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PROCESS MODELING, DYNAMIC DATA RECONCILIATION AND CONTROL OF ACETYLENE HYDROGENATION REACTORS

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A Thesis Submitted in Partial Fulfillment of the Requirements for the Degree of Master of Engineering in Chemical Engineering Department of Chemical Engineering Graduate School Chulalongkorn University Academic year 1998 ISBN 974-639-506-8

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งานวิจัยนี้เสนอแบบจำลองทางคณิตศาสตร์ของกระบวนการอะเซทีลีนไฮโดจีเนชัน โดยเขียนบนโปรแกรมสปีดอัฟ (SPEEDUP) ซึ่งเป็นโปรแกรมสภาวะจำลองแบบไดนามิก เครื่องปฏิกรณ์อะเซทีลีนไฮโดจีเนชันถูกจำลองด้วยชุดลำดับเครื่อง ปฏิกรณ์แบบถังกวนต่อเนื่อง ข้อมูลที่นำมาใช้ในการสร้างแบบจำลองได้จากข้อมูลโรงงานที่ผ่านการปรับให้สอดคล้องของข้อมูล ด้วยตัวแปรเกินจากสมการสมดุลมวลสารและพลังงาน จำนวนชุดลำดับเครื่องปฏิกรณ์แบบถังกวนต่อเนื่องที่เหมาะสมเท่ากับยี่สิบ ถัง ซึ่งให้ผลการเปรียบเทียบการเปลี่ยนแปลงของอุณหภูมิภายในครื่องปฏิกรณ์ และผลการทำนายค่าความเข้มข้นของอะเซทีลีน กับข้อมูลโรงงานเหมือนที่สุด สามารถเขียนชุดสมการการเกิดปฏิกิริยาของกระบวนการได้หกแบบ และทำการเลือกชุดสมการที่ เหมาะสมจากการพิจารณาเปรียบเทียบความเข้มช้นไฮโดเจนกำรังหนึ่ง, ไฮโดเจนมีการแตกตัวก่อนเกิดปฏิกิริยา, ผลิตภัณต์ที่เกิดไม่ถูกดูดชับบนตัว เร่งปฏิกิริยา, และความสามารถของตัวเร่งปฏิกิริยาขึ้นกับผลรวมของปริมาณอะเซทีลีนที่เข้าระบบ ให้ผลเปรียบเทียบผลการทำนาย จากแบบจำลองกับข้อมูลโรงงานดีที่สุด

การปรับให้สอดคล้องของข้อมูลแบบไดนามิกที่สร้างใช้สมการสมดุลมวลสารและพลังงานเป็นเงื่อนไขของการปรับ พบว่าค่าเวลาย้อนหลังที่ใช้ในการปรับข้อมูลที่ดีที่สุดเท่ากับสิบเท่าของช่วงเวลาในการเก็บข้อมูล สามารถลดค่าเบี่ยงเบนมาตราฐาน ของข้อมูลโรงงานได้ 40-70%, สามารถลดค่าเบี่ยงเบนมาตราฐานของข้อมูลที่ได้จากการจำลองได้ 90% และทดสอบโปรแกรมด้วย การใช้ข้อมูลที่มีสัญญาณรบกวน และข้อมูลที่ปรับให้สอดคล้องแล้วในการประมาณค่าตัวแปรของแบบจำลองใหม่ ผลการทำนาย อุณหภูมิของแบบจำลองที่ใช้ข้อมูลที่ปรับให้สอดคล้องแล้วคาดเคลื่อนไป 0.16% และผลการทำนายอุณหภูมิของแบบจำลองที่ใช้ ข้อมูลมีสัญญาณรบกวนคาดเคลื่อนไป 2.13%

แบบจำลองทางคณิตศาสตร์ของกระบวนการอะเซทีลีนไฮโดจีเนชันที่สร้างถูกนำมาใช้ในการออกแบบระบบควบ คุมแบบไดนามิเมทริก และการทดสอบผลการควบคุม ตัวควบคุมที่ดีที่สุดคือ ตัวควบคุมที่มีค่าขั้นการทำนายผลล่วงหน้าเท่ากับ สามเท่าของช่วงเวลาในการเก็บข้อมูล และมีค่าขั้นการควบคุมสำหรับผลในอนาคตเท่ากับสองเท่าของช่วงเวลาในการเก็บข้อมูล ตัว ควบคุมที่ได้มีประสิทธิภาพในการควบคุมกระบวนการ ผลการควบคุมให้ค่าผลร่วมความคาดเคลื่อนจากค่ากำหนด (set point) เท่ากับ 3.33% ของค่าจากตัวควบคุมแบบพีไอดี ช่วยลดปริมาณการสูญเสียเอธธีลีนได้ 80-98% เมื่อเทียบกับตัวควบคุมแบบพีไอ ดี

ภาควิชา	วิศวกรรมเคมี	
ສາ ນາວິชາ	วิศวกรรมเคมี	
ปีการศึกษา		

C817340 : MAJOR CHEMICAL ENGINEERING KEY WORD: ACETYLENE HYDROGENATION / DATA RECONCILIATION / PROCESS MODELING

TARAVIPA SAUROD : PROCESS MODELING, DYNAMIC DATA RECONCILIATION AND CONTROL OF ACETYLENE HYDROGENATION REACTORS. THESIS ADVISOR : MONTREE WONGSRI, D.Sc., THESIS CO-ADVISOR : PAISAN KITTISUPAKORN, Ph. D., 245 pp. ISBN 974-639-506-8

The mathematical model of the industrial acetylene hydrogenation process is developed and formulated on SPEEDUP program which is a dynamic simulation program. The fixed bed acetylene hydrogenation reactor is modeled as the CSTRs connected in series. The data using in modeling, the actual data from the ethylene plant are reconciled first using material and energy balance redundancy. The best number of CSTRs is found to be twenty which is given the best agreement in temperature profile and the output acetylene concentration of each bed. The six kinetic models with the different reaction mechanism are derived and selected for the best one by comparing their predicted outputs with the actual data. The kinetic model which its reaction mechanism are the reaction rate depends on the first order of H_2 , H_2 break to free atoms before react, the product are not adsorbed on the catalytic surface, and the catalytic activity depend on the accumulation of the inlet acetylene, gives the best agreement of the predicted result with the actual data.

The dynamic data reconciliation using the material and energy balance constrains of process is performed. The best time history horizon is found to be ten steps. It can reduce the standard deviation of the actual data and the simulated data to about 40-70% and 90% in series. The noised data and the reconciled data are then used to obtain the new parameters of the model, i.e. the reconstruction of the model. The noised model gives 2.13% of temperature error and the reconciled model gives 0.16% of temperature error.

The obtained model of acetylene hydrogenation reactor is used to demonstrate the design, implementation, and performance of Dynamic Matrix controller by simulation. The Dynamic Matrix controller is tuned for best performance with the control horizon, U=2, the prediction horizon, V=3. The integral error of Dynamic Matrix controller is only 3.33 % of the PID controller's error. The ethylene loss is reduced by 80-98% by using Dynamic Matrix controller over PID controller. The degree of the benefit of using the Dynamic Matrix Control is illustrated.

ภาควิชา	ภาควิชาวิศวกรรมเคมี	ลายมือชื่อนิสิต <u>รรริภา</u> เชื่อรงก
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NOMENCLATURE

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C_{total}	=	the total concentration of sites, sites/weight
C,	=	the concentration of free sites
C _{AS}	=	the A specie adsorbed sites
C _{B.S}	=	the A specie adsorbed sites
CAc	=	acetylene concentration, mol/M ³
C_{Eth}	=	ethylene concentration, mol/M ³
$C_{_{H\!2}}$	=	hydrogen concentration, mol/M ³
C _{MA}	=	methyl acetylene concentration, mol/M ³
C_{PD}	=	propadiene concentration, mol/M ³
C _{co}	=	carbonmonoxide concentration, mol/M ³
C _E	=	ethane concentration, mol/M ³
C_{Pr}	=	propylene concentration, mol/M ³
$C_{M^{o}}$	=	methane concentration, mol/M ³
$C_{_{BD}}$	=	butadiene concentration, mol/M ³
Ср	=	heat capacity of feed stream
C_{m}	=	heat capacity of the reacting system including catalyst
D	=	fluid density
f	=	differential equation constraints
F _i	=	outlet volumetric flow rate of the feed at i th stage, M^3/hr
g	=	inequality constraints including simple upper and lower
		bounds
h		algebraic equality constraints
Н	=	the history horizon time
Hr _{Ac}	=	heat of acetylene hydrogenation
Hr_{Eth}	=	heat of ethylene hydrogenation
$Hr_{MA} =$	he	at of methyl acetylene hydrogenation

Hr _{PD}	=	heat of propadiene hydrogenation
i	=	i th stage
K _{Ao}	=	adsorption equilibrium constant of acetylene
K _{Eth}	=	adsorption equilibrium constant of ethylene
K _{H2}	=	adsorption equilibrium constant of hydrogen
K _{co}	=	adsorption equilibrium constant of carbonmonoxide
k	=	reaction rate constant
M _i	=	mass of reacting system at i th stage including catalyst
N _F	=.	the degrees of freedom
N _v	=	the total number of variables (unspecified inputs plus
		outputs)
N _E	=	the number of independent equations
NV	=	the number of variables in the type section
NI	=	the number of input stream variables
NE	=	the number of equations
NS	=	the number of set variables that should expect to set for a
		simulation.
NDYN	=	the number of the dynamic run
NSS	=	the number of the steady state run
NT	=	the number of the time step of the actual input-output data
NM	=	the number of the the output variables
P _i	=	the elements of the projection vector
Q_i	=	heat loss at i th stage
R _{Ac}	=	rate of acetylene hydrogenation
R_{Eth}	=	rate of ethylene hydrogenation
R _{MA}	=	rate of methyl acetylene hydrogenation
R_{PD}	=	rate of propadiene hydrogenation
R _{co}	=	adsorption rate of carbonmonoxide
t _c	=	the current time

=	time step size
=	the half-life time
=	temperature
=	the control horizon
=	the prediction horizon
=	the variance-covariance matrix
=	volume of one CSTR stage, M^3
=	the weight fraction
= .	discrete measurement
=	estimate function
=	lower limit of y
=	upper limit of y
=	the estimated value of estimation equation at time t_j
=	the measured value at time t_j
=	the actual output data
Ŧ	the predic output
=	the catalyst activity
=	objective function equation
	measurement noise standard deviation

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