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ruthenium-terpyridine-based dyads attached to nanostructured TiO<sub>2</sub>. <u>Inorganic</u> <u>Chemistry</u>. 46 (2007): p. 638-651.

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APPENDIX A



Figure A.2 <sup>13</sup>C-NMR spectrum of compound 1



230 220 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 δ(ppm)

Figure A.4 <sup>13</sup>C-NMR spectrum of compound 2

3815213942



Figure A.6 <sup>13</sup>C-NMR spectrum of compound 3



Figure A.8 <sup>11</sup>H-NMR spectrum of compound 5



Figure A.9 <sup>13</sup>C-NMR spectrum of compound 5

# Mass Spectrum List Report

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20       407 1613       73343       3.0       60831       0.6602       6650         27       362 1650       15542       0.6125       6624         28       427 3047       4593       0.2       42.3       0.6625       6842         29       431 1733       17365       3.1       6917       0.0726       6021         36       432 1755       15601       0.6       130.6       0.0726       6023	25	264 1605	5571	9.2	513	0.0590	5509				
24 4021002 11542 75 1285 7565 884 28 4273047 4593 02 423 00625 6842 29 4371733 77365 31 5917 00726 6021 30 4381755 15601 06 7366 00720 6083	20	457 1613	/3373	30	506 1	0.0609	0690				
26 427 5047 4093 02 423 00665 5642 29 437 5730 21 6917 00726 6021 30 437 1735 15601 0.6 H39.6 0.9720 6083	27	402 1650	15546	0.5	128.9	0.059	5234				
20 404 1700 11000 21 5017 00720 0021 30 408 1765 15601 0.6 30.6 00720 6083	28	427 3047	4593	02	42 J 66-1 7	0.0525	0042				
	29 30	438 1765	17390 1560 1	06	5917 5917	0.0720	6023				





13 12 11 10 9 8 7 6 5 4 3 2 1 0 -i δ(ppm)

**Figure A.12** <sup>1</sup>H-NMR spectrum of compound 6 (DMSO- $d_6$ )



Figure A.14 IR spectra of compound 6

# Mass Spectrum List Report



Bruker Daltonics DataAnalysis 3.3

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Figure A.15 High resolution mass spectrum of compound 6







Figure A.18 IR spectra of compound 7

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#### Mass Spectrum List Report







Figure A.20 High resolution mass spectrum of compound 8



Figure A.21 IR spectra of compound 8

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APPENDIX B



Figure B.1 Absorption spectrum of compound 6 in DMSO



Figure B.2 Calibration curve for the quantitative determination of compound 6 in DMSO (  $\lambda_{\text{abs}}$  = 337 nm)



Figure B.3 Emission spectrum of compound 6 in DMSO ( $\lambda_{\text{ex}}$  = 350 nm)

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Figure B.4 Absorption spectrum of compound 7 in DMSO (  $\lambda_{\text{abs}}$  = 340 nm)



Figure B.5 Calibration curve for the quantitative determination of compound 7 in DMSO (  $\lambda_{abs}$  = 340 nm)



Figure B.6 Emission spectrum of compound 7 in DMSO ( $\lambda_{\text{ex}}$  = 350 nm)



Figure B.7 Absorption spectrum of compound 8 in DMSO (  $\lambda_{abs}$  = 340 nm)



Figure B.8 Calibration curve for the quantitative determination of compound 8 in DMSO (  $\lambda_{abs}$  = 340 nm)

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Figure B.9 Emission spectrum of compound 8 in DMSO ( $\lambda_{ex}$  = 350 nm)

### VITA

Miss Kobkun Sae-pang was born on July 30, 1987 in Phetchabun, Thailand. She got a Bachelor Degree of Petrochemical and Polymer engineering from Faculty of Engineering at Silpakorn University, Nakornprathom in 2008. Then, she was admitted into a Master Degree in the major Petrochemical and Polymer science, Faculty of Science, Chulalongkorn University, Bangkok in 2009 and completed the program in 2013.



