Catalytic Extrusion of Polylactide/Ethylene Vinyl Alcohol Bioplastic Film



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ABSTRACT

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The ring-opening polymerization of lactide was generated by a continuous single-step reactive extrusion process in the presence of 2-ethylhexanoic acid tin(II) salt, Sn(Oct)₂, as a catalyst to obtain high molecular weight polylactide (PLA). For good practical applications of PLA, the softness of the PLA was modified via the graft copolymerization from poly(ethylene-co-vinyl alcohol) EVOH, which is a biocompatible, flexible and soft random copolymer. To investigate the chemical structure of the graft copolymer, the products were characterized by FTIR. The results show that the strong absorption emerged at 1740 cm⁻¹ in the spectra of EVOH-g-PLA and pure PLA was identical, which assigned to carbonyl (C=O) in PLA. Therefore, these results could be confirmed that the ring-opening polymerization of lactide with EVOH by using catalytic extrusion was carried out successfully. Furthermore, the EVOH-g-PLA copolymers gave the number average molecular weight (M_w) ranging from 24.5×10^4 to 36.6×10^4 g/mol. The amount of graft copolymer and the grafting degree showed a maximum at catalyst content around 0.5 wt%. The optimized LA/EVOH content and the screw speed were 50/50 wt% and 40 rpm, respectively. Furthermore, the EVOH-g-PLA copolymers were fabricated into bioplastic films by compression moulding technique for morphological study by SEM and mechanical testing. The elongation of grafted PLA were improved significantly compared to pure PLA. The tradeoff included the reduction of tensile strength.

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บทคัดย่อ

พัชรกมน หนูเอียด : ฟิล์มธรรมชาติ โพลีแลคไทค์/เอทธิลึนไวนิลแอลกอฮอล์ สังเคราะห์ โดยเทกนิคแกททาไลติก เอกทรูชั่น (Catalytic Extrusion of Polylactide/Ethylene Vinyl Alcohol Bioplastic Film) อ. ที่ปรึกษา : รศ. ดร. รัตนวรรณ มกรพันธุ์ 107 หน้า

ปฏิกิริยาพอลิเมอร์ไรเซชั่นแบบเปิควงของแลกไทค์เกิดขึ้นโคยใช้เทคนิครีแอกทีพเอกท รูชั่นซึ่งสามารถทำปฏิกิริยาในขั้นตอนเดียว โดยใช้กรดเกลือของธาตุสแตนนัสเป็นตัวเร่งปฏิกิริยา เพื่อทำการสังเคราะห์โพลีแลกไทด์ที่มีน้ำหนักโมเลกุลสูง แต่เนื่องด้วยความเปราะของโพลีแลก ้ไทค์ซึ่งเป็นข้อจำกัดในการใช้งานหลายประเภท โพลีแลกไทค์จึงจำเป็นอย่างยิ่งต่อการปรับปรุง ้สมบัติทางด้านความอ่อนนุ่มด้วยวิธีการพอลิเมอร์ไรเซชั่นแบบกราฟ โคยเลือกใช้เอทธีลีนไวนิล แอลกอฮอล์ โคพอลิเมอร์ เป็นสายโซ่หลัก เนื่องจากเอทธีลืนไวนิลแอลกอฮอล์ โคพอลิเมอร์ มี ้สมบัติทางด้านความยุืดหยุ่นและเข้ากับวัสดุธรรมชาติด้วยกัน ได้ดี เพื่อศึกษาโครงสร้างทางเคมีของ กราฟโคพอลิเมอร์ที่ได้ เครื่องฟรูเรียทรานสฟอร์มสเปกโทรสโคปีถูกใช้ในการวิเคราะห์ พบว่าการ ปรากฏของพืกที่ 1740 cm⁻¹ แสดงถึงหมู่การ์บอนิลในโพลีแลกไทด์ซึ่งเป็นสายโซ่กิ่งในกราฟโค พอลิเมอร์ ซึ่งสามารถยืนยันได้ว่าปฏิกิริยาพอลิเมอรไรเซชั่นแบบเปิดวงของแลกไทด์โดยใช้หมู่ไฮ ครอกซิลในเอทธีลืนไวนิลแอลกอฮอล์ โคพอลิเมอร์เป็นส่วนที่ทำให้เกิดปฏิกิริยาสามารถประสบ ้ความสำเร็จได้ด้วยเทคนิคแคททาไลติกเอกทรูชั่น ซึ่งกราฟโคพอลิเมอร์ที่สังเคราะห์ได้มีน้ำหนัก โมเลกุลเฉลี่ยโดยน้ำหนักประมาณ 24.5x10⁴ ถึง 36.6x10⁴ กรัม/โมล นอกจากนี้ ยังมีการศึกษา ถึงผลกระทบของอัตราส่วนระหว่างแลกไทด์และเอทธีลีนไวนิลแอลกอฮอล์ โคพอลิเมอร์ ความเร็วรอบหมุนของสกรู และความเข้มข้นของตัวเร่งปฏิกิริยา พบว่าปริมาณของกราฟโคพอลิ เมอร์และประสิทธิภาพในการกราฟให้ค่าสูงสุด เมื่อใช้อัตราส่วนของแลคไทค์ต่อเอทธีลีนไวนิล แอลกอฮอล์ประมาณ 50/50 เปอร์เซ็นน้ำหนัก ความเร็วรอบหมุนเท่ากับ 40 รอบต่อนาที และ ความเข้มข้นของตัวเร่งปฏิกิริยาประมาณ 0.1 เปอร์เซ็นน้ำหนัก และคุณสมบัติเชิงกลของกราฟโค พอลิเมอร์ได้ทำการศึกษาโดยการทคสอบด้วยเครื่องดึงยืด พบว่าสมบัติการยืดออกเพิ่มขึ้น แลกกับ การลดลงของความแข็งแรง

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ABBREVIATIONS

LA Lactide

PLA Polylactide

EVOH Ethylene vinyl alcohol copolymer

EVOH-g-PLA Ethylene vinyl alcohol copolymer graft polylactide