

REFERENCES



- Alawi, M. A., Khalill, F., and Sahili, I. 1994. Determination of trihalomethanes produced from the chlorination of water as a function of humic acid content. Arch. Environ. Contam. Toxicol. 26: 381-386.
- Amy, G., Siddiqui, M., Ozekin, K., Zhu, H.W., and Wang, C. 1998. Empirically-based models for predicting chlorination and ozonation by-products: Trihalomethanes, haloacetic acid, chloral hydrate, and bromate. Cincinnati, Ohio, US Environmental Protection Agency (EPA-815-R-98-005).
- Amy, G.L., Sierka, R.A., Bedessem, J., Price, D., and Tan, L. 1992. Molecular size distribution of dissolved organic matter. J.AWWA. 75(6) : 67.
- APHP, AWWA, and WPCF. 1995. Standard Methods for the examination of water and wastewater. 19th ed., USA
- AWWARF .1991. Disinfection by-products database and model project. AWWARF. Denver, Colorado.
- Barrett, S.E., Krasner, S.W., and Amy, G.L.. 2000. Natural organic matter and disinfection by-products characterization and control in drinking water. American Chemical Society. Washington, DC.
- Bell-Ajy, K., Abbaszadegan, M., Ibrahim, E. , Verges, D., and LeChevallier, M.2000. Conventional and optimized coagulation for NOM removal. J.AWWA. 92(10): 44-58.
- Carlson, M., and Hardy, D. 1998. Controlling DBPs with monochloramine. Effect of water quality conditions on controlling disinfection by-product with chloramines. J.AWWA. 90(2):95-106.
- Chen, R.C., and Yates, R.S., Krasner, S.W., and Liang, S. 1995. Bench -scale evaluation of the effects of seasonal change on TOC removal by enhanced coagulation. Proc. AWWA Ann. Conf. (Water Quality), Anaheim, Calif.
- Chen, W., Westerhoff, P., Leenheer, J.A., and Booksh, K. 2003. Fluorescence Excitation-Emission Matrix reginol intregation to quantity spectra for dissolved organic matter. Environ Sci. Technol. 37: 5701-5710.

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- Edzwald, J.K., and Laffin, A.J. 1983. Alum coagulation of humic substances. Prof. AWWA. Ann. Conf. Las Vegas, Nev.
- Edzwald, J.K., and Van Benschoten, J.E.1990. Aluminum coagulation of natural organic matter. Proc. 4th International Guthenburg Symposium on chemical treatment. Madrid, Spain, (October).
- Enhanced coagulation and enhanced precipitative softening guidance manual. 1999. EPA 815-R-99-012.
- Fearing, A.D., Goslan, H.E., Banks J., Wilson D., Hillis P., Campbell, T.A., and Parsons, A.S. 2004. Staged coagulation for treatment of refractory organics. Journal of Environmental Engineering. 975-982.
- Galapate, R., Baes, A., Ito, K., Iwase, K., and Okada, M. 1999. Trihalomethane formation potential prediction using some chemical functional groups and bulk parameters. Wat. Res. 33(11): 2555-2560.
- Gallard, H., Gunten, V.U. 2002. Chlorination of natural organic matter: kinetics of chlorination and of THM formation. Wat Res. 36:65-74.
- Ghazali, Z.B.Md. 1989. Trihalomethane formation in Bangkok water treatment plant, Thailand. Master's Thesis, Asian Institute of Technology Thailand.
- Hayes, M.H.B., MacCarthy, P., Malcolm, R.L., and Swift, R.S. 1989. The search for the structure: setting the scene, Humic substances II. In search of structure P 4-30. Chichester: John Wiley and Sons.
- Jandi, R., and Sollins, P. 1997. Water-extractable soil carbon in relation to the belowground carbon cycle. Biol. Fertil. Soils. 25:196-201.
- JiJi, D.R., Cooper, A.G., and Booksh, S.K. 1999. Excitation-emission matrix fluorescence based determination of carbamate pesticides and polycyclic aromatic hydrocarbons. Analytica. Chimica. Acta. 397: 61-72.
- John, C., 1998. Trihalomethanes and our water supply [online]. Available from: <http://www.southernstream.com>. Updated on 3/11/2002.
- Kaiser, K., Guggenberger, G., and Zech, W. 2001. Isotopic fractionation of dissolved organic carbon in shallow forest soils as affected by sorption. J. Soil Sci. 52:585-597.

- Kasuga, I., Nakajima, F., and Furumai, H. 2003. Analysis of dissolved organic matter and bacterial community in degradation of algal bloom by EEM and CR-DGGE, Journal Japan Society on Water Environment.26(3):171-174.
- Kavanaugh, M.C., Trussell, A.R., Cromer, J., and Trussell, R.R.1980. An empirical kinetic model of THM formation: application to meet the proposed THM standard. J. AWWA. 71(10): 578-582.
- Kim, H.M., and Yu, J.M.2005. Characterization of NOM in the Han River and evaluation of treatability using UF-NF membrane. Environ. Res. 96 (2), 176-185.
- Koch, B., Krasner, S.W., Scilimenti, M.J., and Schimpff, W.K. 1991. Prediction the formation of DBPs by the simulated distribution system. J.AWWA. 83(10): 62-70.
- Knocke, W.R., West, S., and Hoehn, R.C. 1986. Effects of low temperature on the removal of trihalomethane precursors by coagulation. J.AWWA. 78(4): 189.
- Krasner, S.W., and Amy, G. 1995. Jar test evaluations of enhanced coagulation. J.AWWA. 87(12): 93-107.
- Krasner, S.W., Amy, G., and Zhu, H.W. 1994. Use of some simple NOM characterization techniques to evaluate enhanced coagulation treatability. Proc. AWWA Enhanced Coagulation Research Workshop. Charleston, S.C.
- Kukkonen, J., McCarthy, J.F., and Oikari, A. 1990. Effects of XAD-8 fractions of dissolved organic carbon on the sorption and bioavailability of organic micropollutants. Arch. Environ. Contam. Toxicol. 19:551-557.
- Lange, A.L., and Kawczynski, E. 1978. Controlling organic: the contra costa county water district experience. J. AWWA. 70(11):63.
- Leenheer, J.A.1981. Comprehensive approach to preparative isolation and fractionation of dissolved organic carbon from natural waters and wastewaters. Environ. Sci. Technol. 15(5): 578-587.
- Leenheer, J.A., Noyes, T.I. 1984. A filtration and column-adsorption system for onsite concentration and fractionation of organic substances from large volume of water. U.S. Geological Survey Water Supply Paper 2230, U.S. Govt. Printing.

- Leenheer, J.A., Noyes, T.I., and Steer, H.A. 1982. Determination of polar organic solute in oil-shale report water. Environ. Sci. Technol. 16(10):714-723.
- Lin, C.F., Lin, T.Y., and Hao, O.J. 2000. Effects of humic substance characteristics on UF performance. Wat. Res. 34(4): 1097-1106.
- Lind, C.B. 1995. Experiments in TOC removal by polyaluminium hydroxychloride and enhanced coagulants. Proc. AWWA Ann Conf. Anaheim. Calif.
- Marhaba, T.F., and Pipada, N.S. 2000. Coagulation: Effectiveness in removing dissolved organic matter fractions. Environ. Eng. Sci.17(2):107-115.
- Marhaba, T. F., Pu,Y., and Bengraïne, K. 2003. Modified dissolved organic matter fractionation technique for natural water. J. Hazard Mater. B. 101: 43-53.
- Marhaba, T.F., and Van, D. 2000. The variation of mass and disinfection by-product formation potential of dissolved organic matter fractions along a conventional surface water treatment plant. J. Hazard. Mater, A. 74: 133-147.
- Marhaba, T.F., and Washington, M.B. 1998. Drinking Water Disinfection and by-products: history and current practice. Adv. Environ.Res. 2(1): 103-115.
- Matilainen, A., Lindqvist, N., Korhonen, S., and Tuhkanen, T. 2002. Removal of NOM in the different stages of the water treatment process. Environment International. 28:457-465.
- Miller, W.J., and Uden, P.C.1983. Characterization of nonvolatile aqueous chlorination products of humic substances. Environ Sci Technol.17(3): 150-152.
- Nieminski, E.C., Chaudhuri, S., and Lamoreaux, T. 1993. The occurrence of DBPs in Utah drinking waters. J. AWWA. 85(9): 98-105.
- Notification of the Ministry of Industry, No.332, B.E, 2521. 1978. Issued under the industrial product (1968), published in the Royal Government Gazette, vol 95 part 68, dated July 4, B.E. 2512 (1998).Office Washington, D.C.
- Owen, D.M., Amy, G.L. and Chowdhary, Z.K., Eds. 1993. Characterization of natural organic matter and its relationship to treatability.: AWWARF.
- Page, D.W., Van Leeuwen, J.A., Spark, K.M., Drikas, M., Withers, N., and Mulcahy, D.E. 2002. Effect of alum treatment on the trihalomethane formation and bacterial regrowth potential of natural and synthetic waters. Wat. Res. 36: 4884-4892.

- Peter, C.J., Young, R.J., and Perry, R. 1980. Fractions influencing the formation of haloforms in the chlorination of humic substances. Environ Sci Technol. 14(11):1391-1395.
- Peter, R. J., Wilson, L.B., Amy, G.L., and Brothers, K. 1993. Fate of organochlorine compounds during aquifer storage and recovery the Las Vegas experience. Groundwater. 31(3): 410-416.
- Qualls, R., and Haines, B.L. 1991. Geochemistry of dissolved organic nutrients in water percolating through a forest ecosystem. J.Soil Sci Soc Am. 55: 1112-1123.
- Randtke, S.J. 1988. Organic contaminant removal by coagulation and related process combinations. Research and Technology. 40-56.
- Reckhow, D.A., Bose, P., Bexbarua, B., Hesse, E.M., and McKnight, A.P. 1992. Transformations of natural organic material during preozonation. EPA Report. USEPA, Drinking Water Research Division Cincinnati, Ohio.
- Reckhow, D.A., Singer, P.C., and Malcolm, R.L. 1990. Chlorination of humic materials: by-product formation and chemical interpretation. Environ Sci Technol. 24(11):1655-1664.
- Reckhow, D.A., and Singer, P.C. 1985. Mechanisms of organic halide formation during fulvic acid chlorination and implications with respect to preozonation. In: Jolley RL, Bull RJ and Davis WP ed. Water chlorination: chemistry, environmental impact and health effects. pp 1229-1257. Chelsea, Michigan, Lewis Publisher, Inc.
- Reynold, D.T., and Richards, A.P. 1996. Unit operation and process in environmental engineering. 2nd ed., U.S: PWS publishing Company.
- Rook JJ. 1974. Formation of haloforms during chlorination of natural waters. Water treatment Exam. 23(2):234-243.
- Rook, J.J., Gras, A.A., Vander Heijden, B.G., and De Wee, J. 1978. Bromide oxidation and organic substitution in water treatment. J.Environ Sci Health.A. 13: 91-116.
- Siddiqui, M.S., and Amy, G.L. 1993. Factors affecting DBP formation during ozone-bromide reactions. J AWWA. 85(1): 63-72.

- Stevens, A.A., Slocum, C.J., Seeger, D.R., and Robeck, G.G. 1976. Chlorination of organics in drinking water. J AWWA. 68(11): 615.
- Stuart, M.E., Goody, D.C., Kinniburgh, D.G., and Klinck, B.A. 2001. Trihalomethane formation potential: a tool for detecting non-specific organic groundwater contamination. Urban water.3: 173-184.
- Summers, R.S., Hooper, S.M., Shukairy, H.M., Solarik, G., and Owen, D. 1996. Assessing DBP Yield: Uniform Formation Conditions. J. AWWA. 88: 6: 80.
- Thurman, E.M., and Malcolm, R.L.1981. Preparative isolation of aquatic humic substances. Environ.Sci. Technol. 15(4): 463-466.
- Trussell, R.R., and Umphres, M.D. 1978. The formation of trihalomethanes. J.AWWA. 70(11): 604.
- USEPA, 1996. Drinking water regulations and health advisories. EPA/883/B- 96/002.
- USEPA. 1998. National primary drinking water regulation: disinfection and disinfection by product (D/DBP), notice of data availability: proposed rule. Fed. Reg. 62:212:38668.
- USEPA. 1998 a. National primary drinking water regulations: Disinfectants and disinfection by products (D/DBP). Final rule. Fed. Reg. 63: 69389- 69476.
- USEPA. 1998 b. National primary drinking water regulations: Interim enhanced surface water treatment rule (IESWTR). Fed. Reg. 63: 69478.
- USEPA. 1999. Enhanced coagulation and enhanced precipitative softening guidance manual. Office of water (4607) [computer file]. Available from: <http://www.epa.gov> [2001 November 10]
- Volk, C., Bell, K., Ibrahim, E., Verges, D., Amy, G., and Lechevallier, M. 2000. Impact of enhanced and optimized coagulation on removal of organic matter and its biodegradable fraction in drinking water. Wat. Res. 34(12): 3247-3257.
- White, M.C., Thompson, J.D., Harrington, G.W., and Singer, P.C. 1997. Evaluating criteria for enhanced coagulation compliance. J AWWA. 89:5:64.

WHO. 1993. Guideline values for drinking water quality. Geneva:World Health Organization.

APPENDIX

Table A-1: The coagulation experiment result for removal DOC from Aung-Keaw Reservoir water

pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)
5.0/ 0	2.338	0.0	5.5/ 0	2.338	0.0	6.0/ 0	2.338	0.0
5.0/ 20	1.074	54.1	5.5/ 20	1.123	52.0	6.0/ 20	1.217	47.9
5.0/ 40	1.023	56.2	5.5/ 40	1.032	55.9	6.0/ 40	1.15	50.8
5.0/ 60	0.918	60.7	5.5/ 60	0.959	59.0	6.0/ 60	1.01	56.8
5.0/ 80	0.881	62.3	5.5/ 80	0.841	64.0	6.0/ 80	0.988	57.7
5.0/ 100	0.942	59.7	5.5/ 100	0.826	64.7	6.0/ 100	0.963	58.8

pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)
6.5/ 0	2.338	0.0	7.0/ 0	2.338	0.0
6.5/ 20	1.434	38.7	7.0/ 20	1.709	26.9
6.5/ 40	1.157	50.5	7.0/ 40	1.366	41.6
6.5/ 60	1.155	50.6	7.0/ 60	1.164	50.2
6.5/ 80	1.183	49.4	7.0/ 80	1.169	50.0
6.5/ 100	1.137	51.4	7.0/ 100	1.098	53.0

Table A-2: The coagulation experiment result for removal UV-254 from Aung-Keaw Reservoir water

pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)	pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)	pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)
5.0/ 0	0.1085	0.0	5.5/ 0	0.1085	0.0	6.0/ 0	0.1085	0.0
5.0/ 20	0.0265	75.6	5.5/ 20	0.0298	72.5	6.0/ 20	0.0372	65.7
5.0/ 40	0.0295	72.8	5.5/ 40	0.0237	78.2	6.0/ 40	0.0237	78.2
5.0/ 60	0.0208	80.8	5.5/ 60	0.0195	82.0	6.0/ 60	0.0209	80.7
5.0/ 80	0.0245	77.4	5.5/ 80	0.0177	83.7	6.0/ 80	0.0182	83.2
5.0/ 100	0.0206	81.0	5.5/ 100	0.0208	80.8	6.0/ 100	0.0179	83.5

pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)	pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)
6.5/ 0	0.1085	0.0	7.0/ 0	0.1085	0.0
6.5/ 20	0.0305	71.9	7.0/ 20	0.0634	41.6
6.5/ 40	0.0234	78.4	7.0/ 40	0.0369	66.0
6.5/ 60	0.0231	78.7	7.0/ 60	0.0276	74.6
6.5/ 80	0.0235	78.3	7.0/ 80	0.0290	73.3
6.5/ 100	0.0225	79.3	7.0/ 100	0.0234	78.4

Table A-3: The coagulation experiment result for removal SUVA from Aung-Keaw Reservoir water

pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)	pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)	pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)
5.0/ 0	4.641	0.0	5.5/ 0	4.641	0.0	6.0/ 0	4.641	0.0
5.0/ 20	2.467	46.8	5.5/ 20	2.654	42.8	6.0/ 20	3.057	34.1
5.0/ 40	2.884	37.9	5.5/ 40	2.297	50.5	6.0/ 40	2.061	55.6
5.0/ 60	2.266	51.2	5.5/ 60	2.033	56.2	6.0/ 60	2.069	55.4
5.0/ 80	2.781	40.1	5.5/ 80	2.105	54.6	6.0/ 80	1.842	60.3
5.0/ 100	2.187	52.9	5.5/ 100	2.520	45.7	6.0/ 100	1.859	59.9

pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)	pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)
6.5/ 0	4.641	0.0	7.0/ 0	4.641	0.0
6.5/ 20	2.127	54.2	7.0/ 20	3.710	20.1
6.5/ 40	2.022	56.4	7.0/ 40	2.700	41.8
6.5/ 60	2.000	56.9	7.0/ 60	2.370	48.9
6.5/ 80	1.986	57.2	7.0/ 80	2.481	46.5
6.5/ 100	1.979	57.4	7.0/ 100	2.131	54.1

Table A-4: The coagulation experiment result for removal DOC from Mae-Kuang Reservoir water

pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)
5.0/ 0	2.402	0.0	5.5/ 0	2.402	0.0	6.0/ 0	2.402	0.0
5.0/ 20	1.130	53.0	5.5/ 20	1.413	41.2	6.0/ 20	1.29	46.3
5.0/ 40	1.112	53.7	5.5/ 40	1.252	47.9	6.0/ 40	1.256	47.7
5.0/ 60	1.062	55.8	5.5/ 60	1.172	51.2	6.0/ 60	1.162	51.6
5.0/ 80	1.051	56.2	5.5/ 80	1.124	53.2	6.0/ 80	1.171	51.2
5.0/ 100	1.054	56.1	5.5/ 100	1.095	54.4	6.0/ 100	1.146	52.3

pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)
6.5/ 0	2.402	0.0	7.0/ 0	2.402	0.0
6.5/ 20	1.586	34.0	7.0/ 20	1.878	21.8
6.5/ 40	1.316	45.2	7.0/ 40	1.562	35.0
6.5/ 60	1.316	45.2	7.0/ 60	1.364	43.2
6.5/ 80	1.246	48.1	7.0/ 80	1.308	45.5
6.5/ 100	1.248	48.0	7.0/ 100	1.352	43.7

Table A-5: The coagulation experiment result for removal UV-254 from Mae-Kuang Reservoir water

pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)	pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)	pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)
5.0/ 0	0.0524	0.0	5.5/ 0	0.0524	0.0	6.0/ 0	0.0524	0.0
5.0/ 20	0.0211	59.7	5.5/ 20	0.0261	50.2	6.0/ 20	0.0255	51.3
5.0/ 40	0.0186	64.5	5.5/ 40	0.0206	60.7	6.0/ 40	0.0219	58.2
5.0/ 60	0.0177	66.2	5.5/ 60	0.0200	61.8	6.0/ 60	0.0208	60.3
5.0/ 80	0.0180	65.6	5.5/ 80	0.0188	64.1	6.0/ 80	0.0202	61.5
5.0/ 100	0.0171	67.4	5.5/ 100	0.0177	66.2	6.0/ 100	0.0197	62.4

pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)	pH/Alum dosage (mg/L)	UV-254 residue (1/cm)	percent removal (%)
6.5/ 0	0.0524	0.0	7.0/ 0	0.0524	0.0
6.5/ 20	0.0311	40.6	7.0/ 20	0.037	29.4
6.5/ 40	0.0236	55.0	7.0/ 40	0.0287	45.2
6.5/ 60	0.0225	57.1	7.0/ 60	0.0238	54.6
6.5/ 80	0.0223	57.4	7.0/ 80	0.0225	57.1
6.5/ 100	0.0212	59.5	7.0/ 100	0.0231	55.9

Table A-6: The coagulation experiment result for removal SUVA from Mae-Kuang Reservoir water

pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)	pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)	pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)
5.0/ 0	2.182	0.0	5.5/ 0	2.182	0.0	6.0/ 0	2.182	0.0
5.0/ 20	1.867	14.4	5.5/ 20	1.847	15.3	6.0/ 20	1.977	9.4
5.0/ 40	1.673	23.3	5.5/ 40	1.645	24.6	6.0/ 40	1.744	20.1
5.0/ 60	1.667	23.6	5.5/ 60	1.706	21.8	6.0/ 60	1.790	17.9
5.0/ 80	1.713	21.5	5.5/ 80	1.673	23.3	6.0/ 80	1.725	20.9
5.0/ 100	1.622	25.6	5.5/ 100	1.616	25.9	6.0/ 100	1.719	21.2

pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)	pH/Alum dosage (mg/L)	SUVA residue (L/mg-m)	percent removal (%)
6.5/ 0	2.182	0.0	7.0/ 0	2.182	0.0
6.5/ 20	1.961	10.1	7.0/ 20	1.970	9.7
6.5/ 40	1.793	17.8	7.0/ 40	1.837	20.0
6.5/ 60	1.710	21.6	7.0/ 60	1.745	20.0
6.5/ 80	1.790	18.0	7.0/ 80	1.720	21.1
6.5/ 100	1.699	22.1	7.0/ 100	1.709	21.7

Table A-7: The coagulation experiment result for removal DOC from Mae-Sa River water

pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)
5.0/0	1.657	0.0	5.5/ 0	1.657	0.0	6.0/ 0	1.657	0.0
5.0/20	0.678	59.1	5.5/ 20	0.662	60.0	6.0/ 20	0.791	52.3
5.0/40	0.550	66.8	5.5/ 40	0.614	62.9	6.0/ 40	0.641	61.3
5.0/60	0.571	65.5	5.5/ 60	0.577	65.2	6.0/ 60	0.743	55.2
5.0/80	0.551	66.7	5.5/ 80	0.572	65.5	6.0/ 80	0.66	60.2
5.0/100	0.512	69.1	5.5/ 100	0.57	65.6	6.0/ 100	0.674	59.3

pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	DOC residue (mg/L)	percent removal (%)
6.5/ 0	1.657	0.0	7.0/ 0	1.657	0.0
6.5/ 20	1.091	34.2	7.0/ 20	1.219	26.4
6.5/ 40	0.933	43.7	7.0/ 40	1.119	32.5
6.5/ 60	0.828	50.0	7.0/ 60	1.016	38.7
6.5/ 80	0.858	48.2	7.0/ 80	1.003	39.5
6.5/ 100	0.822	50.4	7.0/ 100	0.951	42.6

Table A-8: The coagulation experiment result for removal UV-254 from Mae-Sa River water

pH/Alum dosage (mg/L)	UV-254 residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	UV-254 residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	UV-254 residue (mg/L)	percent removal (%)
5.0/0	0.0244	0.0	5.5/ 0	0.0244	0.0	6.0/ 0	0.0244	0.0
5.0/20	0.0135	44.7	5.5/ 20	0.0127	48.0	6.0/ 20	0.0137	43.9
5.0/40	0.0139	43.0	5.5/ 40	0.0181	25.8	6.0/ 40	0.0130	46.7
5.0/60	0.0142	41.8	5.5/ 60	0.0151	38.1	6.0/ 60	0.0124	49.2
5.0/80	0.0118	51.6	5.5/ 80	0.0130	46.7	6.0/ 80	0.0121	50.4
5.0/100	0.0102	58.2	5.5/ 100	0.0105	57.0	6.0/ 100	0.0100	59.0

pH/Alum dosage (mg/L)	UV-254 residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	UV-254 residue (mg/L)	percent removal (%)
6.5/ 0	0.0244	0.0	7.0/ 0	0.0244	0.0
6.5/ 20	0.0170	30.3	7.0/ 20	0.0184	24.6
6.5/ 40	0.0132	45.9	7.0/ 40	0.0159	34.8
6.5/ 60	0.0125	48.8	7.0/ 60	0.0141	42.2
6.5/ 80	0.0121	50.4	7.0/ 80	0.0133	45.5
6.5/ 100	0.0117	52.0	7.0/ 100	0.0129	47.1

Table A-9: The coagulation experiment result for removal SUVA from Mae-Sa River water

pH/Alum dosage (mg/L)	SUVA residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	SUVA residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	SUVA residue (mg/L)	percent removal (%)
5.0/0	1.390	0.0	5.5/ 0	1.390	0.0	6.0/ 0	1.390	0.0
5.0/20	1.991	0.0	5.5/ 20	1.918	0.0	6.0/ 20	1.732	0.0
5.0/40	2.527	0.0	5.5/ 40	2.948	0.0	6.0/ 40	2.028	0.0
5.0/60	2.487	0.0	5.5/ 60	2.617	0.0	6.0/ 60	1.669	0.0
5.0/80	2.142	0.0	5.5/ 80	2.273	0.0	6.0/ 80	1.833	0.0
5.0/100	1.992	0.0	5.5/ 100	1.842	0.0	6.0/ 100	1.484	0.0

pH/Alum dosage (mg/L)	SUVA residue (mg/L)	percent removal (%)	pH/Alum dosage (mg/L)	SUVA residue (mg/L)	percent removal (%)
6.5/ 0	1.390	0.0	7.0/ 0	1.390	0.0
6.5/ 20	1.558	0.0	7.0/ 20	1.509	0.0
6.5/ 40	1.415	0.0	7.0/ 40	1.421	0.0
6.5/ 60	1.510	0.0	7.0/ 60	1.388	0.1
6.5/ 80	1.410	0.0	7.0/ 80	1.326	4.6
6.5/ 100	1.423	0.0	7.0/ 100	1.356	2.4

BIOGRAPHY

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