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SYNTHESIS AND CHARACTERIZATION OF ALKYL (METH)ACRYLATE-DIVINYLBENZENE
COPOLYMER BEADS BY SUSPENSION COPOLYMERIZATION

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ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย
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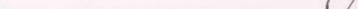
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ได้สังเคราะห์ปีดโคพอลิเมอร์เมทิลเมทาคริเลต-ไดไวนิลเบนซิน หรือบิวทิลเมทาคริเลต-ไดไวนิลเบนซินด้วยวิธีการเกิดโพลิเมอร์แบบแขวนลอย โดยมีเป็นโซ่อิลเพอร์ออกไซด์และพอลิไวนิลแอกโกรอยด์ เป็นตัวริเริ่มปฏิกิริยาแบบอนุมูลอิสระและสารแขวนลอย ตามลำดับ ได้ศึกษาตัวแปรต่างๆ ที่มีผลต่อ สัณฐานวิทยาและความสามารถในการดูดซึมน้ำกลูอินและน้ำมันเครื่องของปีดโคพอลิเมอร์ ได้แก่ อัตราส่วน ของวัฏภาคมอนомерต่อวัฏภาคน้ำ ความเข้มข้นของตัวริเริ่มปฏิกิริยา ความเข้มข้นของตัวเรื่องขาว ความเข้มข้นของสารแขวนลอย เวลาในการทำปฏิกิริยา อุณหภูมิในการทำปฏิกิริยาและองค์ประกอบของตัวเจือจาง ได้ศึกษาจนศาสตร์ของตัวทำละลายเข้าสู่ปีดโคพอลิเมอร์เมทิลเมทาคริเลต-ไดไวนิลเบนซิน เพื่อหาชนิดของกาวพรี่

กลไกการเคลื่อนที่ของโกลุอีนเข้าสู่บีดโคพอลิเมอร์เมทิลเมทาคริเลต-ไดไวนิลเบนซีนเป็นแบบวิธีนอนพิกเกียนชนิดที่ 2 ซึ่งเป็นกระบวนการผ่อนคลายของไขมีเลกุลของพอลิเมอร์ ไดเพิ่มคอมอนอเมอร์ชนิดที่สาม ไดแก่ บิวทิลเมทาคริเลต จอริลเมทาคริเลต สเตียริลอะคริเลต และไดเดซิลอะคริเลต เพื่อปรับปรุงความสามารถในการดูดซึมโกลุอีนและน้ำมันเครื่องของบีดโคพอลิเมอร์เมทิลเมทาคริเลต-ไดไวนิลเบนซีน อัตราส่วนการบวมตัวในโกลุอีนของบีดโคพอลิเมอร์เมทิลเมทาคริเลต-ไดไวนิลเบนซีน-บิวทิลเมทาคริเลตอยู่ในพิสัย 25 เท่าโดยประมาณ และผลกระทบของคอมอนอเมอร์ชนิดที่สามต่อความสามารถในการดูดซึมน้ำมันเครื่องของบีดโคพอลิเมอร์เมทิลเมทาคริเลต-ไดไวนิลเบนซีน พบร่วมความสามารถในการดูดซึมน้ำมันเครื่องเพิ่มขึ้นเมื่อความไม่มีข้อของคอมอนอเมอร์ชนิดที่สามเพิ่มขึ้น

ภาควิชา-

สาขาวิชาปีตรเคมีและ
ปีการศึกษา 2547

ลายมือชื่อนิสิต..... รองคัมภีร์ อ้วนประจุก
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ORNSIRI AUNGSUPRAVATE: SYNTHESIS AND CHARACTERIZATION OF ALKYL (METH)ACRYLATE-DIVINYLBENZENE COPOLYMER BEADS BY SUSPENSION COPOLYMERIZATION. THESIS ADVISOR: PROF. SUDA KIATKUMJORNWONG, Ph.D., THESIS COADVISOR: ASST. PROF. WARINTHORN CHAVASIRI, Ph.D., 178 pp. ISBN 974-17-5928-2.

Methyl methacrylate-divinylbenzene or butyl methacrylate-divinylbenzene copolymer beads were synthesized by suspension polymerization. The reactions were performed in the presence of benzoyl peroxide and poly(vinyl alcohol) as a radical initiator and suspending agent, respectively. The monomer phase weight fraction, initiator concentration, crosslinking agent concentration, suspending agent concentration, reaction time, reaction temperature and diluent composition were studied for the morphology of the resultant copolymer beads and the capacity of toluene and lube oil uptake. The kinetic of toluene uptake of methyl methacrylate-divinylbenzene copolymer bead was studied to validate the diffusion type.

The transport mechanism of toluene into the spherical bead copolymer of methyl methacrylate-divinylbenzene was non-Fickian case II type which was a relaxation control of the polymer molecules. The addition of a third monomer of alkyl (meth)acrylates, namely, butyl methacrylate, lauryl methacrylate, stearyl acrylate and dodecyl acrylate was used to improve the toluene and oil absorbency of the terpolymer beads. The highest capacity of toluene absorption by methyl methacrylate-divinylbenzene-butyl methacrylate terpolymer was 25 times of their original volume. The effect of the third-copolymer on oil absorbency of methyl methacrylate-divinylbenzene copolymer indicated that the oil capacity of terpolymer beads increased with the increasing hydrophobicity of the third-comonomer.

Field of Study Petrochemistry and
Polymer Science
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