

REFERENCES

- [1] Orelup, R.B. Azo dyes having a high solubility in petroleum fuels. U.S. Patent 4000985, 1977.
- [2] Pansare, V.S.; Kulkarni, A.B. Azo dyes from cashew nut shell liquid derivatives. Part I. Azo dyes from 3-pentadecylphenol. Jour. Indian. Chem. Soc. 41(4) (1964): 251–256.
- [3] Suwanprasop, S.; Nhujak, T.; Roengsumran, S.; Petsom, A. Petroleum marker dyes synthesized from cardanol and aniline derivatives. Ind. Eng. Chem. Res. 43 (2004): 4973–4978.
- [4] Hendrickson, J.B.; Cram, D.J., Hammond, G.S. Organic chemistry. 3rd ed. New York: McGraw-Hill, 1972.
- [5] Noller, C.R. Textbook of organic chemistry. 3rd ed. Philadelphia: Saunders, 1966.
- [6] Zollinger, H. Color chemistry. 3rd ed. Switzerland: WILEY-VCH, 2003.
- [7] Wade, L.G. Organic chemistry. 6th ed. United states of America: Pearson education, 2006.
- [8] Silverstein, R.M.; Bassler, G.C.; Morrill, T.C. Spectrometric identification of organic compounds. 4th ed. New York: John Wiley & Sons, 1981.
- [9] Pavia, D.L.; Lampman, G.M.; Kriz, G.S. Introduction to spectroscopy: A guide for students of organic chemistry. Philadelphia: Saunders, 1979.
- [10] Visible and ultraviolet spectroscopy [Online]. Available from: www.cem.msu.edu/~reusch/VirtualText/Spectrpy/UV-Vis/spectrum.htm [2009; January 7].
- [11] UV-vis absorption spectroscopy [Online]. Available from: <http://teaching.shu.ac.uk/hwb/chemistry/tutorials/molspec/uvvisab1.htm> [2009; January 7].
- [12] Fuel dyes [Online]. Available from: http://en.wikipedia.org/wiki/Fuel_dyes [2009; January 9].
- [13] Hunger, K. Industrial dyes. Germany: WILEY-VCH, 2003.
- [14] Thompson, J.W.; Johnson, F.E. Evaluation of alkylated *p*-(*p*-nitrophenylazo)-anilines as gasoline dyes. Ind. Eng. Chem. 43 (1956): 1869–1870.

- [15] Fierz-David, H.E.; Blangey, L. Fundamental processes of dye chemistry. New York: Interscience Pub., 1949.
- [16] Christie, R.M. Colour chemistry. UK: Royal society of chemistry, 2001.
- [17] Schmidt, P.F. Fuel oil manual. 4th ed. New York: Industrial press, 1985.
- [18] Gasoline [Online]. Available from: <http://en.wikipedia.org/wiki/Petrol> [2008; December 10].
- [19] น้ำมันเบนซิน [Online]. Available from:
<http://www.thaigoodview.com/library/studentshow/2549/m6-6/no11-14-16-49/gasoline1.html> [2008; December 10].
- [20] Chemical and physical information, Gasoline [Online]. Available from:
<http://www.atsdr.cdc.gov/toxprofiles/tp72-c3.pdf> [2009; January 8].
- [21] ประกาศกรมธุรกิจพลังงาน เรื่อง กำหนดลักษณะและคุณภาพของน้ำมันเบนซินพื้นฐาน. ราชกิจจานุเบกษา เล่ม 125 ตอนพิเศษ 85 ง. (21 พฤษภาคม 2551): 21–22.
- [22] Bhunia, H.P.; Nando, G.B.; Chaki, T.K.; Basak, A.; Lenka, S.; Nayak, P.L. Synthesis and characterization of polymers from cashew nut shell liquid (CNSL), a renewable resource II. Synthesis of polyurethanes. Eur. Polym. J. 35 (1999): 1381–1382.
- [23] Orazio, A.A.; Berretta, S.; Fiani, C.; Filippone, P.; Mele, G.; Saladino, R. Synthesis and reactions of nitro derivatives of hydrogenated cardanol. Tetrahedron. 62 (2006): 6113.
- [24] Menon, A. R. R.; Pillai, C. K. S.; Sudha, J. D.; Mathew, A.G. Cashew nut shell liquids – its polymeric and other industrial products. J. Sci. Ind. Res. 44 (1985): 324–338.
- [25] Wagner, R. W.; Ruffing, J.; Breakwell, B. V.; Lindsey, J. S. Separation of cashew (*Anarcadium occidentale* L.) nut shell liquid with supercritical carbon dioxide. Bioresource technol. 88 (2003): 1–7.
- [26] Prabhakaran, K.; Narayanan, A.; Pavithran, C. Cardanol as a dispersant plasticizer for an alumina/toluene tape casting slip. J. Europ. Ceram. Soc. 21 (2001): 2873–2878.
- [27] Tyman, J. H. P.; Johnson R. A.; Muir M.; Rekhgar R. The extraction of natural cashew nut shell liquid from the cashew nut (*Anarcadium occidentale*). J. Am. Oil Chem. Soc. 66(4) (1989): 553–557.

- [28] Yadav, R.; Devi, A.; Tripathi, G.; Srivastava, D. Optimization of the process variables for the synthesis of cardanol-based novolac-type phenolic resin using response surface methodology. *Eur. Polym. J.* 43 (2007): 3531–3537.
- [29] Reddy, D.; Chandrashekhar, T.K. Cardanol based matrix biocomposites reinforced with natural fibres. *Compos. Sci. Technol.* 64 (2004): 839–845.
- [30] Leerawan Khaokhum, L.; Sawasdipuksa, N.; Kumthong, N.; Tummatorn, J.; Roengsumran, S. Cardanol polysulfide as a vulcanizing agent for natural rubber. *J. Sci. Res. Chula. Univ.* 30 (2005): 23–30.
- [31] Castro Dantas, T.N.; Dantas, M.S.G.; Dentas Neto, A.A.; D'Ornellas, C.V.; Queiroz, L.R. Novel antioxidants from cashew nut shell liquid applied to gasoline stabilization. *Fuel.* 82 (2003): 1465–1469.
- [32] Rodrigues, F. H. A.; Feitosa, J. P. A.; Nágila M. P. S.; Ricardo, N. M. P. S. de França, F. C. F.; Carioca, J. O. B. Antioxidant activity of cashew nut shell liquid (CNSL) derivatives on the thermal oxidation of synthetic *cis*-1,4-polyisoprene. *J. Braz. Chem. Soc.* 17 (2006): 1–7.
- [33] Chemical composition of CNSL, Sanoor Cashew (Importer & Exporters), India [Online]. Available from: www.adarshsanoor.com [2009; January 10].
- [34] Aggarwal, L. K.; Thapliyal, P. C.; Karade, S. R. Anticorrosive properties of the epoxy–cardanol resin based paints. *Prog. Org. Coat.* 59 (2007): 76–80.
- [35] Kesler, M.L.; Risberg, J.A.; Boro, M.; Dziomba, A. Granular form red oil soluble dye. *U.S. Patent 3056642*, 1962.
- [36] De Feo, F.; Papa, S.; Traverso, E. Organic solvent-soluble azo dyes. *U.S. Patent 4011209*, 1977.
- [37] Zeidler, G.; Dehnert, J.; Hansen, G.; Riedel, G. Oil-soluble azo dye with N-substituted β-naphtholamine as coupling component. *U.S. Patent 4315756*, 1982.
- [38] Hansen, G.; Merger, F.; Nestler, G.; Zeidler, G. Dye mixtures of azo compounds soluble in organic solvents. *U.S. Patent 4473376*, 1984.
- [39] Armbrust, H.; Dix, J.P.; Hansen, G.; Zeidler, G. Disazo dyes. *U.S. Patent 4521216*, 1985.
- [40] Shoair, A.F.; El-Binary, A.A.; El-Sonbati, A.Z.; Younes, R.M. Stereochemistry of new nitrogen containing heterocyclic aldehyde. VI. Novel

- structural and properties models of uranyl with quinoline azo dyes. Spectrochim. Acta Part A 57 (2001): 1683–1691.
- [41] Whitlock, L.R.; Siggia, S.; Smola, J.E. Spectrophotometric analysis of phenols and of sulfonates by formation of an azo dye. Anal. Chem. 44(3) (1972): 532–536.
- [42] Gulati, A.S.; Rao, B.C.S. Relation between structure & chromatographic behaviour of azo dyes derived from *m*-Alkylphenols. Indian J. Chem. 5 (1967): 55–56.

APPENDICES

APPENDIX A

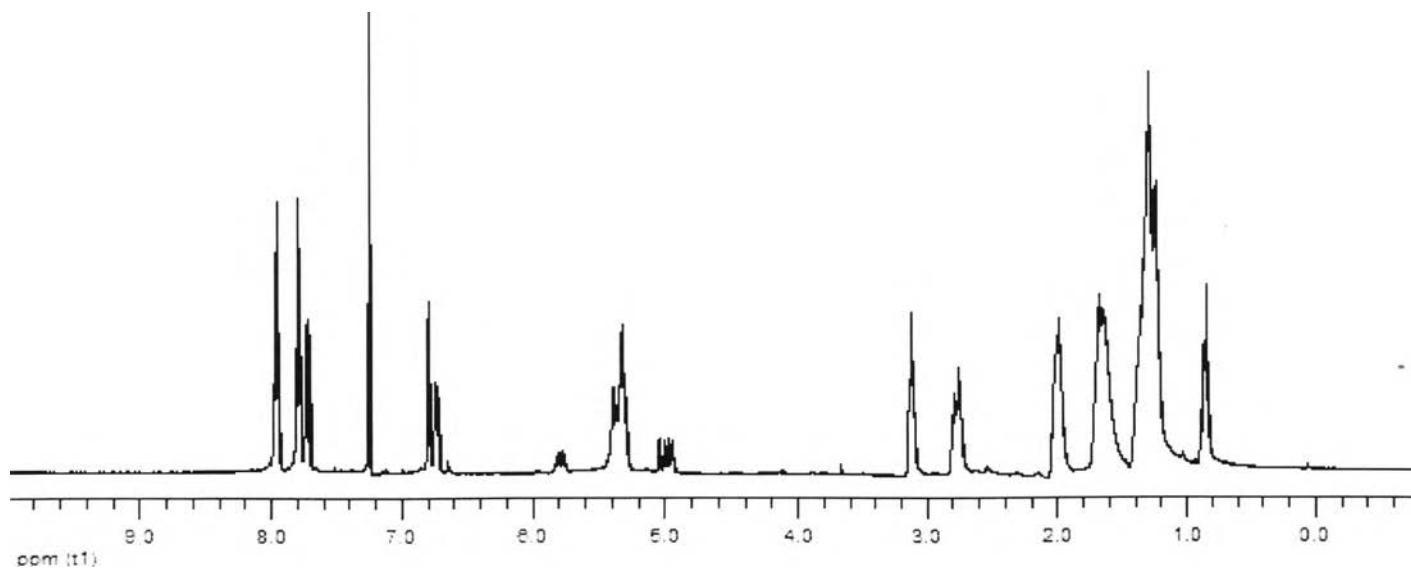


Figure A-1 ^1H -NMR spectrum of Compound **6**.

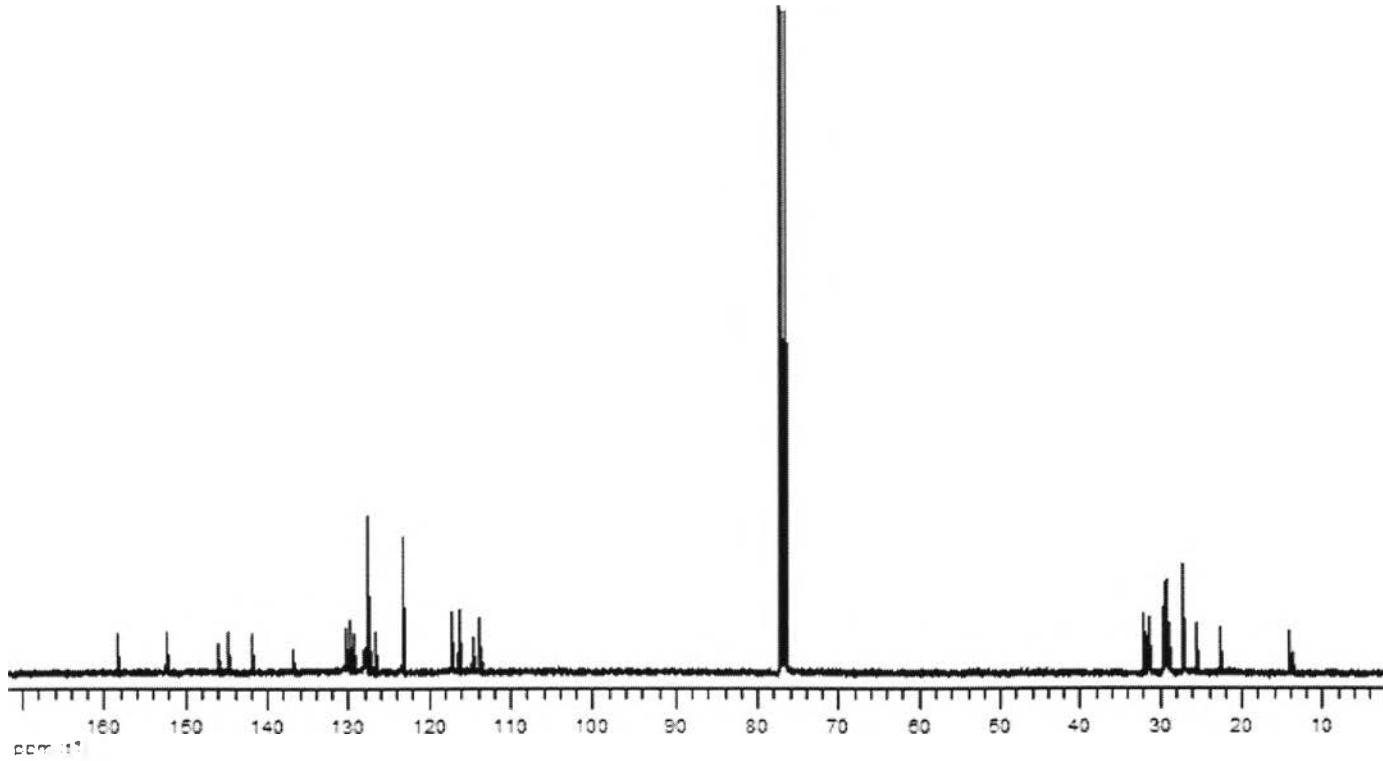


Figure A-2 ^{13}C -NMR spectrum of Compound 6.

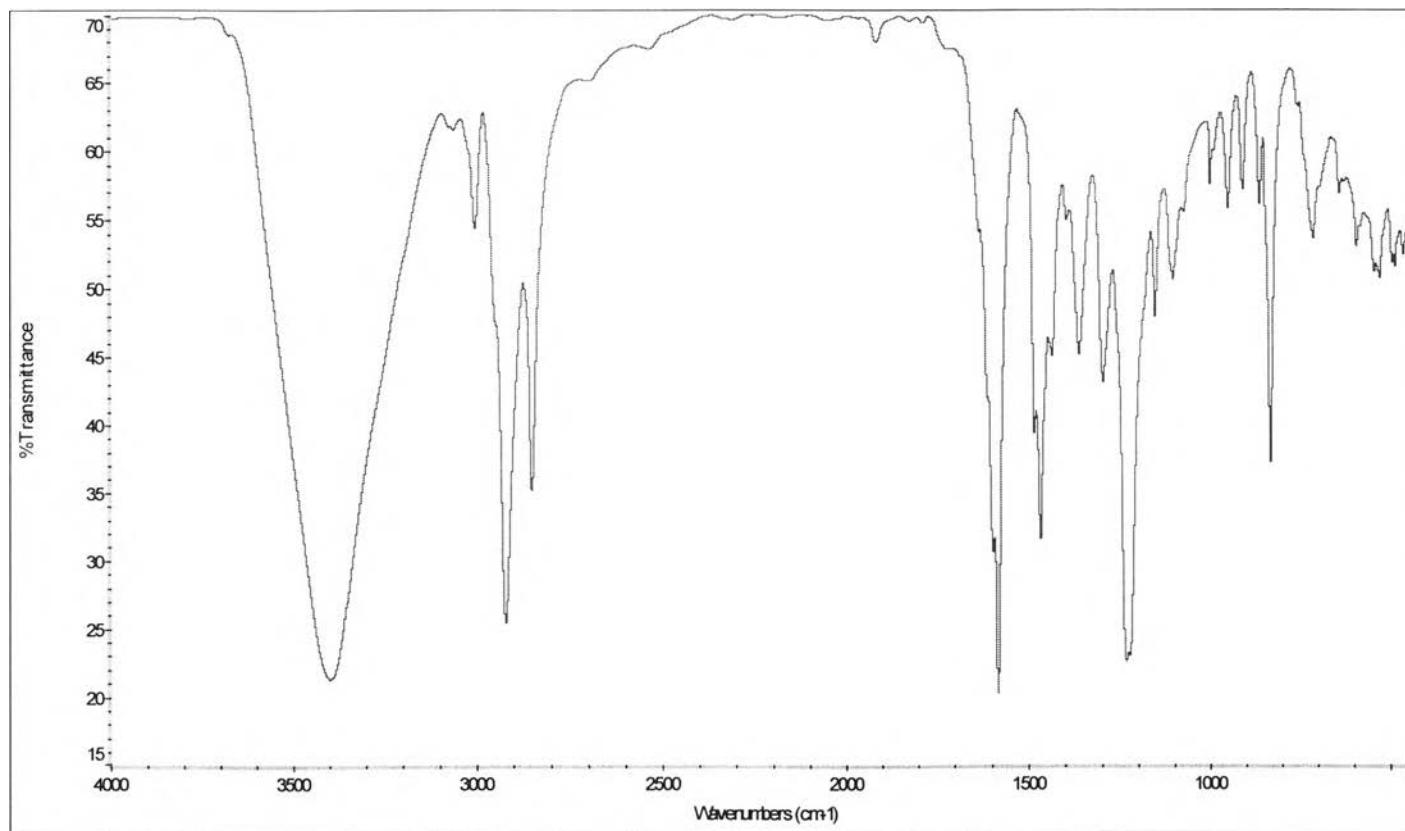


Figure A-3 IR spectrum of Compound 6.

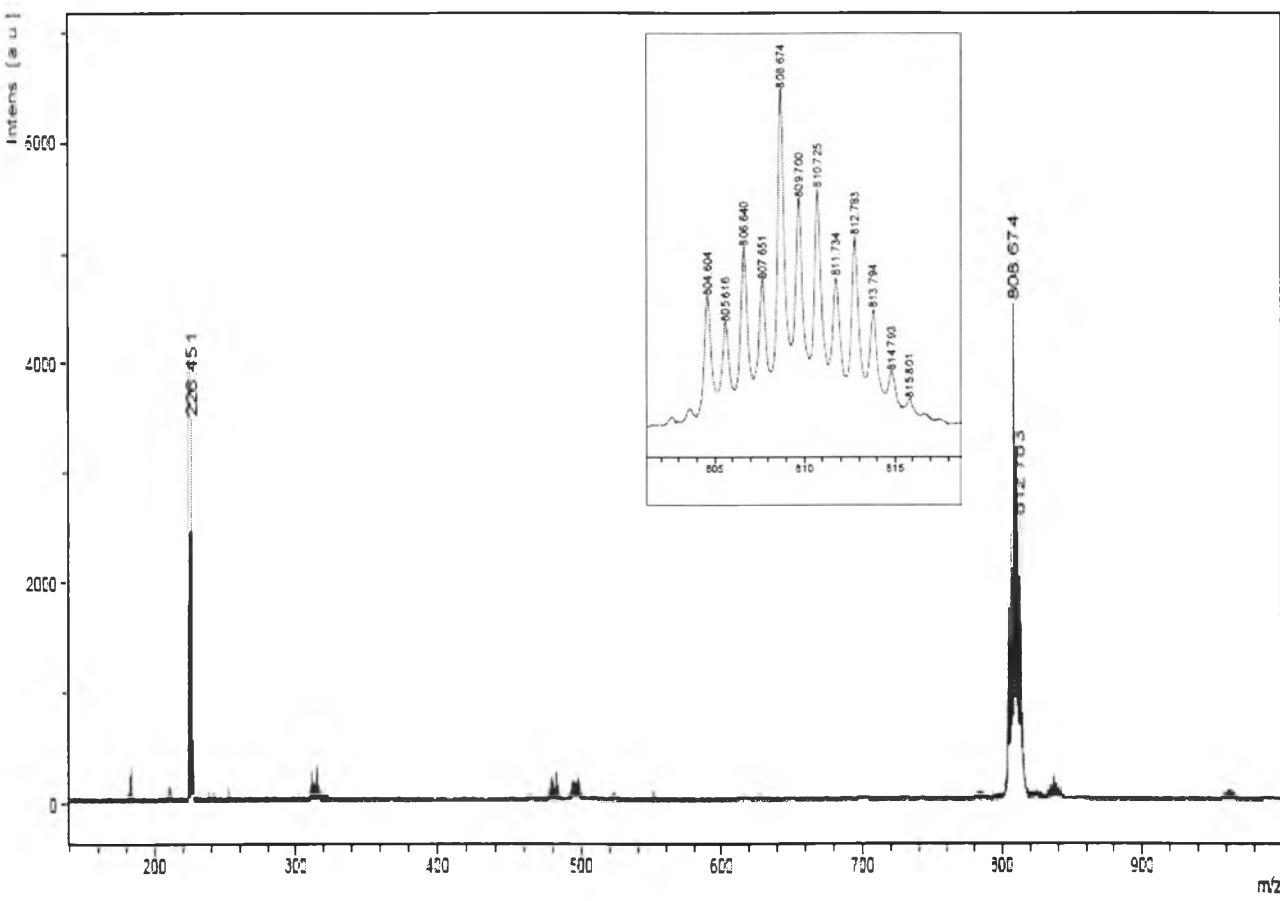


Figure A-4 Mass spectrum of Compound 6.

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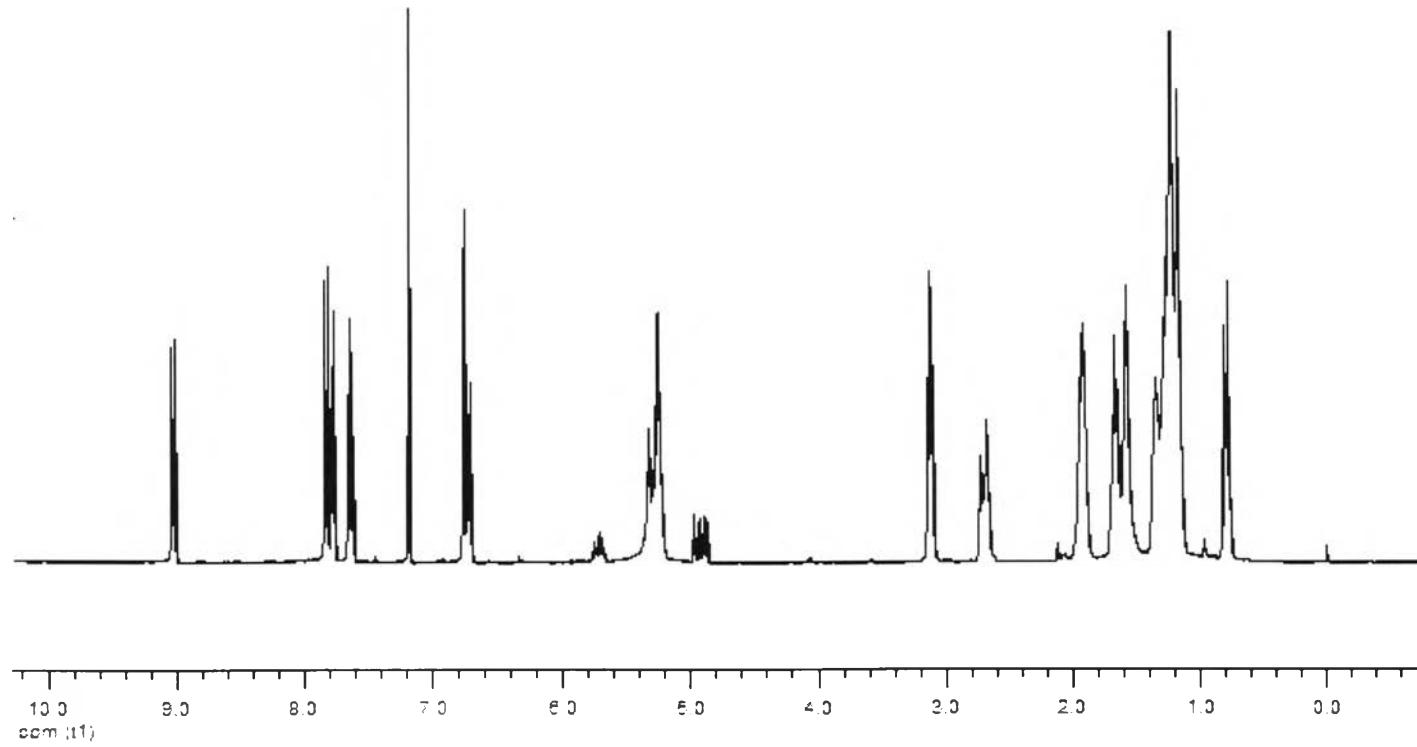


Figure A-5 ^1H -NMR spectrum of Compound 7.

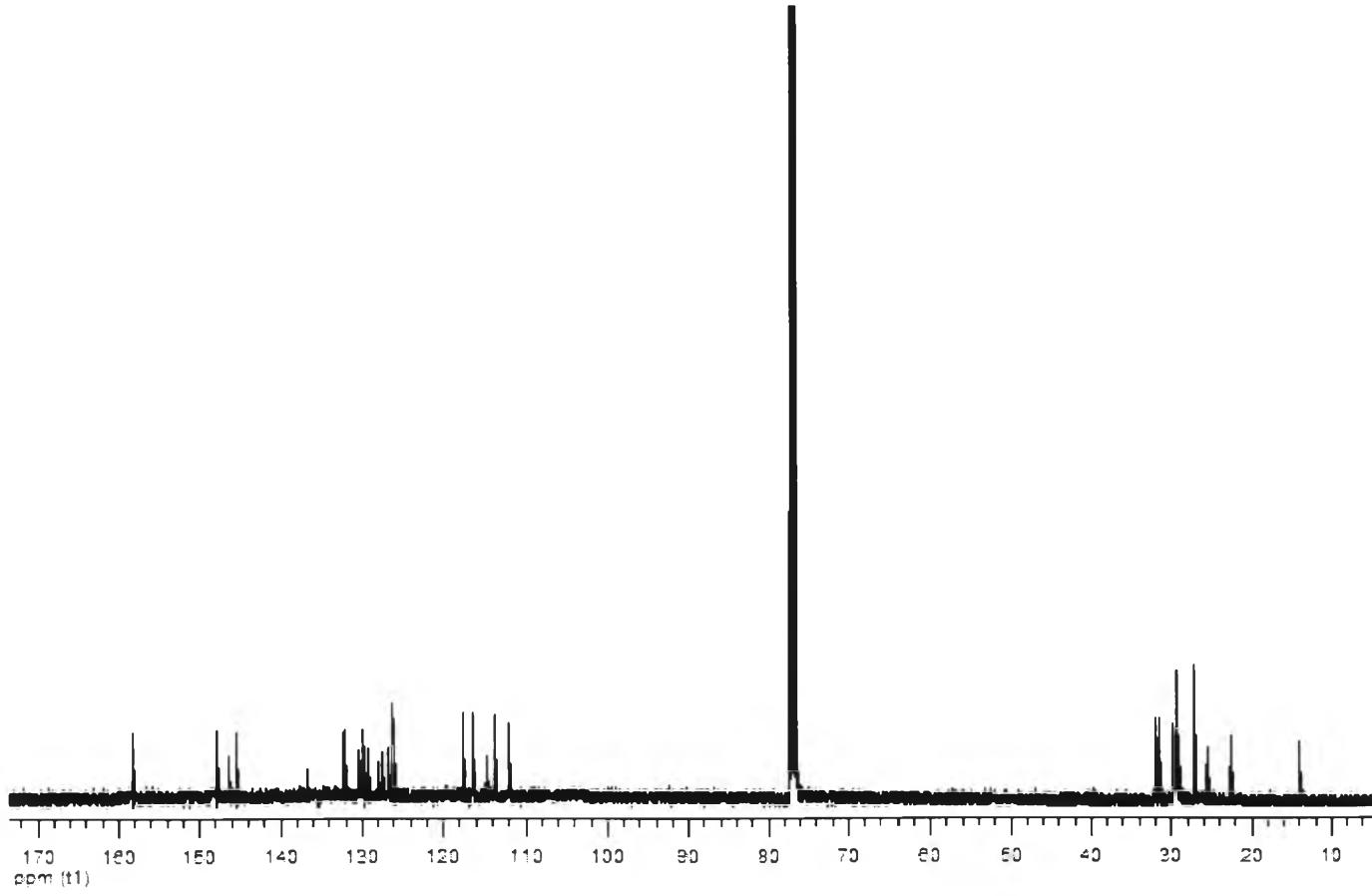


Figure A-6 ^{13}C -NMR spectrum of Compound 7.

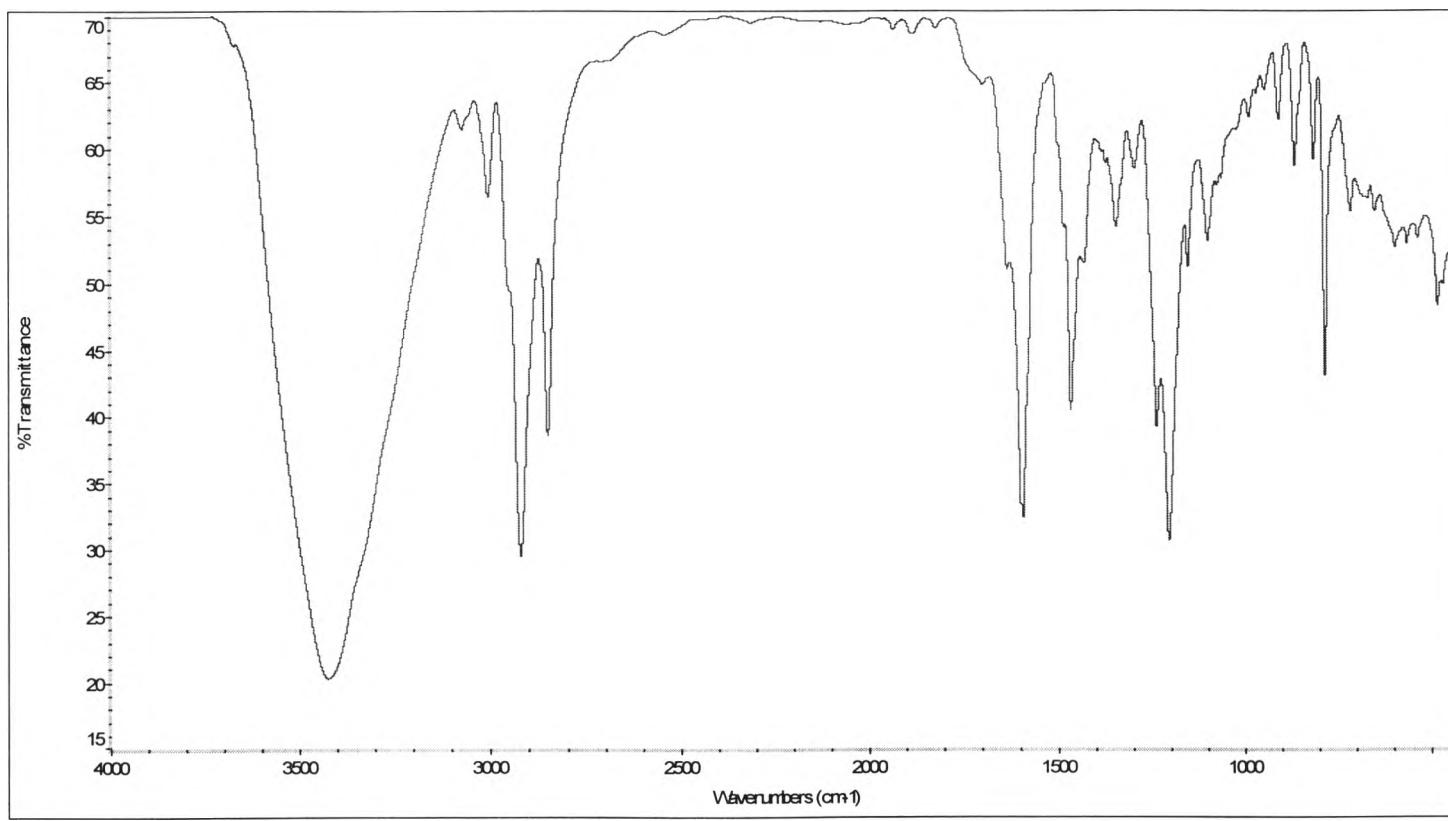


Figure A-7 IR spectrum of Compound 7.

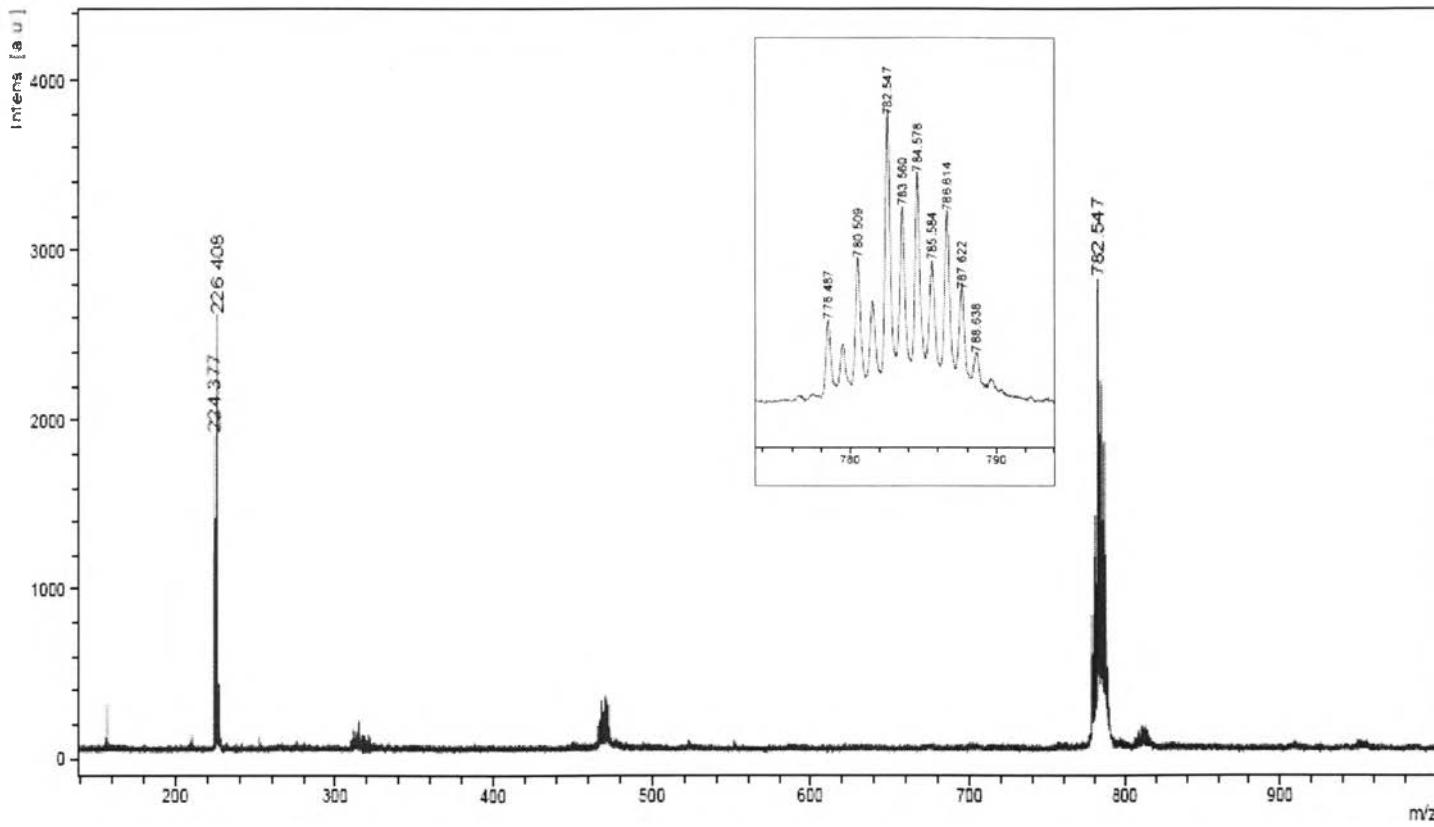


Figure A-8 Mass spectrum of Compound 7.

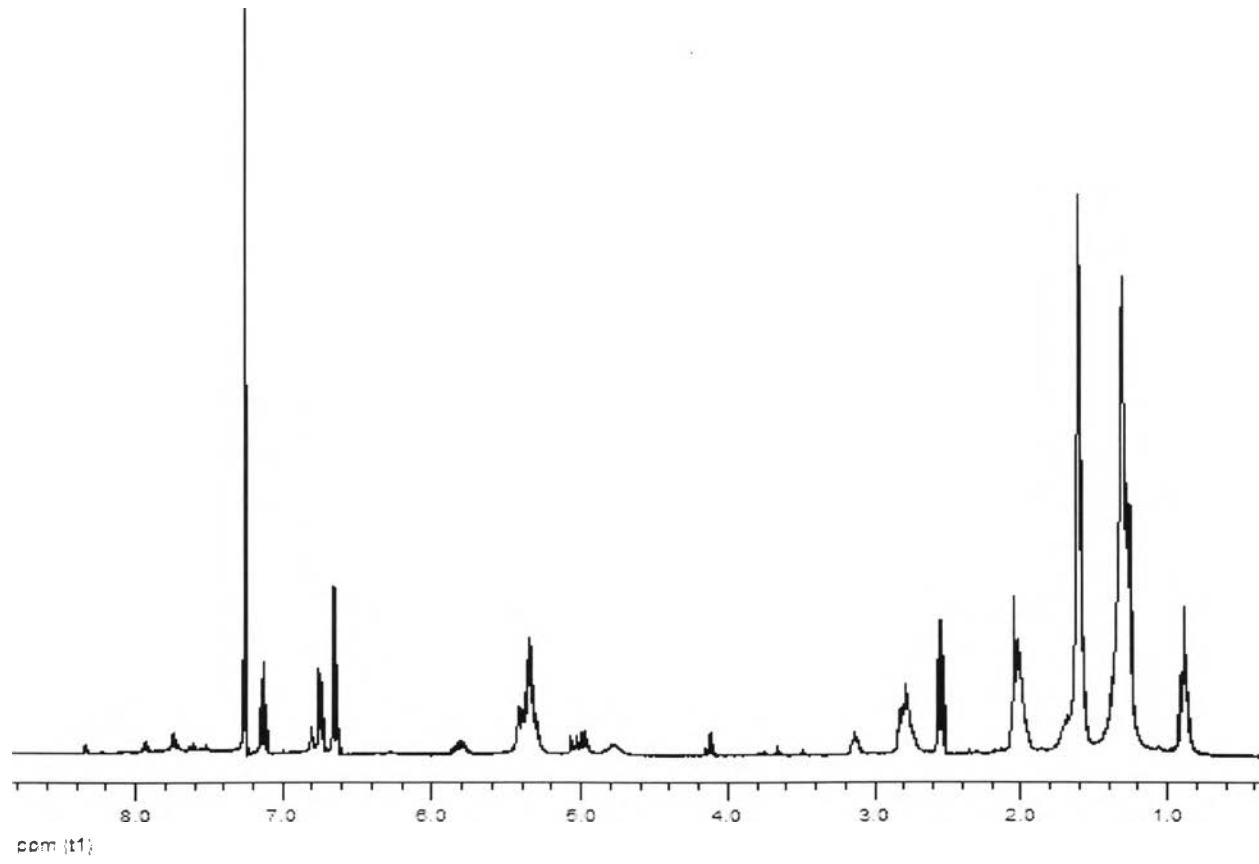


Figure A-9 ^1H -NMR spectrum of crude of Compound 9.

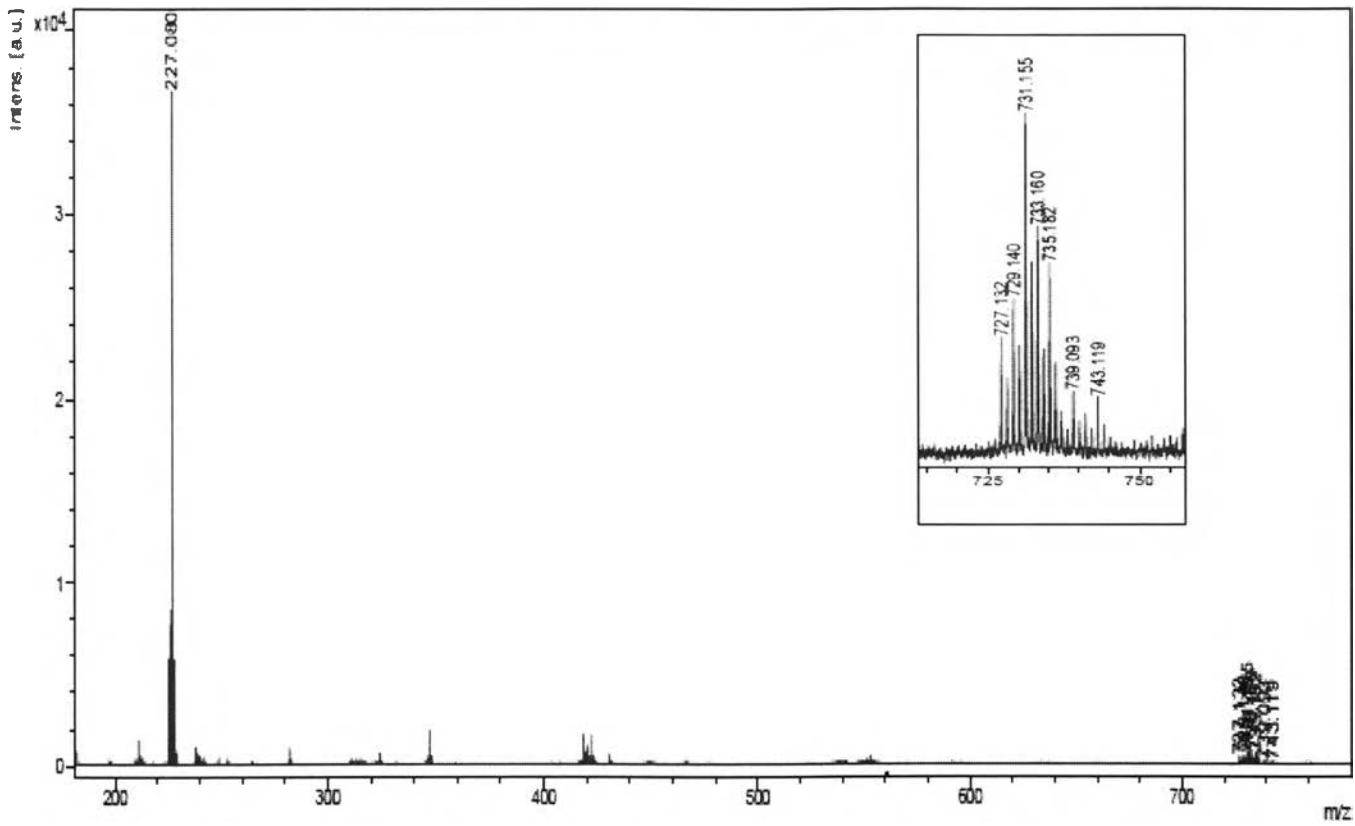


Figure A-10 Mass spectrum of crude of Compound 9.

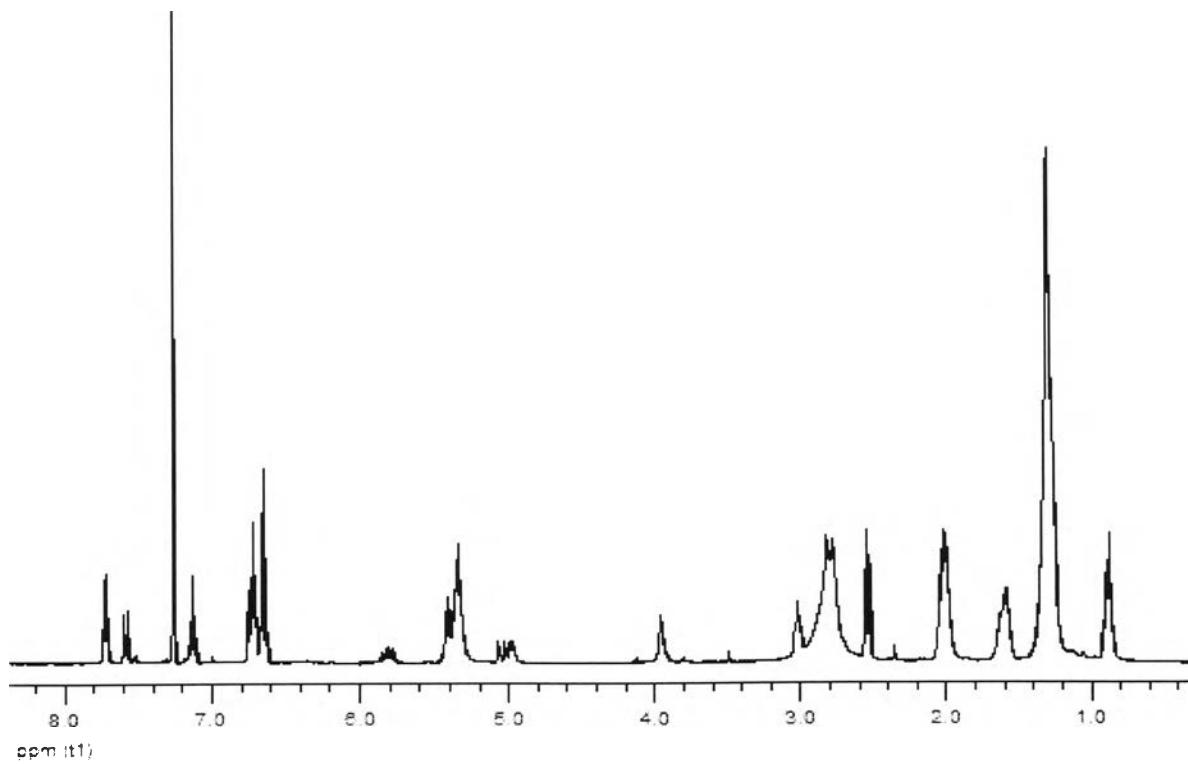


Figure A-11 ^1H -NMR spectrum of crude of Compound **10**.

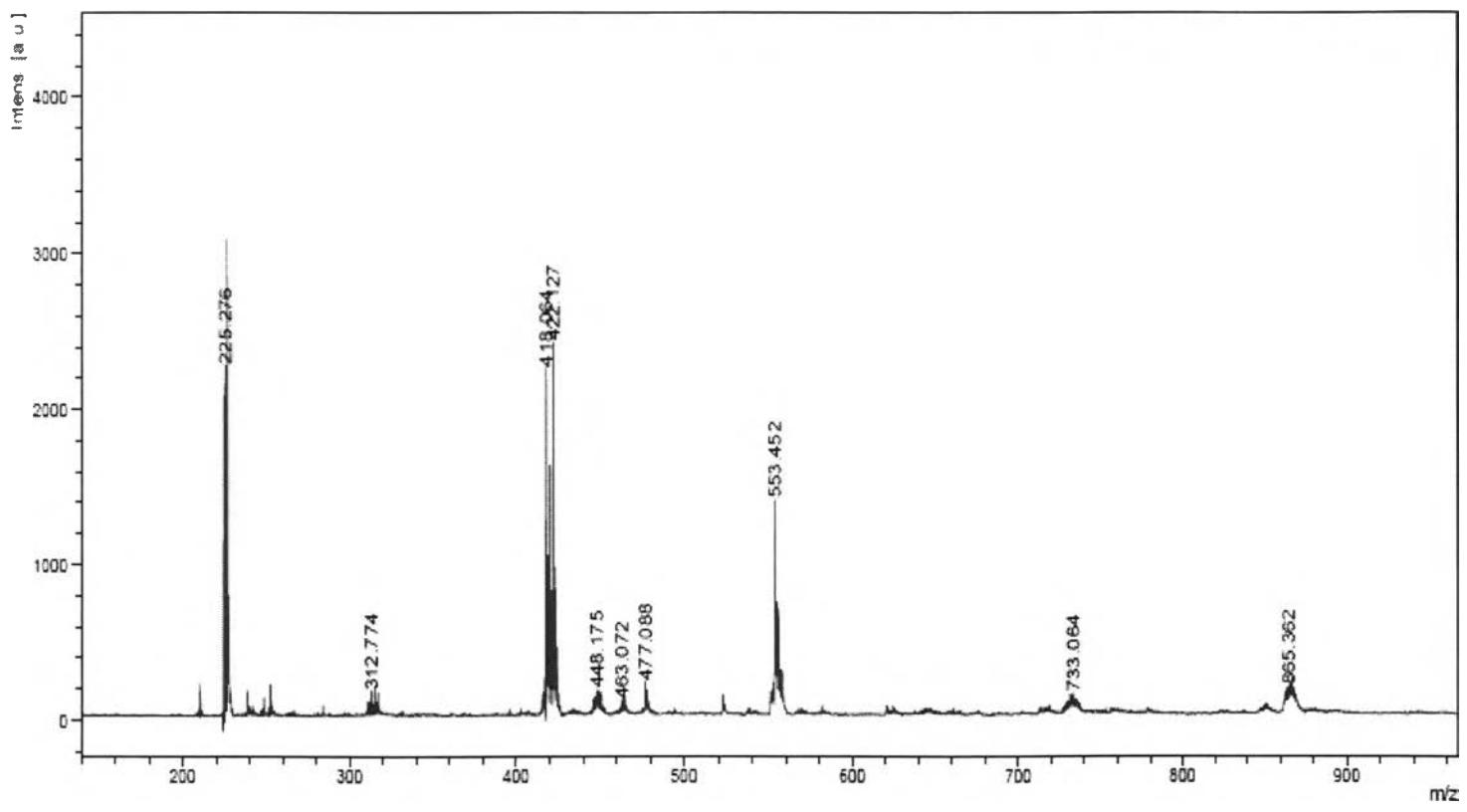


Figure A-12 Mass spectrum of crude of Compound 10.

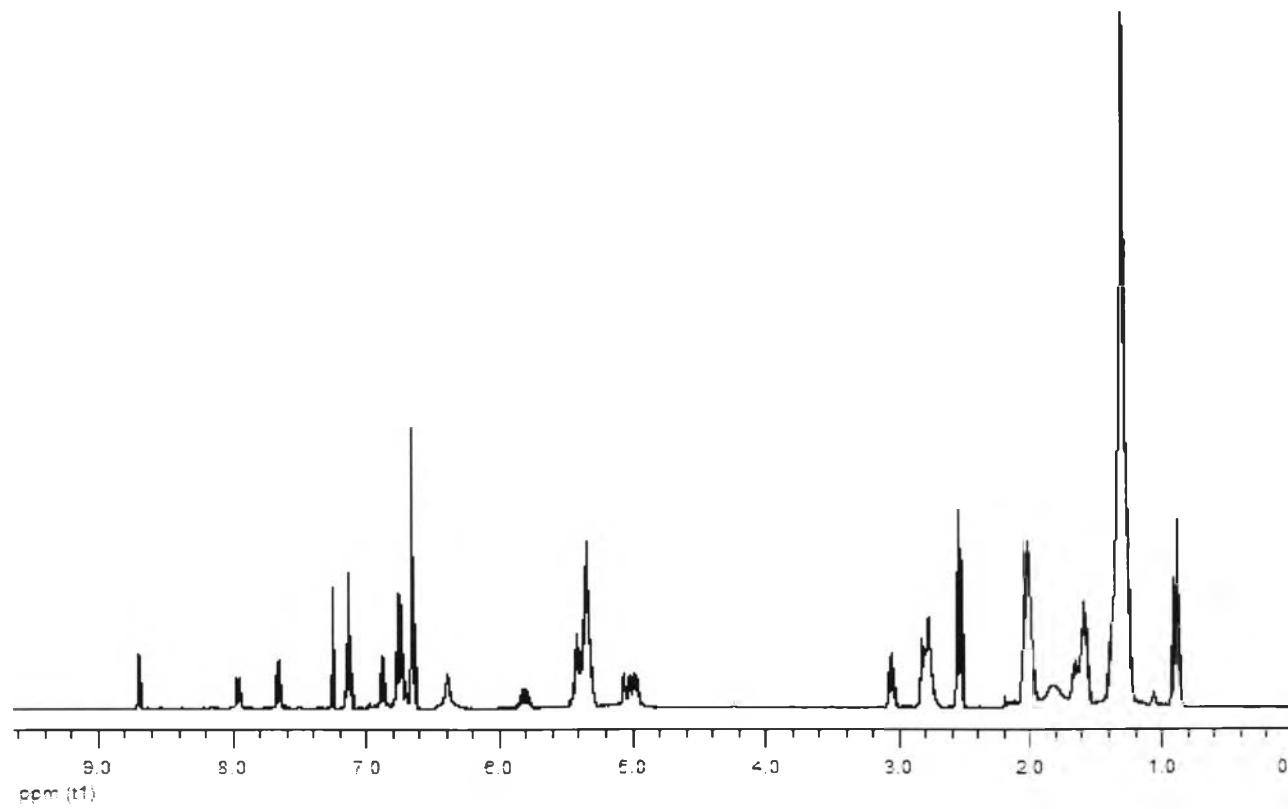


Figure A-13 ^1H -NMR spectrum of crude of Compound 11.

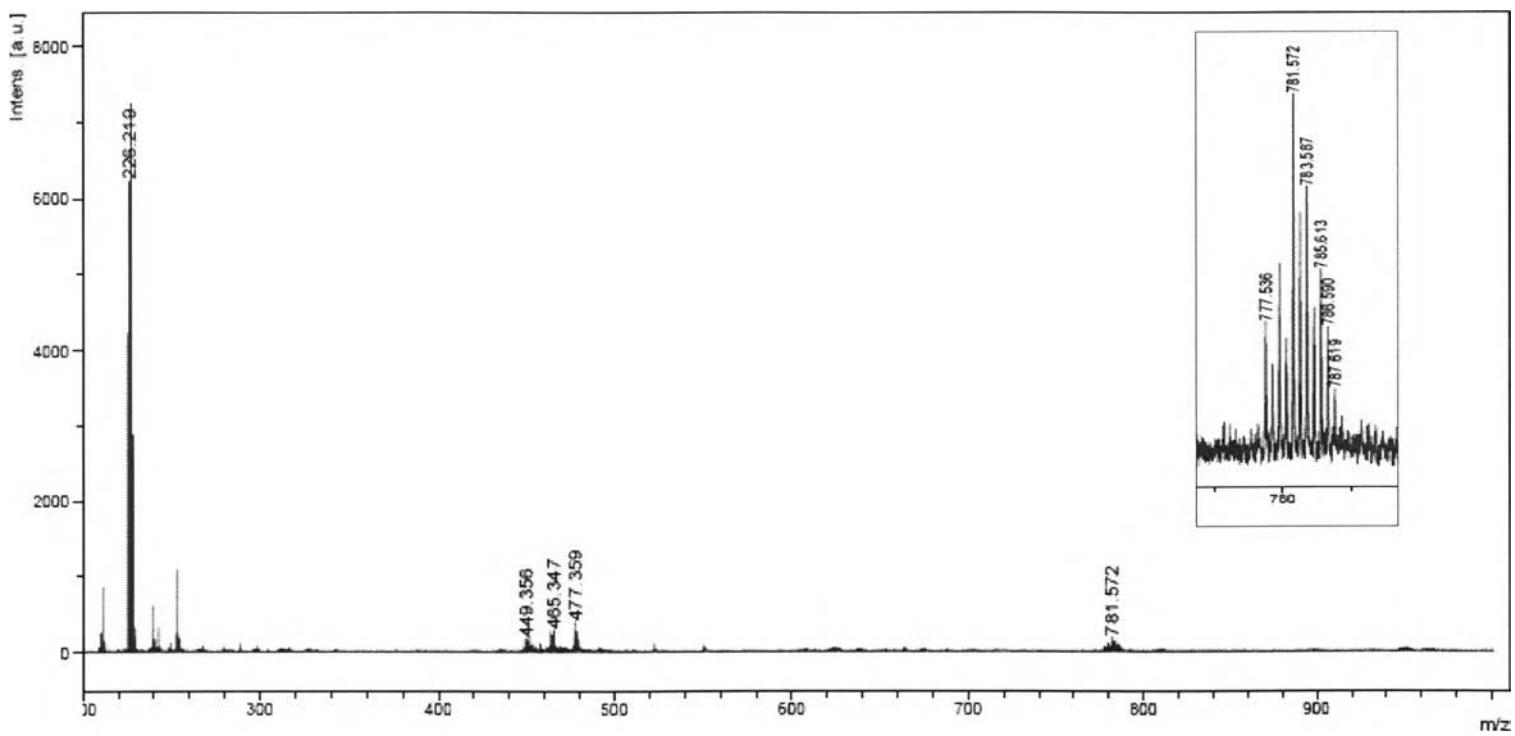


Figure A-14 Mass spectrum of crude of Compound 11.

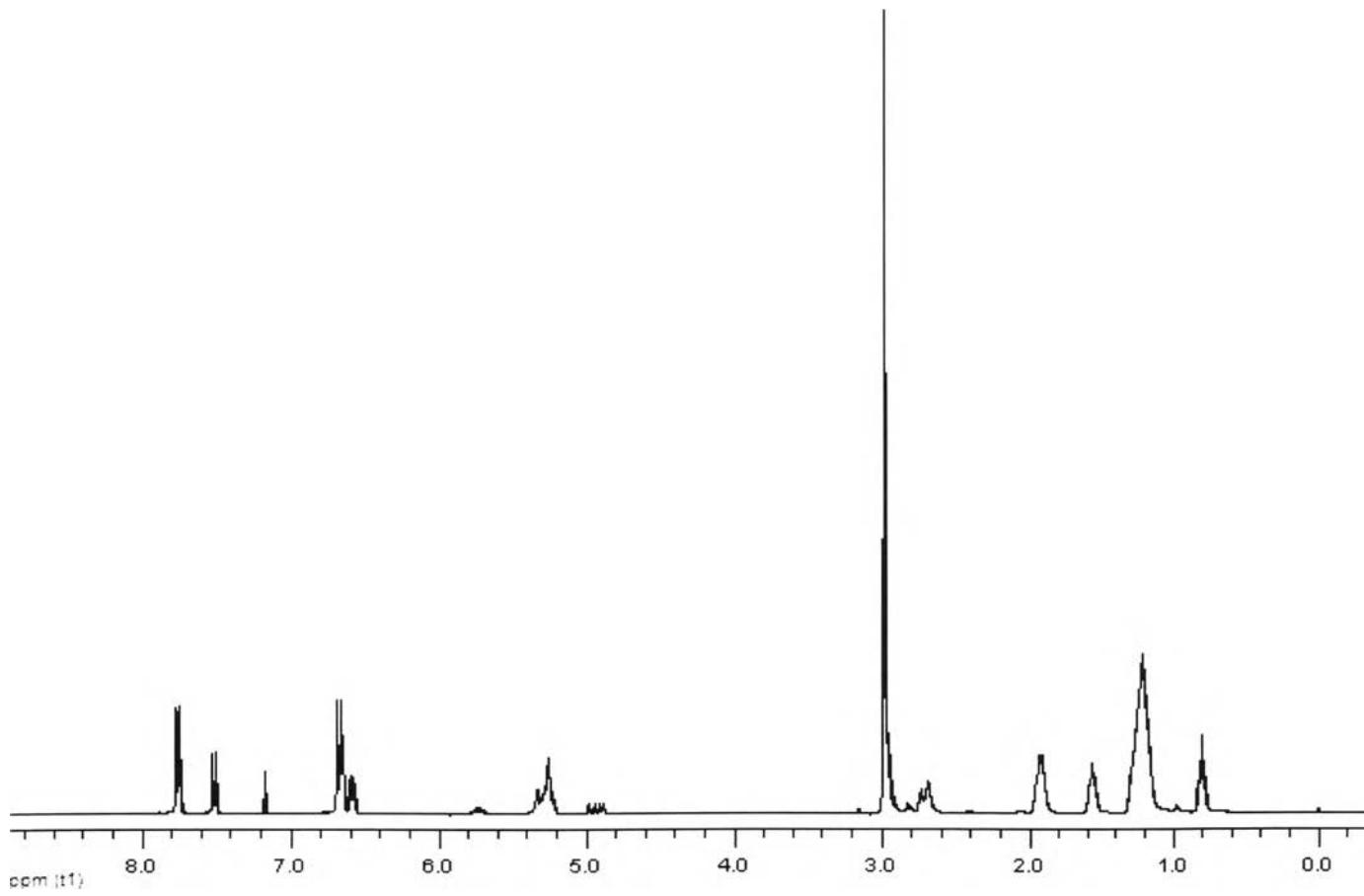


Figure A-15 ^1H -NMR spectrum of Compound 12.

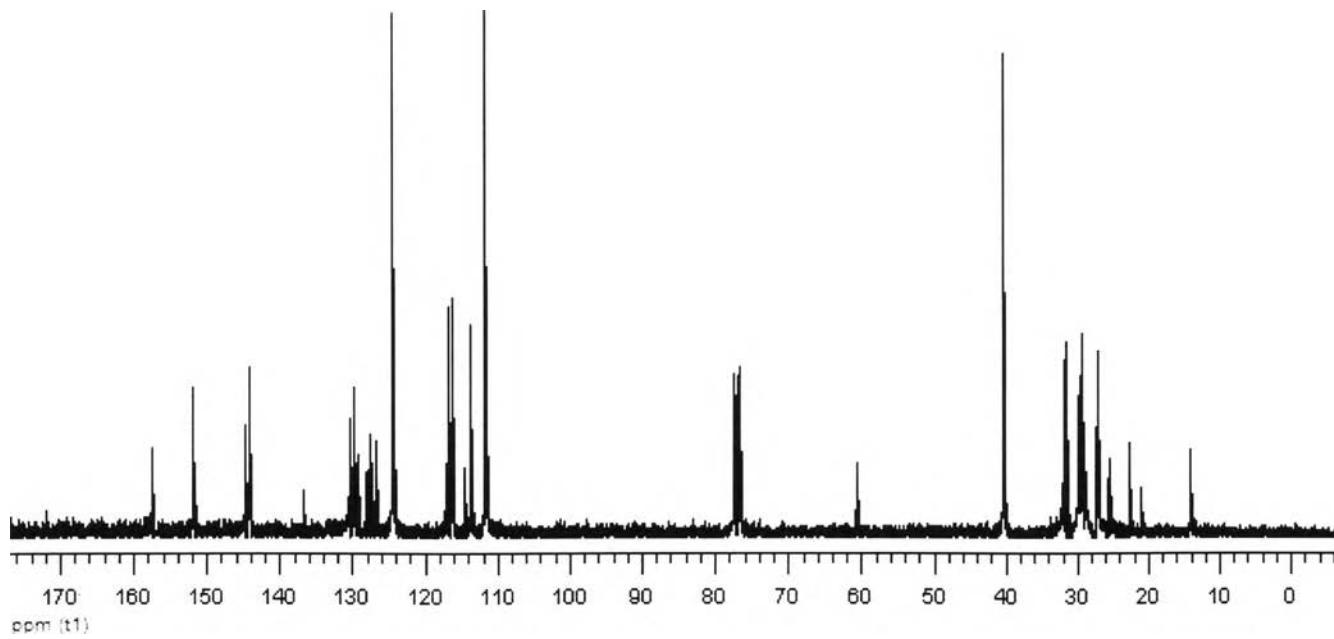


Figure A-16 ^{13}C -NMR spectrum of Compound 12.

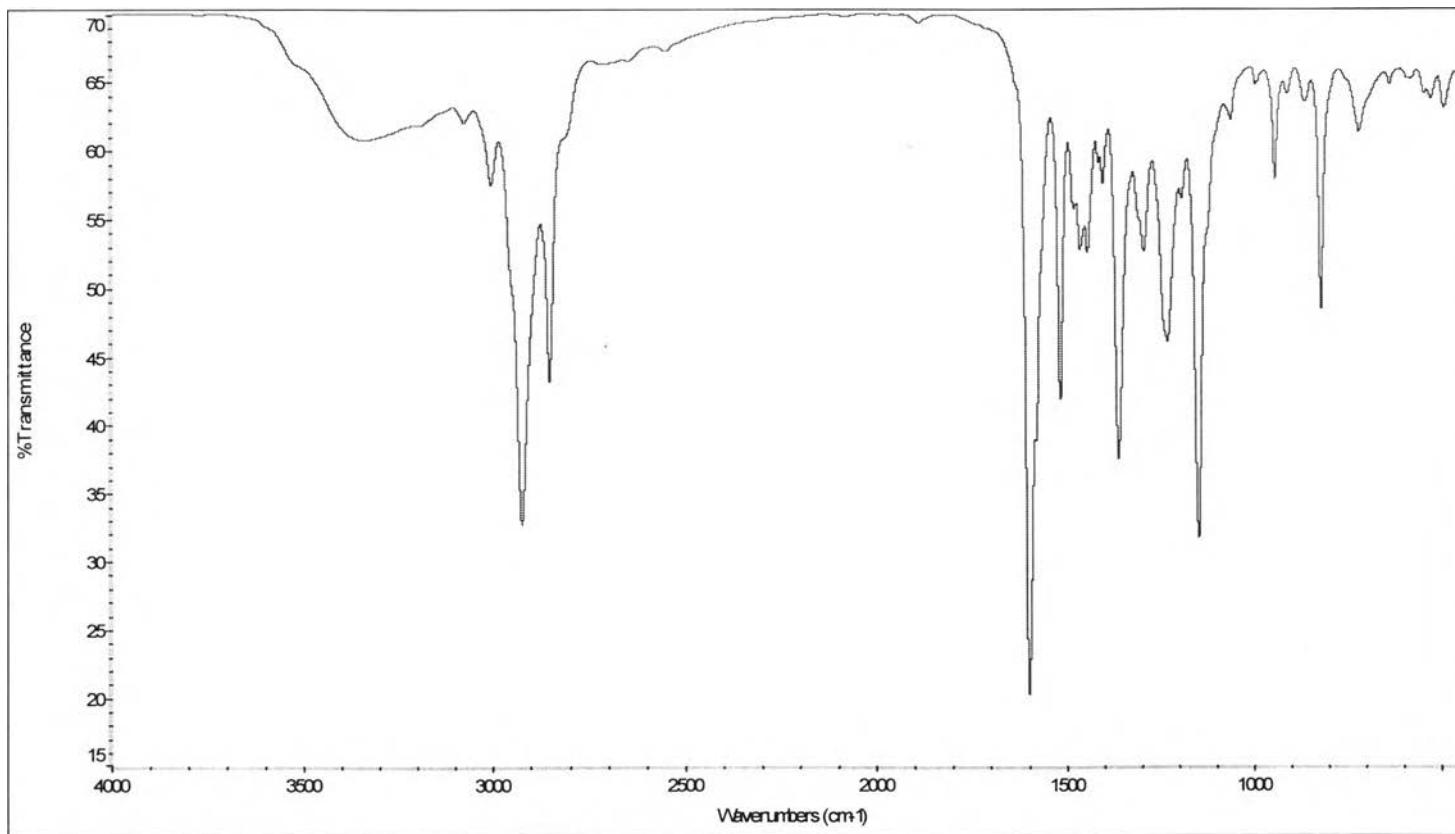


Figure A-17 IR spectrum of Compound 12.

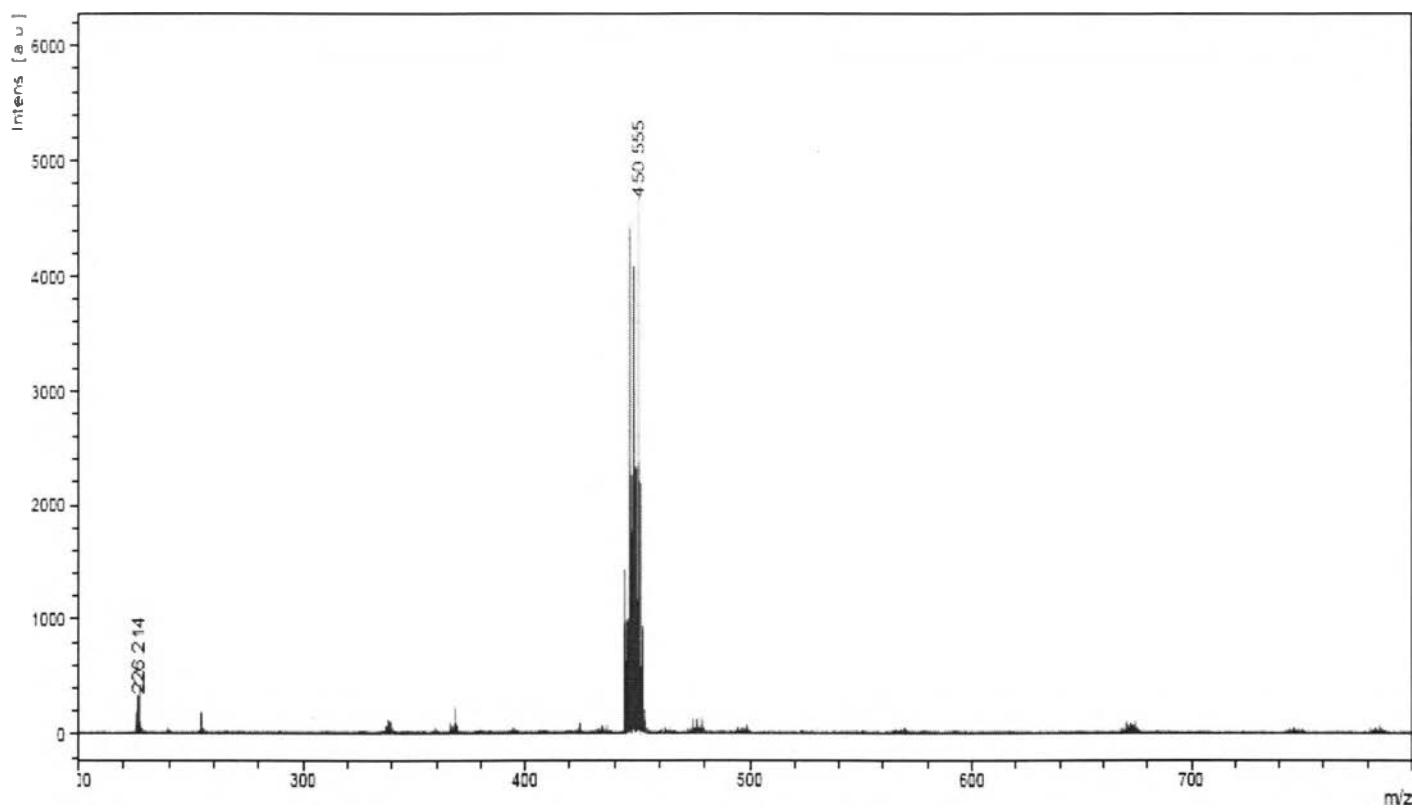


Figure A-18 Mass spectrum of Compound 12.

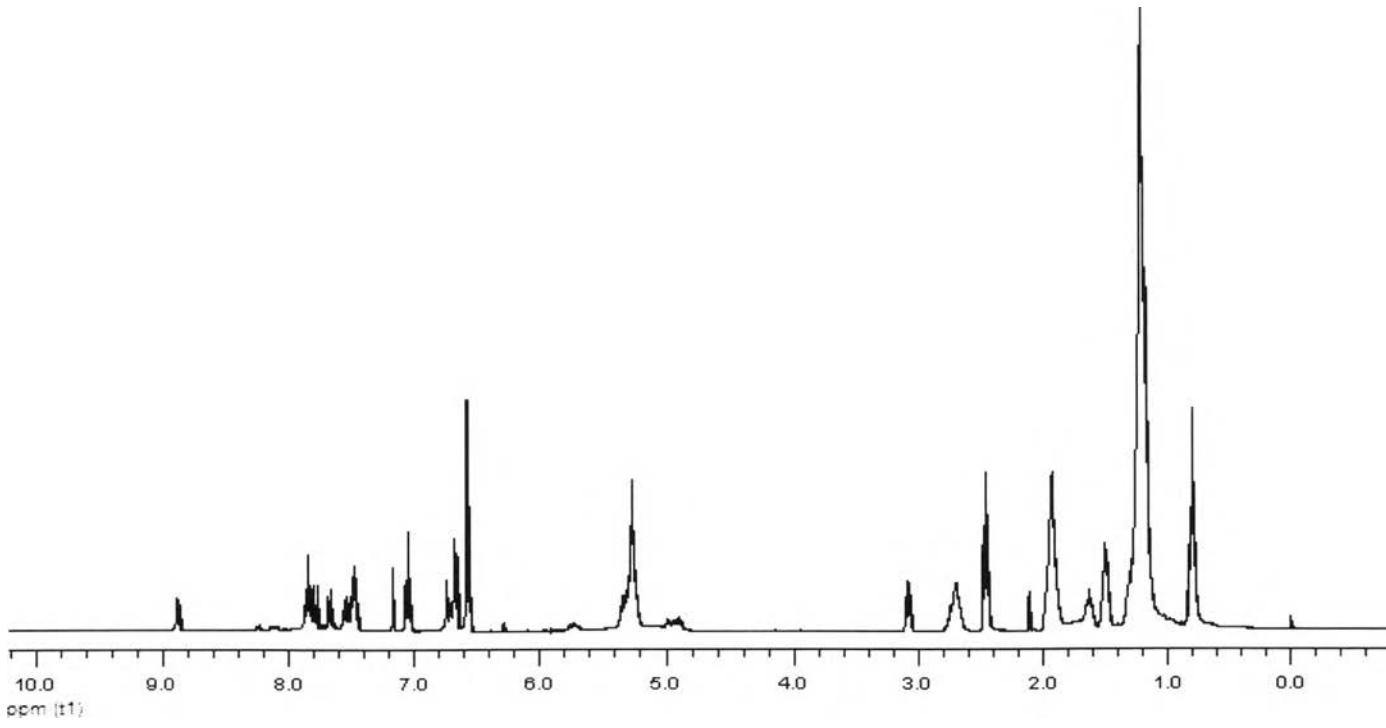


Figure A-19 ¹H-NMR spectrum of Compound 13.

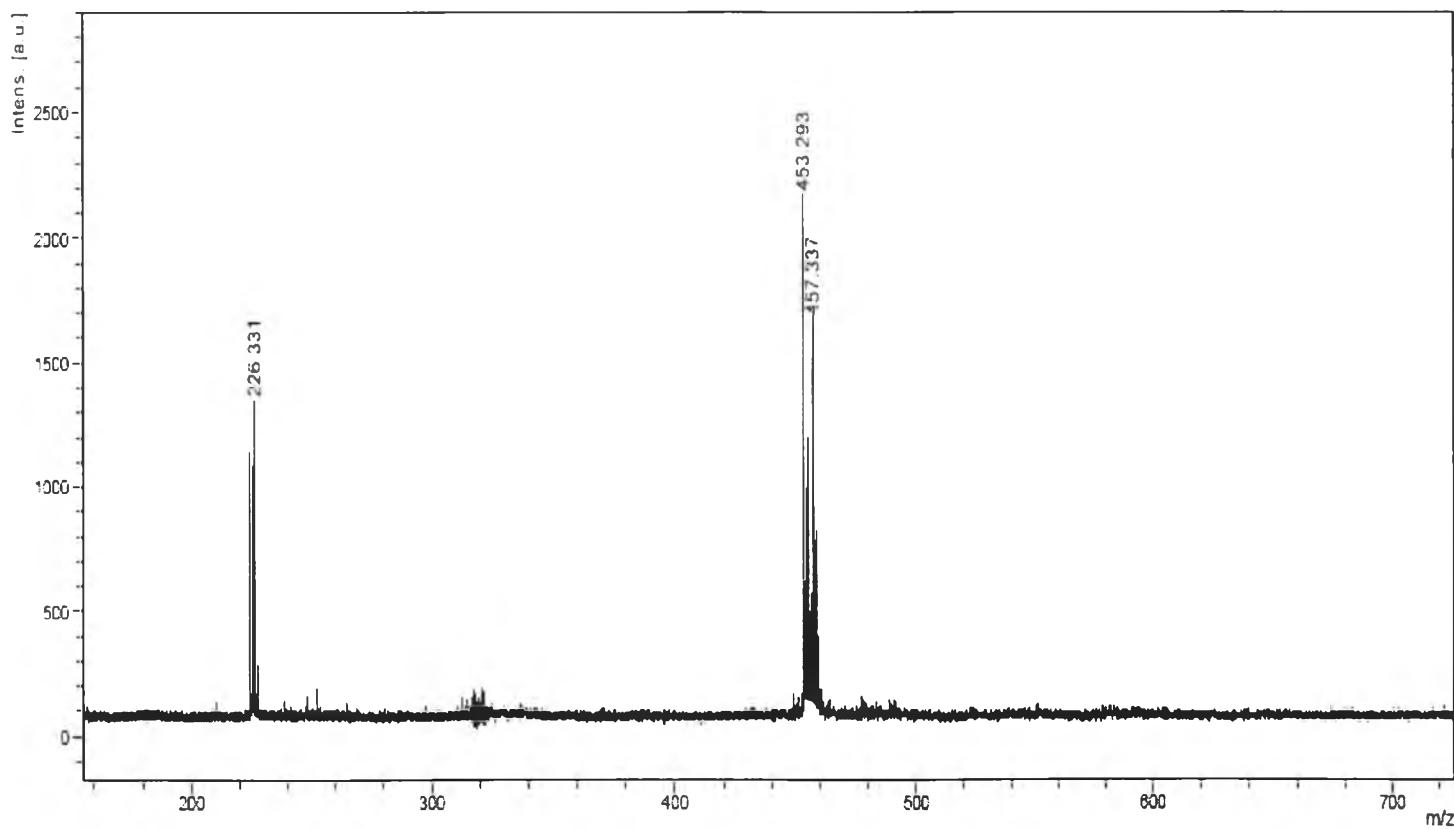


Figure A-20 Mass spectrum of Compound 13.

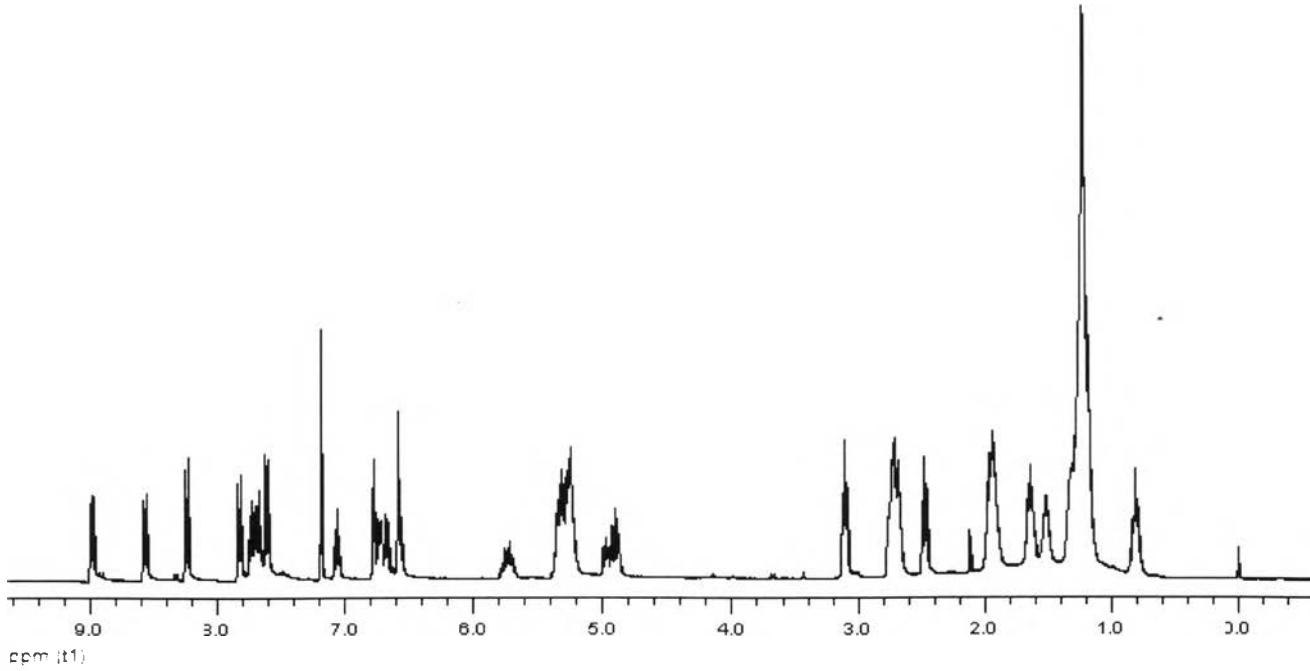


Figure A-21 ¹H-NMR spectrum of Compound 14.

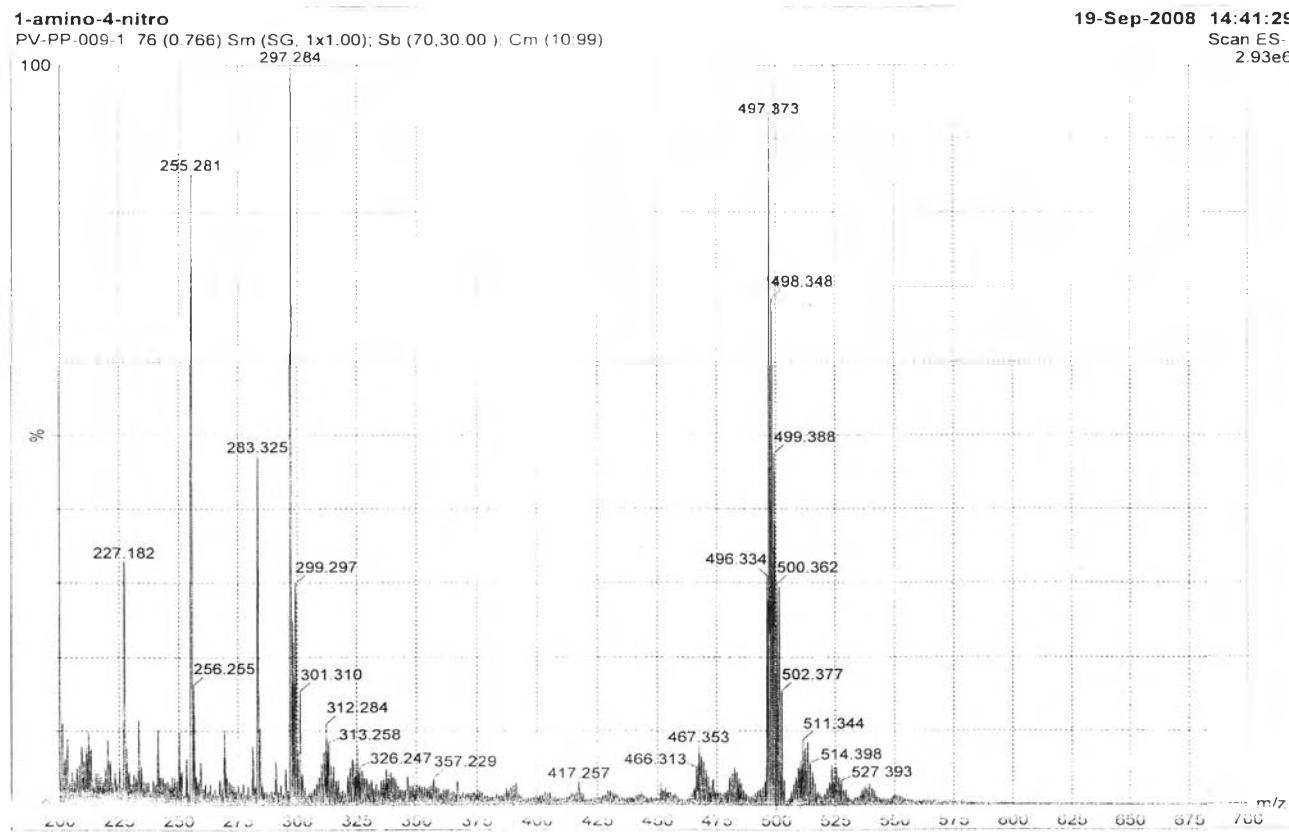


Figure A-22 Mass spectrum of Compound 14.

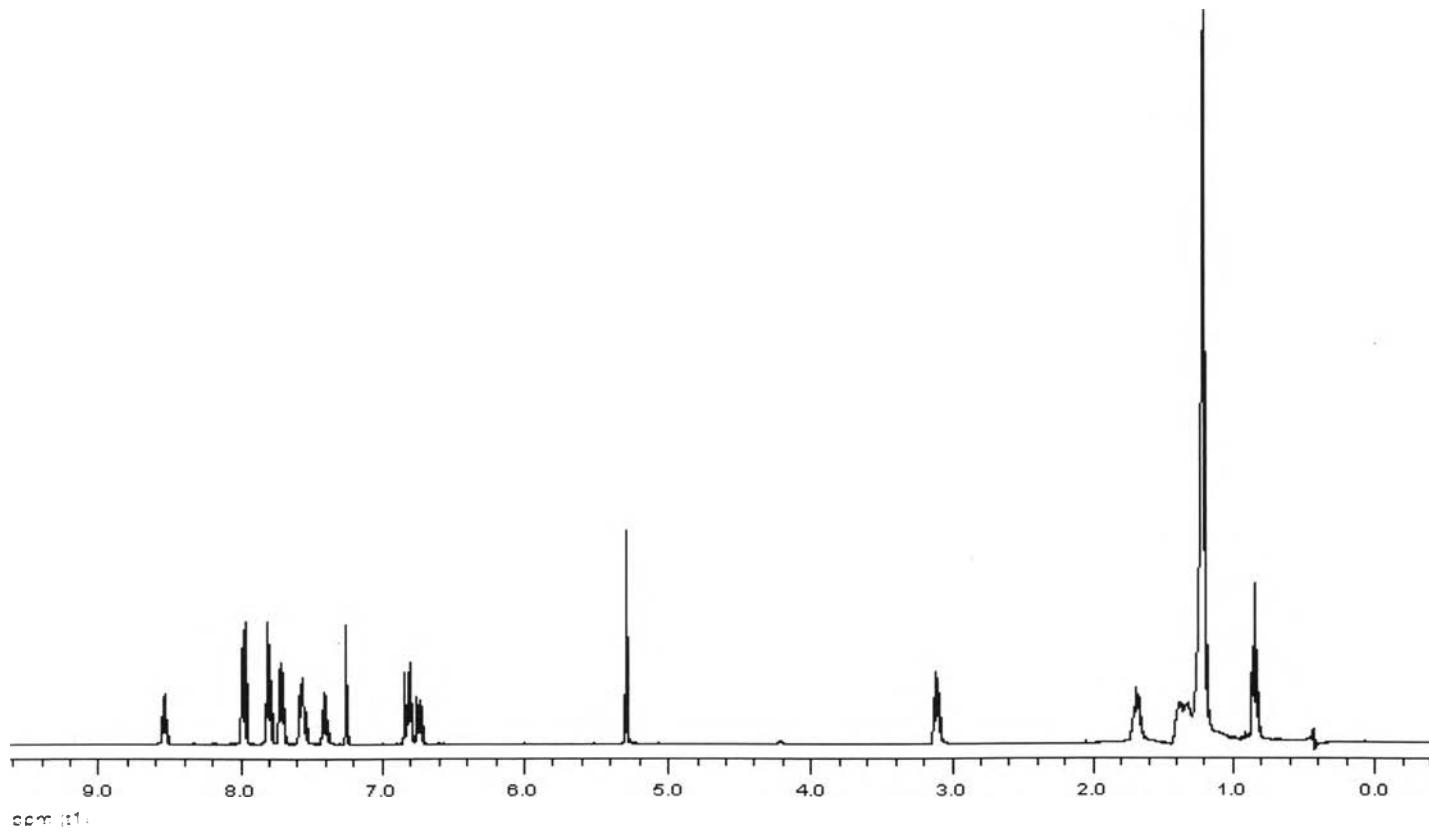


Figure A-23 ^1H -NMR spectrum of Compound 17.

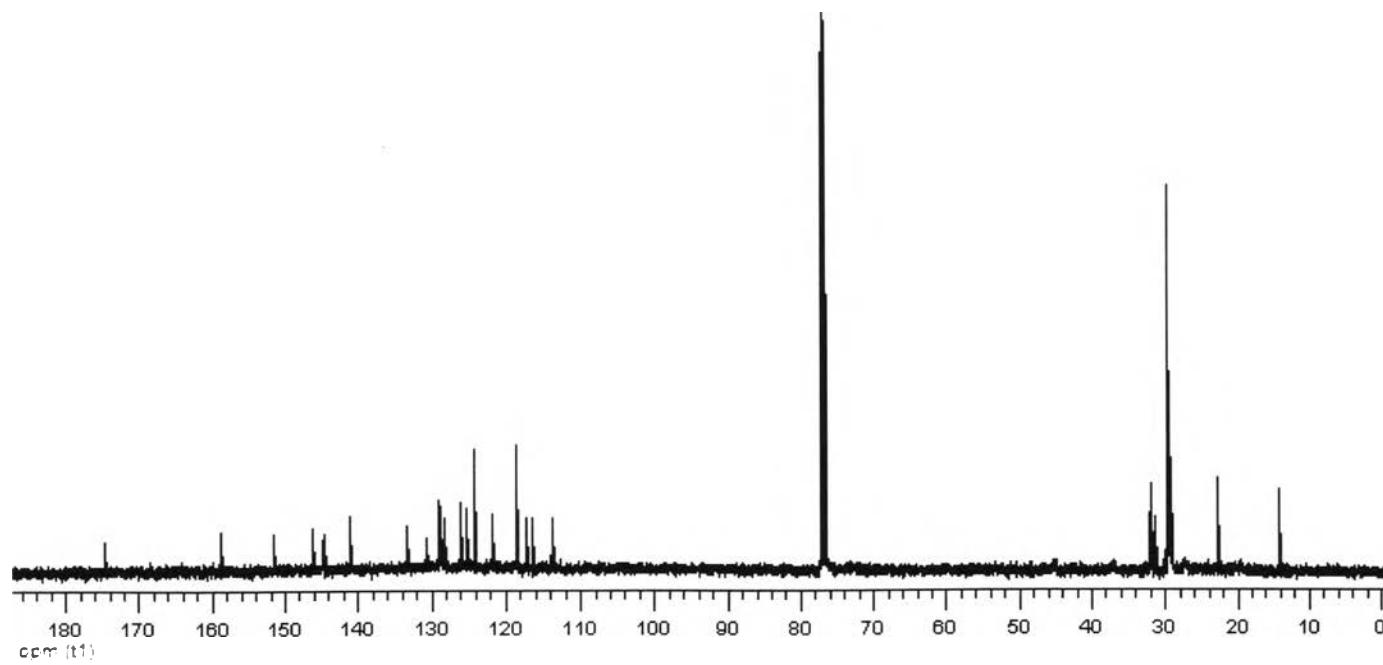


Figure A-24 ^{13}C -NMR spectrum of Compound 17.

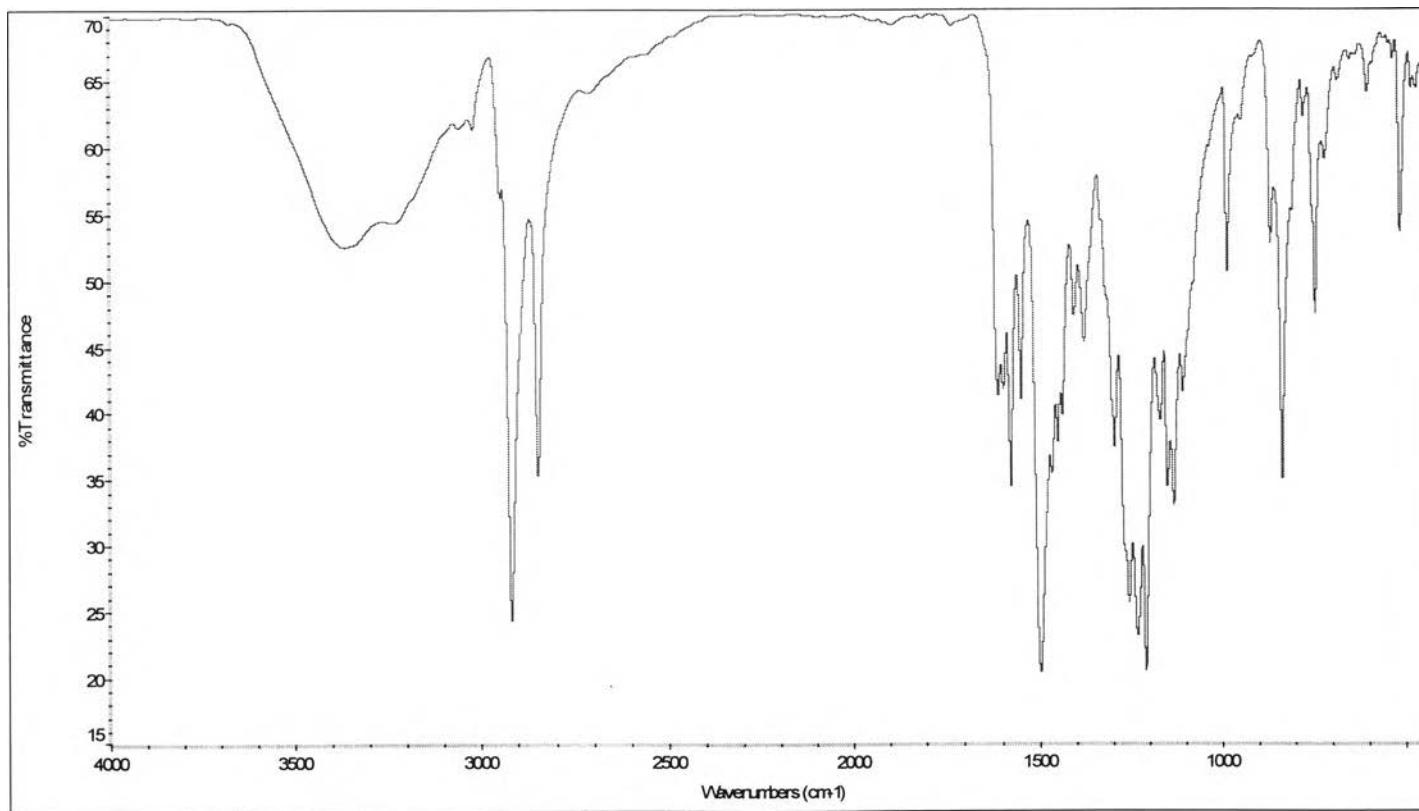


Figure A-25 IR spectrum of Compound 17.

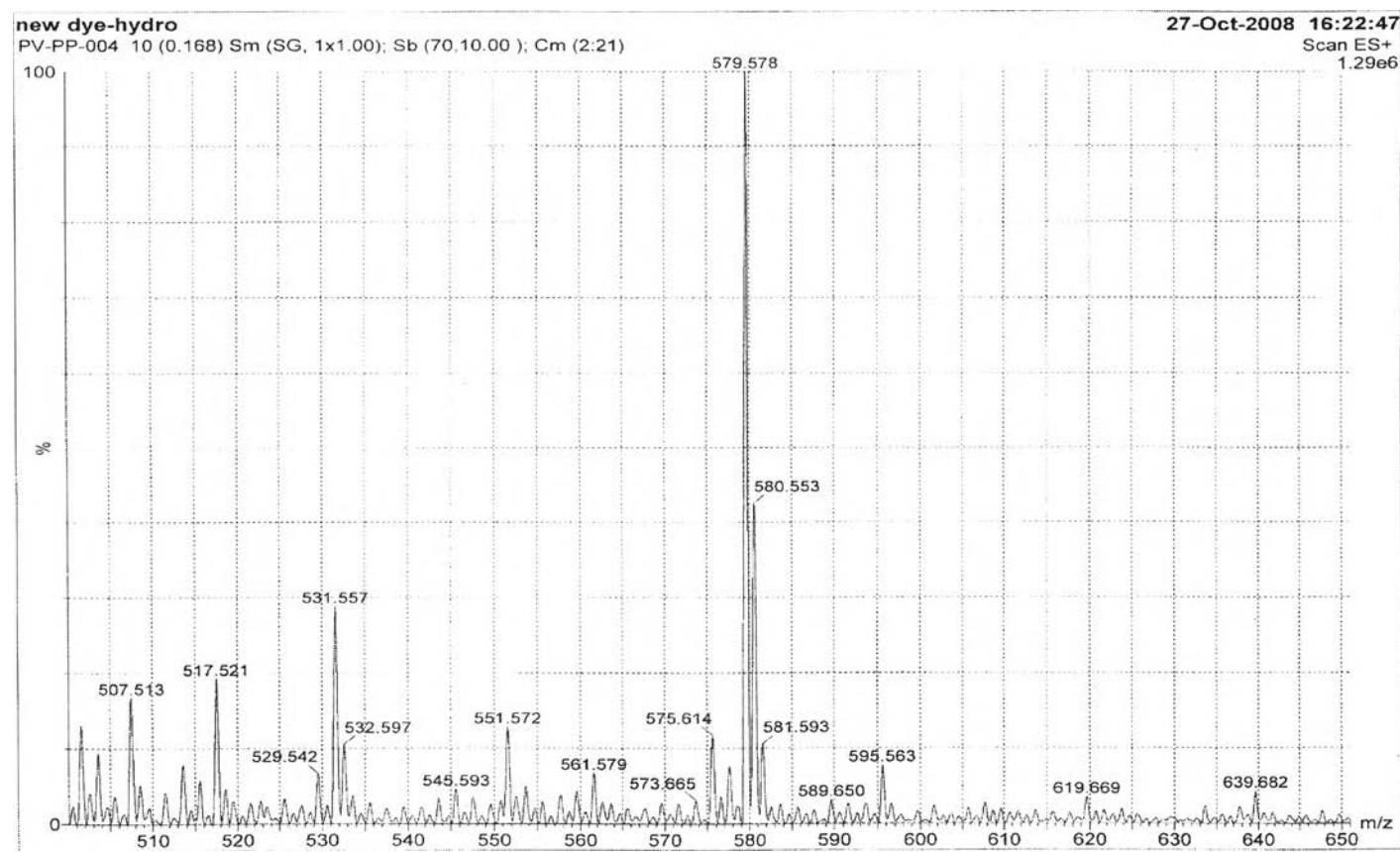


Figure A-26 Mass spectrum of Compound 17.

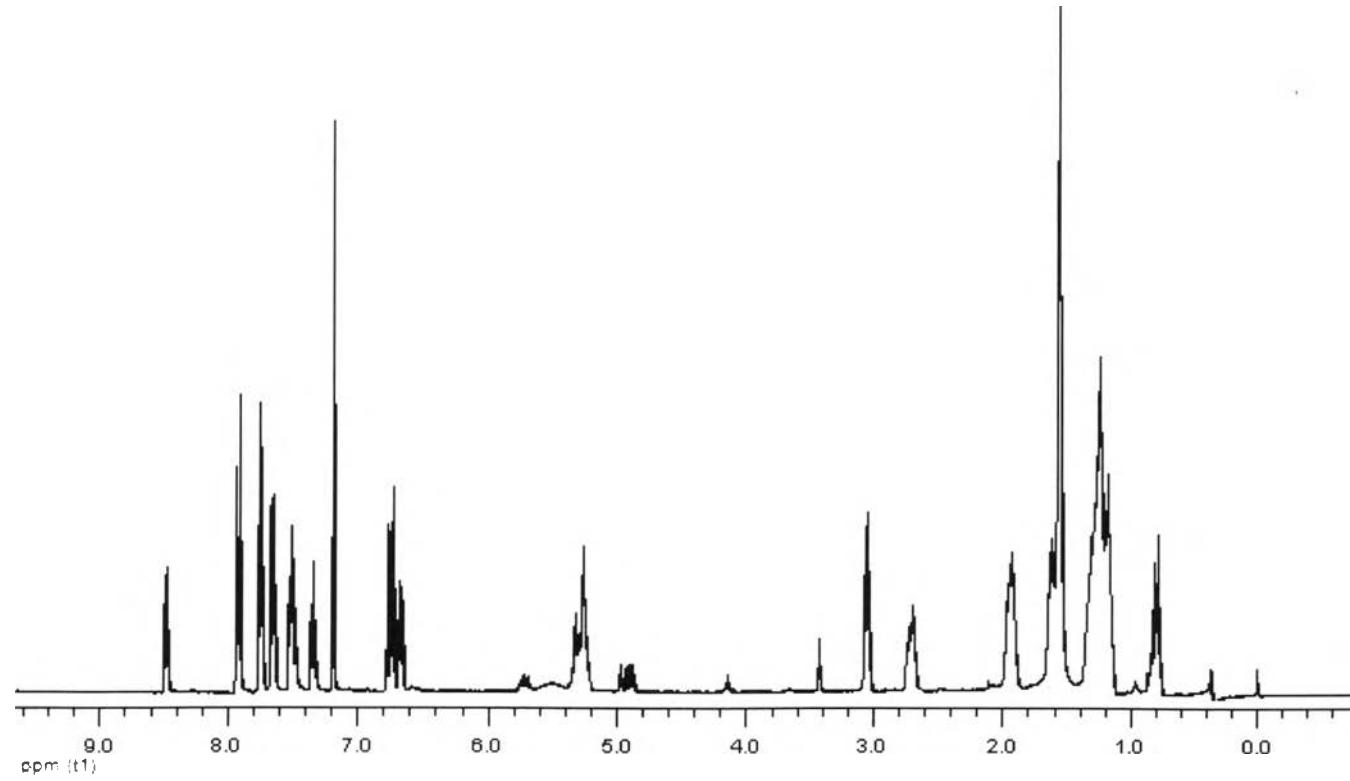


Figure A-27 ^1H -NMR spectrum of Compound 18.

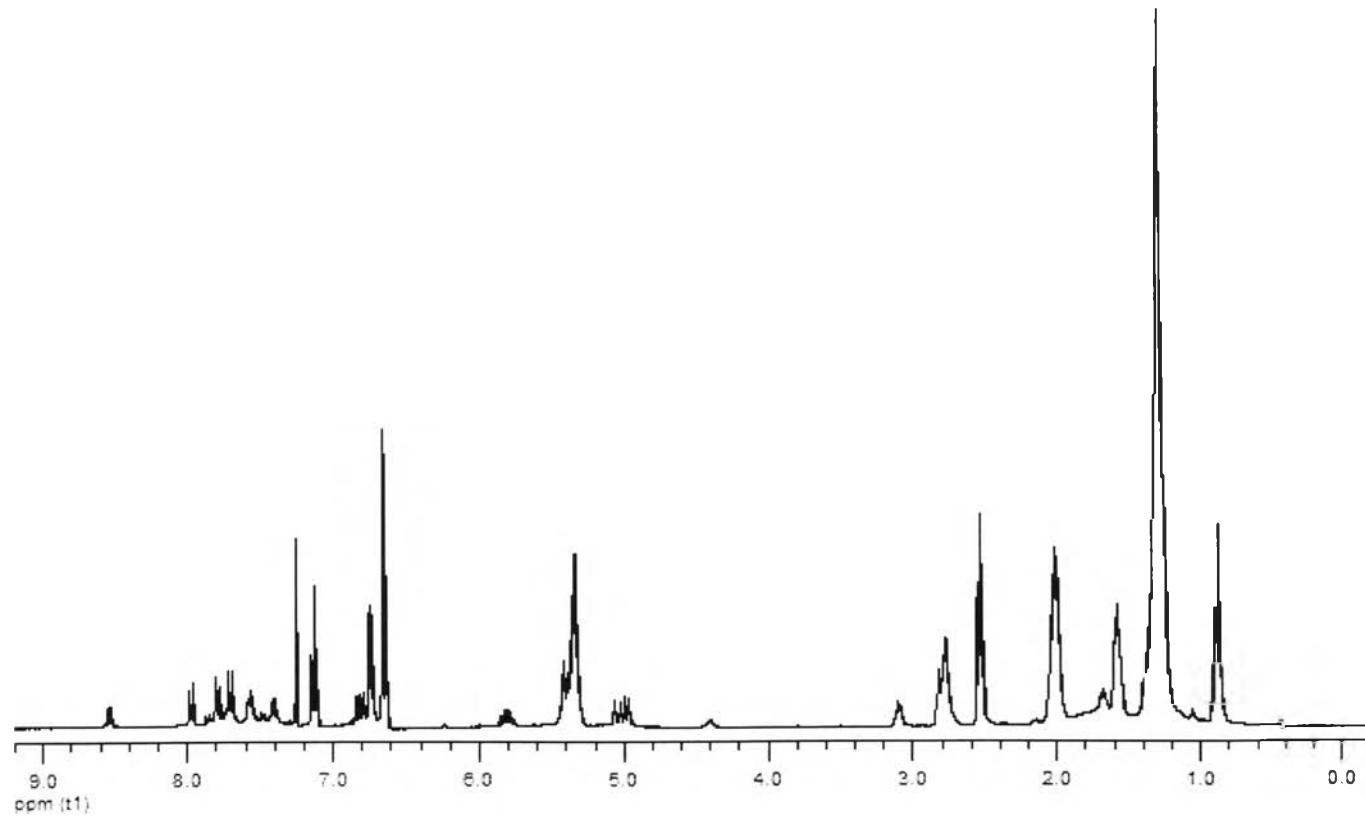


Figure A-28 ^1H -NMR spectrum of crude of Compound 18.

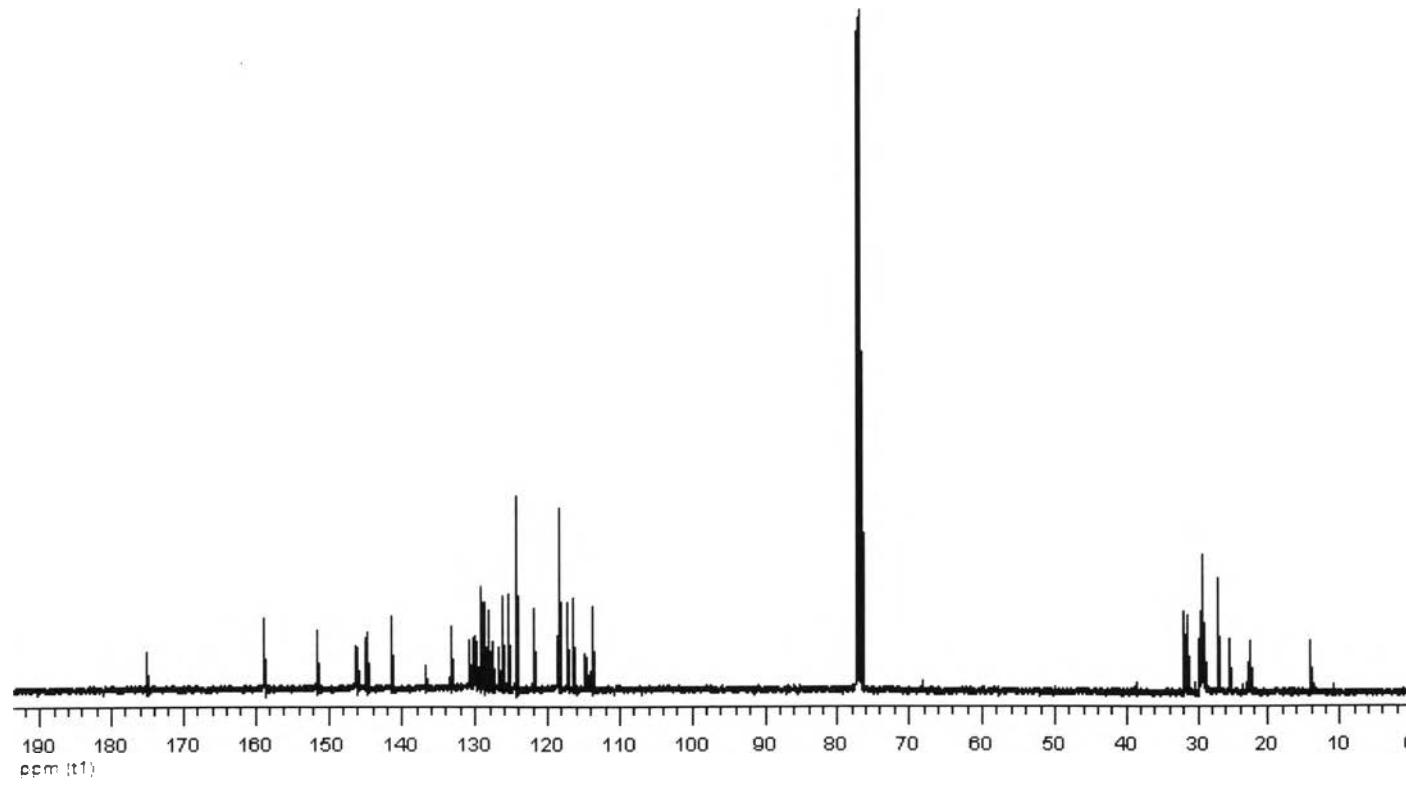


Figure A-29 ^{13}C -NMR spectrum of Compound 18.

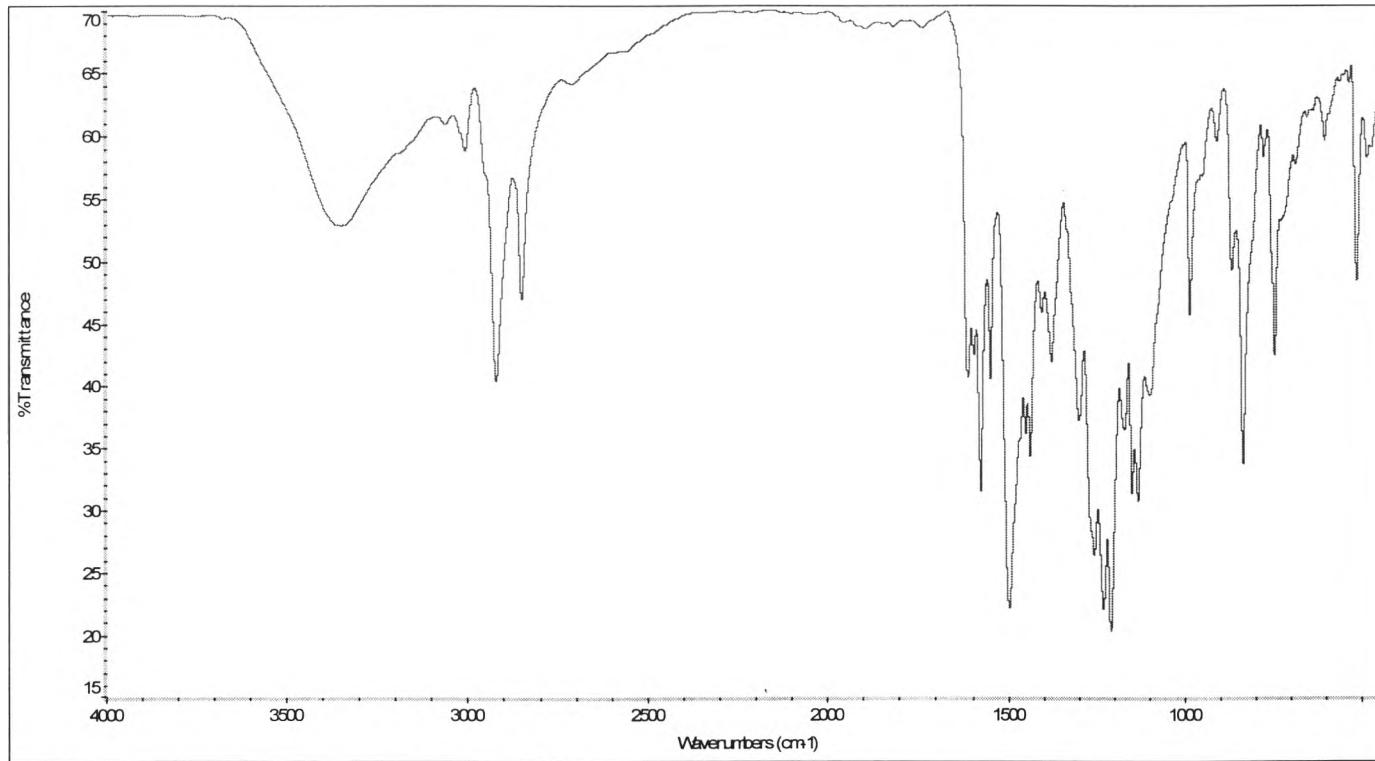


Figure A-30 IR spectrum of Compound 18.

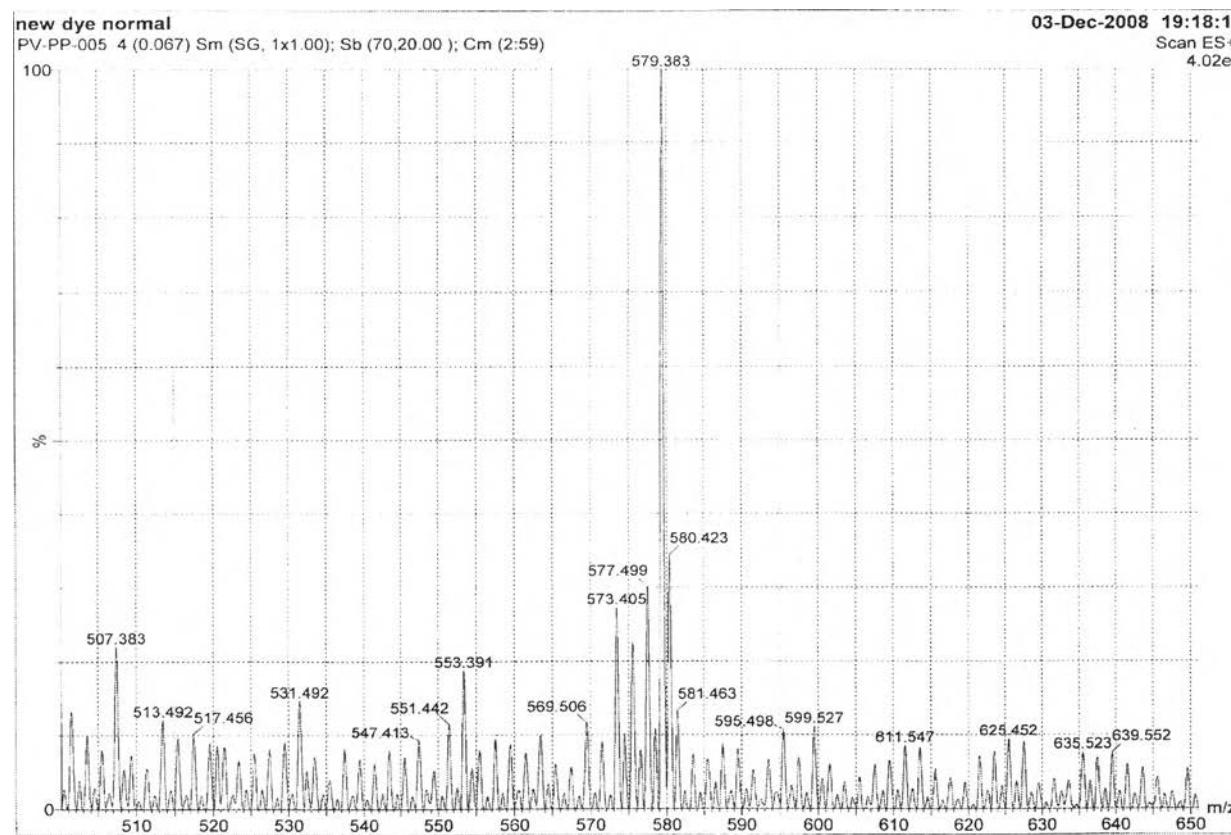


Figure A-31 Mass spectrum of Compound 18.

APPENDIX B

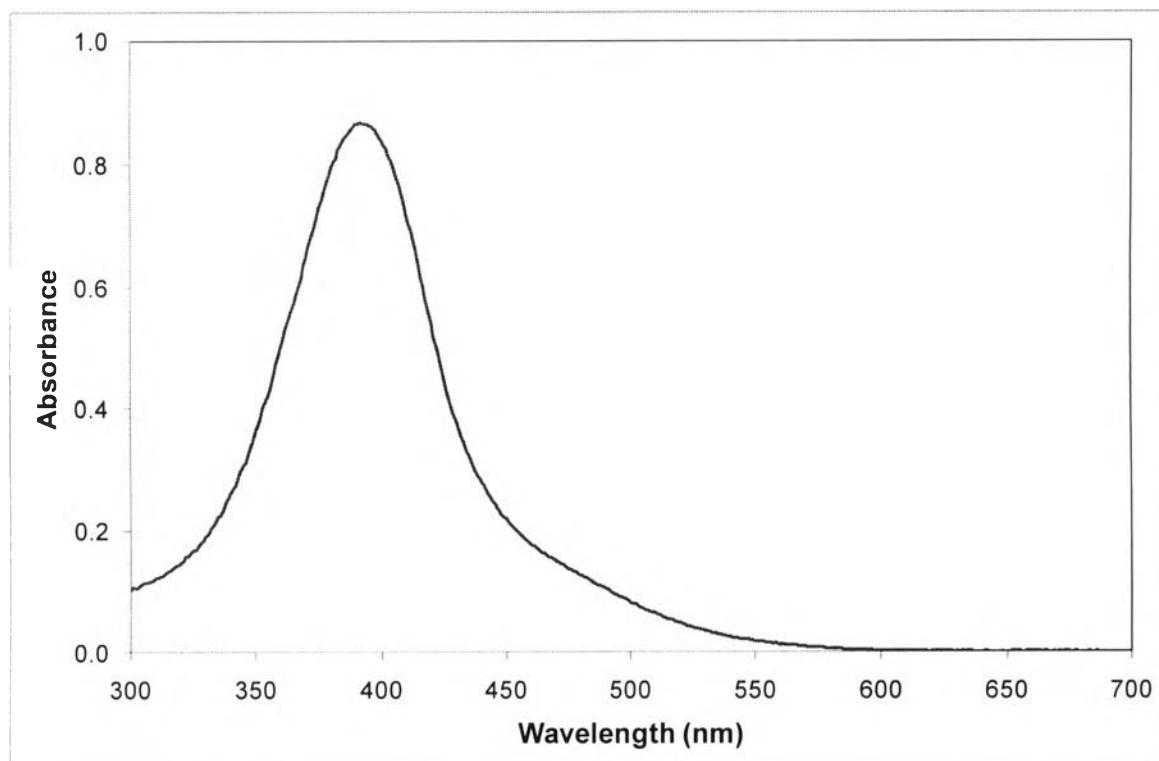


Figure B-1 Absorption spectrum of compound 6 in CH_2Cl_2 .

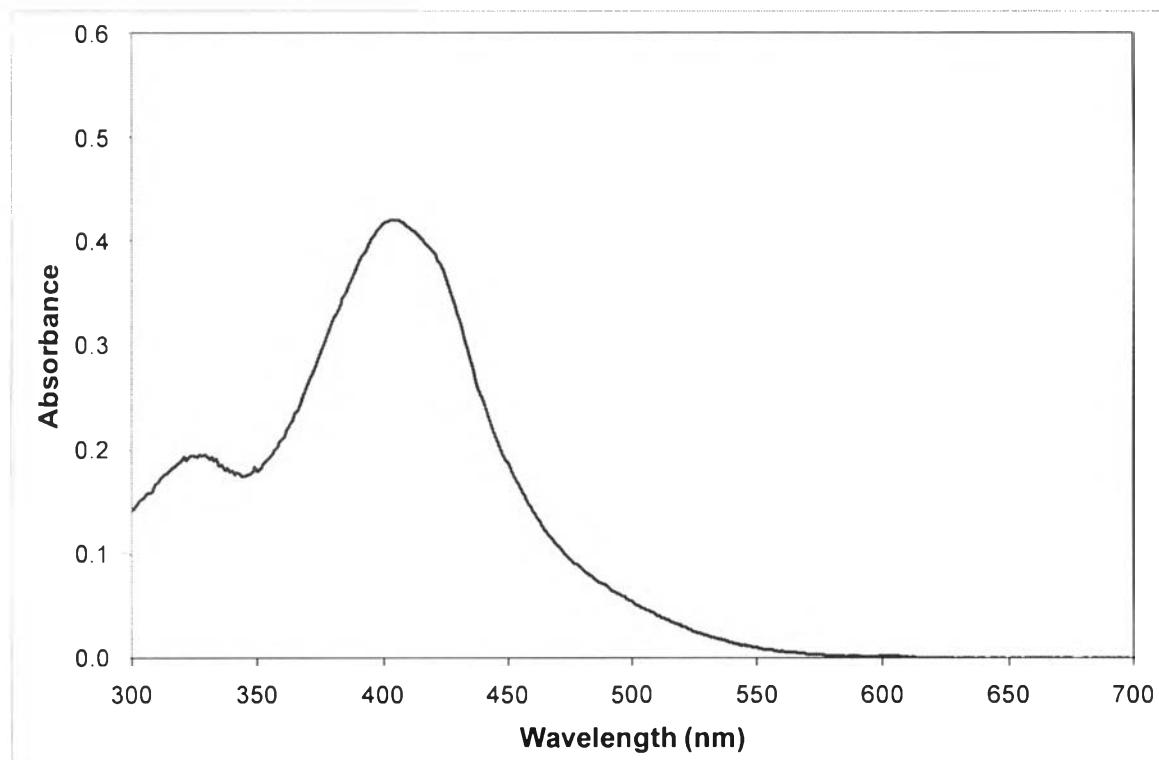


Figure B-2 Absorption spectrum of compound 7 in CH_2Cl_2 .

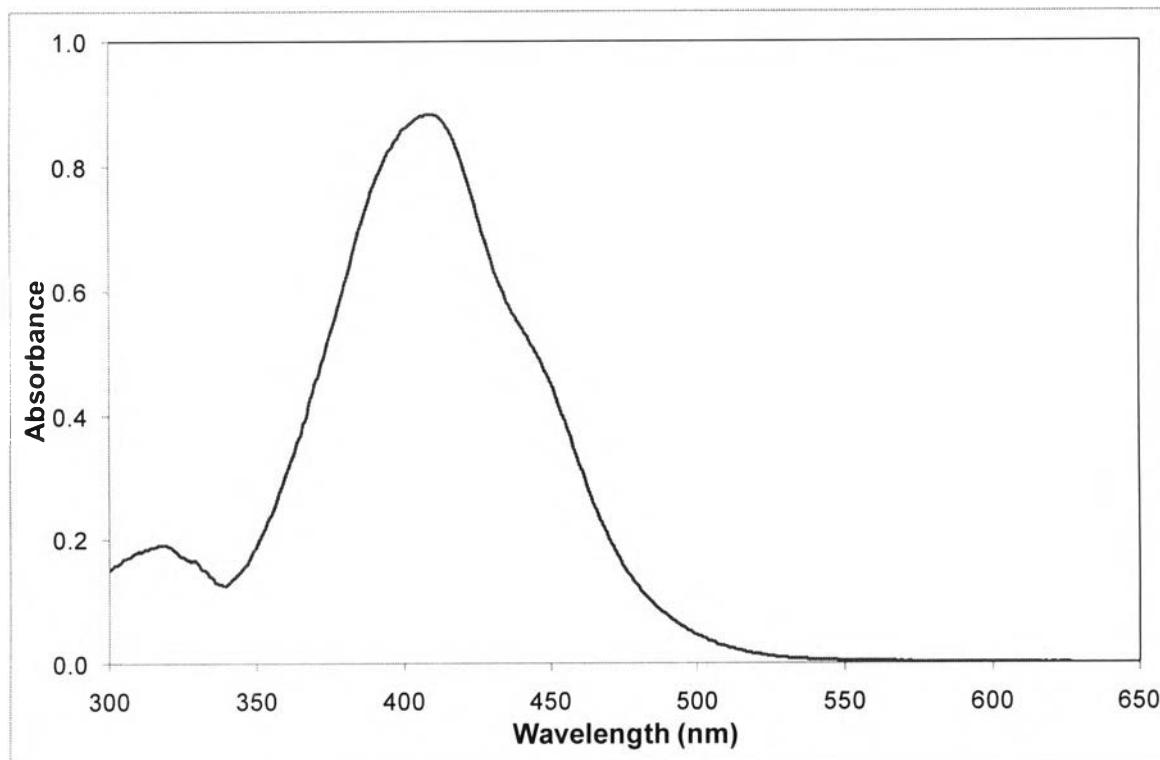


Figure B-3 Absorption spectrum of compound **12** in CH_2Cl_2 .

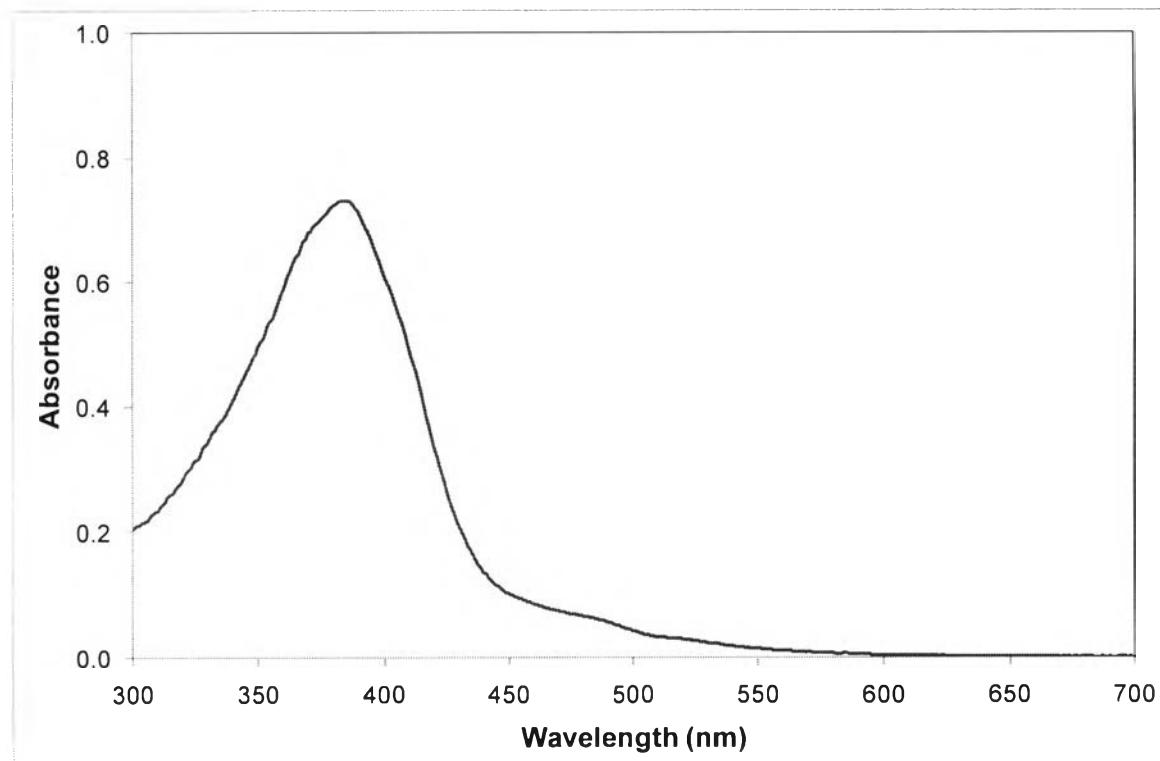


Figure B-4 Absorption spectrum of a mixture of compound **13** and cardanol in CH_2Cl_2 .

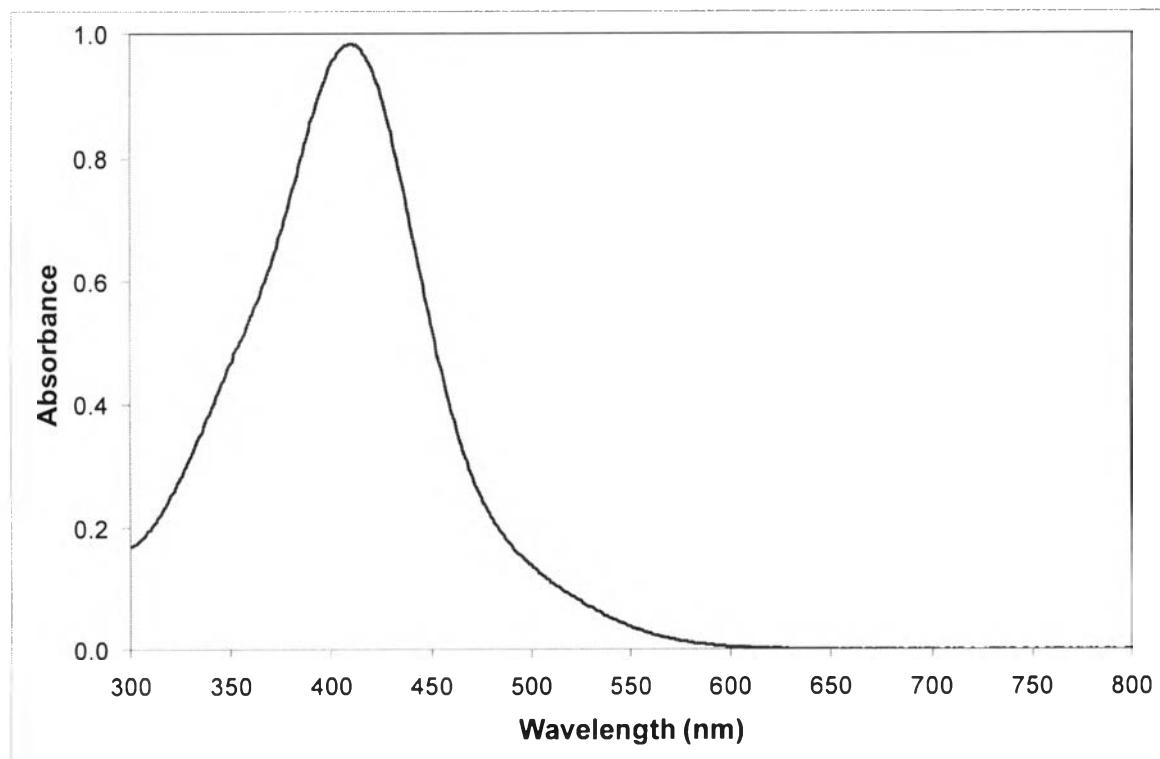


Figure B-5 Absorption spectrum of a mixture of compound **14** and cardanol in CH_2Cl_2 .

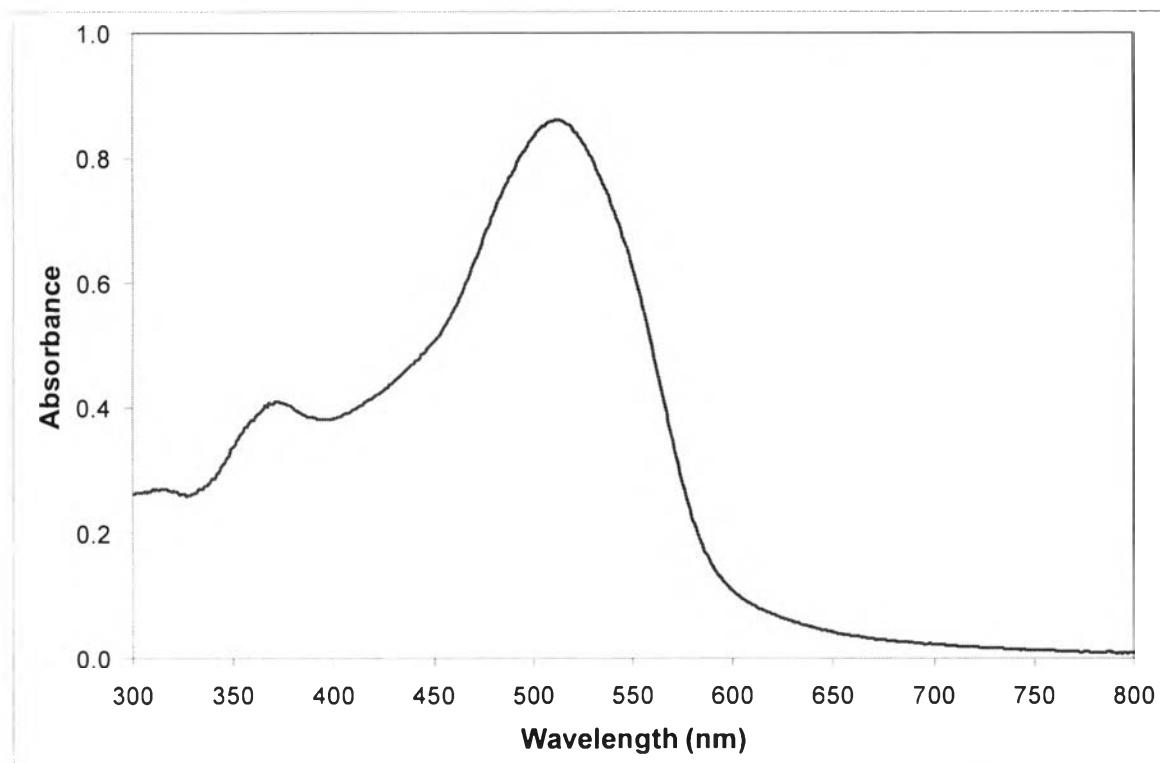


Figure B-6 Absorption spectrum of compound **17** in CH_2Cl_2 .

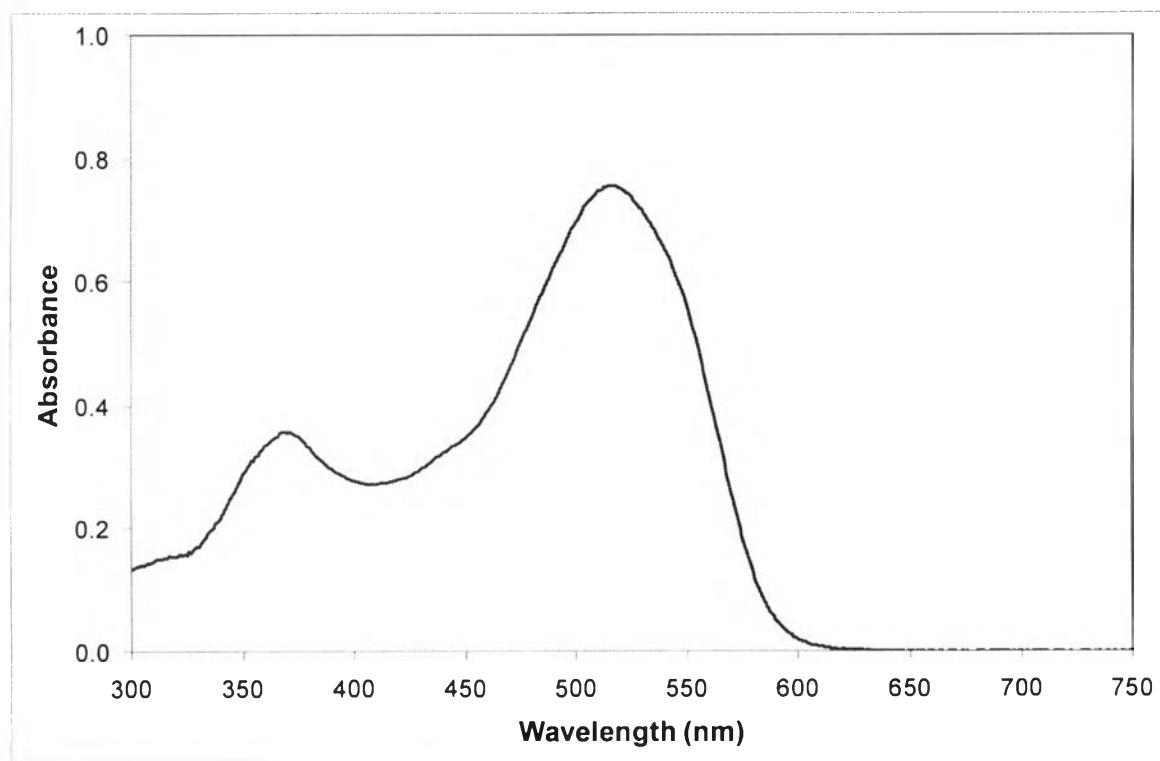


Figure B-7 Absorption spectrum of compound **18** in CH_2Cl_2 .

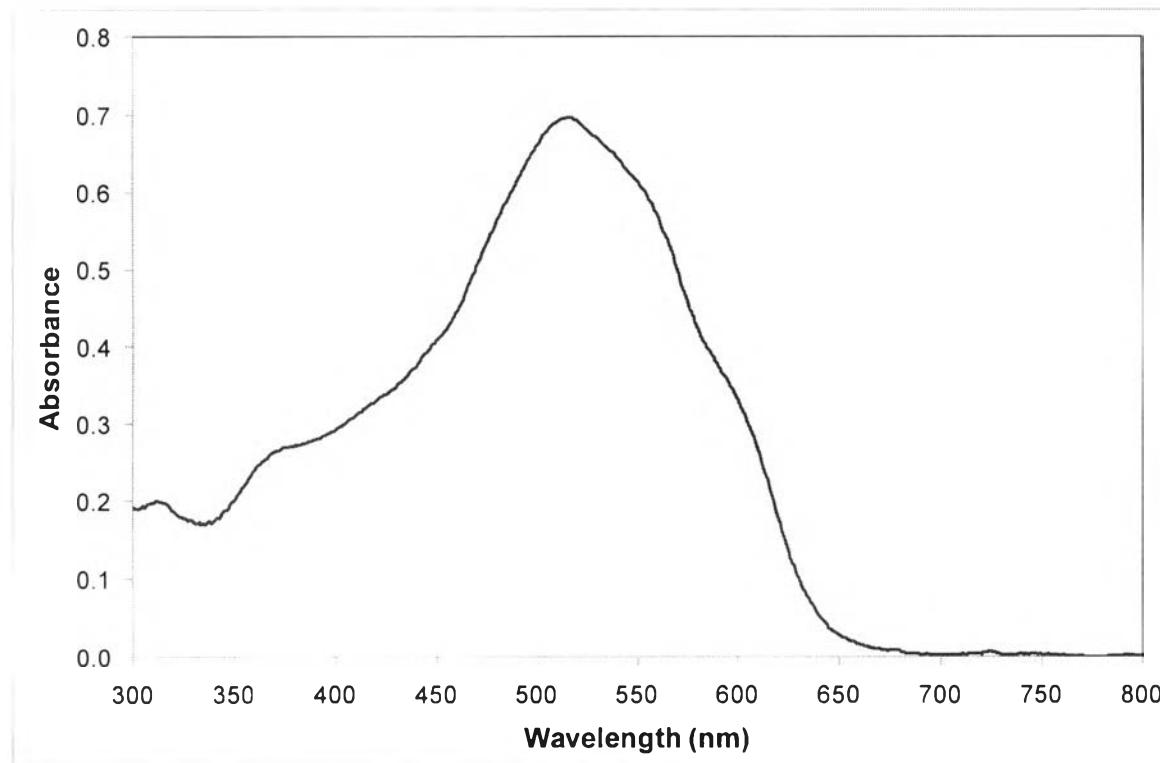


Figure B-8 Absorption spectrum of crude of compound **18** in CH_2Cl_2 .

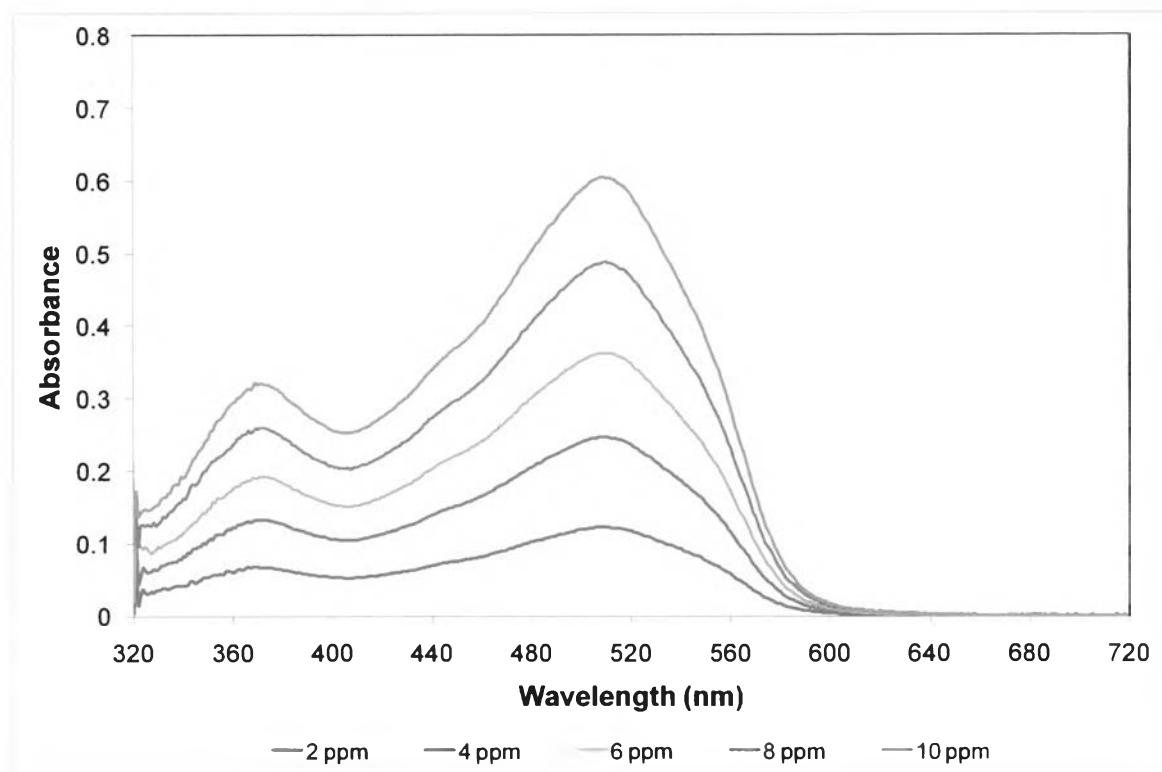


Figure B-9 Absorption spectrum of compound 18 in base gasoline 91 at 2, 4, 6, 8 and 10 ppm ($\lambda_{\text{abs}} = 509$ nm).

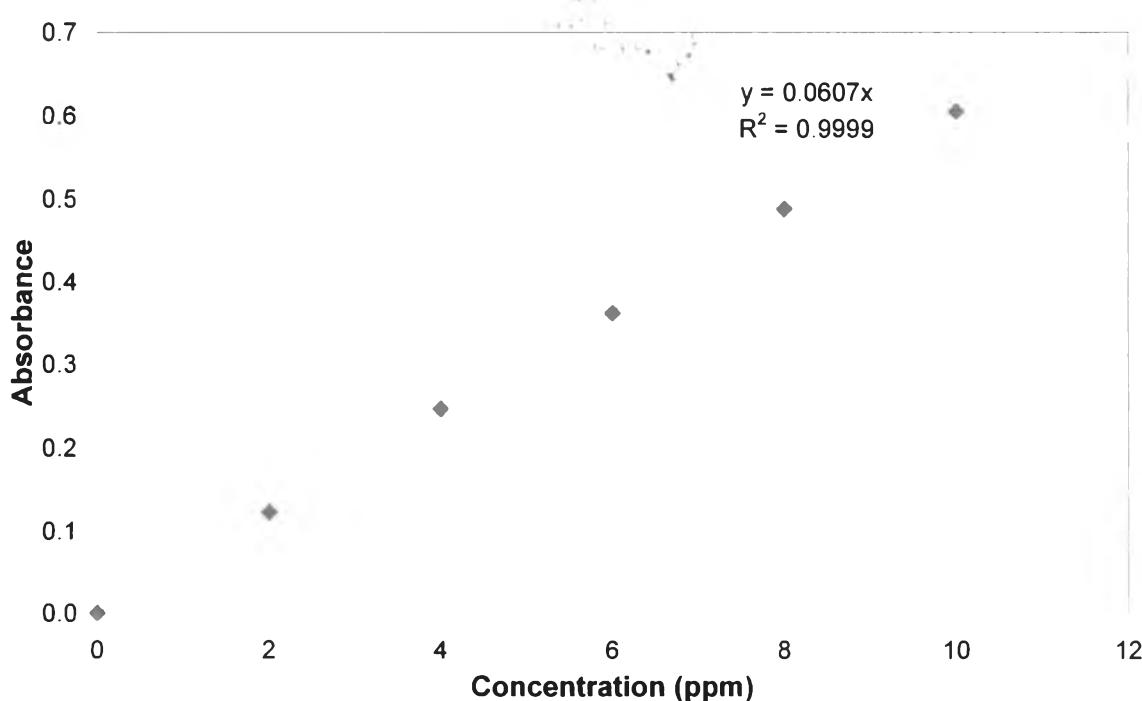


Figure B-10 Calibration curve for the quantitative determination of compound 18 in base gasoline 91 ($\lambda_{\text{abs}} = 509$ nm).

VITA

Miss Pimpaporn Paebumrung was born on September 2, 1983 in Trang province, Thailand. She got a Bachelor's Degree of chemistry in Faculty of Science at Chulalongkorn University, Bangkok in 2006. After that, she was admitted into a Master Degree program in Petrochemistry and Polymer Science, Faculty of Science, Chulalongkorn University, Bangkok in 2006 and completed the program in 2008.

