

เอกสารอ้างอิง

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ภาคผนวก

สูตรอาหารของรองศาสตราจารย์ ดร.ถาวร วัชรภักย์ ซึ่งดัดแปลงมาจากสูตร Knudson B

$\text{Ca}(\text{NO}_3)_2 \cdot 4 \text{H}_2\text{O}$	1000.00 mg.
KH_2PO_4	250.00 mg.
$\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$	250.00 mg.
$(\text{NH}_4)_2\text{SO}_4$	500.00 mg.
Fe^* [จาก NaFe-EDTA]	5.00 mg.
$\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$	0.10 mg.
$\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$	2.00 mg.
H_3BO_3	1.00 mg.
$\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$	0.01 mg.
$\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$	0.02 mg.
$\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$	0.02 mg.
2,4-D	0.10 mg.
Sucrose	20 X 10 ³ mg.
Agar	8000.00 mg.
Glass-distilled water	1000 ml.
pH value	5.2 ± 0.3

*วิธีเตรียม NaFe-EDTA ละลาย Na_2EDTA 22.4 gm. ในน้ำ 186 ml.
และละลาย $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ 13.7 gm. ในน้ำ 364 ml แล้วผสมสารละลายทั้งสองเข้าด้วยกัน
ในสารละลายนี้จะมี Fe 5 mg/ml.

ประวัติการศึกษา

นายวีระวุฒิ มหามนตรี สำเร็จการศึกษาชั้นปริญญาบัณฑิตทางวิทยาศาสตร์ สาขาวิชาพฤกษศาสตร์ เกียรตินิยมอันดับ 2 จากจุฬาลงกรณ์มหาวิทยาลัย ในปี พ.ศ. 2504 ได้รับประกาศนียบัตรชั้นสูงสาขาวิชาพฤกษศาสตร์เมื่อปี พ.ศ. 2510 ปัจจุบัน เป็นอาจารย์ในแผนกวิชาพฤกษศาสตร์ คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย.