

## CHAPTER I

### INTRODUCTION

Normally, flue gas released into the atmosphere from most industries, including petrochemical industry, contains approximately 80 % N<sub>2</sub>, 15 % CO<sub>2</sub>, and 5 % O<sub>2</sub>, as well as 500 ppm SO<sub>2</sub>, 100 ppm CO, and 50 ppm CH<sub>4</sub>. These gases produced by combustion of fossil fuels are regarded as greenhouse gases. Among them, CO<sub>2</sub> has the most adverse impact and causes approximately 55 % of the currently observed global warming. The increasingly accumulated CO<sub>2</sub> in the atmosphere has become a worldwide concerned problem. Therefore, people around the world need to protect the environment and slow down the climate change by reducing emission of CO<sub>2</sub> (Zhang *et al.*, 2008).

To reduce greenhouse gas emission, CO<sub>2</sub> separation and capture technologies are of growing importance. There are four main widely used commercial CO<sub>2</sub> removal processes. They are absorption, adsorption, membrane, and cryogenic processes. However, the most important commercially applied technology for CO<sub>2</sub> removal is the absorption process, which can be called extraction process, by using a liquid solvent.

Liquid solvents used in CO<sub>2</sub> absorption can be divided into two categories: chemical and physical solvents. The processes using amine-based solutions, such as alkanolamines, that chemically react with dissolved CO<sub>2</sub> are commonly used to enhance gas absorption rate and capacity and to improve selectivity. Monoethanolamine (MEA), diethanolamine (DEA), and methyldiethanolamine (MDEA) are some important amines, which are widely used. However, the conventional primary and secondary alkanolamines exhibit a relatively low maximum capacity for CO<sub>2</sub> capture. Moreover, there are other amines, i.e. piperazine (PZ, a diamine), that have been considered as a promoter for amine systems to improve absorption kinetics, such as MEA/PZ and MDEA/PZ blends, and to increase the concentration of PZ in solution allows for increased solvent capacity and faster kinetics (Freeman *et al.*, 2009).

The purposes of this research were to investigate the performance of hybrid solvents blended between primary, secondary, or tertiary amines and PZ for CO<sub>2</sub> removal from flue gas in terms of CO<sub>2</sub> absorption capacity, and to examine the solvent regeneration temperature to evaluate the solvent performance for reuse in CO<sub>2</sub> removal from flue gas.