

APPARATUS AND EXPERIMENTAL METHOD

3.1 Reagent and materials

- 3.1.1 Commercial film grade polyethylene Both LDPE and HDPE used in this study is identified as Polene grade JJ4324 and A3355 respectively supplied by Thai Petrochemical Industry Co.,Ltd. (TPI). Physical properties of the materials are listed in Table 3.1.
- 3.1.2 Photosensitizers Most of additives were divided into two groups which are presented below:
- 3.1.2.1 Aromatic ketones have an ultraviolet absorption maximum between 330 and 400 millimicrons [16], as listed in Table 3.2.
- 3.1.2.1 Aromatic diketones have an ultraviolet absorption maximum between 280 and 330 millimicrons [16], as listed in Table 3.2.
 - 3.1.3 Solvent decahydronaphthalene (purity >98 %)
 - 3.1.4 Heat transfer medium silicone oil

Table 3.1 Physical properties of Polene JJ4324 (LDPE) and A3355 (HDPE)

Physical property	Test method	Unit	JJ4324 (LDPE)	A3355 (HDPE)
Melt index (2.16kg/190°C)	ASTM D1238	g/10 min	5.50	0.06
Melt index (5 kg/190°C)	ASTM D1238	g/10 min	-	0.25
Density	ASTM D1505	g/cm ³	0.922	0.950
Tensile strength	ASTM D638	N/mm ²	>11.0	>40
Yield strength	ASTM D638	N/mm ²	>11.0	>28
Ultimate elongation	ASTM D638	%	>600	>1700
Ball indentation hardness	DIN 53456	N/mm ²	-	>45
Notched impact strength	DIN 53453	mJ/mm ²	-	>12
Vicat softening				
temperature	ASTM D1525	°C	95	_
Haze	ASTM D1003	% max.	<6.5	-
Gloss	ASTM 2457	% min.	>95	-
Impact strength	ASTM D1709B	g	>160	-

Source: Thai Petrochemical Industry Co., Ltd.

Table 3.2 The characterization and properties of photosensitizers [17]

Photosensitizer	Chemical structure	Character	Melting point, °C	Boiling point, °C
Aromatic ketones - benzophenone	O'O	white prisms	48-49	305
- 4-methoxybenzo- phenone	сн3	white prisms	61-64	355
- thioxanthone		yellow needles	210-213	371-373 (at 715 mm)
Aromatic diketones - anthraquinone		yellow needles	283–287	379–381
- 2-methylanthra- quinone	0 CH3	yellow powder	, 177	236-238
- 2-tert-butyl- anthraquinone	CH3	yellow powder	98-100	-

3.2 Apparatus

- 3.2.1 Film blowing machine with extruder Model PE-45-AW from Siam Plastic Machinery Co., Ltd.
 - 3.2.2 Micrometer
 - 3.2.3 Exposure equipments, as recommended in ASTM D1435.
 - 3.2.4 Weathering cabinet
- 3.2.5 Lamp medium pressure mercury vapour lamp (Phillips 125 W) emitting a high degree of ultraviolet radiation



Figure 3.1 Medium pressure mercury vapour lamp

Table 3.3 Absolute spectrum power distribution

Absolute spectral power distribution

Wavelength	nm	248,2	253,7	265,3	269.9	257,3	280.4	289.4	296,7	302.5	313,0	334.1	365,5	404.7/	435.8	491.6	546.1	576.0
Lines	w	0,5	2,5	1,1	0,2	0,2	0,5	0,3	0.9	1,4	3,1	0.4	5.1	1.8	3.5	0,2	4.4	3.5
Region	nm			240-2	80	No.			280-31	15		315	-400			400-600		
Lines Continuum Total	W W W			4.5 4.4 8.9					6.2 2.5 8.7				5.5 2.3 7.8			13.4 1.5 14.9		

- 3.2.6 Tensile tester using an Instron universal testing machine Model 1123 equipped with pneumatic grips by selecting load 5 kg
- 3.2.7 Fourier Transform Infrared Spectroscopy:
 Perkin-Elmer model 1720

3.2.8 Viscosity analysis apparatus, according to ASTM D1601
- Ubbelohde viscometer, K=0.01

3.3 Sample preparation

The samples were prepared by mixing 0.5% by weight of each photosensitizers with polyethylene in a feed hopper and then blow-extruded by having temperatures at zones for LDPE and HDPE as follow:

Table 3.4 Temperatures at zones in the extruder

Feed zone (°C)		Compression	Metering	Die	
		zone (°C)	zone (°C)	zone (°C)	
LDPE	150	155	155	150	
HDPE	180	200	200	210	

The thickness of films was 0.035+0.005 mm.

3.4 Sample irradiation

3.4.1 Natural exposures

Outdoor weathering of PE films was carried out, according to ASTM D1435. The natural exposures were carried out at Bangkok, Thailand which is located at latitude 13° 44'N and longitude 100° 34'E. The exposed surfaces of samples were mounted on ASTM 45° racks, facing the equater as shown in Figure 3.2. The tests were started in September 1990 to February 1991 for 6 months. The meteorological and radiation data at the site were collected from The Monthly Report of the Climatology Division, Meteorological Department. The collected data, including averages of temperature, % relative humidity, total radiation, rainfall amount, are listed in Table 3.5 and shown

in Figure 3.4-3.5.

3.4.2 Irradiation using medium pressure mercury vapour lamp

The samples were irradiated in an UV aging cabinet equipped with a medium pressure mercury vapour lamp (HPK 125 W), as shown in Figure 3.3, and fixed on the sample holders at a distance of 15 cm around the light source. The temperature and % relative humidity were recorded for both LDPE and HDPE during irradiation, as listed in Table 3.6 and shown in Figure 3.6-3.7. When sampling, the PE-film samples were stored in the dark until testing.



Figure 3.2 Exposure equipment





Figure 3.3 Irradiated cabinet with medium pressure mercury lamp

Table 3.5 Meteorological data

	Month	Temperature °C	% R.H.	Rainfall amount, mm	Radiation MJ/M ²
1990 Septemb	September	er 28.1	76.9	257.3	21.62
	October	27.7	78.4	402.3	17.06
	November	27.2	71.7	22.6	17.12
	December	26.5	64.6	0.0	23.48
1991	January	28.0	70.5	3.0	16.67
	February	28.0	66.2	7.1	34.73

Table 3.6 Temperature and % R.H. in UV aging cabinet

Polyethylene	Time,	Temperature,	% R.H.
LDPE	5	35.0	41
	10	33.5	46
	15	32.5	68
	20	34.0	48
	25	31.5	52
	30	35.0	41
HDPE	2	32.0	69
	4	34.0	39
	6	34.5	53
	8	35.5	45

Note: After exposure, the polyethylene films were stored in the dark place until testing.

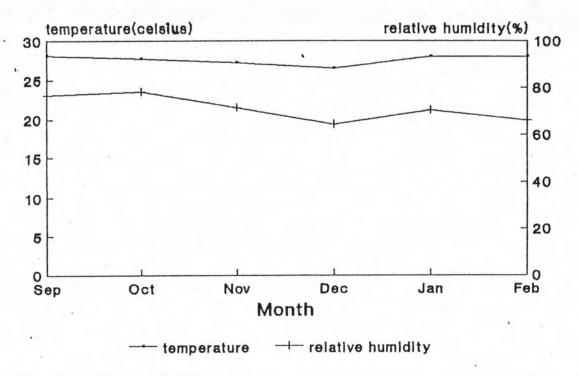


Figure 3.4 Average monthly temperature and relative humidity of Bangkok, Thailand

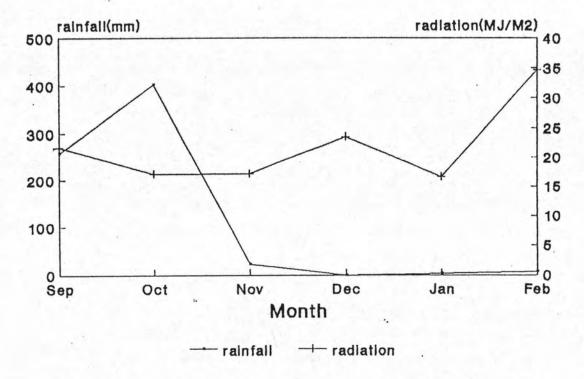


Figure 3.5 Total monthly rainfall and radiation of Bangkok,
Thailand

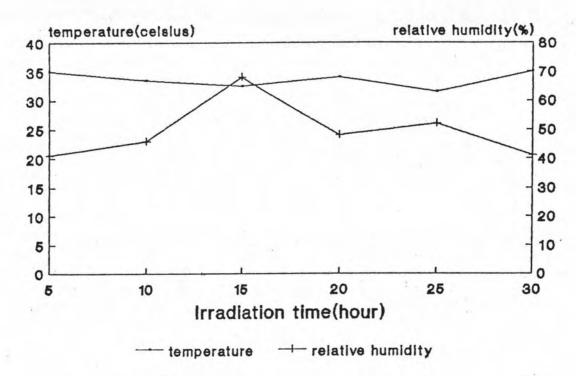


Figure 3.6 Average hourly temperature and relative humudity for LDPE during irradiation with HPK 125 W

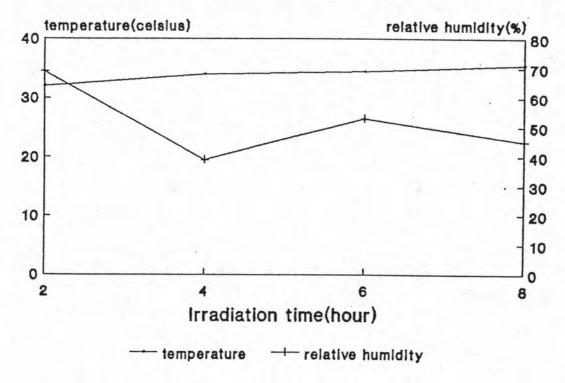


Figure 3.7 Average hourly temperature and relative humudity for HDPE during irradiation with HPK 125 W

3.5 Mechanical measurement

The samples were removed at interval period for testing. According to the ASTM D882, The specimens for tensile testing, cut out along the machine direction, were strips 25.4 mm wide with a guage length of 50 mm. A cross head speed of 100 mm/min was used in the test condition at 27±2 °C and 65±2 % relative humidity, and an average of five specimens was considered as representative value. The resulting stress-strain traces were then determined as an average.

3.6 FTIR measurements

on a FTIR Spectrophotometer and were used for estimating the changes in carbonyl group C=O in the chain at 1715 cm⁻¹ and vinyl group -CH₂-CH₂=CH₂ at 909 cm⁻¹. The polyethylene band at 2019 cm⁻¹ served as an internal standard to which the absorbances of the other bands were related. The results were expressed by carbonyl and vinyl index definded as

$$I_{co} = A_{1715}/A_{2019}$$
 (3.1)

$$I_{viny1} = A_{909}/A_{2019}$$
 (3.2)

where

I_{co} = Carbonyl index

I viny1 = Vinyl index

A₉₀₉ = Absorbance of vinyl group

 A_{1715} = Absorbance of carbonyl group

 A_{2019} = Internal standard band in PE chain

respectively

3.7 Gel content measurements

"Gel content" measurements were used to estimate the extent of crosslinking in irradiated film samples. Samples were weighed, extracted with hot decahydronaphthalene for 1 hour, and then dried under vacuum at 65 °C. The gel fraction (%) was estimated from equation (3.2)

Gel fraction (%) =
$$(W/W_o)$$
x100 (3.2)

where W_o = the initial weight of the polymer
W = the weight of the insoluble polymer

3.8 Molecular weight measurements

Molecular weights of PE were determined by the viscosity method, using decahydronaphthalene solutions and calculated from the Mark-Houwink-Sakurada equation:

$$[N] = kM^a$$
 (3.3)

where [] = intrinsic viscosity

M = viscosity average molecular weight

k,a = constant that can be obtained from the
 literature for a given polymer-solvent

system [18], as listed in Table 3.6

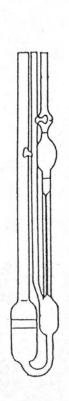
Table 3.7 Constants for viscosity molecular weight calculation [18]

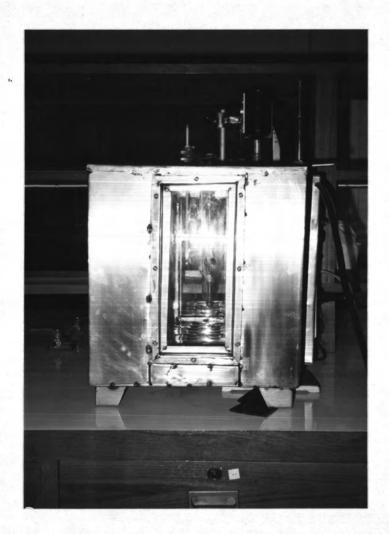
PE	Temperature, °C	kx10 ³ [ml/g]	a
LDPE	70	38.73	0.738
HDPE	135	62.00	0.700

Typical procedure for viscosity determination

Approximately 15 ml of decahydronaphthalene transferred by syringe into an Ubbelohde viscometer which is permanently positioned in the oil bath which was kept at constant given temperature until the solution attained thermal equililbrium (about 5 min.). The liquid level was brought to approximately 10 mm above the upper graduation mark in the viscometer capillary. As the meniscus passed this point, the timer was started and the interval for the solution to drain to the lower mark on the capillary was tuned (see figure 3.7). The efflux time of the solution was measured at least three times. Three consecutive readings should agree within 0.2 second. The solution was then removed from viscometer.

The solution from 3.7 was made up to 50 ml mark with solvent maintained at given temperature by means of a syringe and the flasks were shaken once again. In the same manner as decahydronaphthalene, three consecutive efflux times of the first PE solution in the viscometer were recorded. The first solution was then diluted by adding 15, 10, 5 ml of the first solution into the three 25 ml volumetric flasks, respectively and then all solution were made up to the 25 ml mark with the solvent maintained at given temperature. The efflux time of each times and the average efflux time of undoped HDPE sample and HDPE samples doped with one photosensitizer in a concentration of 0.5 % by weight before and after exposure to the light source were obtained.





(a) Ubbelohde viscometer

(b) Oil bath

Figure 3.8 Viscosity measuring apparatus