

## CHAPTER III

### EXPERIMENTAL

#### 3.1 Chemicals

All chemicals are purchased from commercial sources and used as received without further purification, unless noted otherwise.

1. Benzaldehyde : Merck
2. Boron trifluoride diethyletherate ( $\text{BF}_3 \cdot \text{Et}_2\text{O}$ ) : Fluka
3. Deuterated chloroform ( $\text{CDCl}_3$ ) : Cambridge Isotope
4. 1,8-diazabicyclo (5,4,0) undec-7-ene (DBU) : Sigma-Aldrich
5. 2,3-dichloro-5,6-dicyano benzoquinone (DDQ) : Sigma-Aldrich
6. Ethyl acetate : Distilled from commercial grade
7. Ethyl isocyanoacetate : Sigma-Aldrich
8. Hexanes : Distilled from commercial grade
9. 0.1 M Hydrochloric acid (HCl) : Merck
10. Anhydrous magnesium sulfate ( $\text{MgSO}_4$ ) : Merck
11. Methanol : Distilled from commercial grade
12. 2-methoxyphenylboronic acid : Sigma-Aldrich
13. Methylene chloride ( $\text{CH}_2\text{Cl}_2$ ) : Distilled from commercial grade
14. 1-nitrocyclohexene : Sigma-Aldrich
15. Pyrrole : Merck
16. Silica gel 60 particle size : Merck
17. Silica gel containing gypsum : Merck
18. Anhydrous sodium sulfate ( $\text{Na}_2\text{SO}_4$ ) : Merck
19. Sodium bicarbonate ( $\text{NaHCO}_3$ ) : Merck
20. 2-thiophene carboxaldehyde : Sigma-Aldrich
21. 2,2'-bithiophene-5-carboxaldehyde : Sigma-Aldrich
22. Toluene : RCI Lab-Scan
23. Triethylamine ( $\text{NEt}_3$ ) : Fluka

25. Trifluoroacetic acid (TFA)

: Fluka

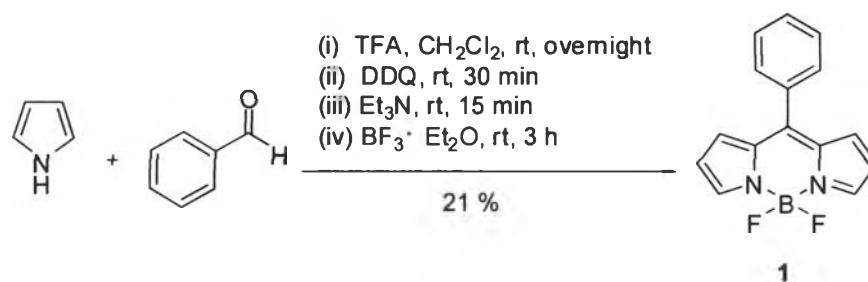
### 3.2 Analytical Instruments

$^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  spectra were obtained in  $\text{CDCl}_3$  at 400 MHz for  $^1\text{H}$  nuclei and 100 MHz for  $^{13}\text{C}$  nuclei (Varian Company, USA). Chemical shifts ( $\delta$ ) are reported in parts per million (ppm) relative to the residual  $\text{CHCl}_3$  peak (7.26 ppm for  $^1\text{H-NMR}$  and 77.0 for  $^{13}\text{C-NMR}$ ). Coupling constant ( $J$ ) is reported in Hertz (Hz). Mass spectra were obtained by high resolution electron spray ionization mass spectrometry (HR-ESI-MS) and matrix-assisted laser desorption ionization, MALDI (Bruker Daltonics, Germany) mass spectrometry with dithranol as a matrix. Absorption spectra were measured in toluene using a Hewlett-Packard 8453 spectrophotometer and absorption extinction coefficient ( $\mathcal{E}$ ) was reported in  $\text{M}^{-1}\cdot\text{cm}^{-1}$ . Fluorescence spectra were measured in toluene using a Perkin-Elmer LS45 luminescence spectrometer.

### 3.3 Experimental procedure

#### Part 1: Synthesis of BODIPY-thiophene derivatives

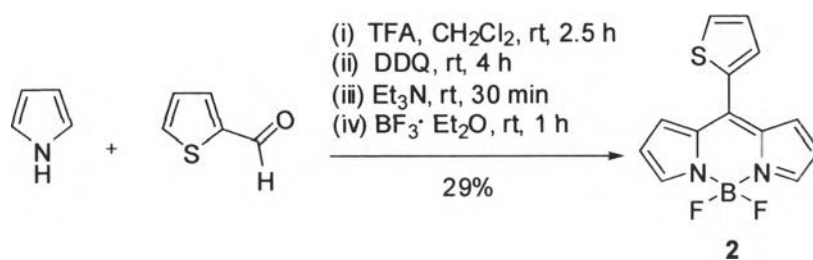
##### 3.3.1 Synthesis of compound 1



Following a previously published procedure [74], to benzaldehyde (0.531 g, 5.00 mmol) and pyrrole (0.671 g, 10.0 mmol) in deoxygenated  $\text{CH}_2\text{Cl}_2$  (150 mL), TFA (0.050 mL, 0.65 mmol) was added and the mixture was stirred at room temperature for overnight under  $\text{N}_2$ . The resulting solution was treated with DDQ (1.1390 g, 5.018 mmol) stirring was continued at room temperature for 30 min, followed by the addition of  $\text{Et}_3\text{N}$  (15.33 mL, 0.1099 mol). After 15 min,  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (15.21 mL, 0.1200 mol) was added at  $0^\circ\text{C}$ , and the mixture was stirred at room temperature for additional 3 h. After washing with a saturated aqueous solution of  $\text{NaHCO}_3$ , the organic phase was collected, dried over  $\text{MgSO}_4$ , filtered, and concentrated to dryness. The residue was purified by silica gel column chromatography (ethyl

acetate/hexanes; 2:1) to afford **1** as an orange solid (113.6 mg, 21%).  $^1\text{H-NMR}$ :  $\delta_{\text{H}}$  6.55 (d,  $J = 2.4$  Hz, 2H), 6.94 (d,  $J = 3.2$  Hz, 2H), 7.50–7.61 (m, 5H), 7.95 (s, 2H) (**Figure A-1**); MALDI-TOF-MS obsd 267.399 ( $[\text{M}]^+$ ), calcd 268.0983 ( $[\text{M}]^+$ ,  $\text{M}=\text{C}_{15}\text{H}_{11}\text{BF}_2\text{N}_2$ ) (**Figure A-2**);  $\lambda_{\text{abs}}$  ( $\mathcal{E}$ ) 344, 503 nm ( $0.5 \times 10^5$ ) (**Figures B-1 and B-2**);  $\lambda_{\text{em}}$  ( $\lambda_{\text{ex}} = 470$  nm) 521 nm (**Figure B-3**). Other spectroscopic data are consistent with those described in the literature.

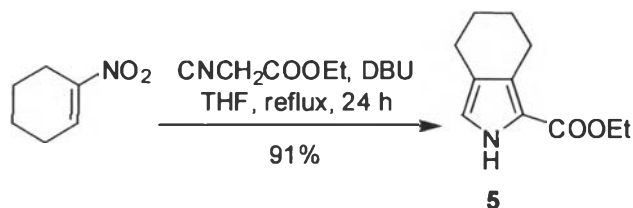
### 3.3.2 Synthesis of compound **2**



Following a previously published procedure [71], 2-thiophene carboxaldehyde (0.200 g, 1.79 mmol) was dissolved in pyrrole (1.79 mL, 25.0 mmol) and TFA (0.27 mL, 3.5 mmol) were added. The reaction was allowed to proceed at room temperature for 2.5 h. A solution of DDQ (0.405 g, 1.79 mmol) in dichloromethane (20 mL) was added and the reaction continued at room temperature for additional 4 h.  $\text{NEt}_3$  (3.09 mL, 23.2 mmol) was added to the reaction mixture, which was stirred at room temperature for 30 min. After that,  $\text{BF}_3 \cdot \text{OEt}_2$  (3.85 mL, 30.4 mmol) was added and the reaction was stirred at room temperature for 1 h. The reaction mixture was then washed with  $\text{H}_2\text{O}$  (2 $\times$ 50 mL) and the aqueous phase was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic layers were dried over anhydrous  $\text{MgSO}_4$  and concentrated to dryness. The resulting crude was purified by silica column chromatography ( $\text{CH}_2\text{Cl}_2$ /hexanes; 4:1) to afford **2** as an orange solid (175.0 mg, 29%).  $^1\text{H-NMR}$ :  $\delta_{\text{H}}$  6.60 (d,  $J = 3.6$  Hz, 2H), 7.26–7.35 (m, 3H), 7.60 (d,  $J = 3.6$  Hz, 1H), 7.74 (d,  $J = 5.2$  Hz, 1H), 7.95 (s, 2H) (**Figure A-3**);  $^{13}\text{C-NMR}$   $\delta_{\text{C}}$  118.5, 128.2, 131.4, 131.5, 133.0, 134.3, 134.5, 139.5, 143.8 (**Figure A-4**); HR-ESI-MS obsd 297.0448 ( $[\text{M} + \text{Na}]^+$ ), calcd 297.0445 ( $[\text{M} + \text{Na}]^+$ ), 274.0548 ( $[\text{M}]^+$ ,  $\text{M}=\text{C}_{13}\text{H}_9\text{BF}_2\text{N}_2\text{S}$ ) (**Figure A-5**);  $\lambda_{\text{abs}}$  ( $\mathcal{E}$ ) 393, 514 nm ( $0.5 \times 10^5$ ) (**Figures B-5 and B-6**);  $\lambda_{\text{em}}$  ( $\lambda_{\text{ex}} = 480$  nm) 617 nm (**Figure B-7**).

## Past 2: Synthesis of benzoBODIPY-thiophene derivaatives

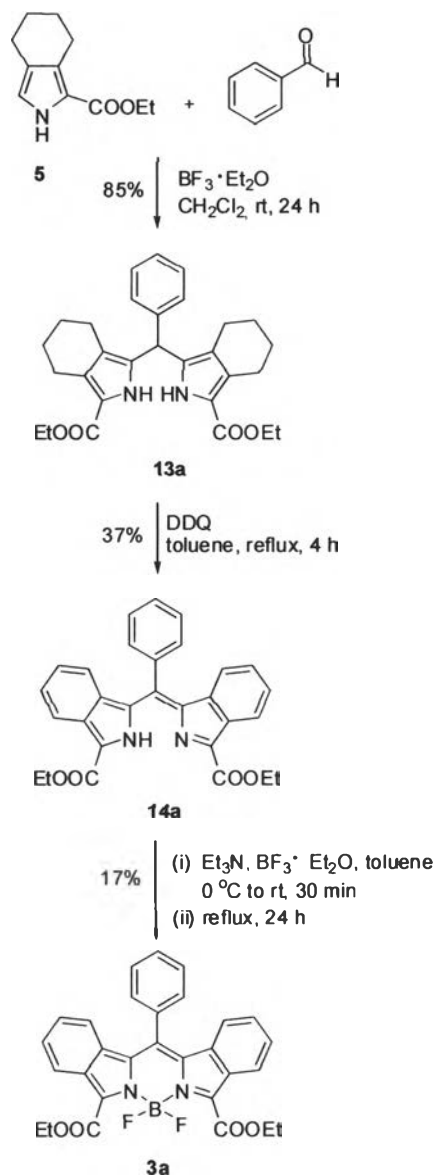
## 3.3.3 Synthesis of 2H-isindole-4,5,6,7-tetrahydro-1-carboxylic ethyl ester (5)



Following a previously published procedure [55], a 3-neck-round bottom flask equipped with a condenser was purged with  $\text{N}_2$ . Then 1-nitrocyclohexene (4.40 mL, 39.2 mmol) and ethyl isocyanoacetate (4.30 mL, 39.2 mmol) were dissolved in dry THF (100 mL). To this solution, DBU (5.50 mL, 39.2 mmol) was slowly added and the reaction was refluxed for 24 h. The solvent was removed under reduced pressure and the crude product was purified on a silica column ( $\text{CH}_2\text{Cl}_2$ /hexanes; 4:1) to afford **5** as pale yellow crystals (6.853 g, 91%).  $^1\text{H-NMR}$ :  $\delta_{\text{H}}$  1.34 (t,  $J = 6.8$  Hz, 3H), 1.51–1.63 (m, 4H), 2.47–2.58 (m, 2H), 2.74–2.86 (m, 2H), 4.29 (q,  $J = 7.2$  Hz, 2H), 6.64 (s, 1H), 8.76 (br s, 1H) (Figure A-26). Other spectroscopic data are consistent with those described in the literature.



## 3.3.4 Synthesis of compound 3a



Following a published procedure with slight modification [73], isoindole 5 (0.500 g, 2.59 mmol) and benzaldehyde (0.13 mL, 1.3 mmol) were dissolved in  $\text{CH}_2\text{Cl}_2$  (50 mL). Then,  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (0.032 mL, 0.26 mmol) was added dropwise and the solution was stirred at room temperature for 24 h. The solvent was evaporated under vacuum and the resulting residue was chromatographed on a silica column ( $\text{CH}_2\text{Cl}_2$ /hexanes, 4:1) to give **13a** as a white solid (0.523 g, 85%).  $^1\text{H-NMR}$ :  $\delta_{\text{H}}$  1.29 (t,  $J = 7.0$  Hz, 6H), 1.58–1.75 (m, 8H), 2.18 (m, 4H), 2.77 (m, 4H), 4.15–4.22 (m, 4H), 5.40 (s, 1H), 7.09 (d,  $J = 6.8$  Hz, 2H), 7.27–7.35 (m, 3H), 8.46 (br s, 1H), 8.58 (br s, 1H) (Figure



**A-18**);  $^{13}\text{C-NMR } \delta_{\text{c}}$  14.5, 21.2, 23.1, 23.3, 40.6, 59.7, 116.7, 119.7, 127.3, 128.2, 128.9, 129.2, 130.8, 139.1, 161.8 (**Figure A-19**);  $\lambda_{\text{abs}} (\mathcal{E})$  289 nm ( $0.4 \times 10^5$ ) (**Figures B-25** and **B-26**). Other spectroscopic data are consistent with those described in the literature.

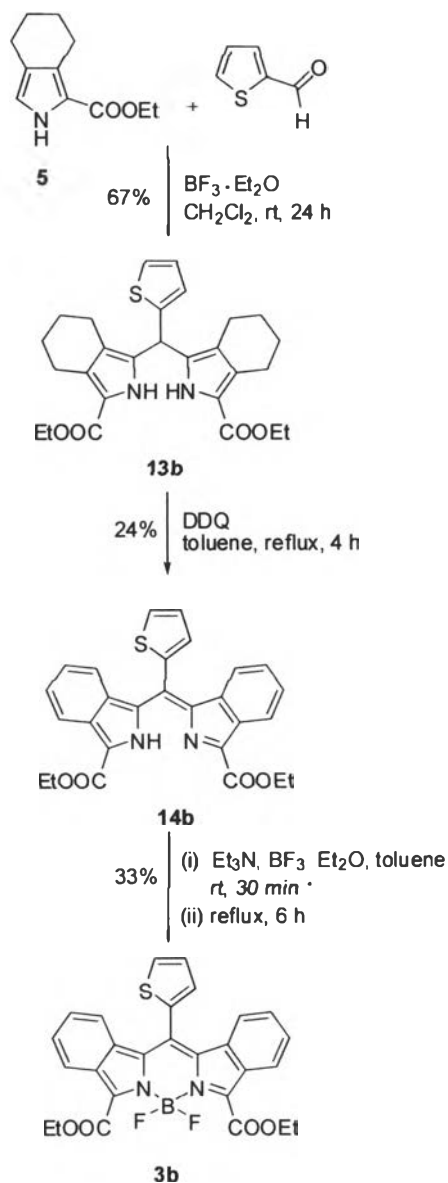
Following a previously published procedure [13], a solution of dipyrromethane **13a** (0.796 g, 1.68 mmol) in toluene (30 mL) was heated to 110°C. A solution of DDQ (2.666 g, 11.74 mmol) in toluene (20 mL) was then added and the mixture was refluxed for 4 h. After that, the solvent was removed under reduced pressure. The resulting residue was dissolved in ethyl acetate and extracted with a 0.1 M solution of HCl to remove traces of DDQ (3×50 mL), washed once with brine (50 mL) and dried over anhydrous  $\text{Na}_2\text{SO}_4$ . The solvent was removed under reduced pressure and the residue was purified by column chromatography ( $\text{CH}_2\text{Cl}_2$ /hexanes, 4:1) to dipyrin **14a** as a purple solid (0.285 g, 37%).  $^1\text{H-NMR: } \delta_{\text{H}}$  1.57 (t,  $J = 7.2$  Hz, 6H), 4.58 (q,  $J = 7.2$  Hz, 4H), 6.12 (d,  $J = 8.0$  Hz, 2H), 6.97 (t,  $J = 8.0$  Hz, 2H), 7.21–7.24 (m, 2H), 7.53 (d,  $J = 6.8$  Hz, 2H), 7.63–7.71 (m, 3H), 8.19 (d,  $J = 8.0$  Hz, 2H) (**Figure A-27**);  $^{13}\text{C-NMR } \delta_{\text{c}}$  14.5, 61.3, 122.1, 122.9, 126.1, 127.0, 129.2, 129.5, 131.8, 135.1, 135.6, 136.5, 137.9, 138.3, 161.9 (**Figure A-28**); MALDI-TOF-MS obsd 463.710 ( $[\text{M}]^+$ ), calcd 464.512 ( $[\text{M}]^+$ ,  $\text{M}=\text{C}_{29}\text{H}_{24}\text{N}_2\text{O}_4$ ) (**Figure A-29**);  $\lambda_{\text{abs}} (\mathcal{E})$  574 nm ( $0.4 \times 10^5$ ) (**Figures B-31** and **B-32**). Other spectroscopic data are consistent with those subscribed in the literature.

Following a previously published procedure [13], a solution of dipyrin **14a** (0.285 g, 0.614 mmol) in toluene (20 mL) was treated with  $\text{Et}_3\text{N}$  (0.55 mL, 4.0 mmol) and  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (0.78 mL, 6.3 mmol) at 0°C and the mixture was stirred at room temperature for 30 min. The reaction mixture was then refluxed for additional 24 h. The solution was washed with a 10% aqueous solution of  $\text{NaHCO}_3$  (20 mL), brine (20 mL) and dried over anhydrous  $\text{Na}_2\text{SO}_4$ . The solvent was evaporated under reduced pressure and purified by silica column ( $\text{CH}_2\text{Cl}_2$ /hexanes, 4:1) to give a mixture containing BODIPY **3a** as a blue solid. Due to impurity having a very similar  $R_f$  value, **3a** could not be completely purified (~90% purity, 53 mg, 17%).  $^1\text{H-NMR: } \delta_{\text{H}}$  1.55 (t,  $J = 7.2$  Hz, 6H), 4.62 (q,  $J = 7.2$  Hz, 4H), 6.17 (d,  $J = 8.4$  Hz, 2H), 7.08 (t,  $J = 8.0$  Hz, 2H), 7.23–7.27 (m, 2H), 7.53 (d,  $J = 7.2$  Hz, 2H), 7.68–7.75 (m, 3H), 8.11 (d,  $J = 8.0$  Hz, 2H) (**Figure A-6**);  $^{13}\text{C-NMR } \delta_{\text{c}}$  14.2, 61.3, 121.8, 124.2, 126.5, 128.5, 129.2, 129.7, 129.8, 130.1, 131.3, 134.3, 134.7, 140.3, 141.2, 160.7 (**Figure A-7**); MALDI-TOF-MS obsd 511.842 ( $[\text{M}]^+$ ), calcd 512.172 ( $[\text{M}]^+$ ,  $\text{M}=\text{C}_{29}\text{H}_{23}\text{BF}_2\text{N}_2\text{O}_4$ ) (**Figure A-8**);  $\lambda_{\text{abs}} (\mathcal{E})$  642 nm



( $1.0 \times 10^6$ ) (Figures B-9 and B-10);  $\lambda_{em}$  ( $\lambda_{ex} = 600$  nm) 663 nm (Figure B-11). Other spectroscopic data are consistent with those subscribed in the literature.

### 3.3.5 Synthesis of compound 3b



Following a published procedure with slight modification [73], isoindole **5** (1.000 g, 5.17 mmol) and 2-thiophene carboxaldehyde (0.24 mL, 2.6 mmol) were dissolved in  $\text{CH}_2\text{Cl}_2$  (70 mL). Then,  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (0.065 mL, 0.52 mmol) was added dropwise and the solution was stirred 24 h at room temperature. The solvent was evaporated under vacuum and the resulting residue was chromatographed on a silica column ( $\text{CH}_2\text{Cl}_2$ /hexanes, 4:1) give **13b** as a white solid (1.669 g, 67%).  $^1\text{H-NMR}$ :  $\delta_{\text{H}}$

1.25 (t,  $J = 7.2$  Hz, 6H), 1.73 (s, 8H), 2.38 (d,  $J = 5.6$  Hz, 4H), 2.80 (s, 4H), 4.13 (q,  $J = 7.2$  Hz, 4H), 5.69 (s, 1H), 6.73 (s, 1H), 6.83–6.89 (m, 1H), 7.16 (d,  $J = 4.6$  Hz, 1H), 9.90 (br s, 2H), (**Figure A-20**);  $^{13}\text{C-NMR}$   $\delta_{\text{c}}$  14.4, 21.3, 23.2, 23.3, 23.4, 29.7, 35.7, 59.8, 117.1, 119.4, 124.8, 125.8, 126.8, 129.0, 130.9, 143.4, 161.1 (**Figure A-21**); HR-ESI-MS obsd 503.1975 ( $[\text{M} + \text{Na}]^+$ ), calcd 503.1980 ( $[\text{M} + \text{Na}]^+$ ), 480.2083 ( $[\text{M}]^+$ ,  $\text{M}=\text{C}_{27}\text{H}_{32}\text{N}_2\text{O}_4\text{S}$ ) (**Figure A-22**);  $\lambda_{\text{abs}}$  ( $\epsilon$ ) 288 nm ( $0.3 \times 10^5$ ) (**Figures B-27 and B-28**).

Following a published procedure [13] with slight modification, a solution of dipyrromethane **13b** (0.488g, 1.02 mmol) in toluene (30 mL) was heated to 110°C. A solution of DDQ (2.077 g, 9.151 mmol) in toluene (20 mL) was then added and the mixture was refluxed for 4 h. After that, the solvent was removed under reduced pressure. The resulting residue was dissolved in ethyl acetate and extracted with a 0.1 M solution of HCl to remove traces of DDQ (50 mL), washed once with brine (50 mL) and dried over anhydrous  $\text{Na}_2\text{SO}_4$ . The solvent was removed under reduced pressure and the residue was purified by column chromatography ( $\text{CH}_2\text{Cl}_2$ /hexanes, 4:1) to give dipyrin **14b** as a purple solid (0.112 g, 24%).  $^1\text{H-NMR}$ :  $\delta_{\text{H}}$  1.48 (t,  $J = 6.5$  Hz, 6H), 4.49 (q,  $J = 6.8$  Hz, 4H), 6.26 (d,  $J = 8.4$  Hz, 2H), 7.00 (t,  $J = 7.2$  Hz, 2H), 7.11–7.24 (m, 3H), 7.29 (s, 1H), 7.65 (d,  $J = 4.4$  Hz, 1H), 8.11 (d,  $J=8.0$  Hz, 2H) (**Figure A-30**);  $^{13}\text{C-NMR}$   $\delta_{\text{c}}$  14.4, 29.3, 29.6, 122.0, 122.9, 126.3, 127.2, 128.0, 128.2, 128.9, 161.7 (**Figure A-31**); HR-ESI-MS obsd 493.1202 ( $[\text{M} + \text{Na}]^+$ ), calcd 493.1198 ( $[\text{M} + \text{Na}]^+$ ), 470.1300 ( $[\text{M}]^+$ ,  $\text{M}=\text{C}_{27}\text{H}_{22}\text{N}_2\text{O}_4\text{S}$ ) (**Figure A-32**);  $\lambda_{\text{abs}}$  ( $\epsilon$ ) 579 nm ( $0.3 \times 10^5$ ) (**Figures B-33 and B-34**).

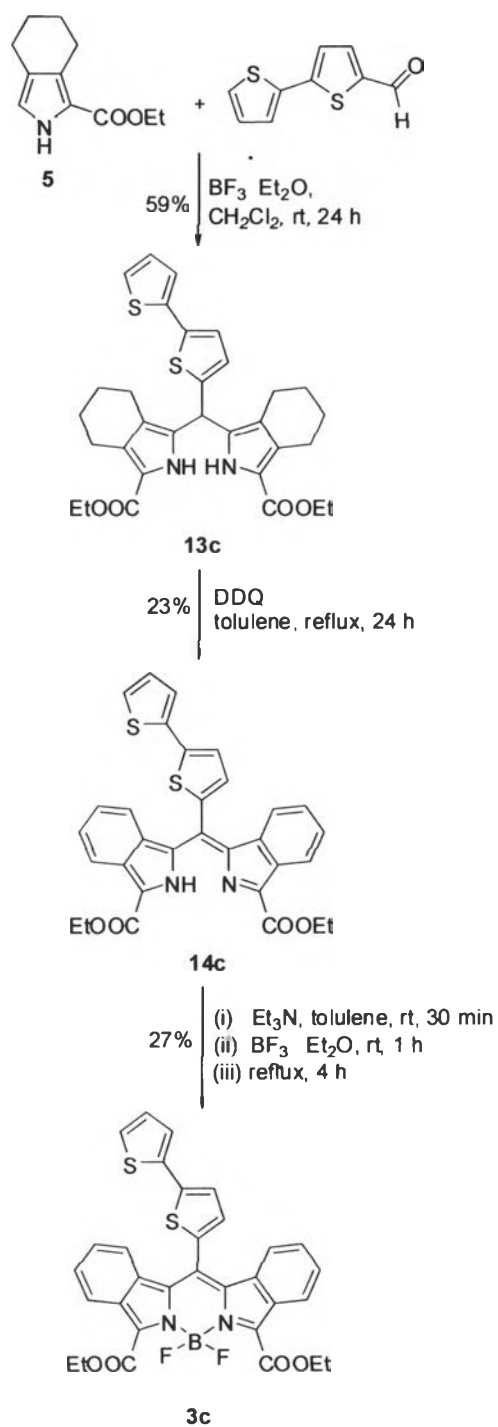
Following a published procedure [13] with slight modification, a solution of dipyrin **14b** (0.147 g, 0.314 mmol) in toluene (20 mL) was treated with  $\text{Et}_3\text{N}$  (0.56 mL, 4.0 mmol) and  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (0.79 mL, 6.4 mmol) at 0°C and the mixture was stirred at room temperature for 30 min. The reaction mixture was then refluxed for additional 6 h. The solution was washed with a 10% aqueous solution of  $\text{NaHCO}_3$  (3×20 mL), brine (1×20 mL) and dried over anhydrous  $\text{Na}_2\text{SO}_4$ . The solvent was evaporated under reduced pressure and purified by a silica column ( $\text{CH}_2\text{Cl}_2$ /hexanes, 4:1) to give a mixture containing BODIPY **3b** as a blue solid. Due to impurity having a very similar  $R_f$  value, **3b** could not be completely purified (~90% purity, 54 mg, 33%).  $^1\text{H-NMR}$ :  $\delta_{\text{H}}$  1.55 (t,  $J = 7.2$  Hz, 6H), 4.61 (q,  $J = 7.2$  Hz, 4H), 6.36 (d,  $J = 8.4$  Hz, 2H), 7.20 (t,  $J = 8.0$  Hz, 2H), 7.27–7.32 (m, 3H), 7.37–7.43 (m, 1H), 7.78 (d,  $J = 4.8$  Hz, 1H), 8.12 (d,  $J = 8.4$  Hz, 2H) (**Figure A-9**);  $^{13}\text{C-NMR}$   $\delta_{\text{c}}$  14.1, 62.3, 121.8, 124.2, 126.8, 128.5, 128.7, 128.9, 130.0, 160.5 (**Figure A-10**); HR-ESI-MS obsd 541.1181 ( $[\text{M} + \text{Na}]^+$ ), calcd





541.1181 ( $[M + Na]^+$ ), 518.1283 ( $[M]^+$ ,  $M=C_{27}H_{21}BF_2N_2O_4S$ ) (Figure A-11);  $\lambda_{abs}$  ( $\mathcal{E}$ ) 656 nm ( $0.7 \times 10^5$ ) (Figures B-13 and B-14);  $\lambda_{em}$  ( $\lambda_{ex}=600$  nm) 676 nm (Figure B-15).

### 3.3.6 Synthesis of compound 3c



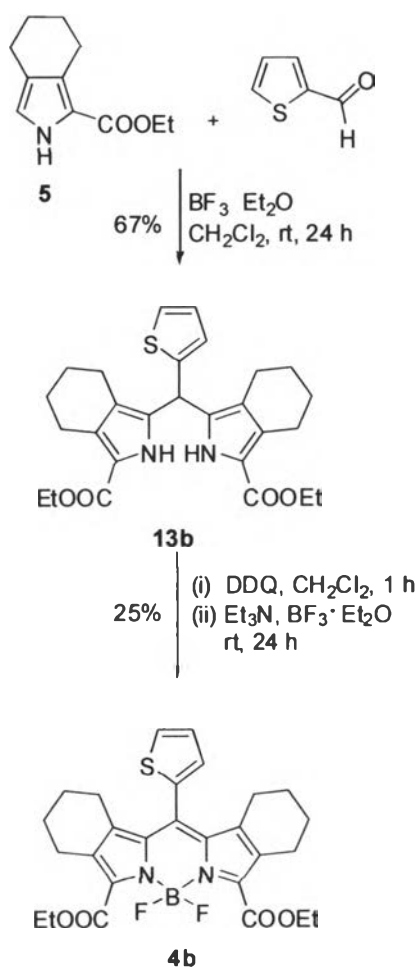
Following a published procedure [73] with slight modification, isoindole **5** (1.000 g, 5.175 mmol) and 2,2'-bithiophene-5-carboxaldehyde (0.503 g, 2.587 mmol) were dissolved in CH<sub>2</sub>Cl<sub>2</sub> (100 mL). BF<sub>3</sub>·Et<sub>2</sub>O (0.065 mL, 0.52 mmol) was added dropwise and the solution was stirred at room temperature for 24 h. The solvent was evaporated under vacuum and the resulting residue was chromatographed on a silica column (CH<sub>2</sub>Cl<sub>2</sub>/hexanes, 4:1) to give **13c** (1.715 g, 59%). <sup>1</sup>H-NMR: δ<sub>H</sub> 1.20–1.26 (m, 6H), 1.68 (s, 8H), 2.39 (d, *J* = 8.8 Hz, 4H), 2.73 (s, 4H), 4.10–4.18 (m, 2H), 4.18–4.26 (m, 2H), 5.61 (s, 1H), 6.60 (d, *J* = 2.0 Hz, 1H), 6.91 (d, *J* = 2.8 Hz, 1H), 6.93–6.98 (m, 1H), 7.00 (s, 1H), 7.16 (d, *J* = 5.2 Hz, 1H), 9.91 (br s, 2H) (**Figure A-23**); <sup>13</sup>C-NMR δ<sub>C</sub> 14.4, 21.2, 23.1, 23.2, 23.3, 24.6, 35.7, 36.6, 59.9, 76.7, 77.0, 77.3, 117.2, 119.6, 123.1, 123.5, 124.2, 126.5, 127.7, 129.1, 130.3, 137.1, 137.2, 142.3, 162.0 (**Figure A-24**); HR-ESI-MS obsd 585.1851 ([M + Na]<sup>+</sup>), calcd 585.1858 ([M + Na]<sup>+</sup>), 562.1960 ([M]<sup>+</sup>, M=C<sub>31</sub>H<sub>30</sub>N<sub>2</sub>O<sub>4</sub>S<sub>2</sub>) (**Figure A-25**); λ<sub>abs</sub> (ε) 289 nm (0.4 × 10<sup>5</sup>) (**Figures B-29 and B-30**).

Following a published procedure [13] with slight modification, a solution of dipyrromethane **13c** (0.494 g, 0.877 mmol) in toluene (30 mL) was heated to 110°C. A solution of DDQ (1.792 g, 7.89 mmol) in toluene (20 mL) was then added and the mixture was refluxed 24 h. After that, the solvent was removed under reduced pressure. The resulting residue was dissolved in ethyl acetate and extracted with a 0.1 M solution of HCl to remove traces of DDQ (3×50 mL), washed once with brine (50 mL) and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. The solvent was removed under reduced pressure and the residue was purified by column chromatography (CH<sub>2</sub>Cl<sub>2</sub>/hexanes, 4:1) to give dibenzo-fused dipyrin **14c** as a purple solid (0.109 g, 23%). <sup>1</sup>H-NMR: δ<sub>H</sub> 1.56 (t, *J* = 7.2 Hz, 6H), 4.57 (q, *J* = 7.2 Hz, 4H), 6.69 (d, *J* = 8.0 Hz, 2H), 7.06–7.15 (m, 3H), 7.19 (d, *J* = 3.2 Hz, 1H), 7.23–7.35 (m, 4H), 7.43 (s, 1H), 8.20 (d, *J* = 8.0 Hz, 2H), (**Figure A-33**); <sup>13</sup>C-NMR δ<sub>C</sub> 14.4, 29.6, 61.3, 122.2, 122.9, 124.6, 125.4, 126.4, 127.4, 128.1, 129.9, 130.6, 131.7, 135.1, 135.3, 135.8, 136.6, 138.4, 140.5, 161.7 (**Figure A-34**); HR-ESI-MS obsd 575.1059 ([M + Na]<sup>+</sup>), calcd 575.1075 ([M + Na]<sup>+</sup>), 552.1177 ([M]<sup>+</sup>, M=C<sub>31</sub>H<sub>24</sub>N<sub>2</sub>O<sub>4</sub>S<sub>2</sub>) (**Figure A-35**); λ<sub>abs</sub> (ε) 589 nm (0.2 × 10<sup>5</sup>) (**Figures B-35 and B-36**).

Following a published procedure [71] with slight modification, a solution of dipyrin **14c** (0.121 g, 0.219 mmol) in toluene (20 mL) was stirred at room temperature. Et<sub>3</sub>N (0.38 mL, 2.7 mmol) was added to the solution. After 30 min. BF<sub>3</sub>·Et<sub>2</sub>O (0.46 mL, 3.6 mmol) was added, and the mixture was stirred was maintained for additional 1h. The reaction mixture was then refluxed for 4 h. After that, the solution was washed with a 10% aqueous solution of NaHCO<sub>3</sub> (20 mL), brine (20 mL)

and dried over anhydrous  $\text{Na}_2\text{SO}_4$ . The solvent was evaporated under reduced pressure and purified by a silica column ( $\text{CH}_2\text{Cl}_2$ /hexanes, 4:1) to give BODIPY **3c** as a blue solid (36.0 mg, 27%).  $^1\text{H-NMR}$ :  $\delta_{\text{H}}$  1.55 (t,  $J = 7.2$  Hz, 6H), 4.61 (q,  $J = 7.2$ , 4H), 6.70 (d,  $J = 8.0$  Hz, 2H), 7.06–7.12 (m, 1H), 7.14–7.27 (m, 3H), 7.28–7.37 (m, 4H), 7.46 (d,  $J = 3.2$  Hz, 1H), 8.13 (d,  $J = 8.0$  Hz, 2H) (**Figure A-12**);  $^{13}\text{C-NMR}$ :  $\delta_{\text{C}}$  14.1, 62.3, 122.0, 124.1, 124.3, 124.9, 125.5, 126.9, 128.0, 128.1, 129.7, 130.0, 131.2, 132.1, 132.7, 134.4, 136.0, 140.8, 141.1, 160.5 (**Figure A-13**); HR-ESI-MS obsd 623.1061 ( $[\text{M} + \text{Na}]^+$ ), calcd 623.1058 ( $[\text{M} + \text{Na}]^+$ ), 600.1160 ( $[\text{M}]^+$ ,  $\text{M} = \text{C}_{31}\text{H}_{23}\text{BF}_2\text{N}_2\text{O}_4\text{S}_2$ ) (**Figure A-14**);  $\lambda_{\text{abs}}$  ( $\mathcal{E}$ ) 658 nm ( $0.5 \times 10^5$ ) (**Figures B-17 and B-18**);  $\lambda_{\text{em}}$  ( $\lambda_{\text{ex}} = 600$  nm) 676 nm (**Figure B-19**).

### 3.3.7 Synthesis of compound 4b



Following a published procedure [13] with slight modification, a solution of DDQ (0.567 g, 2.50 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was added to a solution of **13b** (1.00 g, 2.08 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 mL) at 0°C. After that, Et<sub>3</sub>N (1.74 mL, 12.5 mmol) and BF<sub>3</sub>·Et<sub>2</sub>O (2.64 mL, 20.8 mmol) were added and the solution was stirred at 0°C for 20 min and then at room temperature for overnight. The solution was wash with a 10% aqueous solution of NaHCO<sub>3</sub> (3×20 mL) and brine (1×20mL), and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. The solvent was evaporated under reduced pressure and the resulting residue was purified by column chromatography (ethyl acetate/hexanes; 1:4) to afford BODIPY **4b** (0.271 g, 25%). <sup>1</sup>H-NMR: δ<sub>H</sub> 1.41 (t, *J* = 7.2 Hz, 6H), 1.44–1.50 (m, 4H), 1.60 (d, *J* = 5.6 Hz, 4H), 1.86 (s, 4H), 2.51–2.61 (m, 4H), 4.43 (q, *J* = 7.2 Hz, 4H), 6.94–6.98 (m, 1H), 7.13–7.18 (m, 1H), 7.56 (d, *J* = 4.8 Hz, 1H) (Figure A-15); <sup>13</sup>C-NMR δ<sub>C</sub> 14.07, 14.08, 22.0, 22.3, 22.4, 22.6, 22.9, 23.0, 23.4, 23.5, 29.0, 29.2, 29.3, 29.4, 29.6, 29.7, 31.6, 31.9, 60.6, 61.7, 127.8, 128.0, 128.2, 132.7, 133.9, 134.0, 144.1, 161.4 (Figure A-16); HR-ESI-MS obsd 549.1776 ([M + Na]<sup>+</sup>), calcd 549.1807 ([M + Na]<sup>+</sup>), 526.1909 ([M]<sup>+</sup>, M=C<sub>27</sub>H<sub>29</sub>BF<sub>2</sub>N<sub>2</sub>O<sub>4</sub>S) (Figure A-17); λ<sub>abs</sub> (ε) 439, 556 nm (0.4 × 10<sup>5</sup>) (Figures B-21 and B-22); λ<sub>em</sub> (λ<sub>ex</sub>= 500 nm) 582 nm (Figure B-23).

