# การศึกษาความคงตัวของสารละลายโพวิโดน-ไอโอดีน ๑๐%

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วิทยานิพนธ์นี้ เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญา เภสัชศาสตรมหาบัณฑิต ภาควิชา เภสัช เคมี บัณฑิตวิทยาลัย จุฬาลงกรณ์มหาวิทยาลัย

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### A STUDY OF STABILITY OF 10% POVIDONE-IODINE SOLUTION

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ว่าที่ ร.ต.อ. (ญ) มาลัย ศิริทองถาวร : การศึกษาความคงตัวของสารละลายโพวิโคน-ไอโอดีน 10% (A STUDY OF STABILITY OF 10% POVIDONE-IODINE) อ.ที่ปรึกษา : ผศ. สมเกียรติ รุจิรวัฒน์, อ.ที่ปรึกษาร่วม : อ.ดร. มีตร ปทีปาณิช, 183 หน้า, ISBN 974-581-026-6

การศึกษาความคงตัวของสารละลาย 10% น้ำหนักโดยปริมาตรของโพวิโดน-ไอโอดีนใน สารละลายบัฟเฟอร์ ได้วางแผนการทดลอง เพื่อศึกษาอิทธิพลของบัฟเฟอร์ที่ใช้ในทางเภลัชกรรม แหล่งน้ำ และภาชนะบรรจุ ต่อสารละลายโพวิโดน-ไอโอดีน การวิเคราะห์ทาปริมาณทำโดย วิเคราะห์ available iodine ตามวิธีใน USP.XX พบว่า การเตรียมสารละลายโพวิโดน-ไอโอดีน ที่มี pH เริ่มต้น 5.5 และใช้ความเข้มข้นของบัฟเฟอร์เท่ากัน การเตรียมในฟอสเฟตบัฟเฟอร์ จะมี ความคงตัวที่ดีกว่าการเตรียมในอะซีเตท และซิเตรทบัฟเฟอร์ แต่การใช้ซิเตรทบัฟเฟอร์ จะรักษา pH ของสารละลายได้ดีกว่าฟอสเฟต และอะซีเตทบัฟเฟอร์ ในการเตรียมสารละลายของโพวิโดน – ไอโอดีนควรใช้น้ำกลั่น เป็นตัวทำละลาย และบรรจุในขวดโพลีเอธิลีน

ปริมาณ available iodine ที่เหลืออยู่ จะลดลงอย่างเร็วมากในช่วงต้นของการสลายตัว หลังจากนั้น อัตราการสลายตัวที่ปรากฏจะค่อย ๆ ลดลง จนในที่สุดคาดว่า การสลายตัวอาจเป็นแบบ ปฏิกิริยาอันดับศูนย์ ดังนั้นในการเตรียมสูตรตำรับสารละลาย 10% น้ำหนักโดยปริมาตรของ โพวิโดน-ไอโอดีน ในบัฟเฟอร์ควรเติมโพวิโดน-ไอโอดีนเกินไว้ เพื่อรักษาปริมาณ available iodine หลังจากเกิดการสลายตัวอย่างรวดเร็วในช่วงแรก

ภาควิชา	เภสัชเคมี
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ปีการศึกษา	2534

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## ยินที่ดีแพบทุนถูกกล อินกร แก้จาก การสุดเกษาเล่น แต่การ

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MALAI SIRITONGTHAVORN, POLICE CAPTAIN: A STUDY OF STABILITY 10% POVIDONE-IODINE SOLUTION. THESIS ADVISOR: ASSIST.PROF. SOMKIAT RUJIRAWAT, M.S.(PHARM), THESIS CO-ADVISOR: INSTRUCTOR MITR PATHIPVANICH, Ph.D., 183 PP. ISBN 974-581-026-6.

The stability testing program of 10% w/v buffered solution was designed to assess the influence of pharmaceutical buffers, sources of solvent and packaging materials on the degradation of PVP-I solution. The quantitative analysis was performed by determining of available iodine as official in USP XX.

It was found that, at the same buffer concentration and the initial pH of 5.5, 10% w/v PVP-I solution in phosphate buffer was more stable than in acetate or citrate buffer whereas the solution in citrate buffer could maintain pH better than in the other two. The solvent should be distilled water. The solution of 10% w/v PVP-I buffered solution should be filled in polyethylene bottle.

Available iodine decreased rapidly at the initial state of the degradation. Then, the appearance degradation rate was gradually decreased. Finally, the degradation was postulated to be zero order reaction. Thus, in case of 10% w/v PVP-I solution formulation in buffer, PVP-I should be excessively added to maintain the quantity of available iodine after initial rapid degradation.

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### LIST OF ABBREVIATIONS

#### **Abbreviation**

#### Tern

PVP polyvinylpyrrolidone/povidone
PVP-I povidone-iodine
g gram

mg milligram

ml milliliter

M molar

m mole

ppm part per million

mµ millimicron

ln natural logarithm