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APPENDICES

Appendix A Experimental Data of Liquid Feed Calibration of GC 5890

1. Benzene

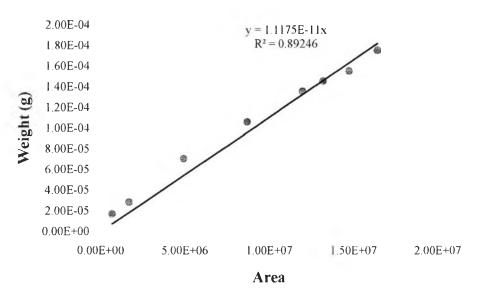


Figure A1 Calibration curve of benzene.

2. Ethanol

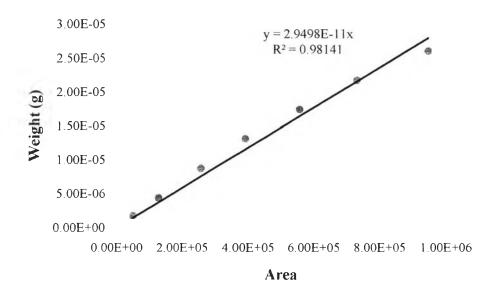
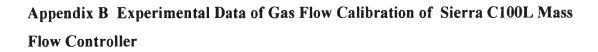


Figure A2 Calibration curve of ethanol.



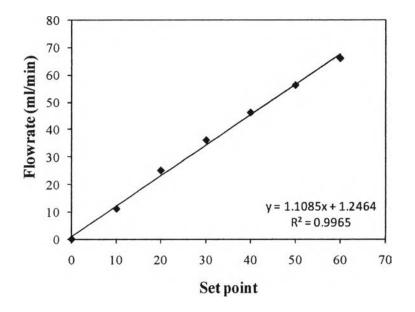
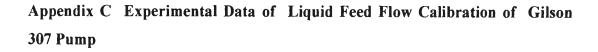


Figure B1 Calibration curve of nitrogen.



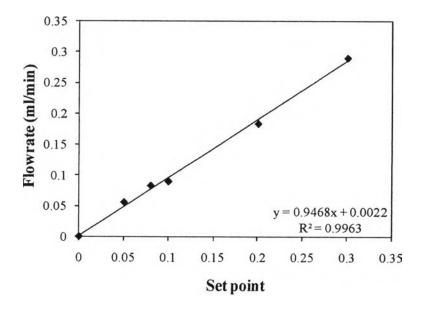


Figure C1 Calibration curve of liquid feed.

Appendix D Calculation of Si/Al Ratio and Theoretical Acidity

From the chemical composition determined by XRF method, the SiO_2/Al_2O_3 ratio is calculated as follows:

In the case of HZ5-195(3),

$SiO_2 =$	98.682 wt%	$Al_2O_3 =$	1.075 wt%
$SiO_2 =$	1.647568 mol	$Al_2O_3 =$	0.009876 mol
$SiO_2/Al_2O_3 =$	166.8204		

From the chemical composition determined by XRF method, the theoretical acidity of zeolite is calculated as follows:

The general formula of HZSM-5 is $H_nAl_nSi_{96-n}O_{192}$ In the case of HZ5-195(3) with,

Si ___ 46.278 wt% Al 10.5273 wt% = Si Al = 1.64775 mol \equiv 0.01954 mol Si/Al =84.3143 From $Al_n Si_{96-n}O_{192}$, Si/Al =84.3143 = (96-n)/n85.3143n 96 = n = 1.12525 So, Si 94.87475 = Al = 1.12525 From formula the above, the general of HZSM-5 is $H_{1,12525}Al_{1,12525}Si_{94,87475}O_{192}$. The weight of unit cell of HZSM-5 (U) is 1.12525(1) + 1.12525(26.98) + 94.87475(28.09) + 192(16.00)U = U = 5768.5162 g

The theoretical acidity $([H^+])$ of HZSM-5 (B1) is

 $[H^+] = 1.12525/5768.5162$

 $[H^+] = 0.19507 \text{ mmol/g}$

Appendix E The Other Catalyst Preparation

In this work, the other method for synthesizing ZSM-5 zeolites was adopted from the work of Rugwong and coworkers (Rugwong*et al.*, 2012) The detailed produre is described below.

This technique used NaOH as a mineralizing agent. HZSM-5 samples were synthesized from hydrogel solutions via hydrothermal synthesis with the following molar composition

195SiO₂:Al₂O₃:20TPABr: xH2O

where x = 3900, 4875, 5850, and 7800.

First, Ludox and 4/5 of the distilled water were stirred at 350 rpm. Under stirring, aqueous solution of NaOH was slowly dropped into the Ludox solution to obtain a pH value of 10.5. The obtained mixture was stirred for 1 h at room temperature while controlling the pH value at 10.5.

The rest of the distilled water and $Al(NO_3)_3 \cdot 9H_2O$ were mixed together until $Al(NO_3)_3 \cdot 9H_2O$ was completely dissolved. The $Al(NO_3)_3 \cdot 9H_2O$ solution and TPABr were placed in the Ludox solution beaker followed by conditioning the pH value to 10.5 with NaOH solution. The mixture was stirred at 350 rpm for 24 h while controlling the pH value at 10.5.

The gel was placed into an autoclave for hydrothermal and heated in an oven at various conditions. After the hydrothermal synthesis, the autoclave was cooled down to room temperature. The gel forms powder after hydrothermal. Next the powder was washed for reducing pH from 10.5 to 7 by distilled water. And then the powder was dried at 80 °C overnight, and calcined to remove the precursor at 550 °C for 5 h.

After first calcination, the as-synthesized ZSM-5 was exchanged with 1 M NH4NO3 solution for three times at 80 °C, and then washed with distilled water to remove the nitrate ions. The resultant zeolite was dried overnight at 80 °C and calcined in flowing dry air at 550 °C for 5 h to obtain the acidic form of the zeolite (HZSM-5).

Appendix F Experimental Data of Catalytic Activity Test for Ethylation of Benzene with Ethanol over synthesized HZSM-5 Catalyst.

Table F1 Catalytic activity testing over HZSM-5 with different synthesis at temperature 500°C, B/E = 4, WHSV = 20 h⁻¹.

		Conversion (%)						
Component	Catalyst	15	60	130	200	270	340	410
		min	min	min	min	min	min	min
	HZ5-195(3)	8.02	8.40	8.35	7.83	8.23	8.13	8.17
Danzana	HZ5-195(7)	6.55	7.51	7.95	7.77	7.46	7.46	7.44
Benzene	HZ5-260(7)	5.73	6.27	7.00	7.14	7.03	7.08	7.13
	HZ5-280(7)	5.82	7.35	6.94	6.43	6.54	6.93	6.36
	HZ5-195(3)	99.96	99.97	99.98	99.98	99.98	99.98	99.99
Ethanal	HZ5-195(7)	99.96	99.96	99.98	99.96	99.98	99.96	99.99
Ethanol	HZ5-260(7)	99.96	99.97	99.97	99.97	98.99	99.95	99.99
	HZ5-280(7)	99.97	99.99	99.98	99.99	99.92	100.00	99.99

Table F2 Catalytic activity testing on different temperature for HZ5-195(3), B/E = 4, WHSV = 20 h⁻¹.

	Temperature	Conversion (%)						
Component	Temperature (°C)	15	60	130	200	270	340	410
	(C)	min	min	min	min	min	min	min
	400	12.89	13.38	13.36	13.31	12.75	13.19	13.82
Benzene	450	10.79	10.95	10.76	11.58	11.20	10.83	11.10
	500	8.02	8.40	8.35	7.83	8.23	8.13	8.17
	400	99.97	99.98	99.99	99.98	99.99	99.98	99.99
Ethanol	450	99.97	99.98	99.97	99.90	99.98	99.99	100.00
	500	99.96	99.97	99.98	99.98	99.98	99.98	99.99

Table F3 Catalytic activity testing on different feed molar ratio for HZ5-195(3), WHSV = 20 h⁻¹, T = 500°C.

		Conversion (%)						
Component	B/E ratio	15	60	130	200	270	340	410
		min	min	min	min	min	min	min
	2	8.65	13.09	13.59	13.08	11.02	12.73	12.76
Benzene	4	8.02	8.40	8.35	7.83	8.23	8.13	8.17
	6	5.06	5.36	5.15	5.28	5.29	4.99	4.86
	2	99.96	99.98	97.99	95.98	99.98	99.97	99.99
Ethanol	4	99.96	99.97	99,98	99.98	99.98	99.98	99.99
	6	98.95	98.98	99.98	99.98	99.98	99.99	99.99

Table F4 Catalytic activity testing on different feed ratio for HZ5-195(3), B/E = 4, T = 500 °C.

	_	Conversion (%)						
Component	WHSV (h^{-1})	15	60	130	200	270	340	410
		min	min	min	min	min	min	min
Bonzono	15	7.34	7.56	7.47	7.03	7.47	7.12	7.23
Benzene	20	8.02	8.40	8.35	7.83	8.23	8.13	8.17
Ethanol	15	99.97	98.99	99.98	99.99	99.99	99.98	100.00
Ethanol	20	99.96	99.97	99.98	99.98	99.98	99.98	99.99

Table F5 Product selectivity of liquid sample over HZSM-5 with different synthesis
at temperature 500°C, $B/E = 4$, $WHSV = 20 h^{-1}$, and TOS 410 min.

Component	Product selectivity (%)						
Component	HZ5-195(3)	HZ5-195(7)	HZ5-260(7)	HZ5-280(7)			
Ethylene	0.22	0.19	0.15	0.14			
Methanol	0.23	0.25	0.36	0.45			
Toluene	2.29	2,53	2.73	3.91			
Ethyl Benzene	93.41	92.76	92.64	89.36			
m-Xylene	0.54	0.62	0.60	0.74			
p-Xylene	1.12	1.03	1.14	2.13			
o-Xylene	-	-	_	_			
Cumene	0.05	0.08	0.06	0.12			
Propyl-benzene	0.11	0.17	0.12	0.21			
p,m-Ethyltoluene	0.03	0.05	0.00	0.09			
o-Ethyltoluene	0.03	0.04	0.03	0.08			
1,2,3 TMB	-	-	-	0.04			
1,3-diethylbenzene	0.09	0.11	0.09	0.20			
1,4-diethylbenzene	0.12	0.16	0.12	0.26			
1,2-diethylbenzene	0.04	0.06	0.04	0.19			
2-butenylbenzene	0.46	0.69	0.47	0.55			
l-butenylbenzene	1.25	1.27	1.45	1.51			

Commont	Produ	ict selectivit	ty (%)
Component	400 (°C)	450 (°C)	500 (°C)
Ethylene	0.17	-	0.22
Methanol	0.17	-	0.23
Toluene	1.31	1.67	2.29
Ethyl Benzene	88.12	92.74	93.41
m-Xylene	0.27	0.36	0.54
p-Xylene	0.19	0.54	1.12
o-Xylene	-	-	-
Cumene	0.14	0.06	0.05
Propyl-benzene	0.32	0.17	0.11
p,m-Ethyltoluene	0.09	-	0.03
o-Ethyltoluene	0.34	0.03	0.03
1,2,3 TMB	0.13	-	•
1,3-diethylbenzene	0.22	0.10	0.09
1,4-diethylbenzene	0.33	0.12	0.12
1,2-diethylbenzene	0.24	0.06	0.04
2-butenylbenzene	2.50	1.29	0.46
1-butenylbenzene	5.44	2.86	1.25

Table F7 Product selectivity of liquid sample over HZ5-195(3) at differenttemperature, B/E = 4, WHSV = 20 h⁻¹, and TOS 410 min.

Table F8	Product selectivity of liquid sample over HZ5-195(3) at different feed
molar ratio	of B/E, Temperature 500 °C, WHSV = 20 h^{-1} , and TOS 410 min.

	Produ	ict selectivit	ty (%)
Component	B/E=2	B/E=4	B/E=6
Ethylene	0.27	0.22	0.35
Methanol	0.63	0.23	0.64
Toluene	1.61	2.29	3.96
Ethyl Benzene	92.80	93.41	89.71
m-Xylene	0.51	0.54	0.82
p-Xylene	0.85	1.12	2.43
o-Xylene	-	-	-
Cumene	0.06	0.05	0.12
Propyl-benzene	0,11	0.11	0.18
p,m-Ethyltoluene	0.04	0.03	0.14
o-Ethyltoluene	0.04	0.03	0.13
1,2,3 TMB	-	-	0.07
1,3-diethylbenzene	0.09	0.09	0.22
1,4-diethylbenzene	0.16	0.12	0.25
1,2-diethylbenzene	0.02	0.04	0.11
2-butenylbenzene	0.91	0.46	0.29
1-butenylbenzene	1.90	1.25	0.58

Table F9 Product selectivity of liquid sample over HZ5-195(3) at different WHSV	,
Temperature 500 °C, $B/E = 4$, and TOS 410 min.	

	Product sel	ectivity (%)
Component	WHSV = 15	WHSV = 20
	h-1	h-1
Ethylene	0.20	0.22
Methanol	0.28	0.23
Toluene	2.56	2.29
Ethyl Benzene	92.58	93.41
m-Xylene	0.59	0.54
p-Xylene	1.48	1.12
o-Xylene	-	-
Cumene	0.05	0.05
Propyl-benzene	0.14	0.11
p,m-Ethyltoluene	0.03	0.03
o-Ethyltoluene	0.04	0.03
1,2,3 TMB	-	-
1,3-diethylbenzene	0.09	0.09
1,4-diethylbenzene	0.15	0.12
1,2-diethylbenzene	0.09	0.04
2-butenylbenzene	0.58	0.46
1-butenylbenzene	1.14	1.25

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