[\alpha\right]^{27}$ _D was +140.0, it was therefore the cis-B/C fusion like compound 4. The ¹H-NMR spectrum (Figure 44) showed all signals corresponding to those in the literature (Puyvelde, 1987). This compound was identified as (+)-12a-hydroxydolineone [8], derivative of compound 4. The furano protons (H-2" and H-3") connected to the A-ring were observed at δ 7.55 (H-2", d, J=2.3 Hz) and 6.74 (H-3", dd, J=2.3, 1 Hz). Four singlet aromatic protons were established as H-1 (δ 6.51 ppm), H-4 (δ 6.47 ppm), H-8 (δ 7.02 ppm) and H-11 (δ 8.19 ppm). The methylene protons at C-6 showed the signals at δ 4.50 (1H, dd, J=15.8, 4 Hz) and 4.63 (1H, dd, J=15.8, 2.6 Hz), whereas the doublet of doublet signals at δ 4.62 (1H, dd, J=4.0, 2.6 Hz) belonged to H-6a. A pair of doublet signals at δ 5.80 and 5.84 (each 1H, each J=1.4 Hz) were assigned to methylenedioxy group connected to the D-ring of rotenoid skeleton. The ¹³C-NMR spectrum (Figure 45) has been reported already (Puyvelde, 1987) but some positions should be revised as C-1 (δ105.8 ppm), C-9 (δ 160.3 ppm), C-3" (δ 106.9 ppm) and C-7a (\delta 158.3 ppm). This assignment was supported by HMQC (Figure 46) and HMBC (Figure 47) experiments.

Table 14 The ¹H- and ¹³C-NMR data of Compound 8 in CDCl₃

	$\delta_{\rm H}(\rm ppm)$), <i>J</i> (Hz)		δ_{C} (ppm)
Н	(+)-12a-	Compound 8	C	(+)-12a-	Compound
	Hydroxydolineone			Hydroxydolineone	8
1	6.52 (1H, s)	6.51 (1H, s)	1	106.8	105.8
		*.	2	142.3	142.3
		- Andrea	3	149.5	149.5
4	6.48 (1H, s)	6.47 (1H, s)	4	99.9	99.3
6	4.50(1H,dd,12.9,2.0)	4.50(1H,dd,15.8,4.0)	6	63.9	64.0
	4.63(1H,dd,12.9,2.4)	4.63(1H,dd,15.8,2.6)			
8	7.02(1H, s)	7.02 (1H, s)	8	100.0	100.1
			9	158.3	160.3
			10	123.3	123.4
11	8.19 (1H, s)	8.19 (1H, s)	11	121.0	121.1
		/// 3. TO A	12	192.9	193.0
			4a	149.6	149.6
6a	4.62 (1H, <i>br s</i>)	4.62 (1H, dd,4.0,2.6)	6a	75.9	75.9
		ANNO GONOSIONA RECELLO CONSTRUITA	7a	160.3	158.3
			11a	114.6	114.6
		100 100 100 100 100 100 100 100 100 100	12a	68.3	68.4
			12b	109.2	109.3
2"	7.55 (1H, d, 2.2)	7.55 (1H, d, 2.3)	2"	146.4	146.5
3"	6.74(1H, dd,2.2, 1.0)	6.74(1H, dd,2.3, 1.0)	3"	105.7	106.9
2""	5.80 and 5.85 (each	5.80 and 5.84 (each	2"'	101.3	101.4
	1H, each <i>d</i> , 1.2)	1H, each d, 1.4)	NE	6171	
ОН	4.46 (1H, s)	4.43 (1H, s)		U_	

^{-:} The bold values are revised assignments.

2.6 Structure Determination of Compound 2

Compound 2 was characterized as white crystals. The EI mass spectrum (Figure 48) demonstrated the molecular ion peak at m/z 336, harmonising with the The UV spectrum (Figure 49) showed the molecular formula C₁₉H₁₂O₆. characteristics of the furanoisoflavonoid chromophore at 303, 237 and 208 nm. The IR absorption spectrum (Figure 50) displayed ν_{max} at 1645-1622 and 1474 $cm^{\text{--}1}$ (C=O stretching and aromatic ring). This compound was identified as dehydroneotenone Its isolation from Neorautanenia mitis and ¹H-NMR data were reported (Puyvelde, 1987). The present work would report the ¹³C-NMR spectral data (Figure **52**) at the first time. The ${}^{1}\text{H-NMR}$ data (Figure 51) showed a singlet signal at δ 7.98 ppm. This was evidence of H-2 of isoflavone. Four singlet aromatic protons of H-5, H-8, H-3' and H-6' showed signals at δ 8.54, 7.57, 6.63 and 6.85 ppm, respectively. Thus, the isoflavone skeleton was substituted by furan ring (7.72 ppm, H-2"and 6.91 ppm, H-3") at positions 6 and 7. The 3 positions on phenyl ring were substituted by OCH₃ (8 3.73 ppm) at 2' and methylenedioxy group (8 5.96 ppm) at 3'and 4'. The complete assignment was managed by performing HMQC (Figure 53) and HMBC (Figure 54) experiments.

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Table 15 The ¹H- and ¹³C-NMR data of Compound 2

	δ_{H} (ppm), J (Hz)		δ_{H} (ppm) of
Н	Dehydroneotenone	Compound 2	C	Compound 2
	(in CDCl ₃ :DMSO- d_6 =1:4)	(in CDCl ₃)		(in CDCl ₃)
2	7.98 (1H, s)	7.98 (1H, s)	2	154.7
			3	121.1
		- Arden	4	176.6
5	8.28 (1H, s)	8.54 (1H, s)	5	119.0
			6	126.0
		9	7	157.2
8	7.63 (1H, <i>br</i> s)	7.57 (1H, s)	8	99.8
			4a	121.1
			8a	154.2
		9 202 (0)	1'	112.8
			2'	153.0
3'	6.64 (1H, s)	6.63 (1H, s)	3'	95.5
	- / // 9	1331031063	4'	148.4
			5'	141.2
6'	6.70 (1H, s)	6.85 (1H, s)	6'	111.3
2"	7.88 (1H, d, 2.0)	7.72 (1H, d, 2.2)	2"	147.4
3"	6.93 (1H, dd, 2.0, 1.0)	6.91 (1H, <i>d</i> , 2.2)	3"	107.0
2"'	5.86 (2H, s)	5.96 (2H, s)	2""	101.4
OCH ₃	3.78 (3H, s)	3.73 (3H, s)	OCH ₃	56.9
	ศูนย์วิท จุฬาลงกรถ	ยทรัพย น์มหาวิ	ากร ทยา	ลัย

2.7 Structure Determination of Compound 11

Compound 11 was acquired as pale yellow crystals. The EIMS (Figure 55) exhibited the molecular ion peak at m/z 382, suggesting the molecular formula $C_{20}H_{14}O_8$. The UV spectrum (**Figure 56**) revealed absorptions at λ_{max} 349, 284, 243 and 206 nm. The IR spectrum (Figure 57) provided absorption bands at 3461 (OH stretching), 2937 (CH stretching), 1682 (C=O stretching) and 1482 and 1621 (aromatic ring) cm⁻¹. The ¹H- and ¹³C-NMR spectral data (Figures 58 and 59, respectively) showed very similar patterns with those of (+)-pachyrrhizone [18], whereas MS data and IR spectrum confirmed the presence of hydroxy group. Furthermore, the connection of B and C-rings was proved to be the cis-B/C fusion as those of compound 18 and (-)-rotenone [19] by exhibiting of $[\alpha]^{27}$ _D at +91.6 and displaying of the singlet signal of H-1 at δ 6.49 ppm, which was highly deshielded Thus, this compound was eventually identified as (+)-12a-hydroxy located. pachyrrhizone [11]. In the ¹H-NMR, two protons at δ 5.80 and 5.84 ppm were characterized as methylenedioxy group and the spin-coupled protons at 8 7.55 and 6.73 ppm (J=2.3 Hz) confirmed the presence of furan moiety connected to A-ring. Three benzenoid protons identified as H-1, H-4 and H-11 showed all singlet signals at δ 6.49, 6.46 and 7.88 ppm, respectively. Additionally, the methoxy substitution at C-8 displayed singlet signal at δ 4.10 ppm, whereas the signal at δ 4.40 ppm belonged to OH group at C-12, and three coupled protons at δ 4.50, 4.61 and 4.71 ppm analyzed for H-6 and H-6a. This compound was known, however, the ¹³C-NMR, HMQC (Figure 60) and HMBC (Figure 61) spectra were presented at the first time in this work.

Table 16 The ¹H- and ¹³C-NMR data of Compound 11 in CDCl₃

	δ_{H} (ppm),	, <i>J</i> (Hz)		δ_{C} (ppm) of
Н	(+)-12a-	Compound 11	C	Compound 11
	Hydroxypachyrrhizone			
1	6.45 (1H, s)	6.49 (1H, s)	1	105.6
			2	142.2
		s. Archele a	3	149.4
4	6.50 (1H, s)	6.46 (1H, s)	4	99.2
6	4.60 (2H, m)	4.50 (1H,dd, 12.1, 1.1)	6	63.8
		4.71 (1H,dd, 12.1, 2.4)		
			8	133.5
			9	151.4
			10	124.3
11	7.85 (1H, s)	7.88 (1H, s)	11	113.9
			12	193.1
			4a	149.7
6a	4.60 (1H, <i>m</i>)	4.61 (1H,dd, 2.4, 0.9)	6a	75.9
			7a	149.3
			11a	115.7
		5 5 5 4 1 1 1 1 5 5 5 5 5 5 5 5 5 5 5 5	12a	68.2
			12b	109.0
2"	7.55(1H, d, 3.0)	7.55(1H, <i>d</i> , 2.3)	2"	146.3
3"	6.70(1H, <i>d</i> , 3.0)	6.73 (1H, d, 2.2)	3"	107.2
2""	5.80 (2H, s)	5.90 and 5.84 (each H,	2""	101.3
	คนยวท	each <i>d</i> , 1.5)		
OCH ₃	4.10 (3H, s)	4.10 (3H, s)	OCH ₃	61.1
ОН	4.60 (1H, s)	4.40 (1H, s)	ยาลั	91

2.8 Structure Determination of Compound 12

Compound 12 was obtained as colourless oil. The EI mass spectrum (Figure 62) showed the molecular ion peak at m/z 410, belonging to the molecular formula $C_{23}H_{22}O_7$. The UV absorption spectrum (Figure 63) showed λ_{max} at 293, 244 and 205 nm. The IR spectrum (Figure 64) exhibited absorption bands at 3446 (OH stretching), 2962 (CH stretching), 1673 (conjugated C=O stretching) and 1507 and 1614 (aromatic ring) cm⁻¹. This compound provided $[\alpha]^{27}_D$ at -145, which was related to that of (-)-rotenone [19]. The ¹H-NMR spectral data (Figure 65) showed the singlet signal of H-1 at δ 6.55 ppm. These data confirmed the junction between B and C rings as cis-B/C ring junction. Compound 12 was determined as rotenone derivative, (-)-12a-hydroxyrotenone [12] previously isolated from this plant and Neorautanenia species (Oberholzer, Rall and Roux, 1974; Puyvelde, 1987). The pattern of ¹H-NMR in the present work was closely similar to that in the literature (Puyvelde, 1987) but the ¹³C-NMR data (Figure 66) should be revised some positions, including C-1 and C-12b. The C-1 should be located at δ 109.3 ppm, while the C-12b should be located at δ 108.7 ppm. The successful elucidation was supported by application of 2D-NMR as HMQC (Figure 67) and HMBC (Figure **68**).

Table 17 The ¹H- and ¹³C-NMR data of Compound 12 in CDCl₃

	δ _H (pp	om), J (Hz)		δ _C (ppn	n)
Н	(+)-12a-	Compound 12	С	(+)-12a-	Compound
	Hydroxyrotenone			Hydroxyrotenone	12
1	6.53(1H, s)	6.55 (1H, s)	1	108.8	109.3
			2	142.9	143.9
		s, Ardeba, a	3	151.2	151.1
4	6.44 (1H, s)	6.48 (1H, s)	4	101.1	101.0
6	4.50 (2H, m)	4.50 (1H, m)	6	63.9	63.8
		4.62 (overlap)			
	1		8	113.2	113.2
	4		9	168.0	168.0
10	6.80 (1H, d, 8.5)	6.53 (1H, d, 8.5)	10	105.3	105.3
11	7.70 (1H, d, 8.5)	7.82 (1H, d, 8.5)	11	130.2	130.1
		/// 3. G.A	12	191.1	191.1
	/	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	4a	148.4	148.3
6a	4.50 (1H, m)	4.58 (overlap)	6a	76.1	76.0
		William Comme	7a	157.7	157.7
			11a	111.8	111.7
		222000 4 300	12a	67.6	67.5
			12b	109.5	108.7
4'	~ 3 (1H, <i>m</i>)	2.94(1H,dd,16.0,9.0)	4'	31.1	31.1
	-	3.29(1H,dd,16.0,9.0)			
5'	5.20 (1H, m)	5.24 (1H, <i>t</i> , 9.0)	5'	88.0	87.9
	19 73 7	PINBILL	6'	142.9	142.8
7'	4.80-5.10(1H, m)	4.94 (1H, s)	7'	112.6	112.7
	ลหาลง	5.07 (1H, s)	779	เมาลม	
8'	1.73 (3H, <i>br s</i>)	1.76 (3H, s)	8'	17.1	17.1
OCH ₃	3.70 (3H, s)	3.72 (3H, s)	OCH ₃	55.9 and 56.4	55.8 and
	3.78 (3H, s)	3.82 (3H, s)			56.3
ОН	4.50 (1H, <i>br s</i>)	4.48 (1H, s)			

^{-:} The bold values are revised assignments.

3. Structure Determination of Compounds Isolated from *Millettia* leucantha

3.1 Structure Determination of Compound 279

Compound 279 was acquired as yellow needles. The EI mass spectrum (Figure 69) displayed the molecular ion peak at m/z 312, agreeing with the molecular formula C₁₈H₁₆O₅, which was supported by elemental analysis (Anal. Calcd for C₁₈H₁₆O₅.1/6H₂O: C, 68.60; H, 5.21. Found: C, 68.66; H, 5.00). The UV spectrum (Figure 70) showed λ_{max} at 348, 304, 245 and 206 nm. The IR absorption spectrum (Figure 71) exhibited v_{max} at 1651 (C=O stretching) and 1489 and 1601 (aromatic This compound was analyzed as 2',4'-dimethoxy-3,4-methylene dioxychalcone. The ¹H-NMR spectrum (Figure 72) showed a set of trans-olefinic protons at δ 7.35 (H- α) and 7.60 (H- β) ppm, each proton exhibited a coupling constant J=16 Hz. This spectrum also revealed the presence of aromatic coupling pattern of two ABX systems, a methylenedioxy group at δ 6.01 (2H, s) and two methoxy substituents at δ 3.87 and 3.91 ppm. Additionally, two fragments of EIMS at m/z 165 [2,4-(MeO)₂C₆H₃-CO⁺, 43%] indicated that two methoxy groups were located on the A ring, while a methylenedioxy group was on the B ring. This compound has already been synthesized for cosmetic purpose (Salem et al., 2000). The present work, however, was the first time to report this compound as a natural product. The precise assignment was proceeded by the ¹³C-NMR (Figure 73) and 2D-NMR as HMQC (Figure 74) and HMBC (Figure 75).

Table 18 The ¹H- and ¹³C-NMR data of Compound 279 in CDCl₃

	$\delta_{\rm H}$ (ppm), J (Hz) of		δ _C (ppm) of
Н	Compound 279	C	Compound 279
		1	129.9
2	7.12 (1H, <i>d</i> , 1.6)	2	106.6
	,	3	148.2
	5.0000	4	149.3
5	6.82 (1H, <i>d</i> , 8.0)	5	108.5
6	7.07 (1H, <i>dd</i> , 8.0,1.6)	6	124.7
		1'	122.3
		2'	160.3
3'	6.50 (1H, <i>d</i> , 2.0)	3'	98.6
		4'	164.0
5'	6.56 (1H, <i>dd</i> , 8.8,2.0)	5'	105.1
6'	7.75 (1H, <i>d</i> , 8.8)	6'	132.8
α	7.35 (1H, <i>d</i> , 16.0)	α	125.3
β	7.60 (1H, <i>d</i> , 16.0)	β	141.0
OCH ₂ O	6.01 (1H, s)	OCH ₂ O	101.4
		C=O	190.3
2'-OCH ₃	3.91 (3H, s)	2'-OCH ₃	55.7
4'-OCH ₃	3.87 (3H, s)	4'-OCH ₃	55.5

3.2 Structure Determination of Compound 280

Compound 280 was obtained as orange needles. The EI mass spectrum (**Figure 76**) displayed [M]⁺ at m/z 344, suggesting the molecular formula $C_{19}H_{20}O_6$. The UV absorption spectrum (Figure 77) provided λ_{max} at 370, 257 and 220 nm. The IR spectrum (Figure 78) exhibited absorption bands at 3445 (OH stretching) and 1622 (C=O stretching) cm⁻¹. In ¹H-NMR spectrum (**Figure 79**), a set of transolefinic protons at δ 7.75 (H- α) and 7.80 (H- β) (each d, J=15.5 Hz) and a chelated hydroxy group at δ 14.40 ppm assigned to OH-2' based on a 2'-hydroxychalcone were observed. This spectrum also exhibited the presence of four methoxy groups at δ 3.83, 3.91, 3.93 and 3.94 ppm. In the aromatic region, a set of two meta-coupled protons at δ 5.96 and 6.11 ppm (J=2 Hz), and one ABX system at δ 6.90(1H, d, J=8Hz), 7.13 (1H, d, J=2 Hz) and 7.21 (1H, dd, J=8, 2 Hz) were observed. The EIMS fragments at m/z 181 and 164 indicated that a hydroxy group and two methoxy groups were on the A ring, while two other methoxy groups were on the B ring. Consequently, this compound was determined to be 2'-hydroxy-3,4,4',6'tetramethoxychalcone [280]. It has already been isolated from Merrilla caloxylon (Rutaceae) (Fraser and Lewis, 1974) and Pongamia pinnata (Leguminosae) (Tanaka et al., 1992). The present work also reported the ¹³C-NMR spectral data (Figure 80) at the first time.

Table 19 The ¹H- and ¹³C-NMR data of Compound 280 in CDCl₃

	δ _H (ppm).	, <i>J</i> (Hz)		δ_{C} (ppm) of
Н	2'-hydroxy-3,4,4',6'-	Compound 280	C	Compound 280
	tetramethoxychalcone			
			1	128.6
2	7.13 (1H, d, 2.0)	7.13 (1H, <i>d</i> , 2.0)	2	110.4
		Andrea.	3	151.1
			4	149.1
5	6.90 (1H, d, 8.0)	6.90 (1H, d, 8.0)	5	111.2
6	7.22 (1H, <i>dd</i> , 8.0,2.0)	7.21 (1H, dd, 8.0,2.0)	6	122.6
			1'	106.3
		'	2'	168.4
3'	5.97 (1H, d, 2.0)	5.96 (1H, d, 2.0)	3'	91.2
		23.23 (8)	4'	166.0
5'	6.12 (1H, d, 2.0)	6.11 (1H, d, 2.0)	5'	93.8
			6'	162.4
α	7.75 (1H, d, 16.0)	7.75 (1H, <i>d</i> , 16.0)	α	125.4
β	7.82 (1H, <i>d</i> , 16.0)	7.80 (1H, d, 16.0)	β	142.6
			C=O	192.4
2'-OH	14.40 (1H, s)	14.40 (1H, s)		
OCH ₃	3.84 (3H, s)	3.83 (3H, s)	OCH ₃	55.5
OCH ₃	3.92 (9H, s)	3.91 (3H, s)	OCH ₃	55.7
OCH ₃	49)	3.93 (3H, s)	OCH ₃	55.8
OCH ₃	สาเย์วิท	3.94 (3H, s)	OCH ₃	56.0

3.3 Structure Determination of Compound 115

Compound 115, $[M+H]^+$ at m/z 293 in the ESI mass spectrum (**Figure 81**), was isolated as colourless plates from CHCl₃. The UV spectrum (Figure 82) showed a characteristic of furanoflavonoid chromophore at λ_{max} 304, 260 and 219 nm. The IR spectrum (Figure 83) displayed absorption bands at 1633 (C=O stretching) and 1625 and 1458 (aromatic ring) cm⁻¹. The ¹H-NMR (Figure 84), ¹³C-NMR (Figure 85) spectra and ESIMS confirmed the molecular formula of this compound as $C_{18}H_{12}O_4$. The *ortho*-coupled protons were observed at δ 8.21 (1H, d, J=8.8 Hz) and 7.56 that located the same position with the other three aromatic protons, including H-3', H-4' and H-5'. Thus, this position showed multiplet signal. Another multiplet signal at δ 8.15 ppm belonged to H-2' and H-6'. The ¹H-NMR spectrum also provided the signal of angular furan ring at δ 7.77 (H-2", d, J=2.4 Hz) and 7.19 (H-3", dd, J=2.4, 1.2 Hz). By comparison with very similar compound, lanceolatin B [103], this compound was identified as karanjin (3-methoxy derivative of lanceolatin B) [115]. This compound was well known from many plant species. In earlier work, no the ¹H-NMR and ¹³C-NMR was reported. Thus, the present work was the first report for these data.

[115]

Table 20 The ¹H- and ¹³C-NMR data of Compound 115 in CDCl₃

	$\delta_{\rm H}$ (ppm), J (Hz) of		$\delta_{C}(ppm)$ of
Н	Compound 115	C	Compound 115
		2	154.9
		3	141.8
		4	175.3
5	8.21 (1H, d, 8.8)	5	121.8
6	7.56 (1H, <i>m</i>)	6	110.0
		7	158.2
		8	117.0
		9	150.0
		10	119.7
		1'	131.1
2'	8.15 (1H, <i>m</i>)	2'	128.4
3'	7.56 (1H, m)	3'	128.7
4'	7.56 (1H, m)	4'	130.7
5'	7.5 <mark>6 (1H, m)</mark>	5'	128.7
6'	8.15 (1H, <i>m</i>)	6'	128.4
2"	7.77 (1H, d, 2.4)	2"	145.6
3"	7.19 (1H, dd, 2.4, 1.2)	3"	104.2
OCH ₃	3.93 (3H, s)	OCH ₃	60.3

3.4 Structure Determination of Compound 281

Compound 281 was obtained as a pale yellow oil and observed a molecular formula as $C_{19}H_{20}O_6$. The EIMS (**Figure 86**) exhibited a molecular ion peak at m/z344 and major fragment at m/z 195 [2,4,6-(MeO)₃C₆H₂CO⁺, 82 %] arose from the ketonic A ring fragment substituted by three methoxy groups, and other major peak at m/z 148 [3,4-OCH₂O-C₆H₃-CH₂=CH⁺, 100%] belonged to the B ring attached by HRFABMS showed m/z 345.1333 (M+H); (cald. For methylenedioxy group. C₁₉H₂₁O₆: 345.1347). The UV spectrum (**Figure 87**) showed maxima absorption bands at 285, 233 and 207 nm. The IR absorption spectrum (Figure 88) displayed v_{max} at 2940 (CH stretching), 1698 (C=O stretching) and 1606 and 1455 (aromatic ring) cm⁻¹. In the ¹H-NMR spectrum (Figure 89), the presence of two sets of methylene protons at δ 3.02 (2H, m, H- α) and 2.91 ppm (2H, m, H- β), the unclear ABX coupling pattern at δ 6.71 (2H,m, H-2 and H-5) and 6.65 (H-6, dd, J=7.5, 2.0 Hz), and symmetry meta-protons located at δ 6.09 ppm (2H, s, H-3' and H-5') were The ¹³C-NMR spectral data (Figure 90) showed three positions of symmetry carbons, including two methoxy carbons at δ 55.8, two methine carbons of the A-ring at δ 90.6 and two quarternary carbons of the A-ring at δ 158.2 ppm. Furthermore, the data from DEPT-135 (Figure 91) provided the appearance of three groups of methylene carbons assigned to one methylenedioxy group at δ 100.7 ppm, and α,β -ethylene carbon, typical of dihydrochalcone, at δ 46.5 and 29.7 ppm, respectively. These data indicated that compound 281 was 2',4',6'-trimethoxy-3,4methylenedioxydihydrochalcone. This work was the first report for this compound. The precise elucidation was achieved by 2D-NMR techniques as HMQC (Figure 92) and HMBC (Figure 93).

[281]

Table 21 The ¹H- and ¹³C-NMR data of Compound 281 in CDCl₃

	$\delta_{\rm H}$ (ppm), J (Hz) of		δ_{C} (ppm) of
Н	Compound 281	C	Compound 281
		1	135.4
2	6.71 (1H, d, 2.0)	2	108.9
		3	147.4
		4	145.5
5	6.71 (1H, <i>d</i> , 7.5)	5	108.0
6	6.65 (1H, dd, 7.5,2.0)	6	121.1
		1'	113.3
		2'	158.2
3'	6.09 (1H, s)	3'	90.5
		4'	162.3
5'	6.09 (1H, s)	5'	90.5
	/// h TO	6'	158.2
α	3.02 (2H, <i>m</i>)	α	46.5
β	2.91 (2H, m)	β	29.7
OCH ₂ O	5.90 (2H, s)	OCH ₂ O	100.7
		C=O	203.4
2'-OCH ₃	3.76 (3H, s)	2'-OCH ₃	55.8
4'-OCH ₃	3.82 (3H, s)	4'-OCH ₃	55.4
6'-OCH ₃	3.76 (3H, s)	6'-OCH ₃	55.8

3.5 Structure Determination of Compound 103

Compound 103 was colourless needles. The EI mass spectrum (**Figure 94**) appeared the molecular ion peak at m/z 262, consistent with $C_{17}H_{10}O_3$ as the molecular formula. The UV spectrum (**Figure 95**) showed λ_{max} at 297, 263 and 219 nm. The IR spectrum (**Figure 96**) displayed conjugated C=O stretching band at 1646 cm⁻¹. The ¹H-NMR spectrum (**Figure 97**) demonstrated that compound 103 was a furanoflavone [δ 6.90 (H-3, s); 7.78 (H-2", d, J=2.4 Hz); 7.22 (H-3", dd, J=2.4, 1.2 Hz)]. This spectrum further exhibited two multiplet signals at δ 7.58 (4H, m) assigned to H-6, H-3', H-4' and H-5', and at δ 7.97 (2H, m) belonged to H-2' and H-6'. Additionally, an aromatic singlet proton at δ 8.17 ppm was observed. This compound eventually identified as lanceolatin B [103], which had already been isolated from many plants, for instance *Pongamia glabra* (Talapatra, Mallik and Talapatra, 1980), P. pinnata (Tanaka et al., 1992) and Millettia sanagana (Mbafor, et al., 1995). Comparison of the ¹H- and ¹³C-NMR (**Figure 98**) spectra between compound 103 from this work and lanceolatin B from the literature (Tanaka et al., 1992; Mbafor, et al., 1995) has been given in **Table 22**.

Table 22 The ¹H- and ¹³C-NMR data of Compound 103 in CDCl₃

	δ_{H} (ppm), J (Hz)			δ _C (ppm)		
Н	Lanceolatin B	Compound 103	С	Lanceolatin B	Compound	
					103	
			2	162.7	162.6	
3	6.90 (1H, s)	6.90 (1H, s)	3	108.1	108.0	
		SMINA	4	178.2	178.2	
5	8.18 (1H, d, 9.0)	8.17(1H, d, 8.8)	5	121.8	121.8	
6	7.58 (1H, <i>m</i>)	7.56 (1H, m)	6	110.2	110.2	
			7	158.4	158.3	
			8	117.2	117.1	
			9	150.9	150.8	
		// back	10	119.4	119.4	
			1'	131.8	131.8	
2'	7.97 (1H, m)	7.97 (1H, m)	2'	126.2	126.2	
3'	7.58 (1H, <i>m</i>)	7.56 (1H, m)	3'	129.1	129.1	
4'	7.58(1H, <i>m</i>)	7.56 (1H, m)	4'	131.5	131.5	
5'	7.58 (1H, m)	7.56 (1H, <i>m</i>)	5'	129.1	129.1	
6'	7.97 (1H, m)	7.97 (1H, m)	6'	126.2	126.2	
2"	7.79 (1H, <i>d</i> , 2.0)	7.78(1H, <i>d</i> , 2.4)	2"	145.8	145.8	
3"	7.26 (1H, <i>d</i> , 2.0)	7.22 (1H,	3"	104.2	104.2	
		dd,2.4,1.2)				

3.6 Structure Determination of Compound 102

Compound 102 was acquired as a colourless oil. The EI mass spectrum (**Figure 99**) showed the molecular ion peak at m/z 344, analyzed for $C_{19}H_{20}O_6$. The UV spectrum (Figure 100) displayed absorptions at 295, 269 and 220 nm. The IR spectrum (Figure 101) exhibited absorption bands at 2935 (CH stretching), 1666 (C=O stretching) and 1600 and 1488 (aromatic ring) cm⁻¹. The base peak of EIMS at m/z 165 was attributed to both [2,4-(OMe)₂-C₆H₃-CO]⁺ and [3,4-OCH₂O-C₆H₃-CH₋ OMe] $^{+}$. The 1 H-NMR spectrum (**Figure 102**) presented an ABX system centred at δ 3.19, 3.48 and 4.74 ppm and a singlet (3H) at 8 3.19 ppm, typical of a methoxy substituent on the aliphatic β-carbon of dihydrochalcone. This spectrum also exhibited two aromatic ABX systems, one was located at δ 6.43 (H-3', d, J=2.4 Hz), 6.51 (H-5', dd, J=8.0,2.4 Hz) and 7.77 (H-6', d, J=8.8 Hz), and another was located at δ 6.88 (H-2, d, J=1.6 Hz), 6.77 (H-5, d, J=8.0 Hz) and 6.82 (H-6, dd, J=8.0, 1.6 Hz). Additionally, this compound also contained two methoxy groups on the A-ring and methylenedioxy group on the B-ring. Thus, this compound was assigned to dihydromilletenone methyl ether [102] by comparison of the above data with those of the literature (Mahmoud and Waterman, 1985). However, this work was the first time to report the ¹³C-NMR spectral data (**Figure 103**).

$$H_3CO$$
 A'
 CO
 $COCH_3$
 C

[102]

Table 23 The ¹H- and ¹³C-NMR data of Compound 102 in CDCl₃

	δ _H (ppm),		δ_{C} (ppm) of	
Н	Dihydromilletenone	Compound 102	C	Compound 102
	methyl ether			
			1	135.9
2	6.88 (1H, d, 2.0)	6.88 (1H, d, 1.6)	2	108.0
		s distribution	3	147.8
			4	146.9
5	6.76 (1H, d, 8.0)	6.77(1H, d, 8.0)	5	107.0
6	6.82 (1H, <i>dd</i> , 8.0,2.0)	6.82 (1H, dd, 8.0,1.6)	6	120.4
			1'	121.2
2'	7.78 (1H, d, 8.0)		2'	160.6
3'	6.51 (1H, dd, 8.0,2.0)	6.43 (1H, d, 2.4)	3'	98.3
			4'	164.4
5'	6.43 (1H, d, 2.0)	6.51 (1H, dd, 8.8,2.4)	5'	105.1
6'		7.77 (1H, d, 8.8)	6'	132.8
$\alpha_{ m eq}$	3.22 (1H,dd, 16.9, 4.9)	3.19(1H, dd,16.8,4.8)	α	52.0
α_{ax}	3.47 (1H, <i>dd</i> , 16.9, 8.0)	3.48(1H, dd,16.8,8.0)		
β_{ax}	4.73 (1H, dd, 8.0, 4.9)	4.74(1H, dd,8.0, 4.8)	β	79.5
OCH ₂ O		5.95 (2H, d, 1.2)	OCH ₂ O	100.9
			C=O	197.2
β- ОСН₃	3.18 (3H, s)	3.19 (3H, s)	OCH ₃	55.4
2'- OCH ₃		3.86 (3H, s)	OCH ₃	55.4
4'-OCH ₃	3.84 (3H, s)	3.84 (3H, s)	OCH ₃	56.5
6'-OCH ₃	3.85 (3H, s)	DAIS MELL		8

-: The bold values should be noticed.

3.7 Structure Determination of Compound 282

Compound 282 was obtained as pale yellow needles. The EI mass spectrum (Figure 104) showed the molecular ion peak at m/z 372, agreeing with $C_{20}H_{20}O_7$, which was supported by elemental analysis (Anal. Calcd for C₂₀H₂₀O₇.1/4H₂O: C, 63.74; H, 5.48. Found: C, 64.04; H, 5.48). Substitution of the ketonic A ring by a methylenedioxy group deduced by the appearance of peaks at m/z 149 [3,4-OCH₂O- $C_6H_3-CO^+$, 27 %] and 121 [3,4-OCH₂O-C₆H₃+, 20 %]. Additionally, a base peak (m/z 341) was reasonably assigned to be a benzopyrilium cation (Kiuchi, Chen and Tsuda, 1990) produced by a loss of a methoxy group as shown in Chart 1. The UV spectrum (Figure 105) displayed λ_{max} at 320, 277 and 211 nm. In the IR spectrum (Figure 106), absorption bands at 2940 (CH stretching), 1659 (C=O stretching), and 1587 and 1487 (aromatic ring) cm⁻¹ were appeared. The ¹H-NMR spectral data (Figure 107) exhibited an olefinic proton as singlet at δ 6.33 assigned as H- α , a singlet (3H) at δ 3.83 belonged to β-methoxy group confirmed by HMBC signal of methoxy proton into C-β, and other three methoxy groups substituted on the B-ring at δ 3.73 (6H, s) and 3.80 (3H, s). In DEPT-135 experiment (Figure 109), the presence of methylene carbon signal at δ 101.4 also supported that compound 282 contains only one methylenedioxy group (¹H-NMR signal of methylenedioxy protons as singlet at δ 5.98 ppm), whereas the signal at δ 90.8 belonged to both C-3 and C-5 methine carbons under symmetrical condition (¹H-NMR signal of these carbons as singlet at δ 6.10 ppm). This compound was suggested to be a new β methoxychalcone, 2,4,6,β-tetramethoxy-3',4'-methylenedioxychalcone [282]. assignment was supported by the ¹³C-NMR (**Figure 108**) resonances and 2D-NMR as HMOC (Figure 110) and HMBC (Figure 111) experiments. Furthermore, the configuration of the olefinic function was determined by NOE experiment (the selected NOE enhancements were shown in structure 283). An E-isomer is known to be the preferred isomer in naturally occurring β-methoxychalcones (Kiuchi, Chen and Tsuda, 1990). Thus, the structure **282** was completely elucidated by this observations and the data from HMBC experiment.

Chart 1 A Possible Formation of Benzopyrilium cation from β -Methoxychalcone 282 in EIMS

Table 24 The ¹H- and ¹³C-NMR data of Compound 282 in CDCl₃

	$\delta_{\rm H}$ (ppm), J (Hz) of		δ_{H} (ppm) of
Н	Compound 282	C	Compound 282
		1	107.0
		2	158.5
3	6.10 (1H, s)	3	90.8
		4	162.1
5	6.10 (1H, s)	5	90.8
		6	158.5
		1'	134.7
2'	7.32 (1H, d, 2.0)	2'	108.2
		3'	147.5
		4'	150.4
5'	6.76 (1H, d, 8.4)	5'	107.4
6'	7.46 (1H, <i>dd</i> , 8.4, 2.0)	6'	123.4
α	6.33 (1H, s)	α	101.2
-		β	165.7
OCH ₂ O	5.98 (2H, s)	OCH ₂ O	101.4
		C=O	188.4
β-ОСН ₃	3.83 (3H, s)	β-ОСН₃	55.9
2-OCH ₃	3.73 (3H, s)	2-OCH ₃	55.9
4-OCH ₃	3.80 (3H, s)	4-OCH ₃	55.2
6-OCH ₃	3.73 (3H, s)	6-OCH ₃	55.9
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3.8 Structure Determination of Compound 284

Compound **284** was obtained as white solids. The EI mass spectrum (**Figure 112**) showed the molecular ion peak at m/z 326, corresponding to the molecular formula $C_{18}H_{14}O_6$. The UV spectrum (**Figure 113**) showed absorption maxima at 339, 314, 242 and 208 nm. The IR absorption spectrum (**Figure 114**) displayed v_{max} at 1619 and 1445 cm⁻¹ belonged to the aromatic region. The ¹H- and ¹³C-NMR spectra (**Figure 115** and **116**, respectively) revealed the presence of methylenedioxy group at δ_H 6.06 (2H, s) and δ_C 101.6 ppm, two series of three coupled aromatic protons showing ABX patterns and two methoxy substituents assigned as 3-OCH₃ at δ 3.88 and 7-OCH₃ at δ 3.91. Furthermore, the EIMS gave ions at m/z 146 [3,4-OCH₂O-C₆H₃-C=CH]^{+•}, typical of a methylenedioxy substituted B-ring. These data were identical with those in the literature (Das *et al.*, 1994), thus requiring the flavone to be assigned as desmethoxykanugin [**284**]. This compound was earlier isolated from *Pongamia glabra* (Subrahmanyam, Rao and Rao, 1977) and *Gelonium multiflorum* (Das *et al.*, 1994).

Table 25 The ¹H- and ¹³C-NMR data of Compound 284 in CDCl₃

11	$\delta_{\rm H}$ (ppm), J (Hz)			δ _C (ppm)	
Н	Desmethoxykanugin	Compound 284	С	Desmethoxykanugin	Compoud 284
			2	154.7	154.6
			3	140.8	140.7
		- Militar	4	174.4	174.4
5	8.14(1H, <i>d</i> , 8.8)	8.14(1H, <i>d</i> , 9.0)	5	127.1	127.0
6	6.96(1H,dd,8.8,	6.95(1H,dd,9.0,	6	114.3	114.3
	2.4)	2.0)			
			7	156.8	156.8
8	6.89 (1H, d, 2.4)	6.89(1H, d, 2.0)	8	99.9	99.8
		///bā/\\	9	164.0	164.0
			10	118.0	118.0
		9. (8) /4	1'	124.8	124.7
2'	7.61 (1H, <i>d</i> , 1.8)	7.61(1H, <i>d</i> , 2.0)	2'	108.6	108.5
		Malalan	3'	147.9	147.8
		(5566)	4'	149.5	149.4
5'	6.94(1H, <i>d</i> , 8.2)	6.94(1H, <i>d</i> , 8.0)	5'	108.4	108.3
6'	7.69(1H, <i>dd</i> ,8.2,	7.69(1H, <i>dd</i> ,8.0,	6'	123.4	123.3
	1.8)	2.0)			
OCH ₂ O	6.06 (2H, s)	6.06 (2H, s)	OCH ₂ O	101.6	101.6
3-OCH ₃	3.88 (3H, s)	3.88 (3H, s)	3-OCH ₃	60.0	60.0
7-OCH ₃	3.91 (3H, s)	3.91 (3H, s)	7-OCH ₃	55.8	55.8

3.9 Structure Determination of Compound 68

Compound **68** was acquired as colourless needles. The EI mass spectrum (**Figure 117**) displayed the molecular ion peak at m/z 296, analysed for $C_{17}H_{12}O_5$. The UV spectrum (**Figure 118**) exhibited absorption maxima at 334, 311, 237 and 220 nm. The IR spectrum (**Figure 119**) showed v_{max} at 1638, 1502 and 1432 cm⁻¹. The ¹H-NMR spectrum (**Figure 120**) showed an aromatic singlet signal at δ 6.64 ppm for H-3, a B-ring spin system for 3',4'-substitution at δ 6.08 (2H, s), one methoxy substituent on the A-ring at δ 3.93 (3H, s), and also showed two typical of ABX systems. In EIMS, the fragment ions at m/z 146 and 134, which were the characteristic of methylenedioxy substituted B-ring, were observed. Thus, the combination of ¹H-NMR, EIMS data and comparison of ¹H-and ¹³C-NMR (**Figure 121**) data with those in the literature, fully supported that this flavone was 3',4'-methylenedioxy-7-methoxyflavone [68] that previously reported from *Millettia hemsleyana* (Mahmoud and Waterman, 1985). However, its ¹³C-NMR spectral data of this compound was reported here for the first.

Table 26 The ¹H- and ¹³C-NMR data of Compound 68 in CDCl₃

	δ _H (ppm),		δ_{C} (ppm) of	
Н	3',4'-methylenedioxy-7-	Compound 68	C	Compound 68
	methoxyflavone	(in CDCl ₃)		(in CDCl ₃)
	(in C ₅ D ₅ N)			
			2	162.7
3	7.08 (1H, s)	6.64 (1H, s)	3	108.7
		AMINA.	4	177.8
5	8.35 (1H, d, 9.0)	8.12 (1H, d, 8.8)	5	127.0
6	7.05 (1H, dd, 9.0, 2.0)	6.99 (1H, dd, 8.8,2.4)	6	114.3
			7	157.9
8	7.15 (1H, <i>d</i> , 2.0)	6.95 (1H, d, 2.4)	8	100.4
			9	164.1
		9.400.0	10	117.8
		A GI A	1'	125.9
2'	7.60 (1H, <mark>d</mark> , 2.0)	7.35 (1H, d, 2.0)	2'	106.6
	//// 8	446(9)220 A	3'	148.4
		11676367/A	4'	150.4
5'	6.99 (1H, d, 8.0)	6.93 (1H, d, 8.4)	5'	106.2
6'	7.54 (1H,dd, 8.0, 2.0)	7.48 (1H, dd, 8.4,2.0)	6'	121.2
OCH ₂ O	6.08 (2H, s)	6.08 (2H, s)	OCH ₂ O	101.9
7-OCH ₃	3.80 (3H, s)	3.90 (3H, s)	7-OCH ₃	55.8

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3.10 Structure Determination of Compound 285

Compound 285 was obtained as pale yellow needles. The EI mass spectrum (Figure 122) exhibited the molecular ion peak at m/z 342, and the analytical calculation gave C, 66.66; H, 5.30 (Found C, 66.49; H, 5.24). These data were consistent with $C_{19}H_{18}O_6$. The UV spectral data (Figure 123) displayed λ_{max} at 344, 296, 254 and 208 nm. The IR spectrum (Figure 124) showed v_{max} at 1646 (C=O stretching) and 1603 and 1489 (aromatic ring) cm⁻¹. The ¹H-NMR spectrum (**Figure**) 125) revealed the presence of a set of *trans*-olefinic protons at δ 6.78 and 7.28 (each d, J=16.0 Hz), a series of three coupled aromatic protons presenting ABX system at δ 7.05 (H-2, d, J=2.0 Hz), 6.79 (H-5, d, J=8.0 Hz) and 6.98 (H-6, dd, J=8.0, 2.0 Hz) in addition to 2',4',6'-trimethoxy substitution on the A-ring and 3,4-methylenedioxy substitution on the B-ring. The DEPT-135 (Figure 127) encouraged the ¹H-NMR and other spectral data as mentioned above by presenting one methylene carbon at δ 101.5 belonged to methylenedioxy carbon and the signal at δ 90.7 ppm assigned to both C-3' and C-5' methine carbons, whilst their ¹H-NMR signals were located at the same position of δ 6.16 (2H, s). Additionally, the ¹H-decoupling experiment (**Figure** 128) was proceeded to confirm the overlapping signals at δ 6.78-6.79 ppm (H- α and H-5) by irradiation of the signals at δ 6.78-6.79 (H- α and H-5) and 7.28 (H- β). These data showed the presence of a chalcone skeleton and a similar substitution pattern of those of dihydrochalcone 281. This compound also exhibited chemical correlation with compound 281 confirmed by the successful process of 1,4-reduction with Et₃SiH/CF₃CO₂H, in which perhydrochalcone **286** was obtained as an over-reduction by-product (Chart 2). This compound was consequently identified as 2',4',6'trimethoxy-3,4-methylenedioxychalcone [285], which was isolated from a natural source at the first time but has been patented for synthetic product without ¹H-NMR and ¹³C-NMR spectral data (Klein, 1993). The assignment of this compound was completely done by ¹³C-NMR (Figure 126), HMQC (Figure 129) and HMBC (Figure 130) experiments.

Table 27 The ¹H- and ¹³C-NMR data of Compound 285 in CDCl₃

	$\delta_{\rm H}$ (ppm), J (Hz) of		δ _C (ppm) of
Н	Compound 285	C	Compound 285
		1	129.4
2	7.05 (1H, <i>d</i> , 2.0)	2	106.7
		3	148.2
		4	149.5
5	6.79 (1H, <i>d</i> , 8.0)	5	108.4
6	6.98 (1H, <i>dd</i> , 8.0, 2.0)	6	124.7
		1'	111.9
		2'	158.7
3'	6.16 (1H, s)	3'	90.7
		4'	162.3
5'	6.16 (1H, s)	5'	90.7
		6'	158.7
α	6.78 (1H, d, 16.0)	α	127.3
β	7.28 (1H, <i>d</i> , 16.0)	β	144.0
OCH ₂ O	5.99 (2H, s)	OCH ₂ O	101.5
		C=O	194.1
2'-OCH ₃	3.77 (3H, s)	2'-OCH ₃	55.9
4'-OCH ₃	3.86 (3H, s)	4'-OCH ₃	55.4
6'-OCH ₃	3.77 (3H, s)	6'-OCH ₃	55.9

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3.11 Structure Determination of Compound 287

Compound **287** was obtained as colourless needles. The EI mass spectrum (**Figure 131**) exhibited the molecular ion peak at m/z 326, agreeing with the molecular formula $C_{18}H_{14}O_6$. This spectrum also showed the fragment ions at m/z 146 (42 %) and 134 (10 %), typical of a methylenedioxy substituted B-ring like compound **68** and **284**. The UV absorption spectrum (**Figure 132**) showed λ_{max} at 334, 265, 240 and 220 nm. The IR spectrum (**Figure 133**) showed ν_{max} at 1653 (C=O stretching) and 1616 and 1456 (aromatic ring) cm⁻¹. On the ¹H-NMR spectrum (**Figure 134**), a B-ring spin pattern for 3',4'-methylenedioxy substitution at δ 6.08 (2H, s), and one ABX system at δ 7.55 (H-2', d, J=1.8 Hz), 6.99 (H-5', d, J=7.9 Hz) and 7.52 (H-6', dd, J=7.9, 1.8 Hz) were appeared. This spectrum also exhibited the signals for meta-coupled A-ring protons at δ 6.55 (H-6, d, d=2.4 Hz) and 6.79 (H-8, d, d=2.4 Hz), and two methoxy groups at δ 3.86 (6H, d). By comparison with the NMR data in the literature (Tomazela et d1, 2000) and its similar compounds (**68** and **284**), and from supporting of the ¹³C-NMR data (**Figure 135**), this compound was assigned to 3',4'-methylenedioxy-5,7-dimethoxyflavone [**287**].

[287]

Table 28 The $^{1}\text{H-}$ and $^{13}\text{C-NMR}$ data of Compound 287 in $C_{5}D_{5}N$

	δ _H (ppm)), <i>J</i> (Hz)		$\delta_{\rm C}$ (ppm)		
Н	3',4'- Compound		С	3',4'-	Compound	
11	methylenedioxy -	287		methylenedioxy -	287	
	5,7-dimethoxy			5,7-dimethoxy		
	flavone			flavone		
			2	160.2	160.2	
3	6.92 (1H, s)	6.95 (1H, s)	3	108.7	108.7	
			4	176.6	176.7	
			5	161.3	161.3	
6	6.55 (1H, d, 2.3)	6.55 (1H, d, 2.4)	6	96.8	96.8	
			7	164.4	164.4	
8	6.78 (1H, d, 2.3)	6.79 (1H, d, 2.4)	8	93.7	93.6	
			9	160.1	160.1	
			10	109.7	109.7	
		1 9.44.000	1'	126.0	126.0	
2'	7.54 (1H, d, 1.8)	7.56 (1H, <i>d</i> , 1.8)	2'	106.6	106.5	
		WARRED TO THE	3'	148.9	148.9	
		250000000000000000000000000000000000000	4'	150.6	150.6	
5'	6.98 (1H, d, 8.1)	6.99 (1H, d, 8.0)	5'	108.9	108.9	
6'	7.51 (1H, <i>dd</i> ,8.1,	7.52 (1H, <i>dd</i> ,8.0,	6'	121.3	121.3	
	1.8)	1.8)				
OCH ₂ O	6.07 (2H, s)	6.08 (2H, s)	OCH ₂ O	101.6	101.6	
OCH_3	3.85 (3H, s)	3.86 (3H, s)	OCH ₃	55.9	55.9	
OCH ₃	3.85 (3H, s)	3.86 (3H, s)	OCH ₃	56.2	56.2	

4. Biological Activities of Compounds from *Pachyrrhizus erosus*

All isolated compounds from *Pachyrrhizus erosus* in the present work, including compounds 2, 4, 8, 11, 12, 15, 17 and 18 were subjected to cytotoxicity test against NCI-H460 cell line, while compounds 4, 8, 11, 15, 17 and 18 were also subjected to antimicrobial activity test using agar diffusion method. In evaluation of COX-2 selectively inhibitory activity using Kirtikara's method (Kirtikara et al., 2001), compounds 17 and 18 were tested at 10 μg/ml. All compounds from the above activity tests were inactive (data not shown). However, anti-HSV activity test using inactivated plaque reduction assay (Abou-Karam and Shier, 1990) for crude CHCl₃ exract of P. erosus seeds and pure compounds 4, 8, 11, 15, 17 and 18 showed that compound (+)-12a-hydroxydolineone [8] could inhibit HSV-1 activity at IC₅₀ of 25.5 μg/ml, and compound (+)-12a-hydroxypachyrrhizone [11] could inhibit both HSV-1 and HSV-2 activities at IC₅₀ of 18.0 and 18.5 µg/ml, respectively. Moreover, the investigation of anti-HSV activity were also observed cytotoxic doses to normal cell (Vero cell) of 8 at concentration >50 μg/ml and 11 at concentration 50 μg/ml. This result (Table 29) demonstrated the importance of the hydroxy substituion at C-12a to the activity.

5. Biological Activities of Compounds from Millettia leucantha

5.1 Antimicrobial Activity Test

Compounds 102, 103, 279, 280, 281, 282, 284, 285 and 287 were subjected to antimicrobial activity test using agar diffusion method. No compound quite showed this inhibition activity (data not shown).

5.2 Anti-HSV Activity Test

Nine compounds, comprising **102**, **103**, **279**, **280**, **281**, **282**, **284**, **285** and **287** were subjected to anti-HSV activity test using inactivated plaque reduction assay (Abou-Karam and Shier, 1990). Compounds **102** and **281** showed moderate activity when compared to the positive control, acyclovir (IC₅₀ at 0.06 μg/ml for HSV-1 and 0.5 μg/ml for HSV-2). Compound **102** displayed inhibition activity at IC₅₀ 17.0 μg/ml for HSV-1 and 36.3 μg/ml for HSV-2, whereas compound **281** inhibited HSV-1 activity at IC₅₀ 15.5 μg/ml and inhibited HSV-2 activity at IC₅₀ 17.0 μg/ml. The cytotoxic dose (CC₅₀) to normal cell of **102** was observed at 45.5 μg/ml, whilst this of **281** was displayed at 38.5 μg/ml. Both compounds **102** and **281** were

dihydrochalcones, suggesting that two saturated carbon bridges might be responsible for this activity. This result was deduced in **Table 30**.

5.3 COX-2 Inhibitory Activity Test

Seven compounds as compounds 103, 279, 280, 282, 284, 285 and 287 were tested for selective COX-2 inhibitory activity by means of Kirtikara's method (Kirtikara *et al.*, 2001). The only compound, desmethoxykanugin [284] exhibited significantly selective COX-2 inhibitory activity at IC₅₀ 0.96 μ M. Aspirin, indomethacin and NS-398 were employed as positive controls. This result was demonstrated in **Table 31**.

5.4 Cytotoxic Activity Test

All isolated compounds from *Millettia leucantha* except compound **115** were subjected to cytotoxicity test against human lung cancer NCI-H460 cell line. 2',4'-Dimethoxy-3,4-methylenedioxychalcone [**279**] and 2',4',6'-trimethoxy-3,4-methylene dioxychalcone [**285**] exhibited activity at IC₅₀ 7.36 and 3.69 μ g/ml, respectively. Both compounds were chalcones without any substituents on α - and β -carbons, suggesting that the *trans*-olefinic protons might be important for the activity. This result was summarized in **Table 32**.

Table 29 Inhibitory Effect of Isolates from *Pachyrrhizus erosus* Against HSV-1 and HSV-2

	Conc.	% Inhibition ^{a)}		IC ₅₀ (μg/ml) ^{b)}			Selectivity	
Tested sample	(µg/ml)					CC_{50}	Index d)	
						(μg/ml) ^{c)}	HSV	HSV
		HSV-1	HSV-2	HSV-1	HSV-2		-1	-2
Crude CHCl ₃ extract	5	84.7	46.6	-	-	-	-	-
(+)-Dolineone [4]	10	0	0	-	-	-	-	-
	50	15.0	24.4	-	-	-	-	-
(+)-12a-Hydroxy	10	31.0	29.0	-	-	-	-	-
dolineone [8]	50	82.7 ^{f)}	42.5	25.5	-	>50	-	ND
(+)-12a-Hydroxy	10	27.5	13.5	-	-	-	-	-
pachyrrhizone [11]	20	56.6 ^{f)}	56.3 ^{f)}	18.0	18.5	35	1.9	1.9
Neotenone [15]	10	0	0	-	-	-	-	-
	50	0	0	-	-	-	-	-
Pachyrrhizin [17]	10	0	0	-	-	-	-	-
	50	26.1	23.7	-	-	-	-	-
Pachyrrhizone [18]	10	0	0	-	-	-	-	-
	50	0	15.5	-	-	-	-	-
Acyclovir ^{e)}	-	<u>-</u>	Ada	0.06	0.50	-	-	-

a)-: Inactivation, Plaque reduction assay

b)-: IC₅₀ (50% Inhibitory concentration, Mean of 3 independent experiments)

c)-: CC₅₀ (50% Cytotoxic concentration, Examined by trypan blue exclusion method, Mean of 3 independent experiments)

d)-: Selectivity Index = CC_{50}/IC_{50}

e)-: Acyclovir as positive control

f)-: Compounds exhibiting more than 50% inhibition at \leq 50 μ g/ml were further determined for IC₅₀.

Table 30 Inhibitory Effect of Compounds from *Millettia leucantha* Against HSV-1 and HSV-2

Compound	Conc.	% Inhibition ^{a)}		IC ₅₀ (μg/ml) ^{b)}		CC ₅₀	Selectivity Index d)	
	(µg/ml)					(μg/ml) ^{c)}	HSV-	HSV-
		HSV-1	HSV-2	HSV-1	HSV-2		1	2
Dihydromilletenone	10	38	16					
methyl ether [102]	20	53	33					
	30	65 ^{f)}	41					
	40	72 ^{f)}	56 ^{f)}	17.0	36.3	45.5	2.7	1.3
Lanceolatin B [103]	10	0	0					
	50	0	0					
2',4'-Dimethoxy-3,4-	10	0	0					
methylenedioxy	50	0	0					
chalcone [279]								ů.
2'-Hydroxy-3,4,4',6'-	10	0	0					
tetramethoxy	50	0	0					
chalcone [280]								
2',4',6'-Trimethoxy-	10	37	33					
3,4-methylenedioxy	20	58 ^{f)}	55 D					
dihydrochalcone	30	80 f)	79 ^{f)}	15.5	17.0	38.5	2.5	2.3
[281]								
2,4,6,β-Tetramethoxy-	10	0	0					
3',4'-methylenedioxy	50	0	0					
chalcone [282]								
Desmethoxykanugin	10	0	0					
[284]	50	0	0					
2',4',6'-Trimethoxy-	10	0	0	UEIO	กร			
3,4-methylenedioxy	50	0	0	10	110			
chalcone [285]		6				0		
3',4'-Methylenedioxy-	10	0	0	291	2171	9 81		
5,7-dimethoxy	50	0	0	0 7 1				
flavone [287]								
Acyclovir ^{e)}	_	_	_	0.06	0.50			

a)-: Inactivation, Plaque reduction assay; b)-: IC_{50} (50% Inhibitory concentration, Mean of 3 independent experiments); c)-: CC_{50} (50% Cytotoxic concentration, Examined by trypan blue exclusion method, Mean of 3 independent experiments); d)-: Selectivity Index = CC_{50}/IC_{50} ;

e)-: Acyclovir as positive control; f)-: Compounds exhibiting more than 50% inhibition at \leq 50 μ g/ml were further determined for IC₅₀.

 Table 31 Selective COX-2 Inhibitory Activity of Compounds from Millettia

 leucantha

compound		bition 1g/ml ^{b)}	IC ₅₀ (μM)		
	COX-1	COX-2	COX-1	COX-2	
Lanceolatin B [103]	-	NI c)	-	-	
2',4'-Dimethoxy-3,4- methylenedioxy	-	NI c)	-	-	
chalcone [279]					
2'-Hydroxy-3,4,4',6'- tetramethoxy	11/-/	NI c)	-	-	
chalcone [280]					
2,4,6,β-Tetramethoxy- 3',4'methylenedioxy	9 -	NI c)	-	-	
chalcone [282]					
Desmethoxykanugin [284]	~50	80	-	0.96±0.003 ^{d)}	
2',4',6'-Trimethoxy-3,4-methylenedioxy	= -\\\	NI c)	-	-	
chalcone [285]	400 0 \ \ \				
3',4'-Methylenedioxy-5,7-dimethoxyflavone		NI ^{c)}	-	-	
[287]					
Aspirin	93.1	35.6	11.41±3.71	19. 80 ±11.20	
Indomethacin	1/2//-	-	0.005±0.003	0.006±0.002	
NS-398	((0)0 - (0)0)	-	NI °)	0.01±0.01	

a):- NS-398 = N-(2-[cyclohexyloxy]-4-nitrophenyl)methanesulfonamide

b):- compounds with \geq 80% inhibition were further analyzed for IC₅₀ value.

c):- NI = no inhibition

d):- mean \pm SE (n)

Table 32 Cytotoxic Activity of Compounds from *Millettia leucantha* Against NCI-H460

Compound	IC_{50} (µg/ml)
3',4'-Methylenedioxy-7-methoxyflavone [68] ^{a)}	>10
Dihydromilletenone methyl ether [102] ^{a)}	>10
Lanceolatin B [103] ^{a)}	>10
2',4'-Dimethoxy-3,4-methylenedioxychalcone [279] b)	7.36
2'-Hydroxy-3,4,4',6'- tetramethoxychalcone [280] ^{a)}	>10
2',4',6'-Trimethoxy-3,4-methylenedioxydihydrochalcone	>10
[281] ^{a)}	
2,4,6,β-Tetramethoxy-3',4'-methylenedioxychalcone [282] ^{b)}	>10
Desmethoxykanugin [284] a)	>10
2',4',6'-Trimethoxy-3,4-methylenedioxychalcone [285] ^{b)}	3.69
3',4'-Methylenedioxy-5,7-dimethoxyflavone [287] ^{a)}	>10

a)-: Dissolved in DMSO

b)-: Dissolved in EtOH, but insoluble material remains