

REFERENCES

- Aboofazeli, R., Patel, N., Thomas, M., and Lawrence, M.J. (1995). Investigations into the Formation and Characterization of Phospholipid Microemulsions. IV Pseudo-ternary Phase Diagrams of Systems Containing Water-lecithin-alcohol and Oil: The Influence of Oil. International Journal of Pharmaceutical. 125, 107-116.
- Acosta, E. (2000). Use of lipophilic and hydrophilic linkers in trichloroethylene microemulsions. M.S. Thesis, University of Oklahoma.
- Aowiriyakul, S. (1998). Alcohol-free microemulsion formation with perchloroethylene and a gemini surfactant, M.S. thesis in Petrochemical Technology, The Petroleum and Petrochemical College, Chulalongkorn University.
- Ash, M. and Ash, I. (1997a). Handbook of Industrial Surfactant Volume 1. New York: Synapse publication.
- Ash, M. and Ash, I. (1997b). Handbook of Industrial Surfactant Volume 2. New York: Synapse publication.
- Benito, I., Garcia, M.A., Monge, C., Saz, J.M., and Marina, M.L. (1997). Spectrophotometric and Conductimetric Determination of the Critical Micellar Concentration of Sodium Dodecyl Sulfate and Cetyltrimethylammonium Bromide Micellar Systems Modified by Alcohols and Salts. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 125, 221-224.
- Bourrel, M. and Schechter, R.S. (1988a). Microemulsions and Related Systems: Formulation, Solvency, and Physical Properties. Surfactant Science Series Vol. 30. New York: Marcel Dekker, 172-177.

- Bourrel, M. and Schechter, R.S. (1988b). Microemulsions and Related Systems: Formulation, Solvency, and Physical Properties. Surfactant Science Series Vol. 30. New York: Marcel Dekker, 229-233.
- Carter, T., Wu, B., Sabatini, D.A., and Harwell, J.H. (1998). Increasing the Solubility Enhancement of Anionic DOWFAX Surfactants. Separation Science and Technology, 33, 2363-2377.
- Deshpande, S., Wesson, L., Wade, D., Sabatini, D.A., and Harwell, J.H. (2000). DOWFAX Surfactant Components for Enhancing Contaminant Solubilization. Water Resources, 34, 1030-1036.
- Gizurason, S., Jonsdottir, V.M., and Heron, I. (1995). Internasal Administration of Diphtheria Toxoid. Selecting Antibody Isotypes using Formulations having Various Lipophilic Characteristics. Vaccine, 13(7), 617-621.
- Graciaa, A., Lachaise, J., Cucuphat, C., Bourrel, M., and Salager, J.L. (1993). Improving Solubilization in Microemulsions with Additives. 2. Long Chain Alcohols as Lipophilic Linkers. Langmuir, 9, 3371-3374.
- Healy, R.N., and Reed, R.L. (1976). Some Physicochemical Aspects of Microemulsion Flooding. In Shah, D.O., and Schechter R.S. (Eds.), A Review Improved Oil Recovery by Surfactant and Polymer Flooding. London: Academic Press.
- Holar, T.P. and Schulman, J.H. (1943). Transparent Water-in-oil Dispersions: The Oleopathic Hydro-micelle. Nature, 152, 102-103.
- Jimenez-Carmona, M.M. and Lague de Castro, M.D. (1998). Reverse-Micelle Formations: A Strategy for Enhancing CO₂-supercritical Fluid Extraction of Polar Analytes. Analytical Chemica Acta, 358, 1-4.
- Kahlweit, M., Busse, G., Faulhaber, B., and Eibl, H. (1995a). Preparing Nontoxic Microemulsions. Langmuir, 11(11), 4185-4187.
- Kahlweit, M., Busse, G., and Winkler, J. (1995b). Electrical Conductivity in Microemulsions, Journal of Chemical Physics, 99, 5605-5614.

- Kahlweit, M. (1995c). How to Prepare Microemulsion at Prescribed Temperature, Oil, and Brine. The Journal of Physical Chemistry, 99 (4), 1281-1284.
- Kahlweit, M., Busse, G., and Faulhaber, B. (1997). Preparing Nontoxic Microemulsion 2. Langmuir, 15, 5249-5251.
- Lang, J., Lalan, N., and Zana, R. (1992). Quaternary Water-in-Oil Microemulsions 2. Effect of Carboxylic Acid Chain Length on Droplet Size and Exchange of Material between Droplets. Journal of Physical Chemistry, 96, 4667-4671.
- Lide, D.R. (Ed.) (2000). Handbook of Chemistry and Physics. Boca Raton: CRC Press.
- Maidment, L.J., Chen, V., and Warr, G.G. (1997). Effect of Added Cosurfactant on Ternary Microemulsion Structure and Dynamics. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 129-130, 311-319.
- Manual for Spinning Drop Tensiometer Site 04. Hamburg: Kruss.
- Meziani, A., Zradba, A., Touraud, D., Clause, M., and Kunz, W. (1997). Can Aldehydes Participate in the Nanostructuring of Liquids Containing Charged Micelles? Journal of Molecular Liquids, 74(73), 107-118.
- Miksik, I., Gabriel, J., and Deyl, Z. (1997). Microemulsion Electrokinetic Chromatography of Diphenylhydrazones of Dicarboxylic Sugars. Journal of Chromatography A, 772, 297-303.
- Osborne, D.W., Middleton, C.A., and Rogers, R.L. (1998). Alcohol-free Microemulsions. Journal of Dispersion Science and Technology, 9 (4), 415-423.
- Pennell, K.D., Jin, M., Abriola, L.M., and Pope, G.A. (1994). Surfactant Enhanced Remediation of Soil Columns Contaminated by Residual Tetrachloroethylene, Journal of Contaminant Hydrology, 16, 35.

- Rosen, M.J. (1989). Surfactants and Interfacial Phenomena. New York: John Wiley & Sons.
- Rouse, J.D., Sabatini, D.A., Brown, R.E., and Harwell, J.H. (1996). Evaluation of the Ethoxylated Alkylsulfate Surfactants for Use in Subsurface Remediation. Water Environment Research, 68, 162.
- Salager, J.L., Microemulsions, in Handbook of Detergents Part A: Properties., Zoller, U. (Ed.). (1999). New York: Marcel Dekker, 253-302.
- Selle, M.H., Sjoblom, J., and Skurtveit, R. (1991). Emulsions under Elevated Pressure and Temperature Conditions II The Model System Water (Electrolyte) - Octanoic Acid - Sodium Octanoate - n-Heptane at 20°C. Journal of Colloid and Interface Science, 144(1), 36-44.
- Shiao, S.Y., Chhabra, V., Patist, A., Free, M.L., Huibers, P.D.T., Gregory, A., Patel, S., and Shah, D.O. (1998). Chain Length Compatibility Effects in Mixed Surfactant Systems for Technological Applications. Advances in Colloid and Interface Science, 74, 1-29.
- Shiau, B.J., Sabatini, D.A., and Harwell, J.H., (1994). Solubilization and Microemulsification of Chlorinated Solvents Using Direct Food Additive (Edible) Surfactants. Ground Water, 32, 561.
- Shiau, B.J., Sabatini, D.A., and Harwell, J.H. (1995). Properties of Food Grade (Edible) Surfactants Affecting Subsurface Remediation of Chlorinated Solvents. Environmental Science & Technology, 29, 2929-2935.
- Sjoblom, J., Lindberg, R., and Friberg, S.E. (1996). Microemulsion-phase Equilibria Characterization Structures, Applications, and Chemical Reactions. Advance Colloid and Interface Science, 95, 125-287.

- Solans, C. and Kunieda, H. (Eds.). (1997). Industrial Applications of Microemulsions. New York: Marcel Dekker.
- Sunwoo, C.K. and Wade, W.H. (1992). Optimal Surfactant Structures for Cosurfactant-free Microemulsion System I. C16 and C14 Guerbet Alcohol Hydrophobes. Journal of Dispersion Science and Technology, 13(5), 491-514.
- Thevinin, M.A., Grossiord, J.L., and Poelman, M.C. (1997). Sucrose Esters/Cosurfactant Microemulsion Systems for Transdermal Delivery: Assessment of Bicontinuous Structure. International Journal of Pharmaceutics, 137, 177-186.
- Tran, S.T. (2000). Enhancement of tetrachloroethylene solubilization in water using linkers in microemulsion. M.S. Thesis, University of Oklahoma.
- Trotta, M., Pattarino, F., and Grosa, G. (1998). Formation of Lecithin-based Microemulsions Containing n-Alkanol Phosphocholides. International Journal of Pharmaceutics, 174, 253-259.
- West, C.C. and Harwell, J.H. (1992). Surfactants and Subsurface Remediation. Environmental Science & Technology, 26, 2324-2330.
- Wu, B. (1997). Formulating microemulsion systems of petroleum hydrocarbons using surfactant/cosurfactant mixtures, M.S. thesis. University of Oklahoma.
- Yao, J. and Romsted, L.S. (1997). Effect of Hydrocarbon and Triglyceride Oils on Butanol Distribution in Water-in-oil Cationic Microemulsion. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 123, 89-105.

APPENDIX A

Experimental Data from Phase Diagram and Solubilization Studies

Table A-1a Winsor type of microemulsion and phase volume at different SDS and NaCl concentrations for microemulsion containing propanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
24.0	1.9908	3010	1990	50	0.250	4.000	I	10.85	10.70	-	5.28	5.43	2.65	2.73	-
				70	0.350	3.600	I	12.30	12.20	-	6.05	6.15	3.04	3.09	-
				80	0.400	3.400	I	11.05	11.00	-	5.48	5.53	2.75	2.78	-
				85	0.425	3.300	IV	12.00	-	-	6.00	6.00	3.01	3.01	-
				90	0.450	3.200	IV	12.10	-	-	6.05	6.05	3.04	3.04	-
				110	0.550	2.800	IV	12.05	-	-	6.03	6.03	3.03	3.03	-
				130	0.650	2.400	IV	12.40	-	-	6.20	6.20	3.11	3.11	-
				150	0.750	2.000	IV	11.15	-	-	5.58	5.58	2.80	2.80	-
				170	0.850	1.600	IV	10.80	-	-	5.40	5.40	2.71	2.71	-
				180	0.900	1.400	IV	12.40	-	-	6.20	6.20	3.11	3.11	-
				185	0.925	1.300	IV	11.88	-	-	5.94	5.94	2.98	2.98	-
				190	0.950	1.200	II	12.25	0.60	-	6.13	5.53	3.08	2.78	-
				210	1.050	0.800	II	11.80	1.70	-	5.90	4.20	2.96	2.11	-
				230	1.150	0.400	II	12.50	2.55	-	6.25	3.70	3.14	1.86	-

Table A-1b Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing propanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
21.6	1.7917	2710	2290	100	0.500	3.000	I	10.90	10.50	-	5.05	5.45	2.82	3.04	-
				110	0.550	2.800	I	11.00	10.70	-	5.20	5.50	2.90	3.67	-
				120	0.600	2.600	I	11.00	10.80	-	5.30	5.50	2.96	3.07	-
				130	0.650	2.400	I	11.00	10.95	-	5.45	5.50	3.04	3.07	-
				135	0.675	2.300	IV	12.00	-	-	6.00	6.00	3.35	3.35	-
				140	0.700	2.200	IV	11.10	-	-	5.55	5.55	3.10	3.10	-
				150	0.750	2.000	IV	10.90	-	-	5.45	5.45	3.04	3.04	-
				170	0.850	1.600	IV	11.20	-	-	5.60	5.60	3.13	3.13	-
				180	0.900	1.400	IV	11.30	-	-	5.65	5.65	3.15	3.15	-
				185	0.925	1.300	IV	10.50	-	-	5.25	5.25	2.93	2.93	-
				190	0.950	1.200	II	11.30	0.22	-	5.65	5.43	3.15	3.03	-
				200	1.000	1.000	II	10.60	0.80	-	5.30	4.50	2.96	2.51	-

Table A-1c Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing propanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
19.8	1.6424	2485	2515	100	0.500	3.000	I	11.70	10.85	-	5.00	5.85	3.04	3.56	-
				120	0.600	2.600	I	11.90	11.35	-	5.40	5.95	3.29	3.62	-
				140	0.700	2.200	I	11.70	11.00	-	5.15	5.85	3.14	3.56	-
				140	0.700	2.200	I	11.70	11.50	-	5.65	5.85	3.44	3.56	-
				150	0.750	2.000	I	11.80	11.60	-	5.70	5.90	3.47	3.59	-
				155	0.775	1.900	IV	12.20	-	-	6.10	6.10	3.71	3.71	-
				160	0.800	1.800	IV	12.10	-	-	6.05	6.05	3.68	3.68	-
				170	0.850	1.600	IV	11.70	-	-	5.85	5.85	3.56	3.56	-
				175	0.875	1.500	IV	11.90	-	-	5.95	5.95	3.62	3.62	-
				180	0.900	1.400	II	12.30	0.02	-	6.15	6.13	3.74	3.73	-
				180	0.900	1.400	II	11.30	0.07	-	5.65	5.58	3.44	3.40	-
				185	0.925	1.300	II	11.80	0.48	-	5.90	5.42	3.59	3.30	-
				190	0.950	1.200	II	12.30	1.35	-	6.15	4.80	3.74	2.92	-
				190	0.950	1.200	II	11.20	0.75	-	5.60	4.85	3.41	2.95	-
				200	1.000	1.000	II	12.10	1.95	-	6.05	4.10	3.68	2.50	-
				220	1.100	0.600	II	11.50	2.40	-	5.75	3.35	3.50	2.04	-

Table A-1d Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing propanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
18.0	1.4931	2260	2740	150	0.750	2.000	I	11.30	10.50	-	4.85	5.65	3.25	3.78	-
				160	0.800	1.800	I	11.00	10.40	-	4.90	5.50	3.28	3.68	-
				170	0.850	1.600	I	11.30	11.00	-	5.35	5.65	3.58	3.78	-
				180	0.900	1.400	I	11.70	11.60	-	5.75	5.85	3.85	3.92	-
				185	0.925	1.300	II	11.10	0.07	-	5.55	5.48	3.72	3.67	-
				190	0.950	1.200	II	10.90	0.40	-	5.45	5.05	3.65	3.38	-
				200	1.000	1.000	II	11.10	1.30	-	5.55	4.25	3.72	2.85	-
15.6	1.2940	1955	3045	150	0.750	2.000	I	11.90	10.70	-	4.75	5.95	3.67	4.60	-
				170	0.850	1.600	I	12.00	11.15	-	5.15	6.00	3.98	4.64	-
				180	0.900	1.400	I	11.40	10.88	-	5.18	5.70	4.00	4.40	-
				185	0.925	1.300	III	11.10	10.70	0.50	5.15	5.05	3.98	3.90	-
				190	0.950	1.200	III	12.05	11.90	1.45	5.88	4.58	4.54	3.54	-
				200	1.000	1.000	III	11.80	11.75	2.20	5.85	3.70	4.52	2.86	-
				205	1.025	0.900	II	11.40	2.20	-	5.70	3.50	4.40	2.70	-
				210	1.050	0.800	II	11.65	2.50	-	5.83	3.33	4.50	2.57	-
				230	1.150	0.400	II	11.15	2.90	-	5.58	2.68	4.31	2.07	-
				250	1.250	0.000	II	12.00	3.90	-	6.00	2.10	4.64	1.62	-

Table A-1e Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing propanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
13.2	1.0949	1655	3345	170	0.850	1.600	I	10.35	8.80	-	3.63	5.18	3.31	4.73	-
				185	0.925	1.300	II	10.60	9.50	0.30	4.20	5.00	3.84	4.57	-
				190	0.950	1.200	III	10.02	9.50	0.40	4.49	4.61	4.10	4.21	-
				210	1.050	0.800	III	10.55	10.00	1.85	4.73	3.43	4.32	3.13	-
				215	1.075	0.700	III	11.80	11.78	3.05	5.88	2.85	5.37	2.60	-
				220	1.100	0.600	II	10.33	2.35	-	5.17	2.82	4.72	2.57	-
				230	1.150	0.400	II	10.80	3.00	-	5.40	2.40	4.93	2.19	-
				240	1.200	0.200	II	10.10	2.75	-	5.05	2.30	4.61	2.10	-
				250	1.250	0.000	II	10.60	3.20	-	5.30	2.10	4.84	1.92	-
10.2	0.8461	1280	3720	150	0.750	2.000	I	10.90	8.47	-	3.02	5.45	3.57	6.44	-
				170	0.850	1.600	I	11.00	8.75	-	3.25	5.50	3.84	6.50	-
				180	0.900	1.400	I	11.10	9.10	-	3.55	5.55	4.20	6.56	-
				185	0.925	1.300	I	11.60	9.60	-	3.80	5.80	4.49	6.85	-
				190	0.950	1.200	III	11.00	9.25	1.02	3.75	4.48	4.43	5.29	-
				230	1.150	0.400	III	11.50	11.35	3.50	5.60	2.25	6.62	2.66	-
				235	1.175	0.300	III	10.90	10.80	3.30	5.35	2.15	6.32	2.54	-
				240	1.200	0.200	II	11.05	3.35	-	5.53	2.18	6.53	2.57	-
				250	1.250	0.000	II	10.80	3.70	-	5.40	1.70	6.38	2.01	-

Table A-1f Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing propanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
7.2	0.5972	905	4095	170	0.850	1.600	I	9.90	7.10	-	2.15	4.95	3.60	8.29	-
				180	0.900	1.400	I	10.20	7.45	-	2.35	5.10	3.94	8.54	-
				190	0.950	1.200	I	10.05	7.60	-	2.58	5.03	4.31	8.41	-
				200	1.000	1.000	III	9.90	7.95	0.01	3.00	4.94	5.02	8.27	-
				205	1.025	0.900	III	10.80	8.70	1.30	3.30	4.10	5.53	6.87	-
				210	1.050	0.800	III	10.30	8.15	0.90	3.00	4.25	5.02	7.12	-
				220	1.100	0.600	III	10.20	8.30	1.70	3.20	3.40	5.36	5.69	-
				230	1.150	0.400	III	10.20	8.70	2.30	3.60	2.80	6.03	4.69	-
				240	1.200	0.200	III	10.00	9.40	2.83	4.40	2.17	7.37	3.63	-
				245	1.225	0.100	III	10.90	10.55	3.30	5.10	2.15	8.54	3.60	-
				250	1.250	0.000	II	10.10	3.10	-	5.05	1.95	8.46	3.27	-

Table A-2a Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing butanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _w	V _o	SP _o	SP _w	Gel
21.6	1.7917	2805	2195	0	0.000	5.000	IV	10.80	-	-	5.40	5.40	3.01	3.01	-
				1	0.005	4.980	IV	11.40	-	-	5.70	5.70	3.18	3.18	-
				2	0.010	4.960	IV	11.05	-	-	5.53	5.53	3.08	3.08	-
				3	0.015	4.940	IV	11.05	-	-	5.53	5.53	3.08	3.08	-
				4	0.020	4.920	IV	10.90	-	-	5.45	5.45	3.04	3.04	-
				5	0.025	4.900	IV	11.20	-	-	5.60	5.60	3.13	3.13	-
				10	0.050	4.800	IV	10.75	-	-	5.38	5.38	3.00	3.00	-
				20	0.100	4.600	IV	10.75	-	-	5.38	5.38	3.00	3.00	-
				40	0.200	4.200	IV	11.15	-	-	5.58	5.58	3.11	3.11	-
				60	0.300	3.800	IV	11.10	-	-	5.55	5.55	3.10	3.10	-
				65	0.325	3.700	II	11.25	0.05	-	5.63	5.58	3.14	3.11	-
				70	0.350	3.600	II	11.40	0.30	-	5.70	5.40	3.18	3.01	-
				80	0.400	3.400	II	10.90	0.65	-	5.45	4.80	3.04	2.68	-
				100	0.500	3.000	II	11.00	1.40	-	5.50	4.10	3.07	2.29	-

Table A-2b Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing butanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
18.0	1.4931	2340	2660	20	0.100	4.600	I	10.80	10.45	-	5.05	5.40	3.38	3.62	-
				30	0.150	4.400	I	10.90	10.85	-	5.40	5.45	3.62	3.65	-
				35	0.175	4.300	IV	11.95	-	-	5.98	5.98	4.00	4.00	-
				40	0.200	4.200	IV	11.20	-	-	5.60	5.60	3.75	3.75	-
				60	0.300	3.800	IV	11.17	-	-	5.59	5.59	3.74	3.74	-
				65	0.325	3.700	II	10.20	0.07	-	5.10	5.03	3.42	3.37	-
				70	0.350	3.600	II	10.20	0.35	-	5.10	4.75	3.42	3.18	-
				80	0.400	3.400	II	11.10	1.00	-	5.55	4.55	3.72	3.05	-
				100	0.500	3.000	II	10.40	1.70	-	5.20	3.50	3.48	2.34	-
15.6	1.2940	2025	2975	20	0.100	4.600	I	10.65	9.25	-	3.93	5.33	3.03	4.12	-
				40	0.200	4.200	I	10.90	10.20	-	4.75	5.45	3.67	4.21	-
				50	0.250	4.000	I	10.25	10.20	-	5.08	5.13	3.92	3.96	-
				55	0.275	3.900	IV	11.25	-	-	5.63	5.63	4.35	4.35	-
				60	0.300	3.800	IV	10.62	-	-	5.31	5.31	4.10	4.10	-
				65	0.325	3.700	IV	10.90	-	-	5.45	5.45	4.21	4.21	-
				70	0.350	3.600	II	10.50	0.07	-	5.25	5.18	4.06	4.00	-
				80	0.400	3.400	II	10.80	0.85	-	5.40	4.55	4.17	3.52	-
				100	0.500	3.000	II	11.05	1.89	-	5.53	3.64	4.27	2.81	-

Table A-2c Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing butanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
13.2	1.0949	1715	3285	60	0.300	3.800	I	10.70	10.02	-	4.67	5.35	4.27	4.89	-
				70	0.350	3.600	I	10.70	10.65	-	5.30	5.35	4.84	4.89	-
				71	0.355	3.580	IV	10.40	-	-	5.20	5.20	4.75	4.75	-
				72	0.360	3.560	IV	10.40	-	-	5.20	5.20	4.75	4.75	-
				73	0.365	3.540	IV	10.30	-	-	5.15	5.15	4.70	4.70	-
				74	0.370	3.520	II	10.20	0.02	-	5.10	5.10	4.66	4.66	-
				75	0.375	3.500	II	11.00	0.25	-	5.50	5.25	5.02	4.79	-
				80	0.400	3.400	II	10.18	0.70	-	5.09	4.39	4.65	4.01	-
				90	0.450	3.200	II	10.70	1.30	-	5.35	4.05	4.89	3.70	-
10.8	0.8959	1405	3595	60	0.300	3.800	I	10.50	8.60	-	3.35	5.25	3.74	5.86	-
				70	0.350	3.600	I	10.15	8.75	-	3.68	5.08	4.10	5.66	-
				75	0.375	3.500	I	11.20	10.00	-	4.40	5.60	4.91	6.25	-
				80	0.400	3.400	III	10.20	9.80	0.02	4.70	5.08	5.25	5.67	-
				85	0.425	3.300	III	10.40	10.00	0.30	4.80	4.90	5.36	5.47	-
				90	0.450	3.200	II	10.00	0.78	-	5.00	4.22	5.58	4.71	-
				100	0.500	3.000	II	10.20	1.53	-	5.10	3.57	5.69	3.98	-

Table A-2d Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing butanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
7.8	0.6470	1015	3985	60	0.300	3.800	I	10.10	7.15	-	2.10	5.05	3.25	7.81	-
				80	0.400	3.400	I	10.30	7.90	-	2.75	5.15	4.25	7.96	-
				100	0.500	3.000	I	9.90	8.50	-	3.55	4.95	5.49	7.65	-
				105	0.525	2.900	III	10.40	9.45	0.66	4.25	4.54	6.57	7.02	-
				110	0.550	2.800	III	10.10	9.45	0.86	4.40	4.19	6.80	6.48	-
				115	0.575	2.700	III	10.45	10.00	1.30	4.78	3.93	7.38	6.07	-
				120	0.600	2.600	II	10.15	1.50	-	5.08	3.58	7.84	5.53	-
				140	0.700	2.200	II	9.95	2.30	-	4.98	2.68	7.69	4.13	-
				4.8	0.3982	625	4375	100	0.500	3.000	I	9.70	6.45	-	1.60
120	0.600	2.600	I					10.00	7.40	-	2.40	5.00	6.03	12.56	-
130	0.650	2.400	I					10.00	8.10	-	3.10	5.00	7.79	12.56	-
135	0.675	2.300	III					10.10	8.45	0.30	3.40	4.75	8.54	11.93	-
140	0.700	2.200	III					10.00	9.00	1.60	4.00	3.40	10.05	8.54	-
145	0.725	2.100	III					10.00	9.30	1.30	4.30	3.70	10.80	9.29	-
150	0.750	2.000	III					10.45	10.40	1.90	5.18	3.33	13.00	8.35	-
155	0.775	1.900	II					10.65	2.85	-	5.33	2.48	13.37	6.22	-
160	0.800	1.800	II					9.10	2.68	-	4.55	1.87	11.43	4.70	-
				170	0.850	1.600	II	10.55	3.40	-	5.28	1.88	13.25	4.71	-

Table A-3a Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing pentanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
13.2	1.0949	1745	3255	10	0.050	4.800	I	11.80	11.50	-	5.60	5.90	5.11	5.39	Gel
				15	0.075	4.760	IV	11.30	-	-	5.65	5.65	5.16	5.16	Gel
				20	0.100	4.600	IV	12.00	-	-	6.00	6.00	5.48	5.48	Gel
				30	0.150	4.400	IV	11.85	-	-	5.93	5.93	5.41	5.41	Gel
				40	0.200	4.200	IV	11.65	-	-	5.83	5.83	5.32	5.32	Gel
				45	0.225	4.100	II	11.40	0.60	-	5.70	5.10	5.21	4.66	Gel
				50	0.250	4.000	II	11.65	1.35	-	5.83	4.48	5.32	4.09	Gel
				70	0.350	3.600	II	11.85	2.60	-	5.93	3.33	5.41	3.04	Gel
				90	0.450	3.200	II	11.90	2.70	-	5.95	3.25	5.43	2.97	Gel
9.6	0.7963	1270	3730	10	0.050	4.800	I	10.80	8.75	-	3.35	5.40	4.21	6.78	Gel
				20	0.100	4.600	I	10.30	8.70	-	3.55	5.15	4.46	6.47	Gel
				30	0.150	4.400	I	11.70	11.00	-	5.15	5.85	6.47	7.35	Gel
				35	0.175	4.300	I	11.50	11.48	-	5.73	5.75	7.20	7.22	Gel
				40	0.200	4.200	IV	10.40	-	-	5.20	5.20	6.53	6.53	Gel
				50	0.250	4.000	IV	11.20	-	-	5.60	5.60	7.03	7.03	Gel
				55	0.275	3.900	II	10.10	0.20	-	5.05	4.85	6.34	6.09	Gel
				60	0.300	3.800	II	10.00	0.60	-	5.00	4.40	6.28	5.53	Gel
				80	0.400	3.400	II	10.60	2.10	-	5.30	3.20	6.66	4.02	Gel

Table A-3b Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing pentanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
7.2	0.5972	955	4045	30	0.150	4.400	I	9.80	7.20	-	2.30	4.90	3.85	8.20	Gel
				50	0.250	4.000	I	10.00	9.50	-	4.50	5.00	7.54	8.37	Gel
				55	0.275	3.900	I	11.20	11.15	-	5.55	5.60	9.29	9.38	Gel
				60	0.300	3.800	IV	11.00	-	-	5.50	5.50	9.21	9.21	Gel
				70	0.350	3.600	IV	9.80	-	-	4.90	4.90	8.20	8.20	Gel
				75	0.375	3.500	II	11.30	0.40	-	5.65	5.25	9.46	8.79	Gel
				80	0.400	3.400	II	10.10	2.30	-	5.05	2.75	8.46	4.60	Gel
				90	0.450	3.200	II	10.40	1.90	-	5.20	3.30	8.71	5.53	Gel
				110	0.550	2.800	II	9.80	3.45	-	4.90	1.45	8.20	2.43	Gel
6.0	0.4977	795	4205	50	0.250	4.000	I	10.85	8.55	-	3.13	5.43	6.28	10.90	Gel
				60	0.300	3.800	I	10.80	9.70	-	4.30	5.40	8.64	10.85	Gel
				65	0.325	3.700	I	9.90	9.50	-	4.55	4.95	9.14	9.95	Gel
				70	0.350	3.600	IV	11.00	-	-	5.50	5.50	11.05	11.05	Gel
				80	0.400	3.400	IV	11.00	-	-	5.50	5.50	11.05	11.05	Gel
				90	0.450	3.200	II	10.80	1.50	-	5.40	3.90	10.85	7.84	Gel
110	0.550	2.800	II	10.80	2.90	-	5.40	2.50	10.85	5.02	Gel				

Table A-3c Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing pentanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
4.8	0.3982	635	4365	50	0.250	4.000	I	11.60	6.65	-	0.85	5.80	2.13	14.57	Gel
				70	0.350	3.600	I	10.90	8.50	-	3.05	5.45	7.66	13.69	Gel
				75	0.375	3.500	III	11.00	9.10	0.70	3.60	4.80	9.04	12.05	Gel
				80	0.400	3.400	III	10.90	9.50	1.00	4.05	4.45	10.17	11.18	Gel
				90	0.450	3.200	(II)	10.90	2.00	-	5.45	3.45	13.69	8.66	Gel
				110	0.550	2.800	II	10.05	3.20	-	5.03	1.83	12.62	4.58	Gel
3.6	0.2986	475	4525	40	0.200	4.200	I	10.20	5.80	-	0.70	5.10	2.34	17.08	Gel
				60	0.300	3.800	I	10.30	6.00	-	0.85	5.15	2.85	17.25	Gel
				70	0.350	3.600	I	10.30	7.10	-	1.95	5.15	6.53	17.25	Gel
				80	0.400	3.400	III	10.05	8.20	2.00	3.18	3.03	10.63	10.13	Gel
				100	0.500	3.000	II	10.10	3.90	-	5.05	1.15	16.91	3.85	Gel
				120	0.600	2.600	II	10.20	4.30	-	5.10	0.80	17.08	2.68	Gel

Table A-4a Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing hexanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V ₁	V ₂	SP ₁	SP ₂	Gel
10.8	0.8960	1450	3550	5	0.025	4.900	I	11.60	10.30	-	4.50	5.80	5.02	6.47	Gel
				10	0.050	4.800	IV	11.40	-	-	5.70	5.70	6.36	6.36	Gel
				20	0.100	4.600	IV	10.35	-	-	5.18	5.18	5.78	5.78	Gel
				25	0.125	4.500	IV	11.40	-	-	5.70	5.70	6.36	6.36	Gel
				30	0.150	4.400	II	11.30	1.50	-	5.65	4.15	6.31	4.63	Gel
				40	0.200	4.200	II	9.95	1.70	-	4.98	3.28	5.55	3.66	Gel
				60	0.300	3.800	II	10.20	1.80	-	5.10	3.30	5.69	3.68	Gel
				80	0.400	3.400	II	10.15	2.50	-	5.08	2.58	5.66	2.87	Gel
				100	0.500	3.000	II	10.70	3.30	-	5.35	2.05	5.97	2.29	Gel
9.6	0.7963	1290	3710	20	0.100	4.600	I	11.40	11.20	-	5.50	5.70	6.91	7.16	Gel
				25	0.125	4.500	IV	10.00	-	-	5.00	5.00	6.28	6.28	Gel
				30	0.150	4.400	IV	10.30	-	-	5.15	5.15	6.47	6.47	Gel
				40	0.200	4.200	IV	11.50	-	-	5.75	5.75	7.22	7.22	Gel
				45	0.225	4.100	II	10.00	0.42	-	5.00	4.58	6.28	5.75	Gel
				50	0.250	4.000	II	10.40	0.97	-	5.20	4.23	6.53	5.31	Gel
				60	0.300	3.800	II	11.40	1.50	-	5.70	4.20	7.16	5.27	Gel
				80	0.400	3.400	II	11.40	4.10	-	5.70	1.60	7.16	2.01	Gel
				100	0.500	3.000	II	11.60	4.60	-	5.80	1.20	7.28	1.51	Gel

Table A-4b Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing hexanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
8.4	0.6968	1125	3875	10	0.050	4.800	I	11.20	8.50	-	2.90	5.60	4.16	8.04	Gel
				15	0.075	4.700	I	11.30	9.80	-	4.15	5.65	5.96	8.11	Gel
				20	0.100	4.600	IV	10.25	-	-	5.13	5.13	7.36	7.36	Gel
				40	0.200	4.200	IV	9.95	-	-	4.98	4.98	7.14	7.14	Gel
				45	0.225	4.100	IV	11.25	-	-	5.63	5.63	8.07	8.07	Gel
				50	0.250	4.000	II	11.10	1.10	-	5.55	4.45	7.96	6.39	Gel
				60	0.300	3.800	II	9.90	1.55	-	4.95	3.40	7.10	4.88	Gel
				80	0.400	3.400	II	10.15	2.50	-	5.08	2.58	7.28	3.70	Gel
7.2	0.5972	965	4035	100	0.500	3.000	II	10.35	3.00	-	5.18	2.18	7.43	3.12	Gel
				20	0.100	4.600	i	11.60	9.30	-	3.50	5.80	5.86	9.71	Gel
				25	0.125	4.500	IV	10.20	-	-	5.10	5.10	8.54	8.54	Gel
				30	0.150	4.400	IV	9.90	-	-	4.95	4.95	8.29	8.29	Gel
				40	0.200	4.200	IV	11.20	-	-	5.60	5.60	9.38	9.38	Gel
				50	0.250	4.000	IV	10.10	-	-	5.05	5.05	8.46	8.46	Gel
				60	0.300	3.800	II	11.20	-	-	5.60	5.60	9.38	9.38	Gel
				80	0.400	3.400	II	11.30	4.00	-	5.65	1.65	9.46	2.76	Gel
				100	0.500	3.000	II	11.35	4.60	-	5.68	1.08	9.50	1.80	Gel

Table A-4c Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing hexanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
6.0	0.4977	805	4195	20	0.100	4.600	I	10.00	6.80	-	1.80	5.00	3.62	10.05	Gel
				30	0.150	4.400	I	10.90	10.05	-	4.60	5.45	9.24	10.95	Gel
				35	0.175	4.300	IV	10.90	-	-	5.45	5.45	10.95	10.95	Gel
				40	0.200	4.200	IV	9.70	-	-	4.85	4.85	9.74	9.74	Gel
				60	0.300	3.800	II	10.10	0.20	-	5.05	4.85	10.15	9.74	Gel
				80	0.400	3.400	II	9.80	2.22	-	4.90	2.68	9.85	5.38	Gel
				100	0.500	3.000	II	10.00	2.60	-	5.00	2.40	10.05	4.82	Gel
4.8	0.3982	645	4355	40	0.200	4.200	I	11.30	8.40	-	2.75	5.65	6.91	14.19	Gel
				45	0.225	4.100	I	10.00	9.40	-	4.40	5.00	11.05	12.56	Gel
				50	0.250	4.000	IV	9.80	-	-	4.90	4.90	12.31	12.31	Gel
				60	0.300	3.800	IV	11.00	-	-	5.50	5.50	13.81	13.81	Gel
				80	0.400	3.400	IV	11.00	-	-	5.50	5.50	13.81	13.81	Gel
				90	0.450	3.200	IV	10.20	-	-	5.10	5.10	12.81	12.81	Gel
				95	0.475	3.100	IV	10.10	-	-	5.05	5.05	12.68	12.68	Gel
				100	0.500	3.000	II	11.20	1.00	-	5.60	4.60	14.06	11.55	Gel
				120	0.600	2.600	II	11.20	2.50	-	5.60	3.10	14.06	7.79	Gel

Table A-4d Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing hexanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
3.6	0.2986	485	4515	20	0.100	4.600	I	10.00	5.40	-	0.40	5.00	1.34	16.74	Gel
				40	0.200	4.200	I	10.00	6.40	-	1.40	5.00	4.69	16.74	Gel
				50	0.250	4.000	I	10.90	7.90	-	2.45	5.45	8.20	18.25	Gel
				55	0.275	3.900	I	10.70	8.30	-	2.95	5.35	9.88	17.92	Gel
				80	0.400	3.400	III	9.80	8.90	4.10	4.00	0.80	13.40	2.68	Gel
				100	0.500	3.000	III	9.70	8.70	3.30	3.85	1.55	12.89	5.19	Gel
				120	0.600	2.600	II	10.80	4.20	-	5.40	1.20	18.08	4.02	Gel
				140	0.700	2.200	II	10.70	2.60	-	5.35	2.75	17.92	9.21	Gel
				160	0.800	1.800	II	10.70	2.60	-	5.35	2.75	17.92	9.21	Gel
2.4	0.1991	320	4680	40	0.200	4.200	I	11.00	6.10	-	0.60	5.50	3.01	27.62	Gel
				50	0.250	4.000	I	9.90	7.50	-	2.55	4.95	12.81	24.86	Gel
				55	0.275	3.900	I	9.90	7.80	-	2.85	4.95	14.31	24.86	Gel
				60	0.300	3.800	III	10.80	9.00	2.00	3.60	3.40	18.08	17.08	Gel
				80	0.400	3.400	III	10.70	10.00	4.80	4.65	0.55	23.36	2.76	Gel
				85	0.425	3.300	II	9.50	4.10	-	4.75	0.65	23.86	3.26	Gel
				90	0.450	3.200	II	9.80	4.45	-	4.90	0.45	24.61	2.26	Gel
				100	0.500	3.000	II	10.90	5.15	-	5.45	0.30	27.37	1.51	Gel
				120	0.600	2.600	II	10.80	5.10	-	5.40	0.30	27.12	1.51	Gel

Table A-5a Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing heptanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
10.8	0.8959	1465	3535	1	0.005	4.980	IV	11.40	-	-	5.70	5.70	6.36	6.36	Gel
				2	0.010	4.960	IV	11.25	-	-	5.63	5.63	6.28	6.28	Gel
				3	0.015	4.940	IV	11.30	-	-	5.65	5.65	6.31	6.31	Gel
				4	0.020	4.920	IV	11.40	-	-	5.70	5.70	6.36	6.36	Gel
				5	0.025	4.900	IV	11.20	-	-	5.60	5.60	6.25	6.25	Gel
				20	0.100	4.600	IV	11.00	-	-	5.50	5.50	6.14	6.14	Gel
				25	0.125	4.500	IV	11.40	-	-	5.70	5.70	6.36	6.36	Gel
				30	0.150	4.400	II	11.40	1.15	-	5.70	4.55	6.36	5.08	Gel
				40	0.200	4.200	II	11.50	2.25	-	5.75	3.50	6.42	3.91	Gel
				60	0.300	3.800	II	11.40	3.50	-	5.70	2.20	6.36	2.46	Gel
9.6	0.7963	1305	3695	100	0.500	3.000	II	11.60	4.25	-	5.80	1.55	6.47	1.73	Gel
				5	0.025	4.900	IV	10.00	-	-	5.19	5.19	6.52	6.52	Gel
				10	0.050	4.800	IV	10.40	-	-	5.39	5.39	6.77	6.77	Gel
				20	0.100	4.600	IV	11.50	-	-	5.62	5.62	7.06	7.06	Gel
				25	0.125	4.500	IV	10.30	-	-	5.34	5.34	6.71	6.71	Gel
				30	0.150	4.400	II	10.30	0.08	-	5.34	4.98	6.71	6.26	Gel
				40	0.200	4.200	II	11.20	2.15	-	5.47	3.41	6.86	4.29	Gel
				60	0.300	3.800	II	11.20	3.35	-	5.47	2.17	6.86	2.73	Gel

Table A-5b Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing heptanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
8.4	0.6968	1140	3860	5	0.025	4.900	IV	10.90	-	-	5.45	5.45	7.82	7.82	Gel
				10	0.050	4.800	IV	10.80	-	-	5.40	5.40	7.75	7.75	Gel
				20	0.100	4.600	IV	11.40	-	-	5.70	5.70	8.18	8.18	Gel
				25	0.125	4.500	II	11.00	3.70	-	5.50	1.80	7.89	2.58	Gel
				30	0.150	4.400	II	10.90	2.75	-	5.45	2.70	7.82	3.87	Gel
				40	0.200	4.200	II	11.20	1.85	-	5.60	3.75	8.04	5.38	Gel
				60	0.300	3.800	II	11.40	3.45	-	5.70	2.25	8.18	3.23	Gel
				80	0.400	3.400	II	11.20	3.95	-	5.60	1.65	8.04	2.37	Gel
				100	0.500	3.000	II	11.30	4.35	-	5.65	1.30	8.11	1.87	Gel
7.2	0.5972	975	4025	10	0.050	4.800	I	10.10	7.80	-	2.75	5.05	4.60	8.46	Gel
				15	0.075	4.700	IV	10.05	-	-	5.03	5.03	8.41	8.41	Gel
				20	0.100	4.600	IV	11.30	-	-	5.65	5.65	9.46	9.46	Gel
				30	0.150	4.400	IV	9.90	-	-	4.95	4.95	8.29	8.29	Gel
				35	0.175	4.300	II	10.00	0.05	-	5.00	4.95	8.37	8.29	Gel
				40	0.200	4.200	II	11.30	0.70	-	5.65	4.95	9.46	8.29	Gel
				60	0.300	3.800	II	11.20	3.20	-	5.60	2.40	9.38	4.02	Gel
				80	0.400	3.400	II	11.30	3.95	-	5.65	1.70	9.46	2.85	Gel

Table A-5c Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing heptanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
6.0	0.4977	815	4185	10	0.050	4.800	I	11.10	6.50	-	0.95	5.55	1.91	11.15	Gel
				15	0.075	4.700	I	10.80	7.70	-	2.30	5.40	4.62	10.85	Gel
				20	0.100	4.600	IV	11.35	-	-	5.68	5.68	11.40	11.40	Gel
				40	0.200	4.200	IV	11.10	-	-	5.55	5.55	11.15	11.15	Gel
				45	0.225	4.100	II	10.80	1.05	-	5.40	4.35	10.85	8.74	Gel
				50	0.250	4.000	II	10.95	3.00	-	5.48	2.48	11.00	4.97	Gel
				60	0.300	3.800	II	11.10	3.35	-	5.55	2.20	11.15	4.42	Gel
				80	0.400	3.400	II	11.15	4.80	-	5.58	0.78	11.20	1.56	Gel
				100	0.500	3.000	II	11.10	4.40	-	5.55	1.15	11.15	2.31	Gel
4.8	0.3982	650	4350	20	0.100	4.600	I	11.30	7.70	-	2.05	5.65	5.15	14.19	Gel
				25	0.125	4.500	I	10.00	7.95	-	2.95	5.00	7.41	12.56	Gel
				30	0.150	4.400	IV	9.80	-	-	4.90	4.90	12.31	12.31	Gel
				40	0.200	4.200	IV	9.90	-	-	4.95	4.95	12.43	12.43	Gel
				50	0.250	4.000	IV	9.80	-	-	4.90	4.90	12.31	12.31	Gel
				55	0.275	3.900	II	10.00	0.60	-	5.00	4.40	12.56	11.05	Gel
				60	0.300	3.800	II	11.00	2.30	-	5.50	3.20	13.81	8.04	Gel
				80	0.400	3.400	II	11.10	3.05	-	5.55	2.50	13.94	6.28	Gel
				100	0.500	3.000	II	11.10	3.00	-	5.55	2.55	13.94	6.40	Gel

Table A-5d Winsor type microemulsion and phase volume at different SDS and NaCl concentration for microemulsion containing heptanoic acid.

SDS (%)	SDS (g)	Fatty acid (μ l)	Hexane (μ l)	NaCl (g/l)	NaCl (g)	H ₂ O (ml)	Winsor type	Labeled volume (ml)			V _o	V _w	SP _o	SP _w	Gel
3.6	0.2986	490	4510	20	0.100	4.600	I	11.10	6.15	-	0.60	5.55	2.01	18.59	Gel
				40	0.200	4.200	I	10.80	9.00	-	3.60	5.40	12.06	18.08	Gel
				45	0.225	4.100	III	10.70	8.00	4.40	2.65	0.95	8.87	3.18	Gel
				50	0.250	4.000	III	10.65	8.00	4.90	2.68	0.43	8.96	1.42	Gel
				60	0.300	3.800	II	10.85	4.80	-	5.43	0.63	18.17	2.09	Gel
				80	0.400	3.400	II	10.95	4.35	-	5.48	1.13	18.33	3.77	Gel
				100	0.500	3.000	II	11.00	4.45	-	5.50	1.05	18.42	3.52	Gel
2.4	0.1991	325	4675	20	0.100	4.600	I	10.80	5.80	-	0.40	5.40	2.01	27.12	Gel
				40	0.200	4.200	I	10.70	6.00	-	0.65	5.35	3.26	26.87	Gel
				45	0.225	4.100	I	9.80	6.15	-	1.25	4.90	6.28	24.61	Gel
				50	0.250	4.000	III	9.50	6.30	3.60	1.55	1.15	7.79	5.78	Gel
				65	0.325	3.700	II	9.00	4.20	-	4.50	0.30	22.60	1.51	Gel
				80	0.400	3.400	II	10.90	5.20	-	5.45	0.25	27.37	1.26	Gel
				100	0.500	3.000	II	10.80	5.00	-	5.40	0.40	27.12	2.01	Gel
				120	0.600	2.600	II	9.50	4.00	-	4.75	0.75	23.86	3.77	Gel
				140	0.700	2.200	II	9.70	4.40	-	4.85	0.45	24.36	2.26	Gel
				160	0.800	1.800	II	9.50	4.20	-	4.75	0.55	23.86	2.76	Gel

Table A-6 Optimum solubilization parameter and optimum salinity at different SDS concentration.

C3 fatty acid			C4 fatty acid			C5 fatty acid			C6 fatty acid			C7 fatty acid		
%SDS	S*	SP*	%SDS	S*	SP*	%SDS	S*	SP*	%SDS	S*	SP*	%SDS	S*	SP*
24	85	3.1	21.6	0	3.1	13.2	15	5.2	10.8	10	6.3	10.8	0	6.3
21.6	135	3.3	18	35	4	9.6	35	7.2	9.6	25	6.2	9.6	0	6.5
19.8	155	3.7	15.6	55	4.3	7.2	60	9.2	8.4	20	7.3	8.4	0	7.8
18	183	3.7	13.2	71	4.7	6	70	10.9	7.2	25	8.3	7.2	15	8.5
15.6	184	3.9	10.8	85.8	5.3	4.8	82	10.7	6	35	11	6	20	11.4
13.2	191	4.1	7.8	108	6.4	3.6	79	11.2	4.8	50	12	4.8	30	12.5
10.2	197	4.7	4.8	138	9.5				3.6	65	11	3.6	43	10.2
7.2	222	5.5							2.4	59	17.7	2.4	49	7

NOTE: Example of the calculation for microemulsion preparations

Example SDS 1.2%, NaCl 100 g/L in microemulsion containing propanoic acid

Assuming the microemulsion composing of 5 mL of distilled water and 5 mL of hexanes

$$\text{g of distilled water} = (5 \text{ mL})(\text{density } 1 \text{ g/mL}) = 5 \text{ g}$$

$$\text{g of hexanes} = (5 \text{ mL})(\text{density } 0.659 \text{ g/mL}) = 3.295 \text{ g}$$

$$\text{g of the system} = 5 \text{ g} + 3.295 \text{ g} = 8.295 \text{ g}$$

$$\text{g of SDS} = (1.2\%)(8.295) = 0.0995 \text{ g}$$

$$\text{g of SDS/ g of propanoic acid} = 40/60$$

$$\text{mL of propanoic acid} = (0.0995 \text{ g SDS})(60 \text{ g propanoic acid}/40 \text{ g SDS})/(\text{density } 0.992 \text{ g/mL})$$

$$= 0.1505 \text{ mL}$$

$$\text{mL of hexanes} = 5 - 0.1505 = 4.8495 \text{ mL}$$

$$\text{g of NaCl} = (100 \text{ g/L})(1 \text{ L}/1000 \text{ mL})(5 \text{ mL of water in system})$$

$$= 0.5 \text{ g}$$

APPENDIX B

Example of pH Calculation

1. pH Calculation from pKa for microemulsion system containing pentanoic acid at 9.6% SDS.

$$\text{pH of DI water} = 5.62$$

$$\text{Initial H}^+ \text{ concentration} = 2.4 \times 10^{-6} \text{ Molar}$$

$$\text{Total amount of SDS in system} = 9.6\% = 0.7963 \text{ g}$$

$$\text{SDS/pentanoic acid} = 40:60 \text{ by wt.}$$

$$\text{Amount of pentanoic acid in system} = 1.1945 \text{ g} = 2.3421 \text{ Molar}$$

$$\text{pK}_a \text{ of pentanoic acid} = 4.84$$

$$\text{K}_a \text{ of pentanoic acid} = 1.445 \times 10^{-5}$$

Assume y molar of pentanoic acid dissociate:

$$\text{K}_a = 1.445 \times 10^{-5} = \frac{(y)(2.4 \times 10^{-6} + y)}{(2.3421 - y)}$$

By solving the above equation,

$$y = 5.8091 \times 10^{-3} \text{ Molar}$$

$$[\text{H}^+] = 2.4 \times 10^{-6} + 5.8091 \times 10^{-3} = 5.8115 \times 10^{-3}$$

$$\text{pH of microemulsion} = -\log(5.8115 \times 10^{-3}) = 2.24$$

2. pH Calculation from pKa and solubility of fatty acid in water for microemulsion system containing pentanoic acid at 9.6% SDS.

$$\text{pH of DI water} = 5.62$$

$$\text{Initial H}^+ \text{ concentration} = 2.4 \times 10^{-6} \text{ Molar}$$

$$\text{Amount of SDS in system} = 9.6\% = 0.7963 \text{ g}$$

$$\text{SDS/pentanoic acid} = 40:60 \text{ by wt.}$$

$$\text{Total amount of pentanoic acid in system} = 1.1945 \text{ g} = 2.3421 \text{ Molar}$$

$$\text{Concentration of pentanoic acid in water} = 3.63 \times 10^{-1} \text{ Molar}$$

$$\text{pK}_a \text{ of pentanoic acid} = 4.84$$

$$\text{K}_a \text{ of pentanoic acid} = 1.445 \times 10^{-5}$$

Assume y molar of pentanoic acid dissociate:

$$K_a = 1.445 \times 10^{-5} = \frac{(y)(2.4 \times 10^{-6} + y)}{(3.63 \times 10^{-1} - y)}$$

By solving the above equation.

$$y = 2.281 \times 10^{-3} \text{ Molar}$$

$$[H^+] = 2.4 \times 10^{-6} + 2.281 \times 10^{-3} = 2.283 \times 10^{-3}$$

$$\text{pH of microemulsion} = -\log(2.283 \times 10^{-3}) = 2.64$$

APPENDIX C

Experimental Data from Conductivity Measurement

Table C-1 Conductivity of microemulsion containing propanoic acid.

SDS (%)	NaCl (g/l)	Conductivity		SDS (%)	NaCl (g/l)	Conductivity	
		Peak area	mS/cm			Peak area	mS/cm
21.6	120	2537.13	97.84	13.2	170	3198.36	101.31
	130	2680.32	104.52		180	3160.98	100.25
	140	3305.45	104.33		185	4289.4	132.08
	150	3489.86	109.53		200	3660.66	114.35
	170	3547.39	111.15		210	3220.59	101.93
	180	3596.55	112.54		215	2255.45	85.06
	185	3424.66	107.69		220	2204.69	82.81
	190	3278.48	103.57	7.2	190	4063.3	125.70
	200	2153.63	80.56		200	3615.51	113.07
18	160	3876.4	120.43		210	5430.5	164.27
	170	2885.52	114.31		220	5773.49	173.94
	181	2846	112.41		230	2010.39	74.33
	182	2742.78	107.47		245	1554.07	55.30
	183	2761.29	108.35		250	1589.34	56.73
	184	2743.4	107.50				
	200	2739.74	107.33				

Table C-2 Conductivity of microemulsion containing butanoic acid.

SDS (%)	NaCl (g/l)	Conductivity		SDS (%)	NaCl (g/l)	Conductivity	
		Peak area	mS/cm			Peak area	mS/cm
21.6	0	1057.13	36.00	10.8	60	1619.40	57.95
	1	1058.90	36.07		70	1579.04	56.31
	2	1060.32	36.12		80	1422.87	50.06
	3	1091.03	37.27		85	1445.39	50.95
	4	1102.49	37.70		90	1188.77	40.97
	5	1129.78	38.73		100	1246.60	43.19
	10	1226.00	42.40	4.8	100	3576.70	111.98
	20	1363.23	47.72		120	2501.50	96.20
	40	1629.52	58.36		130	2119.50	79.06
	60	1891.67	69.26		135	1716.07	61.91
	65	1922.46	70.56		145	1225.00	42.36
	70	1952.91	71.86		150	1154.65	39.67
	80	1963.85	72.33		155	897.70	30.13
	100	1990.01	73.45		160	935.55	31.51
15.6	40	1096.58	37.48		170	1007.42	34.15
	50	1410.46	49.57				
	55	1382.24	48.46				
	60	1452.45	51.23				
	65	1580.81	56.38				
	70	1589.11	56.72				
	80	1534.77	54.52				

Table C-3 Conductivity of microemulsion containing pentanoic acid.

SDS (%)	NaCl (g/l)	Conductivity		SDS (%)	NaCl (g/l)	Conductivity	
		Peak area	mS/cm			Peak area	mS/cm
13.2	10	732.65	24.20	6.0	50	1278.15	44.40
	20	817.39	27.22		60	1135.99	38.97
	30	838.00	27.96		65	1125.80	38.58
	40	1062.00	36.19		70	1024.86	34.80
	50	1191.75	41.09		80	1078.90	36.82
	70	966.35	32.64		85	656.82	21.54
	90	907.95	30.50		110	684.25	22.50
	110	893.50	29.97	3.6	40	1613.05	57.69
9.6	10	872.40	29.21		60	1935.11	71.10
	20	919.35	30.92		80	454.68	14.60
	30	952.60	32.13		100	483.56	15.57
	35	940.15	31.68		120	471.68	15.17
	55	1194.20	41.18				
	60	1114.35	38.15				
	80	953.45	32.16				

Table C-4 Conductivity of microemulsion containing hexanoic acid.

SDS (%)	NaCl (g/l)	Conductivity		SDS (%)	NaCl (g/l)	Conductivity	
		Peak area	mS/cm			Peak area	mS/cm
9.6	20	621.54	20.31	4.8	40	980.81	33.17
	25	659.49	21.63		45	697.67	22.97
	30	675.41	22.19		50	674.67	22.16
	40	809.65	26.95		60	734.37	24.26
	45	884.89	29.66		80	1033.6	35.13
	60	846.69	28.28		90	1139.1	39.09
	80	855.72	28.60		95	1322.5	46.12
	100	897.72	30.13		100	540.86	17.52
7.2	20	628.65	20.55		120	223.93	6.97
	25	609.73	19.90	2.4	40	1956.19	72.00
	30	564.16	18.32		50	1831.60	66.72
	50	840.57	28.06		55	1818.12	66.16
	55	721.52	23.81		60	473.34	15.23
	80	759.68	25.16		80	380.75	12.12
	100	751.59	24.87		85	112.07	3.39
					90	145.27	4.45
					100	149.16	4.57
					120	224.39	6.99

Table C-5 Conductivity of microemulsion containing heptanoic acid.

SDS (%)	NaCl (g/l)	Conductivity		SDS (%)	NaCl (g/l)	Conductivity	
		Peak area	mS/cm			Peak area	mS/cm
9.6	10	500.71	16.15	4.8	20	684.96	22.52
	20	513.24	16.58		25	557.65	18.10
	25	553.18	17.95		30	529.24	17.13
	30	662.44	21.73		40	556.12	18.05
	60	760.96	25.21		50	644.20	21.10
	80	644.35	21.10		55	502.87	16.23
7.2	10	497.56	16.05		60	480.79	15.48
	15	465.57	14.96		80	372.09	11.83
	20	549.89	17.83	2.4	20	1127.54	38.65
	30	573.25	18.64		40	1579.20	56.32
	35	658.07	21.58		45	1960.58	72.19
	40	629.53	20.58		50	751.73	24.88
	60	655.27	21.48		60	257.79	8.07
	80	583.94	19.01		65	124.80	3.80
					70	165.32	5.09
					80	163.83	5.04
					100	194.40	6.02
					120	220.74	6.87

APPENDIX D

Experimental Data from Interfacial Tension Measurement

Table D-1 Interfacial tension of microemulsion containing propanoic acid.

SDS (%)	NaCl (g/l)	Winsor type	Interfacial tension	
			γ_{mo}	γ_{mw}
15.6	150	I	4.12E-02	-
	170	I	2.11E-02	-
	180	I	2.02E-02	-
	185	III	2.76E-02	4.27E-03
	190	III	8.71E-03	1.38E-02
	200	III	4.08E-03	3.28E-02
	205	II	-	3.72E-02
	210	II	-	6.69E-02
	230	II	-	1.16E-01
13.2	170	I	1.00E-02	-
	180	I	7.08E-03	-
	185	III	5.62E-03	2.34E-03
	200	III	3.24E-03	4.07E-03
	210	III	1.91E-03	6.46E-03
	215	II	-	7.94E-03
	220	II	-	1.41E-02
10.2	180	I	1.10E-02	-
	185	I	4.17E-03	-
	190	III	2.57E-03	6.76E-04
	210	III	6.76E-04	1.82E-03
	230	III	-	2.75E-03
	235	III	-	3.55E-03
	240	II	-	4.37E-03
	270	II	-	3.55E-03
7.2	190	I	7.59E-03	-
	195	III	3.39E-03	-
	210	III	2.09E-03	5.25E-04
	220	III	1.32E-03	8.32E-04
	230	III	2.95E-04	1.02E-03
	240	III	-	9.33E-04
	245	II	-	1.62E-03
	250	II	-	4.07E-03

Table D-2 Interfacial tension of microemulsion containing butanoic acid.

SDS (%)	NaCl (g/l)	Winsor type	Interfacial tension	
			γ_{mo}	γ_{mw}
10.8	60	I	1.74E-02	-
	70	I	8.13E-03	-
	75	I	6.46E-03	-
	80	III	5.01E-03	3.55E-03
	85	III	3.24E-03	4.27E-03
	90	II	-	5.98E-03
7.8	100	II	-	6.46E-03
	60	I	1.62E-02	-
	80	I	6.76E-03	-
	100	I	3.09E-03	-
	105	III	2.69E-03	1.66E-03
	110	III	2.51E-03	2.04E-03
	115	III	1.95E-03	2.69E-03
	120	II	-	3.47E-03
	140	II	-	4.47E-03
	4.8	100	I	2.97E-02
120		I	5.75E-03	-
130		I	3.09E-03	-
135		III	2.04E-03	-
140		III	1.58E-03	1.45E-03
145		III	1.29E-03	1.95E-03
150		III	-	2.40E-03
155		II	-	3.09E-02

Table D-3 Interfacial tension of microemulsion containing pentanoic acid.

SDS (%)	NaCl (g/l)	Winsor type	Interfacial tension	
			γ_{mo}	γ_{mw}
4.8	50	I	7.95E-02	-
	70	I	7.46E-02	-
	75	III	7.00E-02	5.95E-02
	80	III	6.40E-02	6.22E-02
	90	II	-	7.47E-02
	110	II	-	1.06E-01
3.6	40	I	1.52E-01	-
	60	I	8.18E-02	-
	70	III	5.62E-02	4.32E-02
	80	III	4.65E-02	6.87E-02
	90	III	-	7.50E-02
	100	II	-	6.87E-02

Table D-4 Interfacial tension of microemulsion containing hexanoic acid.

SDS (%)	NaCl (g/l)	Winsor type	Interfacial tension	
			γ_{mo}	γ_{mw}
3.6	40	I	7.89E-02	-
	50	I	1.35E-01	-
	55	I	1.56E-01	-
	60	III	7.41E-02	4.99E-02
	80	III	5.06E-02	9.12E-02
	100	III	3.78E-02	1.28E-01
	120	II	-	1.39E-01
2.4	140	II	-	1.91E-01
	160	II	-	2.06E-01
	40	I	2.19E-02	-
	50	I	1.95E-02	-
	55	I	1.38E-02	-
	60	III	5.37E-03	5.13E-03
	80	III	3.39E-03	1.35E-02
2.4	85	II	-	2.88E-02
	90	II	-	6.25E-02
	100	II	-	1.73E-01
	120	II	-	2.00E-01

Table D-5 Interfacial tension of microemulsion containing heptanoic acid.

SDS (%)	NaCl (g/l)	Winsor type	Interfacial tension	
			γ_{mo}	γ_{mw}
3.6	20	I	2.24E-01	-
	40	I	1.91E-01	-
	45	III	1.92E-01	1.38E-02
	50	III	4.29E-02	3.63E-02
	60	II	-	1.20E-01
	80	II	-	1.38E-01
	100	II	-	1.62E-01
2.4	20	I	7.35E-02	-
	40	I	7.59E-02	-
	45	I	7.41E-02	-
	50	III	5.75E-02	3.98E-02
	60	III	3.98E-02	6.46E-02
	65	II	-	1.59E-01
	70	II	-	1.21E-01
	80	II	-	2.63E-01
	100	II	-	2.72E-01
	120	II	-	2.68E-01

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