

REFERENCES

1. Lowe, N. J.; Shaath, N. A.; Pathak, M. A. *Sunscreen Development, Evaluation, and Regulatory Aspects*, 2nd ed, New York: Marcel Dekker, 1997, pp 83-85, 268-280.
2. Plastow, S. R.; Harrison, J. A.; Young, J. A. "Early Changes in Dermal Collagen of Mice Exposed to Chronic UVB Irradiation and the Effects of a UVB Sunscreen" *J. Invest. Dermatol.* **1988**, *91*, 590-592.
3. Stickland, K. K.; Krol, E. S.; Liebler, D. C. "UVB Induced Photooxidation of Vitamin E in Mouse Skin" *Chem. Res. Toxicol.* **1999**, *12*, 187-191.
4. Shue, C.; Kang, P.; Khan, S.; Foote, C. S. "Low-Temperature Photosensitized Oxidation of Guanosine Derivative and Formation of an Imidazole Ring-Opened Products" *J. Am. Chem. Soc.* **2002**, *124*, 3905-3913.
5. Pouget, J.-P.; Doukl, T.; Richard, M.-J.; Cadet, J. "DNA Damage Induced in Cells by γ and UVA Radiation as Measured by HPLC/GC-MS and HPLC-EC and Comet Assay" *Chem. Res. Toxicol.* **2000**, *13*, 541-549.
6. Marrot, L.; Belaidi, J. P.; Meunier, J. R.; Perez, P.; Causse, C. A. "The Human Melanocyte as a Particular Target for UVA Radiation and an Endpoint for Photoprotection Assessment" *Photochem. Photobiol.* **1999**, *69*, 686-693.
7. Masaki, H.; Okano, Y.; Sakurai, H. "Generation of Active Oxygen Species from Advanced Glycation End-product (AGEs) during Ultraviolet Light A (UVA) Irradiation and a Possible Mechanism for Cell Damaging" *Biochim. Biophys. Acta* **1999**, *1428*, 45-56.
8. Gasparro, F. P. *Sunscreen Photobiology: Molecular, Cellular and Physiological Aspects*, Berlin, Springer-Verlag, 1997.
9. Hoffmann, M. R.; Martin, S. T.; Choi, W.; Bahnemann, D. "Environmental applications of Semiconductor Photocatalysis" *Chem. Rev.* **1995**, *95*, 69-96.
10. Umbach, W. *Cosmetics and Toiletries Development, Production and Use*, West Sussex, Ellis Horwood, 1991.

11. Agrapidis-Paloympis, L. E.; Nash, R. A. "The Effect of Solvents on the Ultraviolet Absorbance of Sunscreens" *J. Soc. Cosmet. Chem.* **1987**, *38*, 209-221.
12. David, M. R.; Quigley, M. N. "Liquid Chromatography Determination of UV Absorbers in Sunscreen" *J. Chem. Educ.* **1995**, *72*, 279-281.
13. Thompsom, S. C.; Jolley, D.; Marks, R. "Reduction of Solar Keratoses by Regular Sunscreen Use" *N. Engl. J. Med.*, **1993**, *329*, 1147-1151.
14. Bissonauth, V.; Drouin, R.; Mitchell, D. L.; Rhainds, M.; Claveau, J.; Rouabhia, M. "The Efficacy of a Broad-Spectrum Sunscreen to Protect Engineered Human Skin from Tissue and DNA Damage Induced by Solar Ultraviolet Exposure" *Clin. Cancer Res.* **2000**, *6*, 4128-4135.
15. Shaath, N. A.; Fares, H. M.; Klein, K. "Photodegradation of Sunscreen Chemicals" *Cosm. & Toil.* **1990**, *105*, 41-44.
16. Meijer, J.; Loden, M. "Stability Analysis of Three UV-Filters Using HPLC" *J. liq. Chromat.* **1995**, *18*, 1821-1832.
17. Ishigami, T.; Nakazato, K.; Uehara, M.; Endo, T. "Marked Dependence of Multiplicity in Direct Z,E Photoisomerization of a Series of Methyl Cinnamates on their Para-Substituents" *Tetrahedron Lett.* **1979**, 863-866.
18. Tarras-Wahlberg, N.; Stenhagen, G.; Larko, O.; Rosen, A.; Wennberg, A. M.; Wennerstrom, O. "Changes in Ultraviolet Absorption of Sunscreens After Ultraviolet Irradiation" *J. Invest. Dermatol.* **1999**, *113*, 547-553.
19. Pattanaargson, S.; Limphong, P. "Stability of Octyl methoxycinnamate and Identification of Its Photo-Degradation Product" *Int. J. Cosmet. Sci.* **2001**, *23*, 1-7.
20. Morliere, P.; Avice, O.; Sa E Melo, T.; Dubertret, L.; Giraud, M.; Santus, R. "A Study of The Photochemical Properties of Some Cinnamate Sunscreens by Steady State and Laser Flash Photolysis" *Photochem. Photobiol.* **1982**, *36*, 395-399.
21. Marti-Mestres, G.; Fernandez, C.; Parsotam, N.; Nielloud, F.; Mesters, J. P.; Maillols, H. "Stability of UV Filters in Different Vehicles: Solvents and Emulsions" *Drug Dev. Ind. Pharm.* **1997**, *23*, 647-655.

22. Marti-Mestres, G.; Nielloud, F.; Fortune, R.; Fernandez, C.; Maillols, H. "Effect of the Addition of Oxybenzone or Octyl-Methoxycinnamate on Particle Size of Submicron Emulsion" *Drug Dev. Ind. Pharm.* **2000**, *26*, 349-356.
23. Schallreuter, K. U.; Wood, J. M.; Farwell, D. W.; Moore, J.; Edwards, H. G. M. "Oxybenzone Oxidation Following Solar Irradiation of Skin: Photoprotection versus Antioxidant Inactivation" *J. Invest. Dermatol.*, **1996**, *106*, 583-586.
24. Tominaga, N. "Cosmetic Composition": U.S. Pat. 6,077,520 Jun. 20, 2000.
25. Deflandre, A.; Forestier, S.; Lang, G.; Richard, H.; Leduc, M. "Photostable Cosmetic Filter Composition Containing A UV-A Filter and A Substituted Dialkylbenzalmalonate, The Use of Substituted Dialkylbenzalmalonates in Cosmetics as Brod-Band Solar Filters and Novel Substituted Dialkyl malonate": U.S. Pat. 5,624,663 Apr. 29, 1997.
26. Koo, J.; Fish, M. S.; Walker, G. N.; Blake, J. "2,3-Dimethoxycinnamic Acid" *Org. Syn. Coll.*, **1944**, *4*, 327-328.
27. Dictionary of Organic Compounds, 4th ed., Eyre & Spottiswoode, Publisher Ltd. E. & F. N. SPON.
28. Aldrich; *Catalog Handbook of Fine Chemicals*, 1994-1995.
29. Harwood, L. M.; Moody, C. J. *Experimental Organic Chemistry, Principle and Practice*; Oxford: Scientific Publications, 1989, 443-444.
30. Womack, E. B.; McWhirter, J. "Phenyl Cinnamate" *Org. Syn. Coll.*, **1943**, *3*, 714-715.
31. Allen, C. F. H.; Spangler, F. W. "Ethyl Benzalmalonate" *Org. Syn. Coll.*, **1943**, *3*, 377-380.
32. Skoog, D. A.; West, D. M.; Holler, F. J. *Fundamentals of Analytical Chemistry*; 7th ed., New York, Saunders College Publishing, 1997, 510-511.
33. Bourke, J.; Coulson, I.; English, J. "Guidelines for Care of Contact Dermatitis" *Br. J. Dermatol.*, **2001**, *145*, 877-885.
34. Silverstein, R. M.; Bassler, G. C.; Morrill, T. C. *Spectrometric Identification of Organic Compounds*, 4th ed, New York: John Wiley & Sons, 1981.

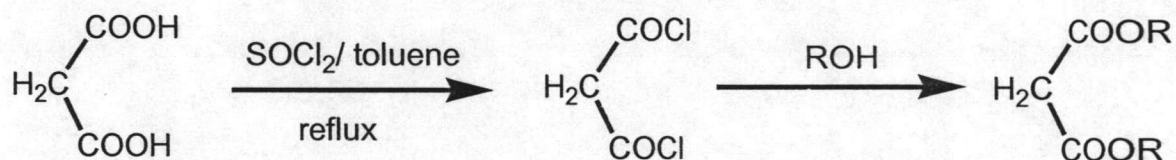
APPENDICES

APPENDIX A

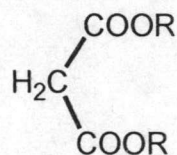
A. Synthesis of malonate Esters

General Procedure²⁴

A mixture of 0.01 mol of malonic acid and 0.02 mol of thionyl chloride were refluxed in toluene at 110 °C. To the top of the condenser is attached an exit tube leading to a gas-absorption trap (K₂CO₃). A mixture was heated until no further evolution of hydrogen chloride was observed (45-60 minutes). An aspirator vacuum was then applied to the top of the condenser to remove any remaining thionyl chloride before 0.01 mol of alcohol was added. The mixture was again refluxed on water bath at 80 °C for 1 hour. It was then placed on a sand bath and refluxed at 110 °C in order to complete the reaction (1.5 hours, checked by TLC). The reaction mixture was cooled and washed with 100 ml of 2% sodium bicarbonate solution. The general equation is shown below:



Two malonate esters were synthesized and their structures are displayed in Fig A.1.



Cpds	Substance	R
M1	Di-(2-ethylhexyl)-malonate	C ₈ H ₁₇ (iso)
M2	Dihexyl malonate	C ₆ H ₁₃

Figure A.1 Structure of the synthesized malonate esters

Malonate esters

Di-(2-ethylhexyl)-malonate (M1): pale yellow oil (58%), IR (neat, cm^{-1}): 2959, 2865, 1748, 1465, 1382, 1324, 1269, 1145 and 1013; $^1\text{H-NMR}$ (CDCl_3) δ (ppm): 4.02-3.99 (d, $J=5.8$ Hz, 4H), 3.30 (s, 2H) and 1.59-0.79 (m, 30H); $^{13}\text{C-NMR}$ (CDCl_3) δ (ppm): 166.4 (COOR), 67.5, 41.4, 38.5, 30.1, 28.7, 23.4, 22.7, 13.7 and 10.7 (alkyl carbons).

Dihexylmalonate (M2): pale yellow oil (71%), IR (neat, cm^{-1}): 2931, 2865, 1743, 1465, 1382, 1324, 1273, 1145 and 1005; $^1\text{H-NMR}$ (CDCl_3) δ (ppm): 4.13-4.04 (t, 4H), 3.32 (s, 2H) and 1.66-0.77 (m, 22H); $^{13}\text{C-NMR}$ (CDCl_3) δ (ppm): 166.3 (COOR), 65.2, 41.3, 31.1, 28.2, 25.2, 22.2 and 13.6 (alkyl carbons).

APPENDIX B

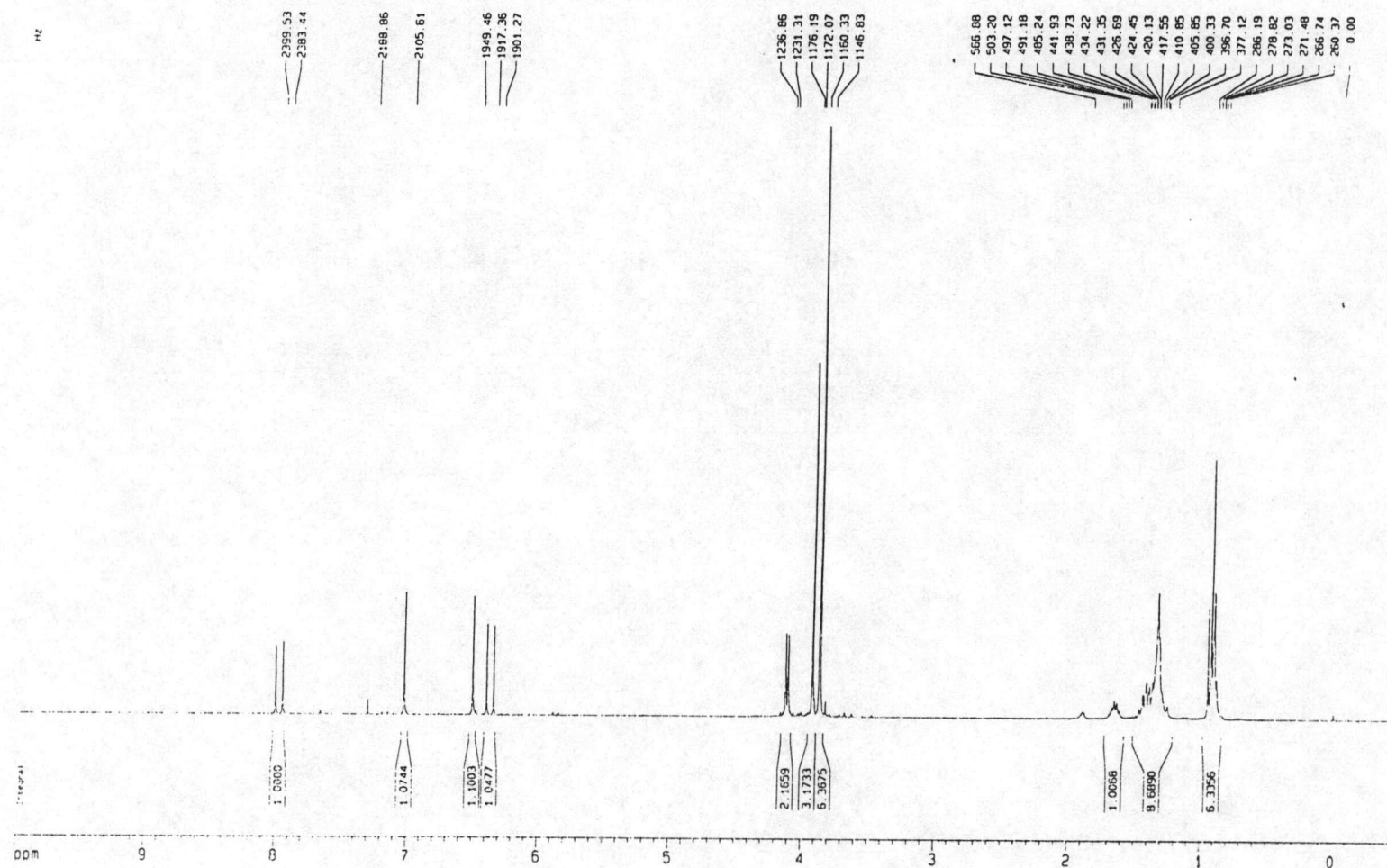


Figure B.1 ¹H-NMR spectrum of 2-ethylhexyl-2,4,5-trimethoxycinnamate (E8)

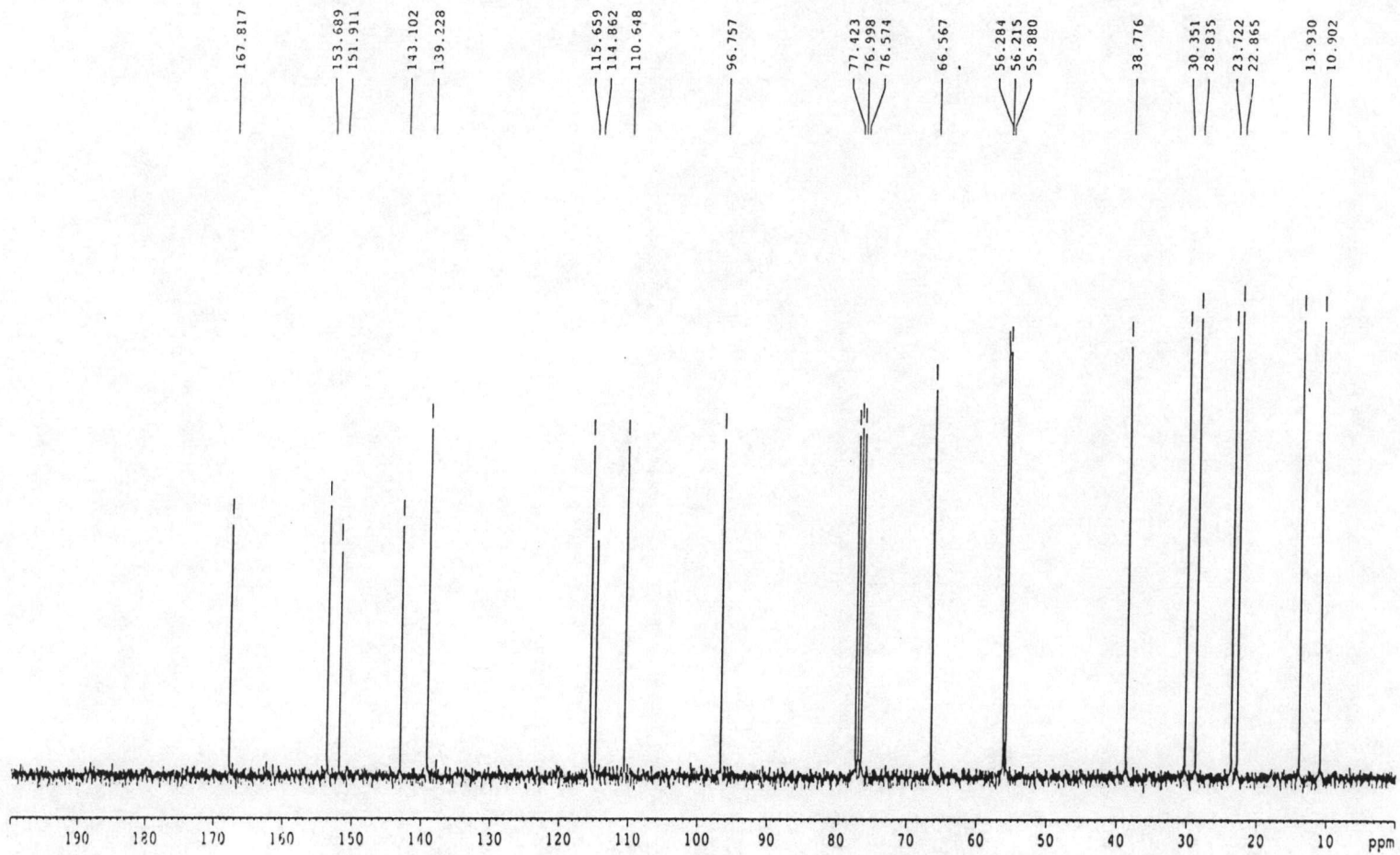


Figure B.2 ^{13}C -NMR spectrum of 2-ethylhexyl-2,4,5-trimethoxycinnamate (**E8**)

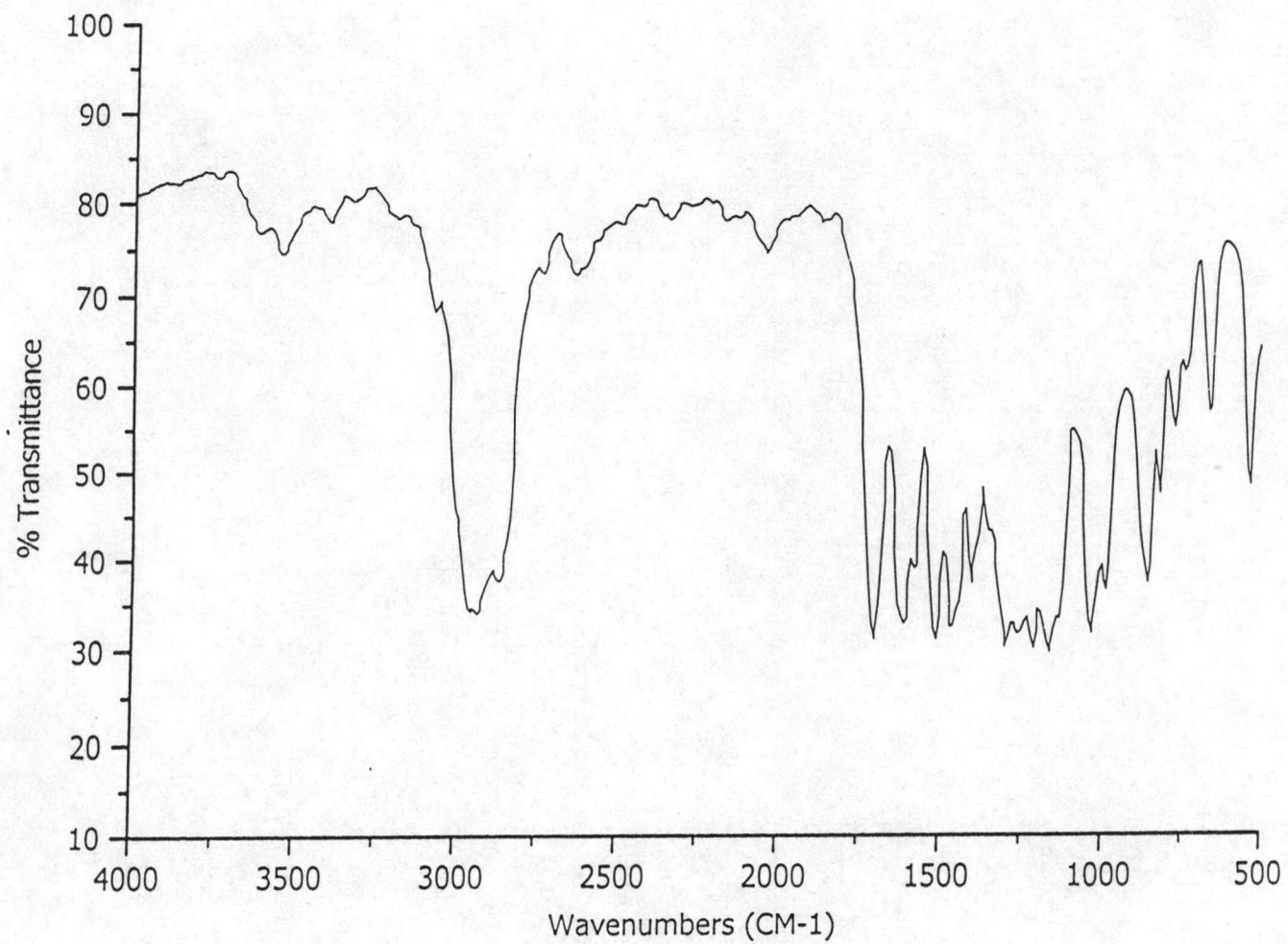


Figure B.3 IR spectrum of 2-ethylhexyl-2,4,5-trimethoxycinnamate (**E8**)

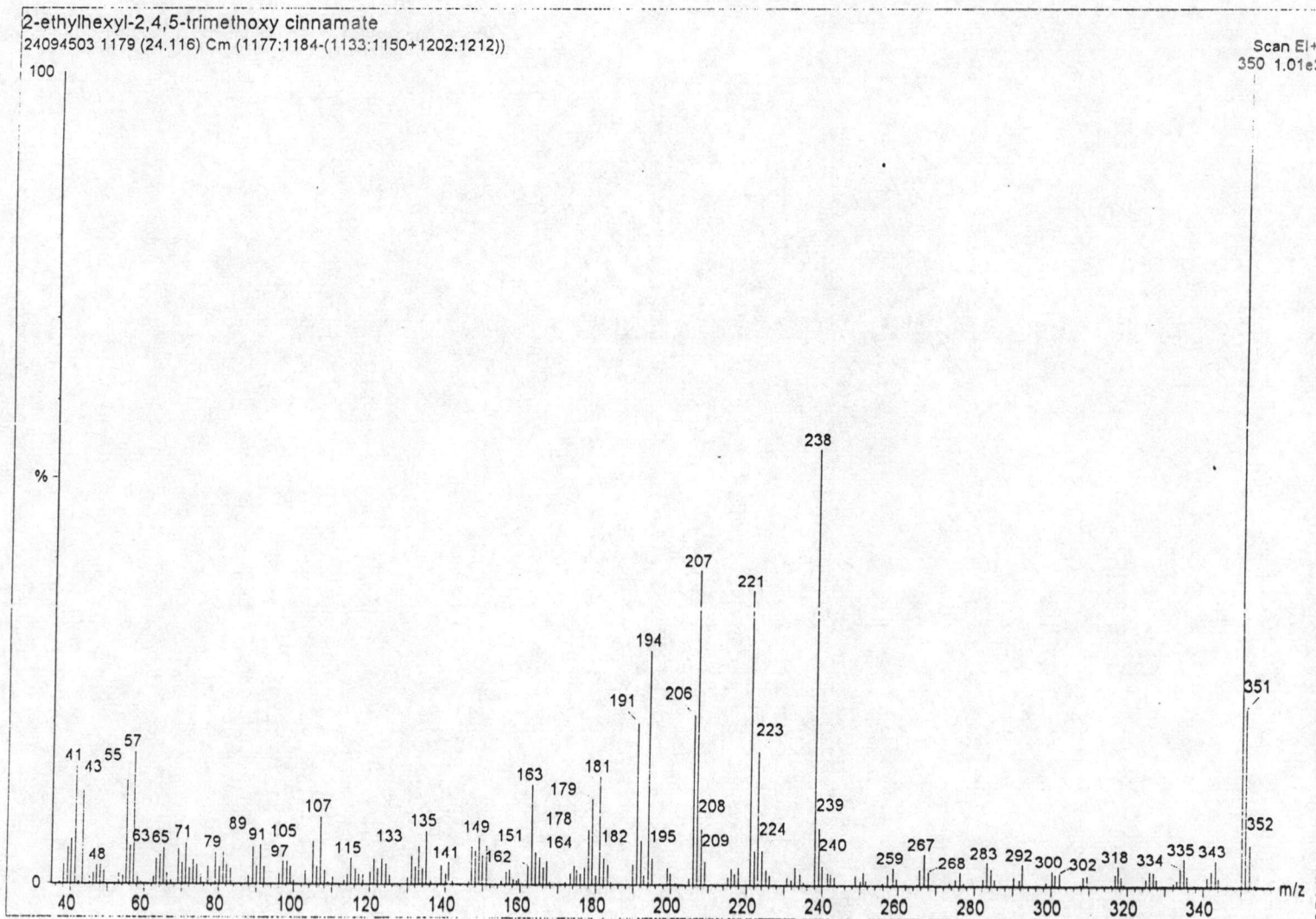


Figure B.4 Mass spectrum of 2-ethylhexyl-2,4,5-trimethoxycinnamate (E8)

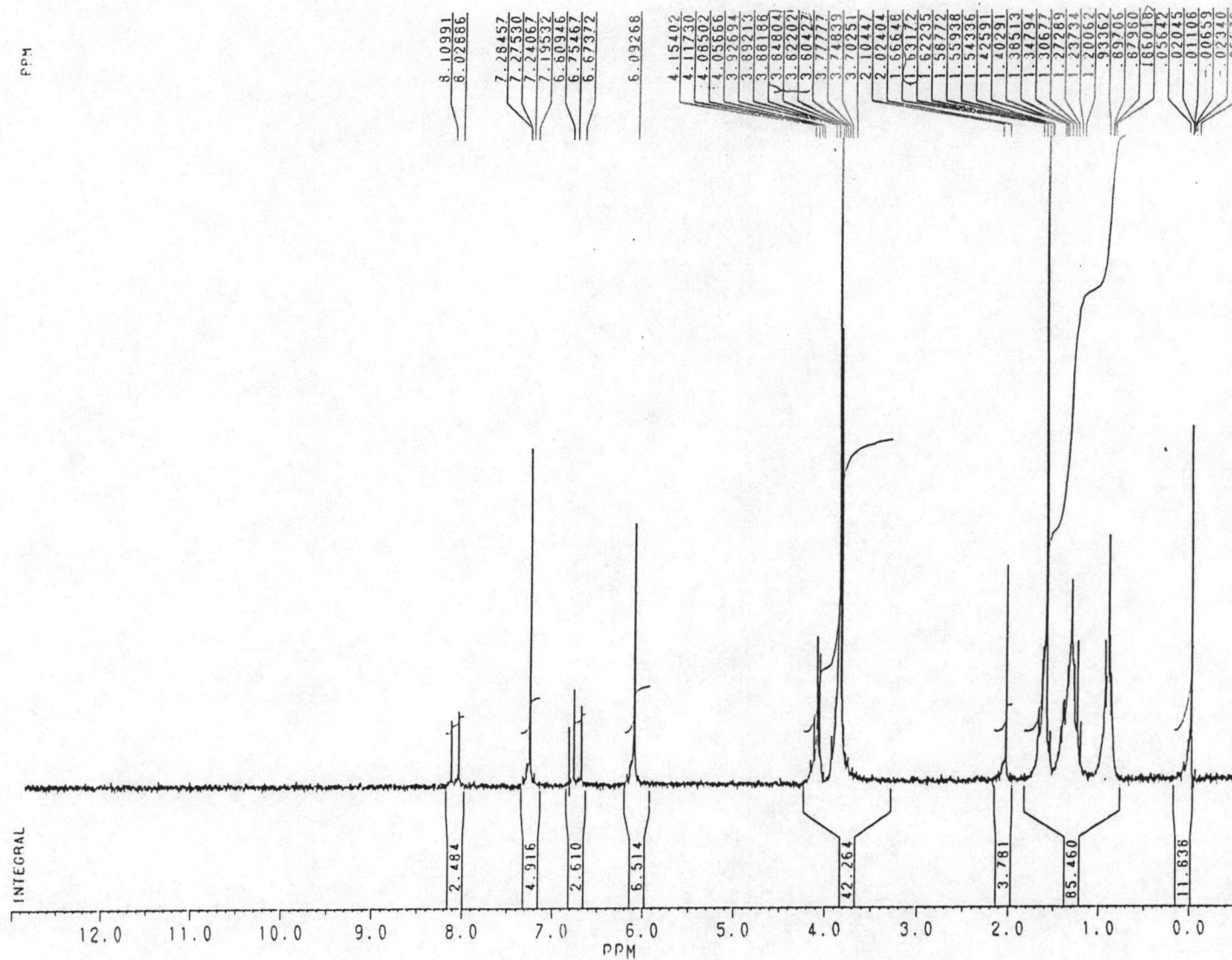


Figure B.5 ¹H-NMR spectrum of 2-ethylhexyl-2,4,6-trimethoxycinnamate (E10)

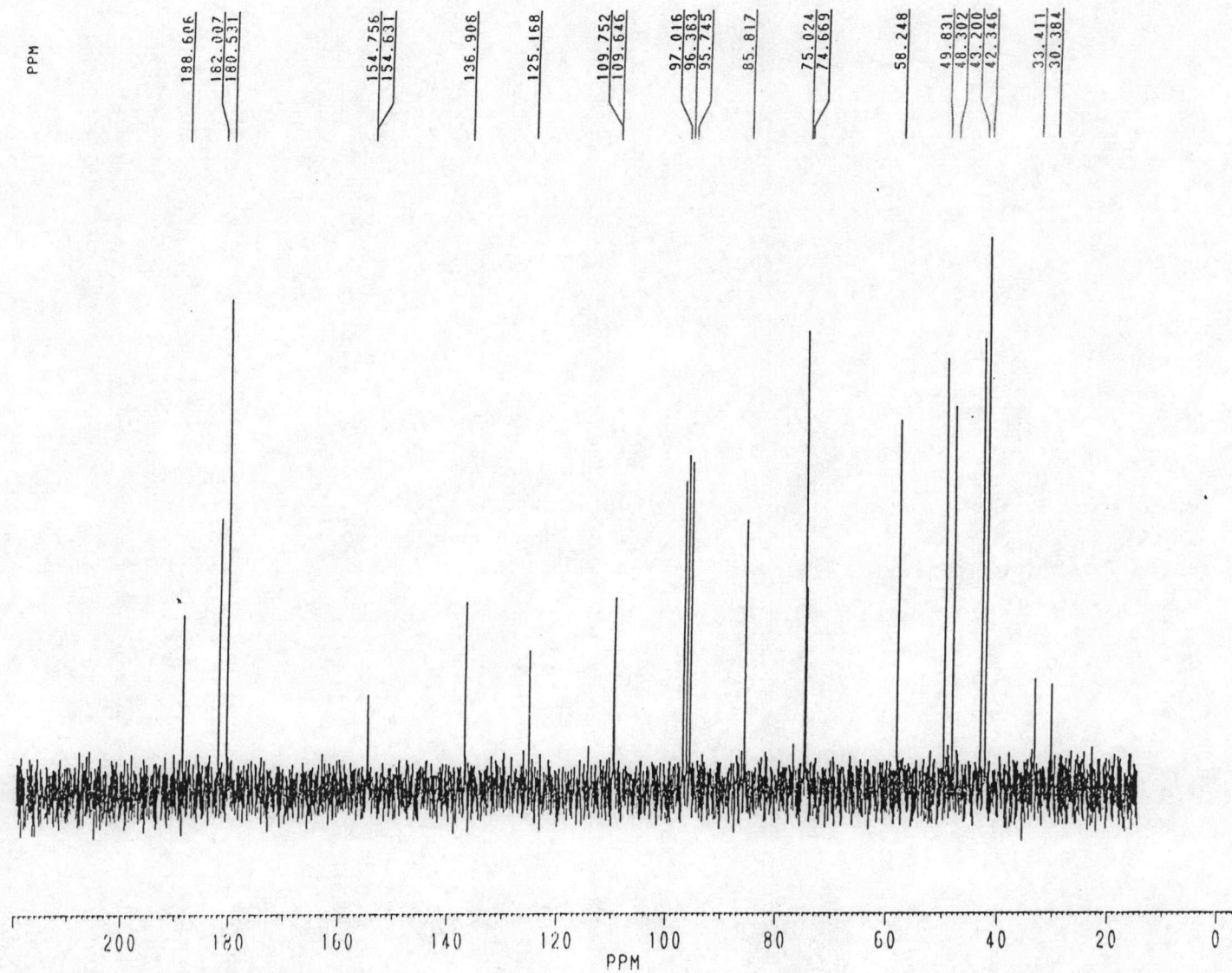


Figure B.6 ^{13}C -NMR spectrum of 2-ethylhexyl-2,4,6-trimethoxycinnamate (E10)

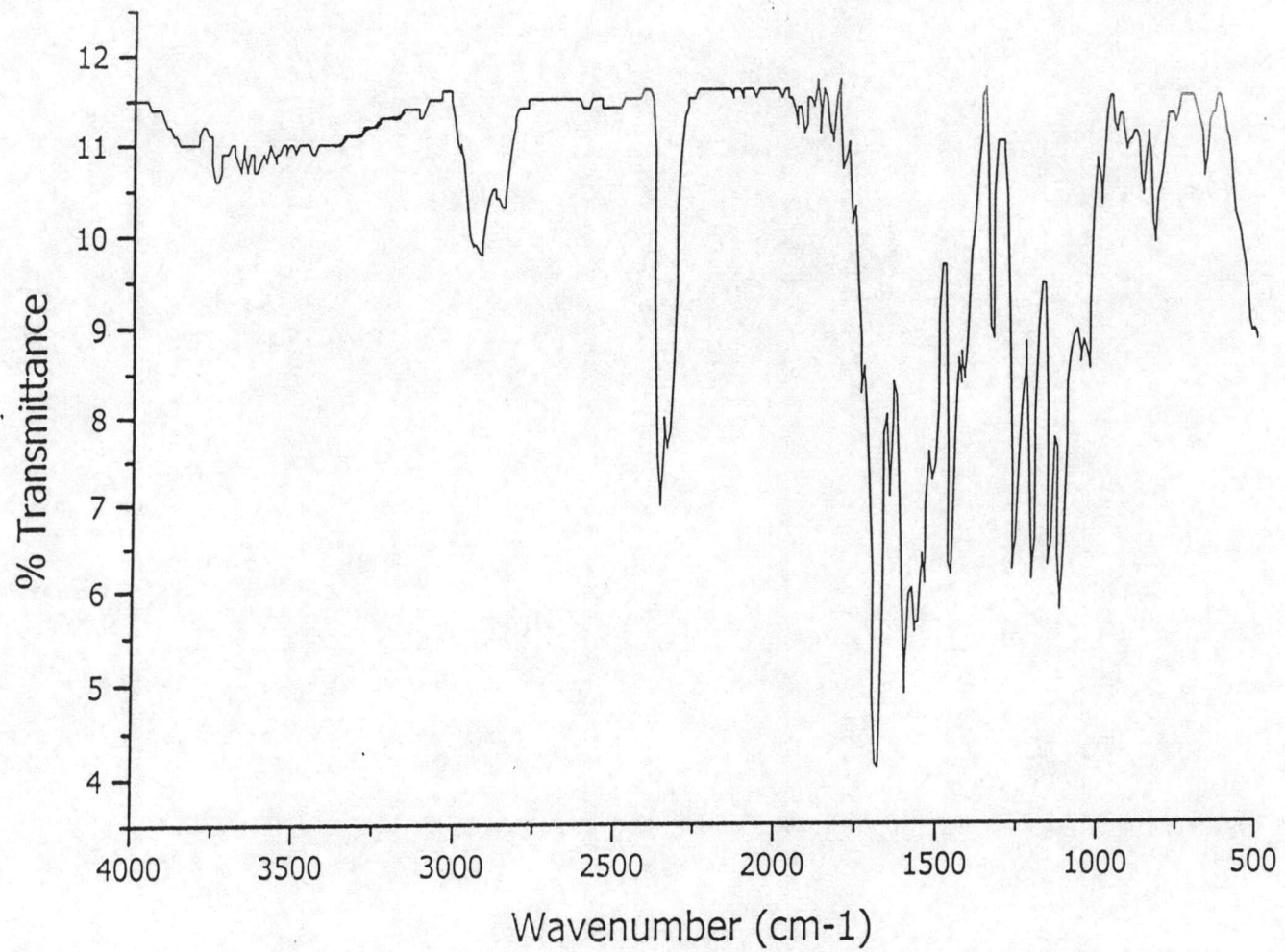


Figure B.7 IR spectrum of 2-ethylhexyl-2,4,6-trimethoxycinnamate (E10)

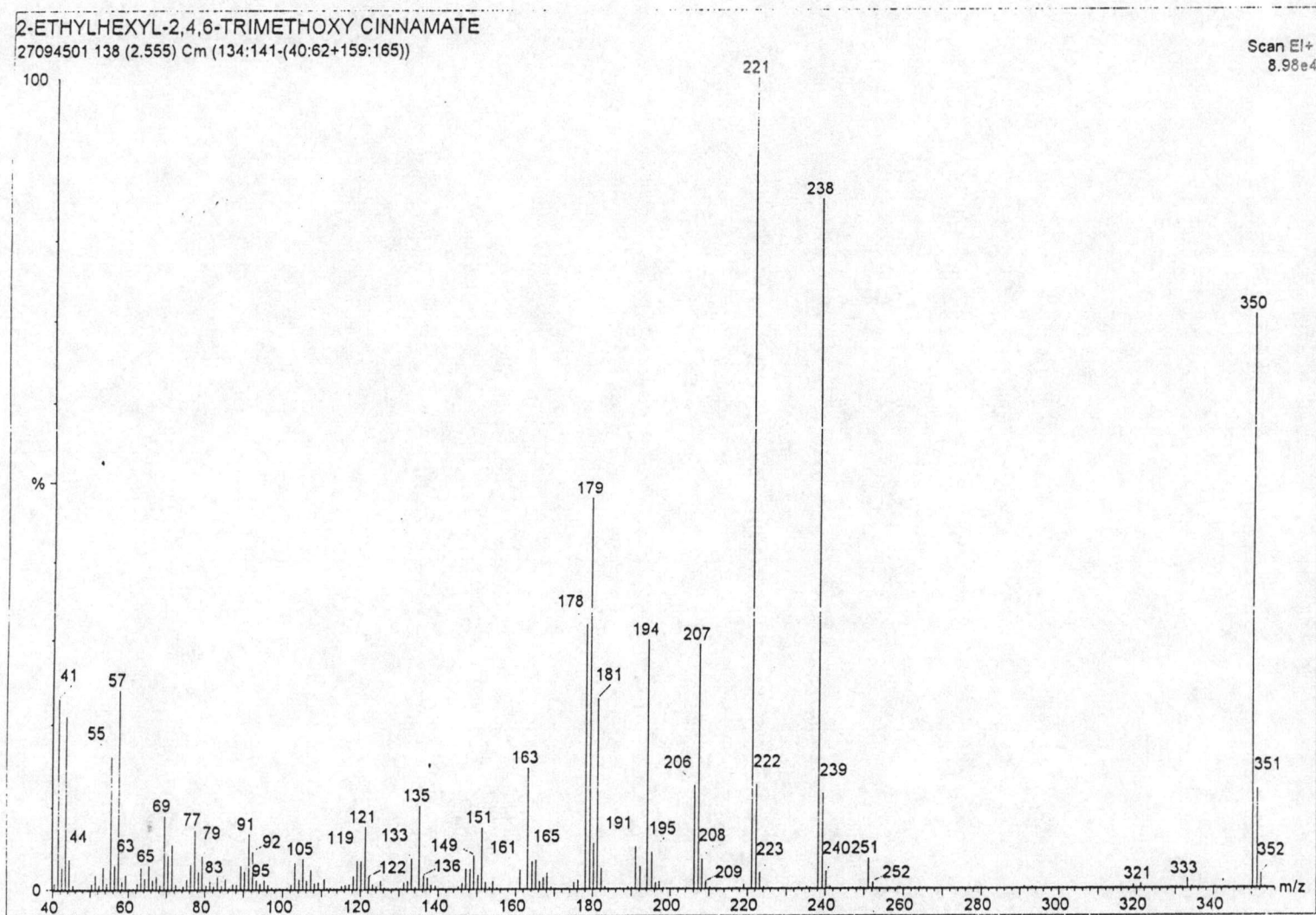


Figure B.8 Mass spectrum of 2-ethylhexyl-2,4,6-trimethoxycinnamate (E10)

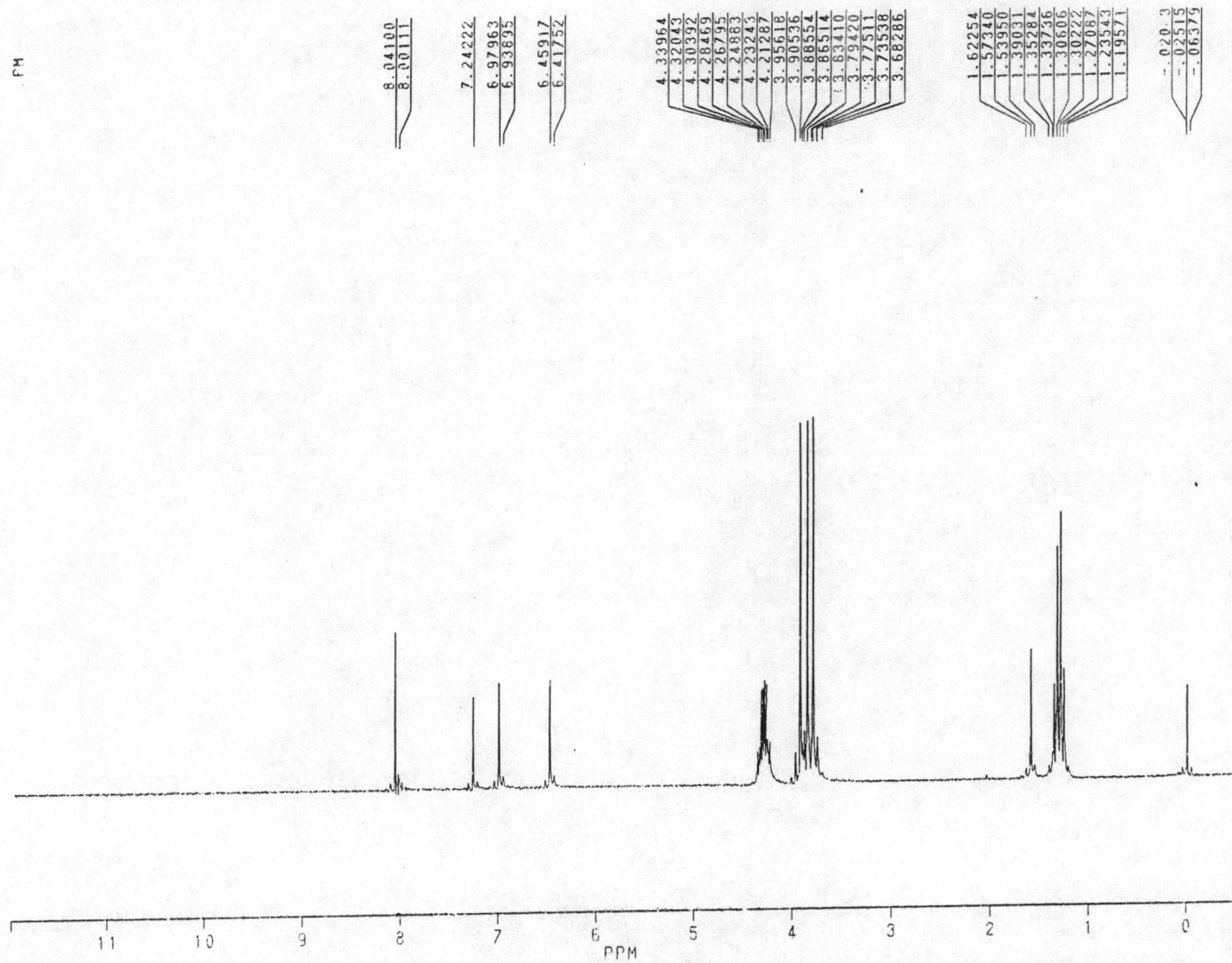


Figure B.9 ¹H-NMR spectrum of diethyl-2,4,5-trimethoxybenzalmalonate (BM8-1)

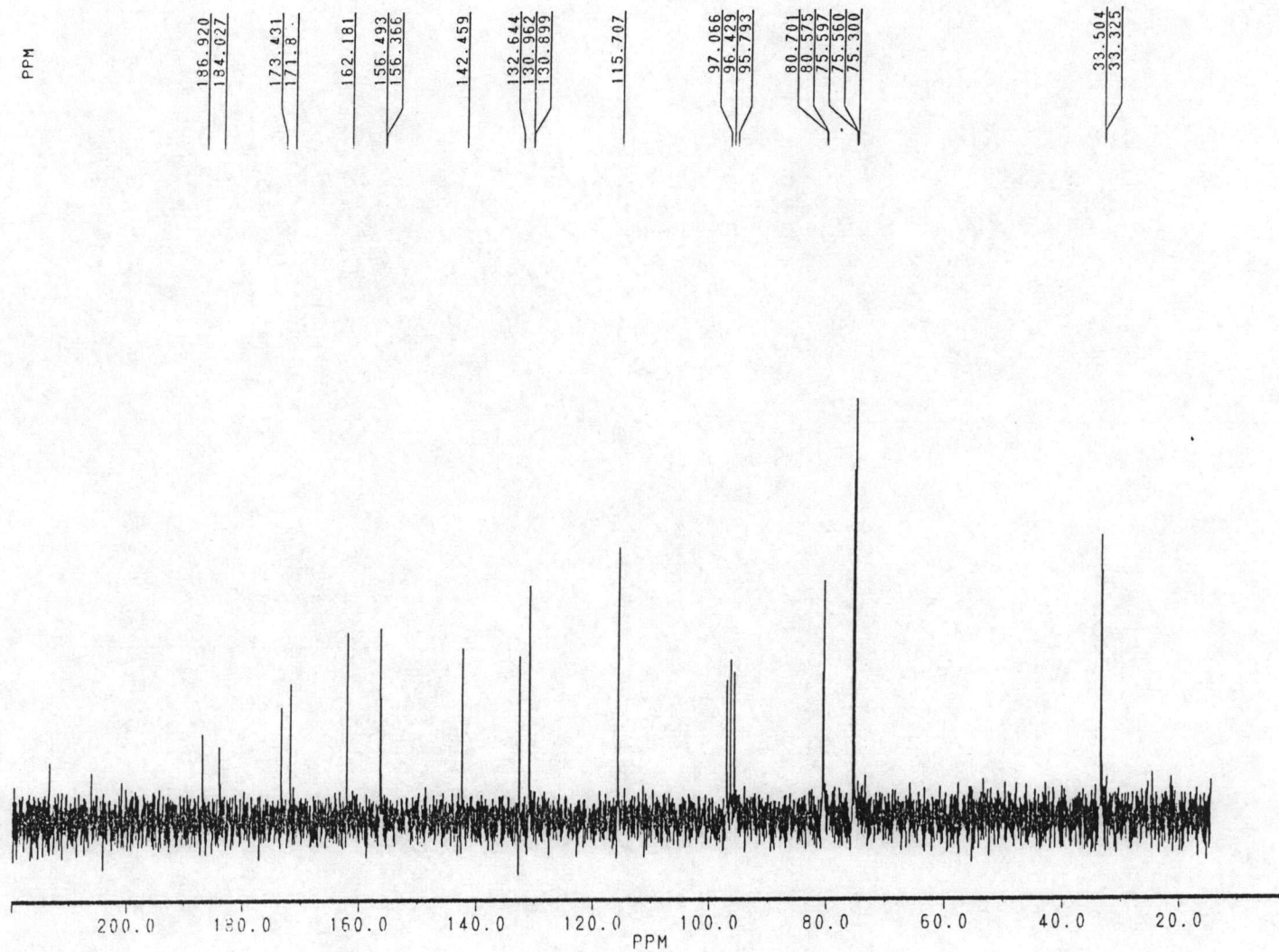


Figure B.10 ¹³C-NMR spectrum of diethyl-2,4,5-trimethoxybenzalmalonate (BM8-1)

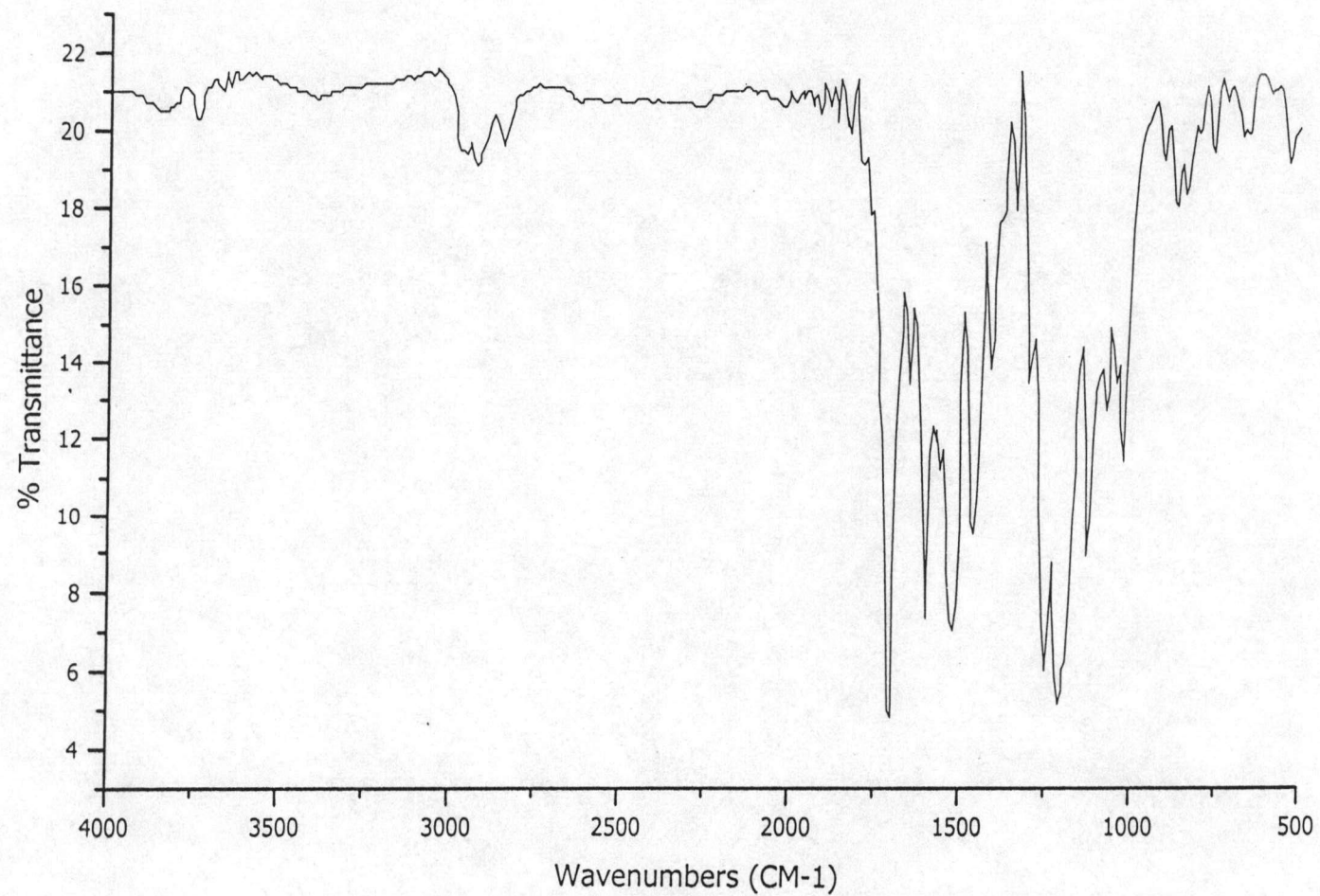


Figure B.11 IR spectrum of diethyl-2,4,5-trimethoxybenzalmalonate (**BM8-1**)

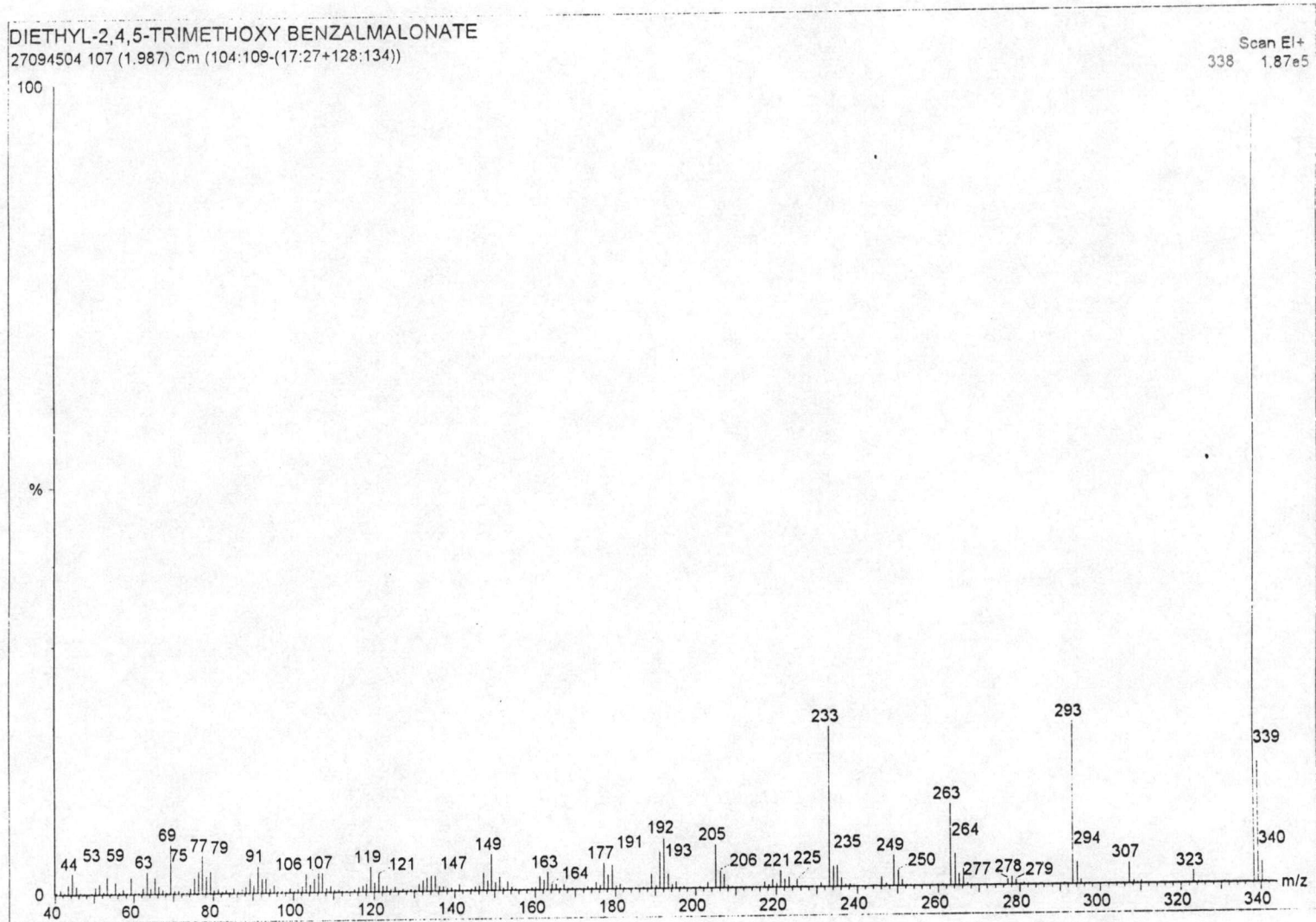


Figure B.12 Mass spectrum of diethyl-2,4,5-trimethoxybenzalmalonate (BM8-1)

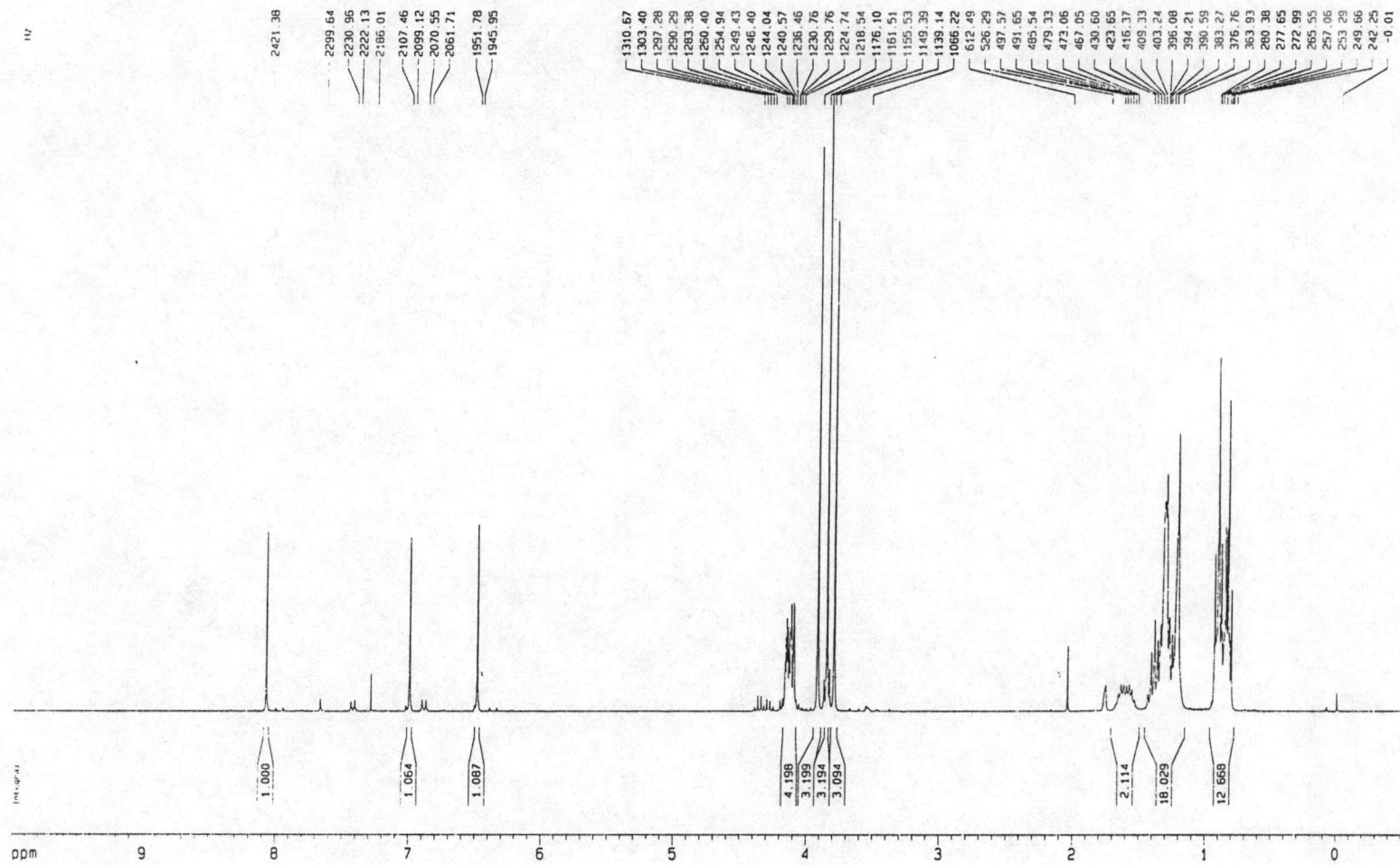


Figure B.13 ^1H -NMR spectrum of di-(2-ethylhexyl)-2,4,5-trimethoxybenzalmalonate (**BM8-2**)

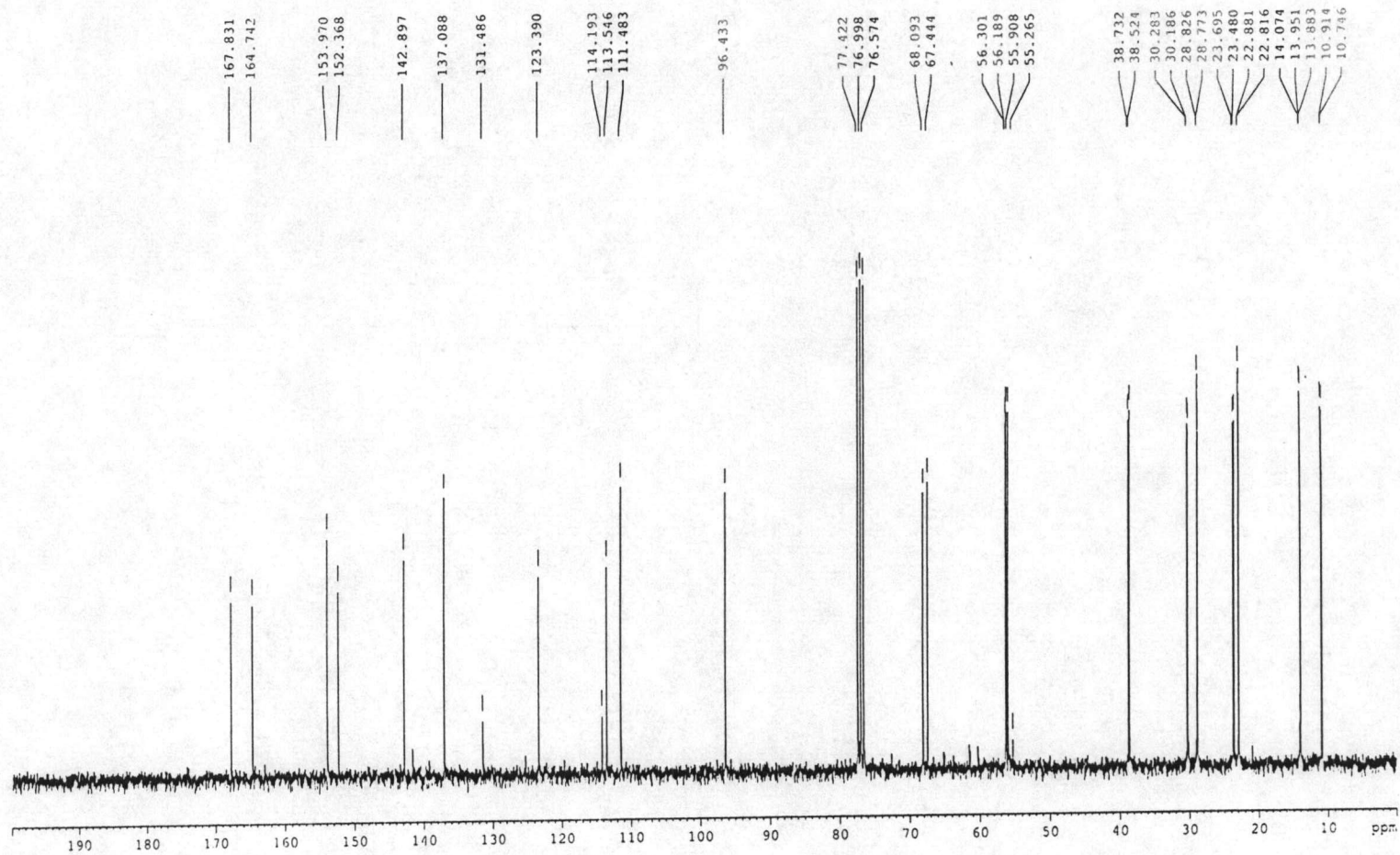


Figure B.14 ^{13}C -NMR spectrum of di-(2-ethylhexyl)-2,4,5-trimethoxybenzalmalonate (BM8-2)

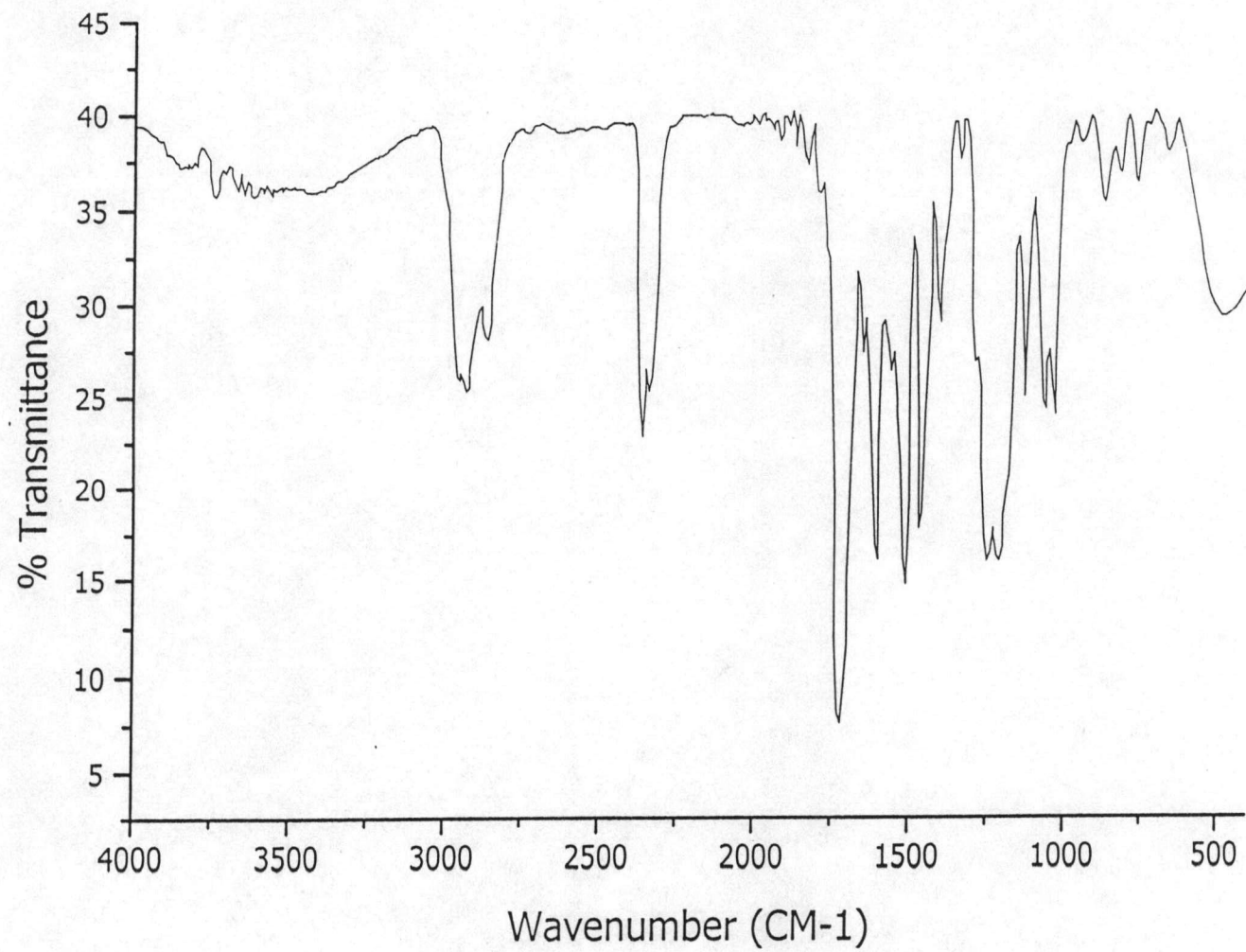


Figure B.15 IR spectrum of di-(2-ethylhexyl)-2,4,5-trimethoxybenzalmalonate (BM8-2)

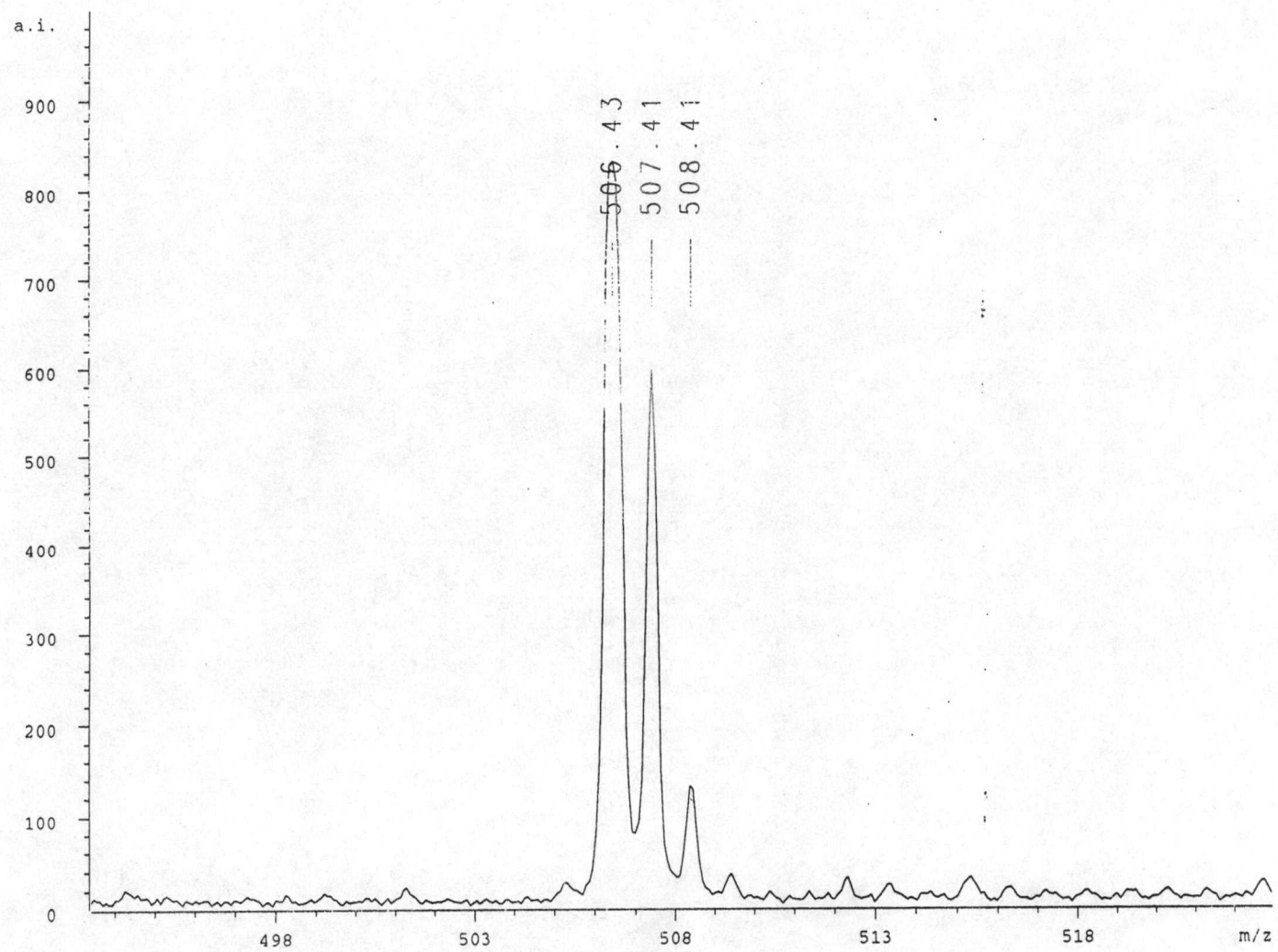


Figure B.16 Mass spectrum (MALDI-TOF) of di-(2-ethylhexyl)-2,4,5-trimethoxybenzalmalonate (**BM8-2**)

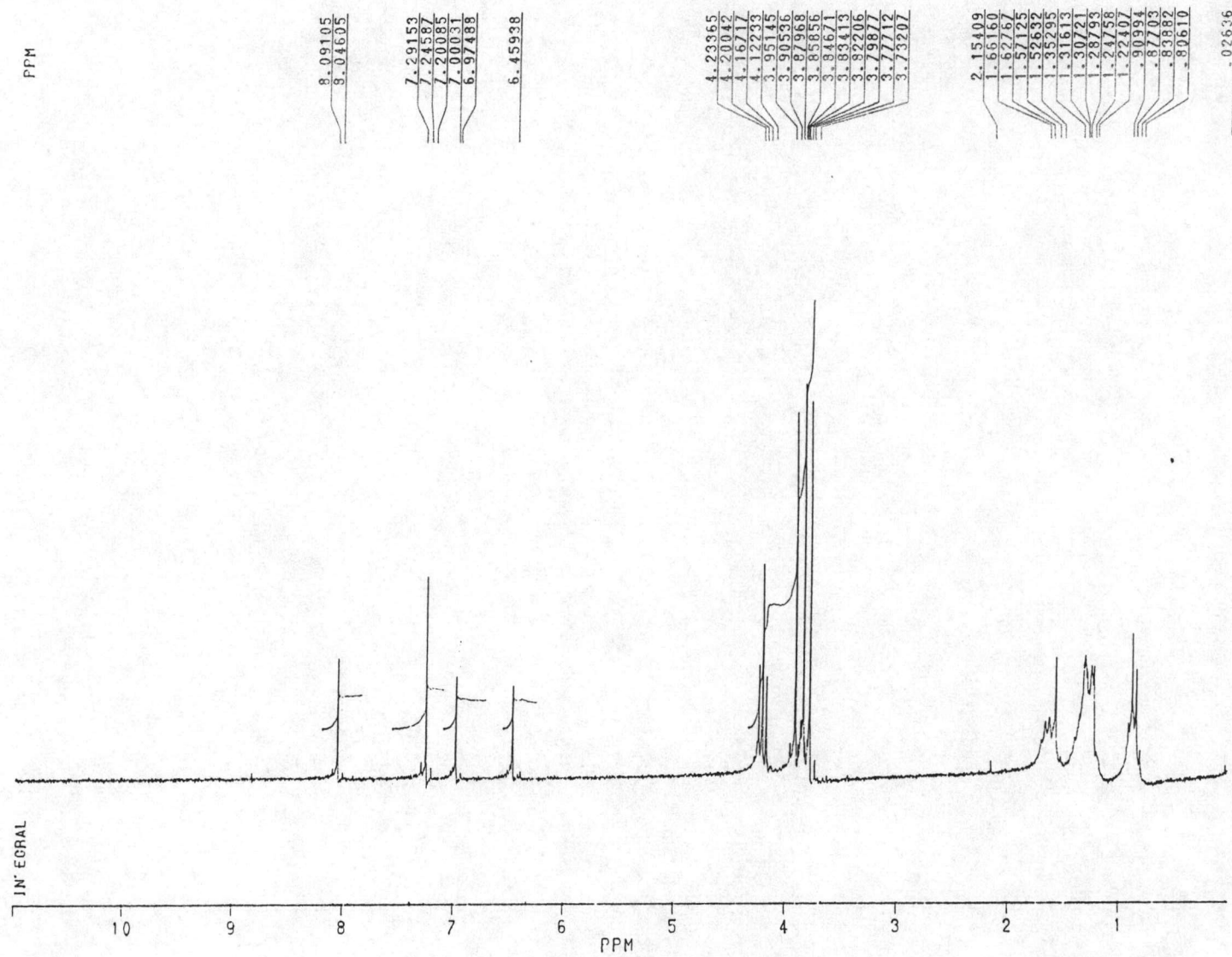


Figure B.17 $^1\text{H-NMR}$ spectrum of dihexyl-2,4,5-trimethoxybenzalmalonate (BM8-3)

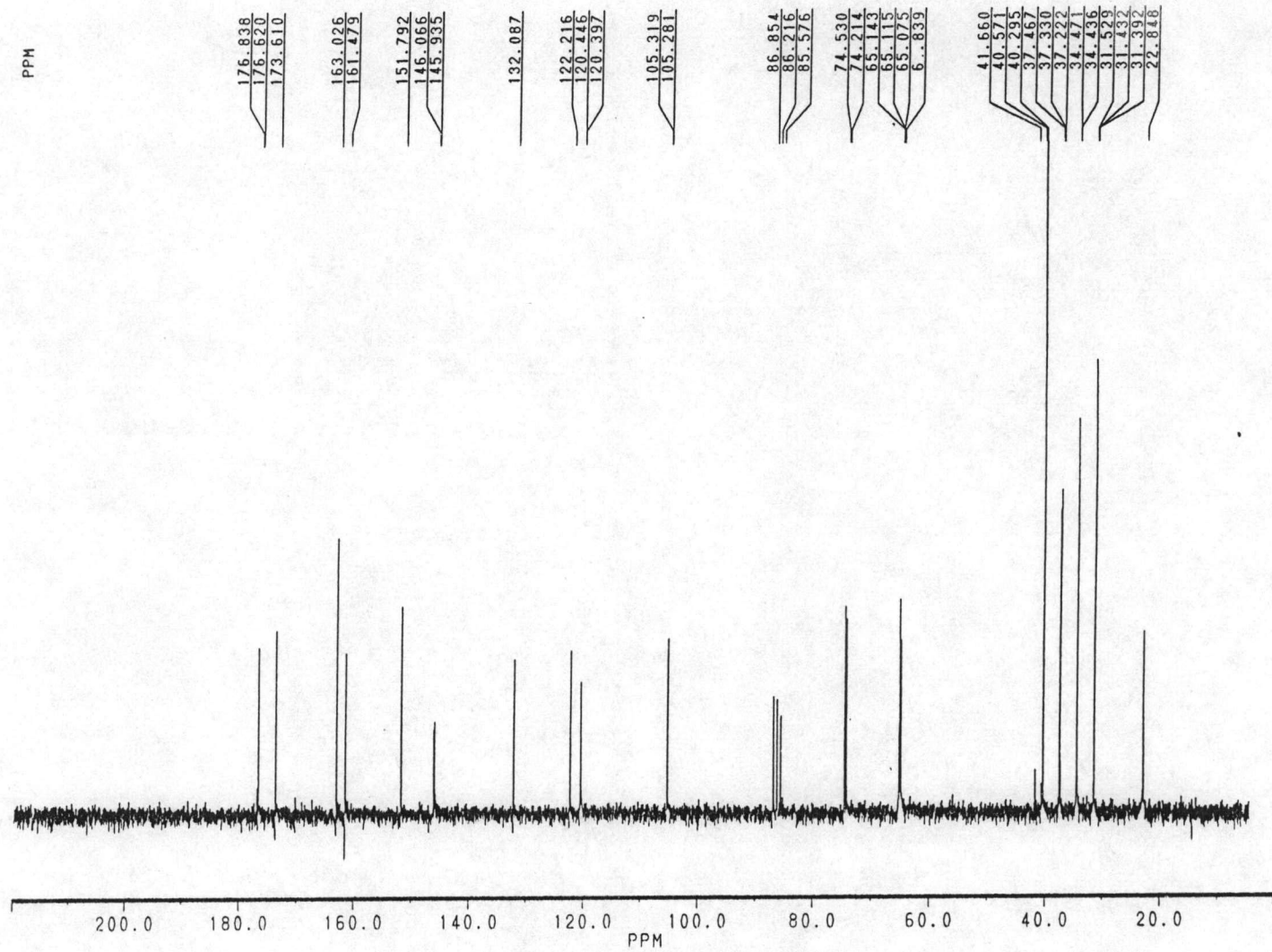


Figure B.18 ^{13}C -NMR spectrum of dihexyl-2,4,5-trimethoxybenzalmalonate (BM8-3)

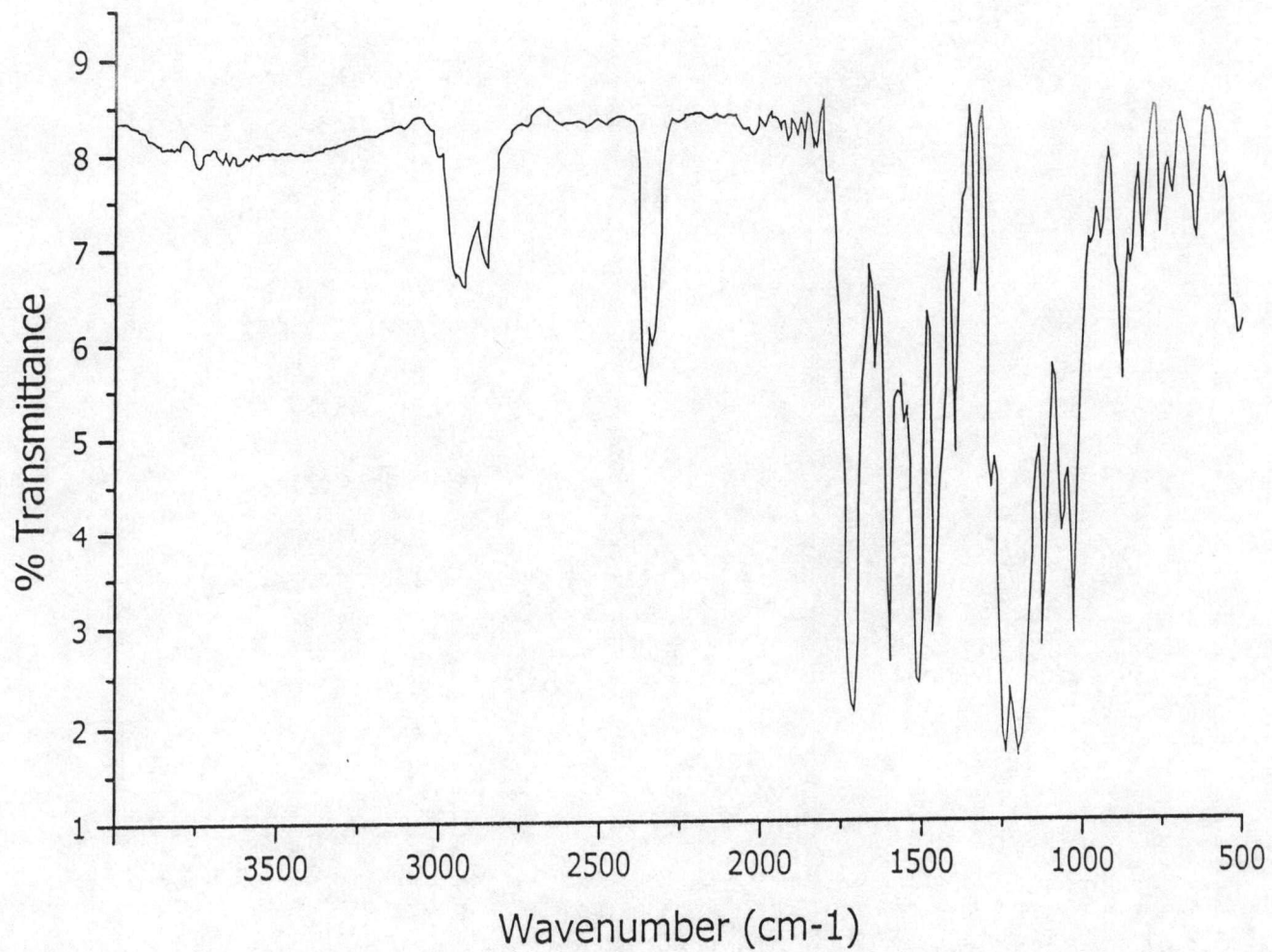


Figure B.19 IR spectrum of dihexyl-2,4,5-trimethoxybenzalmalonate (**BM8-3**)

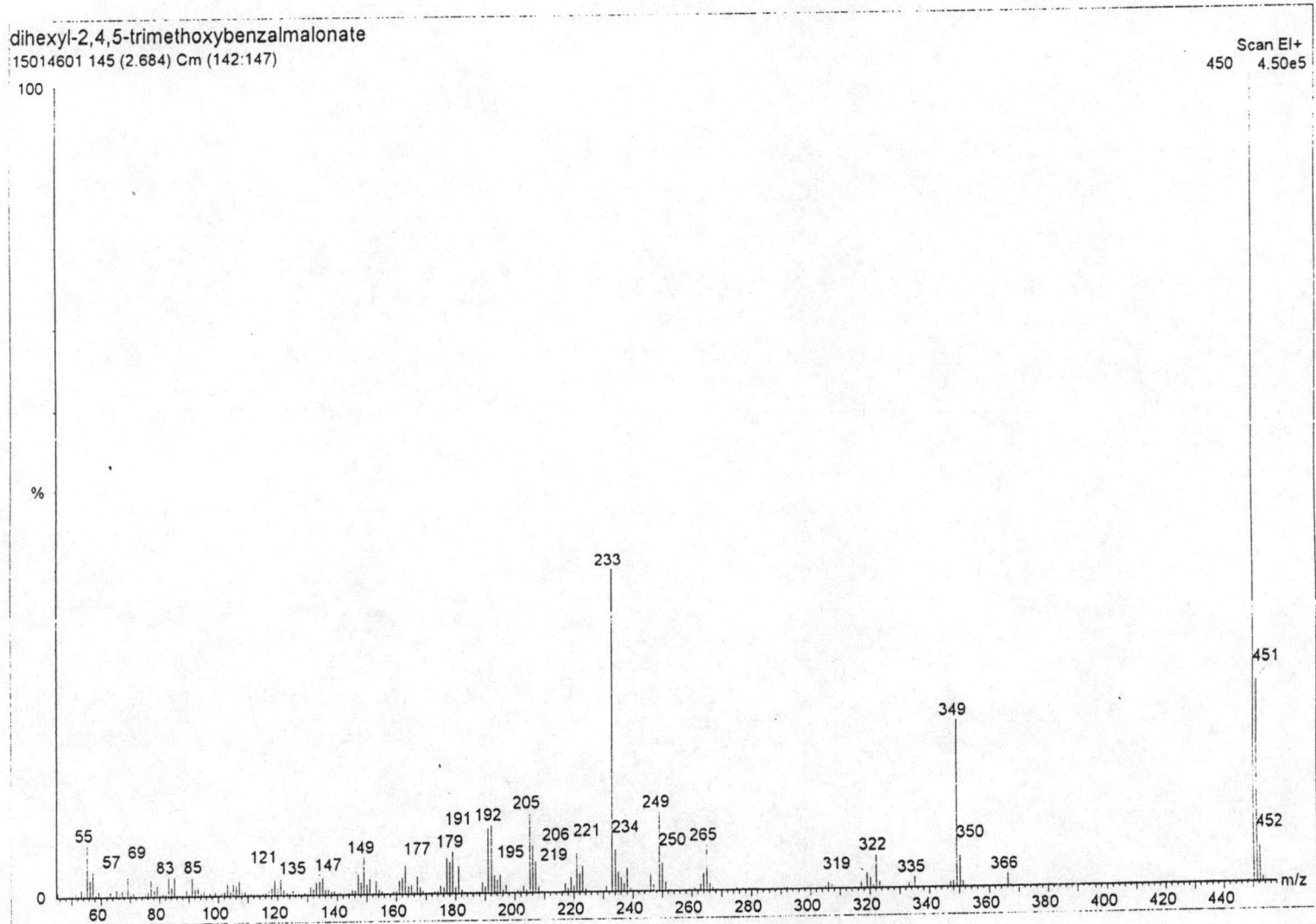


Figure B.20 Mass spectrum of diethyl-2,4,5-trimethoxybenzmalonate (BM8-3)

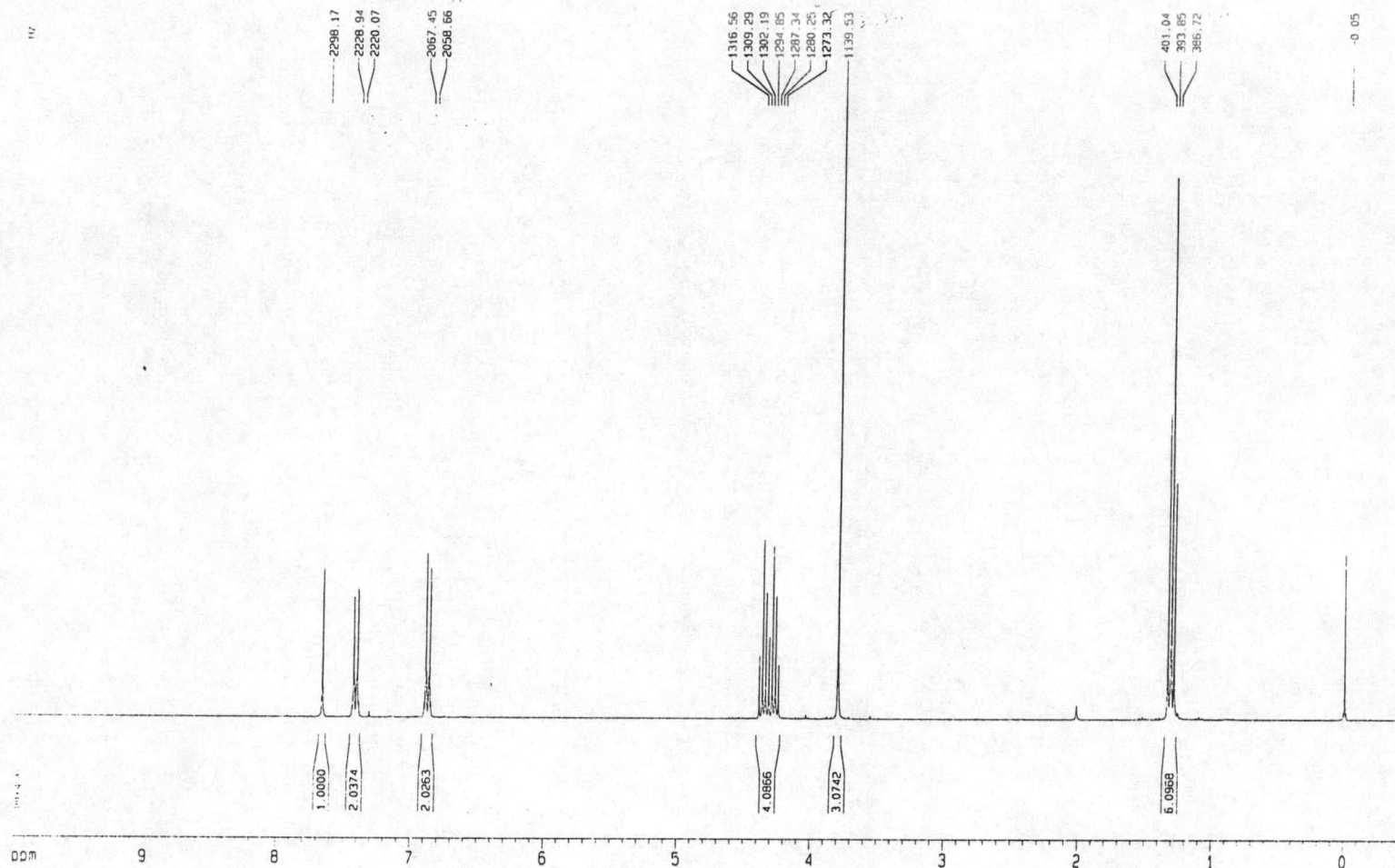


Figure B.21 $^1\text{H-NMR}$ spectrum of diethyl-4-methoxybenzalmalonate (BM1-1)

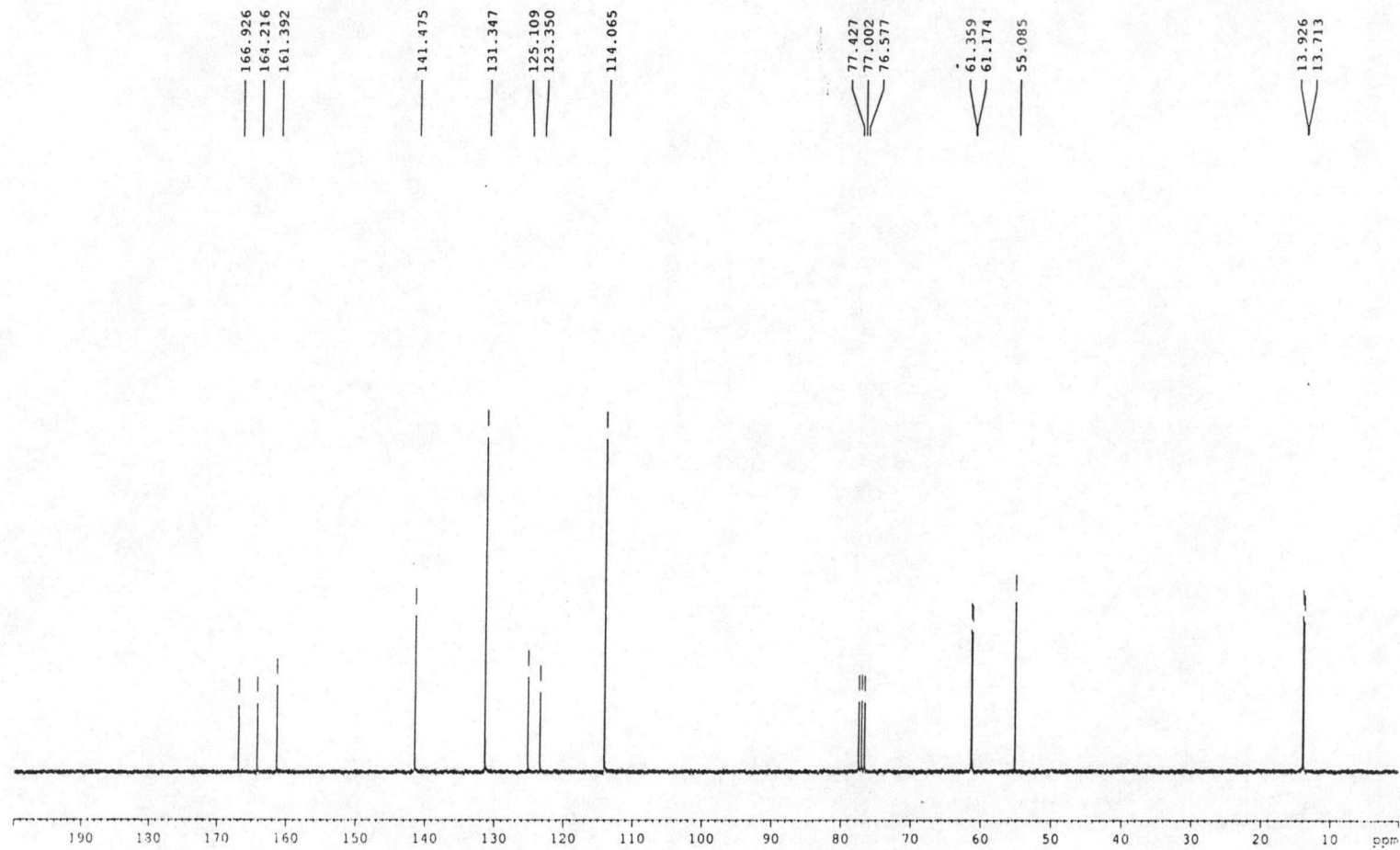


Figure B.22 ^{13}C -NMR spectrum of diethyl-4-methoxybenzalmalonate (BM1-1)

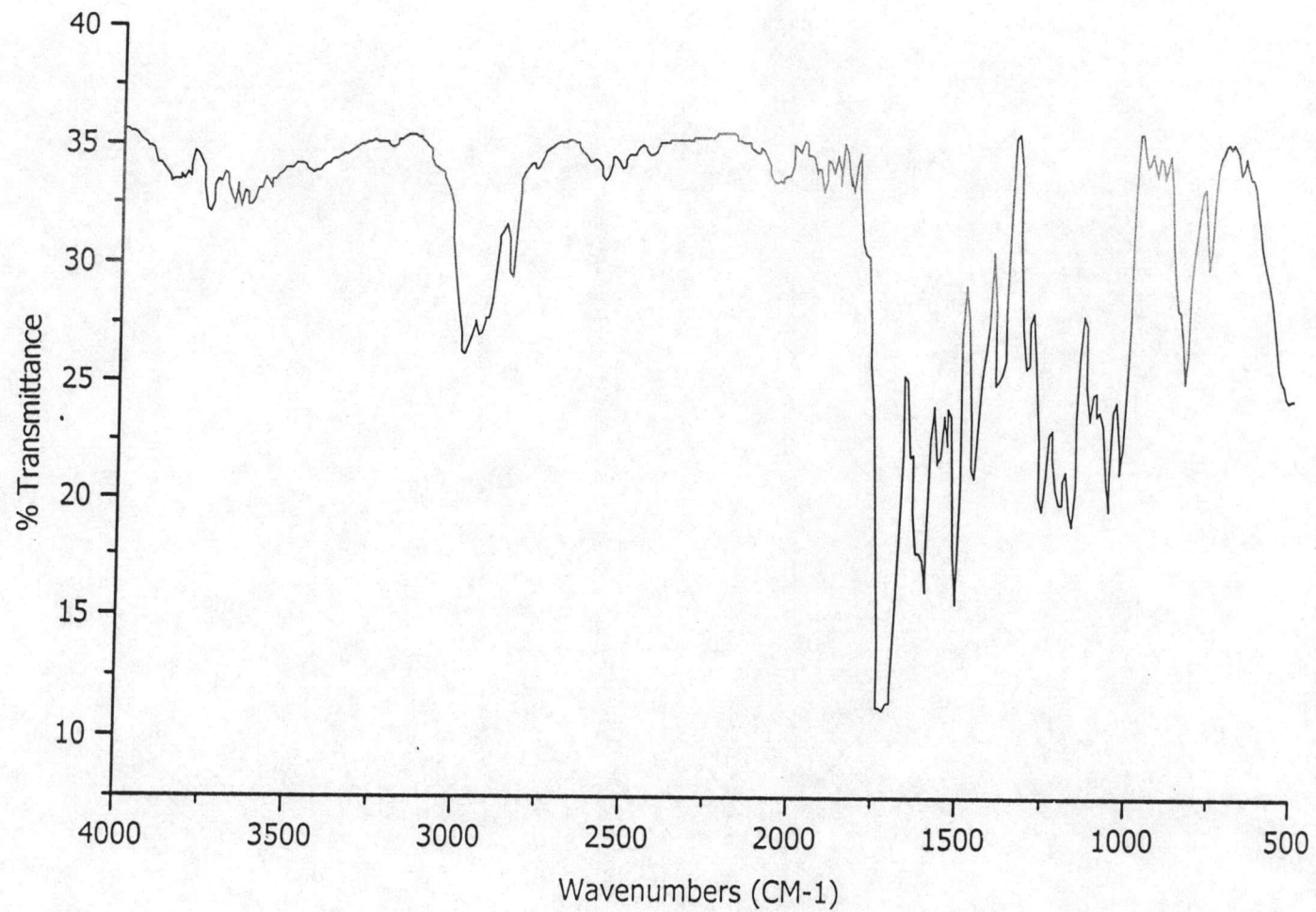


Figure B.23 IR spectrum of diethyl-4-methoxybenzalmalonate (**BM1-1**)

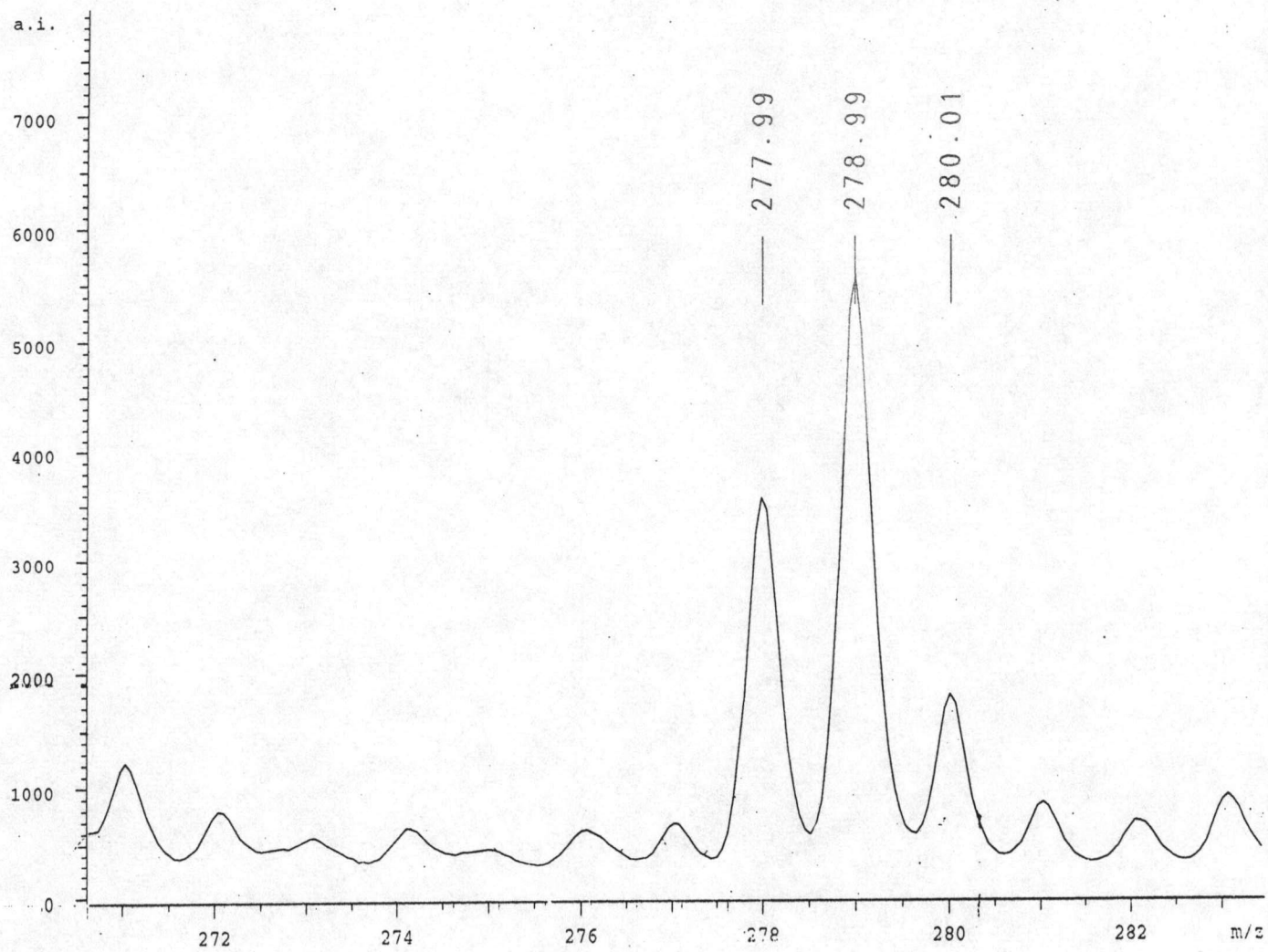


Figure B.24 Mass spectrum (MALDI-TOF) of diethyl-4-methoxybenzalmalonate (**BM1-1**)

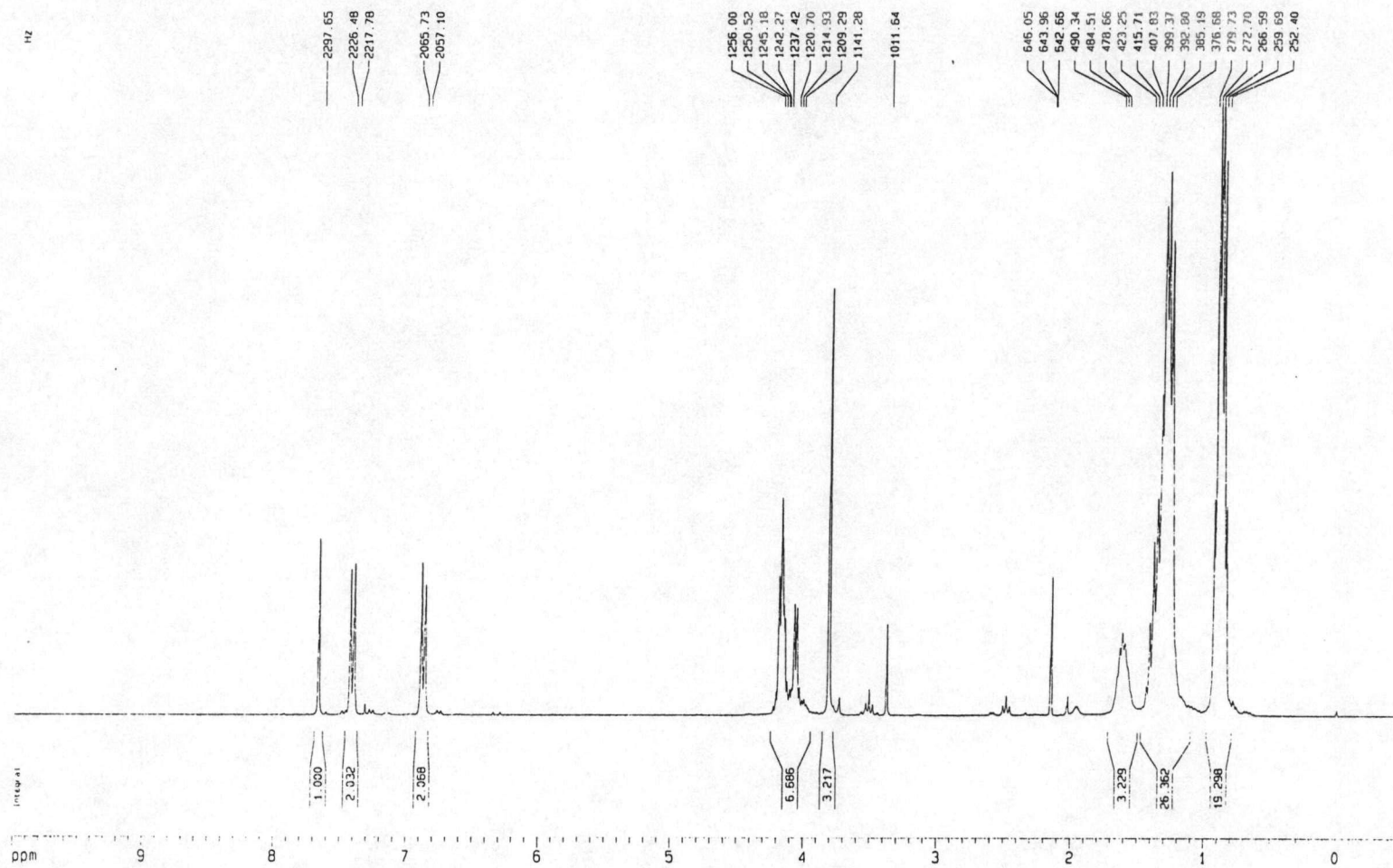


Figure B.25 ¹H-NMR spectrum of di-(2-ethylhexyl)-4-methoxybenzalmalonate (BM1-2)

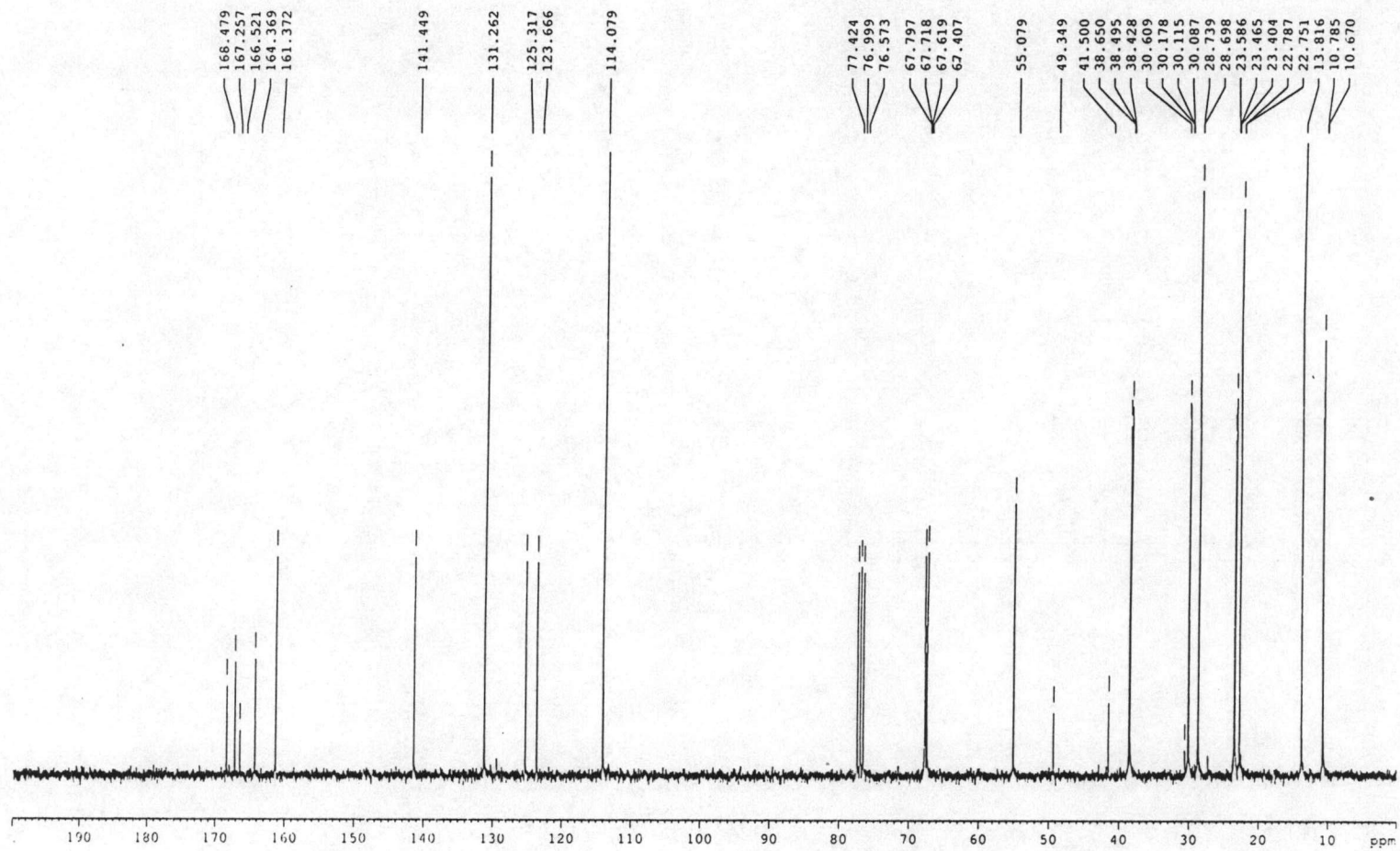


Figure B.26 ^{13}C -NMR spectrum of di-(2-ethylhexyl)-4-methoxybenzalmalonate (BM1-2)

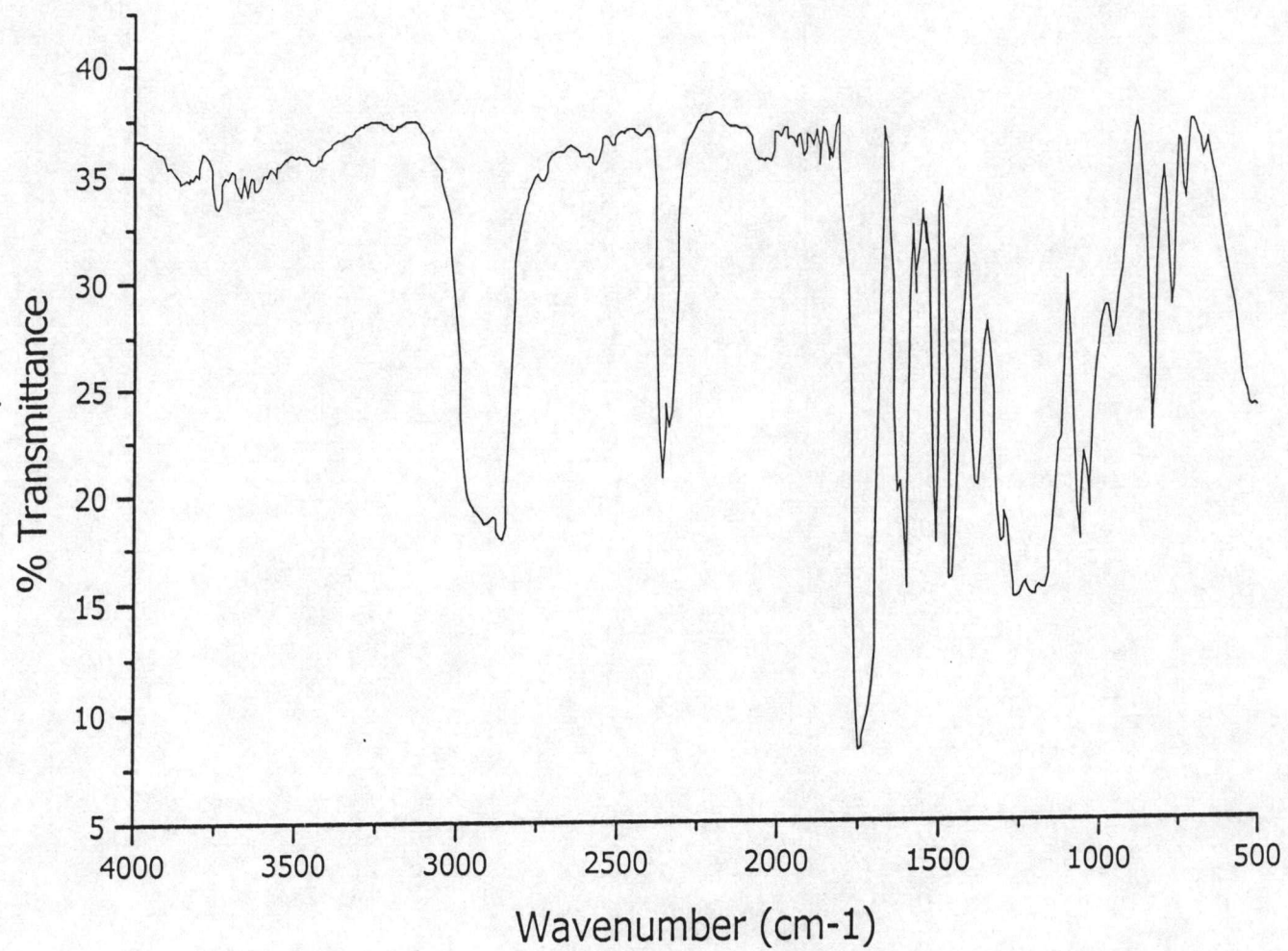


Figure B.27 IR spectrum of di-(2-ethylhexyl)- 4-methoxybenzalmalonate (**BM1-2**)

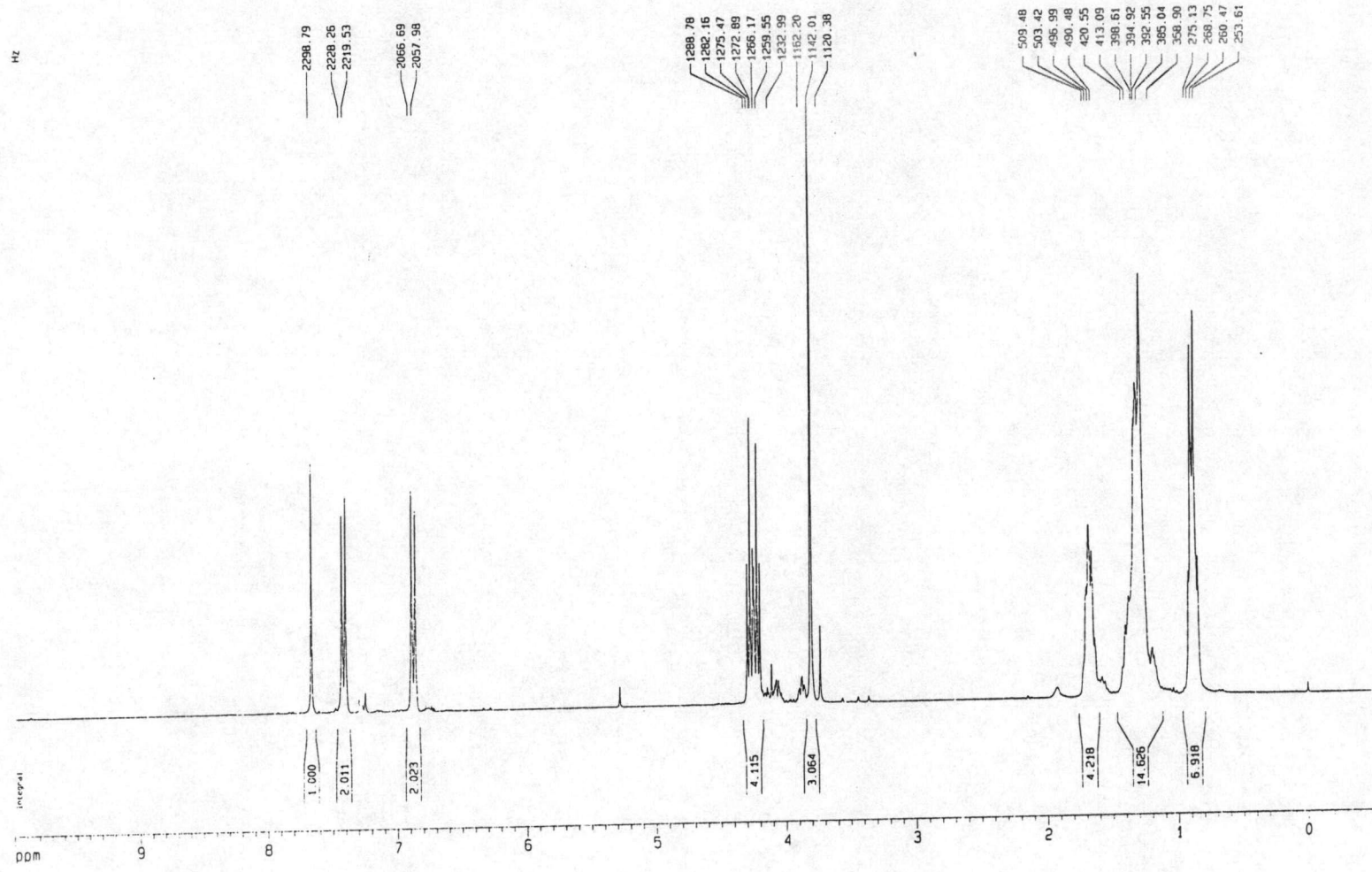


Figure B.28 ¹H-NMR spectrum of dihexyl-4-methoxybenzalmalonate (BM1-3)

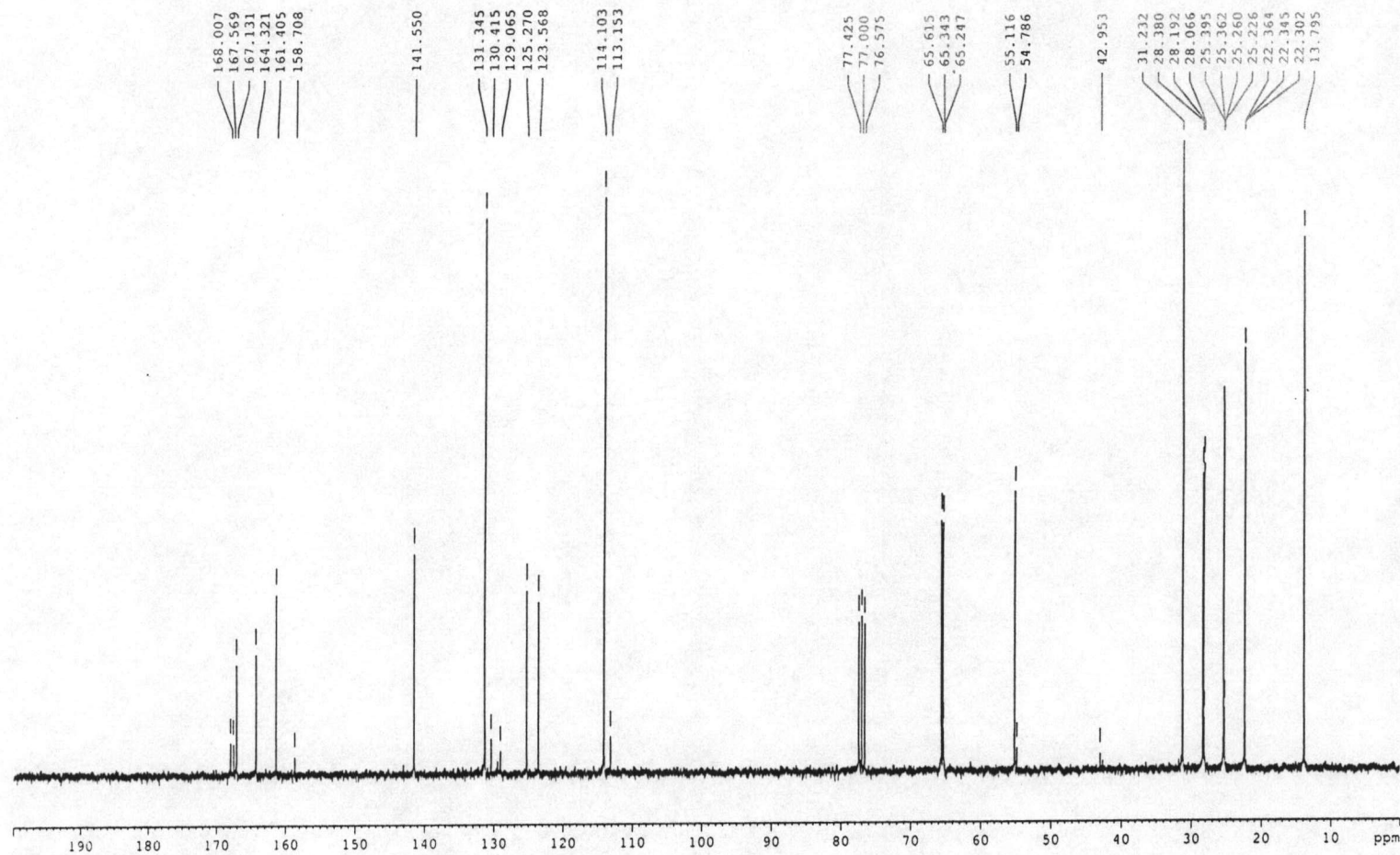


Figure B.29 ^{13}C -NMR spectrum of dihexyl-4-methoxybenzalmalonate (BM1-3)

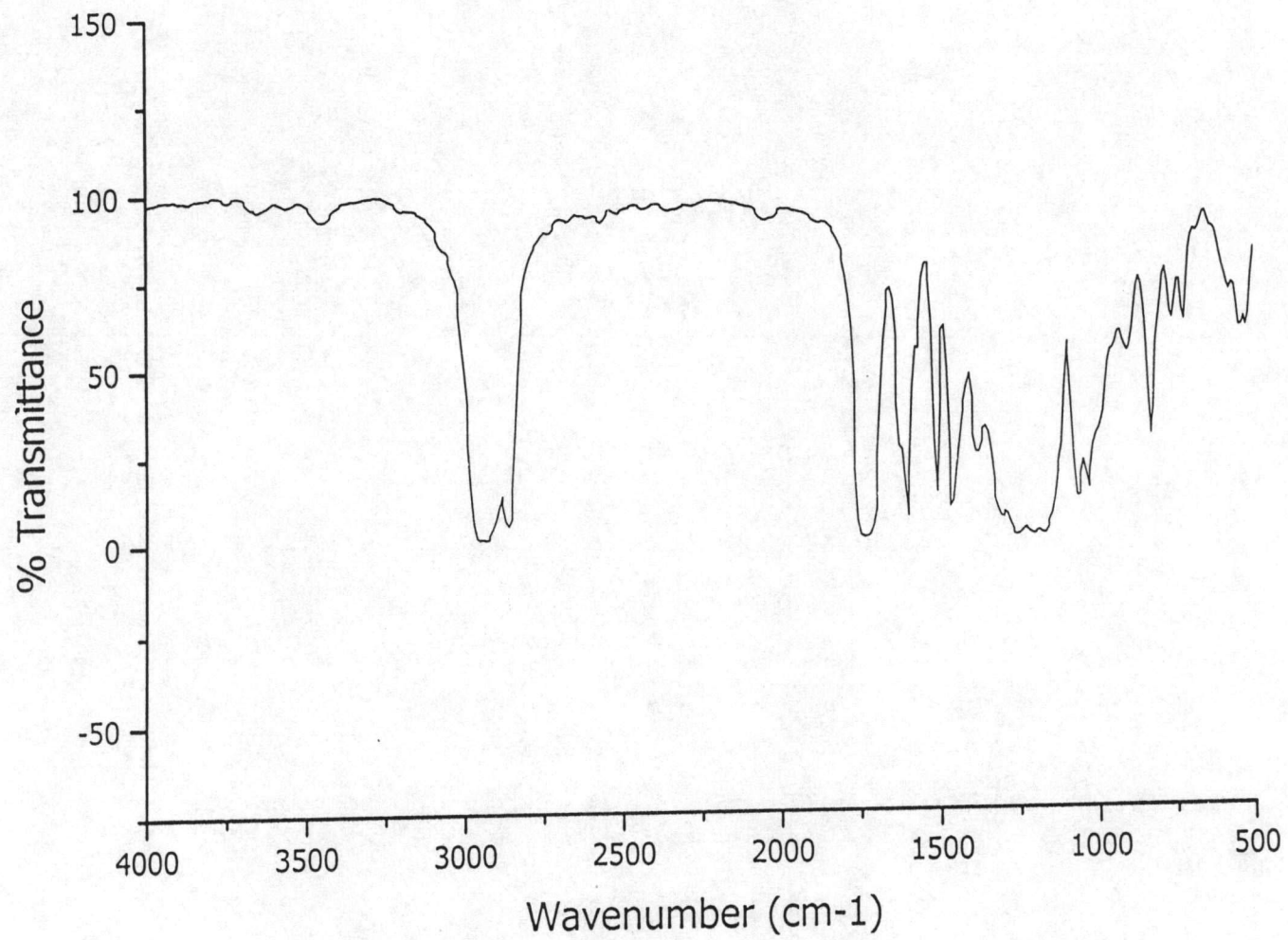
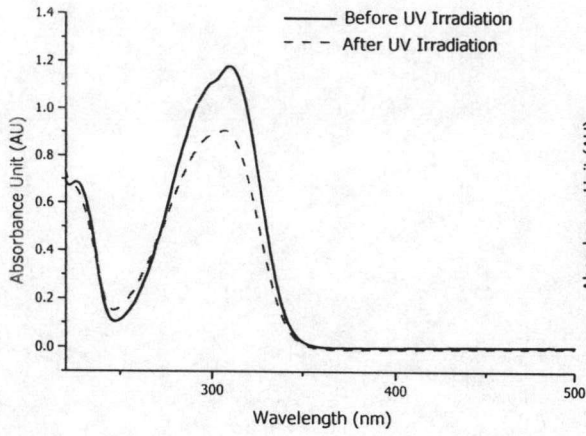
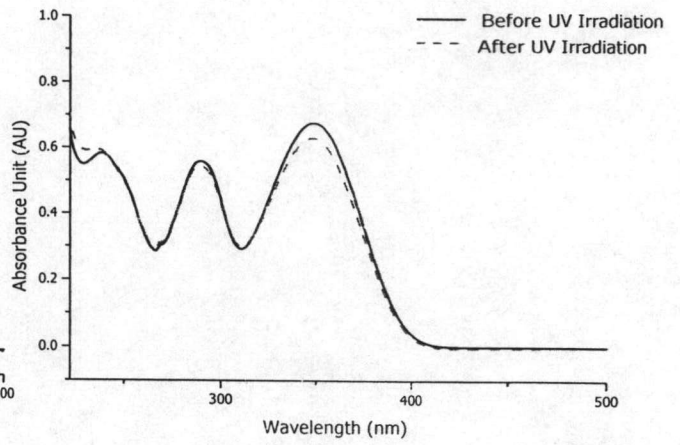


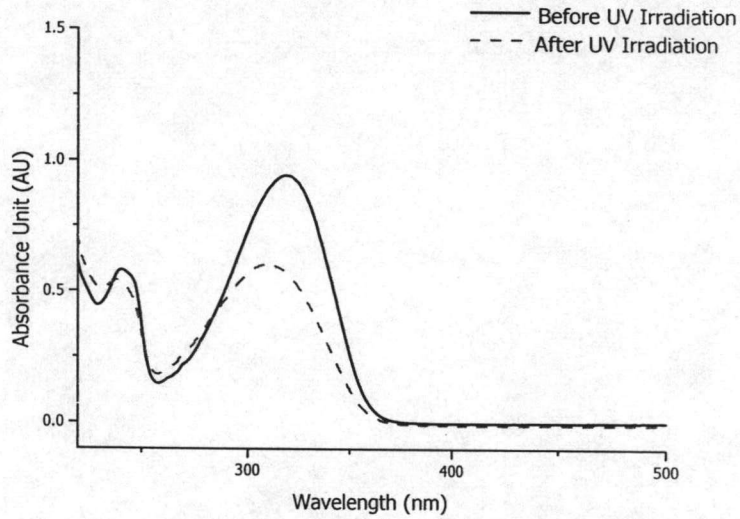
Figure B.30 IR spectrum of dihexyl-4-methoxybenzalmalonate (BM1-3)



a)



b)



c)

Figure B.31 UV spectra of a) OMC, b) E8 and c) E10 before and after UV irradiation for 2 hours

APPENDIX C

C.1 Photostability Test

Table C.1 Percent relative absorbance of synthesized UV-filters in UVA region in methanol

Time (h)	Percent Relative Absorbance			
	E2	BM8-1	BM8-2	BM8-3
0	100	100	100	100
1	-	100	98.25	98.36
1.25	79.98	-	-	-
2	78.74	100	99.07	98.52
3	-	100	99.95	98.57
4	-	100	100	97.81
5.5	78.33	-	-	-
7	-	100	99.11	93.45
10	75.76	-	-	-
11	-	99.4	99.61	97.26
16	76.97	-	-	-
18	-	100	100	96.83
26.5	75.82	-	-	-
28	-	100	96.33	100
34.5	75.12	-	-	-
36	-	100	95.15	100

Table C.2 Percent relative absorbance of synthesized UV-filters in UVB region in methanol

Time (h)	Percent Relative Absorbance								
	E1 ^a	E8	E10	BM8-1	BM8-2	BM8-3	BM1-1	BM1-2	BM1-3
0	100	100	100	100	100	100	100	100	100
1	66.71	-	48.38	100	98.54	97.88	100	99.75	98.72
1.25	-	78.36	-	-	-	-	-	-	-
2	68.66	79.19	47.78	98.72	98.44	98.76	-	-	-
2.5	-	79.93	-	-	-	-	-	-	-
3	69.38	-	50.23	99.75	99.16	99.6	100	99.21	99.68
4	68.47	-	48.56	99.16	99.42	98.81	-	-	-
5	-	-	-	-	-	-	100	99.59	99.87
5.5	-	77.96	-	-	-	-	-	-	-
7	70.46	-	48.35	97.86	98.26	97.96	-	-	-
8	-	-	-	-	-	-	100	98	99.62
10	-	77.72	-	-	-	-	-	-	-
11	69	-	50.78	95.92	98.55	98.13	-	-	-
16	-	77.48	-	-	-	-	100	96.83	99.27
18	69.81	-	52.15	96.24	99.95	97.97	-	-	-
21.5	-	-	-	-	-	-	100	99.74	98.81
26.5	-	76.76	-	-	-	-	-	-	-
28	69.76	-	52.57	99.11	97.28	97.27	-	-	-
34.5	-	76.73	-	-	-	-	-	-	-
36	64.83	-	44.8	99	97.12	100	-	-	-

^a Standard OMC (E1)

Table C.3 Percent relative absorbance of synthesized UV-filters in UVA region in hexanes

Time (h)	Percent Relative Absorbance			
	E8	BM8-1	BM8-2	BM8-3
0	100	100	100	100
1	90.79	100	98	99.5
3	89.9	100	99.42	100
7.25	90.07	100	99.57	99.88
12.25	95.19	101.74	100	100
18	96.1	100.73	100	103.9
22.5	95.01	104.3	100	105.1
27.58	94.69	100	105	105.3

Table C.4 Percent relative absorbance of synthesized UV-filters in UVB region in hexanes

Time (h)	Percent Relative Absorbance								
	E1^a	E8	E10	BM8-1	BM8-2	BM8-3	BM1-1	BM1-2	BM1-3
0	100	100	100	100	100	100	100	100	100
1	82.37	87.09	55.82	97.67	96.15	97.62	100	100	99.91
3	81.87	85.94	55.43	96.94	98.42	96.51	100	99.35	97.98
5	-	-	-	-	-	-	98.26	98.82	96.27
7.25	81.29	84.75	55.14	95.43	97.67	94.45	-	-	-
8	-	-	-	-	-	-	98	100	95.4
12.25	83.49	85.52	56.21	93.32	100	95.26	-	-	-
16	-	-	-	-	-	-	98.78	100	96.27
18	83.6	85.05	54.82	92.88	100	93.94	-	-	-
21.5	-	-	-	-	-	-	96.77	100	97.88
22.5	84.29	83.76	55.06	91.34	100	93.22	-	-	-
27.5	84.29	84.15	55.04	91.74	91.73	100	-	-	-

^a Standard OMC (E1)

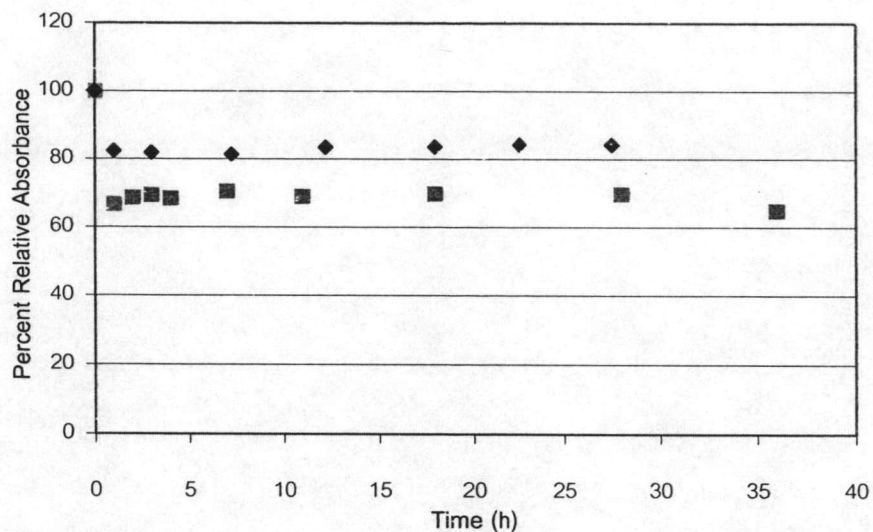


Figure C.1 Photo-equilibrium of OMC (E1); ■ in methanol and ◆ in hexanes

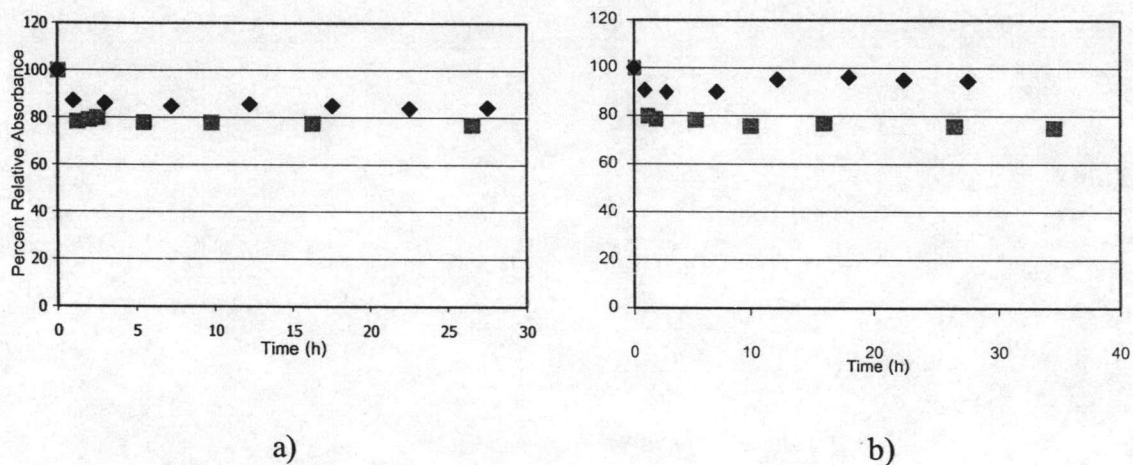


Figure C.2 Photo-equilibrium of 2-ethylhexyl-2,4,5-trimethoxycinnamate (E8); ■ in methanol and ◆ in hexanes; a) absorbed in UVB region and b) absorbed in UVA region

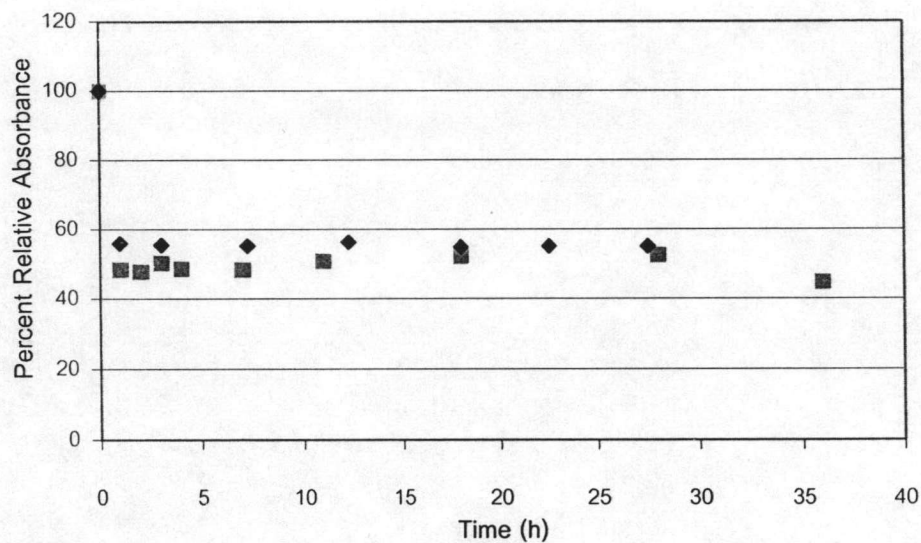


Figure C.3 Photo-equilibrium of 2-ethylhexyl-2,4,6-trimethoxycinnamate (**E10**);

■ in methanol and ◆ in hexanes

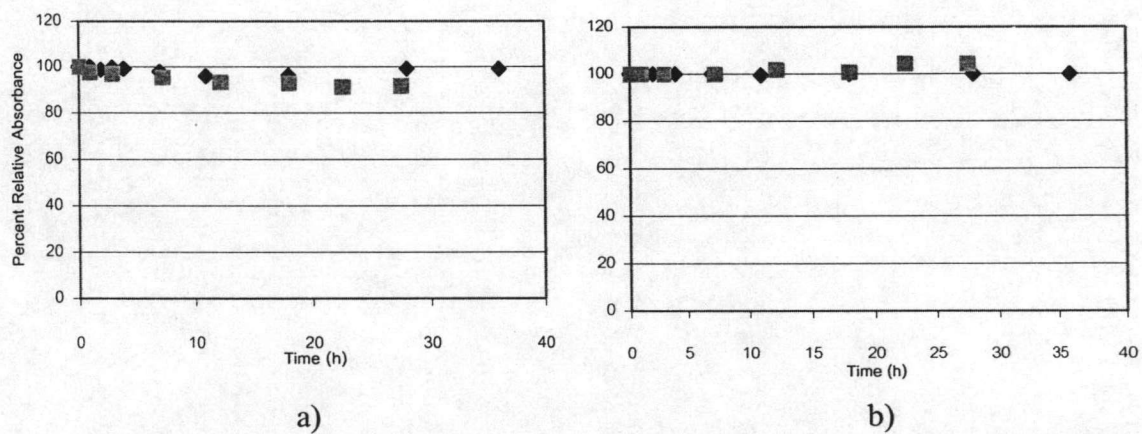


Figure C.4 Photo-equilibrium of diethyl-2,4,5-trimethoxybenzalmalonate (**BM8-1**); ■ in methanol and ◆ in hexanes; a) absorbed in UVB region and b) absorbed in UVA region

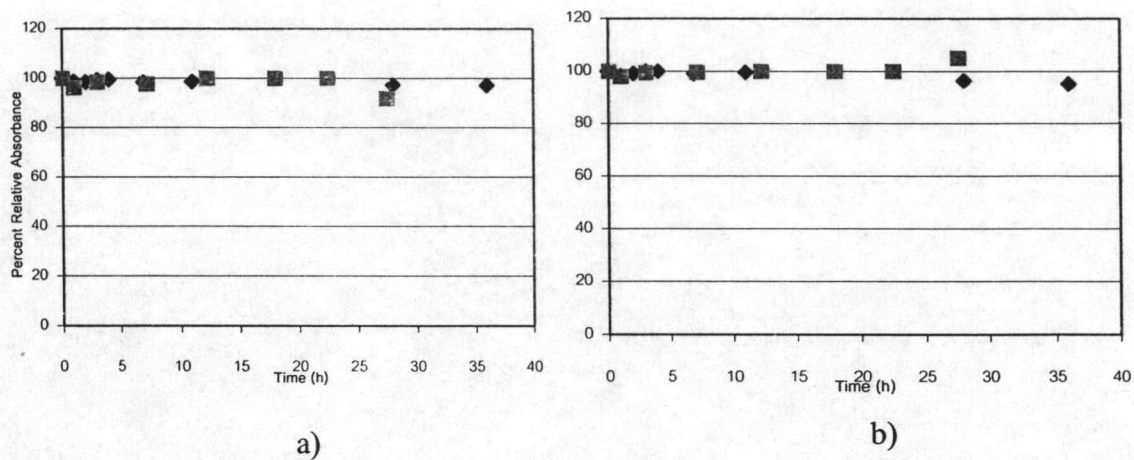


Figure C.5 Photo-equilibrium of di-(2-ethylhexyl)-2,4,5-trimethoxybenzalmalonate (**BM8-2**); ■ in methanol and ◆ in hexanes; a) in UVB region and b) in UVA region

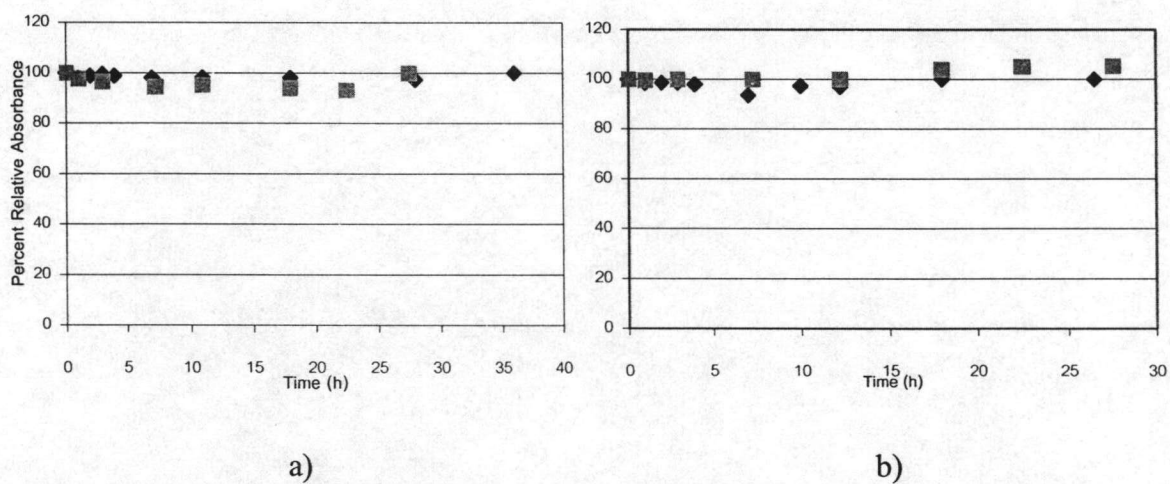


Figure C.6 Photo-equilibrium of dihexyl-2,4,5-trimethoxybenzalmalonate (**BM8-3**); ■ in methanol and ◆ in hexanes; a) in UVB region and b) in UVA region

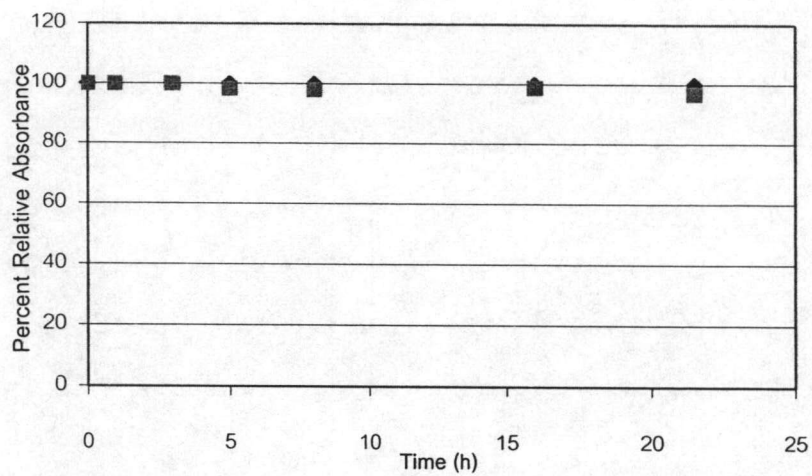


Figure C.7 Photo-equilibrium of diethyl-4-trimethoxybenzalmalonate (BM1-1); ■ in methanol and ◆ in hexanes

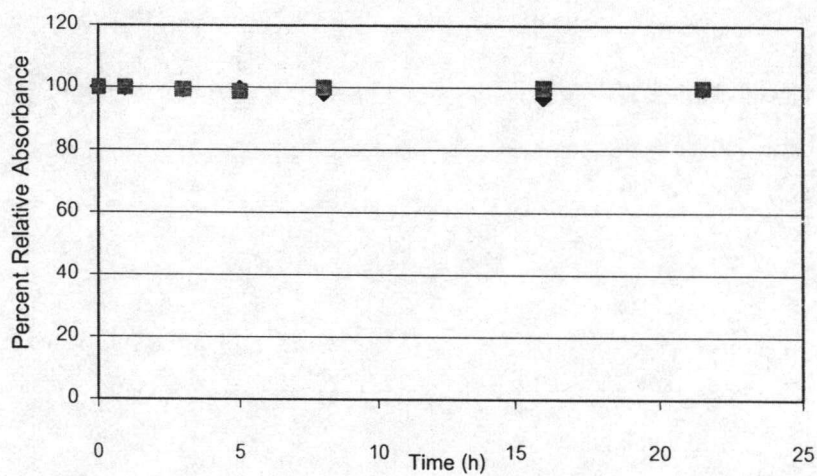


Figure C.8 Photo-equilibrium of di-(2-ethylhexyl)-4-trimethoxybenzalmalonate (BM1-2); ■ in methanol and ◆ in hexanes

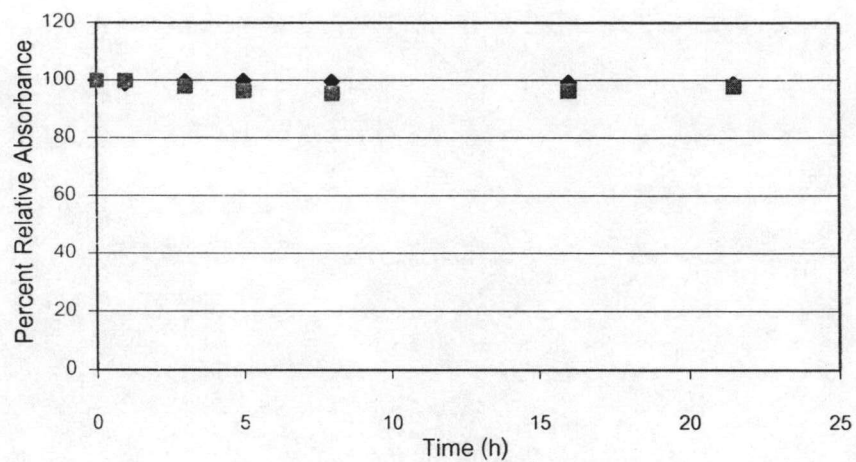


Figure C.9 Photo-equilibrium of dihexyl-4-trimethoxybenzalmalonate (**BM1-3**); ■ in methanol and ◆ in hexanes

VITA

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