

REFERENCE

1. Funk, C.E. New College Standard Dictionary. USA.: Funk&Wagnalls Company, 1956.
2. Lim, C.K., ed. HPLC of small molecules a practical approach. Oxford: IRC Press Limited, 1986. pp. 9.
3. Dean, J.A. Chemical separation method. New York: D.Van Nostrand Company, 1969. pp. 33-34.
4. Pinta, M. Modern methods for trace element analysis. 2nd. ed. Michigan: Ann Arbor Science Publishers, Inc., 1978. pp. 212.
5. International Standard ISO. Water quality - determination of cobalt, nickel, copper, zinc, cadmium and lead - flame atomic absorption spectrometric methods. ISO 8288-1986 (E) (1986): 5-7.
6. Hurst, W.J. Bonded solid-phase extraction columes for the sample preparation of food materials. LC-GC 6 (March 1988): 216-218.
7. Majors, R.E. An overview of sample preparation. LC-GC Internaional 4 (February 1991): 10-14.
8. _____. New chromatography columns and accessories at the 1991 Pittsburgh Conference, PartII LC-GC 9 (April 1991): 256-268.
9. _____. Sample preparation for HPLC and gas chromatography using solid phase extraction. LC-GC 4 (October 1986): 972-984.
10. Tippins, B. Selective sample preparation of endogeneous biological compounds using solid-phase extraction. Am. Lab. 19 (February 1987): 107-114.

11. Tippins, B. Solid phase extraction fundamentals. Nature 334 (July 1988): 273-274.
12. Tippins, B.L. Solid phase extraction: Sample preparation for HPLC, GC, NMR, FTIR, and MS. Am. Lab. News. 19 (June 1987): 8.
13. Van Horne, K.C., ed. Sorbent extraction technology. Harbor City: Analytichem International, Inc., 1985.
14. Yago, L.S. Bonded phase sample preparation technology: A growing trend. Am. Lab. 16 (January 1984): 4-6.
15. Zief, M., and Kiser, R. An overview of solid phase extraction for sample preparation. Am. Lab. 22 (January 1990): 70-83.
16. _____. Solid phase extraction for sample preparation. Phillipsburg: J.T. Baker Inc., 1988.
17. Bidligneyer, B.A. Guidelines for proper usage of solid-phase extraction devices. LC Mag. 2 (August 1984): 578-580.
18. Baker-10 SPE application guide. Vol. I. Phillipsburg: J.T. Baker Chemical Co., 1982.
19. Baker-10 SPE application guide. Vol. II. Phillipsburg: J.T. Baker Chemical Co., 1984.
20. Extract-Clean SPE sample preparation guide. Vol. 1. bulletin no 83. Illinois: Alltech Associates, Inc.
21. Lopez-Avila, V., Milanes, J., Dodhiwala, N.S., and Beckert, W.F. Cleanup of environmental sample extracts using florisil solid phase extraction cartridges. J.Chromatogr.Sci. 27 (May 1989): 209-215.
22. Werkhoven-Goewie, C.E., Brinkman, U.A. Th., and Frel, R.W. Trace enrichment of polar compounds on chemically bonded and carbonaceous sorbents and application to chlorophenols. Anal. Chem. 53 (1981): 2072-2080.

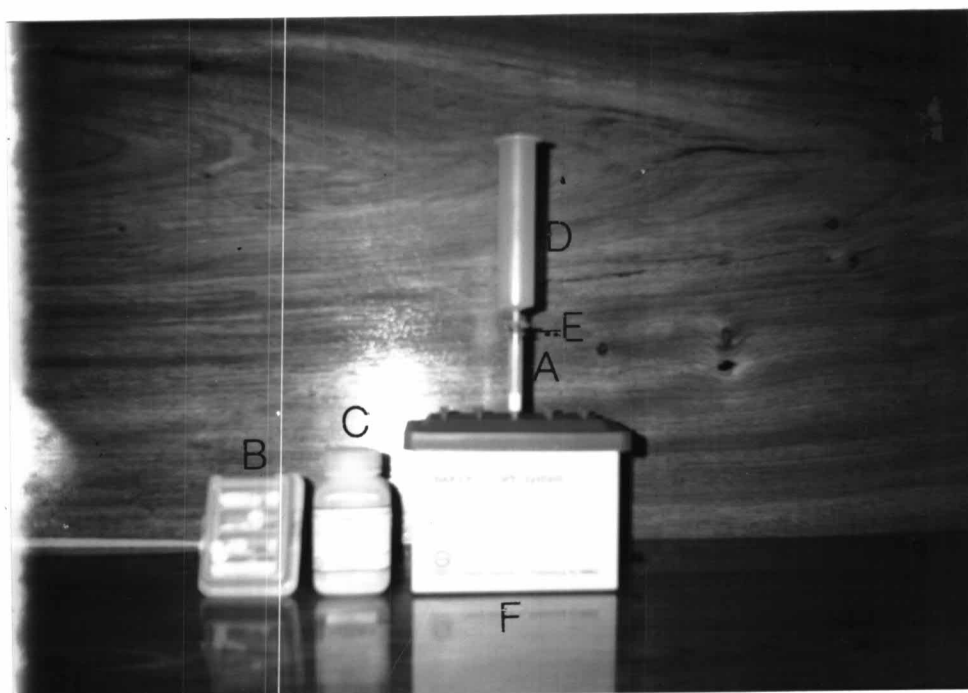
23. Ozretich, R.J., and Schroeder, W.P. Determination of selected neutral priority organic pollutants in marine sediment, tissue, and reference materials utilizing bonded-phase sorbents. Anal. Chem. 58 (1986): 2041-2048.
24. Wells, M.J.M., and Michael, J.E. Reversed-phase solid-phase extraction for aqueous environmental sample preparation in herbicide residue analysis. J. Chromatogr. Sci. 25 (August 1987): 345-350.
25. Brooks, M.W., Tessier, D., Soderstrom, D., Jenkins, J., and Clark, J.M. A rapid method for the simultaneous analysis of chlopyrifos, isofenphos, carbaryl, iprodione, and triadimefon in groundwater by solid phase extraction. J. Chromatogr. Sci. 28 (september 1990): 487-489.
26. Vinuesa, J.M., Cortes, J.C.M., Canas, C.I., and Perez, G.F. Isolation and concentration of organophosphorus pesticides from water using a C₁₈ reversed phase. J. Chromatogr. 472 (1989): 365-370.
27. Johnson, W.E., Fendinger, N.J., and Plimmer, J.R. Solid-phase extraction of pesticides from water: Possible interferences from dissolved organic material. Anal. Chem. 63 (1991): 1510-1513.
28. Marble, L.K., and Delfino, J.J. Extraction and solid phase cleanup methods for pesticides in sediment and fish. Am.Lab. 20 (November 1988): 23-32.
29. Loconto, P.R. Solid phase extraction in trace environmental analysis, Part I-current research. LC-GC International 4 (September 1991): 10-15.

30. Loconto, P.R. Solid phase extraction in trace environmental analysis, current research-part II LC-GC 9 (November 1991): 752-760.
31. Sherma, J., Dryer, J., and Bouvard, J.J. Determination of phthalate esters in water by solid phase extraction and quantitative HPTLC. Am. Lab. 18 (October 1986): 28-32.
32. Junk, G.A., and Richard, J.J. Organics in water: Solid phase extraction on small scale. Anal. Chem. 60 (1988): 451-454.
33. Rostad, C.E., Pereira, W.E., and Ratcliff, S.M. Bonded-phase extraction column isolation of organic compounds in groundwater at a hazardous waste site. Anal. Chem. 56 (1984): 2856-2860.
34. Junk, G.A., Avery, M.T., and Richard, J.J. Interferences in solid phase extraction using C-18 bonded silica cartridges. Anal. Chem. 60 (1988): 1347-1350.
35. Marko, V., Soltes, L., and Radova, K. Polar interactions in solid-phase extraction of basic drugs by octadecyl-silaneized silica. J. Chromatogr. Sci. 28 (August 1990): 403-406.
36. Patel, R.M., Benson, J.R., Hometchko, D., and Jagodzinski, J.J. Mixed-mode sorbent for sample preparation. LC-GC International 3 (November 1990): 49-53.
37. Patel, R.M., Benson, J.R., Hometchko, D., and Marshall, G. Polymeric solid phase extraction of organic acids. Am. Lab. 22 (February 1990): 92-99.
38. Bagchi, R., and Haddad, P.R. Contamination sources in the clean-up of samples for inorganic ion analysis. J. Chromatogr. 351 (1986): 541-547.

39. Saari-Nordhaus, R., Anderson, J.M., Jr., and Henderson, I.K.
Sample preparation for ion chromatography using solid phase extraction. Am. Lab. 22 (August 1990): 18-26.
40. Donat, J.R., Statham, P.J., and Bruland, K.W. An evaluation of a C-18 solid phase extraction technique for isolation metal-organic complexes from central north pacific ocean waters. Marine Chemistry 18 (1986): 85-99.
41. Majors, R.E. New devices and instrumentation for sample preparation in chromatography. LC-GC International 2 (February 1989): 12-16.
42. Yago, L. Automated sample preparation using sorbent extraction. Am. Lab. 17 (October 1985): 118-127.
43. Majors, R.E., Dimson, P., and Brocato, S. Automating solid phase extraction for HPLC sample preparation. Am. Lab. 18 (October 1986): 82-94.
44. Markell, C., Hagen, D.F., and Bunnelle, V.A. New technologies in solid phase extraction. LC-GC 9 (May 1991): 332-337.
45. Glajch, J.L., and Kirkland, J.J. Stable, sterically protected, monofunctional-silane bond-phase columns for high performance liquid chromatography. LC-GC International 3 (April 1990): 50-56.

APPENDIX A

The materials used in this SPE technique



A: Disposable colume 4 mL

B: Glassfibre prefilter, diameter 10 mm

C: Octadecyl (C₁₈) bulk packing

D: Reservoir 75 mL

E: Adapter

F: Baker-10 Extraction System

APPENDIX B

The calculated concentration of each metal

For the study of accuracy,

1. 75.0 mL of the synthetic unknown has been pipetted.
2. 100.0 mL of extracted solution has been passed through a cartridge and eluted with 5.0 mL 1.0 M HNO₃.
3. 5.0 mL of eluent was aspirated to a flame AAS. By using the calibration curve of each metal, the concentration of each metal can be calculated.

The concentration for each metal can be calculated according to this simple equation:

$$\text{Conc. of metal in water} = \frac{\text{Conc. (from calibration curve)} \times \text{final volume}}{\text{pipetted volume}}$$

For the study of accuracy and real water samples, the concentration of each metal from calibration curve (concentration after extraction) can be put in this equation. The concentration of each metal in water can be calculated.

$$\text{Conc.} = \frac{\text{Conc. (from calibration curve)} \times 5.0 \text{ mL}}{75.0 \text{ mL}}$$

APPENDIX C

The calculated concentration by concerning with percent recovery

The percentage of recoveries for each of metals which has been calculated in this study are shown as follows:

For Ag 81.05 %, Cd 89.41 %, Cu 100.86 %, Ni 98.62 %, and Pb 83.57 %.

Example, the calculated concentration of Ag for the study of accuracy (direct calculation by using a calibration curve) is 0.1109 ppm. The concentration concerned with percent recovery can be calculated by using this equation:

$$\text{Conc.} = \frac{\text{Conc. (direct cal.)} \times 100}{\text{percent recovery}}$$

$$\text{Conc.} = \frac{0.1109 \text{ ppm} \times 100 \%}{81.05 \%} = 0.1368 \text{ ppm}$$

The other metals can be calculated by using above equation.

VITA

Miss Chandladda Chotratanadilok was born on October 25, 1965 in Bangkok. She received her Bachelor Degree of Science (Chemistry) from Faculty of Science, Chulalongkorn University in 1988. Since 1989, She has been a graduate student studying Analytical Chemistry in Chulalongkorn University. She was supported by Professor Dr. Buares Kamthong scholarships in 1990 to 1991, teacher assistantship by Graduate School in 1991, The University Development Commission scholarship in 1991 to 1992 and received a research grant for her Master degree's Thesis from the Graduate School of Chulalongkorn University in 1992.

