

## REFERENCES

1. Kiatkamjornwong, S.; Traisaranapong, S.; and Prasassarakich, P. Effect of diluent and nitrogen gas flowrate on organic solvent absorption and desorption properties of styrene-divinylbenzene beads. *J.Sci.Res.Chula.Univ.*, 23 (1998), 65-77.
2. Kuroda, H.; and Osawa, Z. Synthesis and characterization of macroporous glycidyl methacrylate-divinylbenzene copolymer beads. *Eur. Polym. J.*, 31 (1995), 57-62.
3. Jang, J.; and Kim, B. Studies of crosslinked styrene-alkyl acrylate copolymers for oil absorbency application. I. synthesis and characterization. *J. Appl. Polym. Sci.* 77 (2000), 903-913.
4. Jang, J.; and Kim, B. Studies of crosslinked styrene-alkyl acrylate copolymers for oil absorbency application. II. effects of polymerization conditions on oil absorbency. *J. Appl. Polym. Sci.* 77 (2000), 914-920.
5. Zhou, M.; Kim, S.; Park, J.; Ha, C.; and Cho, W. Preparation and oil-absorptivity of crosslinked polymers containing stearylmethacrylate, 4-t-butylstyrene, and divinylbenzene. *Polymer Bulletin*. 44 (2000), 17-24.
6. Odian, G. *Principles of polymerization*. 3<sup>rd</sup> ed., Singapore: John Wiley & Sons, IPN., 1991, pp. 198-217, 452-472.
7. Collins, Edward A. *Experiments in polymer science*. 1<sup>st</sup> ed., United States of America: John Wiley & Sons, INP., 1973, pp. 8-13, 33-34.
8. Rosen, L. T. *Fundamental principles of polymeric materials*. 2<sup>nd</sup> ed., New York: John Wiley & Sons, INC., pp. 82-97.
9. Chang, R. *Physical chemistry with application to biological systems*. 2<sup>nd</sup> ed., Collier Macmillan Ltd., 1981, pp. 83-89.
10. Sperling, L.H. *Introduction to physical polymer science*. 2<sup>nd</sup> ed., New York: John Wiley & Sons, INC., 1992, pp. 65-77,428-433.
11. Rabelo, D.; and Coutinho, F.M.B. Structure and properties of styrene-divinylbenzene copolymers I. pure solvents as pore forming agents. *Polymer Bulletin*., 33 (1994), 479-486.

12. Rabelo, D.; and Coutinho, F.M.B. Structure and properties of styrene-divinylbenzene copolymers II .mixtures of different diluents with heptane as pore forming agents. *Polymer Bulletin.*, 33 (1994), 487-491.
13. Rabelo, D.; and Coutinho, F.M.B. Structure and properties of styrene-divinylbenzene copolymers III. mixtures of different diluents with alcohols as pore forming agents. *Polymer Bulletin.*, 33 (1994), 493-496.kj
14. Okubo, M.; Konishi, Y.; and Minami, H. Production of hollow polymer particles by suspension polymerizations for divinylbenzene/toluene droplets dissolving various polymers. *Colloid Polym Sci.* 278 (2000), 659-664.
15. Aykurt, M.; Kucuk, I.; and Huyulu, A. The Effect of mixed diluent on the porous structure of crosslinked PMMA beads. *Polymer Bulletin.* 44 (2000), 325-330.
16. Ishizu, K.; and Shiratori, S. Microsphere synthesis by dispersion copolymerization using poly(t-butyl methacrylate) macromonomers in nonaqueous media. *Journal of Colloid and Interface Science.* 236 (2001), 266-269.
17. Kangwansupramonkon, W.; Damronglerd, S.; and Kiatkumjornwong, S. Effect of the third acrylated vinyl comonomer on absorption and desorption properties of styrene-divinylbenzene-alkyl acrylate terpolymers, imbibing solvent on a water surface. *J. Appl. Polym. Sci.* 79 (2001), 504-516.
18. Bodugoz, H.; and Guven, O. The synthesis of nonporous poly(isobutyl methacrylate) microspheres by suspension polymerization technique and investigation of their swelling properties. *J. Appl. Polym. Sci.* 83 (2002). 349-356.
19. Kangwansupramonkon, W.; Damronglerd, S.; and Kiatkumjornwong, S. Effect of the crosslinking agent and diluents on bead properties of styrene-divinylbenzene copolymers. *J. Appl. Polym. Sci.* 79 (2001), 654-669.
20. Kiatkumjornwong, S.; Akkarakittimongkol, P.; and Omi, S. Syntheses of acrylate core/shell imbibber beads by seeded suspension copolymerization and one-stage copolymerization for solvent absorption-desorption. *J. Appl. Polym. Sci.* 85 (2002), 670-682.
21. Gawdzik, B.; and Maciejewaka, M. Influence of diluent composition on the porous structure of methacrylate copolymers. *J. polym. Sci., Part A: Polym. Chem.* 40 (2002), 3079-3085.

22. Schidknect, C.E. Polymerization processes. Vol. 29, New York: John Wiley&Sons, 1977, p 106-197.
23. Puig, J.E; Mendizabal, E. Suspension polymerization encyclopedia, Newyork, John Wiley&Sons, 1973, pp. 8215-8220.
24. Wei, J.; Bai, X.Y.; and Yan, J. Water-Swellable Hydrophobic porous copolymers based on divinylbenzene and methyl aceylate: preparation and water-swelling behavior. *Macromolecules*. 36(2003), 4960-4966.
25. Barton, J.; and Capek, J. Radical polymerizationin disperse systems. 1<sup>st</sup> ed., New York: Ellis horwood., 1994, pp. 291-318.
26. Kangwansupramonkon, W. Synthesis of solvent absorption-desorption styrenic imbiber beads by suspension polymerization, Chulalongkorn University, Thailand, 1999.
27. Grulke, E.A. Polymer handbook: solubility parameter values, New york, John Wiley&Sons, 1984, pp. VII518-VII541.
28. Okay, O. Macroporous copolymer networks. *Prog.Polym.Sci.* 25(2000), 711-779.
29. Lee, P.I; and Kim, C.J. Effect of geometry on solvent front penetration in glassy polymer. *Journal of membrane science*, 65(1992), pp.77-92.
30. Kim, B; Flamme, K.L; Peppas, N.A. Dynamic swelling behavior of pH-sensitive anionic hydrogels used for protein delivery. *J. of. Appl. Polym. Sci.* 89(2003), pp.1606-1613.
31. Masaro, L; Zhu, X.X. Physical models of diffusion for polymer solutions, gels and solids. *Prog. Polym. Sci.* 24(1999), pp.731-775.



## **APPENDICES**

ศูนย์วิทยทรัพยากร  
จุฬาลงกรณ์มหาวิทยาลัย

## APPENDIX A

### Data for calculation of diffusion exponent (*n*) and constant characteristic (*k*)

1. Poly(methyl methacrylate-co-divinylbenzene) at 0.025% divinylbenzene

Time(s)	Internal radius (mm.)	External radius (mm.)	$\ln M_t/M_\infty$	$\ln t$
12	0.44	0.52	-3.61	2.48
14	0.44	0.54	-3.33	2.64
17	0.43	0.56	-3.07	2.83
20	0.42	0.58	-2.90	3.00
23	0.41	0.61	-2.60	3.14
26	0.41	0.64	-2.44	3.26
29	0.40	0.64	-2.37	3.37
32	0.20	0.66	-2.05	3.47
35	0.40	0.69	-2.10	3.56
38	0.40	0.73	-1.91	3.64
42	0.40	0.74	-1.83	3.74
45	0.39	0.75	-1.77	3.81
48	0.38	0.77	-1.70	3.87
51	0.38	0.79	-1.59	3.93
54	0.37	0.80	-1.53	3.99
57	0.36	0.82	-1.44	4.04
61	0.35	0.84	-1.36	4.11
64	0.35	0.85	-1.31	4.16
67	0.34	0.86	-1.27	4.20
70	0.34	0.89	-1.18	4.25
73	0.34	0.89	-1.18	4.29
76	0.33	0.90	-1.12	4.33
79	0.33	0.92	-1.05	4.37
82	0.33	0.95	-0.97	4.41
85	0.32	0.96	-0.93	4.44
88	0.31	0.96	-0.92	4.48
92	0.30	0.98	-0.85	4.52
95	0.30	0.99	-0.82	4.55
98	0.30	1.01	-0.76	4.58
101	0.30	1.02	-0.73	4.62
104	0.16	1.04	-0.64	4.64
107	0.27	1.05	-0.63	4.67
110	0.24	1.05	-0.62	4.70
113	0.26	1.06	-0.59	4.73
116	0.24	1.06	-0.60	4.75
122	0.23	1.08	-0.54	4.80

## 2. Poly(methyl methacrylate-co-divinylbenzene) at 0.05 % divinylbenzene

Time,(s)	Internal radius (mm.)	External radius (mm.)	$\ln M_t / M_\infty$	$\ln t$
2	0.42	0.45	-4.24	0.69
4	0.42	0.45	-4.37	1.39
6	0.42	0.45	-4.25	1.79
8	0.44	0.47	-4.37	2.08
10	0.44	0.48	-3.85	2.30
12	0.43	0.49	-3.44	2.48
14	0.43	0.51	-3.15	2.64
16	0.43	0.51	-3.12	2.77
18	0.43	0.53	-2.86	2.89
27	0.40	0.58	-2.21	3.30
28	0.40	0.60	-2.10	3.33
30	0.40	0.60	-2.09	3.40
32	0.40	0.61	-1.97	3.47
34	0.39	0.63	-1.85	3.53
36	0.39	0.64	-1.79	3.58
38	0.38	0.65	-1.69	3.64
40	0.37	0.66	-1.64	3.69
42	0.37	0.67	-1.57	3.74
45	0.36	0.68	-1.52	3.81
47	0.35	0.70	-1.40	3.85
50	0.35	0.71	-1.34	3.91
53	0.34	0.73	-1.25	3.97
55	0.33	0.75	-1.14	4.01
58	0.33	0.76	-1.10	4.06
60	0.31	0.78	-1.00	4.09
63	0.31	0.80	-0.91	4.14
66	0.30	0.81	-0.86	4.19
69	0.29	0.82	-0.82	4.23
72	0.28	0.84	-0.76	4.28
75	0.26	0.85	-0.70	4.32
78	0.25	0.87	-0.63	4.36

## 3. Poly(methyl methacrylate-co-divinylbenzene) at 0.1% divinylbenzene

Time,(s)	Internal radius (mm.)	External radius (mm.)	$\ln M_t / M_\infty$	$\ln t$
9	0.56	0.59	-3.93	2.20
12	0.56	0.60	-3.59	2.48
16	0.55	0.61	-3.30	2.77
20	0.55	0.63	-3.04	3.00
24	0.55	0.64	-2.80	3.18
28	0.54	0.66	-2.57	3.33
32	0.54	0.67	-2.42	3.47
36	0.53	0.68	-2.32	3.58
40	0.52	0.71	-2.09	3.69
44	0.51	0.71	-2.02	3.78
48	0.51	0.72	-1.96	3.87
52	0.51	0.73	-1.84	3.95
56	0.50	0.75	-1.72	4.03
60	0.49	0.76	-1.63	4.09
66	0.48	0.78	-1.51	4.19
72	0.47	0.80	-1.40	4.28
78	0.47	0.82	-1.30	4.36
84	0.45	0.84	-1.20	4.43
90	0.44	0.86	-1.09	4.50
96	0.43	0.88	-1.03	4.56
102	0.42	0.90	-0.93	4.62
109	0.41	0.92	-0.85	4.69
115	0.40	0.94	-0.77	4.74
121	0.38	0.96	-0.69	4.80
127	0.37	0.98	-0.64	4.84
133	0.35	0.99	-0.59	4.89
140	0.33	1.01	-0.51	4.94
150	0.32	1.04	-0.43	5.01
160	0.30	1.06	-0.36	5.08

ศูนย์วิทยทรัพยากร  
จุฬาลงกรณ์มหาวิทยาลัย

4. Poly(methyl methacrylate-co-divinylbenzene) at 0.15 % divinylbenzene

Time,(s)	Internal radius (mm.)	External radius (mm.)	$\ln M_t / M_\infty$	$\ln t$
18	0.58	0.62	-3.56	2.89
22	0.58	0.62	-3.46	3.09
26	0.58	0.63	-3.27	3.26
30	0.58	0.64	-3.11	3.40
34	0.58	0.65	-2.94	3.53
38	0.57	0.65	-2.80	3.64
42	0.57	0.67	-2.57	3.74
46	0.56	0.68	-2.41	3.83
50	0.56	0.69	-2.32	3.91
54	0.56	0.70	-2.19	3.99
58	0.55	0.71	-2.10	4.06
63	0.54	0.71	-2.01	4.14
69	0.54	0.73	-1.87	4.23
80	0.52	0.75	-1.70	4.38
86	0.51	0.78	-1.47	4.45
92	0.50	0.80	-1.37	4.52
98	0.49	0.81	-1.27	4.58
104	0.48	0.84	-1.15	4.64
110	0.46	0.86	-1.03	4.70
116	0.45	0.88	-0.95	4.75
120	0.44	0.89	-0.87	4.83
130	0.41	0.92	-0.75	4.91
140	0.39	0.95	-0.65	4.98
150	0.36	0.97	-0.55	5.04

ศูนย์วิทยทรัพยากร  
จุฬาลงกรณ์มหาวิทยาลัย

## APPENDIX B

**Data for calculation of relaxation rate constant ( $K_2$ ) and  
relaxation rate constant (A)**

1. Poly[(methyl methacrylate)-co-divinylbenzene] at 0.025% divinylbenzene

time(s)	radius internal(mm.)	radius external(mm.)	$\ln\left(1 - \left(r_e^3 - r_{\infty}^3 / r_{e,\infty}^3\right)\right)$
12	0.44	0.52	-3.61
14	0.44	0.54	-3.33
17	0.43	0.56	-3.07
20	0.42	0.58	-2.90
23	0.41	0.61	-2.60
26	0.41	0.64	-2.44
29	0.40	0.64	-2.37
32	0.20	0.66	-2.05
35	0.40	0.69	-2.10
38	0.40	0.73	-1.91
42	0.40	0.74	-1.83
45	0.39	0.75	-1.77
48	0.38	0.77	-1.70
51	0.38	0.79	-1.59
54	0.37	0.80	-1.53
57	0.36	0.82	-1.44
61	0.35	0.84	-1.36
64	0.35	0.85	-1.31
67	0.34	0.86	-1.27
70	0.34	0.89	-1.18
73	0.34	0.89	-1.18
76	0.33	0.90	-1.12
79	0.33	0.92	-1.05
82	0.33	0.95	-0.97
85	0.32	0.96	-0.93
88	0.31	0.96	-0.92
92	0.30	0.98	-0.85
95	0.30	0.99	-0.82
98	0.30	1.01	-0.76
101	0.30	1.02	-0.73
104	0.16	1.04	-0.64
107	0.27	1.05	-0.63
110	0.24	1.05	-0.62
113	0.26	1.06	-0.59
116	0.24	1.06	-0.60
122	0.23	1.08	-0.54
128	0.22	1.13	-0.42
134	0.18	1.13	-0.40
140	0.19	1.14	-0.38
146	0.19	1.16	-0.32
152	0.16	1.16	-0.32

1. Poly[(methyl methacrylate)-co-divinylbenzene] at 0.025% divinylbenzene  
 (continued)

time(s)	radius internal(mm.)	radius external(mm.)	$\ln\left(1 - \left(r_e^3 - r_\infty^3 / r_{e,\infty}^3\right)\right)$
158	0.16	1.17	-0.29
164	0.15	1.19	-0.25
170	0.12	1.20	-0.22
176	0.11	1.21	-0.19
182	0.09	1.23	-0.15
188	0.09	1.23	-0.14
194	0.07	1.23	-0.14
200	0.05	1.24	-0.11
202	0.00	1.24	-0.11
204	0.00	1.24	-0.12
206	0.00	1.25	-0.09

ศูนย์วิทยบรังษยการ  
 จุฬาลงกรณ์มหาวิทยาลัย

## 2. Poly[(methyl methacrylate)-co-divinylbenzene] at 0.05% divinylbenzene

time(s)	radius internal(mm.)	radius external(mm.)	$\ln\left(1 - \left(r_e^3 - r_{\infty}^3 / r_{e,\infty}^3\right)\right)$
2	0.42	0.45	-4.24
4	0.42	0.45	-4.37
6	0.42	0.45	-4.25
8	0.44	0.47	-4.37
10	0.44	0.48	-3.85
12	0.43	0.49	-3.44
14	0.43	0.51	-3.15
16	0.43	0.51	-3.12
18	0.43	0.53	-2.86
27	0.40	0.58	-2.21
28	0.40	0.60	-2.10
30	0.40	0.60	-2.09
32	0.40	0.61	-1.97
34	0.39	0.63	-1.85
36	0.39	0.64	-1.79
38	0.38	0.65	-1.69
40	0.37	0.66	-1.64
42	0.37	0.67	-1.57
45	0.36	0.68	-1.52
47	0.35	0.70	-1.40
50	0.35	0.71	-1.34
53	0.34	0.73	-1.25
55	0.33	0.75	-1.14
58	0.33	0.76	-1.10
60	0.31	0.78	-1.00
63	0.31	0.80	-0.91
66	0.30	0.81	-0.86
69	0.29	0.82	-0.82
72	0.28	0.84	-0.76
75	0.26	0.85	-0.70
78	0.25	0.87	-0.63
82	0.24	0.88	-0.59
85	0.23	0.90	-0.53
88	0.22	0.92	-0.47
90	0.20	0.92	-0.44
93	0.20	0.93	-0.42
96	0.18	0.94	-0.38
99	0.16	0.95	-0.36
105	0.15	0.96	-0.32
113	0.12	0.98	-0.26
115	0.08	1.00	-0.18
120	0.07	1.01	-0.16
125	0.04	1.02	-0.13
130	0.00	1.03	-0.10

## 3. Poly[(methyl methacrylate)-co-divinylbenzene] at 0.1% divinylbenzene

time(s)	radius internal(mm.)	radius external(mm.)	$\ln\left(1 - \left(r_e^3 - r_\infty^3 / r_{e,\infty}^3\right)\right)$
9	0.56	0.59	-3.93
12	0.56	0.60	-3.59
16	0.55	0.61	-3.30
20	0.55	0.63	-3.04
24	0.55	0.64	-2.80
28	0.54	0.66	-2.57
32	0.54	0.67	-2.42
36	0.53	0.68	-2.32
40	0.52	0.71	-2.09
44	0.51	0.71	-2.02
48	0.51	0.72	-1.96
52	0.51	0.73	-1.84
56	0.50	0.75	-1.72
60	0.49	0.76	-1.63
66	0.48	0.78	-1.51
72	0.47	0.80	-1.40
78	0.47	0.82	-1.30
84	0.45	0.84	-1.20
90	0.44	0.86	-1.09
96	0.43	0.88	-1.03
102	0.42	0.90	-0.93
109	0.41	0.92	-0.85
115	0.40	0.94	-0.77
121	0.38	0.96	-0.69
127	0.37	0.98	-0.64
133	0.35	0.99	-0.59
140	0.33	1.01	-0.51
150	0.32	1.04	-0.43
160	0.30	1.06	-0.36
171	0.26	1.09	-0.27
180	0.25	1.10	-0.24
190	0.22	1.12	-0.19
210	0.17	1.14	-0.12
230	0.12	1.16	-0.05
250	0.03	1.17	-0.03
270	0.00	1.18	-0.01
290	0.00	1.18	0.00
300	0.00	1.18	0.00

## 4. Poly[(methyl methacrylate)-co-divinylbenzene] at 0.15% divinylbenzene

time(s)	radius internal(mm.)	radius external(mm.)	$\ln\left(1 - \left(r_e^3 - r_\infty^3 / r_{e,\infty}^3\right)\right)$
18	0.58	0.62	-3.56
22	0.58	0.62	-3.46
26	0.58	0.63	-3.27
30	0.58	0.64	-3.11
34	0.58	0.65	-2.94
38	0.57	0.65	-2.80
42	0.57	0.67	-2.57
46	0.56	0.68	-2.41
50	0.56	0.69	-2.32
54	0.56	0.70	-2.19
58	0.55	0.71	-2.10
63	0.54	0.71	-2.01
69	0.54	0.73	-1.87
80	0.52	0.75	-1.70
86	0.51	0.78	-1.47
92	0.50	0.80	-1.37
98	0.49	0.81	-1.27
104	0.48	0.84	-1.15
110	0.46	0.86	-1.03
116	0.45	0.88	-0.95
125	0.44	0.89	-0.87
135	0.41	0.92	-0.75
145	0.39	0.95	-0.65
155	0.36	0.97	-0.55
165	0.34	1.00	-0.46
175	0.32	1.02	-0.38
190	0.30	1.04	-0.33
210	0.24	1.08	-0.20
230	0.18	1.10	-0.12
250	0.11	1.12	-0.07
270	0.00	1.14	-0.03
300	0.00	1.15	0.00

จุฬาลงกรณ์มหาวิทยาลัย

## APPENDIX C

### Data for calculation of toluene absorbency

#### 1. Methyl methacrylate-divinylbenzene-alkyl (meth)acrylate terpolymer

Runs	Number	Before swelling			After swelling			Toluene absorbency	
		W-tube(g)	W-s+t(g)	W-sample(g)	W-s+t(g)	W-sample(g)	By weight	By weight	By volume
MMA-BMA	1	14.06	14.51	0.45	22.97	8.90	18.78	24.76	24.76
	2	13.43	13.94	0.51	23.86	10.43	19.33	25.45	25.45
	3	15.90	16.41	0.51	26.39	10.49	19.61	25.81	25.81
	average	14.42	14.96	0.53	23.13	8.70	19.24	25.34	25.34
MMA-LMA	1	14.46	14.97	0.50	22.59	8.12	15.36	19.25	19.25
	2	14.46	14.97	0.46	22.13	7.66	15.11	18.96	18.96
	3	14.48	14.94				15.66	19.61	19.61
	average						15.38	19.28	19.28
MMA-DA	1	17.07	17.50	0.44	25.33	8.26	17.89	22.04	22.04
	2	17.20	17.79	0.59	28.47	11.27	18.09	22.28	22.28
	3	15.86	16.31	0.46	24.49	8.63	17.87	22.02	22.02
	average						17.95	22.11	22.11
MMA-SA	1	13.03	13.54	0.51	21.81	8.79	16.11	19.84	19.84
	2	17.14	17.64	0.50	25.73	8.59	16.20	19.94	19.94
	3	16.89	17.34	0.46	24.77	7.88	16.24	19.99	19.99
	average						16.18	19.92	19.92

**VITA**

Miss Ornsiri Aungsupravate was born on November 6, 1979 in Suratthani Province, Thailand. She received her Bachelor's degree of Science in Chemistry, from the Faculty of Science, Mahidol University in 2001. She pursued Master Degree of Science in Petrochemistry and Polymer Science, the Program of Petrochemistry and Polymer Science, Graduate School, Chulalongkorn University in 2001 and finished her study in 2004.



ศูนย์วิทยทรัพยากร  
จุฬาลงกรณ์มหาวิทยาลัย