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APPENDICES

APPENDIX A

1. Preparation calibration curve of N-acetyl-D-glucosamine for chitinolytic enzyme assay by colorimetric method.

Calibration curve for GlcNAc was made by determining the absorbance value at 420 nm of standard GlcNAc according to the method of Schales.

Table A1 The amount of standard solution of GlcNAc and Δ Absorbance

standard No.	amount of GlcNAc (μ mole)	Δ Absorbance
1	0.6021	0.725
2	0.5018	0.599
3	0.4837	0.578
4	0.3870	0.468
5	0.2902	0.354
6	0.1935	0.204
7	0.0967	0.091

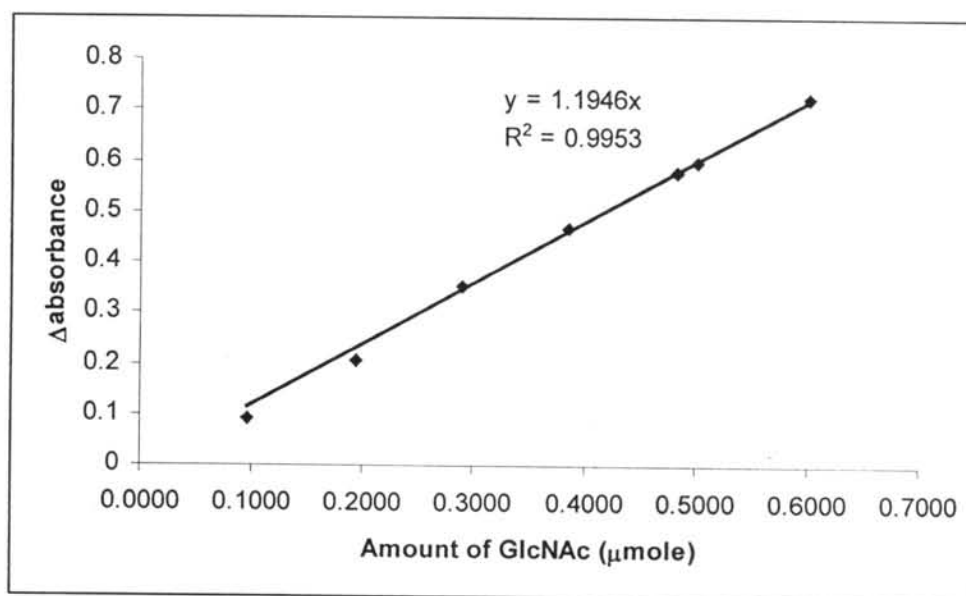


Figure A1 Correlation between amount of standard of *N*-acetyl-D-glucosamine and optical density (absorbance) at 420 nm

2. Preparation the calibration curve of *N*-acetyl-D-glucosamine for HPLC analysis

Calibration curve of GlcNAc was made by varying the concentration and measuring the peak area by HPLC.

Table A2 The concentration of standard solution of GlcNAc and peak area.

Standard No.	Conc. GlcNAc (mM)	Peak Area (mV*Sec)
1	0.12	51.953
2	0.21	92.341
3	0.59	251.211
4	1.07	453.525
5	2.36	1007.929
6	3.20	1383.540

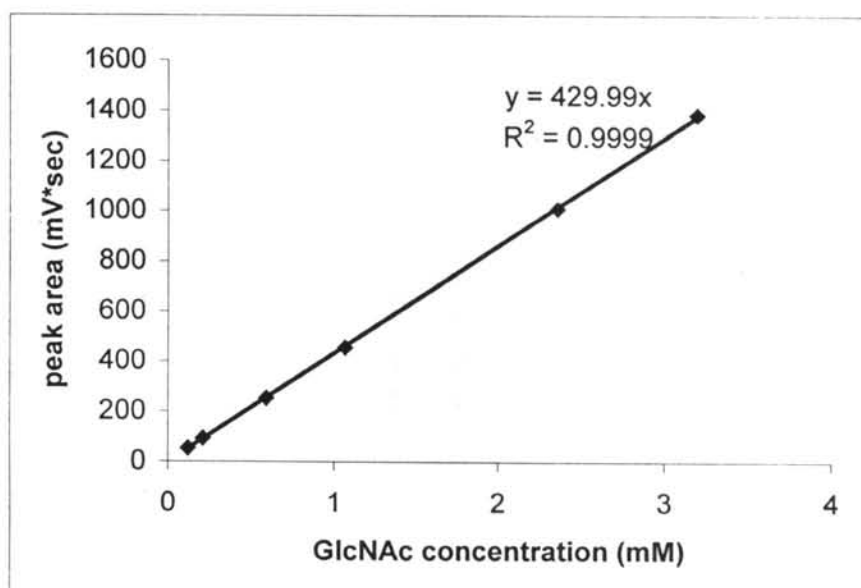


Figure A2 Correlation between concentration of standard *N*-acetyl-D-glucosamine and peak area by HPLC.

3. Preparation the calibration curve of *N,N'*-diacetylchitobiose for HPLC analysis

Calibration curve of (GlcNAc)₂ was made by varying the concentration and measuring the peak area by HPLC.

Table A3 The concentration of standard solution of (GlcNAc)₂ and peak area.

Standard No.	Conc. GlcNAc (mM)	Peak Area (mV*Sec)
1	0.06	37.613
2	0.11	100.833
3	0.30	192.349
4	0.55	391.312
5	1.20	776.494
6	1.65	1191.126

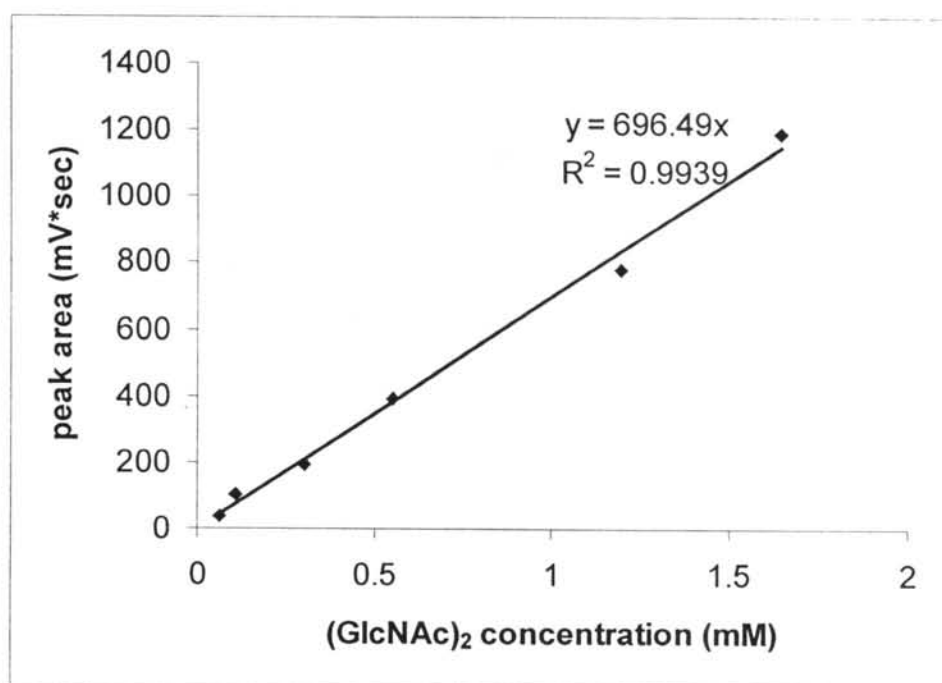


Figure A3 Correlation between concentration of standard *N,N'*-diacetylchitobiose and peak area by HPLC.

APPENDIX B

1. Precipitation of standard GlcNAc by using absolute ethanol, acetone and acetonitrile

Table B3 Precipitation of standard GlcNAc by absolute ethanol at various ratios

H ₂ O:EtOH	Precipitate weight (mg)	%precipitate
1:1	77.1	54
	73.7	
	75.4	
1:2	96.4	69
	96.1	
	96.3	
1:5	115.5	82
	114.9	
	115.2	
1:10	107.6	75
	101.6	
	104.6	
1:15	111.8	80
	112.0	
	111.9	

Various volume of EtOH was added into saturated standard GlcNAc (140 mg) solution (0.5 mL) in water. The average value (bold font) was plotted in the graph **Table 3.1**.

Table B4 Precipitation of standard GlcNAc by acetone at various ratios

H ₂ O:acetone	precipitate weight (mg)	%precipitate
1:1	9.3	9.6
1:2	8.1	8.3
1:4	53.0	54
1:8	57.8	59
1:15	67.6	69

Various volume of dried acetone was added into saturated standard GlcNAc (97 mg) solution (0.5 mL) in water.

Table B5 Precipitation of standard GlcNAc by acetonitrile at various ratios

H ₂ O:acetonitrile	precipitate	
	weight (mg)	%precipitate
1:1	0	0
1:2	0	0
1:4	12.8	13
1:8	18.5	19
1:15	19.4	20

Various volume of dried acetonitrile was added into saturated standard GlcNAc (97 mg) solution (0.5 mL) in water.

2. Purification of GlcNAc and (GlcNAc)₂ by precipitation technique

Table B6 Amount of GlcNAc in precipitate and supernatant obtained by absolute EtOH precipitation.

H ₂ O:EtOH	precipitate		supernatant	
	weight (mg)	%purity	weight (mg)	%purity
1:1	19.9	55	105.9	77
	19.5	48	109.0	69
	19.7	52	107.5	73
1:2	17.0	66	113.6	72
	18.6	66	113.4	73
	17.8	66	113.5	73
1:5	13.1	49	115.2	73
	14.9	59	115.2	74
	14.0	54	115.2	74
1:10	18.8	64	109.8	79
	18.9	67	109.8	78
	18.9	66	109.8	79
1:15	15.3	41	113.6	78
	15.2	49	112.5	80
	15.3	45	113.1	79

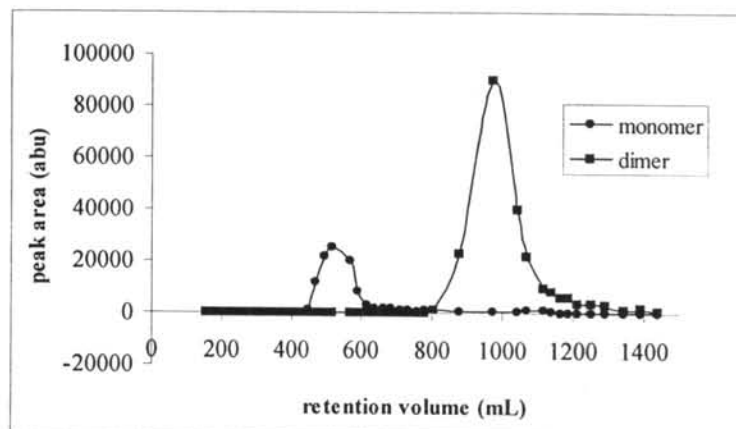
Various volume of EtOH was added into saturated standard GlcNAc (158 mg) solution (0.5 mL) in water. The average value (bold font) was plotted in the graph **Table 3.2**.

Table B7 Amount of (GlcNAc)₂ in precipitate and supernatant obtained by absolute EtOH precipitation.

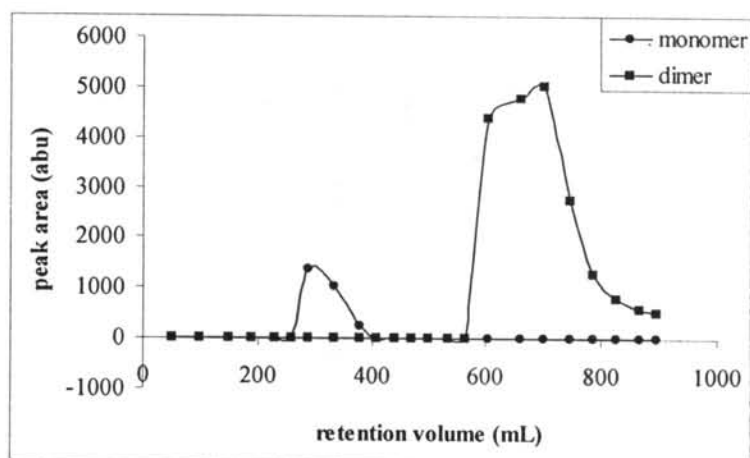
H ₂ O:EtOH	precipitate		supernatant	
	weight (mg)	%purity	weight (mg)	%purity
1:1	14.4	44	92.8	47
	15.5	42	94.3	47
	15.0	43	93.6	47
1:2	16.2	38	94.8	43
	16.0	35	93.4	48
	16.1	37	94.1	46
1:5	14.0	30	86.1	53
	14.0	28	89.5	51
	14.0	29	87.8	52
1:10	14.5	23	94.6	49
	14.7	20	92.8	55
	14.6	22	93.7	52
1:15	18.0	27	86.2	57
	18.3	28	86.0	60
	18.2	28	86.1	59

Various volume of EtOH was added into saturated standard GlcNAc (106 mg) solution (0.5 mL) in water. The average value (bold font) was plotted in the graph **Figure3.2**.

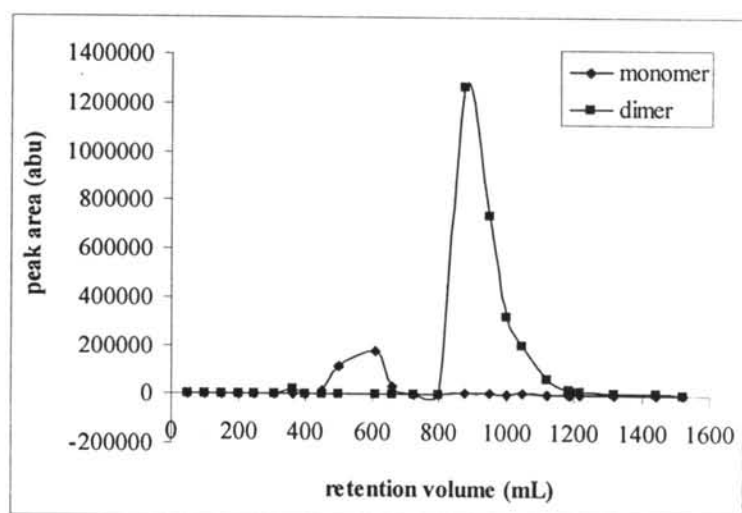
3. Purification of (GlcNAc)₂ by activated charcoal column: determination of loading capacity of charcoal.



(a)



(b)



(c)

Figure A1 chromatogram of GlcNAc and (GlcNAc)₂ from activated charcoal column with loading sugars of 0.44 g (a), 0.89 g (b) and 1.86 g (c)

VITAE

Miss Thitima Maneekul was born on June 19th, 1981 in Nakhornpathom, Thailand. She received a Bachelor Degree of Science, majoring in Chemistry from Chulalongkorn University, in 2002. Since 2003, she has been a graduate student studying Organic Chemistry as her major course at Chulalongkorn University. During her studies towards the Master's Degree, she was awarded a teaching assistant scholarship by the Faculty of Science during 2003-2004.

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