

# จุฬาลงกรณ์มหาวิทยาลัย ทุนวิจัย กองทุนรัชดาภิเษกสมโภช

รายงานวิจัย

# การเตรียมชิ้นงานไบโอเซรามิกเพื่อใช้ทางการแพทย์

โดย

สุพัตรา จินาวัฒน์ ดุจฤทัย พงษ์เก่า คะชิมา พรนภา สุจริตวรกุล

ธันวาคม 2549

จุฬาลงกรณ์มหาวิทยาลัย

ทุนวิจัย

กองทุนรัชดาภิเษกสมโภช

รายงานผลการวิจัย

การเตรียมชิ้นงานไบโอเซรามิกเพื่อใช้ทางการแพทย์

โดย

ร.ศ. ดร. สุพัตรา จินาวัฒน์ ดร. ดุจฤทัย พงษ์เก่า คะชิมา ดร. พรนภา สุจริตวรกุล

เดือน ธันวาคม พ.ศ. 2549

Chulalongkorn University

Research Grant

Ratchadaphiseksomphot Endowment

Report on

Preparation of Bioceramic Specimens for Medical Application

By

Associate Professor Dr. Supatra jinawath Dr. Dujreutai Pongkao Kachima Dr. Pornapa Sujaridworakun

December 2006

### Acknowledgements

The authors gratefully acknowledge Ratchadahiseksomphot Endowment, Chulalongkorn University, for granting the financial support to conduct this research to completion and many thanks to all the anonymous people involved that helped making this task possible and with a beautiful result.



### ชื่อโครงการวิจัย การเตรียมชิ้นงานใบโอเซรามิกเพื่อใช้ทางการแพทย์

ชื่อผู้วิจัย ร.ศ. คร. สุพัตรา จินาวัฒน์ คร. คุจฤทัย พงษ์เก่า กะชิมา คร. พรนภา สุจริตวรกุล

เดือนและปีที่ทำวิจัยสำเร็จ ธันวาคม พ.ศ. 2549

### บทคัดย่อ

ไฮครอกซีแอพาไทต์(HA) และ วัสดุเชิงประกอบไตรแกลเซียมฟอสเฟต/ไฮครอกซีแอพาไทต์ (TCP/HA) ถูกสังเกราะห์ขึ้นในห้องปฏิบัติการจากเถ้ากระดูกวัว/ควาย หรือวัสดุเหลือทิ้งจาก อุตสาหกรรมที่เกี่ยวข้อง ทำการขึ้นรูปเป็นชิ้นทคลองชนิดพรุน มีรูปร่างเป็นแท่งสี่เหลี่ยม และเป็นเม็ด แบน (กวามพรุน30-34 % กวามแข็งแรงคัค17 MPa ขนาครูพรุน <5-30μ) เพื่อนำมาทำการทคสอบ ทางการแพทย์ โดยปลูกฝังในสุนัขทคลอง และต่อมาในผู้ป่วยอาสาสมัคร ผลจากการปลูกฝังในสุนัข เป็นเวลา 3 ถึง 6 เดือน ประสบความสำเร็จ และผลในผู้ป่วยก็ประสบความสำเร็จเป็นที่พอใจของผู้ที่ เกี่ยวข้อง ผลดีที่ได้นี้เป็นการพิสูจน์ว่าชิ้นทคลองไฮครอกซีแอพาไทต์ชนิดพรุนนี้ มีสมบัติที่เหมาะสม ในการที่จะเหนี่ยวนำให้เส้นเลือดและเนื้อเยื่อกระดูกเติบโตและเชื่อมต่อผ่านเข้าไปในรูพรุนได้ดี นอกจากนี้ยังได้ทำการทคลองเพื่อหาสภาวะที่เหมาะสม ในการเกลือบชั้นของไฮครอกซีแอพาไทต์บน ผิวโลหะไทเทเนียม เพื่อการใช้งานที่กว้างมากยิ่งขึ้น

### Project Title Preparation of Bioceramic Specimens for Medical Application

Name of the investigators Associate Professor Dr. Supatra Jinawath Dr. Dujreutai Pongkao Kachima Dr. Pornapa Sujaridworakun

Year December 2006

### ABSTRACT

HA (hydroxyapatite,  $Ca_{10}(PO_4)_6(OH)_2$ , Ca/P = 1.67) and composite TCP/HA (Tricalcium phosphate,  $Ca_3(PO_4)_2$ , Ca/P = 1.5) were synthesized in our laboratory from bovine bone ash or the waste from the manufacture of products derived from it and fabricated into porous bar and disc shaped specimens (30-34 v% porosity, flexural strength of 17 MPa, pore diameters  $<5-30\mu$ ) for *in vivo* histological study and clinical trial, repectively. The histological results of the 3- and 6-month *in vivo* tests in canines were positive and successful. The clinical trials using titanium D-cages with HA inserts after 3-6 months were also proved positive, and with a satisfactory result. This also proved our belief that the practical limit of the smallest pore size for blood vessel or bone tissue ingrowth could be smaller than 100 $\mu$  (macropore) in the case of resorbable HA ceramic with interconnected micro-to-mesopores (estimated as <1, 5-30  $\mu$ ). To explore into a broader application, a coating of biological HA layer on Ti alloy implant was also studied and the results were discussed.

### Contents

		Page
Ac	knowledgements	
บท	เค้ดย่อ	
AF	BSTRACT	
Co	ontents	i
Lis	st of Tables	iii
Lis	st of Figures	iv
1.	Background of the research proposal	1
2.	Introduction	1
	Objective	2
	Scope	2
Pa	rt A	
	Experiment on the application of porous HA and TCP/HA bars and	3
	disc specimens.	
	1. Materials and method	3
	1. sample preparation	3
	2. In vivo experiment and clinical trial	3
	2. Results and discussion	4
	1. Starting materials	4
	2. Porous HA and TCP/HA specimens	4
	3. Results of <i>in vivo</i> test and clinical trial	6
	3. Conclusion	10
Pa	rt B	
	Experiment on the application of the coating of calcium phosphates	11
	on Ti alloy.	
	1. Materials and method	11
	1. Surface treatment and incubation:	
	1.1 Treatment by soaking in $H_2O_2$ solution and followed	11
	by incubating in R-SBF solution	
	1.2 Treatment by autoclaving in NaOH solution and followed	12
	by incubation in R-SBF solution	
	1.3 Treatment by soaking / autoclaving in $H_2O_2$ solution	12
	and followed by incubating in R-SBF solution	
	2. Results and discussion	13
	2.1 Treatment by soaking in $H_2O_2$ solution and followed	13
	by incubating in R-SBF solution (preliminary experiment)	
	2.2 Treatment by autoclaving in NaOH solution and followed	14
	by incubating in R-SBF solution	
	2.3 Treatment by soaking / autoclaving in $H_2O_2$ solution and followed	16
	by incubating in R-SBF solution	-
	3. Conclusion	17
		-

4.	References		18
5.	Presentations		19
6.	Appendix 1	Schematic diagram of experiment set up	
		for electroplating on titanium substrate	
	Appendix 2	Formula of R-SBF solution	21
	Appendix 3	Poster presented at A poster presentation at 50th Birthday Anniversary of H.R.H. Princess	22
		Maha Chakri Sirindthorn, U.S Thai Symposium on Biomedical Engineering in Thailand,	
		Chulalongkorn University, Bangkok, Dec.12th-15th, 2005.	
	Appendix 4	Poster presented at 6 <sup>th</sup> ASIAN BIOCERAMIC	23
		SYMPOSIUM (ABC 2006), Sofitel Central Plaza,	
		Bangkok, Thailand, Nov. 7-10, 2006.	



### List of Tables

Part A	Page
Table 1. Properties of the starting materials	4
Table 2. Properties of the sintered specimens (bars)	5

Part A



### List of Figures

Part A		Page	
Fig. 1A	Flow chart of the porous HA, TCP/HA specimens forming process		
Fig. 2A	XRD of the sintered specimens (bars), at 1200°C, 2h		
Fig. 3A	FT-IR spectrograms of the sintered specimens (bars), at 1200 °C, 2h		
Fig. 4A	A SEM micrographs (JEOL: JDX-8030, CuK $\alpha$ , $\lambda = 1.5405$ A) of the		
	fractured surface of the HA sample (bars)		
Fig. 5A	SEM micrograph of the cross section of the HA bar implanted	7	
	in canine's tibia for 6 m (a), and digital photograph of		
	sintered porous HA bar and discs (b)		
Fig. 6A	Optical micrographs of the thin sections	7	
Fig. 7A	Photograph of the D-cage devices	8	
Fig. 8A	X-ray radiograms (a) after 3-m and (b) after 6-m implantations	8	
Fig. 9A	CT scan after 6-m implantation	9	
Part B			
Fig. 1B	Flow chart for the surface treatment of titanium screw, soaking	11	
	in H <sub>2</sub> O <sub>2</sub> solution and incubating in R-SBF solution		
Fig. 2B	Hydrothermal treatment of Ti screw in NaOH	12	
	and incubating in R-SBF solution		
Fig. 3B	$H_2O_2$ treatment of Ti screw surface	12	
	and incubating in R-SBF solution		
Fig. 4B	Titanium pedicular screw for orthopedic application		
Fig. 5B	XRD patterns of the $H_2O_2$ treated pedicular screws at various 1		
concentrations of $H_2O_2$ (10%, 17% and 35%) after incubation			
	in R-SBF for 72 h	1 5	
F1g. 6B	Photographs of the specimens after autoclaving in NaOH	15	
E' 7D	solution	1.5	
F1g. /B	ARDs of the specimens, autoclaved in NaOH	15	
	and incubated in R-SBF solution	16	
F1g. 8B	Photographs of the specimens after soaking in $H_2O_2$	10	
	and incubating in K-SBF solution	16	
F1g. 9B	in LLO, and insulating in D. SDE solution	10	
Ei~ 10D	III $H_2O_2$ and incubating III K-SDF solution VDDs of specimens often tracting with U O	17	
rig. IUB	and incubating in <b>D</b> SDE solution	1/	
	and incubating in K-SDF solution		

### **Preparation of Bioceramic Specimens for Medical Application**

#### 1. Background of the research proposal

Bioceramics have long been regarded as a potential biomaterial and become increasingly employed in various medical applications. Among the well known bioceramics, calcium phosphates, i.e. hydroxyapatite (HA) and tricalcium phosphates (TCP) have been highly recommended for bone tissue induction, and hence widely applied in orthopedic surgery. There are quite a number of hydroxyapatite products, commercially available in various forms, i.e. granules, dense or porous pieces, injectable suspension, and as a coating layer on shaped metal inserts. In this project, it is required by the surgeons, our collaborators<sup>1</sup>, that the hydroxyapatite should be in (1) the form of porous disc about 1.0 cm in diameter and (2) as a coating layer on Ti alloy.

Previously our research group had successfully synthesized a range of calcium phosphates (Ca/P molar ratios 0.5-2) from chemicals, bovine bone and by-product of bone-gelatin manufacture, and tried several fabrication processes using our synthesized materials<sup>2-10</sup>. The HA produced in our laboratory was in the forms of powder, granule and porous bars. They came out with satisfaction in terms of morphology, chemical purity and mechanical performance. However, in clinical application the shape and size of the HA ceramic to be implanted have to conform with the task of the surgeon, therefore this project is regarded as the extension of the previous one. We highly expect that this collaboration will reveal us more information on the histology of our products and lead to the possibility of independence on import, as well as to further development in the field of biomaterials. Nowadays, we all realized that without the collaboration from researchers in the related fields, most of the research outcomes can never be further developed. The collaboration can open the closed end of the channel.

#### 2. Introduction

Calcium phosphate powders with Ca/P molar ratios of 0.5 (MCPM,  $Ca(H_2PO_4)_2.H_2O)$  to 2.0 (TTCP,  $Ca_4P_2O_9$ ) were successfully synthesized in our laboratory from bovine bone ash or the waste from the manufacture of products derived from it. The most popular calcium phosphates are HA (hydroxyapatite,  $Ca_{10}(PO_4)_6(OH)_2$ ) Ca/P = 1.67) and  $\beta$ -TCP ( $\beta$ -tricalcium phosphate,  $Ca_3(PO_4)_2$ , Ca/P = 1.5). Typically, the applications of porous calcium phosphates are as bone graft, augmentation, infilling and preforms to be shaped during surgery, and practically, the powders have to be fabricated into a compact of designed shape and porosity i.e. a granule, a porous bar of appreciable strength or as a composite with polymer matrix. During an appropriate retention time, an ideal porous calcium phosphate implant should be resorbable, hence conduct bone tissues at its expense. Consequently, the practical limit of the smallest pore size for blood vessel or bone tissue ingrowth can be smaller than  $100 \mu$  (macropore). This point is vital for the application of porous calcium phosphates with micro-to mesoporosity since an appreciable strength can be achieved with this range of porosity. Accordingly, their preforms can be shaped and retained in place long enough to conduct bone tissues effectively. To verify this belief, porous HA bars (30-34 v% porosity, flexural strength of 17 MPa, pore diameters  $<5-30\mu$ ) were implanted in canines' tibias for the periods of 3-6 months for histological study, and later clinical trials were performed using a D-cage filled with a porous HA disc on voluntary patients whose CT scan and X-ray results would be followed up after a period of 6 months.

### Objective

The objective is to prepare HA ceramics in the forms of porous bar and disc, and as a coating on metal implant to support a team of Chulalongkorn Hospital orthopedic surgeons with the expectation that the information on histology systematically studied by our medical collaborators, will clinically justify the properties of our products, and lead to further development.

### Duration

1 year, from May 2005- March 2006.

#### Scope

For producing porous HA bar and disc, the procedure reported in reference 2, 3 and 8 will be followed. The process for producing the coating will be by electroless plating. A histological study using porous HA bars will be performed on canines for the periods of 3 and 6 m. Later a clinical trial on a number of voluntary patients using porous discs will be performed and followed up by X-ray radiography after 3 and 6 m implantations.

For simplicity, the report will be split into 2 parts:

Part A: Experiment on the application of porous HA and TCP/HA bars and disc specimens.

Part B: Experiment on the application of the coating of calcium phosphates on Ti alloy.



# Part A: Experiment on the application of porous HA and TCP/HA bars and disc specimens.

### 1. Materials and method

### **Materials**

Dicalcium phosphate, dicalcium phosphate anhydrous(DCPD/DCPA) synthesized from the by-product of bone-gelatin industry, calcium phosphate glass,  $Ca(PO_3)_{2}$ , and chemical grade  $CaCO_3$  powder were used as the starting materials for the fabrication of porous HA and TCP/HA specimens. The experimental procedure is presented as a flow chart in Fig. 1.

### **Methods**

**1.** *Sample preparation:* The experimental procedure is presented as a flow chart in Fig. 1A.

**2.** *In vivo experiment and clinical trial:* The porous HA bars were implanted in canines' tibias for the periods of 3 and 6 months for histological study, and later clinical trials were performed using D-cages (Ti alloy) filled with porous HA discs on a number of voluntary patients whose CT scan and X-ray results were to be followed up after the periods of 3 and 6 months.



Fig.1A Flow chart of the porous HA, TCP/HA specimens forming process

### 2. Results and discussion

### 1. Starting materials

Chemical compositions (ICP) and particle sizes (Shimadzu SA-CP2) of the synthesized DCPD and DCPA are shown in Table 1.

	DCPD	DCPA	CaCO <sub>3</sub>
MgO (wt%)	0.02	0.04	Purity >99%
MnO (wt%)	< 0.01	< 0.01	Fluka
Fe2O3(wt%)	0.14	0.14	
Zn (ppm)	73	52	
Cu (ppm)	20	24	
Ca/P molar ratio	1.13	1.21	
*Median particle size (µm)	13	7	24

Table 1 Properties of the starting materials

The chemical compositions confirm the high purity and nonstoichiometry of the starting materials. In addition, the concentrations of heavy metals (ppm) are as follows: Cd = 0.5, Pb<5, Hg<1, As<0.5, Ni = 2. The contents of all these heavy metals well pass the specification of American Standard Test Methods for Composition of Ceramic Hydroxyapatite for Surgical Implants (ASTM: F118-88).

### 2. Porous HA and TCP/HA specimens

The phase analysis by XRD (Philips: PW1730/10) and by FT-IR (Perkin Elmer 1760X) of the sintered specimens (bars) is illustrated in Fig. 2A and 3A, respectively.



Fig. 2A XRD of the sintered specimens (bars), at 1200°C, 2h



Fig. 3A FT-IR spectrograms of the sintered specimens (bars), at 1200 °C, 2h

Temp 1200 °C, 2h 3 °C/min	DCPD starting material		DCPA starting material	
29/222	0.5+wt% glass 50MPa	84MPa	0.5+wt% glass 50MPa	8 4MPa
Bulk density, g/cm <sup>3</sup>	1.72	1.67	1.87	1.86
Apparent porosity, vol%	46.6	45.15	36.17	37.35
Flexural strength, MPa	8.05±0.51	9.78±0.20	15.84±0.82	17.51±1.77
Phase	ΗΑ+β-ΤСΡ	HA	ΗΑ+β-ΤСΡ	HA

 Table 2 Properties of the sintered specimens (bars)

The summary of all the analytical data of the sintered specimens is tabulated in Table 2. It is found that small amount of calcium phosphate glass enhances the formation of  $\beta$ -TCP, and single phase HA could be obtained at the sintering temperature of 1200°C.

The SEM micrographs showing the morphology of the porous specimens (bars) are presented in Fig. 4A. A range of mesopores ( $\sim$ 5-30 µ) and that of micropores (<1 µ) are visible and all types of pores are clearly interconnected. This type of morphology is highly resorbable and should be ideal for bone tissue ingrowth.



(a) Mesopores and micropores



(a) Mesopores and micropores (low magnification)



(b) Solid part enlarged, showing micropores

Fig. 4A SEM micrographs (JEOL: JDX-8030, CuK $\alpha$ ,  $\lambda$  =1.5405A) of the fractured surface of the HA sample (bars)

### 3. Results of in vivo test and clinical trial

Fig. 5A is the SEM micrograph of the cross section of the HA bar implanted in a canine's tibia for 6 months. The micrograph reveals a very high compatibility between the implant and the surrounding bone tissue confirmed by the absence of noticeable gaps at the interface. The visible white spots on the surface indicate resorption of the implant over time.



Fig. 5A SEM micrograph of the cross section of the HA bar implanted in canine's tibia for 6 m (a), and digital photograph of sintered porous HA bar and discs (b)

Fig. 6A is the optical micrographs of the thin sections (dyed in toluidine) of the above implant under histological study, in collaboration with the team of orthopedic surgeons of Dr. Piboon Ithiraviwong, Chulalongkorn Hospital and the team of pathologists of Dr. Somboon Thamatakerngkij, Siriraj Hospital, showing old bone tissue in deep red color and new bone tissue (pink) growing into the implant, with an increasing amount over time.



Optical micrographs of implants in canine's tibia, 3 and 6 m



Fig. 6A Optical micrographs of the thin sections

Fig. 7A is the digital photograph of the D-cage devices filled with porous HA discs to be inserted between the cervical bones in patient's neck. The surgery is known as '*anterior cervical disectomy and fusion (ACDF*)'.



Fig. 7A Photograph of the D-cage devices

A clinical trial in collaboration with the team of orthopedic surgeons of Dr. Prakit Tienboon, Chulalongkorn Hospital, had been performed on about 50 cases of voluntary patients so far, and the results after 3 and 6-month implantation are shown in Fig 8A and 9A, respectively.



(a)

(b) bending action

Fig. 8A X-ray radiograms (a) after 3-m and (b) after 6-m implantations



Fig. 9A CT scan after 6-m implantation

The results from X-ray radiography, X-ray radiograms and CT scan (CAT scan) after 3- and 6-m implantation (Fig. 8A and 9A) reveal that the titanium devices have been tightly fixed in place by the aid of the new bone tissues conducted by the porous ceramics. There is no shadow area visible at the interface between the device and the adjacent cortical bone and no any inflammation detected. All the patients heal rapidly and can turn their necks to an angle of almost 180°, moreover the operation can be done in 1 day.

In this application, the porous HA was used as a resorbable bone graft material. It has been well known that there are 4 techniques for bone graft.

1. *Autograft:* using the patient's own bone. This is limited by the size of the wound and long healing time.

2. *Allograft:* using human bone from bone bank. There is a drawback from genetic diseases.

3. *Xenograft:* using bones from other species. The problems encountered are genetic diseases and compatibility.

4. *Synthetic materials:* Biomaterials such as bioceramics, biopolymers and composites of the two, and novel metals come into this one. In the exception of the inert metals, the drawbacks from this group of materials are low compatibility and resorption when compared to the above choices. Nevertheless, this discrepancy has been improved tremendously over the past decade by numerous researchers. Many new materials (monolith and composite) having directed microstructure and function have been introduced. Old concepts of biomaterials have been refined and many new ones proposed.

### 3. Conclusion

The histological results of the 3- and 6-month *in vivo* tests in canines are positive and successful. The clinical trials using titanium D-cages with about 100 pieces of the porous HA specimens on a number of voluntary patients have also been performed and the results of 3- and 6-month implantations in 13 cases have been followed up as scheduled. All the D-cages are firmly fixed so there is no dislodging under neck movement. Hence all the 13 cases prove positive, and with satisfactory results. This also proves our belief that the practical limit of the smallest pore size for blood vessel or bone tissue ingrowth can be smaller than 100  $\mu$  (macropore) in the case of resorbable HA ceramic with interconnected micro-to-mesopores (estimated as <1, 5-30  $\mu$ ).



# Part B: Experiment on the application of the coating of calcium phosphates on Ti alloy.

### 1. Materials and method

### **Materials**

Chemical grade dicalcium phosphate, DCPD, Fluka. Titanium screws (courtesy of Dr. Prakit Tienboon)

### Methods (electroless plating)

### 1. Surface treatment and incubation:

The experimental procedures for a surface treatment of Ti substrates were conducted in either  $H_2O_2$  or NaOH solutions prior to calcium phosphate coating by *biomimetic* process, i.e. soaking in **R-SBF solution**<sup>11</sup> which is regarded as an *electroless plating* process. The procedures are presented as flow charts in Fig. 1B-3B.

# 1.1 Treatment by soaking in $H_2O_2$ solution and followed by incubating in *R-SBF* solution



Fig.1B Flow chart for the surface treatment of titanium screw, soaking in  $H_2O_2$  solution and incubating in R-SBF solution

# 1.2 Treatment by autoclaving in NaOH solution and followed by incubating in R-SBF solution



Fig. 2B Hydrothermal treatment of Ti screw in NaOH and incubating in R-SBF solution

1.3 Treatment by soaking / autoclaving in  $H_2O_2$  solution and followed by incubating in R-SBF solution





### 2. Results and discussion

### Improving the bioactivity of titanium screw by electroless plating

Titanium screw used in orthopedic application is so called as "pedicular screw". It is treated in either  $H_2O_2$  solution or  $H_2O_2$  solution with fluoride ion (from NaF) addition. Pedicular screw is a complex shape screw made from an alloy of titanium metal used in the fixing of human spinal bone as shown in Fig. 3B.



Fig.4B Titanium pedicular screw for orthopedic application

# 2.1 Treatment by soaking in $H_2O_2$ solution and followed by incubating in R-SBF solution (preliminary experiment)

Generally, titanium is classified as a bio-inert metal which is not able to induce bone formation. Hence an appropriate surface treatment technique is employed to modify its surface property. However, due to its complex shape, the surface treatment in solution should be a right way to bring about the expected outcome, i.e. surface treatment in a strong oxidizing solution of  $H_2O_2$  would allow the development of a homogeneous titanium dioxide (TiO<sub>2</sub>) layer on the surface of titanium metal. This surface treatment was reported<sup>12</sup> to be a means to improve the bioactivity of titanium since the surface TiO<sub>2</sub> in body fluid (R-SBF solution) would act as nucleation site for HA.

The result from our preliminary experiment showed that  $H_2O_2$  treated titanium could induce the apatite formation after incubation in R-SBF (Revised-Simulated Body Fluid) for 72 hours as shown in the XRD patterns of the screw surface Fig. 5B.



Fig. 5B XRD patterns of the  $H_2O_2$  treated pedicular screws at various concentrations of  $H_2O_2$  (10%, 17% and 35%) after incubation in R-SBF for 72 h

From Fig. 5, it can be observed that apatite (HA) could form on the surface of 35% H<sub>2</sub>O<sub>2</sub> treated titanium screw after incubation in R-SBF for 72 hours. However, we can neither observe the peak of apatite on 17% and 10% H<sub>2</sub>O<sub>2</sub>-treated Ti screws nor on the ones treated with F<sup>-</sup> added H<sub>2</sub>O<sub>2</sub>. Therefore fluoride ion did not play any role to induce the apatite formation on titanium screw.

From the literature<sup>12</sup>, hydrothermal treatment has been reported as another means to enhance the surface treatment of substrate so the content of apatite on the Ti surface is expected to increase by hydrothermal treatment in an autoclave.

# 2.2 Treatment by autoclaving in NaOH solution and followed by incubating in R-SBF solution

This method is aimed to improve the bioactivity of titanium screw by hydrothermal process in homogeneous solution

#### Results

Physical appearance:

As-received screw, shiny. After soaking in NaOH, dull. After incubating in R-SBF solution, dull. The

photographs of the specimens are shown in Fig. 6B.



Fig. 6B. Photographs of the specimens after autoclaving in NaOH solution

Phase analysis by XRD of the surface of the specimens(cut into 3 pieces) after autoclaving in NaOH and incubating in R-SBF solution are shown in Fig. 7B.



Fig. 7B XRDs of the specimens, autoclaved in NaOH and incubated in R-SBF solution

No HA phase  $(2\theta \sim 32^{\circ})$  was detected in Fig. 5B. Therefore, the next experiment was performed by further treatment of Ti screw surface by either soaking or autoclaving in 35 % H<sub>2</sub>O<sub>2</sub> solution, and is schematically presented in Fig. 3B.

# 2.3 Treatment by soaking / autoclaving in $H_2O_2$ solution and followed by incubating in R-SBF solution

### Results (soaking in H<sub>2</sub>O<sub>2</sub>)

Physical appearance: As-received screw, shiny. After soaking in H<sub>2</sub>O<sub>2</sub>, rusty brown.

After firing , dark blue.

After incubating in R-SBF solution, dark blue.

The photographs of the specimens after soaking in  $H_2O_2$  and incubating in R-SBF solution are shown in Fig. 8B.



Fig. 8B Photographs of the specimens after soaking in  $H_2O_2$ and incubating in R-SBF solution

### **Results** (autoclaving in H<sub>2</sub>O<sub>2</sub>)

Physical appearance: As-received screw, shiny. After autoclaving in H<sub>2</sub>O<sub>2</sub>, black.

After incubating in R-SBF solution, black.

The photographs of the specimens after autoclaving in  $H_2O_2$  and incubating in R-SBF solution are shown in Fig. 9B.



Fig. 9B Photograph of the specimens after autoclaving in  $H_2O_2$  and incubating in R-SBF solution



Fig. 10B XRDs of specimens after treating with  $H_2O_2$  and incubating in R-SBF solution

It was found that there was HA (Apatite) layer detected around  $2\theta = 31.77-32.90^{\circ}$ . Therefore, the attempt to get an HA coating on the Ti screw surface by biomimetic coating has been accomplished within 1-year duration of the project. Moreover, there is some information obtained from this experiment that will be of help in pursuing further experiment (on both biomimetic and electroplating) and reproducibility.

#### 3. Conclusion

The results from XRD showed that 35%  $H_2O_2$ -treated titanium screws, 60°C, for 24 hour by hydrothermal process could induce the apatite phase as clearly shown in the XRD pattern (Fig.10 B) at 2 theta = 32°. This result implied that the hydrothermal process could replace heat treatment at 600°C which was reported by Kokubo group<sup>13, 14</sup>. However, the hydrothermal process in  $H_2O_2$  was too strong and it could corrode the autoclave equipment. Moreover, the color of Ti screw was also changed to the dull black color which might be from the oxidizing power of  $H_2O_2$  solution.

### Future suggestion

To pursue a success in coating, the right condition for Ti surface treatment has to be found and the following experiments should be tried:

- 1. Using new chemical additives to enhance the formation of biological HA on Ti surface.
- 2. Electroplating method on the Ti substrate using calcium phosphate solution.

We also would like to inform that, due to the time limit, the coating by electroplating was not included in this research.

#### References

1. Tianboon, P. et al., Strength effect between the hydroxyapatite coating on pedicular screw and vertebral body-pedicular bone integration, a research proposal granted financial support from Ratchadapiseksomphot endowment in 2005.

2. Lorprayoon, C., and Jinawath, S., A comparative study on hydroxyapatite from different origins. Proc. 1<sup>st</sup> International Symposium on Apatite, Tokyo, Japan, Vol.1 (1992).

3. Jinawath, S., and Trakarnvichit, S., Preparation of dicalcium phosphate dihydrate from cattle bone. Proc. 2<sup>nd</sup>International Symposium on Apatite, Mishima, Japan,Vol.2 (1997), 48-52.

4. Jinawath, S., and Ratanachan, S., Synthesis of tetracalcium phosphate (TTCP). Proc. 3 <sup>rd</sup> Far-Eastern Symposium on Biomedical Materials, Chengdu, China (1997), 148-149.

5. Jinawath, S., Pongkao, D., Suchanek, W., and Yoshimura, M., Hydrothermal synthesis of monetite and hydroxyapatite from monocalcium phosphate monohydrate, J. Inorganic Materials 3 (2001) 997-1001.

6. Jinawath, S., Pongkao, D., and Yoshimura, M., Hydrothermal systhesis of hydroxyapatite from natural source, J. Mats. Sc.: Materials in Medicine 13 (2002) 491-494.

7. Jinawath, S., Pholchai, D., and Yoshimura, Y., Low-temperature, hydrothermal transformation of aragonite to hydroxyapatite, J. Materials Science & Engineering C 22 (2002) 35-39.

8. Jinawath, S., and Sujaridworakun, P., Fabrication of porous calcium phosphates, J. Materials Science & Engineering C 22 (2002) 41-46.

9. Dujreutai PONGKAO, Supatra JINAWATH, and Masahiro YOSHIMURA: Characterization and Stability of the New Biocompatible Monetite Whiskers Derived from Bio-bearing Calcium Phosphate (BCP); to be submitted

10. Jinawath, S., Hengst, M., and Heimann, R.B., Plasma-sprayed DCPD/CaCO<sub>3</sub> coating on Ti6Al4V-substrate, J. Sci. Res. Chula. Univ., Vol.29, No. 1(2004), pp 33-44.

11. Kokubo, T. and Takadama, H., How useful is SBF in predicting *in vivo* bone bioactivity, J. Biomaterials 27 (2006), 2907-2915.

12. NAKAGAWA, M., ZHANG, L., UDOH, K., MATSUYA, S., ISHIKAWA, K., Effects of hydrothermal treatment with CaCl<sub>2</sub> solution on surface property and cell response of titanium implants, J. Mats. Sc.: Materials in Medicine 16 (2005) 985-991.

13. Kim, H.M., Miyaji, F., Kokubo, T., and Nakamura, T., Effect of heat treatment on apatite-forming ability of Ti metal induced by alkali treatment, Journal of Materials Science : Materials in medicine 8(1997) 341-347

14. Nishiguchi, S., Nakamura, T., Kobayashi, M., Kim, H.M., Miyaji, F., and Kokubo, T., The effect of heat treatment on bone-bonding ability of alkali-treated titanium, Biomaterials 20 (1999) 491-500.

### **Presentations**

1. Jinawath, S., Kashima P.D., Sujaridworakun, P., Jaruwungsunti, N., Tienboon, P., and Thamatakerngkit, S., Porous calcium phosphates: In-vivo and clinical studies (I). A poster presentation at 50th Birthday Anniversary of H.R.H. Princess Maha Chakri Sirindthorn, U.S.-Thai Symposium on Biomedical Engineering in Thailand, Chulalongkorn University, Bangkok, Dec.12th-15th, 2005.

2. Sujaridworakun, P., Kashima P. D., Tienboon, Jaruwungsunti, N., Itiravivong, P., Thamatakerngkit, S., and Jinawath, S., Porous calcium phosphates: In-vivo and clinical studies (II). A poster presentation at 6<sup>th</sup> ASIAN BIOCERAMIC SYMPOSIUM (ABC 2006), Sofitel Central Plaza, Bangkok, Thailand, Nov. 7-10, 2006.



### Appendix 1

The experimental set-up for electroplating cell.



Schematic diagram of experiment set up for electroplating on titanium substrate.

### Appendix 2

The chemical amount and their purities for preparation of 1L of R-SBF (Revised-Simulated Body Fluid).

Reagent	Purities	Chemical Amount for preparation of R-SBF
NaCl	99.5 %	5.403 g
NaHCO <sub>3</sub>	99.7 %	0.736 g
Na <sub>2</sub> CO <sub>3</sub>	99.9 %	2.036 g
KCl	99.5 %	0.225 g
K <sub>2</sub> HPO <sub>4</sub> •3H <sub>2</sub> O	99.0 %	0.238 g
MgCl <sub>2</sub> •H <sub>2</sub> O	98.0 %	0.311 g
HEPES*	99.5 %	11.928 g
CaCl <sub>2</sub>	94.0 %	0.293 g
Na <sub>2</sub> SO <sub>4</sub>	99.0 %	0.072 g
1M-NaOH	1116/2018	1.5 mL

### **Formula of R-SBF solution**

HEPES : [4-(2-Hydroxyethyl)piperazine-1-ethanesulfonic acid]



<sup>\*\*</sup> HEPES = 2-(4-(2-hydroxyethyl0-1-piperazinyl)ethane sulfonic acid)

## Porous Calcium Phosphates: In-vivo and Clinical studies

P. Sujaridworakun (1) D. Kashima Pongkao (1) P. Tienboon(2), N. Jaruwungsunti(2), P. Itiravivong(2), S. Thematekerngkti(2) and S. Jineweth(1)

<sup>27</sup> Research Unit of Advanced Ceramics, Department of Materials Science, Faculty of Science, Chulalongkom University <sup>27</sup> Department of Orthopedic Surgery, Faculty of Medicine, Chulalongkom University, Bangkok, Thailand <sup>30</sup> Department of Forensic Medicine, Faculty of Medicine, Striraj, Hospital, Mahidol University, Bangkok, Thailand

**Trunt**: HA and HA/TCP composite were synthesized from DCPD/DCPA and CaCO<sub>p</sub> shaped and sintered at 1200°C to obtain porous nens. Chemical compositions of the materials involved were determined by ICP, phase analysis by XRD, and morphology by SEM. The porous nens having 37 vol% porosity composed of interconnected micro (< 5 µm) to meso (>20 µm) pores and a flexural strength of 17 MPa, were need in a canine for 3 to 6 months. Histological study of the implant was performed and the result showed that there were new bone tissues rated into the implants which still survived after 6 months. Also some resorption of the implants was observed. The results of clinical trial on uts' anterior cervical neck bones after 6 month implantation, followed up by x-ray and CT scan radiography proved positive.

-----

تشلتا

### terials and Methods



### HA/TCP specimens

### Results of histological study (In Vivo)

The porous HA specimen implanted in the canine tibla



regraph of canine bone tissues indicated that there was no gap between the alcrum phosphate spectrum after implantation in the canine tibus for 6 and there was some notionable resorretion solaries in the implant.



we tissue was dyed in taluiding

nicrographs showed that the generation of new bone (paler color) livnes was alle and more clearly detected in the 6-month specimen, interpretenting into the of specimen and interconnecting with the old tissues (deep red color). The m showed a good, howe conduction and biocompatibility.

#### Specimens for In Vivo and Clinical studies







D-cage substitute for human cervical neck bone. Porous HA artificial bone used instead of bone graft from patient's hip bone.

### Results of Clinical trial

Ti Cervical D-cage was used as a replacement of the patient's anterior cervical neck bone





The CT scan and x-ray results of clinical trial after 6-m implantation showed no inflammation ad newly conducted bone itasues had grown into the gaps. Moreover, the patients recovered ery quickly and could mixe their necks almost 180%

**Conclusion :** Porous HA and HA/TCP ceramics with flexural strength and porosity ranging from 9-17 MPa, 45-37 vol% and 8-15 MPa, 46-36 vol%, respectively were successfully produced in our laboratory. The results of in-vivo test on animal and the clinical trial on patients' anterior cervical neck bones (13 cases , 24 pieces of ceramics) using porous HA specimens proved positive.

the collaborations from the team of orthopedic surgeons from Chuldongkorn Hospital and the team of pathologists from Sinraj Hospital are much appreciated.

run provinces (American Tel 1002/18-555) (DE-018-555) (DE-018-555) (Remain otherses) (supporte (Remaining the

# Porous D. Pongle

### **Porous Calcium Phosphates: Fabrication, in-vivo and clinical studies**

D. Pongkao Kashima, P. Sujaridworakun, P. Tienboon<sup>\*</sup>, N. Jaruwungsunti<sup>\*</sup>, S. Thamatakerngkit<sup>\*\*</sup> and S.Jinawath Research Unit of Advanced Ceramics, Department of Materials Science, Faculty of Science, Chulalongkorn University \* Department of Orthopedic, Chulalongkorn Hospital.

\*\*Department of Forensic Medicine, Siriraj Hospital, Bangkok, Thailand

**ract**: High purity DCPD and DCPA were synthesized from bovine bone ash or the waste from the manufacture of acts derived from it. Then DCPD and DCPA were each mixed with  $CaCO_3$  in a stoichiometric molar ratio to form HA A/ $\beta$ -TCP composite. After pressing into shaped specimens and sintering at 1200°C, the resulted specimens became as HA or HA/ $\beta$ -TCP depending on the employed conditions. Chemical composition of the materials involved was mined by ICP, phase analysis by XRD and FT-IR, and morphology by SEM. The porous specimens having 32-37 is porosity composed of interconnected micro (< 5  $\mu$ m) to meso (>20  $\mu$ m) pores and a flexural strength of 17 MPa, were anted in a canine for 3 to 6 months. Histology study of the implant was performed and the result showed that there are bone tissues penetrated into the implants which still survived after 6 months. Also some resorption of the ants was observed.

ical application : D-cage was used as a replacement of the patient's anterior cervical neck bone

Ti alloy D- ring



ter 3 months

### In vivo study and histology test

SEM micrograph indicated that there was no gap between the porous calcium phosphate specimen after implantation in the canine tibia for 6 months, and there was some noticeable resorption shown in the implant.



**Porous HA specimen** 



The generation of new bone (paler color) tissues was appreciable and more clearly detected in the 6-month specimen, interpenetrating into the implanted specimen and interconnecting with the old tissues (deep red color). The specimens showed a good bone conduction and biocompatibility.

rch Unit of Advanced Ceramics, Tel: 02-218-5554, 02-218-5548, 02-218-5552, E-mail address : supatra.j@chula.ac.th